

Molecular, Local, and Network-Level Basis for the Enhanced Stiffness of Hydrogel Networks Formed from Coassembled Racemic Peptides: Predictions from Pauling and Corey

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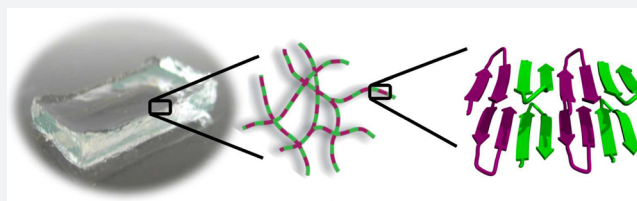
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Supporting Information

ABSTRACT: Hydrogels prepared from self-assembling peptides are promising materials for medical applications, and using both L- and D-peptide isomers in a gel's formulation provides an intuitive way to control the proteolytic degradation of an implanted material. In the course of developing gels for delivery applications, we discovered that a racemic mixture of the mirror-image β -hairpin peptides, named MAX1 and DMAX1, provides a fibrillar hydrogel that is four times more rigid than gels formed by either peptide alone—a puzzling observation. Herein, we use transmission electron microscopy, small angle neutron scattering, solid state NMR, diffusing wave, infrared, and fluorescence spectroscopies, and modeling to determine the molecular basis for the increased mechanical rigidity of the racemic gel. We find that enantiomeric peptides coassemble in an alternating fashion along the fibril long axis, forming an extended heterochiral pleat-like β -sheet, a structure predicted by Pauling and Corey in 1953. Hydrogen bonding between enantiomers within the sheet dictates the placement of hydrophobic valine side chains in the fibrils' dry interior in a manner that allows the formation of nested hydrophobic interactions between enantiomers, interactions not accessible within enantiomerically pure fibrils. Importantly, this unique molecular arrangement of valine side chains maximizes inter-residue contacts within the core of the fibrils resulting in their local stiffening, which in turn, gives rise to the significant increase in bulk mechanical rigidity observed for the racemic hydrogel.



INTRODUCTION

Self-assembling peptides can form a broad array of supramolecular structures including sheets,^{1–5} disks,⁶ spheres,⁷ barrels,⁸ tubes,^{9–12} and fibrils.^{13–22} Fibrils are privileged structures capable of higher order assembly leading to the formation of networks that constitute the formation of macroscopic hydrogels. The mechanical rigidities,^{23–25} enzymatic degradability,^{19,26,27} antibacterial activity,^{28–30} and cellular compatibility^{31–34} of peptide gels can be finely tuned through simple chemical and material-based strategies to enable targeted applications. For example, we recently reported the design of a UV-sensitive peptide gel that enables micro-anastomosis, the suturing of ultrasmall blood vessels.³⁵

Chirality is now being employed as a design modality in the fabrication of hydrogels.^{36–39} Heterochiral systems, containing both L- and D-isomers, offer unique material morphologies^{40–43} as well as a control over hydrogel degradability⁴⁴ as peptides composed of D-amino acids are resistant to proteolysis.^{45,46} As such, we initially sought to control the biodegradation of hydrogels formed by the peptide hydrogelator, MAX1, by

preparing mixed gels of the parent peptide with its enantiomer. In the course of analyzing the rheological properties of the mixed hydrogels, a large, nonadditive, synergistic enhancement in material rigidity was unexpectedly observed compared to gels formed from enantiomerically pure peptides with the greatest increase found in the racemic hydrogel.⁴⁷

MAX1 is a 20-amino acid peptide designed to undergo triggered folding and self-assembly into β -sheet rich fibrils that ultimately lead to the formation of a self-supporting hydrogel.^{24,48} The peptide's primary sequence contains two segments of alternating hydrophobic (valine) and hydrophilic (lysine) residues that flank a tetrapeptide motif designed to adopt a reverse turn in the folded state, Table 1. In neutral pH and low ionic strength buffers, the peptide adopts an ensemble of random coil conformations due to the electrostatic repulsion between protonated lysine side chains affording a solution of monomeric peptide. Increasing the ionic strength of the

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Table 1. Sequences of Peptides Utilized in These Studies^a

peptide	sequence
MAX1	VKVKVKVKV ^D P ^L PTKVKVKVKV-NH ₂
DMAX1	VKVKVKVKV ^L P ^D PTKVKVKVKV-NH ₂
MAX1-Azide	VKVKVKVKV ^D P ^L PTKVKVKVKV _(azide) -NH ₂
DMAX1-Biotin	Biotin-KVKVKVKVKV ^L P ^D PTKVKVKVKV-NH ₂
MAX1-EDANS	VKVKVE _(Edans) VKV ^D P ^L PTKVKVKVKV-NH ₂
DMAX1-DabcyI	VKVKVK _(DabcyI) VKV ^L P ^D PTKVKVKVKV-NH ₂
MAX1-IR1	VKVKVKVKV ^D P ^L PTKVKVKVKV-NH ₂
MAX1-IR2	VKVKVKVKV ^D P ^L PTKVKVKVKV-NH ₂
DMAX1-NMR1	VKVVKVKV ^L P ^D PTKVKVKVKV-NH ₂
MAX1-NMR1	<u>V</u> KVKVKVKV ^D P ^L PTKVKVKVKV-NH ₂
MAX1-NMR2	VKVKVKVKV ^D P ^L PT TKVKVKVKV-NH ₂
MAX35	IKVKIKVKV ^D P ^L PTKIKVKIKV-NH ₂
DMAX35	IKVKIKVKV ^L P ^D PTKIKVKIKV-NH ₂
MARK4	VRVKVRVKV ^D P ^L PTRVKVRVKV-NH ₂
DMARK4	VRVKVRVKV ^L P ^D PTRVKVRVKV-NH ₂

^aD-Amino acid residues are italicized. Uniformly ¹³C and ¹⁵N isotopically labeled residues are bold and underlined. ¹³C=¹⁸O isotopically labeled amino acid residues are in bold.

solution by the addition of NaCl screens the lysine borne charges and allows the peptide to fold and self-assemble into fibrils.²⁴ Increasing the temperature also favors folding and assembly by driving the hydrophobic effect and the burial of valine side chains into the dry interior of the fibrils.²¹ Solid state NMR shows that MAX1, in its folded and assembled state, adopts an amphiphilic β -hairpin structure that contains a type II' β -turn as designed, and that the peptide assembles with high fidelity into monomorphic fibrils composed of a bilayered cross- β structure. Folded hairpins assemble in-register with a *Syn* arrangement of their β -turns aligned along a given fibril monolayer. However, across the bilayer, hairpins adopt an *Anti* orientation.⁴⁹ The resulting network formed by these fibrils constitutes a mechanically rigid hydrogel.^{50,51}

As stated earlier, when MAX1 is mixed with its enantiomer, DMAX1, a gel is formed that is significantly more rigid. When we initially reported this behavior the physical basis for this enhancement in material rigidity was unknown.⁴⁷ According to theory for semiflexible biopolymer networks, hydrogel mechanical rigidity is dependent on both the number of cross-links made between fibrils in the network and the stiffness of the fibrils themselves.^{52–54} In the context of MAX1 and DMAX1, we envisioned that mixtures of these enantiomers could either self-sort to form an interpenetrating network of enantiomerically pure fibrils or coassemble into mixed fibrils, Figure 1. It had been unclear which scenario was more likely to occur and how either outcome would lead to an increase in network cross-link density, fibril stiffness, or both. Further, the molecular basis guiding the assembly process for these mirror-image peptides was not known. Herein, we use an arsenal of microscopy, scattering, spectroscopic techniques, rational peptide design, and modeling to uncover the physical basis for this material enhancement at the molecular, local-fibril, and hydrogel network length scales.

RESULTS AND DISCUSSION

Rheological, Microscopic, and Spectroscopic Evidence of Enantiomer Co-Assembly into Mixed Fibrils. Frequency-sweep rheology experiments measuring the storage modulus (G' , a measure of a gel's mechanical rigidity) show that an equimolar mixture of MAX1 and DMAX1 self-

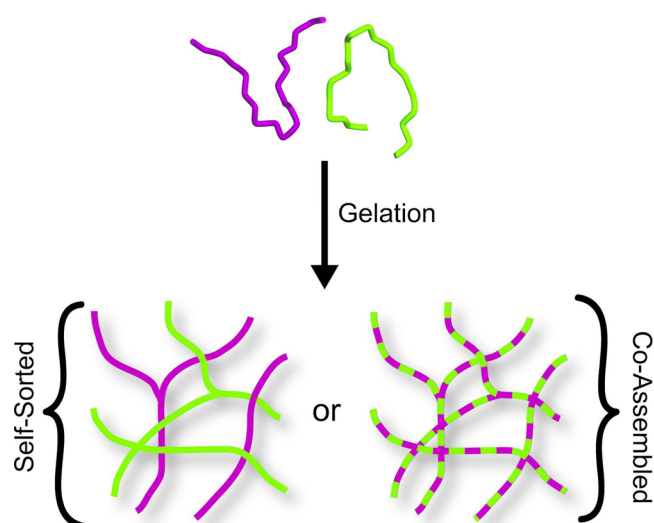


Figure 1. Possible modes of self-assembly for enantiomeric peptides MAX1 (green) and DMAX1 (purple) include self-sorting (solid-colored fibrils) or coassembly (dual-colored fibrils) into mixed fibrils.

assembles to form a hydrogel that is four times more stiff than gels formed by either pure enantiomer at identical peptide concentrations, Figure 2A. We previously reported that a racemic mixture of these peptides undergoes triggered folding to adopt β -hairpins that self-assemble into β -sheet rich fibrils where each fibril is comprised of a bilayer of hairpins.⁴⁷ Thus, the local fibril morphology in the racemic network was identical to that of both pure MAX1 and DMAX1 fibrils, with measured fibril heights of ~ 2.5 nm and widths of ~ 3 nm. Although the local morphology of the fibrils comprising the racemic network was known, the composition and arrangement of enantiomers within those fibrils was a mystery. Interestingly, both assemblies (self-sorted or coassembled) illustrated in Figure 1 would lead to hydrogel networks composed of fibrils having identical local dimensions.

Rheology was used to probe the nature of the racemic assembly process. We had previously determined that the rate of peptide folding, self-assembly, and thus fibril formation leading to gelation could be controlled by temperature.⁵⁵ The higher the temperature, the faster the rate of gelation. Figure 2B shows experiments where temperature and self-assembly conditions were controlled to mimic the formation of a self-sorted interpenetrating network of fibrils to assess if this type of network would lead to an enhancement in material rigidity. Here, individual peptide solutions in separate containers were triggered to assemble at 25 °C. At this temperature, each peptide can slowly fold and assemble to form local regions of soluble fibrils in solution but not a fully percolated network. The two fibrillar solutions were then mixed in equal volumes, and the temperature was raised to 37 °C where the evolution of an interpenetrating network could be formed. The time-sweep measurement shown in panel B demonstrates that the gel resulting from this network realizes a storage modulus that is not significantly greater than gels prepared from pure enantiomers alone ($G' \approx 300$ Pa). A control experiment was performed in which an equimolar mixture of the enantiomers was initially prepared in the same container at 5 °C, where assembly is prohibited. The temperature of the mixed solution was then increased to 37 °C to trigger folding and self-assembly. Under these conditions, coassembly is possible. Data

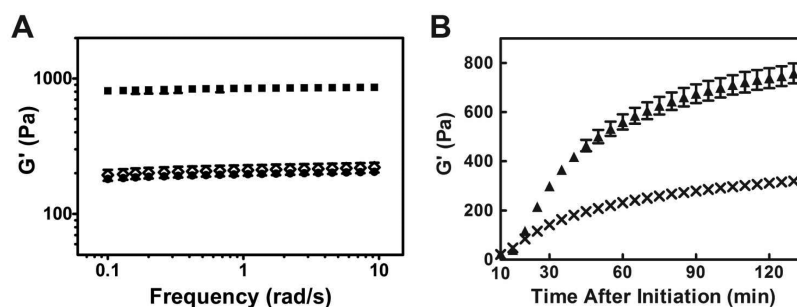


Figure 2. Dynamic frequency-sweep oscillatory rheology measuring the storage moduli of 1 wt % hydrogels as a function of frequency. A 4-fold increase in storage modulus is achieved in the racemic hydrogel (■) with respect to either pure MAX1 (◇) or DMAX1 (●) gels ($N = 3$) (A). Time-sweeps of two racemic gels. The first prepared by allowing independent solutions of MAX1 and DMAX1 to self-assemble independently at 25 °C for 10 min, then mixed and further gelled at 37 °C (×). The second gel prepared by premixing solutions of monomeric unfolded MAX1 and DMAX1 and then triggering gelation at 37 °C (▲) ($N = 3$) (B).

in panel B show that the gel resulting from these conditions displays the 4-fold increase in material rigidity ($G' \approx 750$ Pa). These data suggest that the enantiomers do not self-sort, but rather coassemble to form a network of mixed fibrils. Figure S1 shows CD data defining the temperature-dependent folding and assembly process.

We further investigated possible enantiomer coassembly by TEM. Here, two enantiomeric derivatives, namely, MAX1-Azide and DMAX1-Biotin, were designed to bind 10 and 5 nm gold nanoparticles (GNP) respectively while in the fibrillar state. Their sequences are shown in Table 1 along with the other peptides employed in this study. Peptides were prepared by standard Fmoc solid phase peptide synthesis and purified to homogeneity (Figures S2–S16). MAX1-Azide contains an azide moiety that can react with a dibenzylcyclooctyne (DBCO)-conjugated 10 nm GNP. DMAX1-Biotin is designed to bind to streptavidin-conjugated 5 nm GNPs. In the experiment depicted in Figure 3A, a fibrillar hydrogel was prepared from an equimolar mixture of these two peptides, diluted and deposited onto a TEM grid, and incubated with

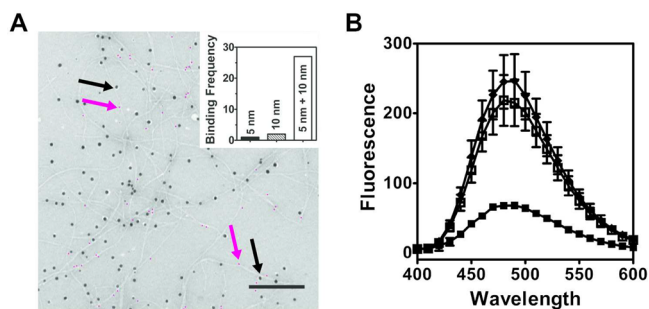


Figure 3. Coassembly of enantiomers MAX1 and DMAX1 is accessed by nanoparticle labeling of peptide fibrils using TEM (A) and fluorescence quenching (B). (A) DMAX1-Biotin and MAX1-Azide peptides were visualized within individual fibrils by the binding of 5 or 10 nm gold nanoparticles, respectively (scale bar = 200 nm). Five nanometer particles are false colored magenta for clarity. Magenta and black arrows indicate examples of bound 5 and 10 nm GNPs, respectively. The original noncolored micrograph is provided as Figure S18E. Inset: frequency of particles binding to distinct individual fibrils ($N = 30$). (B) Fluorescence spectra of MAX1-EDANS fibrils (◆), MAX1-EDANS fibrils and DMAX1-Dabcyl fibrils formed separately, then mixed (□) and a premixed solution of 1:1 MAX1-EDANS:DMAX1-Dabcyl that was allowed to assemble together (■) ($N = 3$).

sequential additions of each gold nanoparticle. Should these peptides coassemble to form mixed fibrils, each individual fibril should be adorned by both 5 and 10 nm GNPs. Conversely, if the peptides self-sort into independent fibrils, distinct populations of fibrils decorated with only 5 nm or only 10 nm GNPs will exist. The micrograph shows fibrils that are decorated with both 5 and 10 nm nanoparticles. The inset shows a statistical evaluation of binding where it is clear that the frequency of fibrils observed bound by both nanoparticles is much greater than fibrils bound by a single-sized GNP, supporting a coassembly mechanism. Although the two enantiomeric derivatives are not strictly non-superimposable mirror images, the gel formed by an equimolar mixture of these derivatives also displays the 4-fold increase in mechanical rigidity (Figure S17). Additionally, TEM control experiments were performed on fibrils composed of either pure MAX1-Azide or, separately, pure DMAX1-Biotin which showed that these homogeneous fibrils preferentially bound 10 or 5 nm GNPs respectively (Figure S18).

The coassembly of the peptide enantiomers within racemic fibrils was further confirmed by fluorescence spectroscopy, Figure 3B. Here, derivatives of MAX1 and DMAX1, namely, MAX1-EDANS and DMAX1-Dabcyl respectively (Table 1), were prepared to obtain spectroscopic evidence of the peptide coassembly via fluorescence quenching. EDANS and Dabcyl are often exploited as a fluorophore/quencher pair in biological assays^{56–58} and have a distance-limited Förster's radius of ~ 3 nm. If these peptide derivatives coassemble, solutions containing the mixed fibrils should display diminished fluorescence compared to fibrils formed by pure MAX1-EDANS. As in the peptide pair used in the TEM experiments discussed above, the hydrogel produced from a 1:1 mixture of MAX1-EDANS and DMAX1-Dabcyl retains the 4-fold increase in storage modulus with respect to either MAX1-EDANS or DMAX1-Dabcyl hydrogels (Figure S19).

Solutions of pure MAX1-EDANS fibrils are highly fluorescent as shown in Figure 3B with an emission maximum centered at 494 nm. When a gel is formed from an equimolar mixture of MAX1-EDANS and DMAX1-Dabcyl and diluted, the fluorescence of the resulting solution is highly attenuated, again suggesting that the peptides have coassembled into a given fibril where the enantiomeric pairs are in close proximity to each other. If the peptides were to self-sort, one would expect a solution of these fibrils to be highly fluorescent because the distance between the fluorophore and quenchers is greater than 3 nm. A control experiment was performed where

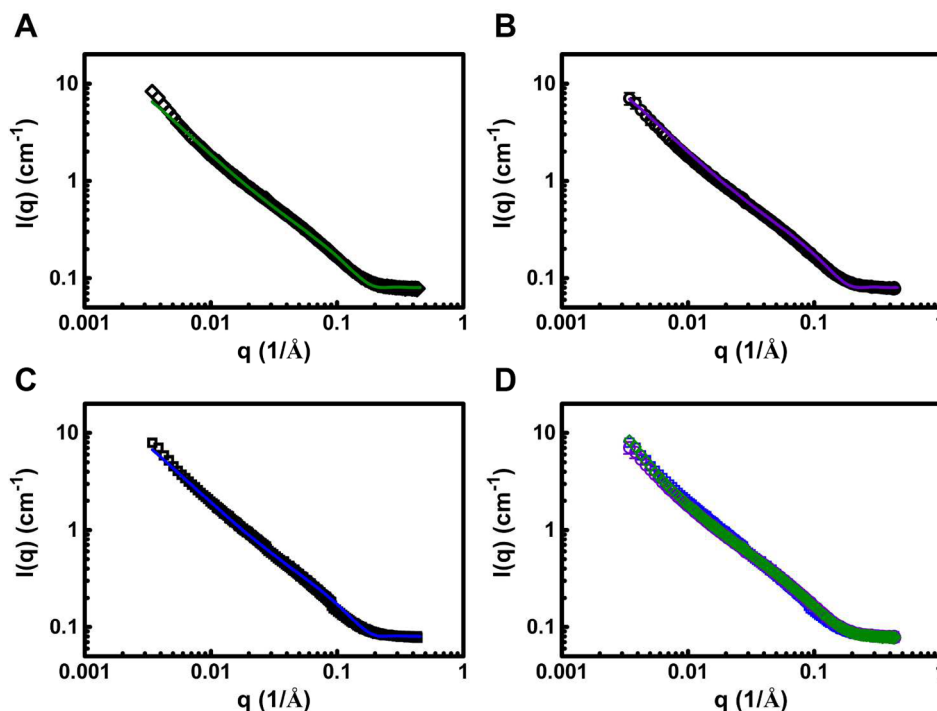


Figure 4. SANS scattering for 1 wt % hydrogels of pure MAX1, (\diamond) (A); pure DMAX1, (\circ) (B); and a mixture of 1:1 MAX1/DMAX1, (\square) (C). Each scattering profile is an average of three independent experiments on separate gels performed on separate days. Error bars are smaller than the symbol where not evident. Colored lines represent fractal model fits for the individual hydrogels. (D) Scattering profiles of MAX1 (green), DMAX1 (purple), and racemic (blue) hydrogels are coplotted to emphasize that the enantiomeric assemblies have similar mesh sizes.

separate gels of MAX1-EDANS and DMAX1-DabcyI were prepared and mixed upon dilution mimicking the formation of a self-sorted fibrillar network. Figure 3B shows that the resulting solution is nearly as fluorescent as MAX1-EDANS fibrils alone. Taken together, the rheology, TEM, and fluorescence spectroscopy data show that the parent peptides MAX1 and DMAX1 coassemble into mixed fibrils where enantiomers are in close proximity within the solid state.

Why Do Co-Assembled Fibrils Afford a Stiffer Hydrogel Network? Theory for semiflexible polymers^{52–54} predicts that the rigidity of a densely cross-linked gel is dependent on both the mesh size, ξ , and fibril bending modulus, κ , as in eq 1.

$$G' \approx \frac{\kappa^2}{kT\xi^5} \quad (1)$$

Thus, it is likely that changes in the mesh size and/or fibril bending modulus result in the enhancement in rigidity that is observed in the racemic hydrogel. Mesh size is directly related to the cross-link density: more cross-links lead to a smaller mesh size and a more rigid gel. With respect to the fibril bending modulus, if the fibrils comprising the network are themselves more stiff, the resulting gel will be more rigid.

Small angle neutron scattering (SANS) interrogates the hydrogel network structure at length scales relevant to the mesh size of a 1 wt % hydrogel, which was estimated to be in the tens of nanometers.⁵⁰ This corresponds to a regime centered at $q \approx 0.01 \text{ \AA}^{-1}$ in the SANS experiment. If the racemic gel contains more cross-links and a smaller mesh, then qualitatively, one should expect the scattering intensity to be greater than the pure enantiomeric gels in this regime. Figure 4 shows SANS data for MAX1, DMAX1, and racemic hydrogels; panel D clearly shows that all the scattering profiles are superimposable, suggesting that the mesh sizes for these three gels are very

similar. Further, each scattering profile (Figure 4A–C) was fit to a fractal model, where the scattering intensity $I(q) = P(q)S(q)$, to determine the mesh size of the hydrogels.⁵¹ Here, the form factor, $P(q)$, is calculated for spherical particles

$$P(q) = \varphi V_p (\rho_{\text{sph}} - \rho_{\text{solv}})^2 F(qR_0)^2 \quad (2)$$

where R_0 is the radius of the spherical particles with volume, $V_p = 4/3\pi R_0^3$ and

$$F(x) = \frac{3[\sin x - x \cos x]}{x^3} \quad (3)$$

In this model, the peptide fibrils within the hydrogel network are approximated as an assembly of spherical building blocks that further associate to form fractal-like clusters with a defined average size, namely, the correlation length, λ , and self-similarity dimension, D_f . These clusters are characterized by the structure factor

$$S(q) = 1 + \frac{\sin[(D_f - 1) \tan^{-1}(q\lambda)]}{(qR_0)^{D_f}} \frac{D_f \Gamma(D_f - 1)}{[1 + 1/q^2 \lambda^2]^{(D_f - 1)/2}} \quad (4)$$

Fitting these data for MAX1 and DMAX1 hydrogels results in a fractal dimension of 1.33 and correlation length of 125 nm for both gels. The racemic gel has a very similar fractal dimension and correlation length of 1.36 and 118 nm. Thus, the SANS data show that the observed increase in mechanical rigidity enjoyed by the racemic gel is not a result of a more cross-linked network.

On the basis of eq 1 above, the alternative to a decrease in mesh size to bring about an increase in gel rigidity is an increase

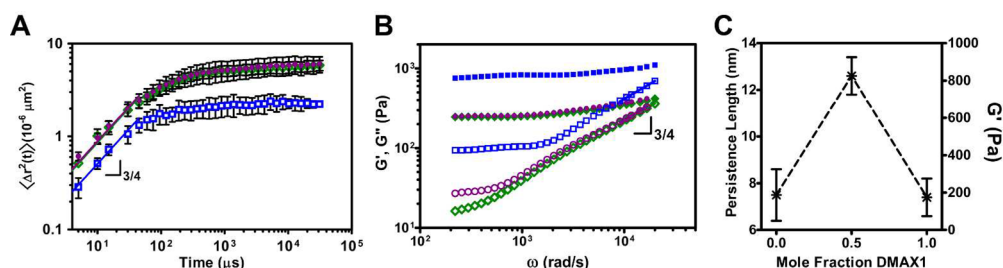


Figure 5. Diffusing wave spectroscopy monitoring the mean squared displacement, $\langle \Delta r^2(t) \rangle$, of particles embedded in 1 wt % MAX1 (\diamond), DMAX1 (\bullet), and 1:1 MAX1/DMAX1 (\square) hydrogels ($N = 3$) (A) (B) Corresponding storage, G' (closed symbols) and loss, G'' (open symbols) moduli of 1 wt % MAX1 (green), DMAX1 (purple) and 1:1 MAX1/DMAX1 hydrogels (blue) obtained from DWS measurements. Nonlinear fit with logarithmic slope of $3/4$ of early time regime in (A) and high frequency regime in (B) is indicated by colored lines. Loss moduli, G'' , at high frequencies are fit to calculate persistence length. In panel C, the persistence length and G' are coplotted against the mole fraction of DMAX1 used in the gel formulation ($N = 3$). Line is included to guide the eye.

Table 2. Fibril Parameters Derived from Diffusing Wave Spectroscopy

gel	τ_c (μs) ^a	δ^2 ($\times 10^{-6} \mu\text{m}^2$) ^a	d (nm) ^b	L (nm) ^c	ρ ($\times 10^4 \mu\text{m}^{-2}$) ^d	l_p (nm) ^a	κ (pN \cdot nm ²) ^a
MAX1	118 (5)	5.45 (0.70)	3	125	2.36	7.5 (1.1)	30.8 (4.5)
DMAX1	119 (18)	5.66 (1.27)	3	125	2.36	7.4 (0.8)	30.4 (3.3)
racemate	62 (14)	2.21 (0.37)	3	118	2.23	12.6 (0.8)	51.8 (3.3)

^aMeasured from DWS; errors in parentheses. ^bMeasured from TEM. ^cDetermined by SANS. ^dCalculated from polymer concentration.

in the bending modulus of individual fibrils comprising the racemic gel. However, the small widths (~ 3 nm) of the racemic fibrils make it incredibly difficult to measure their stiffness by established methods such as atomic force microscopy (AFM).^{59,60} Thus, we turned our attention to diffusing wave spectroscopy (DWS).^{61–63} DWS offers a noninvasive method to access the high frequency linear rheology of biomaterials, beyond the capabilities of conventional mechanical rheometers and particle tracking microrheology methods, enabling precise characterization of their mechanical properties.⁶⁴ This technique has previously been applied to colloidal gels,^{65,66} emulsions,⁶⁵ worm-like micelles,^{67–69} and F-actin solutions.^{70,71} Here, we use DWS for the first time to measure the stiffness of peptide fibrils, namely, those formed by pure MAX1, DMAX1, and their corresponding racemate.

In DWS experiments, the fibril stiffness is extracted by monitoring the mean-squared displacement, MSD $\langle \Delta r^2(t) \rangle$ of polystyrene microspheres embedded within the hydrogels at high frequencies, Figure 5A. The MSD was fit to a stretched exponential of the form $\langle \Delta r^2(\tau) \rangle = \delta^2(1 - e^{-(\tau/\tau_c)^a})$.⁶⁶ Here, a is the short-time logarithmic slope, whose scaling (0.75) is characteristic of semiflexible polymer networks. τ_c is the characteristic relaxation time of the particle in the gel, and δ^2 (defined by the plateau region) corresponds to the maximal particle displacement in the network (Table 2). The relaxation time for particles in the racemic gel is faster (62 μs) than that in either of the enantiomeric gels ($\sim 118 \mu\text{s}$), and the maximal displacement is correspondingly smaller for the racemate. This indicates that particle movement is significantly more restricted in the racemic gel. In panel B, MSD data are converted to the storage (G') and loss (G'' , a measure of the material's viscous response to stress) moduli numerically through calculation of the complex modulus, $G^*(\omega)$, using the approximate form of the Generalized Stokes–Einstein Relation given in eqs S5–S7 in the Materials and Methods section. The 4-fold enhancement in material rigidity for the racemic hydrogel is observed in the low frequency plateau regime, which is in agreement with the bulk rheology measurements.

Importantly, at lag times $< 100 \mu\text{s}$, corresponding to the high frequency regime in panel B, the probe particles are displaced on the order of 1 nm and their response is dictated by the single filament mechanics of the peptide network. This regime is used to determine the persistence length and bending modulus of the fibrils that comprise the hydrogel networks. Here, the theory for high frequency rheology of semiflexible polymers is invoked, and data are fit to eq 5 below:^{52,72,73}

$$G''(\omega) - \omega\eta_s = \frac{1}{15}\rho\kappa l_p(-2i\zeta/\kappa)^{3/4}\omega^{3/4} \quad (5)$$

where ρ is the fibril length per unit volume, $l_p = \kappa/kT$ is the fibril persistence length, and $\zeta \approx 4\pi\eta/\ln(0.6L/d)$ is the lateral drag coefficient per unit length for filaments of length L and diameter d . Previous work (see Table 2) has determined all relevant parameters in eq 5 except the persistence length for MAX1, DMAX1, and racemic assemblies.^{47,50,51} This allows the loss modulus to be fit in the high frequency regime with the persistence length as the only adjustable parameter. All peptide samples exhibit the characteristic $\omega^{3/4}$ scaling as predicted by semiflexible polymer theory. Resulting values for fibril persistence length are shown in Table 2.

As expected, average persistence length (l_p) values for fibrils formed from MAX1 and DMAX1 are nearly identical, 7.5 ± 1.1 nm and 7.4 ± 0.8 nm, respectively. However, the average persistence length of fibrils formed within the racemic network is nearly twice as long (12.6 ± 0.8 nm). The corresponding bending moduli for fibrils within each hydrogel can be calculated from $\kappa = l_p kT$ (Table 2), and it is clear that the fibrils in the racemic gel are significantly more stiff (51 vs 31 pN nm²). Further, the Job plot in Figure 5C nicely shows that the persistence length of the fibrils and the corresponding rigidity of their respective gels measured by bulk rheology track with the mole fraction of enantiomer used in the gel formulation. Taken together, the SANS and DWS show that the racemic hydrogel is mechanically more rigid because the fibrils that comprise the gel are, themselves, more rigid.

Molecular Arrangement of Enantiomers within Racemic Fibrils. It is likely that unique molecular-level interactions,

made between MAX1 and DMAX1 in the assembled state, are responsible for the enhanced stiffness of the racemic fibrils. Three arrangements can be envisioned for the coassembly of MAX1 and DMAX1, Figure 6. In the first arrangement (panel

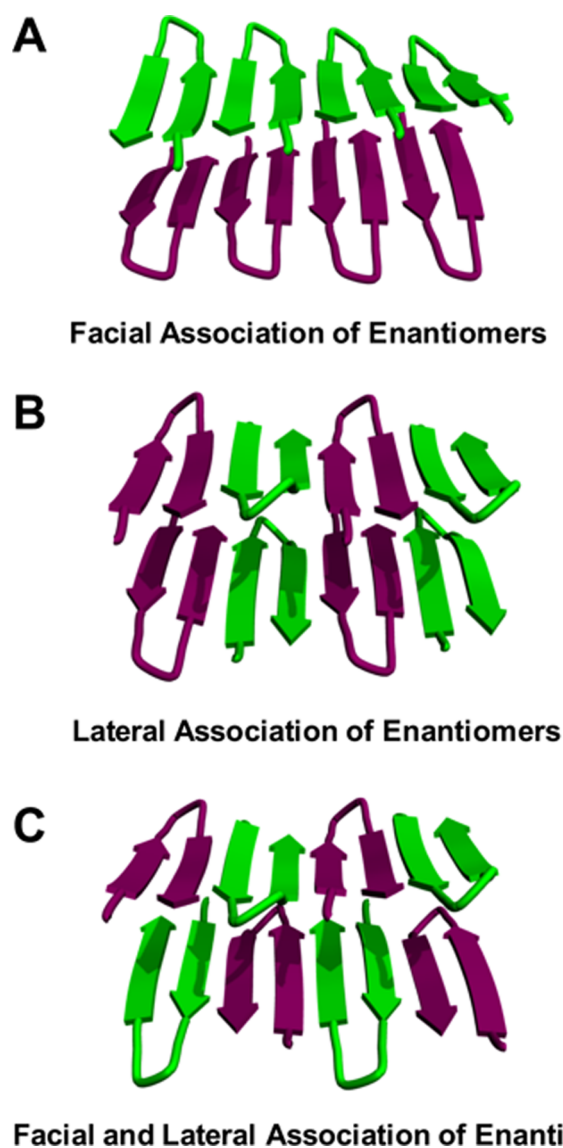


Figure 6. Possible arrangements of enantiomeric coassembly within racemic fibrils. (A) MAX1 and DMAX1 self-segregate into distinct monolayers. (B) Each enantiomer pairs with itself across the bilayer, but alternates down the long axis of the fibril. (C) MAX1 and DMAX1 pair across the bilayer and laterally within each monolayer.

A), each enantiomer self-segregates into its respective monolayer. Here, MAX1 and DMAX1 make facial contact across the bilayer. However, within each monolayer, each enantiomer only makes lateral contact with itself as it traverses down the long axis of the fibril. In panel B, each enantiomer pairs with itself across the bilayer, but alternates down the long axis of the fibril. In the last arrangement (panel C), enantiomers pair across the bilayer and alternate down the fibril long axis.

The possibility of enantiomers assembling as depicted in Figure 6A was investigated using isotope-edited Fourier transform infrared spectroscopy (IE-FTIR). This technique has been utilized to monitor the assembly and alignment of β -sheet peptides.^{41,74–78} The fibrils that constitute the MAX1 and

DMAX1 hydrogel networks exhibit a characteristic strong amide I stretch at $\sim 1615\text{ cm}^{-1}$ due to their large β -sheet secondary structure content.⁷⁹ Spectra also included a weaker band centered at 1680 cm^{-1} . This band can be assigned either to the intramolecular arrangement of antiparallel β -strands within each hairpin of the assembly or the β -turn secondary structure also contained with each hairpin.⁸⁰ In the experiment presented in Figure 7A, $^{13}\text{C}/^{18}\text{O}$ -labeled amino acids were incorporated into MAX1 at Lys6 and Val14 (namely, MAX1-IR1, Table 1). A control gel prepared from pure MAX1-IR1 exhibits a red-shifted vibrational frequency stretch at 1571 cm^{-1} due to the heavy-atom labeled carbonyl groups⁸¹ (Figure 7A). The large magnitude of this frequency shift is indicative of the extended vibrational coupling between labeled amides that traverse uninterrupted along the long axis of the fibrils constituting the gel. Conceptually, if one prepares a racemic gel comprising equimolar mixtures of MAX1-IR1 and DMAX1 and the enantiomers self-segregate into monolayers, then a similar shift should be observed.

A second control gel was prepared using MAX1-IR2 (Table 1). This peptide incorporates only one $^{13}\text{C}=^{18}\text{O}$ group at Val7. As such, a gel prepared from this peptide alone will consist of fibrils in which only alternating β -strands contain the label, and thus the extended coupling is lost. This results in a stretch at 1582 cm^{-1} , blue-shifted relative to the pure MAX1-IR1 gel, Figure 7B. Again, conceptually, if one prepares a racemic gel comprised of MAX1-IR1 and DMAX1 and the peptides now alternate within each of the monolayers of a racemic fibril, the extended coupling would be lost. This would result in a blue-shift relative to that observed for a pure MAX1-IR1 gel.

Figure 7C shows the IE-FTIR spectrum of a hydrogel resulting from the equimolar mixture of MAX1-IR1 with unlabeled DMAX1. The lower vibrational frequency stretch at 1579 cm^{-1} is blue-shifted relative to that in pure MAX1-IR1 (dotted line), indicating that extended vibrational coupling does not occur. On the basis of these results, the arrangement depicted in Figure 6A is unlikely. Rather, MAX1 and DMAX1 enantiomers coassemble within each monolayer that defines the racemic fibril, as shown in Figure 6B or C. These two arrangements differ in the formation of facial associations made between the enantiomers across the bilayer. In panel B, homochiral facial contacts are made, whereas in panel C, MAX1 and DMAX1 make heterochiral facial contact. A further complication exists in that enantiomers within a given fibril monolayer can adopt a *Syn* orientation, where β -hairpin turns align along the monolayer, or an *Anti* arrangement, where turns alternate down the long axis of the fibril as shown in panels B and C.

To examine the arrangement of enantiomers within the fibril monolayer as well as across the fibril bilayer, solid state nuclear magnetic resonance (ssNMR) was employed. We recently determined the molecular structure of pure MAX1 in its fibrillar state (PDB file 2N1E) through a series of REDOR, PITHIRDS-CT, and two-dimensional (2D) ^{13}C - ^{13}C radio frequency-assisted diffusion (RAD) measurements.⁴⁹ MAX1 adopts a *Syn/Anti* arrangement where the β -turns of each hairpin align *Syn* along a given monolayer. Across the bilayer, hairpins are arranged *Anti* where β -turns occupy opposite fibril edges, much like the arrangement depicted in Figure 6A if all the peptides were MAX1. With the structural knowledge of pure MAX1 fibrils in hand, isotopically labeled peptides were designed to identify relative enantiomer orientation within the racemic fibril monolayer and across the bilayer. Peptides

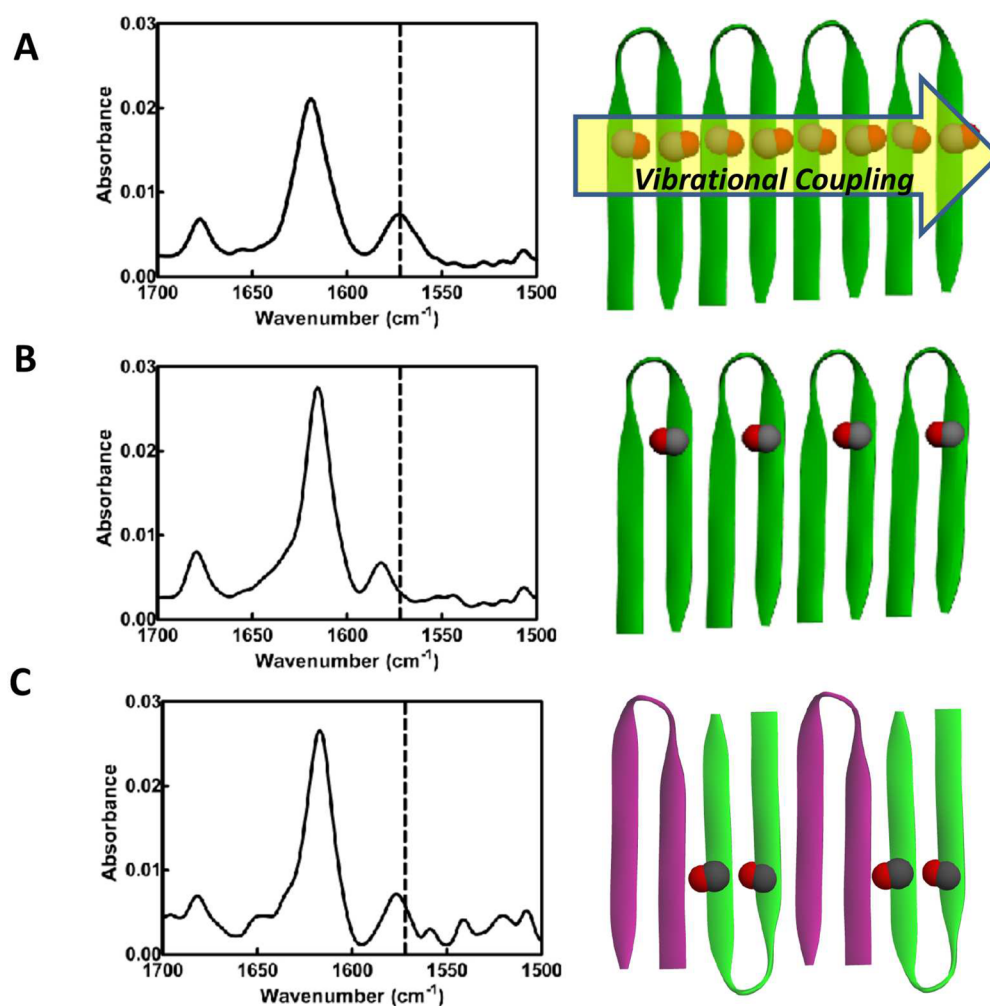


Figure 7. Isotope edited FTIR spectra of MAX1-IR1 (A), MAX1-IR2 (B), and 1:1 MAX1-IR1/DMAX1 (C) hydrogels. Dotted line indicates the coupled stretch at 1571 cm⁻¹ for reference.

DMAX1-NMR1 and MAX1-NMR1 (Table 1, Figure 8) were designed with ¹³C-labeled ^DP11 and V1 respectively to allow for the determination of enantiomeric orientation within the fibril monolayer using PITHIRDS-CT measurements. MAX1-NMR2 was designed with uniformly labeled P11, T12, and V20 residues to enable the identification of bilayer partners through the absence or presence of through-space cross-peaks between isotopically labeled residues in 2D RAD measurements.

To test if enantiomers adopt an *Anti* or *Syn* orientation within the racemic monolayer, a gel comprising a 1:1 mixture of MAX1-NMR1/DMAX1-NMR1 was prepared. PITHIRDS-CT measurements of this sample reveal signal decays consistent with a close proximity between isotopically labeled ^DP11 and V1, Figure 8B. This experiment affords a measured distance of approximately 6.5 Å between the labeled proline and valine, which can only be achieved if the enantiomers are oriented in an *Anti* arrangement where β-turns alternate along the fibril long-axis, Figure 8A.

Next, the relative enantiomer pairing across the bilayer was examined by preparing a gel comprising an equimolar mixture of labeled MAX1-NMR2 and unlabeled DMAX, Table 1. If MAX1-NMR2 makes homochiral interfacial contact as depicted in Figures 6B and 8C, intermolecular coupling between two MAX1-NMR2 hairpins should be observed across the bilayer. The absence of interfacial-residue coupling would argue against

this arrangement and argue for heterochiral pairing as depicted in Figure 6C, where the distances between spin labels are too long (>8 Å) to observe intermolecular coupling. Figure 8D shows one-dimensional (1D) slices at the C α (green) and C β (blue) chemical shifts of labeled valine 20 (V20) derived from 2D RAD ¹³C–¹³C spectra of fibrils comprised of a 1:1 MAX1-NMR2/DMAX mixture. Careful inspection of the 2D spectrum with a 500 ms mixing period suggests the possibility of weak Val20-Thr12 and Val20-Pro11 cross-peaks, which are not observed in the 2D RAD spectrum with a 25 ms mixing period (Figure S20). Although these observations support the arrangement depicted in Figure 6B, we cannot definitively rule out heterochiral pairing as shown in panel C. This is because the Val20 signals in the spectra are not fully resolved from Pro11 and Thr12 signals, making it difficult to clearly distinguish these weak intermolecular cross-peaks from stronger intrasidue Pro11-Thr12 cross-peaks. At any rate, the IR and NMR studies support the formation of racemic fibrils comprised of an *Anti* arrangement of alternating enantiomers in each monolayer, and the possible homochiral pairing of enantiomers across the bilayer as depicted in Figure 6B.

Molecular Basis for Enhanced Fibril Stiffness: Predictions from Pauling and Corey. Thus far, the data indicate that the racemic gel of MAX1 and DMAX1 is more

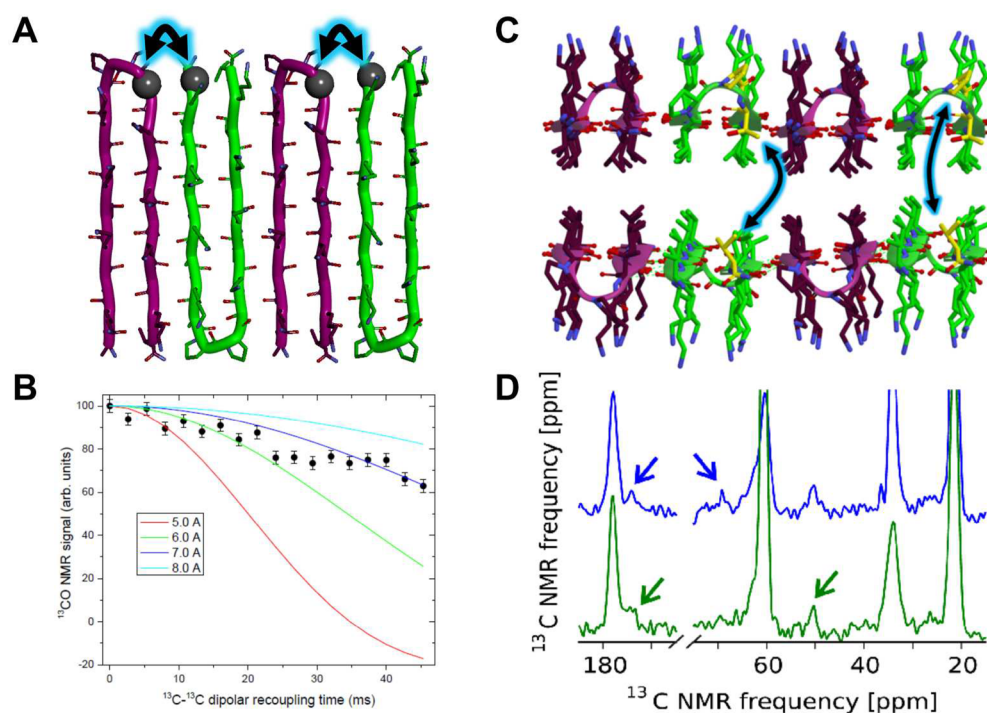


Figure 8. ssNMR of racemic fibrils. (A, B) PITHIRDS-CT experiments indicate an *Anti* arrangement of enantiomers within each fibril monolayer employing isotopically labeled DMAX1-NMR1 and MAX1-NMR1 peptides. Color-coded lines in panel B are simulations of PITHIRDS-CT data for ^{13}C spin pairs with the indicated distances. (C, D) Expected couplings between Val20 and Thr12 (shorter interaction at left) and between Val20 and Pro11 (longer interaction at right) if peptides are paired across the bilayer in a homochiral fashion. Spectra and arrows are defined in the main text.

mechanical rigid because the individual fibrils that comprise the gel are themselves more rigid. Further, these fibrils contain both enantiomers arranged in a unique alternating molecular orientation within each monolayer of a fibril's bilayered β -sheet rich structure. The question remains as to why this molecular arrangement leads to the formation of fibrils that are more stiff than fibrils comprised of either enantiomer alone. Using the structural insights gained from the spectroscopic data, an energy-minimized model of the racemic assembly was constructed and compared to a model of an assembly formed by pure MAX1 based on ssNMR.⁴⁹

Shown in the top panel of Figure 9A is a dimer formed by two enantiomers within a monolayer of the racemic fibrillar assembly. Although each individual hairpin contains two intramolecular β -strands arranged in an antiparallel fashion, the β -sheet interface between the two hairpins comprises β -strands aligned in a *parallel* manner. Thus, MAX1 and DMAX1 form an extended network of *intermolecular parallel* β -strands that traverse the long axis of a given fibril. In contrast, hairpins within enantiomerically pure MAX1 fibrils are defined by an extended array of *anti-parallel* hydrogen bonded β -strands as shown in Figure 9B. Both the racemic and pure enantiomeric fibrils maximize the hydrogen-bonding potential of every strand in their respective systems. Although the exact geometry of the H-bonds is different in each assembly (bent versus linear), the number of bonds formed is identical. Thus, it is unlikely that enthalpic contributions from H-bonding are responsible for the large enhancement in fibril stiffness for the racemate. The central image in panel A shows a tetrameric portion of a fibril monolayer. The formation of bent H-bonds between enantiomeric strands is evident and is responsible for defining the alignment of strands within the β -sheet assembly. Importantly, these bent H-bonds not only define strand

alignment but also dictate the exact placement of the hydrophobic valine side chains into the dry interior core of the fibril. As shown in the CPK rendering, valine side chains form intimate nested hydrophobic interactions where the entire isopropyl group of one residue nestles between the isopropyl groups of its valine neighbors. This facilitates tight packing and maximizes side-side/side-chain van der Waals contacts. The bottom panel in Figure 9A shows the entire fibril viewed axially down its long axis showing the interior of the bilayer. The staggered arrangement of the nested valines is clearly evident with the side chains of each successive enantiomer translated relative to its mirror image neighbor. These nested interactions are not realized in enantiomerically pure fibrils, Figure 9B (center and bottom images). Here, the antiparallel arrangement of β -strands, with its linear H-bonds, dictates that the valine side chains are arranged in a head-to-head manner within the core of the fibril. Although packing can occur between the valines where one C_γ atom of a valine side chain can penetrate the fork of a neighboring isopropyl group, this interaction does not pack the interior of the fibril as efficient as does nesting of an entire isopropyl side chain. This is especially apparent when comparing the axial views of the fibril interiors, Figure 9 (bottom panels). Interestingly, nested hydrophobic interactions between β -branched residues in sheet-rich proteins frequently occur, as first noted by Curmi et al.⁸² As in naturally occurring proteins where effective packing of side chains stabilizes their folded state, the well-packed interior of the racemic fibril is also stabilizing and results in its local stiffening (Table 2).

Remarkably, the arrangement of β -strands afforded by the racemic assembly results in a supersecondary structure predicted by Pauling and Corey over 60 years ago, which they termed the rippled sheet.⁴⁰ They proposed models of both antiparallel and parallel extended rippled sheets formed by the

arguments with little mention of side chain packing. In fact, the models drawn by Pauling and Corey only show the main chain atoms of the peptide's backbone, omitting all side chains entirely. Our work demonstrates that side-chain packing is indeed important to the overall formation of this unique sheet topology and is responsible for the enhanced material properties enjoyed by the hydrogels formed by the racemic fibrils described here.

The model shown in Figure 9A, which underscores the importance of side chain interactions, makes several predictions. The first is that design can be used to modulate the bulk mechanical properties of the gel by incorporating amino acids that either enhance or diminish side-chain interactions between enantiomers. This prediction was first tested by preparing MAX35 and DMAX35, enantiomers that incorporate four isoleucine residues on their hydrophobic faces among a background of valines, Table 1. When assembled into racemic fibrils, both Ile–Ile and Val–Val pairwise interactions can form between neighboring enantiomers. Ile–Ile cross-pairs within β -sheets are generally rare in natural proteins especially when compared to Ile–Val or Val–Val pairs,⁸² suggesting that they are less energetically favorable. The rheological time-sweep experiment in Figure 10A shows that the MAX35/DMAX35 racemic gel displays only a 2-fold enhancement in rigidity compared to either of the pure enantiomeric gels. Importantly, the enhancement displayed by the MAX35/DMAX35 gel is 2-fold less than that displayed by the MAX1/DMAX1 gel, suggesting weaker inter-enantiomer interactions in the former.

To strengthen the enantiomeric interaction, MARK4 and DMARK4 were designed with residue changes on the hydrophilic face of the hairpin, Table 1. Here, arginine residues were incorporated to enhance inter-enantiomer interactions in the fibrillar state through the formation of inter-guanidino hydrogen bonds. These interactions are not possible in the pure enantiomer assemblies. Figure 10B shows that the racemic gel displays an impressive 11-fold enhancement in its mechanical rigidity. Thus, modulating side-chain–side-chain interactions between enantiomers, whether in the fibril's core or on its solvent exposed exterior, provide a means to tune the gel's bulk properties via molecular-level engineering.

The second prediction our model makes is that the hairpins comprising the “racemic” fibrils need not be related as true enantiomers (e.g., non-superimposable mirror images) to realize the enhancement in fibril stiffness and corresponding gel rigidity. As long as important side-chain/side-interactions are not disrupted, alterations that break mirror image symmetry are tolerated. In fact, the fluorophore-, azide-, and biotin-labeled derivatives used in this study assembled readily with their nonmirror image partners. These hairpins were chemically modified at their strand termini or on their hydrophilic faces; regions of the amphiphile that do not participate in the formation of critical pairwise interactions that stabilize the fibril.

CONCLUSION

Heterochiral peptide assembly affords materials with unique macroscopic traits that are dependent on the sequence and composition of monomers used in their preparation. A racemic mixture of enantiomeric peptides MAX1 and DMAX1 assembled into a fibrillar hydrogel that is 4-fold more rigid than enantiomerically pure gels formed from either peptide alone. Herein, we thoroughly characterize the racemic hydrogel across molecular, local-fibril, and hydrogel network length scales using an arsenal of spectroscopy, microscopy, and

scattering techniques to uncover the physical basis for this enhancement in mechanical rigidity. MAX1 and DMAX1 coassemble into fibrils comprised of a bilayer of peptides that H-bond along the long axis of a given fibril. Within each monolayer, enantiomers coassemble in an alternating fashion forming an extended heterochiral rippled sheet, a structure predicted by Pauling and Corey in 1953. Importantly, valine residues display their side chains into the fibrils' dry interior, forming nested hydrophobic interactions between enantiomers, interactions not accessible within enantiomerically pure fibrils. This arrangement maximizes inter-residue hydrophobic contacts within the interior of fibrils that results in their local stiffening, which in turn, gives rise to the significant increase in bulk mechanical rigidity observed for the racemic hydrogel. We go on to show that modulation of inter-enantiomer interactions within fibrils can be used as a design tool to control the bulk mechanical properties of gels.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acscentsci.7b00115.

Experimental methods, temperature-dependent circular dichroism of MAX1, analytical HPLC and mass spectra of all peptides, oscillatory rheology of hydrogels, micrographs of gold nanoparticle controls, solid NMR spectra of peptide gels and NMR characterization of labeled amino acids (PDF)

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Notes

The authors declare no competing financial interest.

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