

Photosynthetic Oxygen Reduction in Isolated Intact Chloroplasts and Cells from Spinach¹

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ABSTRACT

The time course of light-induced O₂ exchange by isolated intact chloroplasts and cells from spinach was determined under various conditions using isotopically labeled O₂ and a mass spectrometer. In dark-adapted chloroplasts and cells supplemented with saturating amounts of bicarbonate, O₂ evolution began immediately upon illumination. However, this initial rate of O₂ evolution was counterbalanced by a simultaneous increase in the rate of O₂ uptake, so that little net O₂ was evolved or consumed during the first ~ 1 minute of illumination. After this induction (lag) phase, the rate of O₂ evolution increased 3- to 4-fold while the rate of O₂ uptake diminished to a very low level. Inhibition of the Calvin cycle, e.g. with DL-glyceraldehyde or iodoacetamide, had negligible effects on the initial rate of O₂ evolution or O₂ uptake; both rates were sustained for several minutes, and about balanced so that no net O₂ was produced. Uncouplers had an effect similar to that observed with Calvin cycle inhibitors, except that rates of O₂ evolution and photoreduction were stimulated 40 to 50%.

These results suggest that higher plant photosynthetic preparations which retain the ability to reduce CO₂ also have a significant capacity to photoreduce O₂. With near-saturating light and sufficient CO₂, O₂ reduction appears to take place primarily via a direct interaction between O₂ and reduced electron transport carriers, and occurs principally when CO₂ fixation reactions are suboptimal, e.g. during induction or in the presence of Calvin cycle inhibitors. The inherent maximum endogenous rate of O₂ reduction is approximately 25 to 50% of the maximum rate of noncyclic electron transport coupled to CO₂ fixation. Although the photoreduction of O₂ is coupled to ion transport and/or phosphorylation, this process does not appear to supply significant amounts of ATP directly during steady-state CO₂ fixation in strong light.

Efficient photosynthetic CO₂ fixation is dependent upon the light-driven production of NADPH and ATP in appropriate stoichiometric amounts. One means for assuring that correct stoichiometries are maintained would be to vary the rate of photosynthetic electron flow through pathways differing in their ATP:NADPH production ratios (10). The possible occurrence of coupled electron flow pathways other than the noncyclic transport from water to NADPH, e.g. pseudocyclic or cyclic electron transport, has been recognized for some time (2, 16). The extent to

which these pathways operate *in vivo*, however, is still unresolved.

Studies with washed broken chloroplast preparations have shown that rates of electron transport to O₂ are generally quite low unless certain exogenous electron acceptors are added. In contrast, algae or intact isolated chloroplasts display a potentially significant endogenous capacity for pseudocyclic (and cyclic) transport (see, e.g. 1, 8, 19, 20, 24). Under certain conditions, algae have been shown to carry out pseudocyclic electron transport (O₂ reduction) at rates approaching saturated rates of noncyclic electron transport (24). A number of direct and indirect measurements with isolated intact chloroplasts have suggested that higher plant photosynthetic tissues also possess a substantial endogenous capacity for O₂ reduction (5, 6, 12, 14, 15, 17). Here, we determined the magnitude and kinetics of O₂ evolution and reduction in isolated intact chloroplasts and whole cells obtained from spinach using a mass spectrometer with a fast mass stepper system. Significant rates of O₂ reduction are shown to occur principally when CO₂ fixation reactions are suboptimal.

MATERIALS AND METHODS

Intact chloroplasts were isolated from greenhouse-grown spinach as described previously (26). Preparations contained more than 70% intact chloroplasts as determined by the ferricyanide reduction method (13) and fixed CO₂ at rates in excess of 100 μmol/mg Chl · h in saturating light. Chloroplasts were routinely assayed in a 0.33 M sorbitol, 50 mM Hepes-KOH (pH 8.0) medium containing 10 mM NaHCO₃, 5 mM Na₂P₂O₇, 2 mM EDTA, 0.25 mM K₂HPO₄, and catalase (195 units/ml). Other additions are indicated in the figure legends.

Intact cells were prepared from freshly harvested spinach leaves sliced into small (0.5 × 2.0 cm) strips. Approximately 2 g of leaf strips were vacuum-infiltrated in 20 ml of media containing 0.8 M sorbitol, 20 mM Mes (pH 5.8) buffer, 12.5 mM K₂SO₄, and 0.75% Macerase (obtained from Calbiochem). Leaf strips were digested in 75 ml of the infiltration medium (maintained at 15 C) in an apparatus similar to that described by Servaites and Ogren (23). Released cells were pelleted (100g), washed twice with 50 mM Hepes-KOH (pH 7.8) buffer containing 0.8 M sorbitol and 1 mM MgCl₂, and subsequently resuspended in a small volume of the washing medium containing 5 mM DTT. Preparations contained ≥ 85% intact (plasmolysed) cells as judged microscopically after staining with 2.5% Evans Blue. Cells were normally assayed in a medium consisting of 0.7 M sorbitol, 50 mM Hepes-KOH (pH 8.0) buffer, 10 mM NaHCO₃, and 1 mM MgCl₂. Saturated rates of net O₂ evolution were routinely 50 to 70 μmol O₂/mg Chl · h.

O₂ exchange was measured polarographically or with a mass spectrometer using isotopically labeled O₂ (99 atom% ¹⁸O₂, obtained from Bio-Rad Laboratories). The instrumentation and

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expressions used for calculating rates of O₂ evolution and O₂ uptake have been described earlier (20, 21). All experiments were done using broad-band saturating orange-red light (Schott OG530 and appropriate heat filters) at approximately 20 C. Experiments with the mass spectrometer were generally run at elevated O₂ tensions (see figure legends).

RESULTS

O₂ Reduction in Isolated Chloroplasts. Figure 1A shows the calculated rates of O₂ evolution and O₂ uptake characteristically observed when dark-adapted chloroplasts were illuminated with near-saturating light. O₂ evolution began immediately upon illumination (within the ≤ 3 -s response time of the instrument). However, this initial rate (V_0) of O₂ evolution was counterbalanced by a comparable light-induced rate of O₂ uptake, so that little net O₂ was evolved. After 1 to 2 min in continuous light, the rate of O₂ evolution increased 3- to 4-fold to a maximum steady-state rate (V_m) while the rate of O₂ uptake diminished to a relatively insignificant level. Figure 1B shows the integrated time course of net O₂ evolution computed from the data of Figure 1A. These results are consistent with the typical induction lag measured with dark-adapted intact chloroplasts (or with whole cells) using a standard O₂ concentration electrode.

One could postulate that the observed light-driven O₂ uptake occurred via O₂ uptake reactions associated with the Calvin cycle, e.g. ribulose bis-P carboxylase/oxygenase. The results shown in Figure 2A indicate that this is not the case. Inhibition of the Calvin cycle with DL-glyceraldehyde (3, 25) had minimal effects on the light-induced rate of O₂ uptake or evolution (compared to V_0 rates observed in Fig. 1A). Both rates were sustained in continuous light for several minutes so that, observed polarographically, little net O₂ would be evolved. Similar results were obtained when net CO₂ fixation (and net O₂ evolution) was inhibited with high phosphate concentrations (data not shown). As illustrated in Figure 2B, light-induced rates of O₂ evolution and concomitant O₂ uptake were increased approximately 50% by the uncoupler methylamine.

Results equivalent to those of Figure 2B were also obtained with methylamine in the absence of DL-glyceraldehyde (data not shown); the uncoupler eliminated CO₂ fixation (and net O₂ evolution), and O₂ evolution and an equivalent O₂ uptake were sustained during illumination at a rate $\sim 50\%$ higher than that observed during the induction lag or in the presence of Calvin

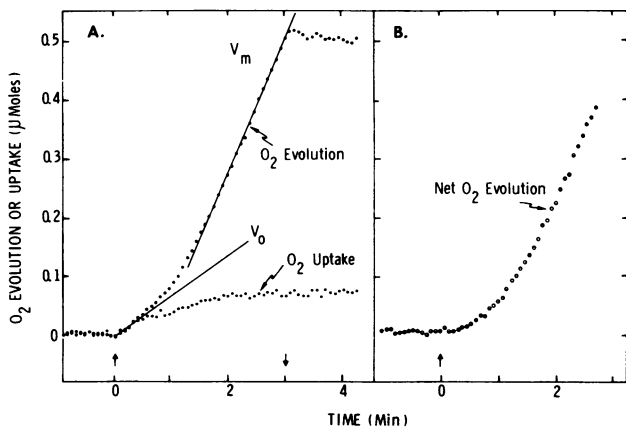


FIG. 1. A: light-induced O₂ evolution and O₂ uptake in intact chloroplasts. B: time course of net O₂ evolution computed from the data of Figure 1A. Standard chloroplast assay medium. Saturated net rates of O₂ evolution for this chloroplast preparation were 110 $\mu\text{mol O}_2/\text{mg Chl} \cdot \text{h}$. Actinic light on (\uparrow) and off (\downarrow). Total Chl concentration, 100 $\mu\text{g}/\text{ml}$. Initial total O₂ concentration 0.610 mM. Each point in this and the succeeding figures represents the value of O₂ evolution and O₂ uptake measured and computed as described in reference 21.

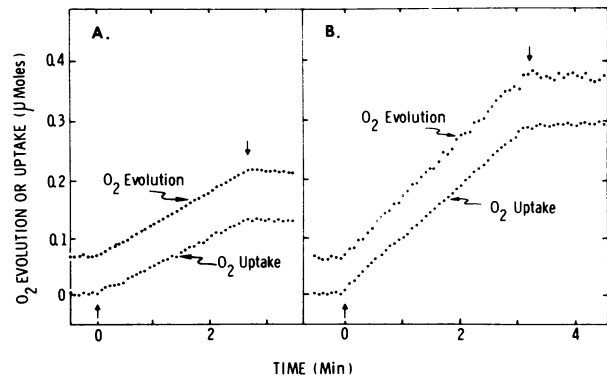


FIG. 2. Light-induced O₂ evolution and O₂ uptake in intact chloroplasts: (A) in the presence of 10 mM DL-glyceraldehyde, and (B) in the presence of 10 mM DL-glyceraldehyde plus 30 mM methylamine. Other conditions as in Figure 1. Actinic light on (\uparrow) and off (\downarrow). Initial O₂ concentration of 0.814 mM and 0.516 mM in A and B, respectively. Note that the plot of O₂ evolution was offset for clarity.

cycle inhibitors. These results are in agreement with numerous suggestions that pseudocyclic electron transport is coupled to ion transport and/or ATP synthesis (5-7, 10).

Our results indicate that the broken chloroplasts ($\leq 30\%$) present in our intact chloroplast preparations do not contribute significantly to the observed rates of O₂ uptake. The light-induced rate of O₂ uptake decreased to a low steady-state following induction (Fig. 1A). This decrease correlates with the increase in the rate of O₂ evolution, suggesting that electron transport to O₂ is diminished by noncyclic electron transport leading to CO₂ reduction. This conclusion is further corroborated by the results shown in Figure 3. Intact chloroplasts which were osmotically shocked (in the reaction vessel) had an O₂ uptake rate that was 32% of that observed with intact chloroplasts in the presence of DL-glyceraldehyde; washed broken chloroplasts preparations (type C[9]) had an even lower rate (14% of the DL-glyceraldehyde control). As expected, neither of these broken chloroplast preparations showed net O₂ evolution. These observations are consistent with the suggestions (1, 20) that the catalyst(s) involved in electron transport to O₂ is loosely membrane-bound.

O₂ Reduction in Intact Cells. We also measured the magnitude and kinetics of O₂ reduction in isolated intact spinach cells to determine whether the results obtained with isolated chloroplasts accurately reflected O₂ reduction *in vivo*. The results of a set of experiments, performed under conditions similar to those with isolated chloroplasts, are shown in Figures 4 and 5. It was necessary in these whole cell experiments to substitute the Calvin cycle inhibitor iodoacetamide for DL-glyceraldehyde and the uncoupler CCCP² for methylamine. With minor exceptions the whole cell data are comparable to those obtained with intact chloroplasts. Under optimum conditions for CO₂ fixation (Fig. 4A), O₂ evolution and O₂ uptake again began immediately upon illumination; O₂ uptake subsequently declined in the light to a very low steady-state rate; the initial rates of O₂ evolution and uptake did not quite balance. This is also apparent in the integrated time course for net O₂ evolution plotted in Figure 4B. Similar net O₂ kinetics (albeit with a slightly longer induction) were observed for this cell preparation when measured polarographically under approximately the same conditions (dashed curve in Fig. 4B). Presumably in both cases, CO₂ fixation was not completely inactivated at the onset of illumination.

A sustained light-induced rate of O₂ uptake, similar to that observed initially with bicarbonate, was obtained when the Calvin cycle was inhibited with iodoacetamide (Fig. 5A). Likewise, the rate of O₂ uptake was stimulated 40 to 50% by the uncoupler

² Abbreviation: CCCP: carbonylcyanide *m*-chlorophenylhydrazone.

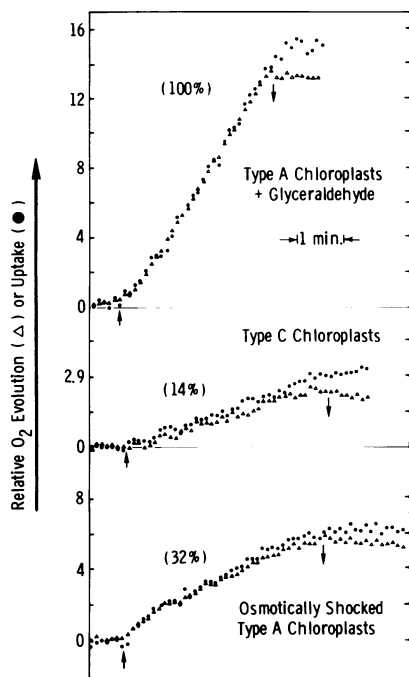


FIG. 3. Light-induced O_2 evolution (open markers) and O_2 uptake (solid markers) in three types (9) of chloroplast preparations. Broken (type C) chloroplasts were prepared according to reference 22. Osmotically shocked chloroplasts were prepared by placing intact (type A) chloroplasts in distilled H_2O for 5 min and then diluting with an equal volume of twice-concentrated assay medium. Standard assay medium containing 10 mM DL-glyceraldehyde and 50 $\mu\text{g}/\text{ml}$ total Chl. Actinic light on (\uparrow) and off (\downarrow). Note that the middle trace was plotted on a different relative scale and that the numbers in parentheses refer to relative absolute rates. Initial O_2 concentration of 0.431 mM, 0.202 mM, and 0.306 mM in upper, middle, and lower experiments, respectively.

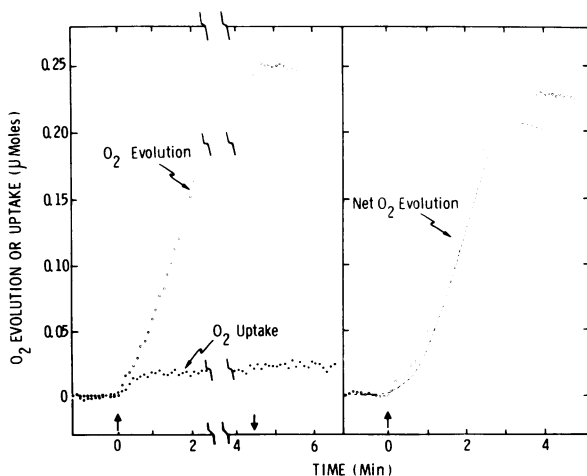


FIG. 4. A: light-induced O_2 evolution and O_2 uptake in isolated spinach cells. B: time course of net O_2 evolution computed from the data of Figure 4A (open markers) or recorded polarographically (dashed curve). Standard cell assay medium. Net rates of O_2 evolution for this cell preparation were 55 $\mu\text{mol } O_2/\text{mg Chl} \cdot \text{h}$. Actinic light on (\uparrow) and off (\downarrow). Total Chl concentration, 75 $\mu\text{g}/\text{ml}$. Initial O_2 concentration of 0.323 mM in A and 0.237 mM in polarographic measurement.

CCCP (Fig. 5B). In both cases, rates of O_2 uptake were accompanied by comparable (slightly greater) rates of O_2 evolution such that, as observed polarographically, little net O_2 was evolved.

Preliminary studies with isolated soybean cells have shown O_2 exchange kinetics similar to those described above for spinach. We should note, however, that our soybean preparations had

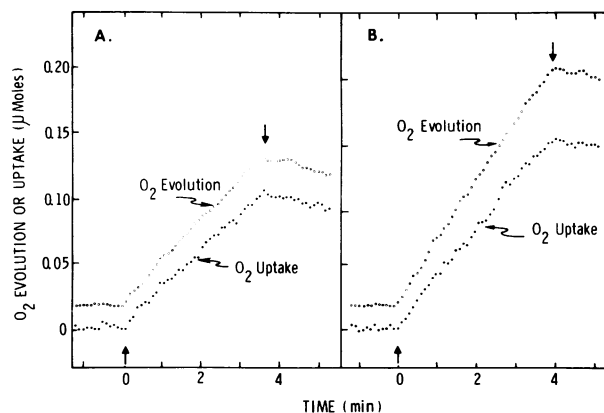


FIG. 5. Light-induced O_2 evolution and O_2 uptake in isolated spinach cells: (A) in the presence of 5 mM iodoacetamide and (B) in the presence of 1 μM CCCP. Other conditions as in Figure 4. Actinic light on (\uparrow) and off (\downarrow). Initial O_2 concentration of 0.408 mM and 0.468 mM in A and B, respectively. Note that the plot of O_2 evolution was offset for clarity.

higher maximum net rates of photosynthesis (70–100 $\mu\text{mol } O_2/\text{mg Chl} \cdot \text{h}$) and required turgid cells, *i.e.* lower sorbitol concentrations, for maximum rates of CO_2 fixation, compared to spinach (see also 23).

DISCUSSION

Our results with intact chloroplasts and cells from spinach support the earlier suggestion (20) that the direct photoreduction of O_2 appears to be a reaction common to all oxygenic photosynthetic organisms. Under the conditions of our experiments, *i.e.* near-saturating light, ample CO_2 and relatively high O_2 tensions, significant endogenous rates of O_2 reduction appear to occur principally when rates of CO_2 fixation are limited, *e.g.* during the initial lag phase following the onset of illumination or in the presence of components which interfere with normal CO_2 assimilation (DL-glyceraldehyde, methylamine, etc.). Thus, as in algae, O_2 appears to compete with CO_2 for photochemically generated reducing equivalents. The coupling of O_2 reduction to ion transport and/or phosphorylation and the normally low steady-state rate of O_2 uptake (in the presence of saturating amounts of CO_2) suggest that the electron flow via this pathway is tightly controlled and may principally function to prime the CO_2 reduction system. In weak light the direct photoreduction of O_2 has been reported to account for approximately 40% of the net steady-state rate of O_2 evolution during CO_2 reduction in intact spinach chloroplasts (5), although a more recent report from the same laboratory (11) indicates a much lower relative steady-state rate of O_2 uptake ($\leq 16\%$) with increased light intensities. Consistent with our findings, it was suggested that the relative direct contribution of pseudocyclic electron transport to the ATP demands of the Calvin cycle at steady-state in high light is minimal.

The most striking difference we observed between algae and higher plant systems is the considerably lower rates of O_2 photoreduction in spinach. We typically find endogenous O_2 reduction rates of approximately 25 to 50% of the maximum rate of noncyclic electron transport compared to O_2 uptake rates of $\geq 80\%$ of V_{max} in algae. Our calculated rates of O_2 reduction in both cases represent minimum estimates. The sampling system used in these experiments does not necessitate equilibration across a gas-liquid interface, but incomplete equilibration of the O_2 isotopes between the inside and outside of the chloroplasts (or cells) will result in an underestimation of the rate of O_2 reduction. However, the relatively simpler isolated chloroplast system and the reported (27) rapid equilibration of O_2 across chloroplast membranes particularly at elevated O_2 tensions as used in the present experiments would suggest that the observed differences in maximum rates of O_2 reduction are not an artifact.

The lower rates of O₂ reduction in spinach are not a result of substrate (O₂) limitations; preliminary experiments (unpublished) suggest an apparent K_m for O₂ similar to that found for *Scenedesmus* (21). It is interesting to note that previous studies with whole leaves (17, 18) have shown that rates of O₂ evolution are roughly equivalent to rates of O₂ uptake when measured in strong light at CO₂ compensation, and that these rates are approximately 30% of the rates of O₂ evolution observed at high CO₂ concentrations. These relative rates of O₂ exchange are strikingly similar to those we found in isolated chloroplasts and cells (comparing steady-state O₂ exchange rates with sufficient CO₂ in the presence and absence of Calvin cycle inhibitors), although one cannot exclude a possible involvement of carbon metabolism, e.g. glycolate metabolism in the whole leaf experiments. In this regard, recent studies with leaves and intact chloroplasts from spinach have suggested that a considerable fraction of O₂ uptake (at CO₂ compensation) can be associated with glycolate formation (4).

Under the conditions of our experiments, uncouplers weakly stimulated (~50%) electron transport to O₂. This increase is considerably less than can typically be demonstrated for noncyclic electron transport to other acceptors in intact chloroplasts, e.g. NO₂⁻ and oxaloacetate. We assume that this simply reflects a lower turnover rate for O₂ reduction in higher plants.

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