# NATURAL OF PRODUCTS

## Potential Antiosteoporotic Natural Product Lead Compounds That Inhibit $17\beta$ -Hydroxysteroid Dehydrogenase Type 2

Anna Vuorinen,<sup>†,#</sup> Roger T. Engeli,<sup>†,#</sup> Susanne Leugger,<sup>†</sup> Fabio Bachmann,<sup>†</sup> Muhammad Akram,<sup>‡</sup> Atanas G. Atanasov,<sup>§,⊥</sup> Birgit Waltenberger,<sup>||</sup> Veronika Temml,<sup>||</sup> Hermann Stuppner,<sup>||</sup> Liselotte Krenn,<sup>§</sup> Sylvin B. Ateba,<sup>¶</sup> Dieudonné Njamen,<sup>¶</sup> Rohan A. Davis,<sup>□</sup> Alex Odermatt,<sup>\*,†</sup> and Daniela Schuster<sup>\*,‡</sup>

<sup>†</sup>Division of Molecular & Systems Toxicology, University of Basel, Klingelbergstraße 50, 4056 Basel, Switzerland

<sup>‡</sup>Computer-Aided Molecular Design Group, Institute of Pharmacy/Pharmaceutical Chemistry and Center for Molecular Biosciences Innsbruck, and <sup>∥</sup>Institute of Pharmacy/Pharmacognosy and Center for Molecular Biosciences Innsbruck, University of Innsbruck, Innrain 80-82, 6020 Innsbruck, Austria

<sup>§</sup>Department of Pharmacognosy, University of Vienna, Althanstraße 14, 1090 Vienna, Austria

<sup>1</sup>Institute of Genetics and Animal Breeding of the Polish Academy of Sciences, Postępu 36A Street, 05-552, Jastrzebiec, Poland

<sup>II</sup>Laboratory of Animal Physiology, Department of Animal Biology and Physiology, Faculty of Science, University of Yaounde I, P.O. Box 812, Yaounde, Cameroon

<sup>□</sup>Griffith Institute for Drug Discovery, Griffith University, Brisbane, QLD 4111, Australia

**Supporting Information** 



**ABSTRACT:**  $17\beta$ -Hydroxysteroid dehydrogenase type 2 ( $17\beta$ -HSD2) converts the active steroid hormones estradiol, testosterone, and  $5\alpha$ -dihydrotestosterone into their weakly active forms estrone,  $\Delta^4$ -androstene-3,17-dione, and  $5\alpha$ -androstane-3,17-dione, respectively, thereby regulating cell- and tissue-specific steroid action. As reduced levels of active steroids are associated with compromised bone health and onset of osteoporosis,  $17\beta$ -HSD2 is considered a target for antiosteoporotic treatment. In this study, a pharmacophore model based on  $17\beta$ -HSD2 inhibitors was applied to a virtual screening of various databases containing natural products in order to discover new lead structures from nature. In total, 36 hit molecules were selected for biological evaluation. Of these compounds, 12 inhibited  $17\beta$ -HSD2 with nanomolar to low micromolar IC<sub>50</sub> values. The most potent compounds, nordihydroguaiaretic acid (1), IC<sub>50</sub> 0.38 ± 0.04  $\mu$ M, (–)-dihydroguaiaretic acid (4), IC<sub>50</sub> 0.94 ± 0.02  $\mu$ M, isoliquiritigenin (6), IC<sub>50</sub> 0.36 ± 0.08  $\mu$ M, and ethyl vanillate (12), IC<sub>50</sub> 1.28 ± 0.26  $\mu$ M, showed 8-fold or higher selectivity over  $17\beta$ -HSD1. As some of the identified compounds belong to the same structural class, structure–activity relationships were derived for these molecules. Thus, this study describes new  $17\beta$ -HSD2 inhibitors from nature and provides insights into the binding pocket of  $17\beta$ -HSD2, offering a promising starting point for further research in this area.

17β-Hydroxysteroid dehydrogenase type 2 (17β-HSD2) belongs to a large family of short-chain dehydrogenase/ reductase (SDR) enzymes with the systematic name SDR9C2.<sup>1</sup> It is mainly expressed in the placenta, endometrium, breast, prostate, small intestine, liver, and bone.<sup>2–5</sup> This NAD<sup>+</sup>dependent enzyme converts active sex steroid hormones such as estradiol, testosterone, and 5α-dihydrotestosterone into their respective inactive forms, namely, estrone,  $\Delta^4$ -androstene-3,17dione (androstenedione), and 5α-androstane-3,17-dione (androstanedione), thereby protecting tissues from excessive sex steroid hormone action (Figure 1).<sup>67</sup> Furthermore, 17β-HSD2 catalyzes the oxidation of  $\Delta^{5}$ -androstene- $3\beta$ ,17 $\beta$ -diol (androstenediol) to dehydroepiandrosterone (DHEA). The enzyme shares considerable structural and functional similarity with other extensively studied SDR enzymes such as 17 $\beta$ -HSD1 and 17 $\beta$ -HSD3.<sup>8</sup> In contrast to 17 $\beta$ -HSD2, the enzymes 17 $\beta$ -HSD1, 17 $\beta$ -HSD3, and the aldo-keto-reductase 17 $\beta$ -HSD5 (also known as AKR1C3) are oxidoreductases converting the weak

 Received:
 October 14, 2016

 Published:
 March 20, 2017





Figure 1. Enzymatic reactions catalyzed by  $17\beta$ -HSD2 and reverse reactions catalyzed by other HSD enzymes.

estrogen estrone to the potent estradiol and the weak androgens androstenedione and androstanedione to testosterone and  $5\alpha$ -dihydrotestosterone, respectively.<sup>9–11</sup> Whereas  $17\beta$ -HSD3 is responsible for the last step of testosterone synthesis in the testes,  $17\beta$ -HSD5 is responsible for the production of extratesticular testosterone and plays a crucial role in androgen maintenance in the elderly.<sup>9,10</sup>

Owing to its favorable localization and its role as a main contributor to the inactivation of estradiol, testosterone, and  $5\alpha$ -dihydrotestosterone in bone cells,<sup>2</sup> 17 $\beta$ -HSD2 has been proposed as a promising target for the treatment of osteoporosis.<sup>12</sup> This condition, where decreased bone density leads to an increased fracture risk, is in the majority of cases

Article

linked with the age-related decrease of sex steroid hormones.<sup>13</sup> The age-related onset of osteoporosis in postmenopausal women<sup>14</sup> and men with low testosterone levels<sup>15</sup> can be explained, at least in part, by a decline in the concentrations of estradiol and testosterone, which inhibit bone degradation.<sup>16</sup> Thus, by inhibiting  $17\beta$ -HSD2, the amount of active steroids can be locally increased in the bones, thereby improving bone health. This hypothesis is supported by an in vivo study, where a  $17\beta$ -HSD2 inhibitor was administered to ovariectomized cynomolgus monkeys.<sup>17</sup> In this study, the  $17\beta$ -HSD2 inhibitor was shown to improve bone strength by increasing bone formation and decreasing bone resorption, although the effects were rather weak and only observed at the highest dose of 25 mg/kg/day.

Although multiple synthetic  $17\beta$ -HSD2 inhibitors have already been reported,<sup>18-21</sup> natural products inhibiting this enzyme are currently underexplored. There are only a few reports on natural product inhibitors of  $17\beta$ -HSD2 and other steroid-metabolizing enzymes, and the majority of these compounds are flavonoids.<sup>22-24</sup> Flavonoids share certain functional similarities with steroids and can be considered as steroid mimetics (Figure S1, Supporting Information). However, most of these compounds are not selective. They also inhibit other members of the SDR enzyme family, and, additionally, they frequently show activity toward estrogen and androgen receptors. Nevertheless, natural compounds play an important role in providing new structures as potential lead candidates in drug discovery, and hence they are of high general interest.<sup>25,26</sup> Remarkably, from 1999 to 2008, 28% of all new FDA-approved, first-in-class small-molecule drugs were natural products or compounds derived thereof.<sup>2</sup>

Despite the fact that osteoporosis is not well represented among the conditions treated with plants and phytotherapy,<sup>28</sup> there are many other conditions related to bone homeostasis and fractures that are reported in the literature on ethnopharmacology. Interestingly, an ethnopharmacological study has been reported that shows that plants such as *Pholidota articulate* Lindl. and *Coelogyne cristata* Lindl. (both of the Orchidaceae family) contain several flavonoids that are used to treat bone fractures in India.<sup>29</sup> Even though part of the observed effects of these compounds may be due to direct



Figure 2. Pharmacophore models for  $17\beta$ -HSD2 inhibitors. (A) Chemical features of models 1 and 2 describing the types, locations, and tolerance spheres of inhibitory chemical functionalities. Pharmacophore features are colored as follows: red, hydrogen-bond acceptor; green, hydrogen-bond donor; yellow, hydrophobic; and blue, aromatic ring. Optional features are depicted in scattered style. (B) Full versions of models 1 and 2 with gray exclusion volumes as steric restraints for inhibitor size (forbidden areas). A 3D video view of model 1 is available as Supporting Information.

### Table 1. Active Hit Compounds of Natural Origin, Databases, Mapping Pharmacophore Models, and Activities against $17\beta$ -HSD2

compound	database	pharmacophore models	remaining activity at 20 $\mu M$ (% of control) or $IC_{50}$
nordihydroguaiaretic acid (1)	Atanasov	models 1 and 2	$0.38\pm0.04\;\mu\mathrm{M}$
oleanolic acid (2)	Atanosov	model 1 omf <sup>a</sup>	49 ± 6%
curcumin (3)	Atanosov	models 1 and 2 omf	$1.73 \pm 0.2 \ \mu \text{M}$
(–)-dihydroguaiaretic acid (4)	Davis	models 1 and 2	$0.94 \pm 0.02 \ \mu \mathrm{M}$
jaceosidin (5)	Davis	models 1 and 2 omf	$9.3 \pm 2.3 \ \mu M$
isoliquiritigenin (6)	Davis	models 1 and 2	$0.36 \pm 0.08 \ \mu\mathrm{M}$
pinoresinol (7)	Waltenberger	models 1 and 2	$42 \pm 5\%$
lupinalbin A (8)	Krenn	model 2 omf	$1.52 \pm 0.15 \ \mu M$
2'-hydroxygenistein (9)	Krenn	model 2 omf	$2.03 \pm 0.37 \ \mu\mathrm{M}$
butein (10)	Sigma	model 1	$7.3 \pm 2.7 \ \mu M$
rosmarinic acid (11)	Sigma	model 1	$3.72\pm0.17\mu\mathrm{M}$
ethyl vanillate (12)	Sigma	model 1	$1.28\pm0.26\;\mu\mathrm{M}$
<sup>a</sup> omf, screening by allowing one omit	ted feature.		



compound	database	pharmacophore models	remaining activity at 20 $\mu {\rm M}$ (% of control) or ${\rm IC}_{\rm 50}$
2-(3-chloro-4-hydroxyphenyl)-N-phenethylacetamide (13)	Davis	model 1	1.57 $\pm$ 0.16 $\mu$ M
2-(3-chloro-4-hydroxyphenyl)-N-(2-methoxyethyl)acetamide (14)	Davis	model 1 omf <sup>a</sup>	37 ± 3%
N-butyl-2-(3-chloro-4-hydroxyphenyl)acetamide (15)	Davis	model 1	33 ± 6%
N-benzyl-2-(3-chloro-4-hydroxyphenyl)acetamide (16)	Davis	model 1	$3.42 \pm 0.74 \ \mu \mathrm{M}$
N-(2-(1H-indol-3-yl)ethyl)-2-(3-chloro-4-hydroxyphenyl)acetamide (17)	Davis	model 1	0.98 $\pm$ 0.24 $\mu$ M
2-(3-chloro-4-hydroxyphenyl)-N-(2-chlorobenzyl)acetamide (18)	Davis	model 1	0.78 $\pm$ 0.16 $\mu M$
<sup><i>a</i></sup> omf, screening by allowing one omitted feature.			



Figure 3. Structures of natural products identified in this study that inhibit  $17\beta$ -HSD2.

modulation of estrogen and androgen receptor activities, the mechanism of action of these compounds in the treatment of bone-related conditions is largely unknown. Accordingly,  $17\beta$ -HSD2 inhibition might well contribute to the effects of these herbal remedies.

As natural compounds represent a rich source of potential lead structures, novel  $17\beta$ -HSD2 inhibitors of natural origin were searched using in silico methods. Previously, a procedure

to discover new synthetic chemicals that inhibit  $17\beta$ -HSD2 was established.<sup>19</sup> In this previous study, pharmacophore models representing the chemical functionalities and steric requirements essential for the activity of small molecules toward  $17\beta$ -HSD2 were constructed and employed for virtual screening of a commercial synthetic chemical database. From this previous experimental validation, the two pharmacophore models 1 and 2 (Figure 2) showed good predictive power, with positive hit



Figure 4. Semisynthetic fungal natural products that inhibit  $17\beta$ -HSD2.

#### Table 3. Selectivity of the Most Active $17\beta$ -HSD2 Inhibitors toward $17\beta$ -HSD1

compound	$17\beta$ -HSD2 activity (IC <sub>50</sub> )	17 $\beta$ -HSD1 activity (IC <sub>50</sub> or remaining activity at 20 $\mu$ M)	selectivity factor
nordihydroguaiaretic acid (1)	$0.38 \pm 0.04 \; \mu \mathrm{M}$	$5.5 \pm 1.3 \ \mu M$	15
curcumin (3)	$1.73\pm0.20\;\mu\mathrm{M}$	$52.2 \pm 7.1\%$	~12
(–)-dihydroguaiaretic acid (4)	$0.94 \pm 0.02 \ \mu \mathrm{M}$	$7.7 \pm 2.2 \ \mu M$	8
isoliquiritigenin (6)	$0.36 \pm 0.08 \; \mu \mathrm{M}$	$2.83 \pm 0.80 \ \mu \mathrm{M}$	8
lupinalbin A (8)	$1.52 \pm 0.15 \; \mu \mathrm{M}$	$0.049 \pm 0.019 \ \mu M$	0.03
2'-hydroxygenistein (9)	$2.03\pm0.37\;\mu\mathrm{M}$	$1.09 \pm 0.06 \ \mu M$	0.5
rosmarinic acid (11)	$3.72\pm0.17\;\mu\mathrm{M}$	n.i. <sup>a</sup>	>5
ethyl vanillate (12)	$1.28\pm0.26\;\mu\mathrm{M}$	n.i.	>15
2-(3-chloro-4-hydroxyphenyl)-N-phenethylacetamide (13)	$1.57$ $\pm$ 0.16 $\mu \rm M$	n.i.	>12
N-benzyl-2-(3-chloro-4-hydroxyphenyl)acetamide (16)	$3.42 \pm 0.74 \; \mu \mathrm{M}$	n.i.	>5
N-(2-(1H-indol-3-yl)ethyl)-2-(3-chloro-4- hydroxyphenyl)acetamide (17)	$0.98 \pm 0.24 \ \mu\mathrm{M}$	n.i.	>20
2-(3-chloro-4-hydroxyphenyl)-N-(2-chlorobenzyl)acetamide (18)	0.78 $\pm$ 0.16 $\mu {\rm M}$	54.8 ± 5.8%	~25
<sup>a</sup> n.i., no inhibition.			

rates of 50% and 10%, respectively. Although the models are very similar in feature types and distribution, they differ slightly in feature location, which is why they may lead to somewhat different virtual hits. Thus, both of these models were selected for virtual screening of selected natural product databases.

#### RESULTS AND DISCUSSION

In-house natural product databases based on input from several academic institutions (total of 439 entries) and the Sigma-Aldrich catalogue (Sigma-Aldrich, St. Louis, MO, USA), containing natural products and synthetic compounds, were screened virtually using the two pharmacophore models. The virtual screening procedure and its results are described in detail in the Supporting Information (text and Table S1). As the full models were quite restrictive, most databases were also screened in models where one omitted feature (omf) was applied during the pharmacophore mapping.

The 36 selected virtual hits were evaluated in an in vitro assay using lysates of cells expressing the recombinant human enzyme 17 $\beta$ -HSD2. Initially, all compounds were tested at a final concentration of 20  $\mu$ M. Compounds showing more than 50% inhibition at that concentration are shown in Tables 1 and 2 as well as Figures 3 and 4. For all compounds inhibiting 17 $\beta$ -HSD2 activity by at least 70% (remaining activity  $\leq$ 30% of vehicle control), IC<sub>50</sub> values were determined. The complete list of the compounds tested is provided in Table S2, Supporting Information.

From the selected 36 tested compounds, 12 were active with IC<sub>50</sub> values of <5  $\mu$ M, six were moderately active showing at least 50% inhibition at a compound concentration of 20  $\mu$ M, and the remaining compounds were considered inactive. Altogether, this corresponds to a 50% hit rate, indicating that the pharmacophore models performed explicitly well, not only

for synthetic molecules but also for natural compounds. This is an important aspect, because natural products often differ from synthetic drug-like structures. From the 33 in-house databasederived test compounds, 10 fit into model 1 and four into model 2, respectively, without omitted features during the screening (Tables 1 and 2). Remarkably, all these hits were active in vitro. Additionally, the strategy of allowing one pharmacophore feature to be left out during the natural product database screening proved successful: The hits obtained by allowing one omitted feature additionally included the active compounds oleanolic acid (2), curcumin (3), jaceosidin (5), lupinalbin A (8), 2'-hydroxygenistein (9), and the semisynthetic derivative 2-(3-chloro-4-hydroxyphenyl)-N-(2methoxyethyl)acetamide (14). Although, admittedly, all inactive compounds from this study have also been identified in the screenings with one omitted feature, these additional active hits encourage this screening mode, when a wider range of chemically diverse  $17\beta$ -HSD2 inhibitors is sought and a higher number of false positive virtual hits is acceptable.

For a possible therapeutic use of a  $17\beta$ -HSD2 inhibitor, a compound must be selective over  $17\beta$ -HSD1, which catalyzes the reverse reaction. Therefore, the most active newly identified  $17\beta$ -HSD2 inhibitors were screened at a final concentration of 20  $\mu$ M in vitro using lysates of cells expressing the recombinant human  $17\beta$ -HSD1 enzyme. For all compounds inhibiting  $17\beta$ -HSD1 by 70% or more, IC<sub>50</sub> values and corresponding selectivity factors were determined. The results are shown in Table 3. Follow-up experiments should include additional SDR enzymes such as  $11\beta$ -HSDs,  $3\alpha/\beta$ -HSDs, and retinol dehydrogenases as well as a careful assessment of the cytotoxic potential of the identified compounds.

Most of the active hits found in this study belong to compound classes associated with steroidogenic activities. This includes the triterpene oleanolic acid (2), which belongs to a compound class containing several  $11\beta$ -HSD inhibitors.<sup>30–33</sup> Compounds 5, 8, and 9 are flavonoids, a class known to have estrogenic activity. Nordihydroguaiaretic acid (1) is a lignan found at high concentrations in the leaves of Larrea tridentata (Sessé & Moc. ex DC.) Coville, a common shrub in the United States and in Mexico.<sup>34</sup> The leaves have been used in the preparation of a tea for the treatment of cancer, arthritis, and tuberculosis. Compound 1 is an antioxidant that also inhibits lipoxygenase, thus influencing the leukotriene cascade and suppressing ovulation in rats.<sup>35</sup> Thereby, it may pose a potential risk for reproductive toxicity if ingested in large amounts. Compound 1 was proposed to be converted into a phytoestrogen by gut flora.<sup>36</sup> In addition, it was shown to have estrogenic effects, being an ER $\alpha$ -agonist, with a tendency to be selective over  $ER\beta$ .<sup>37</sup> Additionally, compound 1 was shown to inhibit the formation of  $\beta$ -amyloid fibrils in the central nervous system and the accumulation of  $\beta$ -peptides. These properties suggest that 1 is an interesting compound for the development of potential anti-Alzheimer disease (AD) pharmaceuticals.<sup>38</sup> Similar anti-amyloidogenic effects were also reported in studies with mice for 1, 3, and 11, supporting the potential preventive properties of these natural compounds against AD.39

Curcumin (3) is a tautomeric diarylheptanoid compound that is found in the roots of Curcuma longa L. and has a great variety of potential therapeutic activities.<sup>40,41</sup> It is one of the main ingredients of curry spice mixtures and is responsible for the yellow color.<sup>42</sup> Many papers have been published in the past few decades describing anti-inflammatory,<sup>43</sup> anti-cancer,<sup>44,45</sup> and antioxidant properties of 3.<sup>40</sup> In Asian medicine, 3 was used for topical or oral application to treat a variety of diseases for thousands of years. Despite the low bioavailability and rapid hepatic metabolism, 3 was shown to be therapeutically active against several diseases.<sup>46</sup> There is debate as to whether 3 may be an invalid bioactive compound because of its PAINS properties<sup>47-49</sup> or may still have some potential as a lead structure candidate for certain conditions.<sup>50</sup> According to the experiments and observations from this study, 3 directly and specifically inhibits  $17\beta$ -HSD2 and  $17\beta$ -HSD1. A detailed discussion on this issue is provided in the Supporting Information (p S9). Although 3 may not be a suitable lead compound for various reasons, it still reflects the ability of the virtual screening workflow to detect structurally diverse  $17\beta$ -HSD2 inhibitors.

Dihydroguaiaretic acid (4) is another lignan that is present in various plant extracts, such as those derived from the bark of Machilus thunbergii<sup>51</sup> Siebold & Zucc. and the seeds of Myristica fragrans Houtt.<sup>52</sup> These plants are found predominantly in tropical and subtropical Asian countries. Compound 4 was reported to possess antibacterial,<sup>53</sup> antioxidative,<sup>54</sup> and potential anticancer properties.55 Little is known about the potential interference of 4 with estrogen-metabolizing hormones. In 2001, Filleur et al. reported that 4 showed no effects on 17 $\beta$ -HSD activity in placenta microsomes.<sup>56</sup> This is in contrast with the potent inhibition (IC<sub>50</sub>, 940  $\pm$  20 nM) of 17 $\beta$ -HSD2 by 4 found in the present study. The reason for this discrepancy is unclear but may be due to experimental differences, as in the present study recombinant human enzyme was used. In contrast, in the study by Filleur et al. placenta microsomes that also express other steroid-metabolizing enzymes were applied.

Isoliquiritigenin (6) is a hydroxylated chalcone found in *Glycyrrhiza uralensis* Fisch. ex. DC.<sup>57</sup> and other various plant preparations. Many pharmacological effects of 6 have been described in the literature such as antitumor, antioxidative, and antibacterial properties.<sup>58</sup> Using a recombinant protein, it was reported that 6 inhibits aromatase activity with an IC<sub>50</sub> value of 3.8  $\mu$ M.<sup>59</sup> This would lower the amount of estrogens produced from androgens, which may aggravate osteoporosis. Nevertheless, 6 is a moderately potent inhibitor of aromatase, and efficient inhibition of 17 $\beta$ -HSD2 was achieved at concentrations 10 times lower. Importantly, 6 did not inhibit 17 $\beta$ -HSD1. Using yeast strains expressing human receptors, 6 was shown to bind to ER $\alpha$  (IC<sub>50</sub> to displace estradiol of 1.87  $\mu$ M) and ER $\beta$  (IC<sub>50</sub> of 269 nM), however, with much lower affinity than estradiol.<sup>60</sup>

Compounds 8 and 9 are major constituents contained in a methanolic extract of the aerial parts of Eriosema laurentii De Wild, which was shown to have protective effects against femur mass loss and significantly increased calcium and inorganic phosphorus content in the femur in ovariectomized rats.<sup>61,62</sup> Inhibition of  $17\beta$ -HSD2 by these compounds may enhance local levels of estradiol, thereby potentiating estrogen receptor  $\alpha$  (ER $\alpha$ )-mediated signaling. However, some of these effects may be explained by direct effects of the compounds on steroid receptors and/or helix-loop-helix transcription factors. In yeast systems expressing the human  $ER\alpha$  and the human aryl hydrocarbon receptor, 8 showed agonistic effects with EC<sub>50</sub> values of 21.4 nM and 1.34  $\mu$ M, respectively.<sup>63</sup> Additionally, 9 was reported to activate ER $\alpha$  with an EC<sub>50</sub> value of 6.1  $\mu$ M. Regarding 8 and 9, it needs to be noted that these compounds exert more potent inhibitory effects against  $17\beta$ -HSD1 than 17 $\beta$ -HSD2. In fact, 8 potently inhibited 17 $\beta$ -HSD1 with an  $IC_{50}$  of 49  $\pm$  19 nM and an approximately 30-fold selectivity over 17 $\beta$ -HSD2. This in vitro information suggests that 8 most potently activates  $ER\alpha$  and potently inhibits estrone to estradiol conversion by  $17\beta$ -HSD1 but shows weaker effects on  $17\beta$ -HSD2-mediated estradiol inactivation. Depending on the tissue and cell type, ER $\alpha$  is expressed together with either 17 $\beta$ -HSD1 or  $17\beta$ -HSD2, which may result in cell-specific estrogenic effects of 8.

Rosmarinic acid (11) was first isolated from an extract of *Rosmarinus officinalis* L.<sup>64</sup> This compound was studied for many years and showed antinociceptive and anti-inflammatory effects in animal studies.<sup>65</sup> In addition, several clinical trials showed positive effects of comfrey roots containing 11 as a topical treatment against pain.<sup>66</sup> Antinociceptive effects would clearly be beneficial in the treatment of osteoporosis because of increasing pain with progression of the disease. Compound 11 selectively inhibited  $17\beta$ -HSD2 over  $17\beta$ -HSD1, although with rather moderate activity. It therefore remains to be seen whether such concentrations can be reached in bone cells. Alternatively, paracrine effects from neighboring cells may affect estrogen availability and therefore bone metabolism.

Ethyl vanillate (12) is an antioxidative<sup>67</sup> compound that has been found in hedge mustard [*Sisymbrium officinale* (L.) Scop.] and also in Pinot noir wine.<sup>68</sup> Although 12 has been known for quite some time, due to its intense vanilla taste and its use as a flavoring additive, its biological properties remain poorly investigated.

Most of the newly discovered  $17\beta$ -HSD2 inhibitors were already known as phytoestrogens or compounds that are converted into phytoestrogen by gut flora (e.g., pinoresinol (7) and 1).<sup>36</sup> The rationale why the pharmacophore model found

#### Journal of Natural Products

these ER-active compounds was that the substrate (estradiol) of 17 $\beta$ -HSD2 is the endogenous ER agonist, and thus the binding pockets of ER and 17 $\beta$ -HSD2 are obviously able to accommodate similar compounds that may be considered as steroid mimetics. This was reflected by the pharmacophore model that is based on the properties of compounds binding to 17 $\beta$ -HSD2: the compounds that share features needed for binding to 17 $\beta$ -HSD2 are likely to bind to ER $\alpha$  and ER $\beta$  as well.

Many of the active hits share considerable structural similarity. Interestingly, the most active substance, **6**, has one phenolic hydroxy group less than **10**. This difference led to a drastic effect on the activity of these compounds: **6** gave an IC<sub>50</sub> value of  $0.36 \pm 0.08 \ \mu$ M, whereas **10** was 20-fold less active, with an IC<sub>50</sub> of 7.3  $\pm$  2.7  $\mu$ M. However, the difference in the overall lipophilicity of these compounds may also play a role in their different activities.

The semisynthetic fungal natural products  $(13-18^{69})$ followed a clear structure-activity relationship (SAR), with the activity shown to increase when a second aromatic ring was present. The parent compound (i.e., natural product) for this semisynthetic series, 2-(3-chloro-4-hydroxyphenyl)acetamide (S11), and the related natural products 2-(3-chloro-4hydroxyphenyl)acetic acid (S15) and 2-(4-hydroxyphenyl)acetamide (S16) (see Table S1, Supporting Information, for their chemical structures), did not inhibit  $17\beta$ -HSD2, whereas compounds 2-(3-chloro-4-hydroxyphenyl)-N-(2methoxyethyl)acetamide (14) and N-butyl-2-(3-chloro-4hydroxyphenyl)acetamide (15) were moderately active. The most active compounds from this series were 2-(3-chloro-4hydroxyphenyl)-N-phenethylacetamide (13) and 16-18, which all shared a similar interaction pattern (Figure 5A). However, if the acetamide fragment is extended with, for example, an Nbutyl chain, the compound can form additional hydrophobic interactions with the enzyme, resulting in an increased activity (Figure 5B). In addition to the alkyl chain, the most active compounds have a second aryl ring that can form aromatic interactions with the enzyme (Figure 5C). On the basis of the activities of these compounds, it can be proposed that  $17\beta$ -HSD2 has a hydrophobic ligand binding pocket and aromatic amino acid residues in the active site that may contribute to the affinities of these ligands.

Most of the tested compounds inhibited selectively  $17\beta$ -HSD2 over  $17\beta$ -HSD1, except for compounds **8** and **9**. The semisynthetic compounds **13** and **16–18** also showed good selectivity in terms of the inhibition of  $17\beta$ -HSD2. The two most potent compounds, **1** and **6**, were 15 and 8 times more active toward  $17\beta$ -HSD2 than  $17\beta$ -HSD1. Both compounds are potential natural lead structures that could be used for the development of  $17\beta$ -HSD2 drug candidates. Unlike many other related compounds that are possibly rapidly metabolized due to the presence of several hydroxy groups, 2-(3-chloro-4-hydroxyphenyl)-*N*-(2-chlorobenzyl)acetamide (**18**) has only a single hydroxy group and might therefore be less prone to rapid biotransformation. Compound **18** still potently and selectively inhibited  $17\beta$ -HSD2 with an IC<sub>50</sub> of 0.78 ± 0.16  $\mu$ M.

Among the most active compounds identified during these studies were also the flavonoids 5 and 9. Schuster et al. earlier reported several flavonoids inhibiting  $17\beta$ -HSD2. Taking the data together (Table 4),<sup>24</sup> a SAR model for the flavonoids that inhibit this enzyme could be established (Figure 6).

In general, the active flavonoids share a typical pharmacophore model containing hydrogen bond acceptors and donors



Figure 5. Illustration of the SAR of semisynthetic natural product derivatives (Table 3). (A) The core structure with compounds S12 (gray), S15 (red), and S11 (blue) with a pharmacophore model illustrating the interaction pattern. (B) The moderately active compounds 14 (yellow) and 15 (gray) with the additional hydrophobic feature. (C) The most active compounds 13 (orange), *N*-benzyl-2-(3-chloro-4-hydroxyphenyl)acetamide (16, green), *N*-(2-(1*H*-indol-3-yl)ethyl)-2-(3-chloro-4-hydroxyphenyl) (17, purple), and 18 (gray) with the additional aromatic ring feature.

and hydrophobic and aromatic features (Figure 6A). The hydrogen bond acceptor in position C-3 (scaffold A) was found to be beneficial for activity, as the most active flavonoids, **30** and **31**, contain a hydroxy group at this position (Figure 6B). If this feature was absent, the activity decreased or the compound was inactive. Furthermore, the hydrogen bond acceptor unit at the C-4'-position is important and shared by all active compounds. If the hydrogen-bonding feature at this position was deleted, active and inactive compounds were no longer distinguished (Figure 6C).

To learn more about the general properties of  $17\beta$ -HSD2 inhibitors, model 1 and the flavonoid model were aligned (Figure 7). Every model contains an aromatic ring feature next to a hydrogen bond donor/acceptor feature. Among the compounds mapped, this combination was often represented by a phenolic hydroxy group. Another common feature was the hydrophobic/aromatic group in a certain distance from the first feature group. Interestingly, in between these aligned hydrophobic/aromatic features, there were hydrogen bond acceptor features. These indicate that in the binding pocket there may be two hydrophobic regions that tolerate aromatic interactions, and in between these pockets, there was most likely a hydrogen-bonding partner. This feature arrangement is in line with the architecture of already crystallized  $11\beta$ -HSD1 and  $17\beta$ -HSD1, where inhibitors are anchored to the catalytically active amino acids by central hydrogen bonds and form further, adjacent hydrophobic contacts (e.g., the PDB structures 4c7j<sup>7</sup> and  $3hb5^{71}$ ).

#### Table 4. Flavonoid Structures and Activities Used for Deriving a Flavonoid SAR Model of $17\beta$ -HSD2 Inhibitors

$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array}{} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\$				$ \begin{array}{c} 7 & \mathbf{A} & \mathbf{C} \\ 6 & \mathbf{C} & 2 \\ 6 & \mathbf{C} & 3 \\ 6 & \mathbf{C} &$		
compound	scaffold	ring A	ring B	ring C	inhibition of 17 $\beta$ -HSD2: % residual activity at 40 $\mu$ M or IC <sub>50</sub> ( $\mu$ M); mean $\pm$ SD	classification
jaceosidin (5)	A	OH-5, OMe- 6, OH-7	OMe-3', OH-4'	$\Delta^{2,3}$	$9.27 \pm 2.28 \ \mu M$	active
2'-hydroxygenistein (9)	В	OH-5, OH-7	OH-2', OH-4'		$2.03 \pm 0.37 \ \mu M$	active
flavanone (19)	A				70.7 ± 2.6 %	inactive24
2'-hydroxyflavanone (20)	A		OH-2'		$50.7 \pm 0.5$ %	inactive24
4'-hydroxyflavanone (21)	A		OH-4'		63.6 ± 3.5 %	inactive24
6-hydroxyflavanone (22)	A	OH-6			$42.5 \pm 1.8$ %	inactive24
naringenin (23)	A	OH-5, OH-7	OH-4'	2-( <i>S</i> )	$14.4\pm4.5~\mu M$	active24
hesperetin (24)	A	OH-5, OH-7	OH-3', OMe-4'	2-( <i>S</i> )	36.4 ± 1.7 %	inactive24
flavone (25)	A			$\Delta^{2,3}$	$63.9 \pm 0.6$ %	inactive <sup>24</sup>
7-hydroxyflavone (26)	A	OH-7		$\Delta^{2,3}$	$23.1 \pm 1.1 \% (> 20 \ \mu M)$	weak <sup>24</sup>
chrysin (27)	A	OH-5, OH-7		$\Delta^{2,3}$	$32.8 \pm 5.1 \%$	inactive <sup>24</sup>
daidzein (28)	В	OH-7	OH-4'		$58.3 \pm 6.1$ %	inactive <sup>24</sup>
apigenin (29)	A	OH-5, OH-7	OH-4'	$\Delta^{2,3}$	$25.2 \pm 0.9 \% (> 20 \ \mu M)$	weak <sup>24</sup>
kaempferol (30)	A	OH-5, OH-7	OH-4'	$\Delta^{2,3}$ , OH-3	$0.36\pm0.04~\mu M$	active <sup>24</sup>
quercetin (31)	A	OH-5, OH-7	OH-3', OH-4'	$\Delta^{2,3}$ , OH-3	$1.54\pm0.22~\mu M$	active <sup>24</sup>
genistein (32)	В	OH-5, OH-7	OH-4'		$16.5 \pm 2.7 \ \mu M$	active <sup>24</sup>
biochanin A (33)	В	OH-5, OH-7	OMe-4'		$9.90\pm1.87~\mu M$	active <sup>24</sup>

The present virtual screening approach for the identification of natural products-derived  $17\beta$ -HSD2 inhibitors was productive. Thus, only 38 compounds had to be tested to yield 17 active hits with sub- and low-micromolar IC<sub>50</sub> values. The most potent bioactive compound, **6**, exhibited an IC<sub>50</sub> value of  $360 \pm$ 80 nM. Thus, the present approach had a success rate of 47%within the virtual hit lists. The fact that so many interesting  $17\beta$ -HSD2 inhibitors were obtained within this relatively small natural product collection points toward the probable presence of more potent active compounds among other natural products.

Furthermore, SAR information was derived for two compound classes, providing more detailed insight into the binding pocket of the enzyme. Only 8 and 9, which were identified by model 2 with one omitted feature, were not selective and even preferentially inhibited  $17\beta$ -HSD1. Consequently, both compounds seem not to be suitable lead structures for further development as antiosteoporosis leads. All other newly discovered  $17\beta$ -HSD2 inhibitors were preferentially selective over  $17\beta$ -HSD1, and therefore they could serve as lead structures for further optimization. It needs to be noted that the activities of these compounds toward  $17\beta$ -HSD2 are at least an order of magnitude lower than that of reported synthetic, chemically optimized compounds.<sup>18,20,21</sup> To further develop potential lead candidates, additional investigations into the bioavailability, metabolism, and tissue distribution of the identified natural compounds are needed. Inhibition of  $17\beta$ -HSD2 is expected to result in tissue-specific elevated levels of estradiol, and potential adverse effects include endometrial hyperplasia and impaired growth control of the glandular epithelium of the breast.<sup>72–74</sup> Thus, compounds that are primarily active in the bone would be preferred for future drug development.

#### EXPERIMENTAL SECTION

**Databases.** The Davis Compound Library (Griffith Institute for Drug Discovery, Griffith University) consisted of 352 compounds, of which the majority were obtained from Australian natural sources, such as endophytic fungi,<sup>75</sup> macrofungi,<sup>76</sup> plants,<sup>77</sup> and marine invertebrates.<sup>78,79</sup> Approximately 15% of the entries of this library were semisynthetic natural product analogues,<sup>80,76</sup> while a small percentage (~5%) are known commercial drugs or synthetic compounds inspired by natural products. The Atanasov and Krenn databases consisted of 51 and 13 in-house available natural products, respectively, from the Department of Pharmacognosy at the University of Vienna, Austria. From the University of Innsbruck, 23 selected plant- and lichen-derived compounds<sup>81–84,62</sup> available in-house at the Institute of Pharmacy/Pharmacognosy were collected in the Waltenberger database. Finally, the Sigma-Aldrich catalogue was also screened, as it includes some commercially available natural products.

**Virtual Screening.** The databases were prepared for virtual screening by deleting counterions and generating multiconformational databases using OMEGA implemented in LigandScout 3.03b. For the relatively small in-house databases used, BEST settings were employed with a maximum of 500 conformers per molecule. For the larger Sigma-Aldrich database, FAST settings were used, which allowed for a maximum of 50 conformations per compound.

Origin, Isolation, and Purification of the Natural Compounds. All natural products from the Davis Compound Library were isolated from plants, marine invertebrates, or endophytic fungi archived at the Griffith Institute for Drug Discovery, Griffith University, Australia, or purchased from Sigma-Aldrich. The extraction and isolation of the natural products featured in this paper have been previously reported by Davis et al.<sup>69,85–88</sup> The synthesis and characterization of the semisynthetic fungal analogues 13-18 have also been previously reported in the literature.<sup>69</sup> All compounds from the Davis collection were analyzed for purity prior to screening and were shown by LC-MS or <sup>1</sup>H NMR analysis to have purities of >95%. The compounds from the Atanasov library were obtained from Sigma-Aldrich, except for 2, 3, and butyl gallate (S3), which were purchased from Fisher, Molekula, and ABCR GmbH & Co. KG, respectively. All



Figure 6. SAR of the flavonoids inhibiting  $17\beta$ -HSD2. (A) The three most active compounds, 9 (2'-hydroxygenistein, blue), kaempferol (**30**, red), and quercetin (**31**, green), share a combined hydrogen bond acceptor/donor at position C-4', a hydrophobic (aromatic) ring (ring B), two neighboring hydrogen bond acceptors on rings A and C, and the aromatic ring A. (B) The moderately active inhibitors **5** (magenta), naringenin (**23**, gray), genistein (**32**, black), and biochanin A (**33**, yellow) fit into the SAR-pharmacophore illustrating the importance of the hydrogen bond acceptor features on the B and C rings, respectively. (C) For comparison, the general flavonoid model not distinguishing active from inactive compounds is shown with all 17 flavonoids from Table 4.



**Figure** 7. Alignment of the  $17\beta$ -HSD2 inhibitor model 1 from Vuorinen et al.<sup>19</sup> and the SAR model (features highlighted by grid) for highly active flavonoids. The pharmacophore features are color-coded: hydrogen bond acceptor, red; hydrogen bond donor, green; hydrophobic, yellow; aromatic ring, blue. The alignment centers are indicated with orange spheres.

compounds were purchased at a purity of  $\geq$ 90%. Compounds 8 and 9 were isolated in an activity-guided approach from a MeOH extract from *Eriosema laurentii* de Wild and unambiguously identified by following MS and NMR analysis. HPLC was applied to determine purity and resulted in 98.7% purity for 8 and 92.1% purity for 9. The compounds from the Waltenberger library were isolated from different plant and lichen species in the course of the project "Drugs from Nature Targeting Inflammation" (DNTI).<sup>89</sup> Compound 7 was isolated

from a MeOH extract of the bark material of *Himatanthus sucuuba* (Spruce) Woodson as described elsewhere.<sup>62</sup> The purity of this compound was determined by HPLC and NMR experiments as >95%.

Activity Assays for  $17\beta$ -HSD1 and  $17\beta$ -HSD2 Using Cell **Lysates.** The  $17\beta$ -HSD1 and  $17\beta$ -HSD2 activity assays were performed as described previously.<sup>19</sup> Briefly, lysates of human embryonic kidney cells (HEK-293, ATCC, Manassas, VA, USA) expressing either human  $17\beta$ -HSD1 or human  $17\beta$ -HSD2 were incubated for 10 min at 37 °C in TS2 buffer (100 mM NaCl, 1 mM EGTA, 1 mM EDTA, 1 mM MgCl<sub>2</sub>, 250 mM sucrose, 20 mM Tris-HCl, pH 7.4) in a final volume of 22  $\mu$ L containing either solvent (0.1% DMSO) or the inhibitor at the respective concentration.  $17\beta$ -HSD1 activity was measured in the presence of 200 nM estrone, containing 50 nCi of [2,4,6,7-3H]-estrone, and 500 µM NADPH. In contrast,  $17\beta$ -HSD2 activity was determined in the presence of 200 nM estradiol, containing 50 nCi of  $[2,4,6,7^{-3}H]$ -estradiol, and 500  $\mu$ M NAD<sup>+</sup>. Reactions were stopped after 10 min by adding an excess of unlabeled estradiol and estrone (2 mM of each in methanol). Unlabeled steroids and cofactors were purchased from Sigma-Aldrich and radiolabeled compounds from PerkinElmer (Boston, MA, USA). The steroids were separated by TLC, followed by scintillation counting and calculation of substrate conversion. Data were collected from at least three independent measurements. Compound 29<sup>24</sup> was used as a positive control for  $17\beta$ -HSD1 assays and compound 22 from Vuorinen et al.<sup>19</sup> as a positive control for  $17\beta$ -HSD2 tests.

**Structure–Activity-Relationship Modeling.** The SAR models were generated using LigandScout 4.09 with default settings (Wolber 2005 JCIM;<sup>90</sup> LigandScout 4.09, 2005–2016, Inte:Ligand GmbH, Vienna, Austria, www.inteligand.com). For all compounds, BEST conformational models using iCon (max 500 conformers per entry) were calculated and overlaid by chemical features using the pharmacophore-based alignment algorithm of the program.

#### ASSOCIATED CONTENT

#### **S** Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jnat-prod.6b00950.

Additional information (PDF) 3D video view of model 1 (AVI)

#### AUTHOR INFORMATION

#### **Corresponding Authors**

\*Biochemistry: A. Odermatt, Tel: +41 (0)61 267 15 30. Fax: +41 (0)61 267 15 15. E-mail: alex.odermatt@unibas.ch. \*Molecular modeling: D. Schuster, Tel: +43-512-507-58253. Fax: +43-512-507-58299. E-mail: daniela.schuster@uibk.ac.at.

### ORCID 💿

Daniela Schuster: 0000-0002-9933-8938

#### **Author Contributions**

<sup>#</sup>A. Vuorinen and R. T. Engeli contributed equally to this work. **Notes** 

The authors declare no competing financial interest.

#### ACKNOWLEDGMENTS

This study was supported by the Swiss National Science Foundation (31003A-159454 to A.O.), the Novartis Research Foundation (A.O.), the Austrian Science Fund (P26782 to D.S. and P25971-B23 to A.G.A.), the Hochschuljubiläumsfond (H-297322/2014 to A.G.A.), the Ernst Mach Stipendium (to S.B.A.), the National Health and Medical Research Council (NHMRC) (APP1024314 to R.A.D.), and the Australian Research Council (ARC) (LE0668477, LE0237908, LP120200339 to R.A.D.). D.S. is an Ingeborg Hochmair professor of the University of Innsbruck. We thank P. Schuster

and G. Begg for help in preparing the manuscript and Inte:Ligand GmbH for providing LigandScout software free of charge.

#### REFERENCES

(1) Persson, B.; Kallberg, Y.; Bray, J. E.; Bruford, E.; Dellaporta, S. L.; Favia, A. D.; Duarte, R. G.; Jornvall, H.; Kavanagh, K. L.; Kedishvili, N.; Kisiela, M.; Maser, E.; Mindnich, R.; Orchard, S.; Penning, T. M.; Thornton, J. M.; Adamski, J.; Oppermann, U. *Chem.-Biol. Interact.* **2009**, *178*, 94–98.

(2) Dong, Y.; Qiu, Q. Q.; Debear, J.; Lathrop, W. F.; Bertolini, D. R.; Tamburini, P. P. J. Bone Miner. Res. **1998**, *13*, 1539–1546.

(3) Mustonen, M.; Poutanen, M.; Kellokumpu, S.; de Launoit, Y.; Isomaa, V.; Vihko, R.; Vihko, P. J. Mol. Endocrinol. **1998**, 20, 67–74.

- (4) Mustonen, M. V.; Isomaa, V. V.; Vaskivuo, T.; Tapanainen, J.;
  Poutanen, M. H.; Stenback, F.; Vihko, R. K.; Vihko, P. T. J. Clin. Endocrinol. Metab. 1998, 83, 1319–1324.
- (5) Takeyama, J.; Sasano, H.; Suzuki, T.; Iinuma, K.; Nagura, H.; Andersson, S. J. Clin. Endocrinol. Metab. **1998**, 83, 3710–3715.

(6) Puranen, T. J.; Kurkela, R. M.; Lakkakorpi, J. T.; Poutanen, M. H.; Itaranta, P. V.; Melis, J. P.; Ghosh, D.; Vihko, R. K.; Vihko, P. T. *Endocrinology* **1999**, *140*, 3334–3341.

(7) Wu, L.; Einstein, M.; Geissler, W. M.; Chan, H. K.; Elliston, K. O.; Andersson, S. J. Biol. Chem. **1993**, 268, 12964–12969.

(8) Lukacik, P.; Kavanagh, K. L.; Oppermann, U. Mol. Cell. Endocrinol. 2006, 248, 61–71.

(9) Dufort, I.; Rheault, P.; Huang, X. F.; Soucy, P.; Luu-The, V. Endocrinology **1999**, 140, 568-574.

(10) Geissler, W. M.; Davis, D. L.; Wu, L.; Bradshaw, K. D.; Patel, S.; Mendonca, B. B.; Elliston, K. O.; Wilson, J. D.; Russell, D. W.;

Andersson, S. Nat. Genet. **1994**, 7, 34–39. (11) Ghosh, D.; Vihko, P. Chem.-Biol. Interact. **2001**, 130–132, 637–650.

(12) Soubhye, J.; Alard, I. C.; van Antwerpen, P.; Dufrasne, F. Future Med. Chem. 2015, 7, 1431–1456.

(13) Compston, J. E. Physiol. Rev. 2001, 81, 419-447.

(14) Riggs, B. L.; Khosla, S.; Melton, L. J. J. Bone Miner. Res. 1998, 13, 763–773.

(15) Chin, K.-Y.; Ima-Nirwana, S. Int. J. Endocrinol. 2012, 2012, 208719.

(16) Michael, H.; Härkönen, P. L.; Väänänen, H. K.; Hentunen, T. A. J. Bone Miner. Res. **2005**, 20, 2224–2232.

(17) Bagi, C. M.; Wood, J.; Wilkie, D.; Dixon, B. J. Musculoskelet. Neuronal. Interact. 2008, 8, 267–280.

(18) Perspicace, E.; Cozzoli, L.; Gargano, E. M.; Hanke, N.; Carotti, A.; Hartmann, R. W.; Marchais-Oberwinkler, S. *Eur. J. Med. Chem.* **2014**, *83*, 317–337.

(19) Vuorinen, A.; Engeli, R.; Meyer, A.; Bachmann, F.; Griesser, U. J.; Schuster, D.; Odermatt, A. J. Med. Chem. **2014**, *57*, 5995–6007.

(20) Wetzel, M.; Marchais-Oberwinkler, S.; Perspicace, E.; Möller, G.; Adamski, I.; Hartmann, R. W. J. Med. Chem. 2011, 54, 7547-7557.

(21) Xu, K.; Al-Soud, Y. A.; Wetzel, M.; Hartmann, R. W.; Marchais-Oberwinkler, S. *Eur. J. Med. Chem.* **2011**, *46*, 5978–5990.

(22) Deluca, D.; Krazeisen, A.; Breitling, R.; Prehn, C.; Möller, G.; Adamski, J. J. Steroid Biochem. Mol. Biol. 2005, 93, 285-292.

(23) Le Bail, J. C.; Laroche, T.; Marre-Fournier, F.; Habrioux, G. Cancer Lett. 1998, 133, 101–106.

(24) Schuster, D.; Nashev, L. G.; Kirchmair, J.; Laggner, C.; Wolber, G.; Langer, T.; Odermatt, A. J. Med. Chem. 2008, 51, 4188-4199.

(25) Atanasov, A. G.; Waltenberger, B.; Pferschy-Wenzig, E. M.; Linder, T.; Wawrosch, C.; Uhrin, P.; Temml, V.; Wang, L.; Schwaiger, S.; Heiss, E. H.; Rollinger, J. M.; Schuster, D.; Breuss, J. M.; Bochkov,

V.; Mihovilovic, M. D.; Kopp, B.; Bauer, R.; Dirsch, V. M.; Stuppner, H. Biotechnol. Adv. 2015, 33, 1582-614.

(26) Newman, D. J.; Cragg, G. M. J. Nat. Prod. 2012, 75, 311–335.
(27) Eder, J.; Sedrani, R.; Wiesmann, C. Nat. Rev. Drug Discovery 2014, 13, 577–587.

(28) Williamson, E. M.; Heinrich, M.; Jäger, A. K., Eds. *Ethnopharmacology*; John Wiley & Sons Ltd: Chichester, West Sussex, UK, 2015; pp 213–226.

(29) Sharma, C.; Kumari, T.; Arya, K. R. Int. J. Pharm. Res. Health Sci. 2014, 2, 185–190.

(30) Blum, A.; Favia, A. D.; Maser, E. Mol. Cell. Endocrinol. 2009, 301, 132-136.

(31) Kratschmar, D. V.; Vuorinen, A.; Da Cunha, T.; Wolber, G.; Classen-Houben, D.; Doblhoff, O.; Schuster, D.; Odermatt, A. J. Steroid Biochem. Mol. Biol. 2011, 125, 129–142.

(32) Rollinger, J. M.; Kratschmar, D. V.; Schuster, D.; Pfisterer, P. H.; Gumy, C.; Aubry, E. M.; Brandstötter, S.; Stuppner, H.; Wolber, G.; Odermatt, A. *Bioorg. Med. Chem.* **2010**, *18*, 1507–1515.

(33) Vuorinen, A.; Seibert, J.; Papageorgiou, V. P.; Rollinger, J. M.; Odermatt, A.; Schuster, D.; Assimopoulou, A. N. *Planta Med.* **2015**, *81*, 525–532.

(34) Lambert, J. D.; Zhao, D.; Meyers, R. O.; Kuester, R. K.; Timmermann, B. N.; Dorr, R. T. *Toxicon* **2002**, *40*, 1701–1708.

(35) Mikuni, M.; Yoshida, M.; Hellberg, P.; Peterson, C. A.; Edwin, S. S.; Brännström, M.; Peterson, C. M. *Biol. Reprod.* **1998**, *58*, 1211–1216.

(36) Benassayag, C.; Perrot-Applanat, M.; Ferre, F. J. Chromatogr. B: Anal. Technol. Biomed. Life Sci. 2002, 777, 233–248.

(37) Fujimoto, N.; Kohta, R.; Kitamura, S.; Honda, H. *Life Sci.* **2004**, 74, 1417–1425.

(38) Ono, K.; Hasegawa, K.; Yoshiike, Y.; Takashima, A.; Yamada, M.; Naiki, H. *J. Neurochem.* **2002**, *81*, 434–40.

(39) Yamada, M.; Ono, K.; Hamaguchi, T.; Noguchi-Shinohara, M. Adv. Exp. Med. Biol. 2015, 863, 79–94.

(40) Gupta, S. C.; Patchva, S.; Koh, W.; Aggarwal, B. B. Clin. Exp. Pharmacol. Physiol. 2012, 39, 283-299.

(41) Manolova, Y.; Deneva, V.; Antonov, L.; Drakalska, E.; Momekova, D.; Lambov, N. Spectrochim. Acta, Part A 2014, 132, 815–820.

(42) Eigner, D.; Scholz, D. J. Ethnopharmacol. 1999, 67, 1-6.

(43) Ghosh, S.; Banerjee, S.; Sil, P. C. Food Chem. Toxicol. 2015, 83, 111–24.

(44) Epstein, J.; Sanderson, I. R.; Macdonald, T. T. Br. J. Nutr. 2010, 103, 1545–1557.

(45) Ko, E. Y.; Moon, A. J. Cancer Prev. 2015, 20, 223-231.

(46) Anand, P.; Kunnumakkara, A. B.; Newman, R. A.; Aggarwal, B. B. Mol. Pharmaceutics **2007**, *4*, 807–818.

(47) Baell, J. B. J. Nat. Prod. 2016, 79, 616-28.

(48) Baell, J. B.; Holloway, G. A. J. Med. Chem. 2010, 53, 2719-40.

(49) Bisson, J.; McAlpine, J. B.; Friesen, J. B.; Chen, S. N.; Graham,

J.; Pauli, G. F. J. Med. Chem. 2016, 59, 1671-90.

(50) Nelson, K. M.; Dahlin, J. L.; Bisson, J.; Graham, J.; Pauli, G. F.; Walters, M. A. J. Med. Chem. 2017, 60, 1620.

(51) Ma, C. J.; Sung, S. H.; Kim, Y. C. *Planta Med.* 2004, 70, 79–80.
(52) Kwon, H. S.; Kim, M. J.; Jeong, H. J.; Yang, M. S.; Park, K. H.;

Jeong, T. S.; Lee, W. S. *Bioorg. Med. Chem. Lett.* **2008**, *18*, 194–198. (53) Favela-Hernandez, J. M.; Garcia, A.; Garza-Gonzalez, E.; Rivas-

Galindo, V. M.; Camacho-Corona, M. R. *Phytother. Res.* **2012**, *26*, 1957–1960.

(54) Yamauchi, S.; Masuda, T.; Sugahara, T.; Kawaguchi, Y.; Ohuchi, M.; Someya, T.; Akiyama, J.; Tominaga, S.; Yamawaki, M.; Kishida, T.; Akiyama, K.; Maruyama, M. *Biosci., Biotechnol., Biochem.* **2008**, *72*, 2981–2986.

(55) Choi, M. S.; Jeong, H. J.; Kang, T. H.; Shin, H. M.; Oh, S. T.; Choi, Y.; Jeon, S. *Life Sci.* **2015**, *141*, 81–89.

(56) Filleur, F.; Le Bail, J. C.; Duroux, J. L.; Simon, A.; Chulia, A. J. *Planta Med.* **2001**, *67*, 700–704.

(57) Li, S.; Li, W.; Wang, Y.; Asada, Y.; Koike, K. Bioorg. Med. Chem. Lett. 2010, 20, 5398-401.

(58) Peng, F.; Du, Q.; Peng, C.; Wang, N.; Tang, H.; Xie, X.; Shen, J.; Chen, J. *Phytother. Res.* **2015**, *29*, 969–977.

(59) Ye, L.; Gho, W. M.; Chan, F. L.; Chen, S.; Leung, L. K. Int. J. Cancer 2009, 124, 1028-36.

#### Journal of Natural Products

(60) Choi, S. Y.; Ha, T. Y.; Ahn, J. Y.; Kim, S. R.; Kang, K. S.; Hwang, I. K.; Kim, S. *Planta Med.* **2008**, *74*, 25–32.

- (61) Ateba, S. B.; Njamen, D.; Medjakovic, S.; Hobiger, S.; Mbanya, J. C.; Jungbauer, A.; Krenn, L. *J. Ethnopharmacol.* **2013**, *150*, 298–307.
- (62) Waltenberger, B.; Rollinger, J. M.; Griesser, U. J.; Stuppner, H.; Gelbrich, T. Acta Crystallogr., Sect. C: Cryst. Struct. Commun. 2011, 67, 0409–12.

(63) Ateba, S. B.; Njamen, D.; Medjakovic, S.; Zehl, M.; Kaehlig, H.; Jungbauer, A.; Krenn, L. *BMC Complementary Altern. Med.* **2014**, *14*, 294.

- (64) Petersen, M.; Simmonds, M. S. Phytochemistry 2003, 62, 121–125.
- (65) Boonyarikpunchai, W.; Sukrong, S.; Towiwat, P. Pharmacol, Biochem. Behav. 2014, 124, 67–73.
- (66) Staiger, C. Phytother. Res. 2012, 26, 1441-1448.
- (67) Tai, A.; Sawano, T.; Ito, H. Biosci., Biotechnol., Biochem. 2012, 76, 314-318.
- (68) Blazevic, I.; Radonic, A.; Mastelic, J.; Zekic, M.; Skocibusic, M.; Maravic, A. *Chem. Biodiversity* **2010**, *7*, 2023–2034.
- (69) Davis, R. A.; Pierens, G. K.; Parsons, P. G. Magn. Reson. Chem. 2007, 45, 442–445.
- (70) Goldberg, F. W.; Dossetter, A. G.; Scott, J. S.; Robb, G. R.;
- Boyd, S.; Groombridge, S. D.; Kemmitt, P. D.; Sjögren, T.; Gutierrez,
- P. M.; deSchoolmeester, J.; Swales, J. G.; Turnbull, A. V.; Wild, M. J. J. Med. Chem. 2014, 57, 970–986.
- (71) Mazumdar, M.; Fournier, D.; Zhu, D. W.; Cadot, C.; Poirier, D.; Lin, S. X. Biochem. J. **2009**, 424, 357–366.
- (72) Gunnarsson, C.; Hellqvist, E.; Stal, O. Br. J. Cancer 2005, 92, 547–52.
- (73) Gunnarsson, C.; Olsson, B. M.; Stal, O. Cancer Res. 2001, 61, 8448–51.
- (74) Kitawaki, J.; Koshiba, H.; Ishihara, H.; Kusuki, I.; Tsukamoto, K.; Honjo, H. J. Clin. Endocrinol. Metab. **2000**, 85, 3292–6.
- (75) Davis, R. A.; Carroll, A. R.; Andrews, K. T.; Boyle, G. M.; Tran, T. L.; Healy, P. C.; Kalaitzis, J. A.; Shivas, R. G. *Org. Biomol. Chem.* **2010**, *8*, 1785–1790.
- (76) Choomuenwai, V.; Andrews, K. T.; Davis, R. A. Bioorg. Med. Chem. 2012, 20, 7167-7174.
- (77) Levrier, C.; Balastrier, M.; Beattie, K. D.; Carroll, A. R.; Martin, F.; Choomuenwai, V.; Davis, R. A. *Phytochemistry* **2013**, *86*, 121–126.
- (78) Barnes, E. C.; Said, N. A. B. M.; Williams, E. D.; Hooper, J. N. A.; Davis, R. A. *Tetrahedron* **2010**, *66*, 283–287.
- (79) Liberio, M. S.; Sooraj, D.; Williams, E. D.; Feng, Y.; Davis, R. A. *Tetrahedron Lett.* **2011**, *52*, 6729–6731.
- (80) Barnes, E. C.; Choomuenwai, V.; Andrews, K. T.; Quinn, R. J.; Davis, R. A. Org. Biomol. Chem. 2012, 10, 4015–4023.
- (81) Atanasov, A. G.; Wang, J. N.; Gu, S. P.; Bu, J.; Kramer, M. P.; Baumgartner, L.; Fakhrudin, N.; Ladurner, A.; Malainer, C.; Vuorinen, A.; Noha, S. M.; Schwaiger, S.; Rollinger, J. M.; Schuster, D.; Stuppner, H.; Dirsch, V. M.; Heiss, E. H. *Biochim. Biophys. Acta, Gen. Subj.* **2013**, *1830*, 4813–9.

(82) Bauer, J.; Waltenberger, B.; Noha, S. M.; Schuster, D.; Rollinger, J. M.; Boustie, J.; Chollet, M.; Stuppner, H.; Werz, O. *ChemMedChem* **2012**, *7*, 2077–2081.

(83) Fakhrudin, N.; Ladurner, A.; Atanasov, A. G.; Heiss, E. H.; Baumgartner, L.; Markt, P.; Schuster, D.; Ellmerer, E. P.; Wolber, G.; Rollinger, J. M.; Stuppner, H.; Dirsch, V. M. *Mol. Pharmacol.* **2010**, *77*, 559–566.

(84) Oettl, S. K.; Gerstmeier, J.; Khan, S. Y.; Wiechmann, K.; Bauer, J.; Atanasov, A. G.; Malainer, C.; Awad, E. M.; Uhrin, P.; Heiss, E. H.; Waltenberger, B.; Remias, D.; Breuss, J. M.; Boustie, J.; Dirsch, V. M.;

Stuppner, H.; Werz, O.; Rollinger, J. M. PLoS One 2013, 8, e76929.
(85) Barnes, E. C.; Kavanagh, A. M.; Ramu, S.; Blaskovich, M. A.;

Cooper, M. A.; Davis, R. A. *Phytochemistry* **2013**, *93*, 162–166. (86) Baron, P. S.; Neve, J. E.; Camp, D.; Suraweera, L.; Lam, A.; Lai,

J.; Jovanovic, L.; Nelson, C.; Davis, R. A. Magn. Reson. Chem. 2013, 51, 358–363.

(87) Davis, R. A.; Barnes, E. C.; Longden, J.; Avery, V. M.; Healy, P. C. Bioorg. Med. Chem. 2009, 17, 1387–1392.

(88) Healy, P. C.; Hocking, A.; Tran-Dinh, N.; Pitt, J. I.; Shivas, R. G.; Mitchell, J. K.; Kotiw, M.; Davis, R. A. *Phytochemistry* **2004**, *65*, 2373–2378.

(89) Waltenberger, B.; Atanasov, A. G.; Heiss, E. H.; Bernhard, D.; Rollinger, J. M.; Breuss, J. M.; Schuster, D.; Bauer, R.; Kopp, B.; Franz, C.; Bochkov, V.; Mihovilovic, M. D.; Dirsch, V. M.; Stuppner, H. *Monatsh. Chem.* **2016**, *147*, 479–491.

(90) Wolber, G.; Langer, T. J. Chem. Inf. Model. 2005, 45, 160-169.