

SCIENTIFIC REPORTS



OPEN

Physalins V-IX, 16,24-cyclo-13, 14-seco withanolides from *Physalis angulata* and their antiproliferative and anti-inflammatory activities

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Five new physalins, including a novel 1,10-seco one, physalin V (1), a tricarboxylic acid cycle one, physalin VIII (5), a rare 11,15-cyclo one, physalin IX (6), and two new ones, physalins VI (2) and VII (4) were isolated from stems and leaves of *Physalis angulata* together with eleven known analogues (3 and 7–16). Their structures were established by MS, IR, UV, and NMR spectroscopic analysis, together with the X-ray diffraction analysis of neophysalin, physalin P (12), and the structure of physalin D₁ (3) has been revised here. These isolated compounds were evaluated for their antiproliferative activities against human cancer cells (C4-2B, 22Rv1, 786-O, A-498, ACHN, and A375-S2) and inhibitory effects on nitric oxide production. Compounds 9 and 10 showed antiproliferative activities against all tested human cancer cells with IC₅₀ values of 0.24–3.17 μM. Compounds 1, 3, 4, 9, 10, 13, 14, and 16 exhibited inhibitory activities against NO production. The IC₅₀ values of compounds 9, 10, 13, and 16 were between 0.32 and 4.03 μM, while compounds 1, 3, 4, and 14 had IC₅₀ values of 12.83–34.19 μM. Herein, plausible biosynthetic pathways for rare structures 1 and 6 and structure–activity relationships on the inhibition of NO production for all isolated compounds are discussed.

The withanolides are a group of natural C₂₈ steroids with a γ- or δ-lactone based on an ergostane skeleton, which are derived from a parent 23-hydroxy-26-oic or 22-hydroxy-26-oic acid. They can be further divided into 22 subtypes based on the difference of the structural skeleton, such as normal withanolides, physalins, withaphysalins, neophysalins, jaborols, and so on^{1,2}. Physalins, commonly termed 16,24-cyclo-13,14-seco steroids, are classified as a group of withanolides with the most advanced oxidation level, from which they are formally derived from oxidative bond cleavage between C-13 and C-14 to produce a nine-membered ring, formation of a new six-membered carbocycle between C-16 and C-24, oxidation of the C-18 methyl group to a COOH group, which leads to 18,20-lactonization, and formation of an oxo bridge between C-14 and C-17, resulting in an oxygen heterocyclic system across rings C and D^{3–5}.

The genus *Physalis* has attracted attention from scientists due to the occurrence of the first 16,24-cyclo-13,14-seco withanolide physalin A from *Physalis alkekengi* var. *franchetii* in 1969⁶. Over the past 47 years, about 60 physalins have been isolated from this genus. The genus *Physalis* (Solanaceae) comprising approximately 120 species, is widely distributed in subtropical and tropical regions all over the world. *Physalis angulata* L., known as *ku-zhi* in China⁷, is a folk medicine that has been used to treat a variety of illnesses in many countries, such as dermatitis, trachitis, impaludism, rheumatism, and hepatitis. It is also used as a diuretic, antipyretic, antileukemic, anticancer, and immuno-modulatory agents⁸. Phytochemical investigations of *P. angulata* have led to the isolation of many physalins and normal withanolides, such as physalins A, B, D, E, F, G, I, and H, and physagulins A, B, C, and F, and some of them displayed remarkable anti-inflammatory^{9–12}, antitumor^{13–16}, antinociceptive¹⁷,

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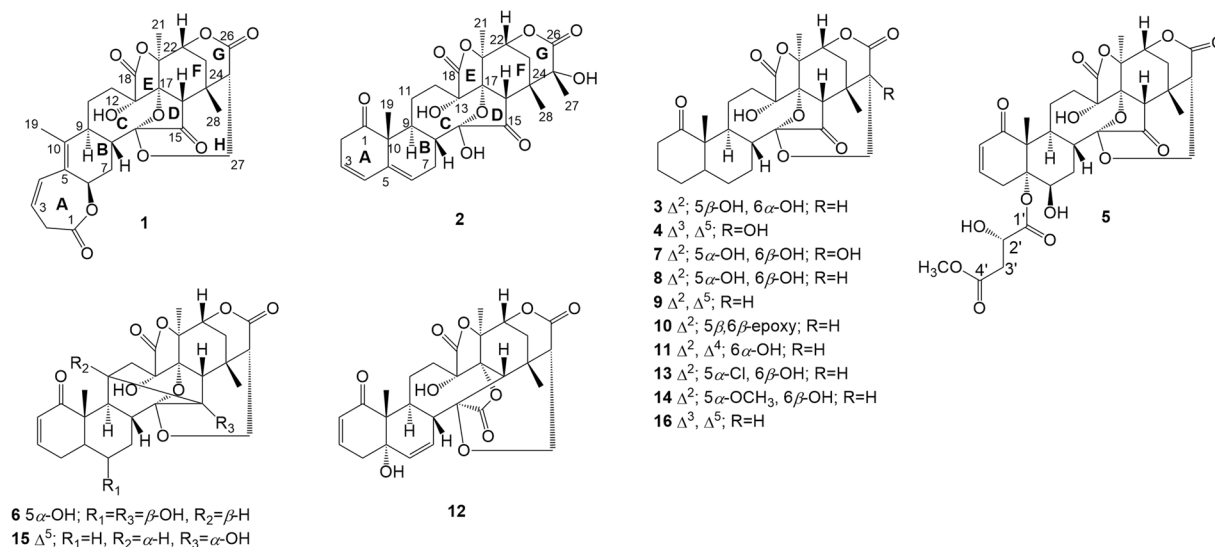


Figure 1. Chemical constituents of *P. angulata*.

and immunomodulatory¹⁸ activities. As part of our ongoing research on isolating bioactive physalins from the genus *Physalis* to provide potential anticancer and anti-inflammatory medicines^{11, 12, 15, 16}, the EtOH extracts of the dried stems and leaves of *P. angulata* were isolated to afford a novel 1,10-*seco* physalin, physalin V (**1**), a tricarboxylic acid cycle one, physalin VIII (**5**), a rare 11,15-*cyclo* one, physalin IX (**6**), and two new ones, physalins VI (**2**) and VII (**4**), together with eleven known analogues (Fig. 1), and the structure of the known physalin D₁ (**3**) was revised. In this paper, we describe the isolation and structural elucidation of these compounds together with their antiproliferative and anti-inflammatory evaluations *in vitro*. Furthermore, biosynthetic pathways for the rare physalins **1** and **6** are proposed, and structure–activity relationships for all isolated compounds are preliminarily discussed.

Results and Discussion

The EtOH extracts of stems and leaves of *P. angulata* were separated by silica gel, Sephadex LH-20, ODS open column chromatography, preparative TLC, and preparative HPLC to yield five new physalins, including an uncommon 1,10-*seco* one, physalin V (**1**), physalins VI (**2**), VII (**4**), and VIII (**5**), and a novel 11,15-*cyclo* one, physalin IX (**6**), together with eleven known analogues, 25 β -hydroxyphysalin D (**7**)¹⁹, physalins D₁ (**3**), D (**8**)²⁰, B (**9**)²¹, F (**10**)²², G (**11**)²³, P (**12**)²⁴, H (**13**)²⁵, I (**14**)²⁰, and R (**15**)²⁶, and isophysalin B (**16**)²⁵ (Fig. 1).

Structure elucidation. Physalin V (**1**) was isolated as an amorphous powder, and HRESIMS spectrum (Supplementary Fig. 8) showed that the quasi-molecular peak at m/z 549.1750 $[M + Na]^+$ (calcd for $C_{28}H_{30}O_{10}Na$, 549.1737), requiring the molecular formula $C_{28}H_{30}O_{10}$. The IR spectrum (Supplementary Fig. 9) of **1** showed the presence of hydroxy (3400 cm^{-1}), carbonyl (1728 cm^{-1}), and olefinic (1646 cm^{-1}) functionalities. The ¹H NMR spectrum (Supplementary Fig. 1) of **1** displayed signals for three methyl groups at δ_H 1.83 (3H, s, Me-19), 1.82 (3H, s, Me-21), and 1.17 (3H, s, Me-28), and two olefinic protons at δ_H 6.49 (1H, dd, $J = 11.2, 2.0\text{ Hz}$, H-4) and 5.49 (1H, td, $J = 9.3, 2.0\text{ Hz}$, H-3). The ¹³C NMR data (Supplementary Fig. 2) showed 28 carbon resonances, including one carbonyl carbon (δ_C 209.0), three hydroxycarbonyl carbons (δ_C 171.80, 171.79, and 167.2), four olefinic carbons (δ_C 141.0, 128.8, 125.7, and 118.3), one ketal carbon (δ_C 105.2), six oxygenated carbons (δ_C 80.5, 80.3, 78.8, 76.4, 71.2, and 61.2), and three methyl carbons (δ_C 24.3, 21.3, and 17.1). The ¹H and ¹³C NMR data of **1** were closely resembled those of physalin G (**11**) isolated from *P. angulata*²³, with the exception that signals of ring A [δ_H 6.49 (1H, dd, $J = 11.2, 2.0\text{ Hz}$, H-4), 5.49 (1H, ddd, $J = 11.2, 9.3, 2.0\text{ Hz}$, H-3), and 5.25 (1H, br s, H-6); δ_C 171.8 (C-1), 141.0 (C-10), 128.8 (C-4), 125.7 (C-5), 118.3 (C-3), 71.2 (C-6), and 35.6 (C-2)] were present in **1**, indicating that **1** was a 1,10-*seco* physalin^{27, 28}. This conclusion was confirmed by the HMBC correlations from H-2a to C-1/C-3/C-4, H-3 to C-1/C-2/C-5, H-4 to C-2/C-5/C-6/C-10, and H-6 to C-1/C-5/C-8/C-10 (Supplementary Fig. 3–5), suggesting the presence of the seven-membered β,γ -unsaturated lactonic ring (Fig. 2). The relative configuration of **1** was established through the NOESY correlation of H-6 with H-2a (Fig. 3, Supplementary Fig. 6 and 7) in combination with the coupling constant ($J_{2a,3} = 4.7\text{ Hz}$), requiring *a*-orientations of H-6 and H-2a^{27, 28}. According to X-ray diffraction data (Cu K α) of physalin P (**12**) (Fig. 4), an acid induced benzenic acid-type rearranged product of physalins²⁹, in combination with the biogenetic grounds and previous literatures^{30–34}, **1** was assigned as (6*R*,8*R*,9*S*,13*S*,14*R*,16*S*,17*R*,20*S*,22*R*,24*S*,25*S*)-1,10-*seco* physalin G.

Physalin VI (**2**) was obtained as an amorphous powder with the molecular formula $C_{28}H_{32}O_{10}$ based on HRESIMS m/z 527.1918 $[M - H]^-$ (calcd for $C_{28}H_{31}O_{10}$, 527.1917; Supplementary Fig. 17) and ¹³C NMR data (Supplementary Fig. 11). The IR spectrum (Supplementary Fig. 18) displayed absorption bands corresponding to hydroxy (3397 cm^{-1}), carbonyl (1716 cm^{-1}), and olefinic (1646 cm^{-1}) functionalities. The ¹H NMR data (Supplementary Fig. 10) of **2** observed three olefinic protons at δ_H 6.07 (1H, br d, $J = 11.2\text{ Hz}$, H-4), 5.70 (1H, br d, $J = 4.8\text{ Hz}$, H-6), and 5.67 (1H, m, H-3), four methyl groups at δ_H 1.75 (3H, s, Me-21), 1.43 (3H, s, Me-28), 1.36

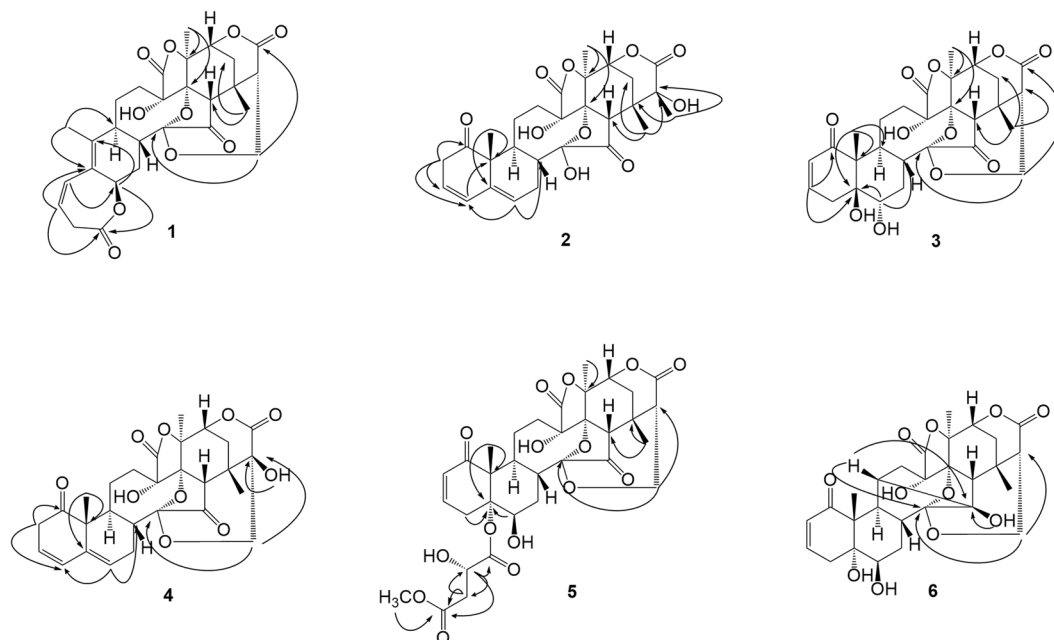


Figure 2. Selected HMBC correlations of compounds 1–6.

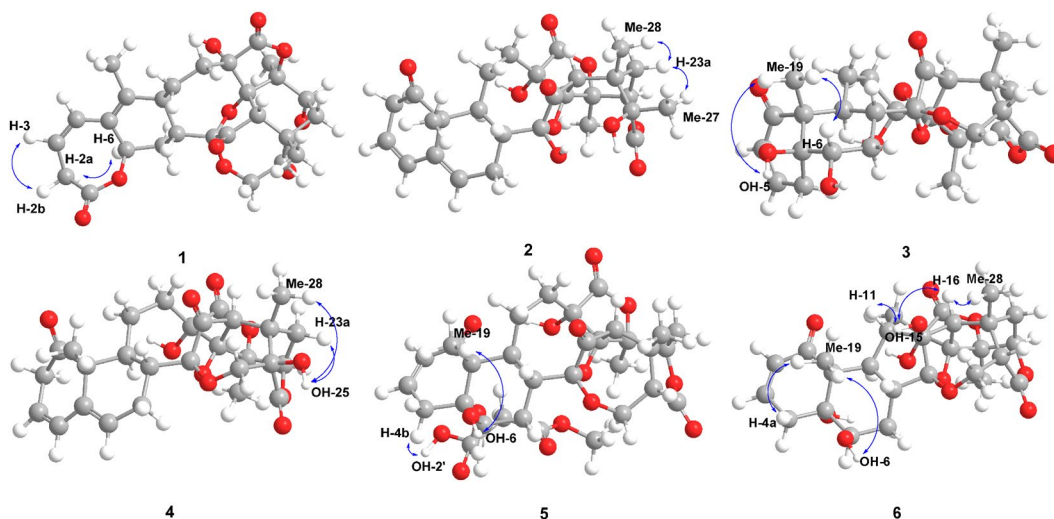


Figure 3. Selected NOESY correlations of compounds 1–6.

(3 H, s, Me-27), and 1.17 (3 H, s, Me-19), and the reduced two geminal oxymethylene protons in conjunction with the ^{13}C NMR data [δ_{C} 209.6 (C-1), 140.4 (C-5), 128.0 (C-4), 126.6 (C-6), 122.6 (C-3), 101.0 (C-14), 22.4 (C-28), 20.7 (C-21), 19.4 (C-27), and 17.8 (C-19)], indicating the existence of a 1-oxo-3,5-diene unit and the nonexistence of a C(14)-O-C(27) cyclization moiety. The above NMR spectroscopic data of 2 were related to those of physalin M isolated from *P. alkekengi* L. var. *franchetii*³⁵, except for the presence of a methyl group [δ_{H} 1.36 (3 H, s, Me-27); δ_{C} 19.4 (C-27)] and an oxygenated carbon [δ_{C} 72.6 (C-25)], with the assumption for the presence of a hydroxy group at C-25. The key HMBC correlations from Me-27 to C-24/C-25/C-26 and Me-28 to C-16/C-23/C-24/C-25 (Fig. 2 and Supplementary Fig. 12–14) confirmed the hypothesis. The orientation of OH-25 was deduced to be α by NOESY correlations of Me-27 with H-23a and Me-28 with H-23a (Fig. 3 and Supplementary Fig. 15 and 16). Therefore, 2 was determined as (8*R*,9*S*,10*R*,13*S*,14*R*,16*S*,17*R*,20*S*,22*R*,24*S*,25*R*)-25-hydroxyphysalin M.

Physalin D₁ (3) had the molecular formula $\text{C}_{28}\text{H}_{31}\text{O}_{11}$ identified by HRESIMS m/z 543.1870 [$\text{M} - \text{H}$]⁻ (calcd for $\text{C}_{28}\text{H}_{31}\text{O}_{11}$, 543.1866; Supplementary Fig. 26). Compound 3 exhibited the ^1H and ^{13}C NMR data (Supplementary Fig. 19–23) completely identical to those of physalin D₁ isolated from *P. alkekengi* L. var. *franchetii*³⁶, indicating that they were the same compound. However, the NOESY spectrum of 3 displayed a key correlation between OH-5 and Me-19 (Fig. 3 and Supplementary Fig. 24 and 25), indicating a β -orientation of OH-5. This conclusion was further confirmed by the ^{13}C NMR chemical shift value of the methyl group at C-10

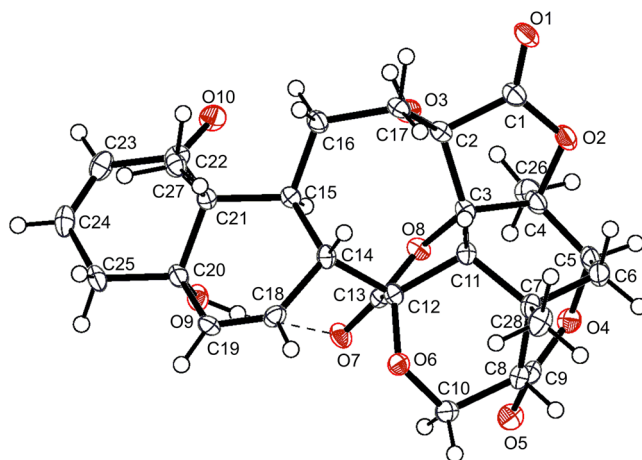


Figure 4. ORTEP drawing of compound **12**.

$[\delta_C$ 8.5 (C-19)], since this value in 5,6-dihydroxy or 4,5,6-trihydroxy withanolides could indicate the relationships between rings A and B to be *cis*- (around δ_C 10) or *trans*-fusion (around δ_C 15)³⁷. Hence, its structure was revised as (5*S*,6*S*,8*R*,9*S*,10*R*,13*S*,14*R*,16*S*,17*R*,20*S*,22*R*,24*S*,25*S*)-5,6-dihydroxyphysalin D.

The molecular formula of physalin VII (**4**) was determined as $C_{28}H_{30}O_{10}$ based on HRESIMS m/z 525.1769 $[M - H]^-$ (calcd for $C_{28}H_{29}O_{10}$, 525.1761; Supplementary Fig. 35) and ^{13}C NMR data (Supplementary Fig. 29). Comparison of the NMR data of **4** and isophysalin B (**16**)²⁵ indicated that the proton at C-25 was replaced by a hydroxy group, since the chemical shift value of C-25 was deshielded from δ_C 50.9 in **16** to δ_C 73.6 in **4**, which was further supported by the HMBC correlations from OH-25 to C-25, H-27a/H-27b to C-14/C-24/C-25/C-26, and Me-28 to C-16/C-23/C-24/C-25 (Fig. 2 and Supplementary Fig. 30–32). Furthermore, the NOESY correlations of OH-25 with H-23a/Me-28 (Fig. 3 and Supplementary Fig. 33 and 34) suggested that OH-25 has the same orientation as Me-28 with β . Thus, **4** was established as (8*R*,9*S*,10*R*,13*S*,14*R*,16*S*,17*R*,20*S*,22*R*,24*S*,25*R*)-25-hydroxyisophysalin B.

HRESIMS analysis of physalin VIII (**5**) established the molecular formula $C_{33}H_{38}O_{15}$ [m/z 673.2127 $[M - H]^-$ (calcd for $C_{33}H_{37}O_{15}$, 673.2132)] (Supplementary Fig. 45), indicating extra five carbons except for the C_{28} skeleton of the physalins. The detailed analysis of 1H and ^{13}C NMR data (Supplementary Fig. 37 and 38) for **5** and physalin D (**8**)²⁰ revealed that **5** possessed a similar structure to **8**, and the only difference was the presence of five carbon resonances (δ_C 170.9, 170.1, 67.0, 51.3, and 38.1) in **5**. The HMBC correlations from OMe-4' to C-4', H-2' to C-3'/C-4', and OH-2' to C-1' (Fig. 2 and Supplementary Fig. 39–43) indicated **5** had a methyl malate moiety (Supplementary Fig. 44). Its linkage was deduced at C-5 based on the downfield shift of C-5 from δ_C 76.5 in **8** to δ_C 90.3 in **5**. Moreover, its configuration was established as L-configuration by a polarimetric analyses for the hydrolyzed product of **5** with optical value of -2.0 . Thus, **5** was characterized as (2'*S*,8*R*,9*S*,10*R*,13*S*,14*R*,16*S*,17*R*,20*S*,22*R*,24*S*,25*S*)-5-L-methyl malatephysalin D.

Physalin IX (**6**) was isolated as an amorphous powder, and the molecular formula was established as $C_{28}H_{32}O_{11}$ according to HRESIMS m/z 567.1842 $[M + Na]^+$ (calcd for $C_{28}H_{32}O_{11}Na$, 567.1837; Supplementary Fig. 54) and ^{13}C NMR data (Supplementary Fig. 48). The ^{13}C NMR spectrum observed characteristic resonances at δ_C 112.6 (C-14), 85.7 (C-15), 82.0 (C-17), 75.8 (C-13), and 47.1 (C-11), indicating that **6** was an unusual 11,15-*cyclo* physalin in which C-15 was an oxy-carbon rather than a ketonic carbon²⁶. The long-range correlations from H-11 to C-9/C-10/C-13/C-14/C-15 and OH-15 to C-11/C-15/C-16 in the HMBC spectrum (Supplementary Fig. 49–51) confirmed the above deduction. A comparison of 1H and ^{13}C NMR data for **6** and physalin R (**15**)²⁶ indicated the absence of two carbons [δ_C 135.5 (C-5) and 123.4 (C-6) in **15**] and the presence of two oxygenated carbons [δ_C 77.2 (C-5) and 74.1 (C-6) in **6**]. The HMBC correlations from OH-5 to C-5, OH-6 to C-6, and H-6 to C-5/C-8/C-10 (Fig. 2) suggested that two hydroxy groups were linked at C-5 and C-6, respectively. The NOESY correlations of H-4a with Me-19, H-4b with OH-5, OH-6 with Me-19, OH-15 with H-11/H-16, Me-28 with H-16 (Fig. 3 and Supplementary Fig. 52 and 53) revealed an α -orientation of OH-5, and the β -orientations of OH-6, H-11, and OH-15. Accordingly, **6** was identified as (5*R*,6*R*,8*R*,9*S*,10*R*,11*S*,13*S*,14*R*,15*S*,16*R*,17*R*,20*S*,22*R*,24*S*,25*S*)-11-hydro-5,6,15-trihydroxyphysalin R.

Plausible biogenetic pathway. The discovery of compounds **1** and **6** in the genus *Physalis* is rather uncommon from the viewpoint of chemotaxonomy. Compounds **1** and **6** could have been produced from physalin B, one of the major constituents of *P. angulata* (Fig. 5). Epoxidation of physalin B followed by Michael addition reaction from C-3 could give intermediate i. Then nucleophile attack on intermediate i by C-1, followed by ring cleavage between C-1 and C-10, lactonization, and dehydration could yield **1**²⁷. The cyclization of physalin B between C-11 and C-15 could produce intermediate ii as well as physalin R except for the orientations of H-11 and OH-15, which could be further epoxidized and hydrated to give **6**.

Antiproliferative assay. The isolated compounds were examined for their antiproliferative activities against human prostate cancer cells (C4-2B and 22Rv1), human renal cancer cells (786-O, A-498, and ACHN), human

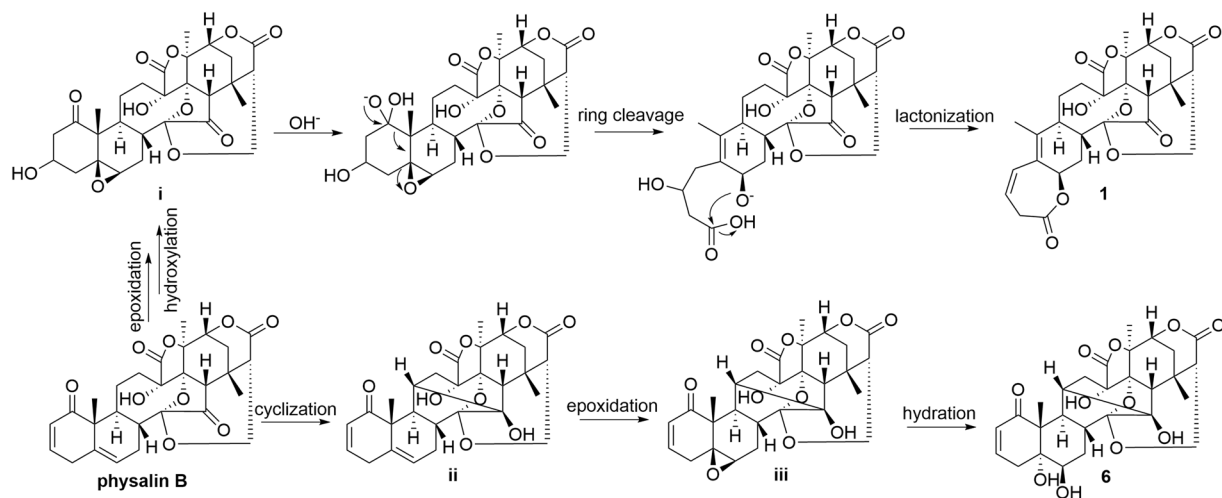


Figure 5. A plausible biogenetic pathway for compounds **1** and **6**.

melanoma cancer cells (A375-S2), and their inhibitory effects on NO production induced by LPS in macrophages. In the antiproliferative experiment (Table 3), 5-fluorouracil was used as a positive control drug. Compounds **9** and **10** displayed significant inhibitory effects against all tested cancer cells with IC_{50} values of 0.24 and 3.17 μM , respectively.

Inhibitory effects of all the compounds on NO production induced by LPS in macrophages. Nitric oxide (NO) is famous as a cellular signaling molecule, and considered as an important regulator in many physiological mechanisms^{38–40}. Pharmacological studies have indicated that inflammation is related to overproduction of NO⁴¹. The inhibitory effects of all isolated compounds on NO production induced by LPS in macrophages were assayed (Table 4). As shown in Table 4, compounds **9**, **10**, **13**, and **16** exhibited significant inhibitory activities against NO production with IC_{50} values of 0.32–4.03 μM , while compounds **1**, **3**, **4**, and **14** showed moderate inhibitory activities with IC_{50} values of 12.83–34.19 μM . A comparison of the inhibitory efficiency of physalins revealed that the $\Delta^{5,6}$ double bond and the proton at C-25 were of pivotal importance. Compound **16** displayed significant inhibitory effect, while **4** showed moderate inhibitory activity, indicating that H-25 could increase inhibitory activity. An analogous case was observed for compounds **8** and **9**, the former showed weak inhibitory effect since the 5-ene unit was hydroxylated. However, compound **3** displayed moderate inhibitory effect, indicating that the configurations of OH-5 and OH-6 could influence inhibitory activity. All isolated compounds were evaluated for their cytotoxic effects against RAW 264.7 macrophages, but did not exhibit any at their effective concentration.

Conclusion

In summary, a 1,10-*seco* physalin, physalin V (**1**), physalin VIII (**5**), a novel 11,15-*cyclo* physalin, physalin IX (**6**) and two other new ones (**2** and **4**) were isolated from the stems and leaves of *P. angulata* together with eleven known ones. The absolute configuration of physalin P (**12**) was established by single crystal X-ray crystallography, and it is the first report about the absolute configuration of the neophysalins. To our knowledge, the 1,10-*seco* normal withanolides were previously isolated from *P. minnina*, *P. peruviana* and *Flos Daturae*, while it is the first report about the presence of 1,10-*seco* physalin in nature and the genus *Physalis*. The inhibitory effects on nitric oxide production and antiproliferative activities against human cancer cells of the isolated compounds were evaluated. Compounds **9** and **10** showed significant antiproliferative activities against all tested human cancer cells. Compounds **9**, **10**, **13**, and **16** showed significant inhibitory activities against NO production. These results indicated that they are promising candidates that could be further researched on and developed as antitumor and anti-inflammatory agents.

Methods

General experimental procedures. Optical rotations were recorded on a Perkin-Elmer 241 polarimeter. UV spectra were measured on a Shimadzu UV 2201 spectrophotometer. IR spectra were recorded on a Bruker IFS 55 spectrometer. Bruker AV-400 and AV-600 spectrometers were used in the NMR experiments. Chemical shift values were expressed in δ (ppm) using the peak signals of the solvent $\text{DMSO}-d_6$ (δ_{H} 2.50 and δ_{C} 39.51) as references, and coupling constants (J in Hz) were given in parentheses. HRESIMS data were acquired on an Agilent 6210 TOF mass spectrometer. Silica gel GF₂₅₄ prepared for TLC was purchased from Qingdao Marine Chemical Factory (Qingdao, China). Silica gel (200–300 mesh, Qingdao Marine Chemical Factory, Qingdao, China), Sephadex LH-20 (Pharmacia, USA), and octadecyl silica gel (Merck Chemical Company Ltd., German) were used for column chromatography (CC). RP-HPLC separations were conducted using an LC-6AD liquid chromatograph and a SPD-20A UV detector (Shimadzu, Kyoto, Japan) with a RP-C₁₈ column (250 × 20 mm, 120 Å, 5 μm , YMC Co. Ltd.).

no.	1 ^a		2 ^b		3 ^b	
	δ_C	δ_H (J in Hz)	δ_C	δ_H (J in Hz)	δ_C	δ_H (J in Hz)
1	171.79		209.6		202.7	
2	35.6	4.09, br d (17.6)	39.6	3.42, d (20.0)	127.3	5.79, dd (10.0, 2.0)
		3.02, dd (17.6, 9.3)		2.61, dd (20.0, 4.3)		
3	118.3	5.49, ddd (11.2, 9.3, 2.0)	122.6	5.67, m	143.4	6.69, ddd (10.0, 5.1, 2.0)
4	128.8	6.49, dd (11.2, 2.0)	128.0	6.07, br d (11.2)	29.7	2.60, dt (20.2, 2.0)
						2.33, dd (20.2, 5.1)
5	125.7		140.4		77.4	
6	71.2	5.25, br s	126.6	5.70, br d (4.8)	70.4	3.58, m
7	26.1	2.29, br d (14.3)	26.1	2.44, dd (14.3, 4.8)	29.8	2.08, m
		1.36, td (14.3, 3.9)		2.07, m		1.34, m
8	40.0	2.09, td (9.4, 3.9)	40.5	2.06, m	41.2	2.01, td (12.4, 4.5)
9	38.4	3.38, t (9.4)	32.5	2.95, dd (10.4, 7.4)	33.4	2.98, br t (12.4)
10	141.0		55.3		55.7	
11	23.7	2.19, m	23.7	1.36, m	21.1	1.33, m
		1.43, m		1.08, m		0.80, m
12	24.1	2.19, m	28.8	2.15, m	24.3	1.76, m
		1.43, m		1.86, dd (15.5, 6.5)		1.27, m
13	80.5		79.3		78.3	
14	105.2		101.0		105.9	
15	209.0		215.2		208.9	
16	53.9	2.86, s	51.3	2.78, s	54.1	2.75, s
17	78.8		82.3		80.4	
18	171.80		172.3		171.6	
19	17.1	1.82, s	17.8	1.05, s	8.5	0.81, s
20	80.3		82.8		80.2	
21	21.4	1.83, s	20.7	1.75, s	21.4	1.78, s
22	76.4	4.61, t (2.6)	76.7	4.42, br d (5.3)	76.3	4.56, t (2.9)
23	31.2	2.13, dd (14.6, 2.6)	26.9	2.71, dd (14.3, 5.3)	31.2	2.09, m
		1.94, br d (14.6)		1.38, m		1.89, m
24	30.5		41.1		30.5	
25	49.2	2.96, d (4.4)	72.6		49.2	2.90, d (4.4)
26	167.2		169.8		167.2	
27	61.2	4.31, dd (13.4, 4.4)	19.4	1.36, s	60.8	4.25, dd (13.4, 4.4)
		3.66, br d (13.4)				3.57, dd (13.4, 4.4)
28	24.3	1.17, s	22.4	1.43, s	24.4	1.13, s
OH-5						4.40, s
OH-6						4.53, d (5.0)
OH-13		6.86, s		6.08, s		6.36, s
OH-14				6.43, s		
OH-25				5.93, s		

Table 1. ¹H and ¹³C NMR data of compounds 1–3. ^a¹H NMR spectra recorded at 600 MHz, ¹³C NMR spectra recorded at 150 MHz, DMSO-*d*₆. ^b¹H NMR spectra recorded at 400 MHz, ¹³C NMR spectra recorded at 100 MHz, DMSO-*d*₆.

Plant material. The stems and leaves of *P. angulata* were collected from Nanning, Guangxi Province, China, in July 2013, and identified by Jia-Fu Wei, Guangxi Institute for Food and Drug Control. A voucher specimen (PA-20130826) has been deposited in the herbarium of the Department of Natural Products Chemistry, Shenyang Pharmaceutical University.

Extraction and isolation. The dried stems and leaves of *P. angulata* (9.5 kg) were extracted with 75% EtOH (2 × 2 h × 110 L) and concentrated *in vacuo*. The resulting extracts (1.3 kg) were suspended in H₂O (5 L), and partitioned successively with petroleum ether (3 × 5 L), EtOAc (3 × 5 L), and *n*-BuOH (3 × 5 L). The EtOAc extracts (116 g) were subjected to silica gel CC (10 × 80 cm) eluted with CH₂Cl₂–MeOH (100:1, 80:1, 60:1, 40:1, 20:1, 10:1, 8:1, 5:1, 3:1, 1:1, and 0:1, v/v) to afford compound 8 (500 mg) and six fractions (E1–E6). E3 (35 g) was subjected to silica gel CC (6 × 80 cm) eluted with petroleum ether–acetone (10:1 to 0:1) to produce seven sub-fractions (E31–E37). E33 (4.0 g) was separated by ODS CC (3 × 50 cm) using a gradient of increasing MeOH in H₂O (1:9 to 1:0) to yield three subfractions (E331–E333). E331 (2 g) was chromatographed over silica gel

no.	4 ^a		5 ^b		6 ^a	
	δ_C	δ_H (J in Hz)	δ_C	δ_H (J in Hz)	δ_C	δ_H (J in Hz)
1	209.8		203.1		203.9	
2	39.6	3.42, d (20.0)	126.7	5.75, dd (10.1, 2.2)	127.3	5.70, dd (10.0, 2.0)
		2.62, dd (20.0, 4.7)				
3	122.6	5.66, m	142.6	6.68, ddd (10.1, 5.1, 2.2)	143.9	6.70, ddd (10.0, 5.3, 2.0)
4	128.0	6.06, dd (9.5, 2.1)	28.6	3.27, dd (20.6, 5.1)	34.8	3.03, dt (19.5, 2.0)
				3.11, dt (20.6, 2.2)		2.06, dd (19.5, 5.3)
5	140.4		90.3		77.2	
6	125.7	5.67, m	65.3	4.54, t (3.6)	74.1	3.52, br s
7	24.8	2.33, dd (14.5, 3.8)	26.8	1.89, dt (14.1, 3.6)	28.9	1.85, m
		2.03, m		1.39, td (14.1, 3.6)		1.52, dt (12.5, 2.1)
8	38.5	2.10, td (11.1, 4.3)	37.8	2.22, td (12.2, 3.6)	40.0	2.41, td (12.5, 2.1)
9	31.7	3.08, br t (11.1)	29.9	3.32, td (12.2, 5.4)	41.0	2.75, dd (12.7, 6.7)
10	54.6		53.5		54.1	
11	24.6	1.57, m	24.3	1.85, m	47.1	2.23, t (6.0)
		1.03, m		1.01, m		
12	25.1	2.34, m	25.8	2.16, m	31.3	2.55, br d (14.8)
		1.42, dd (15.8, 9.9)		1.46, dd (16.0, 9.7)		1.87, m
13	78.3		78.7		75.8	
14	106.2		106.5		112.6	
15	209.0		209.7		85.7	
16	54.2	2.98, s	54.0	2.82, s	49.8	1.79, s
17	80.4		80.8		82.0	
18	171.5		171.6		173.9	
19	18.4	1.21, s	13.2	1.18, s	13.9	1.17, s
20	79.5		80.5		81.9	
21	21.8	1.80, s	21.2	1.81, s	20.5	1.61, s
22	76.6	4.59, dd (3.5, 2.1)	76.3	4.56, t (2.9)	75.9	4.39, br t (2.6)
23	28.0	2.35, dt (14.2, 3.5)	31.3	2.10, dd (14.5, 2.9)	34.7	1.89, m
		1.76, dd (14.2, 2.1)		1.93, dd (14.5, 2.9)		1.77, m
24	35.5		30.5		31.4	
25	73.6		49.4	2.89, d (4.2)	50.2	2.69, d (3.8)
26	168.4		167.3		168.9	
27	64.6	3.95, br d (12.7)	60.6	4.25, dd (13.4, 4.2)	59.9	4.70, br d (11.8)
		3.37, br d (12.7)		3.59, br d (13.4)		4.02, dd (11.8, 3.8)
28	18.9	1.12, s	24.4	1.16, s	28.5	1.34, s
1'			171.1			
2'			67.0	4.13, ddd (7.8, 5.1, 3.6)		
3'			38.1	2.52, dd (16.0, 3.6)		
				2.36, dd (16.0, 7.8)		
4'			170.1			
OMe-4'			51.3	3.52, s		
OH-2'				5.52, d (5.3)		
OH-5						4.15, s
OH-6				5.41, d (4.9)		4.88, d (4.0)
OH-13		6.37, s		6.03, s		5.95, s
OH-15						5.50, s
OH-25		6.45, s				

Table 2. ¹H and ¹³C NMR data of compounds 4–6. ^a¹H NMR spectra recorded at 600 MHz, ¹³C NMR spectra recorded at 150 MHz, DMSO-*d*₆. ^b¹H NMR spectra recorded at 600 MHz, ¹³C NMR spectra recorded at 100 MHz, DMSO-*d*₆.

CC (2 × 50 cm, CHCl₃–MeOH, 80:1 to 1:1) and preparative TLC (CH₂Cl₂–acetone, 4:1), yielding compound 6 (12 mg). E332 (1.5 g) was separated by silica gel CC (2 × 50 cm, petroleum ether–acetone, 50:1 to 1:1), preparative TLC (CH₂Cl₂–acetone, 4:1), and preparative HPLC (60% MeOH–H₂O, 6 mL min⁻¹) to afford compounds 2 (10 mg, *t*_R = 24 min) and 14 (21 mg, *t*_R = 26 min). Compound 9 (140 mg) was recrystallized from E34 (4.2 g) using MeOH. E36 (3.8 g) was subjected to silica gel CC (5 × 70 cm), eluted with petroleum ether–acetone (80:1 to 1:1),

Compound	C4-2B	22Rv1	786-O	A-498	ACHN	A375-S2
	(mean \pm SD, μ M)	(mean \pm SD, μ M)	(mean \pm SD, μ M)	(mean \pm SD, μ M)	(mean \pm SD, μ M)	(mean \pm SD, μ M)
9	0.30 \pm 0.04	0.36 \pm 0.02	0.72 \pm 0.11	0.57 \pm 0.03	0.96 \pm 0.08	3.17 \pm 0.17
10	0.50 \pm 0.03	0.30 \pm 0.02	0.24 \pm 0.06	0.33 \pm 0.02	0.82 \pm 0.09	2.61 \pm 0.14
5-fluorouracil ^c	5.64 \pm 0.45	3.83 \pm 0.16	>10	8.83 \pm 0.88	2.73 \pm 0.79	1.91 \pm 0.54

Table 3. IC₅₀ values^a of tested compounds against human cancer cell lines. ^aResults were expressed as IC₅₀ values in μ M. ^bCompounds **1–8** and **11–16** were inactive for all cell lines used (IC₅₀ > 10 μ M). ^cPositive control.

compound	IC ₅₀ (mean \pm SD, μ M)	compound	IC ₅₀ (mean \pm SD, μ M)
1	19.72 \pm 1.16	10	0.32 \pm 0.14
2	>100	11	>100
3	12.83 \pm 0.93	12	78.87 \pm 5.01
4	34.19 \pm 2.26	13	3.01 \pm 0.20
5	69.81 \pm 4.44	14	25.91 \pm 1.53
6	>100	15	70.69 \pm 4.93
7	>100	16	4.03 \pm 0.26
8	54.06 \pm 3.89	hydrocortisone ^a	58.79 \pm 3.32
9	0.88 \pm 0.22		

Table 4. Inhibitory effects of compounds **1–16** on NO production induced by LPS in macrophages. ^aPositive control.

to afford five subfractions (E361–E365). E363 (1.5 g) was subjected to silica gel CC (3.5 \times 70 cm) using CH₂Cl₂–acetone (80:1 to 4:1) to produce compound **10** (200 mg). E364 (210 mg) was separated by silica gel CC (2 \times 50 cm, CH₂Cl₂–acetone, 80:1 to 4:1) and preparative TLC (CH₂Cl₂–acetone, 4:1) to yield compound **12** (28 mg). E37 (3.2 g) was purified by preparative TLC (CH₂Cl₂–EtOAc, 1:1) to give compounds **1** (19 mg) and **11** (25 mg). E4 (15 g) was subjected to silica gel CC (5 \times 70 cm), eluted with CHCl₃–acetone (80:1 to 1:1), to afford five subfractions (E41–E45). E43 (1.0 g) was chromatographed over ODS CC (3 \times 50 cm, MeOH–H₂O, 1:9 to 1:0) and preparative HPLC (60% MeOH–H₂O) to give compound **7** (10 mg, *t*_R = 24 min). E45 (4 g) was separated by an ODS column (3 \times 50 cm, MeOH–H₂O, 1:9 to 1:0) and preparative TLC (CH₂Cl₂–acetone, 2:1), yielding compound **3** (50 mg) and an impure subfraction, which was further purified by preparative HPLC (65% MeOH–H₂O) to afford compounds **5** (8 mg, *t*_R = 20 min) and **15** (10 mg, *t*_R = 23 min). E6 (7 g) was subjected to silica gel CC (5 \times 70 cm) eluted with CHCl₃–acetone (50:1 to 1:1) to afford four subfractions (E61–E64). E61 (900 mg) was separated by Sephadex LH-20 CC (3 \times 80 cm, MeOH) and preparative HPLC (60% MeOH–H₂O) to produce compound **13** (5 mg, *t*_R = 16 min). E64 (1.5 g) was subjected to ODS CC (3 \times 50 cm, MeOH–H₂O, 1:9 to 1:0) and preparative TLC (CH₂Cl₂–acetone, 10:1), yielding compounds **4** (8 mg) and **16** (28 mg).

Spectroscopic data of 1–6. Physalin V (**1**): amorphous powder; $[\alpha]_D^{25}$ –12.0 (*c* 0.05, MeOH); UV (MeOH) λ_{\max} (log ϵ) 234 (4.0) nm; IR (KBr) ν_{\max} 3400, 2921, 2850, 1782, 1765, 1728, 1646, 1385, 1143 cm⁻¹; ¹H (600 MHz, DMSO-*d*₆) and ¹³C NMR (150 MHz, DMSO-*d*₆) data, see Table 1; HRESIMS *m/z* 549.1750 [M + Na]⁺ (calcd for C₂₈H₃₀O₁₀Na, 549.1737).

Physalin VI (**2**): amorphous powder; $[\alpha]_D^{25}$ –64.0 (*c* 0.05, MeOH); UV (MeOH) λ_{\max} (log ϵ) 216 (3.8) nm; IR (KBr) ν_{\max} 3397, 2921, 2850, 1716, 1646, 1467, 1384, 1111 cm⁻¹; ¹H (400 MHz, DMSO-*d*₆) and ¹³C NMR (100 MHz, DMSO-*d*₆) data, see Table 1; HRESIMS *m/z* 527.1918 [M – H][–] (calcd for C₂₈H₃₁O₁₀, 527.1917).

Physalin D₁ (**3**): amorphous powder; $[\alpha]_D^{25}$ –43.6 (*c* 0.05, MeOH); UV (MeOH) λ_{\max} (log ϵ) 220 (3.9) nm; IR (KBr) ν_{\max} 3396, 2921, 2850, 1765, 1734, 1648, 1468, 1384, 1134 cm⁻¹; ¹H (400 MHz, DMSO-*d*₆) and ¹³C NMR (100 MHz, DMSO-*d*₆) data, see Table 1; HRESIMS *m/z* 543.1870 [M – H][–] (calcd for C₂₈H₃₁O₁₁, 543.1866).

Physalin VII (**4**): amorphous powder; $[\alpha]_D^{25}$ (*c* 0.055, MeOH) –138.2; UV (MeOH) λ_{\max} (log ϵ) 218 (4.0) nm; IR (KBr) ν_{\max} 3431, 2920, 2850, 1767, 1740, 1697, 1645, 1465, 1384, 1138 cm⁻¹; ¹H (600 MHz, DMSO-*d*₆) and ¹³C NMR (150 MHz, DMSO-*d*₆) data, see Table 2; HRESIMS *m/z* 525.1769 [M – H][–] (calcd for C₂₈H₂₉O₁₀, 525.1761).

Physalin VIII (**5**): amorphous powder; $[\alpha]_D^{25}$ –76.4 (*c* 0.055, MeOH); UV (MeOH) λ_{\max} (log ϵ) 216 (3.5) nm; IR (KBr) ν_{\max} 3442, 2921, 2850, 1791, 1753, 1729, 1687, 1647, 1441, 1383, 1260, 1167 cm⁻¹; ¹H (600 MHz, DMSO-*d*₆) and ¹³C NMR (100 MHz, DMSO-*d*₆) data, see Table 2; HRESIMS *m/z* 673.2127 [M – H][–] (calcd for C₃₃H₃₇O₁₅, 673.2132).

Physalin IX (**6**): amorphous powder; $[\alpha]_D^{25}$ –53.0 (*c* 0.055, MeOH); UV (MeOH) λ_{\max} (log ϵ) 218 (3.7) nm; ¹H (600 MHz, DMSO-*d*₆) and ¹³C NMR (150 MHz, DMSO-*d*₆) data, see Table 2; HRESIMS *m/z* 567.1837 [M + Na]⁺ (calcd for C₂₈H₃₂O₁₁Na, 567.1842).

X-ray crystal structure determination of compound 12. The data were collected on an Xcalibur, Eos, Gemini diffractometer using monochromatized Cu K α radiation. The structure was solved by direct methods using SHELXL. Crystallographic data have been hosted in the Cambridge Crystallographic Data Centre (CCDC)

number 1465139). Copies of the data can be obtained, free of charge, from the CCDC website (www.ccdc.cam.ac.uk). Crystal Data: $C_{29}H_{35.30334}O_{11.65167}$, $M = 570.30$, orthorhombic, size $0.16 \times 0.09 \times 0.05 \text{ mm}^3$, $a = 7.65075(19) \text{ \AA}$, $b = 17.3380(6) \text{ \AA}$, $c = 19.8734(5) \text{ \AA}$, $\alpha = \beta = \gamma = 90^\circ$, $V = 2636.18(13) \text{ \AA}^3$, $T = 103.2$, space group $P2_12_12_1$ (no. 19), $Z = 4$, $\mu(\text{Cu K}\alpha) = 0.937$, completeness $\theta_{\text{max}} = 100.0\%$, $F(000) = 1210$, 2θ range for data collection from 6.766 to 143.782° , 9459 reflections measured, 5071 unique ($R_{\text{int}} = 0.0293$) which were used in all calculations. The final $wR(F_2)$ was 0.0974 (all data). The Flack parameter was $0.04(11)$. The largest difference peak and hole were 0.293 and $-0.200 \text{ e \AA}^{-3}$.

Optical rotation analysis for hydrolyzed product of compound 5. Compound 5 (3 mg) and hydrochloric acid (2 M, 4 mL) were added into the flask (10 mL) with cover, and stirred at 90°C for 3 h. The reaction mixture was extracted thrice with CHCl_3 , then the water layer was freeze-dried *in vacuo* to afford the residue. The hydrolyzed product (0.3 mg), purified over a Sephadex LH-20 column from the residue using CHCl_3 -MeOH (1:1), was analyzed by a Perkin-Elmer 241 polarimeter (Perkin-Elmer, Waltham, MA, USA) and an API3200 mass spectrometer (AB SCIEX, Framingham, MA, USA).

Antiproliferative assay. Compounds were evaluated by the MTT method for antiproliferative activities against human prostate cancer cells (C4-2B and 22Rv1), human renal carcinoma cells (786-O, A-498, and ACHN), and human melanoma cells (A375-S2)⁴². All these cells were incubated in RPMI-1640 or EMEM medium with 10% fetal bovine serum at a humidified atmosphere (5% CO_2 , 37°C). Cells (1×10^4 cells/well) were added into the 96-well plates for 12 h before drug addition. The test compounds with various concentrations were added into the 96-well plates, then incubated for 48 h. 5-Fluorouracil was used as the positive control, and every assay was repeated three times. Cell viability was evaluated by the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) reduction assay.

NO production bioassay. All compounds were assayed for the inhibition of NO production according to the Griess method^{43,44}. 1×10^6 Cells/well of RAW 264.7 cells were added into the 96-well plates, and incubated at 37°C for 24 h by the stimulation of LPS ($1 \mu\text{g/mL}$) with or without test compounds. After the addition of Griess reagent [0.1% *N*-(1-naphthyl)-ethylenediamine ($50 \mu\text{L}$); 1% sulfanilamide in 5% H_3PO_4 ($50 \mu\text{L}$)], absorbance (540 nm) was recorded by using a microplate reader. The standard curve was used to calculate the NO concentrations and inhibitory rates.

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Acknowledgements

This work was financially supported by grants from the National Natural Science Foundation of China (NSFC) (Grant No. 81430095, 21472138, and 31270399), Fund of the Educational Department of Liaoning Province (Grant No. L2011177), Liaoning Baiqianwan Talents Program (Grant No. 2013921043), and 2015 Career Development Program for Young and Middle-aged Teachers of Shenyang Pharmaceutical University (ZQN2015015).

Author Contributions

F.Q. and L.X.C. initiated the project. F.Q. and L.X.C. designed and coordinated the project. C.P.S. and C.Y.Q. performed the extraction, isolation, and structural identification of compounds. F.Z. performed the anti-inflammatory assay. K.N. and L.X.C. performed the antiproliferative assay. All authors approved the final version of the manuscript.

Additional Information

Supplementary information accompanies this paper at doi:10.1038/s41598-017-03849-9

Competing Interests: The authors declare that they have no competing interests.

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