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### PERFLUORINATED ALKYL ACIDS IN PLASMA OF AMERICAN ALLIGATORS (*ALLIGATOR MISSISSIPPIENSIS*) FROM FLORIDA AND SOUTH CAROLINA

Jacqueline T. Bangma<sup>†</sup>, John A. Bowden<sup>‡</sup>, Arnold M. Brunell<sup>§</sup>, Ian Christie<sup>||</sup>, Brendan Finnell<sup>#</sup>, Matthew P. Guillette<sup>†</sup>, Martin Jones<sup>††</sup>, Russell H. Lowers<sup>‡‡</sup>, Thomas R. Rainwater<sup>§§</sup>, Jessica L. Reiner<sup>\*,‡</sup>, Philip M. Wilkinson<sup>||||</sup>, and Louis J. Guillette Jr<sup>†,\*\*</sup> <sup>†</sup>Department of Obstetrics and Gynecology, Medical University of South Carolina, Charleston, South Carolina, USA

<sup>‡</sup>Hollings Marine Laboratory, Chemical Sciences Division, National Institute of Standards and Technology, Charleston, South Carolina, USA

§Florida Fish and Wildlife Conservation Commission, Eustis, Florida, USA

<sup>II</sup>Grice Marine Laboratory, College of Charleston, Charleston, South Carolina, USA

<sup>#</sup>Illinois Wesleyan University, Bloomington, Illinois, USA

<sup>††</sup>Department of Mathematics, College of Charleston, Charleston, South Carolina, USA

<sup>‡‡</sup>Integrated Mission Support Service, Kennedy Space Center, Titusville, Florida, USA

§§Baruch Institute of Coastal Ecology and Forest Science, Clemson University, Georgetown, South Carolina, USA

III Tom Yawkey Wildlife Center, Georgetown, South Carolina, USA

#### Abstract

The present study aimed to quantitate 15 perfluoroalkyl acids (PFAAs) in 125 adult American alligators at 12 sites across the southeastern United States. Of those 15 PFAAs, 9 were detected in 65% to 100% of samples: perfluorooctanoic acid, perfluorononanoic acid, perfluorodecanoic acid (PFDA), perfluoroundecanoic acid (PFUnA), perfluorododecanoic acid, perfluorotridecanoic acid (PFTriA), perfluorotetradecanoic acid, perfluorohexanesulfonic acid (PFHxS), and perfluorooctane sulfonate (PFOS). Males (across all sites) showed significantly higher concentrations of 4 PFAAs: PFOS (p = 0.01), PFDA (p = 0.0003), PFUnA (p = 0.021), and

Deceased

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<sup>\*</sup>Address correspondence to jessica.reiner@nist.gov.

Supplemental Data—The Supplemental Data are available on the Wiley Online Library at DOI: 10.1002/etc.3600.

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Data availability—Data are available upon request from the author (jessica.reiner@nist.gov). Additional sample information and data are provided in Supplemental Data.

PFTriA (p = 0.021). Concentrations of PFOS, PFHxS, and PFDA in plasma were significantly different among the sites in each sex. Alligators at both Merritt Island National Wildlife Refuge (FL, USA) and Kiawah Nature Conservancy (SC, USA) exhibited some of the highest PFOS concentrations (medians of 99.5 ng/g and 55.8 ng/g, respectively) in plasma measured to date in a crocodilian species. A number of positive correlations between PFAAs and snout–vent length were observed in both sexes, suggesting that PFAA body burdens increase with increasing size. In addition, several significant correlations among PFAAs in alligator plasma may suggest conserved sources of PFAAs at each site throughout the greater study area. The present study is the first to report PFAAs in American alligators, to reveal potential PFAA hot spots in Florida and South Carolina, and to provide a contaminant of concern when assessing anthropogenic impacts on ecosystem health. *Environ Toxicol Chem* 2016;9999:1–9. Published 2016 Wiley Periodicals Inc. on behalf of SETAC. This article is a US government work and, as such, is in the public domain in the United States of America.

#### Keywords

Perfluorooctane sulfonate (PFOS); Perfluorohexanesulfonic acid (PFHxS); Alligator; Crocodilian; Plasma

#### INTRODUCTION

Despite being manufactured for more than 50 yr [1], it was not until 2000 that the class of chemicals known as perfluoroalkyl acids (PFAAs) entered the scientific spotlight as a major environmental contaminant of concern [2]. The 2 most commonly known PFAAs, perfluorooctane sulfonate (PFOS) and perfluorooctanoic acid (PFOA), were produced by 3M in 1948 and 1947 [1], respectively, the latter of which was subsequently purchased by DuPont in 1951 [3]. A variety of new PFAAs have steadily been introduced to the market since then. Structurally, PFAAs can widely vary; but as a whole, they typically consist of carbon chains of varying length (linear and branched isomers), an acid functional group, and hydrogen atoms substituted with fluorine atoms [4]. The carbon–fluorine bonds are the unique feature of PFAAs and provide chemical and thermal stability [5]. Two well-studied families of PFAAs are carboxylic acids and sulfonic acids [2,6].

The usage of PFAAs has become widespread since the introduction of these chemicals in the 1940s, largely because they exhibit unique surfactant properties that make them attractive components for many consumer-related products, such as nonstick pans, water-repellant surfaces, hair products, plastics, and lubricants [2], as well as firefighting products known as aqueous film-forming foams [7]. Active manufacturing and use of certain PFAAs, such as PFOS and PFOA, have largely ceased as a result of a voluntary phaseout by industry. Current production of fluorinated chemicals includes shorter-chained carboxylic and sulfonic acid substitutes, such as perfluorobutanesulfonic acid (PFBS) and perfluorobutyric acid (PFBA) [8]. In addition, precursor chemicals that have a nonfluorinated structural component attached to a perfluorinated chain may be amenable to microbial or chemical transformation and have the potential to degrade into perfluorinated carboxylic and sulfonic acids over time [9].

The same properties that make PFAAs commercially valuable (e.g., the highly stable carbon–fluorine bonds) also enable them to persist in the environment by resisting chemical, microbial, and photolytic degradation. However, unlike the more lipophilic environmental contaminants, such as organo-chlorine pesticides, polychlorinated biphenyls (PCBs), and brominated flame retardants (PBDEs) that are sequestered in adipose tissue, PFAAs accumulate in the blood and blood-rich organs, such as the liver [10,11]. Conversely, like organochlorine pesticides, PCBs, and PBDEs, PFAAs have also been shown to bioaccumulate and biomagnify in food webs [6]. Increasing PFAA chain length has been shown to correlate with an increasing ability to bioaccumulate [12], and the greatest PFAA concentrations detected in wildlife have been in species occupying high trophic positions [13]. Because PFAAs are bioaccumulative and often observed in higher concentrations in fish-eating marine species [13], humans who consume more fish in their diet may be at higher risk of PFAA exposure than those who consume less fish [14].

Animal studies reveal a wide range of PFAA-related effects that include alterations in liver physiology and serum cholesterol, as well as resulting hepatomegaly, wasting syndromes, neurotoxicity, and immunotoxicity [15–17]. In addition, PFAAs have been mentioned as possible obesogens because of their interaction with peroxisome proliferator–activated receptors[18]. However, although species-specific variations in PFAA excretion rates have been observed [2], the actual mechanism of action of PFAA toxicity is not well understood across species.

Few reports exist on PFAA distribution and body burdens in North American wildlife, and studies of PFAAs in wild reptiles and amphibians have been limited almost exclusively to frogs and sea turtles [19]. Globally, only 3 studies have examined PFAAs in crocodilians [20–22]. Because of their high trophic status, long life span, and high site fidelity, crocodilians are attractive study species for ecotoxicological investigations, particularly those involving exposure and accumulation of persistent environmental contaminants [23–25]. As such, studies examining PFAAs in crocodilians can provide insight into exposure and potential effects in focal species and can identify potential hot spots of PFAA contamination.

In the present study, we examined PFAA concentrations in plasma of wild American alligators (*Alligator mississippiensis*) from 12 sites in Florida and South Carolina, USA (Figure 1). Because factors such as sex, body size, and location may influence PFAA concentrations in alligators, the relationships between PFAA body burdens and these parameters were also examined.

#### MATERIALS AND METHODS

#### Study area

American alligator plasma samples (n = 125) were collected between 2012 and 2015 as part of multiple ongoing projects examining the biology and ecotoxicology of alligators in Florida and South Carolina [23–25]. In South Carolina, alligator blood samples were collected from the following sites (in order of north to south): Tom Yawkey Wildlife Center (YK; n = 10), Kiawah Island (KA; n = 10), and Bear Island Wildlife Management Area (BI;

n = 10) (Figure 1; Supplemental Data, Table S1). In Florida, samples were collected from the following sites (in order from north to south): Lochloosa Lake (LO; n = 10), Lake Woodruff (WO; n = 10), Lake Apopka (AP; n = 10), Merritt Island National Wildlife Refuge (MI; n = 15), St. Johns River (JR; n = 10), Lake Kissimmee (KS; n = 10), Lake Trafford (TR; n = 10), Everglades Water Conservation Area 2A (2A; n = 10), and Everglades Water Conservation Area 3A (3A; n = 10) (Figure 1; Supplemental Data, Table S1).

#### Sample collection

Immediately following capture of the alligator, a blood sample was collected from the postoccipital sinus of the spinal vein of each animal using a sterile needle and syringe [23–25]. Whole-blood samples were then transferred to 8-mL lithium–heparin Vacutainer blood-collection tubes (BD), stored on ice in the field, and later centrifuged at 2500 rpm at 4 °C for 10 min to obtain plasma, which was stored at –80 °C until analysis. Snout–vent length was measured for each animal as a proxy for size, and sex was determined by cloacal examination of the genitalia [26].

The National Institute of Standards and Technology (NIST) Standard Reference Material<sup>®</sup> (SRM) 1958 Organic Contaminants in Fortified Human Serum (Freeze-Dried) was used as a control material during PFAA analysis. The freeze-dried human serum SRM 1958 was reconstituted with deionized water according to the instructions on the certificate of analysis [27] and analyzed alongside collected alligator plasma.

#### Chemicals

Calibration solutions were created by combining 2 solutions produced by the NIST reference materials 8446 Perfluorinated Carboxylic Acids and Perfluoroctane Sulfonamide in Methanol and 8447 Perfluorinated Sulfonic Acids in Methanol. Together, the solution contained 15 PFAAs as follows: PFBA, perfluoropentanoic acid (PFPeA), perfluorohexanoic acid (PFHxA), perfluoroheptanoic acid (PFHpA), PFOA, perfluorononanoic acid (PFNA), perfluorodecanoic acid (PFDA), perfluorodecanoic acid (PFDA), perfluorotridecanoic acid (PFTriA), perfluorotetradecanoic acid (PFTA), PFBS, perfluorohexanesulfonic acid (PFHxS), PFOS, and perfluoroctanesulfonamide (PFOSA).

Internal standards were purchased from Cambridge Isotope Laboratories, RTI International, and Wellington Laboratories to create an internal standard mixture comprised of 11 isotopically labeled PFAAs, as follows: <sup>13</sup>C<sub>4</sub>-PFBA, <sup>13</sup>C<sub>2</sub>-PFHxA, <sup>13</sup>C<sub>8</sub>-PFOA, <sup>13</sup>C<sub>9</sub>-PFNA, <sup>13</sup>C<sub>9</sub>-PFDA, <sup>13</sup>C<sub>2</sub>-PFUnA, <sup>13</sup>C<sub>2</sub>-PFDoA, <sup>18</sup>O<sub>2</sub>-PFBS, <sup>18</sup>O<sub>2</sub>-PFHxS, <sup>13</sup>C<sub>4</sub>-PFOS, and <sup>18</sup>O<sub>2</sub>-PFOSA.

#### Sample preparation

Samples were extracted using a method described by Reiner et al. [28]. Approximately 1 mL of each alligator plasma sample and SRM 1958 aliquots were thawed and gravimetrically weighed. All samples were then spiked with the internal standard mixture (approximately 600  $\mu$ L) and gravimetrically weighed. After brief vortex-mixing and 90 min of equilibration, 4 mL of acetonitrile were used to extract the PFAAs from each sample. After sonication and

centrifugation, the supernatant was removed from all samples. The collected supernatant was then solvent-exchanged to methanol and further purified using an ENVI-Carb solid-phase extraction cartridge (Supelco). Resulting extracts were evaporated to 1 mL using nitrogen gas prior to being analyzed by liquid chromatography-tandem mass spectrometry.

Samples were analyzed using an Agilent 1100 high-performance liquid chromatographic system coupled to an Applied Biosystems API 4000 triple-quadrupole mass spectrometer with electrospray ionization in negative mode. Samples were examined by liquid chromatography using an Agilent Zorbax Eclipse Plus C18 analytical column (2.1 mm × 150 mm × 5  $\mu$ m). A ramping liquid chromatography solvent gradient was employed using methanol and deionized water, both containing 20 mmol/L ammonium acetate [28]. Two multiple reaction monitoring transitions for each PFAA were monitored to ensure no interferences with measurements. One was employed for quantitation, and the other was used for confirmation [28].

#### **Quality control**

All alligator plasma samples were processed alongside quality control material NIST SRM 1958 to determine the accuracy and precision of the method. The PFAA levels of SRM 1958, processed during our extraction, had to be within the 95% confidence interval as reported on the certificate of analysis. All samples were also processed alongside blanks to assess any background contamination that might be present in the laboratory or a result from the extraction method. Compounds were considered to be above the reporting limit if the mass of an analyte in the sample was greater than the mean plus 3 standard deviations of all blanks.

#### Statistical methods

All statistical analyses were performed using SPSS statistic 22 (IBM). Statistical tests were performed for the compounds detected in more than 75% of the samples: PFNA, PFDA, PFUnA, PFDoA, PFTriA, PFTA, PFHxS, and PFOS. Unlike previous environmental studies, where PFOA had the second highest concentration measured, PFOA was detected at much lower frequency (detected in only 65% of the samples analyzed). With a full one-third of PFOA measurements falling below the limit of detection, PFOA was excluded from statistical analysis, along with the remaining chemicals (PFHpA, PFHxA, PFPeA, PFBS, and PFBA), which were detected in <2% of the samples. For those PFAAs included in statistical analysis, compounds below the limit of detection were set equal to one-half of the limit of detection prior to running the statistical tests [29].

Sex-based differences of PFAAs in Florida and South Carolina were investigated using univariate analysis of variance with log normally distributed concentration values, and a Friedman's test was used for the PFAAs with non-normally distributed concentration values. Site was set as the nuisance factor, sex as the treatment, and PFAA concentration as the dependent variable. These tests simulated a randomized block design for the collected data. Other parametric tests employed for data analysis of sex-based differences, on a site-by-site basis, and analysis of site differences for PFAA levels included a *t* test and one-way analysis of variance when data were normal or log-normal and Friedman rank test, Mann-Whitney *U* 

test, and Kruskal-Wallis test when data remained non-normal following log transformation. Pearson correlation and Spearman correlation were used when applicable for correlative measures.

#### **RESULTS AND DISCUSSION**

In the present study, we collected a total of 125 plasma samples from alligators at multiple sites in Florida and South Carolina to examine PFAA concentrations in animals from different localities. Of the 15 PFAAs included in the present analysis, all samples contained at least 6 PFAAs. The following 5 PFAAs were detected in every plasma sample analyzed (in order of highest median concentration to lowest median concentration, among all sites): PFOS (median, 11.2 ng/g; range, 1.36-452 ng/g), PFUnA (median, 1.58 ng/g; range, 0.314-18.4 ng/g), PFDA (median; 1.20 ng/g, range; 0.169–15.1 ng/g), PFNA (median, 0.528 ng/g; range, 0.155–1.40 ng/g), and PFHxS (median, 0.288 ng/g; range, 0.057–23.3 ng/g) (Table 1; Supplemental Data, Table S2). Also, PFDoA, PFTriA, PFTA, and PFOA were detected frequently in alligator plasma samples (in more than 96%, 94%, 75%, and 65%, respectively): PFDoA (median, 0.363 ng/g; range, <0.009–7.27 ng/g), PFTriA (median, 0.416 ng/g; range, <0.026-2.60 ng/g), PFTA (median, 0.050 ng/g; range, <0.008-1.38 ng/g), and PFOA (median, 0.064 ng/g; range, <0.008–0.412 ng/g) (Table 1; Supplemental Data, Table S2). The 9 PFAAs commonly measured over the limit of detection resulted in unique fingerprints for each site (Supplemental Data, Figure S1), which are discussed in the section Site differences. The shorter-chain PFAAs (PFHpA, PFHxA, PFPeA, PFBS, and PFBA) were detected infrequently (<2% of the samples) and, therefore, not included in any statistical analysis.

#### Sex differences

As a whole, across all sites, male alligators exhibited significantly higher concentrations of several PFAAs in plasma compared with females as a group: PFOS (p = 0.01), PFDA (p = 0.0003), PFUnA (p = 0.021), and PFTriA (p = 0.021) (Supplemental Data, Figure S2). At some individual sites, however, PFAA concentrations were significantly higher in females (e.g., PFOS at AP, PFUnA at KA).

In a population of captive Chinese alligators (*Alligator sinensis*), Wang et al. [21] found the highest PFAA concentration in serum to be that of PFUnA rather than PFOS, the PFAA with the highest concentrations in the present study. However, similar to the present study, male Chinese alligators contained significantly higher concentrations of PFOS and PFUnA compared with females. Wang et al. [21] did not find a sex-based difference for PFDA in Chinese alligators. Christie et al. [22] did not find any sex-based differences in their population of Nile crocodiles (*Crocodylus niloticus*) in South Africa. It is possible that sex-based differences observed for certain PFAAs in alligators are the result of a differential clearance of these contaminants between males and females, as has been observed in rats [30], mice [31], and other mammals [32]. It is also possible that females may offload PFAAs during oviposition, reducing their PFAA body burden compared with males at the same locality. This possibility is supported by studies reporting measurable concentrations of PFAAs in eggs of herring gulls (*Larus argentatus*) [33] and Nile crocodiles [20], confirming

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maternal transfer of PFAAs in oviparous species. Sex-specific differences in PFAA concentrations may also be the result of differential habitat use by adult males and females, a phenomenon common among crocodilians [34–37]. In such cases, differences in prey availability and contamination between and among habitats within a site could result in different PFAA exposures in males and females.

Because no sex-specific differences in PFOA, PFNA, PFHxS, PFDoA, and PFTA concentrations were observed as a whole (all sites combined), sex-based differences were examined on a site-by-site basis (Supplemental Data, Table S3). Overall, only a few sites exhibited sex-based differences for these 5 PFAAs (Supplemental Data, Figure S3). At LO, male alligators had significantly higher PFNA (p = 0.016), PFTA (p = 0.032), and PFDoA (p = 0.032) plasma concentrations compared with females; and at MI males had significantly higher PFOA (p = 0.047) plasma concentrations than females. Interestingly, PFHxS was the only PFAA for which females exhibited significantly higher plasma concentrations (YK, p = 0.008; TR, p = 0.008) when compared with males (Supplemental Data, Figure S3). It is important to note that our examination of sex-based differences in PFAA concentrations may have been influenced by small sample sizes; in almost all cases, only 5 males and 5 females were sampled per site.

#### Site differences

Because sex-based differences in PFAA concentrations were observed among alligator plasma samples, site differences were determined separately for males and females. All of the 9 detected PFAAs displayed at least some minor site differences. The PFAAs that displayed the most notable site differences (the highest number of statistically significant groups between the 12 sites) were PFOS, PFDA, and PFHxS. Of those, PFOS exhibited the greatest statistical difference across sites (Figure 2). This is likely because PFOS is generally the most abundant PFAA in the environment. For male alligators only, concentrations of PFOS in plasma ranged from 1.57 ng/g to 452 ng/g. Concentrations of PFOS were highest at MI (median, 106 ng/g) and KA (median, 56.4 ng/g). Males from MI exhibited significantly higher PFOS concentrations compared with males from all other sites, with the exception of KA. In addition, the individual alligator with the highest overall PFOS concentration measured in the present study (452 ng/g plasma) was from MI. After MI, males from South Carolina (KA, YK, and BI) exhibited higher PFOS concentrations observed in males, with the exception of WO. Some of the lowest PFOS concentrations observed in males in the present study were measured at sites 2A, 3A, LO, and JR.

For female alligators, PFOS concentrations in plasma ranged from 1.36 ng/g to 206 ng/g. Similar to males, females from sites MI (median, 85.5 ng/g) and KA (median, 51.3 ng/g) exhibited significantly higher PFOS concentrations compared with the other sites examined, and the individual female with the highest PFOS concentration was from MI (206 ng/g plasma). After MI and KA, females from the 2 other South Carolina sites (YK and BI) exhibited higher PFOS concentrations than females from Florida, with the exception of WO and AP. Some of the lowest PFOS concentrations observed in females in the present study were measured at sites 2A, 3A, LO, JR, and TR.

The concentrations of PFHxS detected in alligator plasma in the present study exhibited a similar trend to PFOS across sites but on a reduced scale (Supplemental Data, Table S4). For males, PFHxS plasma concentrations ranged from 0.057 ng/g to 23.3 ng/g. Males from MI (median, 3.95 ng/g) had significantly higher PFHxS concentrations than those at any other site examined, and the individual male with the highest PFHxS concentration was from Merritt Island National Wildlife Refuge (23.3 ng/g). Males from Kiawah Island and Lake Kissimmee followed closely but were still statistically grouped with other sites (Lake Apopka, Lake Woodruff, and Bear Island). The lowest PFHxS concentrations in males were typically measured at sites 2A, 3A, Lochloosa Lake, and Lake Trafford. For female alligators, PFHxS concentrations in plasma ranged from 0.069 ng/g to 10.0 ng/g. Similar to males, MI females exhibited significantly higher PFHxS concentrations than those at all other sites. Females from KA and KS had the next highest concentrations but were still statistically grouped with those from other sites (AP, WO, and YK). The lowest PFHxS concentrations in females were typically observed at sites 2A, 3A, and LO.

Across the sampling sites, PFDA had a unique signature, one that varied from the patterns observed for plasma PFOS and PFHxS concentrations (Supplemental Data, Table S4). For male alligators, PFDA concentrations ranged from 0.498 ng/g to 15.1 ng/g. KA males had significantly higher PFDA concentrations overall (median, 6.21 ng/g) compared with all sites, with the exception of YK (median, 6.20 ng/g). Males from YK exhibited the next highest PFDA concentrations, but these were not significantly different from those detected in WO males (median, 2.02 ng/g). Males from many of the remaining sites had similarly low concentrations of PFDA. Overall, LO males (median, 0.792 ng/g) had some of the lowest PFDA concentrations in females were also in South Carolina: KA (median, 6.32 ng/g) and YK (median, 5.55 ng/g). Concentrations of PFDA at BI (median, 1.18 ng/g) and WO (median, 1.84 ng/g) followed closely behind but were not significantly different from the bill of the other sites sampled. Like males, LO females (median, 0.501 ng/g) had some of the lowest PFDA concentrations across all sites.

Overall, male and female alligators from both MI and KA exhibited some of the highest PFOS concentrations measured to date in a crocodilian species (median PFOS concentrations in plasma: MI males = 106 ng/g; MI females = 85.5 ng/g; KA males = 56.4 ng/g; KA females = 51.3 ng/g). In comparison, the mean PFOS concentration in serum from captive Chinese alligators was 28.7 ng/mL (28.0 ng/g) [21], whereas the median concentrations in wild Nile crocodiles at several sites in South Africa ranged from 4.31 ng/g to 50.3 ng/g [22]. In a study of other reptiles, loggerhead sea turtles (*Caretta caretta*) along the coast of South Carolina and Florida exhibited median PFOS plasma concentrations of 2.87 ng/g and 3.80 ng/g, respectively [38]. In another study, hawksbill sea turtles (*Eretmochelys imbricata*) off of Juno Beach (FL, USA), south of MI, showed higher than expected PFOS levels at 11.9 ng/g [39] when compared with several other species of turtles along the east coast. The authors discuss geographic differences as possible reasons for the higher levels of PFOS in the hawksbill. Comparing the results Keller et al. [39] found for hawksbill with the present results would suggest that there might be a potential source of PFOS off the east coast of Florida. It is possible that, for the present study, the high

concentrations of PFOS and PFHxS detected in male and female alligators at MI may be related to the aeronautic facilities located in and around MI, which comprise a large part of Florida's Kennedy Space Center. The past use of aqueous film-forming foams at Kennedy Space Center may have played a role in PFAAs in the surrounding environment and wildlife. Historically, aqueous film-forming foams have been shown to contain PFAAs, such as PFOS and PFHxS, as well as a number of other proprietary PFAA mixtures [7], and can be resistant to remediation [9]. Perfluoroalkyl acids have been measured in firefighters [40], in wildlife [6], and downstream of their use [41]. Potential sources of PFOS and PFHxS at KA are more speculative. In addition, it should be noted that, with the exclusion of MI, alligators from the South Carolina sites (BI, YK, and KA) had some of the highest PFOS concentrations compared with the Florida sites. In Florida, WO exhibited mid to high concentrations of PFOS, PFHxS, and PFDA compared with other sites sampled. For many years, WO has been used as a reference site for multiple studies on alligator ecotoxicology because of its relatively low concentrations of organochlorine contaminants, such as DDT, its metabolites, and other organochlorine pesticides [42]. The results of the present study indicate that WO would not be a suitable reference site for future studies involving PFAAs. In contrast to WO, sites 2A and 3A, which are located in the Everglades, exhibited some of the lowest concentrations of PFOS and PFHxS measured in Florida. Surprisingly similar levels of PFOS were found in loggerhead sea turtles (3.67 ng/g) from Florida Bay close to the 2A and 3A alligator sampling sites [38]. Interestingly, whereas PFAA concentrations appear to be relatively low in 2A and 3A alligators, the same adult alligators at these sites have been reported to contain some of the highest mercury concentrations in Florida and throughout the range of the species [25,43,44].

#### Correlations

For all alligators included in the present study, snout-vent length was uniform across sites for males and nearly uniform across sites for females (Supplemental Data, Figure S4). Thus, data from all sites were combined within each sex to investigate relationships between PFAA concentration and alligator snout-vent length. Because MI had a very different pattern of PFAAs compared with the other sites, correlations were run with and without this site included. The significance did not change based on the inclusion or exclusion of this site, and thus it was included in the analysis. Correlations comparing both male snout-vent length and female snout-vent length to PFAAs resulted in a number of significant positive correlations (Table 2). Overall, females exhibited higher correlation coefficients between PFAA concentration and snout-vent length when compared with males. The highest correlation coefficients for females were with PFTriA, which explained 57.0% of the variation, followed closely by PFOS, which explained 55.1% of the variation. In contrast, the highest correlation coefficients for male snout-vent length and PFAA concentration were for PFUnA, which explained 35.5% of the variation, followed closely by PFOS, which explained 33.1% of the variation. This would seem to refute the hypothesis that female alligators may have a lower PFAA burden than male alligators as a result of maternal transfer of PFAAs during oviposition. Collectively, these data suggest that concentrations of some PFAAs in adult American alligators increase with increasing body size in both males and females. Conversely, Wang et al. [21] found that PFAA (specifically PFUnA, PFDA, and PFNA) concentrations decreased with increasing body size (total length). These observed

differences between American and Chinese alligators may be the result of many factors, including the combination of including animals from all sites in the present study, interspecific differences in food consumption, growth rate differences, and differences in body size [45], as well as differences in toxicodynamics and toxicokinetics of PFAAs. In addition, differences in diet and numerous environmental variables between wild (present study) and captive [22] alligators may influence growth and body burdens of PFAAs.

With all sites combined for each sex, significant correlations were observed between different PFAAs measured in plasma, suggesting somewhat similar sources of PFAA contamination across the sampling localities. The varying levels of PFAA contamination from site to site are likely the result of varying distances from these potential PFAA sources. Some correlative relationships between the PFAAs were stronger than others (Table 3). Of all the PFAAs, correlations between PFUnA and PFDoA for male (p < 0.01, r = 0.920) and female (p < 0.01, r = 0.938) alligators across the sites were the most highly significant relationships observed in the present study.

#### CONCLUSIONS

The present study is the first to quantitate PFAA concentrations in American alligators and one of the few studies to quantitate PFAAs in crocodilians [20–22]. All alligator samples (*n* = 125) contained the 5 following PFAAs: PFOS (median, 11.2 ng/g; range, 1.36–452 ng/g), PFUnA (median, 1.58 ng/g; range, 0.314–18.4 ng/g), PFDA (median, 1.20 ng/g; range, 0.169–15.1 ng/g), PFNA (median, 0.528 ng/g; range, 0.155–1.40 ng/g), and PFHxS (median, 0.288 ng/g; range, 0.057–23.3 ng/g). The present findings support sex-based differences in PFOS and PFUnA concentrations previously observed in captive Chinese alligators [21], while demonstrating opposite relationships between PFAA concentration and body size for American (wild) and Chinese (captive) alligators. The high number of significant PFAA-to-PFAA correlations suggests common point sources throughout the sampling sites in Florida and South Carolina. The present study also reveals potential hot spots for various PFAAs (e.g., PFOS at KA and MI) that warrant further investigation and provides another contaminant of concern to be combined with organochlorines, metals, and others when assessing overall anthropogenic impacts on ecosystem health.

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#### Figure 1.

Map showing the 12 sites (SC and FL, USA) from which American alligators (*Alligator mississippiensis*) were sampled in the present study (n = 125) during the years 2012 to 2015. Collection sites are listed in decreasing latitude. WCA = Everglades Water Conservation Area.



#### Figure 2.

Site comparison of mean (±standard deviation) perfluorooctane sulfonate concentrations (log) in (**A**) male and (**B**) female American alligator (*Alligator mississippiensis*) plasma from multiple sites in Florida and South Carolina. Letters above bars represent statistically significant differences between groups (p < 0.05). Samples are listed from left to right in decreasing latitude. 2A/3A = Everglades Water Conservation Areas 2A and 3A; AP = Lake Apopka; BI = Bear Island Wildlife Management Area; JR = St. Johns River; KA = Kiawah Island; KS = Lake Kissimmee; LO = Lochloosa Lake; MI = Merritt Island National Wildlife Refuge; TR = Lake Trafford; WO = Lake Woodruff; YK = Tom Yawkey Wildlife Center.

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	Yawk	ey ( <i>n</i> = 10)		Kiawah I	sland $(n = 1)$	10)	Bear Isl	and $(n = 1)$	(0
	Range	Median	n >RL <sup>a</sup>	Range	Median	n >RL <sup>a</sup>	Range	Median	n >RL <sup>a</sup>
PEOA	$< 0.008^{b-0.193}$	0.050	4	0.028-0.298	0.126	10	$< 0.008^{b-0.193}$	<0.100	3
FNA	0.272-1.32	0.620	10	0.446 - 1.38	1.19	10	0.155 - 1.14	0.472	10
FDA	2.27–15.1	5.88	10	3.72-13.6	6.26	10	0.998 - 3.21	1.57	10
FUnA	1.89 - 18.4	6.25	10	1.87–7.53	3.93	10	1.05-5.02	2.32	10
FDoA	0.362-3.45	1.01	10	1.32–7.27	3.05	10	0.231 - 1.88	0.559	10
FTriA	$< 0.070^{b-1.85}$	0.646	8	0.420 - 2.60	0.919	10	$< 0.070^{b-1.83}$	0.674	6
FTA	<0.082 <sup>b</sup> -0.774	0.241	L	0.198 - 1.38	0.476	10	$< 0.081^{b} - 0.733$	0.095	L
FHxS	0.099-0.566	0.353	10	0.313 - 1.86	0.620	10	0.077 - 0.824	0.304	10
FOS	4.50–57.0	20.2	10	38.4–98.2	55.8	10	10.0-44.9	19.5	10
	Lochloos	a Lake ( <i>n</i> =	10)	Lake Woo	odruff $(n = 1)$	(0)	Lake Apo	opka ( <i>n</i> = 1	(0
	Range	Median	n>RL <sup>a</sup>	Range	Median	n>RL <sup>a</sup>	Range	Median	n>RL <sup>a</sup>
FOA	$< 0.008^{b} - 0.132$	0.071	6	$< 0.097^{b} - 0.184$	0.062	5	$< 0.096^{b-0.152}$	0.126	7
FNA	0.328-1.19	0.676	10	0.282 - 1.34	0.578	10	0.251 - 1.40	0.648	10
FDA	0.238 - 1.00	0.615	10	0.350 - 5.06	2.01	10	0.169 - 2.44	1.12	10
FUnA	0.580 - 1.56	1.03	10	0.633 - 3.33	1.43	10	0.614–3.39	1.65	10
FDoA	0.105 - 0.309	0.182	10	$< 0.166^{b} - 0.810$	0.317	6	$< 0.157 b_{-0.831}$	0.315	6
FTriA	0.181 - 0.580	0.309	10	$< 0.070^{b} - 0.854$	0.259	8	0.189 - 1.00	0.450	10
FTA	$<0.008^{b_{-0.060}}$	0.018	L	$< 0.008^{b} - 0.146$	0.029	4	$< 0.080^{b} - 0.194$	0.049	L
FHxS	0.069-0.201	0.093	10	0.130 - 0.623	0.445	10	0.166 - 0.449	0.332	10
FOS	2.19–6.16	4.21	10	5.89-41.2	16.0	10	1.98–15.8	11.4	10
	Merrit Is	sland $(n = 1)$	2)	St. Johns	River $(n = 1)$	(0)	Lake Kissi	mmee ( <i>n</i> =	10)
	Range	Median	<i>n</i> >RL <sup><i>a</i></sup>	>Range	Median	<i>n</i> >RL <sup>a</sup>	Range	Median	n>RL <sup>a</sup>

	Yawko	ey ( <i>n</i> = 10)		Kiawah I	sland $(n = 1)$	(0)	Bear Isk	and $(n = 10)$	()
	Range	Median	$n > \operatorname{RL}^{d}$	Range	Median	n >RL <sup>a</sup>	Range	Median	n >RL <sup>a</sup>
PFOA	$< 0.096^{b} - 0.412$	0.155	7	0.010-0.160	0.080	10	$< 0.008^{b-0.142}$	0.104	6
PFNA	0.298 - 1.10	0.611	15	0.250 - 1.04	0.471	10	0.275 - 1.18	0.642	10
PFDA	0.395 - 3.50	1.02	15	0.492 - 1.72	1.17	10	0.417 - 3.15	1.26	10
PFUnA	0.844–5.45	1.82	15	0.655 - 2.20	1.28	10	0.314-2.47	1.03	10
PFDoA	$< 0.543^{b-1.07}$	0.418	14	0.156-0.591	0.362	10	$< 0.009^{b-0.382}$	0.147	6
PFTriA	$< 0.026^{b-1.42}$	0.654	14	0.173-0.739	0.267	10	0.122-0.677	0.251	10
PFTA	$< 0.080^{b} - 0.257$	0.076	9	$< 0.008^{b-0.131}$	0.022	8	$< 0.009^{b-0.104}$	0.025	6
PFHxS	0.684–23.3	3.83	15	0.100 - 0.308	0.166	10	0.338 - 1.50	0.505	10
PFOS	38.6-452	99.5	15	3.41-10.2	7.13	10	6.51–25.1	12.2	10
	Lake Tra	fford $(n = 1)$	(0	WCA-:	2A ( <i>n</i> = 10)		WCA-ŝ	3A(n=10)	
	Range	Median	$n > RL^{a}$	Range	Median	$n > RL^{a}$	Range	Median	$n > RL^{a}$
PFOA	0.021-0.117	0.091	10	<0.008 <sup>b</sup> -0.077	0.036	2	$< 0.008^{b-0.042}$	0.033	9
PFNA	0.239-0.936	0.484	10	0.189 - 0.382	0.234	10	0.172 - 0.388	0.301	10
PFDA	0.275-2.05	0.885	10	0.641 - 2.26	0.900	10	0.406-1.46	0.912	10
PFUnA	0.463–2.19	0.953	10	0.958 - 3.15	1.43	10	0.719–2.48	1.45	10
PFDoA	0.073-0.737	0.210	10	0.277 - 0.949	0.392	10	0.172 - 0.631	0.371	10
PFTriA	0.111 - 0.528	0.304	10	0.232-0.702	0.370	10	0.162 - 0.594	0.280	10
PFTA	$<0.008^{b-0.096}$	0.039	6	0.031 - 0.188	0.109	10	0.011 - 0.148	0.042	10
PFHxS	0.071 - 0.320	0.119	10	0.080 - 0.172	0.112	10	0.057-0.303	0.105	10
PFOS	4.21-14.3	7.82	10	1.36-6.23	2.65	10	1.57-4.71	3.81	10
a a									

 $a^{a}$  >RL indicates the number of samples above the reporting limit (RL).

b Values were calculated with one-half the reporting limit substituted for nondetects, as described in Materials and Methods; for values shown as "<," however, a specified number describe the actual reporting limit. PFOA = perfluorooctanoic acid; PFNA = perfluorononanoic acid; PFDA = perfluorodecanoic acid; PFUA = perfluorondecanoic acid; PFDA = perfluorodecanoic acid; PFDA = perfluorononanoic acid; acid; PFTriA = perfluorotridecanoic acid; PFHxS = perfluorohexanesulfonic acid; PFOS = perfluorooctane sulfonate; WCA = Water Conservation Area.

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## Table 2

Correlation coefficients between plasma perfluoroalkyl acid concentrations and snout-vent length for American alligators (Alligator mississippiensis) sampled in Florida and South Carolina,  $USA^a$ 

Snout-vent length	PFOA	PFNA	PFDA	PFUnA	PFD <sub>0</sub> A	PFTA	PFTriA	PFHxS	PFOS
Male ( $n = 65$ )	0.072	$0.252bc^*$	0.206	$0.355^{b^{**}}$	$0.279b^{*}$	0.209	$0.273^{b^{*}}$	$0.273b^{*}$	$0.331 b.c^{**}$
Female $(n = 60)$	0.133	$0.261 b.c^*$	$0.443 \ b.c^{**}$	$0.489^{b^{**}}$	$0.469^{b^{**}}$	$0.468 b. c^{**}$	$0.570^{b^{**}}$	$0.412^{b^{**}}$	$0.551 b.c^{**}$

<sup>a</sup>All significant results were positively correlated. Values were calculated using log normal concentrations.

bIndicates significant correlation coefficients.

<sup>c</sup>Indicates a correlation coefficient determined using the Pearson correlation. All other correlation coefficients were determined using the Spearman correlation.

\* Correlation is significant at the 0.05 level (2-tailed).

\*\* Correlation is significant at the 0.01 level (2-tailed). PFOA = perfluorooctanoic acid; PFNA = perfluorononanoic acid; PFDA = perfluorodecanoic acid; PFUA = perfluorondecanoic acid; PFDA = perfluorodecanoic acid; PFDA = perfluorononanoic acid; acid; PFTriA = perfluorotridecanoic acid; PFHxS = perfluorohexanesulfonic acid; PFOS = perfluorooctane sulfonate. NIST Author Manuscript

# Table 3

Correlation coefficients between concentrations of various perfluoroalkyl acids in plasma of American alligators (Alligator mississippiensis) sampled in Florida and South Carolina  $(n_{\text{male}} = 65, n_{\text{female}} = 60)^a$ 

	PFOA	PFNA	PFDA	PFUnA	PFD0A	PFTriA	PFTA	PFHxS	PFOS
Aale									
PFOA		$0.615^{b^{**}}$	0.226	0.092	0.036	0.152	0.181	0.260	$0.386^{b^{**}}$
PFNA			$0.550^{b^{**}}$	$0.322b^{*}$	0.144	$0.313^{b^{*}}$	$0.273^{b^{**}}$	$0.339^{b^{**}}$	$0.541^{b^{**}}$
PFDA				$0.840^{b^{**}}$	$0.743^{b^{**}}$	$0.439^{b^{**}}$	$0.654^{b^{**}}$	$0.307^{b^{*}}$	$0.550^{b^{**}}$
PFUnA					$0.920^{b^{**}}$	$0.783^{b^{**}}$	$0.826^{b^{**}}$	$0.445^{b^{**}}$	$0.654^{b^{**}}$
PFDoA						$0.751^{b^{**}}$	$0.846^{b^{**}}$	$0.316^{b*}$	$0.528^{b^{**}}$
PFTriA							$0.770^{b^{**}}$	$0.395^{b^{**}}$	$0.489^{b^{**}}$
PFTA								0.238	$0.399^{b^{**}}$
PFHxS									$0.827^{b^{**}}$
PFOS									
temale									
PFOA		$0.648^{b^{**}}$	$0.332^{b^{*}}$	0.186	0.098	0.064	-0.003	$0.441^{b^{**}}$	$0.440^{b^{**}}$
PFNA			$0.585^{b^{**}}$	$0.444^{b^{**}}$	$0.365^{b^{**}}$	$0.339^{b^{**}}$	0.196	$0.387^{b^{**}}$	$0.538^{b^{**}}$
PFDA				$0.890^{b^{**}}$	$0.827^{b^{**}}$	$0.529^{b^{**}}$	$0.560^{b^{**}}$	$0.337^{b^{**}}$	$0.595^{b^{**}}$
PFUnA					$0.938^{b^{**}}$	$0.684^{b^{**}}$	$0.578^{b^{**}}$	$0.331^{b*}$	$0.691^{b^{**}}$
PFDoA						$0.763^{b^{**}}$	$0.708^{b^{**}}$	0.226	$0.635^{b^{**}}$
PFTriA							$0.713^{b^{**}}$	0.190	$0.598^{b^{**}}$
PFTA								0.130	$0.454^{b^{**}}$
PFHxS									$0.654^{b^{**}}$
PFOS									

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b Indicates significant correlation coefficients.

\* Correlation is significant at the 0.05 level (2-tailed).

\*\* Correlation is significant at the 0.01 level (2-tailed).

PFOA = perfluorooctanoic acid; PFNA = perfluorononanoic acid; PFDA = perfluorodecanoic acid; PFUnA = perfluoroundecanoic acid; PFDoA = perfluorodoecanoic acid; PFTriA = perfluorotridecanoic acid; PFTA = perfluorotetradecanoic acid; PFTA = perfluo