

# Polarographic Study of Oxaloacetate Reduction by Isolated Pea Chloroplasts

Received for publication February 1, 1979 and in revised form July 18, 1979

JOHN W. ANDERSON AND COLIN M. HOUSE  
*Botany Department, La Trobe University, Bundoora, Victoria 3083, Australia*

## ABSTRACT

Suspensions of pea chloroplasts, prepared by differential centrifugation, catalyzed oxaloacetate-dependent  $O_2$  evolution (mean rate of 29 determinations 10.9 micromoles per milligram of chlorophyll per hour, SD 3.2) with the concomitant production of malate. At optimum concentrations of oxaloacetate, both reactions were light-dependent, inhibited by 3-(3,4-dichlorophenyl)-1,1-dimethylurea and oxalate, and enhanced 2.5- to 4-fold by 10 millimolar  $NH_4Cl$ . At concentrations of oxaloacetate (<50 micromolar), 10 millimolar  $NH_4Cl$  was inhibitory. The ratio of  $O_2$  evolved to malate produced was 0.39 to 0.58. The ratio of  $O_2$  evolved to oxaloacetate supplied was commensurate with the theoretical value of 0.5.

Chloroplast suspensions contained both NAD- and NADP-malate dehydrogenase activities. It was concluded from oxalate inhibition studies and the promotion of oxaloacetate-dependent  $O_2$  evolution by shocked chloroplasts by NADPH (but not NADH) that the reaction was mediated via the NADP enzyme.

---

Crude suspensions of isolated chloroplasts prepared by differential centrifugation exhibit OAA<sup>1</sup>-dependent  $O_2$  evolution in the light (7–9, 18). This reaction is attributed to MDH activity coupled to the oxidation of a reductant produced by the light reactions (7, 9). However, with the exception of some important studies of the quantum requirements (6–8), few details are known about OAA-dependent  $O_2$  evolution and the presumed associated synthesis of malate.

The light-coupled MDH activity of chloroplasts has been implicated as an essential component of the OAA/malate shuttle which represents one of several shuttles for the transfer of reducing equivalents from chloroplasts to the cytoplasm and, under certain conditions, *vice versa* (7, 14, 17, 23). Addition of OAA to illuminated intact chloroplasts should initiate a stoichiometric evolution of  $O_2$  dependent on the light reactions, chloroplast-MDH, and pyridine nucleotides. According to the OAA/malate shuttle, addition of OAA to illuminated chloroplasts in the presence of a soluble extrachloroplast system to reoxidize malate exported from the chloroplasts, should stimulate continuing  $O_2$  evolution due to recycling of OAA via chloroplast-MDH. This paper describes a study of OAA-dependent  $O_2$  evolution (and the associated malate synthesis) by intact and osmotically shocked suspensions of crude washed chloroplasts under noncycling conditions.

In recent reviews of the OAA/malate shuttle (7, 14, 23) it is not clear whether NADH or NADPH serves to reduce OAA in the

chloroplast. This uncertainty stems primarily from early reports that NAD-MDH (see Yamazaki and Tolbert [24] for references), but not NADP-MDH (9), was associated with chloroplasts prepared by differential centrifugation and that the activity of NAD-MDH associated with chloroplast suspensions prepared by this technique was sufficient to account for the coupled and uncoupled rates of OAA-dependent  $O_2$  evolution (10). Density gradient centrifugation studies suggest that the NAD-MDH activity associated with crude suspensions of chloroplasts is associated with contaminating organelles (19, 24). The low activity of chloroplast NADP-MDH reported by Heber and Krause (9) can be attributed to activation *in vivo* by light and *in vitro* by various thiols (1, 4, 13, 21). A secondary aim of this study was to demonstrate that the OAA-dependent  $O_2$  evolution catalyzed by crude chloroplast suspensions was mediated via NADP-MDH.

## MATERIALS AND METHODS

**Chemicals.** LDH (in 7 M  $(NH_4)_2SO_4$  and MDH (in glycerol) were obtained from Boehringer-Mannheim, West Germany. GSH reductase (in 3.6 M  $(NH_4)_2SO_4$ ) was obtained from Sigma. All acidic substrates were adjusted to pH 7 with KOH. The concentration of solutions of OAA was ascertained immediately prior to use by reduction of NADH at 340 nm in 0.1 M Tris-HCl buffer (pH 7.6) in the presence of MDH.

**Chloroplasts and  $O_2$  evolution.** Pea seedlings (*Pisum sativum* cv. Massey Gem) were raised in a growth cabinet as before (11) and harvested 13 or 14 days after imbibition. Chloroplasts were prepared by a modification of the method used previously (2) but which incorporated certain of the procedures of Nakatani and Barber (16). In summary, pea shoots (100 g) were blended in 250 ml of an icy slush of extracting medium (2) but with  $MgCl_2$  at 0.2 mM. Chloroplasts were recovered by centrifugation at 2,500g for 30 s. A solution (50 ml) containing 0.33 M glucose, 0.1% (w/v) BSA, and 0.04 volumes of incubating medium (2) was added and gently swirled over the pellet. The washings were discarded and the pellet resuspended in 0.33 M sorbitol adjusted to pH 7.6 with Tris. After centrifuging at 2,500g for 20 s the supernatant solution was aspirated and the washed chloroplasts resuspended in 1 to 2 ml of incubating medium (2). Unless specified otherwise, these chloroplasts (referred to as crude washed chloroplasts) were used in all experiments. Purified chloroplasts were prepared by layering crude washed chloroplasts onto a discontinuous gradient of Ficoll 400 (30, 32, and 40%, w/v) in incubating medium and centrifuging at 5,000g for 7.5 min. Broken chloroplasts from the top of the gradient and purified chloroplasts from the 30/32% Ficoll interface were diluted 4-fold in incubating medium and, after centrifuging at 5,000g for 3 min, resuspended in incubating medium.

$O_2$  evolution by chloroplasts was measured polarographically at 25°C in a pair of  $O_2$  electrodes supplied by Hansatech, Norfolk, England (model DW) and manufactured to the design of Delieu and Walker (5). Chloroplast intactness was measured by the ratio of the  $O_2$  evolution rates of osmotically shocked and unshocked

---

<sup>1</sup> Abbreviations: OAA: oxaloacetate; NAD-MDH: NAD-specific malate dehydrogenase; NADP-MDH: NADP-specific malate dehydrogenase; LDH: lactate dehydrogenase; GSH: reduced glutathione; GSSG: oxidized glutathione;  $\alpha$ -kG:  $\alpha$ -ketoglutarate.

chloroplasts in the presence of 10 mM  $\text{NH}_4\text{Cl}$  using ferricyanide as electron acceptor (15). Chl was measured in ethanol as before (2).

Sonicated chloroplasts were prepared by treating intact chloroplasts, suspended in incubating medium, with a Rapidis 600 sonicator (Ultrasonics Ltd., Yorks, England) for 40 s. The soluble fraction (chloroplast extract) was recovered by centrifugation at 12,000g for 10 min.

**Composition of Incubation Mixtures for  $\text{O}_2$  Evolution by Illuminated Chloroplasts.** Reactions were conducted in incubating medium containing 0.33 M sorbitol, 50 mM Hepes, 0.1% (w/v) BSA, 2 mM  $\text{Na}_2\text{EDTA}$ , 1 mM  $\text{MgCl}_2$ , 1 mM  $\text{MnCl}_2$  adjusted to pH 7.6 with KOH (2). Correct concentrations were maintained by addition of double strength incubating medium where necessary. Standard incubations were conducted in a volume of 0.5 ml containing 100  $\mu\text{g}$  Chl/ml. DL-Glyceraldehyde (10 mM) was added to inhibit endogenous  $\text{O}_2$  evolution (20). OAA-dependent  $\text{O}_2$  evolution by intact chloroplasts was initiated with 0.4 to 0.8 mM OAA. The uncoupled rate was determined in the presence of 10 mM  $\text{NH}_4\text{Cl}$ .

(OAA plus NADPH)-dependent  $\text{O}_2$  evolution was determined with osmotically shocked chloroplasts in the presence of 10 mM DL-glyceraldehyde. Intact chloroplast suspensions (1 volume) were diluted with approximately 2.5 volumes of water. NADPH (0.1 mM) was added to incubation mixtures (0.5 ml) containing shocked chloroplasts (220–300  $\mu\text{g}$  Chl/ml). This effected the evolution of a small amount of  $\text{O}_2$  which ceased within 30 s. (OAA plus NADPH)-dependent  $\text{O}_2$  evolution was initiated by addition of OAA (0.8 mM).

**Determination of Enzyme Activities in Chloroplast Extract.** Unless specified otherwise, chloroplast extract was preincubated at 0 C for 40 min with 5 mM DTT and 5 mM  $\text{MgCl}_2$  (13) prior to determining enzyme activities (treated extract). NAD-MDH and NADP-MDH were determined by monitoring OAA-dependent oxidation of NADH and NADPH at 340 nm at 25 C. Reaction mixtures (1 ml) contained 2 mM OAA, 0.15 mM NADH or 0.15 mM NADPH, 87.5 mM Tris-HCl (pH 7.6), and 10  $\mu\text{l}$  treated extract (equivalent to approximately 8–10  $\mu\text{g}$  Chl).

Two methods were used to determine transhydrogenase, and involved transhydrogenation from NADPH to NAD and NADH to NADP, respectively. For the first method, reaction mixtures (1 ml) contained 0.15 mM NADPH, 0.15 mM NAD, 2 mM pyruvate, 94 mM Tris-HCl (pH 7.6), and approximately 4 units LDH. For the second method reaction mixtures (1 ml) contained 0.15 mM NADH, 0.15 mM NADP, 0.3 mM GSSG, 90 mM Tris-HCl (pH 7.6), and approximately 5 units GSH reductase. For both methods, reactions were initiated with 10  $\mu\text{l}$  chloroplast extract and monitored at 340 nm at 25 C and activity corrected for controls lacking chloroplast extract.

**Determination of Malate.** Samples (0.35 ml) from chloroplast incubation mixtures were treated with 0.117 ml 6% (w/v)  $\text{HClO}_4$  and adjusted to pH 7 with KOH. Insoluble material was removed by centrifugation and malate in the supernatant solution was determined by the method of Hohorst (10).

## RESULTS

**Properties of Chloroplasts.** The mean uncoupled rates of  $\text{O}_2$  evolution for crude washed chloroplasts from cv. Massey Gem for ferricyanide and OAA were 205 (SD 30) and 37.2 (SD 8.6)  $\mu\text{mol}/\text{mg Chl}\cdot\text{h}$ , respectively. The coupled rates for OAA and nitrite were 10.9 (SD 3.2) and 8.8 (SD 1.9)  $\mu\text{mol}/\text{mg Chl}\cdot\text{h}$ , respectively, and the mean intactness 79% (SD 8). The rates are similar to those determined for cv. Massey Gem by the method used previously (11) but in general are 50 to 60% less than those determined for chloroplasts from cv. Feltham First (2, 3). However, the intactness was appreciably greater than that determined by the previous method for either cv.

**Some Properties of OAA-dependent  $\text{O}_2$  Evolution by Intact Chloroplasts.** OAA-dependent  $\text{O}_2$  evolution did not proceed in the dark and was inhibited (100%) by 1.7  $\mu\text{M}$  DCMU. In the presence of 0.8 mM OAA the rate was constant for approximately 3 to 5 min and thereafter gradually decreased. The initial rate increased with the concentration of OAA up to approximately 50  $\mu\text{M}$  (Fig. 1A);  $V_{\text{max}}/2$  was estimated at approximately 15  $\mu\text{M}$ . The rate was essentially independent of OAA concentration between 0.5 and 2.5 mM but higher concentrations were slightly inhibitory (e.g. 16% inhibition at 4.5 mM). When small amounts of OAA were supplied (up to 200 nmol/ml),  $\text{O}_2$  evolution commenced in the usual way but ceased abruptly after a few minutes. Further additions of OAA reinitiated  $\text{O}_2$  evolution (Fig. 2A) implying that the reaction had come to equilibrium after cessation of  $\text{O}_2$  evolution. For the first addition of OAA, the ratio of  $\text{O}_2$  evolved to OAA supplied approximated the theoretical value of 0.5 for the reduction of OAA to malate but both the ratio and the rate decreased with each successive addition of OAA (Fig. 2A). The abrupt cessation of  $\text{O}_2$  evolution after each addition of OAA is in agreement with the high affinity of the reaction for OAA (Fig. 1A). Intact chloroplasts commenced  $\text{O}_2$  evolution at the maximum rate within 5 s of supplying OAA.

Neither malate (10–40 mM) nor the transaminase inhibitor amino oxycetate (5 mM) significantly affected the rate of OAA-dependent  $\text{O}_2$  evolution; amino oxycetate was also without significant effect on the  $\text{O}_2$  to OAA ratio. However, oxalate inhibited  $\text{O}_2$  evolution by intact chloroplasts (Fig. 3C); the concentration required to effect 50% inhibition was approximately 4 to 5 mM. Chloroplasts did not catalyze  $\text{O}_2$  evolution when OAA was replaced with  $\alpha$ -kG or pyruvate (2 mM) nor did these compounds affect the rate of OAA-dependent  $\text{O}_2$  evolution.

The effect of  $\text{NH}_4\text{Cl}$  on OAA-dependent  $\text{O}_2$  evolution varied with the concentration of OAA (Fig. 1B) and  $\text{NH}_4\text{Cl}$  (Fig. 4). In the presence of 0.3 to 2 mM OAA, the rate increased 2.5- to 4-fold with  $\text{NH}_4\text{Cl}$  concentrations up to approximately 2 mM; higher concentrations (up to 20 mM) were not inhibitory. At lower

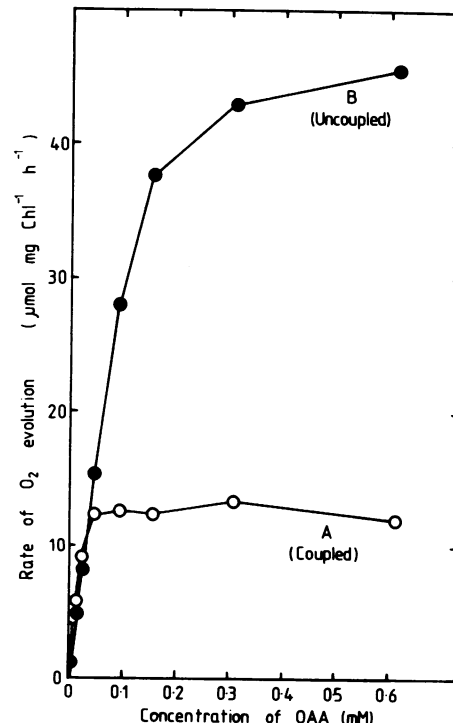


FIG. 1. Effect of concentration of OAA on the rate of  $\text{O}_2$  evolution by illuminated intact chloroplasts in (A) the absence and (B) the presence of 10 mM  $\text{NH}_4\text{Cl}$ . Chloroplast intactness: 76%; Chl concentration: 100  $\mu\text{g}/\text{ml}$ .

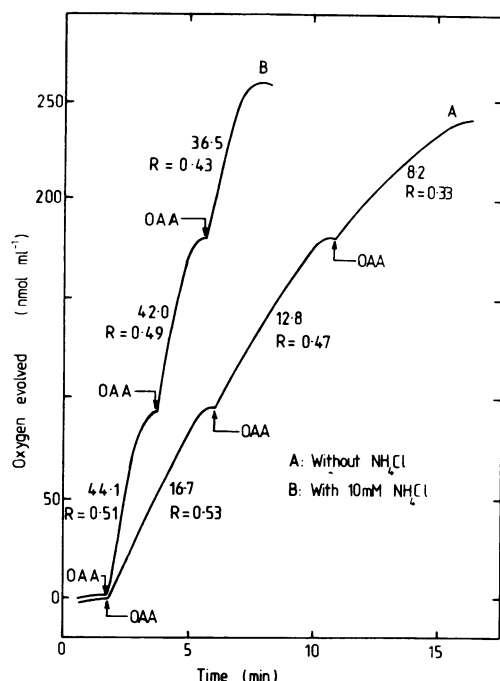


FIG. 2. Effect of limiting amounts of OAA on  $O_2$  evolution by illuminated intact chloroplasts in (A) the absence and (B) the presence of 10 mM  $NH_4Cl$ . Reactions were timed from the moment of illumination and OAA (179 nmol/ml) was added as shown. Values beside the curves represent rates of  $O_2$  evolution in  $\mu\text{mol}/\text{mg Chl}\cdot\text{h}$ . Molar ratios of  $O_2$  evolved to OAA supplied (R) for each successive addition of OAA are also shown. Chloroplast intactness: 75%; Chl concentration: 100  $\mu\text{g}/\text{ml}$ .

concentrations of OAA, however, the enhancement effect was proportionately less and at concentrations less than 50  $\mu\text{M}$  OAA,  $NH_4Cl$  (10 mM) was slightly inhibitory (Fig. 1).  $NH_4Cl$  (10 mM) did not significantly affect the  $O_2$  to OAA ratio when OAA was supplied at 0.179 mM but the decrease in the ratio and rate of  $O_2$  evolution for subsequent additions of OAA (Fig. 2B) was relatively much less than for reaction mixtures lacking  $NH_4Cl$ .

**Light-dependent Synthesis of Malate from OAA by Intact Chloroplasts.** OAA-dependent  $O_2$  evolution was accompanied by the synthesis of malate. The ratio for  $O_2$  evolved to malate synthesized by coupled chloroplasts was 0.39 to 0.58 (Table I). Malate synthesis was light-dependent, inhibited by 1.7  $\mu\text{M}$  DCMU and 61 mM oxalate, and did not occur in the absence of OAA.  $NH_4Cl$  (10 mM) enhanced the rate of malate synthesis approximately 2-fold.

**Enzyme Activities of Sonicated Chloroplasts.** The activities of several enzymes associated with the soluble fraction prepared from sonicated chloroplasts were compared with the coupled and uncoupled rates of OAA-dependent  $O_2$  evolution. The rates of NAD-MDH, after correcting for the appropriate stoichiometry, were sufficient to account for the uncoupled rates of OAA-dependent  $O_2$  evolution (Table II). The NADP-MDH activity of untreated chloroplast extract was very low but when the extract was pretreated with 5 mM DTT and 5 mM  $MgCl_2$  at 0 C for 40 min, enzyme activity was enhanced approximately 25- to 50-fold. The activity was sufficient to account for the coupled rate of  $O_2$  evolution but in most experiments the activity did not fully account for the uncoupled rate (Table II). Transhydrogenase activity (NADH to NADP and NADPH to NAD) was not detected in treated chloroplast extracts using the glutathione reductase or LDH assays.

Both the NAD-MDH and NADP-MDH activities of treated chloroplast extract were inhibited by oxalate (Fig. 3, A and B); NADP-MDH (50% inhibition at approximately 4 to 5 mM) was more sensitive than NAD-MDH (50% inhibition at approximately

45 to 50 mM).  $NH_4Cl$  (10 mM) had no significant effect on the NAD-MDH and NADP-MDH activities of treated extract as determined by the spectrophotometric assay.

**Properties of Purified Chloroplasts.** The initial rate of OAA-

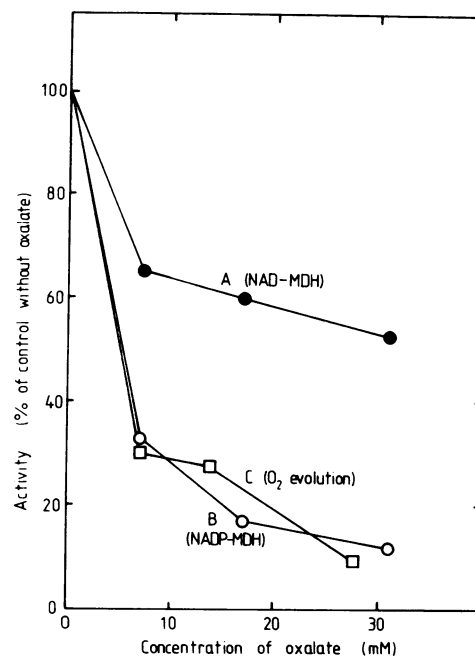


FIG. 3. Effect of concentration of oxalate on (A) NAD-MDH and (B) NADP-MDH activity of chloroplast extract and (C) uncoupled rate of OAA-dependent  $O_2$  evolution by illuminated intact chloroplasts. Chloroplast extract (from 77% intact chloroplasts) was preincubated with 5 mM DTT and 5 mM  $MgCl_2$  at 0 C for 40 min prior to determining the enzyme activities (A and B). The activities of NAD-MDH and NADP-MDH in the absence of oxalate were 209 and 52  $\mu\text{mol NAD(P)H oxidized}/\text{mg Chl}\cdot\text{h}$ , respectively. For  $O_2$  evolution, chloroplasts (72% intact) were preincubated in the light at 25 C for 5 min in the presence of 10 mM  $NH_4Cl$  and oxalate before initiating the reaction with 0.8 mM OAA. The uncoupled rate of OAA-dependent  $O_2$  evolution in the absence of oxalate was 19.0  $\mu\text{mol}/\text{mg Chl}\cdot\text{h}$ . All results are expressed as a percentage of the rates of appropriate controls lacking oxalate.

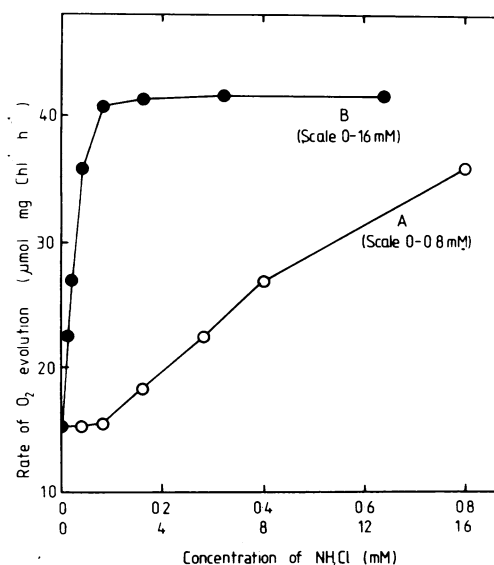


FIG. 4. Effect of concentration of  $NH_4Cl$  on OAA-dependent  $O_2$  evolution by illuminated intact chloroplasts. Reaction mixtures contained 100  $\mu\text{g Chl}/\text{ml}$  (chloroplast intactness 67%) and  $O_2$  evolution was initiated with 0.8 mM OAA. A: scale 0 to 0.8 mM; B: scale 0 to 16 mM.

Table I. OAA-dependent Malate Synthesis and O<sub>2</sub> Evolution by Illuminated Intact Chloroplasts

Reaction mixtures contained 100  $\mu\text{g}$  Chl/ml and were initiated with 0.8 to 0.96 mM OAA. Rates for O<sub>2</sub> evolution represent average values for the time periods specified.

Experiment No.	Chloroplast Intactness %	Additions	Time Sampled min	Malate Synthesized $\mu\text{mol/mg Chl}\cdot\text{h}$	O <sub>2</sub> Evolved $\mu\text{mol/mg Chl}\cdot\text{h}$	Ratio O <sub>2</sub> Evolved to Malate Synthesized
1	71	Nil	8	20.3	11.8	0.58
		NH <sub>4</sub> Cl (10 mM)	4	43.0	24.5	0.57
		DCMU (3.3 $\mu\text{M}$ )	9	2.3	0.9	(0.39)
		Oxalate (61 mM)	8	7.6	0	
		Dark	8	1.4	0	
2	88	Nil	16	10.2	4.5	0.44
		NH <sub>4</sub> Cl (10 mM)	7.2	23.1	9.1	0.39
3	79	Nil	14	14.4	7.0	0.49
		Dark	14	1.1	0	

Table II. NAD-MDH and NADP-MDH Activities Associated with Chloroplast Extracts in Relation to Coupled and Uncoupled Rates of OAA-dependent O<sub>2</sub> Evolution of Intact Chloroplasts

Chloroplast extracts were prepared from sonicated chloroplasts and O<sub>2</sub> evolution was measured under standard conditions. Extracts were pretreated with 5 mM DTT and 5 mM MgCl<sub>2</sub> at 0°C for 40 min where indicated. Values in parentheses in columns 3 and 4 represent the rate of OAA reduction as calculated from the observed rate of O<sub>2</sub> evolution (theoretical stoichiometry 2.0).

Experiment No.	Intact Chloroplasts		Chloroplast Extract			
	Intactness %	O <sub>2</sub> evolution $\mu\text{mol/mg Chl}\cdot\text{h}$		Pretreatment	Enzyme activities $\mu\text{mol/mg Chl}\cdot\text{h}$	
		Without NH <sub>4</sub> Cl	With 10 mM NH <sub>4</sub> Cl		NAD-MDH	NADP-MDH
1	84	13.1 (26.2)	40.2 (80.4)	Nil	271	1
2	86	9.6 (19.2)	35.3 (70.6)	DTT + MgCl <sub>2</sub>	283	49
3	77	9.2 (18.4)	38.9 (77.8)	DTT + MgCl <sub>2</sub>	283	105
4	69	14.7 (29.4)	41.2 (82.4)	Nil	217	0
				DTT + MgCl <sub>2</sub>	209	46
				DTT + MgCl <sub>2</sub>	270	2.3
				DTT + MgCl <sub>2</sub>	272	60

dependent O<sub>2</sub> evolution varied from 0 to 17.7  $\mu\text{mol/mg Chl}\cdot\text{h}$  and, when active, decreased to zero after 5 to 7 min. NH<sub>4</sub>Cl did not enhance OAA-dependent O<sub>2</sub> evolution. Relative to crude chloroplasts, purification enhanced the NADP-MDH to NAD-MDH ratio up to 4-fold but the activity of either enzyme was invariably sufficient to account for the observed OAA-dependent O<sub>2</sub> evolution. Broken chloroplasts, which also contained relatively high levels of NADP-MDH and NAD-MDH, did not catalyze OAA-dependent O<sub>2</sub> evolution.

**(OAA plus NADPH)-dependent O<sub>2</sub> Evolution by Osmotically Shocked Chloroplasts.** Partially shocked chloroplast suspensions were prepared by diluting crude chloroplast suspensions in approximately 2.5 volumes of water. This procedure, which decreased the intactness to 15 to 20%, was adopted to obtain a system which, in the absence of any additions, did not catalyze high rates of O<sub>2</sub> consumption in the light. Completely shocked or sonicated chloroplasts (0% intact) catalyzed high rates of O<sub>2</sub> consumption (up to 12  $\mu\text{mol/mg Chl}\cdot\text{h}$ ). (OAA plus NADPH)-dependent O<sub>2</sub> evolution was not detected at less than 220  $\mu\text{g}$  Chl/ml.

Partially shocked chloroplasts failed to catalyze OAA-dependent O<sub>2</sub> evolution but when NADPH (0.02–0.2 mM) was supplied O<sub>2</sub> evolution commenced immediately. The rate decreased during the first 30 s but then stabilized at 2.5 to 4.1  $\mu\text{mol/mg Chl}\cdot\text{h}$  for a period of 3 to 7 min before gradually decreasing to zero (Fig. 5A). NADH would not replace NADPH in this reaction (Fig. 5C).

In the absence of OAA, NADPH initiated O<sub>2</sub> evolution which ceased within 30 s (Fig. 5B). The amount of O<sub>2</sub> evolved varied with the amount of NADPH supplied (Table III, column 2). Subsequent addition of OAA effected an immediate resumption of O<sub>2</sub> evolution at rates of 1.9 to 5.4  $\mu\text{mol/mg Chl}\cdot\text{h}$ . These rates were sustained for 3 to 7 min, the time varying with the amount of NADPH supplied (Table III, column 4), before the rate declined. The initial rates of O<sub>2</sub> evolution were not highly dependent on the concentration of NADPH although 0.06 mM was optimal (Table III). Except at 0.2 mM NADPH, the amount of OAA-dependent O<sub>2</sub> evolution exceeded the amount of NADPH supplied (Table III, column 6), thus demonstrating that NADPH served a catalytic function. The results in Figure 5C and Table III demonstrate that the high but rapidly declining rate of O<sub>2</sub> evolution

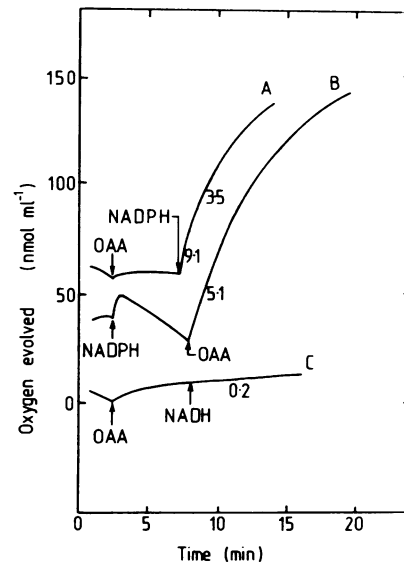


FIG. 5. Requirement for NADPH and OAA to effect O<sub>2</sub> evolution by partially shocked chloroplasts. Chloroplasts (80% intact) were shocked as described under "Materials and Methods" and incubated in the light in the presence of 10 mM DL-glyceraldehyde. Reaction mixtures (A–C) contained 233  $\mu\text{g}$  Chl/ml and additions were made as shown; OAA was supplied at 0.8 mM and NADPH and NADH at 0.1 mM. Values beside the curves represent the rate of O<sub>2</sub> evolution in  $\mu\text{mol/mg Chl}\cdot\text{h}$ .

Table III. Effect of Concentration of NADPH on OAA-dependent O<sub>2</sub> Evolution by Partially Shocked Chloroplasts

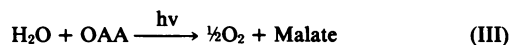
Chloroplasts (71% intact) were partially shocked as described under "Materials and Methods" and incubated with 10 mM DL-glyceraldehyde and NADPH at the concentrations shown. A small amount of O<sub>2</sub> was evolved during this period (column 2) but ceased after approximately 0.5 min. After a further 0.5 min, O<sub>2</sub> evolution was initiated by addition of 0.8 mM OAA. The initial rate of O<sub>2</sub> evolution (column 3) and the total amount of O<sub>2</sub> evolved at the time the reaction ceased (column 5) were recorded. All reactions were conducted in the light and contained 224  $\mu\text{g}$  Chl/ml. A progress curve for a similar experiment is shown in Figure 5B.

Concentration of NADPH	O <sub>2</sub> Evolved Prior to Adding OAA	Initial Rate OAA-dependent O <sub>2</sub> Evolution	Time Initial O <sub>2</sub> Evolution Rate Sustained	Total OAA-dependent O <sub>2</sub> Evolution at Cessation of Reaction	Ratio OAA-dependent O <sub>2</sub> Evolution to NADPH Supplied
nmol/ml		$\mu\text{mol/mg Chl}\cdot\text{h}$	min	nmol/ml	
0	0	0	0	0	
20	0	3.3	3	97	4.9
60	1.9	4.1	3	103	1.7
80	3.2	2.9	7	105	1.3
100	3.8	2.8	6	127	1.3
200	9.5	2.5	7	95	0.5

observed after addition of NADPH in Figure 5A was due to a short but rapid NADPH-dependent O<sub>2</sub> evolution (presumably due to an endogenous substrate [s]) in addition to the more sustained rate of OAA plus NADPH-dependent O<sub>2</sub> evolution. When OAA was supplied to partially shocked chloroplasts which had been preincubated in the light for various times, the rates of OAA-dependent O<sub>2</sub> evolution increased with the duration of the preincubation period (Table IV).

### DISCUSSION

OAA-dependent O<sub>2</sub> evolution in the light was attributed to photosynthetically coupled NADP-MDH activity.



The ratio of O<sub>2</sub> evolved to OAA consumed as determined by OAA-limiting experiments (Fig. 2A) and the ratio of O<sub>2</sub> evolved to malate produced (Table I) were in approximate agreement with the theoretical stoichiometry (reaction III). Like OAA-dependent O<sub>2</sub> evolution, the associated production of malate was light-dependent, inhibited by DCMU, and stimulated by 2 to 20 mM NH<sub>4</sub>Cl (Table I). The value of 15 μM for  $V_{max}/2$  for coupled OAA-dependent O<sub>2</sub> evolution is in approximate agreement with the  $K_m$  (apparent) for thiol-activated NADP-MDH from maize (25 μM) and spinach (30 μM) chloroplasts (13, 21). It is somewhat less than the value of 50 μM for the  $K_m$  (apparent) of the OAA translocator of spinach chloroplasts (7). This suggests that although the OAA translocator is operating at a smaller proportion of its maximum potential rate than NADP-MDH, it nevertheless does not affect the kinetics of the over-all reaction as judged by O<sub>2</sub> evolution. The coupled and uncoupled rates of OAA-dependent O<sub>2</sub> evolution (Table I) are commensurate with other values for pea chloroplasts (2, 18) but less than those reported for spinach (7, 8). The enhanced rate of O<sub>2</sub> and malate synthesis by 2 to 20 mM NH<sub>4</sub>Cl at high concentrations of OAA (0.5–2.5 mM), implies that photophosphorylation is not required for the reaction *per se*.

Since purified chloroplasts are reported to contain NADP-MDH (21) but not NAD-MDH (19, 24), NADP-MDH is most probably the reductant in reaction II. Crude washed chloroplasts,

which are effective in OAA-dependent O<sub>2</sub> evolution, contain presumably contaminating NAD-MDH (19, 24) at sufficient levels to account for the observed rates of OAA-dependent O<sub>2</sub> evolution whereas the NADP-MDH activity was not always sufficient (Table III). Rapid purification of chloroplasts on a discontinuous Ficoll gradient enhanced the ratio of NADP-MDH to NAD-MDH but purified chloroplasts did not consistently support OAA-dependent O<sub>2</sub> evolution and any activity could be accounted for by either enzyme. The relatively low levels of NAD-MDH associated with intact spinach chloroplasts purified by sucrose density gradient centrifugation, attributed to contamination (24), are also sufficient to account for the OAA-dependent O<sub>2</sub> evolution by crude pea chloroplasts. It was therefore necessary to seek other evidence to establish which enzyme was involved in OAA-dependent O<sub>2</sub> evolution in crude pea chloroplast preparations. Participation of NAD-MDH was made unlikely or ruled out by a series of observations. Addition of NADPH, but not of NADH, effected continuing OAA-dependent O<sub>2</sub> evolution by shocked chloroplasts (Table III and Fig. 5A). The kinetic data of oxalate inhibition of OAA-dependent O<sub>2</sub> evolution and NADP-MDH, but not NAD-MDH were similar (Fig. 3). The enhanced initial rates of OAA-dependent O<sub>2</sub> evolution by shocked chloroplasts with added NADPH following incubation in the light (Table IV) is consistent with the light activation of NADP-MDH of chloroplasts (1, 13). The absence of transhydrogenase activity in the chloroplasts demonstrates that transfer of a reducing equivalent from light-generated NADPH to NAD is not involved in OAA-dependent O<sub>2</sub> evolution. We presume that the discrepancy between the uncoupled rate of OAA-dependent O<sub>2</sub> evolution by intact chloroplasts and NADP-MDH of chloroplast extract (Table II) was due to incomplete extraction of the enzyme, incomplete activation of the enzyme by DTT and/or nonoptimum conditions for determining NADP-MDH activity.

As judged by spectrophotometric assay of chloroplast extracts prepared from chloroplasts following a period of illumination, complete activation of NADP-MDH is achieved in 60 s (1). We invariably observed commencement of OAA-dependent O<sub>2</sub> evolution at maximum rates within 1 to 5 s of supplying OAA to intact chloroplasts, irrespective of the period of preillumination. This method of monitoring NADP-MDH in intact chloroplasts, which does not require time to prepare extracts for estimating activity in secondary incubations, suggests that complete activation was achieved within 5 s. The relatively long preillumination periods required to effect OAA-dependent O<sub>2</sub> evolution at maximum rates of shocked chloroplasts suggest that the concentration of a membrane-bound or soluble factor(s) (1, 4, 12, 22) is rate-limiting in effecting light modulation of NADP-MDH in this system.

Table IV. Effect of Preillumination of Partially Shocked Chloroplasts on OAA-dependent O<sub>2</sub> Evolution

Chloroplasts were partially shocked as described under "Materials and Methods" and preincubated in the light for the times shown in the presence of 10 mM DL-glyceraldehyde and NADPH. O<sub>2</sub> evolution was initiated at the end of the preillumination period by addition of OAA. The chloroplast preparations prior to osmotic shock for experiments 1 and 2 were 72 and 84% intact, respectively. Reaction mixtures for both experiments contained 300 μg Chl/ml. For experiments 1 and 2 the concentrations of NADPH were 0.1 and 0.2 mM, respectively. For OAA, the corresponding values were 0.8 and 0.32 mM.

Experiment No.	Period of Preillumination	Initial Rate of OAA-dependent O <sub>2</sub> Evolution
	min	
1	4	2.6
	7	3.5
	10	5.1
	13	5.4
2	4	1.9
	7	2.3
	12	3.2
	20	3.4

### LITERATURE CITED

- ANDERSON LE, M AVRON 1976 Light modulation of enzyme activity in chloroplasts. Generation of membrane-bound vicinal dithiol groups by photosynthetic electron transport. *Plant Physiol* 57: 209–213
- ANDERSON JW, J DONE 1977 Polarographic study of glutamate synthase activity in isolated chloroplasts. *Plant Physiol* 60: 354–359
- ANDERSON JW, J DONE 1978 Light-dependent assimilation of nitrite by isolated pea chloroplasts. *Plant Physiol* 61: 693–697
- ANDERSON LE, SC NEHRlich, M-L CHAMPIGNY 1978 Light modulation of enzyme activity. Activation of the light effect mediators by reduction and modulation of enzyme activity by thiol-disulfide exchange. *Plant Physiol* 61: 601–605
- DELIEU T, DA WALKER 1972 An improved cathode for the measurement of photosynthetic oxygen evolution by isolated chloroplasts. *New Phytol* 71: 201–225
- HEBER U 1973 Stoichiometry of reduction and phosphorylation during illumination of intact chloroplasts. *Biochim Biophys Acta* 305: 140–152
- HEBER U 1974 Metabolite exchange between chloroplasts and cytoplasm. *Annu Rev Plant Physiol* 25: 393–421
- HEBER U, MR KIRK 1975 Flexibility of coupling and stoichiometry of ATP formation in intact chloroplasts. *Biochim Biophys Acta* 376: 136–150
- HEBER U, GH KRAUSE 1971 Transfer of carbon, phosphate energy and reducing equivalents across the chloroplast envelope. In MD Hatch, CB Osmond, RO Slatyer, eds, *Photosynthesis and Photorespiration*. Wiley-Interscience, New York, pp 218–225
- HOHORST H-J 1963 L(-)-Malate: determination with malic dehydrogenase and DPN. In HU

- Bergmeyer, ed, *Methods of Enzymatic Analysis*. Academic Press, New York, pp 328-332
11. JABLONSKI PP, JW ANDERSON 1978 Light-dependent reduction of oxidized glutathione by ruptured chloroplasts. *Plant Physiol* 67: 221-225
  12. JACQUOT JP, J VIDAL, P GADAL 1976 Identification of a protein factor involved in dithiothreitol activation of NADP malate dehydrogenase from french bean leaves. *FEBS Lett* 71: 223-227
  13. JOHNSON HS, MD HATCH 1970 Properties and regulation of leaf nicotinamide adenine dinucleotide phosphate malate dehydrogenase and "malic" enzyme in plants with the C<sub>4</sub>-dicarboxylic acid pathway of photosynthesis. *Biochem J* 119: 273-280
  14. KRAUSE GH, U HEBER 1976 Energetics of intact chloroplasts. *In* J Barber, ed, *The Intact Chloroplast*. Elsevier, Amsterdam, pp 171-214
  15. LILLEY RMCC, MP FITZGERALD, KG RIENITS, DA WALKER 1975 Criteria of intactness and photosynthetic activity of spinach chloroplast preparations. *New Phytol* 75: 1-10
  16. NAKATANI HY, J BARBER 1977 An improved method for isolating chloroplasts retaining their outer membranes. *Biochim Biophys Acta* 461: 510-512
  17. RATHNAM CKM, BA ZILINSKAS 1977 Reversal of 3-(3,4-dichlorophenyl)-1,1-dimethyl urea inhibition of carbon dioxide fixation in spinach chloroplasts and protoplasts by dicarboxylic acids. *Plant Physiol* 60: 51-53
  18. ROBINSON SP, JT WISKICH 1976 Stimulation of carbon dioxide fixation in isolated pea chloroplasts by catalytic amounts of adenine nucleotides. *Plant Physiol* 58: 156-162
  19. ROCHA V, IP TING 1970 Preparation of cellular plant organelles from spinach leaves. *Arch Biochem Biophys* 140: 398-407
  20. STOKES DM, DA WALKER 1972 Photosynthesis by isolated chloroplasts. Inhibition by DL-glyceraldehyde of carbon dioxide assimilation. *Biochem J* 128: 1147-1157
  21. TING IP, V ROCHA 1971 NADP-specific malate dehydrogenase of green spinach leaf tissue. *Arch Biochem Biophys* 147: 156-164
  22. VIDAL J, JP JACQUOT, H MEMBRE, P GADAL 1978 Detection and study of protein factors involved in dithiothreitol activation of NADP-malate dehydrogenase from a C<sub>4</sub> plant. *Plant Sci Lett* 11: 305-310
  23. WALKER DA 1976 Chloroplast and cell—the movement of key substances, etc. across the chloroplast envelope. *In* CR Stocking, U Heber, eds, *Encyclopaedia of Plant Physiology (New Series)*, Vol 3. Springer Verlag, Berlin, pp 1-49
  24. YAMAZAKI RK, NE TOLBERT 1969 Malate dehydrogenase in leaf peroxisomes. *Biochim Biophys Acta* 178: 11-20