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Development of Small Molecules with a Non-Canonical Binding Mode to HIV-1 Trans Activation Response (TAR) RNA

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Abstract

Small molecules that bind to RNA potently and specifically are relatively rare. The study of molecules that bind to the HIV-1 transactivation response (TAR) hairpin, a cis-acting HIV genomic element, has long been an important model system for the chemistry of targeting RNA. Here we report the synthesis, biochemical and structural evaluation of a series of molecules that bind to HIV-1 TAR RNA. A promising analog, **15**, retained the TAR binding affinity of the initial hit and displaced a Tat-derived peptide with an IC_{50} of 40 $>M$. NMR characterization of a soluble analog, **2**, revealed a non-canonical binding mode for this class of compounds. Finally, evaluation of **2** and **15** by Selective 2′-hydroxyl acylation analyzed by primer extension (SHAPE) indicates specificity in binding to TAR within the context of an *in vitro*-synthesized 365-nt HIV-1 $5'$ untranslated region (UTR). Thus, these compounds exhibit a novel and specific mode of interaction with TAR, providing important implications for RNA ligand design.

Graphical Abstract

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Molecular formula strings and associated biochemical data, procedure for the synthesis of **20**–**26** and biochemical data, liganddetected full 1H NMR spectra of **2** and **14**, full TOCSY spectra of free TAR and in presence of different concentration of **2**, 1H NMR and ¹³C NMR spectra for all compounds.

Notes

The authors declare no competing financial interest.

Supporting Information

Keywords

HIV-1 transactivation response RNA (TAR); RNA-binding small molecules; thienopyridine compounds; NMR; SHAPE

Introduction

RNA-binding small molecules are an important and highly challenging area of therapeutic research. The discovery of non-coding RNAs and their widespread role in human disease strongly suggests that opportunities to redress disease conditions by targeting RNA should be plentiful.¹ Structural and analytical techniques over the last few decades have provided information about RNA structure and function, yet examples of small drug-like molecules that interact with RNA are still rare relative to proteins. The polyanionic nature of RNA, coupled with its highly flexible and dynamic structure and repetitive nature of its surface, present challenges for inhibitor design. Many known RNA-binding small molecules are polycationic, binding tightly (with K_d s in the mid nM range) but with limited specificity. Furthermore, they tend to be larger than desirable for typical drugs and do not satisfy the characteristic physicochemical properties of modern successful drug candidates.

A well-characterized system to study small molecule RNA antagonists is transactivation response (TAR) element from HIV type 1 (HIV-1), a 59-nucleotide stem-loop RNA located in the 5′-untranslated region (UTR) of the viral genome. TAR plays a pivotal role in HIV replication via its interaction with Tat, a virus-encoded protein that recruits the host transcriptional machinery to facilitate efficient transcription of the viral $DNA.^{2–5}$ While the TAR structure is evolutionary conserved, sequence heterogeneity exists among various virus isolates.⁶ However, in the case of HIV-1 subtypes, base pair co-variations, e.g., A-U conversion into G-C has been shown to maintain TAR structural integrity.⁷ In addition to being a transcriptional activator, TAR has also been shown to play a role in initiation of reverse transcription.^{8–10} More recently, several groups have demonstrated that TAR also functions as a miRNA that downregulates cellular apoptotic genes and leads to extended cellular proliferation and continued virus production.^{11–13} In sum, The central role of TAR in HIV-1 biology underscores its potential as a therapeutic target.¹⁴ Thus, inhibition of the Tat-TAR interaction has been pursued as a potential anti-HIV therapeutic strategy. While numerous classes of small molecules have been reported^{15–17}, the lack of anti-HIV drugs targeting TAR in the clinic highlights the challenges in developing small molecule RNA binders. TAR RNA remains both an important anti-viral target and a model system to develop RNA-targeting molecules.

Previously, we utilized a small molecule microarray (SMM) screening platform to rapidly screen immobilized small molecules against nucleic acid targets.18 In one example, we reported the discovery of **1**, a thienopyridine compound that interacts with the TAR hairpin.¹⁹ However, at that time little was known about the mode of interaction between the compound and RNA. Here, we have developed a chemical synthesis of **1** to generate a focused library of 25 compounds derived from the initial hit structure. Biochemical and biophysical analyses of this series of compounds, including affinity measurements, NMR

analysis, and SHAPE, provided information about structure-activity relationships as well as insights into the mode of binding of this class of compounds to HIV-1 TAR.

Results and Discussion

Chemistry

To prepare a focused compound library, we developed a modular and convergent synthetic route to **1** and its analogs. For preparation of **1**, 2-chloro-6-methylnicotinonitrile (**24**) was treated with thiourea in refluxing n-butanol to provide 2-mercapto-6-methylnicotinonitrile (25) in essentially quantitative yield (Scheme 1).²⁰ In parallel, reaction of 3-(trifluoromethyl)aniline (**26**) with bromoacetyl bromide in the presence of catalytic amounts of 4-N,N-dimethylamino-pyridine (DMAP) provided 2-bromo-N-(3- (trifluoromethyl)phenyl)acetamide (27) in high yield (Scheme 1).²¹ Reaction of mercaptopyridine **25** with bromoacetamide **27** in refluxing ethanol in the presence of excess sodium ethoxide promoted bromide displacement by the thiol group of the mercaptopyridine intermediate followed by cyclization to afford the thienopyridine ring structure, providing **1** in 58% yield (Scheme 1).²²

We next used this route to design and synthesize several analogs of **1** to evaluate how structural changes would affect affinity for TAR. **2**–**4**, containing a second pyridine ring, were prepared by reaction of the associated aminopyridines (**28g-i**) with bromoacetyl bromide to provide the bromoacetamide as an HBr salt (**29g-i**) (Scheme 2A).23 Treatment of bromoacetamides **29g-i** with 2-mercapto-6-methylnicotinonitrile (**25**) in refluxing ethanol in the presence of excess base yielded the assembled thienopyridine ring system (Scheme 2A). **5–9** were prepared via a similar synthetic route to **1** (Scheme 2B). In the case of **10**, with a bulkier isopropyl group at the 6-position of the pyridine core, reaction of 2-chloro-6 isopropylnicotinonitrile (**30c**) with thiourea failed to yield the desired mercaptopyridine. Therefore, Lawesson's reagent was used to synthesize the desired mercaptopyridine (**31c**) in modest yield.24 The reaction of mercaptopyridine **31c** with the desired bromoacetamide (**27**) proceeded as described for **1** to provide **10** (Scheme 2B). Preparation of **11**, lacking an amine, was achieved by a modified Sandmeyer reaction, employing NaNO₂/H₂SO₄/Cu₂O with hypophosphorous acid (Scheme 3).²⁵ Additionally, reaction of 1 with triphosgene at room temperature provided cyclized **12** (Scheme 3).26 Analogs containing a sulfonamide linker were synthesized by the reaction of desired anilines (**26** and **34**) and chloromethanesulfonyl chloride in the presence of triethylamine followed by a basic workup27 to provide the sulfonamide intermediate (**35**–**36**) (Scheme 3). Sulfonamides **35**–**36** were treated with the desired mercaptopyridine (**25**) in the presence of excess base in N,Ndimethylformaide (DMF) at 130 °C to afford **13** and **14** (Scheme 3).²⁷

We also prepared **15**–**19**, containing oxadiazolyl linkers between the thienopyridine core and the aryl group (Scheme 4). For **15** and **16**, which contain 1,3,4-oxadiazol-2,5-yl and 1,2,4 oxadiazol-3,5-yl rings (respectively), commercially available chloromethyl phenyl substituted oxadiazoles (**37I**-**m**) were treated with 2-mercapto-6-methylnicotinonitrile (**25**) in the presence of excess base in DMF at 130 °C, producing the thienopyridine scaffold in good yield. **17**–**23** were synthesized similar to **16** (Scheme 4 and Scheme S1). In the case of

24–**26**, the desired chloromethyloxadiazoles were conveniently prepared as reported previously28 and further treated with **25** to yield the desired products (Scheme S2). Collectively, the convergent route reported here furnished a library of 25 analogs of the initial hit structure.

Biochemical evaluation of compound library

a. Affinity measurement—To evaluate the TAR binding properties of analogs described above, we utilized a series of biochemical assays that provide information on both compound affinity for TAR and its ability to displace a Tat-derived peptide from TAR, revealing a clear picture of structure-activity relationship. To probe direct binding of compounds to TAR, we utilized two different 2-aminopurine (2-AP) assays. 2-AP is an environmentally-sensitive fluorophore that reports on the local environment of the base.²⁹ Conformational changes in the RNA upon binding of a ligand result in changes in fluorescence intensity, which can be used to derive accurate affinities. Two distinct constructs were used where either U25 or U23 (both of which are located in the bulge region of TAR hairpin) was replaced with a 2-AP fluorophore (Figure 1A). U25 is a flexible residue³⁰ and its replacement by 2AP has been previously utilized in probing TAR-ligand interactions29. To evaluate larger numbers of compounds, we miniaturized the 2-AP assay to be compatible with plate reader-based analysis. This format required a higher concentration of RNA, but enabled a significantly reduced assay volume of 100 >L per sample in a 96 well plate. Once optimized, this assay generated reproducible apparent binding constants (K_{damp}) , albeit with slightly higher values than obtained under previous assay conditions.¹⁹ Values obtained under these conditions were both internally consistent and comparable to recently published K_d s of several TAR-binding small molecules.³¹

 K_{dapp} obtained from 2-AP assays with both constructs were in good agreement (Table 1). Synthetic 1 showed an apparent $K_{\text{dapp 25-2AP}}$ of 47.7 \pm 11.6 >M. Next, the trifluoromethylphenyl group was replaced by different pyridine isomers (**2**–**4**) which dramatically improved compound solubility, making them usable at high concentrations in NMR experiments (see below). Interestingly, analog **2,** containing 4-pyridyl substitution, exhibited a similar binding affinity ($K_{dapp 25-2AP} = 40.0 \pm 0.7$ >M) to 1 (Figure 1B). Analogs **3**, containing 3-pyridyl (K_{dapp 25-2AP} = 68.0 \pm 26.0 >M) and **4** with 2-pyridyl (K_{dapp 25-2AP} = 50.3 ± 10.3 >M) substituents showed ~2-fold weaker binding affinity compared to **2**. Changing the position of the *meta*-trifluoromethyl group in 1 to *para* (7, $K_{\text{dapp 25-2AP}} = 75.3$) \pm 9.2 >M) or *ortho* (8, K_{dapp 25-2AP} = 90.1 \pm 4.7 >M) decreased binding affinity approximately 2-fold. Additionally, extending the meta-trifluoromethyl phenyl by a methylene linker (9, $K_{\text{dapp 25-2AP}} = 66.4 \pm 0.9$ >M) maintained a similar binding affinity to **1**.

We further investigated the role of substitution on the pyridyl ring of thienopyridine scaffold. Moving the methyl group to the 5-position $(6, K_{\text{dapp 25-2AP}} = 48.5 \pm 5.4$ >M) maintained a similar binding affinity as 1 while the 4-methyl substituent $(5, K_{\text{dapp 25-2AP}} = 66.5 \pm 4.5 > M)$ showed ~2-fold weaker binding affinity. Replacing the methyl substituent with a bulkier isopropyl group in 10 ($K_{\text{damp 25-2AP}} = 43.3 \pm 13.3$) maintained similar affinity compared to **1**. Next, Analog **11**, a desamino thienopyridine, showed inferior binding affinity,

highlighting the importance of the amino group. Additionally, replacing amide for a sulfonamide in analog **13** and **14** or a cyclized analog, **12**, completely abolished binding.

Next, we evaluated the oxadiazole series. Interestingly, isosteric replacement of the amide linker in **1** with 1,3,4-oxadiazol-2,5-yl in **15** (K_{dapp 25-2AP} = 49.1 \pm 7.8 >M) (Figure 1B) as well as 1,2,4-oxadiazol-3,5-yl in **16** ($K_{\text{dapp 25-2AP}} = 47.7 \pm 17.8$ >M) retained binding affinity as observed for **1**. Analogs **17–19**, containing the 1,3,4-oxadiazol-2,5-yl but with different substitution pattern, retained the binding affinity of **15**. However, evaluation of 1,3,4-oxadiazol-2,5-yl analogs **20**–**26** either decreased solubility or abolished binding affinity (Table S1).

b. Competitive binding assay measurements—In addition to the 2-AP direct binding assays, we utilized an orthogonal FRET-based assay to evaluate disruption of the Tat-TAR interaction.32 In this assay, a 16-mer peptide derived from Tat protein is labeled with fluorescein at its N-terminus and TAMRA at its C-terminus. In solution, the two fluorophores in the free peptide are in close proximity and quench each other. Upon binding to TAR, the peptide adopts an extended conformation, inducing FRET. In this assay, a compound capable of disrupting the Tat-TAR interaction will cause a loss of FRET signal. All synthesized analogs were evaluated in FRET assay. While many, including **2,** did not display activity in this assay up to 400 >M, it is notable that oxadiazole analogs **15** (Figure 1C) and **16** (IC₅₀ = 40.1 \pm 5.6 $>$ M and IC₅₀ = 45.1 \pm 25.8 $>$ M respectively) displaced the Tat peptide with IC_{50} values similar to their apparent $K_{d}s$ (Table 1). Additionally, while removal of meta-trifluoromethyl group on the phenyl ring in **17** or its replacement with meta-chloro in **18** abolished FRET signal, the absence of 6-methyl substituent in **19** maintained FRET activity similar to **15** (Table 1) suggesting the importance of phenyl substituent on the ability of the oxadiazole analogs to displace the Tat peptide.

Taken together, our thienopyridine analogs displayed mid micromolar binding to the TAR hairpin under the assay conditions. While the majority of the structural changes in this study resulted in subtle effects, isosteric replacement of the amide with oxadiazoles provided analogs which not only maintained binding affinity to TAR but displaced the Tat peptide in the FRET assay at concentrations in the mid micromolar range. It is also noteworthy that the compounds in this series with similar binding affinity show differing behavior in the FRETbased assay. However, these observations are consistent with a report in which peptides with similar TAR binding affinity also exhibited very different IC_{50} values in displacing a Tat peptide.³³ This seemingly contradictory observation suggests that the determinants of binding may differ from those required for disrupting an associated RNA/peptide interaction, at least for some ligands.

NMR studies

To better understand the mode of interaction of the compounds with TAR, we investigated the small molecule-RNA complex by NMR. Many compounds in this series exhibited poor aqueous solubility, which limited high-resolution structure analysis. Compound **2** displayed increased solubility relative to **1** and could be studied at concentrations amenable for liganddetected NMR experiments and some limited RNA-detected experiments. **2** was first

evaluated using line broadening methods (Figure 2A and Figure S1A–C).³⁴ In this experiment, its proton spectra was recorded in the presence and absence of 10μM TAR. Upon binding, the ligand resonances broaden in response to the increase in rotational correlation time while bound to target RNA.34 In the case of **2**, the two doublets corresponding to protons on the pyridine ring of the thienopyridine scaffold (protons a and b in Figure 2A) are completely broadened in presence of TAR suggesting the pyridine ring directly interacts with the hairpin while other proton peaks do not change (Figure S1A). In contrast, ligand-detected NMR of a non-binding analog **14** shows no line broadening in neither the aromatic (Figure 2B) nor aliphatic region (Figure S1B). The line-broadening pattern in the 1D-1H NMR experiments is consistent with the 2-AP fluorescent-based affinity measurements indicating mid micromolar binding.

Ligand-detected NMR experiments require lower ligand concentration to measure binding, which is useful for low solubility molecules. However, these methods are also susceptible to nonspecific aggregation effects, which generate false positive results. Therefore, to identify and verify binding of **2**, we evaluated ligand-induced chemical shift perturbations in total correlation spectroscopy (TOCSY) experiments. Here, free TAR hairpin was evaluated upon the addition of **2** (Figure 3A–B and Figure S1D–E). Specific changes in the TOCSY spectrum suggest single site binding with the largest changes occurring at bases C39, U40 and C41. These residues are located directly below the -UCU-bulge residues U23, C24 and U25. Resonances within residues U23 and C24 also show significant chemical shift changes. However, unlike many TAR ligands, U25 does not show significant changes in chemical shift, nor do any of the remaining pyrimidines, indicating a new binding mode compared to previously reported TAR-binding ligands such as argininamide and acridine, which disrupt stacking of A22 and U23 and bind to the major groove of TAR similar to Tat-derived peptides.^{35–37} On the other hand, the solution NMR structure of Neomycin B bound to TAR indicates that this aminoglycoside binds in the minor groove instead.38 However TAR residues G21, U42 and G26 that are shifted in presence of Neomycin B are not the same residues perturbed by **2**. Additionally, Hoescht 33258 has been reported to bind to the TAR hairpin, though it binds to the major groove/upper bulge region primarily at and around bases G36-U40.³⁹

The concentration-dependent chemical shift perturbations of TAR residues induced by **2** suggest that complex formation occurs in the fast exchange regime of the NMR time scale. However, residues U40 and C41 also experience broadening in addition to chemical shift changes specific to these residues. These results are entirely consistent with binding in the mid micromolar range, in agreement with binding affinity determined by biochemical assays. The perturbed residues are concentrated in the bulge and lower bulge region of the TAR hairpin, suggesting that **2** may be a lower bulge binder.

In sum, the NMR studies indicate that **2** binds to TAR and is likely a lower bulge/minor groove binder. This binding mode does not perturb many residues in the Tat binding site itself, which provides insight into the varying ability of compounds in this series to displace the Tat peptide. Attempts to derive a full structure of TAR bound to **2** by intermolecular NOESY experiments were hampered by the limited solubility of the ligand under the required NMR conditions.

SHAPE

Selective 2′-hydroxyl acylation analyzed by primer extension (SHAPE) can be used to establish RNA secondary structure and identify sites of interaction of RNA with other molecules by examining backbone flexibility.40 Electrophilic reagents, e.g. 1M7 (1 methyl-7-nitroisatoic anhydride) modify the 2′-OH of less-structured RNA nucleotides, which causes reverse transcription (RT) to halt one nucleotide before the modification. Modification frequencies are then used to estimate a 'reactivity profile' that quantifies the propensity of each nucleotide to be modified by the chemical probe. High reactivity reflects nucleotides that are flexible and unstructured, while low reactivity provides structural constraining, such as formation of base pairing, stacking or RNA–ligand interactions.

We applied SHAPE to probe RNA conformational changes that would be diagnostic of compound binding in the context of an in vitro-synthesized 365-nt HIV-1 5′ UTR RNA. Changes in 1M7 reactivity at each nucleotide of the HIV-1 5′ UTR RNA in the presence of **2** and **15** are presented in Figure 4 (relative to the analogous experiment performed in its absence). The *decrease* in nucleotide reactivity upon analog addition is indicated in gradients of blue, while increase in reactivity is represented in gradients of red; unchanged nucleotide reactivity is in gray. While the global topology of the 5′ UTR was unaffected upon incubation with **2** and **15**, substantial changes in chemical reactivity were observed mostly within TAR (Figure 4A–B). In particular, addition of **2** caused a decrease in 1M7 reactivity for residue U25 within the TAR bulge and U31, G33 and A35 within the apical loop (Figure 4A). Reactivity changes observed in the apical loop residues for **2** could reflect ligand binding to the lower bulge region (as discussed above) causing disruption of a direct interaction between the apical loop and the bulge as recently demonstrated by Krawczyk et al.41 Addition of **15** to RNA showed a similar 1M7 reactivity reduction throughout TAR compared to **2**, and in particular with residues U13–14, C19 and G32 (Figure 4B). Taken together, reduced 1M7 reactivity suggests that analog binding causes structural perturbations to TAR due to a direct or, for loop residues, indirect interaction with **2** and **15**. Furthermore, these results are consistent with our observations for **1**, which we showed also bound selectively to the TAR hairpin.¹⁹ Although the global topology of the $5'$ UTR was unchanged upon compound binding, subtle changes in chemical reactivity were observed downstream of the TAR hairpin. These "off target" effects may reflect the influence of compound binding on long-range tertiary interactions^{42, 43} and/or indirect structural rearrangements in the 5^{\degree} UTR.⁴⁴ In conclusion, the SHAPE data suggest specific binding of both **2** and **15** to the TAR region of HIV-1 UTR in the presence of other structured motifs.

Conclusions

We have characterized the interaction of a series of thienopyridine compounds to the HIV-1 TAR RNA hairpin. The initial structure, **1**, was discovered in a screen of immobilized small molecules, which identified several structures binding to TAR specifically over other RNA stem-loops. Compound **1** and a focused library of 25 analogs were synthesized using a concise and modular synthetic route. These compounds were then evaluated in several biochemical assays for binding to the TAR hairpin as well as to disrupt the Tat-TAR interaction. A clear structure-activity relationship was observed in the binding assays. A number of analogs, but not all those that bound to TAR, were able to displace a Tat-derived

peptide from the RNA. Replacing the amide linker with either of two oxadiazolyl groups yielded analogs with similar affinity to the original hit that were also able to disrupt the Tat/TAR interaction. The most potent analog, **15**, showed a similar binding affinity to **1** and exhibited an IC_{50} of 40 >M for displacement of Tat-derived peptide. SHAPE studies showed that both **2** and **15** retained global specificity for binding to the TAR motif within the 365-nt HIV-1 5′ UTR RNA. NMR studies further confirmed that **2** binds to TAR and causes chemical shift perturbations exclusively within and below the bulge region. These results suggest that **2** binds to the lower bulge region, possibly within the minor groove. The perturbed residues differ from those interacting with a Tat-derived peptide, neomycin, Hoechst 33258 or argininamide, which could provide a basis for the varying ability of compounds in this class to displace the peptide. Future efforts will focus on improving solubility of analogs for cell-based studies and further development of a bifunctional molecule to tether our lower bulge binder with an apical loop binder such as 5-(N,N) dimethyl amiloride 31 to achieve both increased specificity and enhanced affinity through synergistic binding.

Experimental Section

Biochemical assays

Nucleic acids were purchased from Dharmacon with the following sequences: TAR RNA: 5′-GGCAGAUCUGAGCCUGGGAGCUCUCUGCC-3′ and 25-2AP TAR: 5′- GGCAGAUC(2AP)GAGCCUGGGAGCUCUCUGCC-3 and 23-2AP TAR: 5′- GGCAGA(2AP)CUGAGCCUGGGAGCUCUCUGCC-3. TAR and 2AP TAR constructs were reconstituted in 1x PBS pH 7.4 to a concentration of \sim 100 >M and annealed by heating to 95 °C for 3 min. The constructs were allowed to cool to room temperature for at least 30 min. Tat FRET peptide, a 16-mer peptide labeled at its N-terminus by FAM and C-terminus by TAMRA, was purchased from AnaSpec (catalog #: AS-64346-05). A fluorescence-based assay employing TAR constructs labeled with 2-AP either at bulge residues U23 or U25 was used. 2-AP TAR constructs were used at 1 >M. 5 >L of the tested compound in DMSO and 95 $>$ L of 2-AP TAR in the assay buffer (1 \times PBS pH 7.4) were added to assay plate (Nunc black plates) and incubated at room temperature with rocking for 30 min. Fluorescence values were measured at an excitation wavelength of 310 ± 9 nm and an emission wavelength of 365 ± 9 nm using a microplate reader and values were normalized to free 2-AP TAR. Most analogs were not fluorescent at this wavelength, and in the case of fluorescent compounds, the inherent fluorescence was subtracted from that of the complex (TAR and compound) and then normalized to free 2-AP TAR. Apparent K_d values were determined by non-linear regression by fitting the measured fluorescence (GraphPad Prism 6.0 software). FRET studies were performed as previously reported 32 with modification. Unlabeled TAR was used at 100 nM and FAM-Tat-TAMRA at 12 nM. 5 μL of the tested compound (in DMSO) and 95 μL of unlabeled RNA/peptide complex in assay buffer (50 mM Tris pH 7.4; 20 mM KCl; 0.01% Triton X-100) were added to Nunc black assay plates and incubated at room temperature for 30 min with rocking. It should be noted that each plate contained four wells for negative control (equivalent to 0% inhibition and highest fluorescent/FRET) which included 100 nM unlabeled TAR and 12 nM FAM-Tat-TAMRA in presence of 5% DMSO. Additionally, three wells were assigned for the positive control

(equivalent to 100% inhibition and lowest fluorescence/FRET) which included 12 nM of FAM-Tat-TAMRA in presence of 5% DMSO. Fluorescence values were measured at an excitation wavelength of 485 ± 20 nm and an emission wavelength at 590 ± 35 nm using a microplate reader. Most analogs were not fluorescent at this wavelength; however for analogs **2**, **3**, **4**, **16**, **23** and **25**, the fluorescence of complex (unlabeled TAR/labeled Tat) in presence of compound exceeded the maximum fluorescence as indicated by the negative control wells. Therefore, the inherent fluorescence of compound was subtracted from that of the complex plus compounds. IC_{50} values were determined by non-linear regression fitting of the competition curves.

NMR

HIV-1 TAR RNA hairpin was transcribed and purified as described previously.⁴⁵⁻⁴⁷ Ligand detected NMR experiments were collected on a Bruker DRX 500MHz equipped with TCI cryoprobe or Bruker Avance III 600 MHz spectrometer equipped with TCI cryoprobe. NMR screening buffer consisted of 20 mM d^{19} -Bis-Tris at pH6.5, 10 mM sodium chloride, and 11.1 >M DSA as chemical shift reference (integrates to 100 >M)⁴⁸, prepared in 99.99% D₂O or 95%H₂O/5%D₂O. Each ligand was first dissolved in DMSO to a 10 mM stock concentration in d_6 -DMSO. Free ligand screening samples were prepared at 150 $>$ M in NMR screening buffer with final DMSO concentration of 15%. We used this concentration as the safe upper limit beyond which the RNA structure might be affected. For RNA detected experiments, the lyophilized RNA pellets were dissolved in DMSO-free NMR screening buffer and heated for 4 min at 90°C, then snapped cooled for 5 min at −20 °C. 1D-1H experiments for both ligand and RNA were collected using the Bruker "zgesgp" excitation sculpting pulse sequence to suppress background water signal. Data was collected with 64 scans and 16K data points. Data was processed using 16K zero points, Fourier transformed followed by manual phase and baseline correction. 2D-1H TOCSY experiments were collected using the Bruker "mlevesgpph" pulse sequence with excitation sculpting water suppression, data collected with 8 scans, $2K \times 512$ data points, a recycle delay of 1.2 s and spin-lock mixing time of 80ms. Data was processed using NMRPipe and visualized using SPARKY.

SHAPE

RNA preparation for SHAPE analysis—DNA template for *in vitro* transcription were generated by PCR amplification of the HIV-1 molecular clone pNL4-3 using the following primers: T7L 5'-TAATACGACTCACTATAGGTCTCTCTG -3' (containing T7 promoter, underlined) 369R 5′-GCTTAATACCGACGCTCTCGC -3′

All PCR experiments were performed using Invitrogen Platinum® Taq DNA polymerase High Fidelity. Transcripts were synthesized with the MEGAscript T7 transcription kit (Thermo Fisher Scientific) following the manufacturer's protocol, and RNAs were purified from denaturing 8 M urea/6% polyacrylamide gels, followed by electro-elution and ethanol precipitation. Purified RNAs were dissolved in sterile water and stored at −20 ºC.

Renaturation and 1M7 treatment of RNA—5 pmoles of RNA were heated to 90 °C for 3 min, and gradually cooled to 4 °C (0.1 °C/sec). The volume was adjusted to $10 > L$ in a

final buffer of 50 mM Tris-HCl (pH 8.0), 100 mM NaCl, 5 mM MgCl₂. Samples were then divided into 5 μl experimental (+) and control (−) aliquots (2.5 pmoles each) and incubated at 37 ºC for 15 min. Analog **2** and **15** were added to both (+) and (−) RNA solutions to a final concentration of 5 >M and 0.5 >M respectively, and incubated for 10 min at room temperature. Chemical modification was initiated by addition of 1 μL of 1M7 (in anhydrous DMSO) to the (+) RNA solution to a final concentration of 3.5 mM, or DMSO alone to the (−) RNA reaction, followed by incubation at 37 ºC for 5 min. Completed reactions were precipitated at −20 ºC with 10 ng/>L glycogen, 0.3M sodium acetate pH 5.2 and 2.5 volumes of cold ethanol. RNA was collected by centrifugation, and resuspended in 10 >L of water. A parallel experiment lacking analogs was performed as a control.

Primer extension—For detection of 2[']-O-adducts, 2.5 pmoles of modified and unmodified RNAs were mixed, respectively, with 5 pmoles of Cy5-labelled (for 1M7 modified samples) or 6 pmoles of Cy5.5-labelled (for unmodified samples) DNA primer complementary to their 3′ end. Samples were incubated at 85 ºC for 1 min, 60 ºC for 5 min, 35 ºC for 5 min and 50 ºC for 2 min to anneal primers to initiate reverse transcription. RNA was reverse transcribed at 50 °C for 50 min with 100U RT (Invitrogen Superscript III), $1^{\wedge}p$ RT buffer (Invitrogen), 5 mM DTT and 500 mM dNTPs (Promega). RNA was hydrolyzed with 200 mM NaOH for 5 min at 95 °C and samples were neutralized with an equivalent volume of HCl. Sequencing ladders were prepared using the Epicentre cycle sequencing kit according to the manufacturer's instructions and primers labeled with WELLRed D2 and LICOR IR-800 dyes. Modified and control samples were mixed with the sequencing ladders, precipitated as above, dried and resuspended in 30 >L deionized formamide. Primer extension products were analyzed on a Beckman CEQ8000 Genetic Analysis System programmed with the separation method described previously.⁴⁹

SHAPE data analysis—Electropherograms were processed using the program SHAPEfinder program, following the software developer's protocol and included the required pre-calibration for matrixing and mobility shift for each set of primers as described.40, 50 The area under each negative peak was subtracted from that of the corresponding positive peak. The resulting peak area difference at each nucleotide position was then divided by the average of the highest 8% of peak area differences, calculated after discounting any results greater than the 3rd quartile plus 1.5× the interquartile range. RNAstructure software version 5.7 was used to predict RNA secondary structure(s) on the basis of pseudo-free energy constraints derived from SHAPE reactivity values. All reactivity data used in 2D structure analysis was averaged from three independent experiments. Varna (ver. 3–9), the visualization applet for RNA secondary structure, was used to produce highquality images.⁵¹

Chemistry

Chemicals were purchased from commercial sources and used without further purification unless otherwise noted. Anhydrous solvents were prepared by passage over activated alumina. Thin-layer chromatography was conducted with E. Merck silica gel 60 F254 precoated plates (0.25 mm) and visualized by exposure to UV light (254 nm). Flash column chromatography was performed using normal phase or reverse phase on a CombiFlash® Rf

200i (Teledyne Isco Inc). 1H NMR spectra were recorded at 400 or 500 MHz, and are reported relative to deuterated solvent signals. Data for 1 H NMR spectra are reported as follows: chemical shift (δ ppm), multiplicity, coupling constant (Hz), and integration. ¹³C NMR spectra were recorded at 100 or 125 MHz. Data for ¹³C NMR spectra are reported in terms of chemical shift. All NMR spectra were standardized to the NMR solvent signal as reported previously.52 Analytical LC/MS was performed using a Shimadzu LCMS-2020 Single Quadrupole utilizing a Kinetex 2.6 μm C18 100 A $(2.1 \times 50 \text{ mm})$ column obtained from Phenomenex Inc. Runs employed a gradient of 0→90% ACN/0.1% aqueous formic acid over 4 minutes at a flow rate of 0.2 mL/min. Purities of final compounds were assessed by analytical reversephase HPLC using Agilent Proshell 120 EC-C18 (4.6 mm \times 50 mm; particle size 2.7 μm). Runs employed a gradient of 10% ACN/water (1 min), 10–90% ACN/ water (6 min), and 90% ACN/water (2 min) flow $= 1$ mL/min. High-resolution LC/MS analyses acquired on an Agilent 6520 Accurate-Mass Q-TOF LC/MS System, (Agilent Technologies, Inc., Santa Clara, CA) equipped with a dual electrospray source, operated in the positive-ion mode. Separation was performed on Zorbax 300SB-C18 Poroshell column $(2.1 \text{ mm} \times 150 \text{ mm})$; particle size 5 µm). All final compounds were purified to >95% purity unless otherwise stated.

Representative procedure for thienopyridine carboxamide preparation

3-Amino-6-methyl-N-(3-(trifluoromethyl)phenyl)thieno[2,3-b]pyridine-2 carboxamide (1)—Synthesized using reported procedure with modification.²² To a suspension of **25** (51 mg, 0.34 mmol) in anhydrous EtOH (2 mL) was added **27** (92 mg, 0.32 mmol) and sodium ethoxide (54 mg, 0.75 mmol). The mixture was heated to reflux for 1 h 15 min. The mixture was diluted with EtOAc (10 mL) and washed with H₂O (10 mL \times 2) and brine (10 mL). The organic layer was dried (Na₂SO₄), filtered, and concentrated in vacuo. Crude was purified by flash column chromatography (hexanes/EtOAc) on silica gel to give 1 (65 mg, 58%) as a light yellow solid. Purity 100% .¹H NMR (500 MHz, DMSO- d_6) δ 9.70 (s, 1H), 8.41 (d, J = 8.3 Hz, 1H), 8.23 (s, 1H), 7.99 (d, J = 8.3 Hz, 1H), 7.56 (t, J = 8.0 Hz, 1H), 7.47 (s, 2H), 7.41 (d, $J = 7.7$ Hz, 1H), 7.35 (d, $J = 8.3$ Hz, 1H), 2.61 (s, 3H). 13C NMR (125 MHz, DMSO-^d6) δ 164.23, 159.90, 158.54, 147.83, 140.00, 131.15, 129.58, 129.14 (q, $J = 31.5$ Hz), 125.31 (q, $J = 272.4$ Hz), 124.20, 123.60, 119.61, 119.41 (q, $J = 3.8$ Hz), 116.81 (q, $J = 4.0$ Hz), 94.75, 24.29. ESI HRMS: m/z 352.07332 (M+H)⁺.

Representative procedure for pyridinylthienopyridine carboxamide preparation

3-Amino-6-methyl-N-(pyridin-4-yl)thieno[2,3-b]pyridine-2-carboxamide (2)—

Synthesized using a similar procedure used to prepare **1** except using **29g**. Mixture was heated to reflux overnight. Upon cooling to room temperature and addition of H_2O , yellow solid precipitated. Solid was filtered and washed with cold EtOH. Crude was purified by flash column chromatography (hexanes/EtOAc) on silica gel to give **2** (18 mg, 22% over two steps) as a yellow solid. Purity 100%.¹H NMR (400 MHz, DMSO- d_6) δ 9.72 (s, 1H), 8.44 – 8.38 (m, 3H), 7.76 (d, $J = 6.3$ Hz, 2H), 7.53 (s, 2H), 7.35 (d, $J = 8.3$ Hz, 1H), 2.60 (s, 3H).13C NMR (100 MHz, DMSO-^d6) δ 164.49, 160.15, 158.63, 149.99, 148.33, 146.11, 131.24, 123.44, 119.65, 114.19, 94.63, 24.29. ESI HRMS: 285.08048 (M+H)+.

6-Methyl-2-(pyridin-3-ylcarbamoyl)thieno[2,3-b]pyridin-3-aminium 2,2,2 trifluoroacetate (3)—Synthesized using a similar procedure used to prepare **1** except using **29h**. Mixture was heated to reflux overnight. Upon cooling to room temperature and addition of H₂O, yellow solid precipitated. Solid was filtered and washed with H₂O to give the title compound (26 mg, 46%) as a light yellow solid. Purity 98% .¹H NMR (400 MHz, DMSO- d_6) δ 9.60 (s, 1H), 8.88 (d, J = 2.2 Hz, 1H), 8.39 (d, J = 8.3 Hz, 1H), 8.27 (dd, J = 4.7, 1.4 Hz, 1H), 8.11 – 8.06 (m, 1H), 7.43 (s, 2H), 7.38 – 7.32 (m, 2H), 2.60 (s, 3H).¹³C NMR (100 MHz, DMSO-d₆) δ 164.21, 159.84, 158.52, 147.61, 144.13, 142.59, 135.73, 131.12, 127.95, 123.60, 123.27, 119.59, 94.87, 24.27. ESI HRMS: 285.08043 (M+H)+.

3-Amino-6-methyl-N-(pyridin-2-yl)thieno[2,3-b]pyridine-2-carboxamide 2,2,2 trifluoroacetate (4)—Synthesized using a similar procedure used to prepare **1** except using **29i**. Mixture was heated to reflux overnight. Upon cooling to room temperature and addition of H2O, yellow solid precipitated. Solid was filtered and washed with cold EtOH. Crude was purified by flash column chromatography (hexanes/EtOAc) on silica gel to give **4** (18 mg, 32%) as a yellow solid. Purity 100% .¹H NMR (400 MHz, DMSO- d_6) δ 9.54 (s, 1H), 8.39 (d, $J = 8.3$ Hz, 1H), 8.35 (d, $J = 4.4$ Hz, 1H), 8.05 (d, $J = 8.3$ Hz, 1H), 7.83 – 7.77 $(m, 1H), 7.42$ (s, 2H), 7.34 (d, $J = 8.3$ Hz, 1H), 7.12 (dd, $J = 6.8$, 5.2 Hz, 1H), 2.60 (s, 3H).13C NMR (100 MHz, DMSO-^d6) δ 163.99, 159.86, 158.62, 151.96, 147.88, 147.62, 137.86, 131.15, 123.66, 119.51, 119.35, 114.70, 95.67, 24.28. ESI HRMS: 285.08095 (M $+H)^+$.

3-Amino-4-methyl-N-(3-(trifluoromethyl)phenyl)thieno[2,3-b]pyridine-2-

carboxamide (5)—Synthesized using a similar procedure used to prepare **1** except using **31a**. Crude was purified by flash column chromatography (hexanes/EtOAc) on silica gel to give **5** (44 mg, 37%) as a light yellow solid. Purity 96%. ¹H NMR (500 MHz, DMSO- d_6) δ 9.78 (s, 1H), 8.49 (d, $J = 4.7$ Hz, 1H), 8.17 (s, 1H), 8.00 (d, $J = 8.3$ Hz, 1H), 7.57 (t, $J = 8.0$ Hz, 1H), 7.43 (d, J = 7.8 Hz, 1H), 7.21 (d, J = 5.4 Hz, 1H), 7.08 (s, 2H), 2.81 (s, 3H).¹³C NMR (125 MHz, DMSO-d₆) δ 164.40, 159.28, 150.00, 149.70, 145.16, 139.76, 129.64, 129.19 (q, J = 31.0 Hz), 125.28, 124.93, 124.49, 123.12, 122.01, 119.69, 117.08, 96.93, 20.00. ESI HRMS 352.07318 (M+H)+.

3-Amino-5-methyl-N-(3-(trifluoromethyl)phenyl)thieno[2,3-b]pyridine-2-

carboxamide (6)—Synthesized using a similar procedure used to prepare **1** except using **31b**. Crude was purified by silica gel chromatography (hexanes/EtOAc) to give **6** (36 mg, 30%) as a light yellow solid. Purity 99%. ¹H NMR (500 MHz, DMSO- d_6) δ 9.72 (s, 1H), 8.55 (s, 1H), 8.36 (s, 1H), 8.21 (s, 1H), 7.99 (d, $J = 8.1$ Hz, 1H), 7.56 (t, $J = 7.9$ Hz, 1H), 7.46 – 7.39 (m, 3H), 2.43 (s, 3H). ¹³C NMR (125 MHz, DMSO- d_6) δ 164.17, 156.13, 151.54, 147.34, 139.93, 130.86, 129.60, 129.16 (q, $J = 31.4$ Hz), 128.89, 125.80, 124.24, 124.22 (q, $J = 272.2$ Hz), 119.49 (d, $J = 3.5$ Hz), 116.85 (q, $J = 3.8$ Hz), 96.12, 17.96. ESI HRMS: 352.07272 (M+H)+.

3-amino-6-methyl-N-(4-(trifluoromethyl)phenyl)thieno[2,3-b]pyridine-2 carboxamide (7)—Synthesized using a similar procedure used to prepare **1** except using **33d**. Crude was purified by silica gel chromatography (hexanes/EtOAc) to give **7** (28 mg,

23%) as a light yellow solid. Purity 97%. ¹H NMR (400 MHz, DMSO- d_6) δ 9.72 (s, 1H), 8.40 (d, $J = 8.3$ Hz, 1H), 7.96 (d, $J = 8.5$ Hz, 2H), 7.67 (d, $J = 8.6$ Hz, 2H), 7.47 (s, 2H), 7.35 (d, $J = 8.3$ Hz, 1H), 2.60 (s, 3H).¹³C NMR (100 MHz, DMSO- d_6) δ 164.23, 159.94, 158.57, 147.91, 142.90, 131.14, 125.62 (q, J = 3.5 Hz), 123.55, 123.21, 120.48, 119.61, 94.91, 24.27. ESI HRMS: 352.07336 (M+H)+.

3-Amino-6-methyl-N-(2-(trifluoromethyl)phenyl)thieno[2,3-b]pyridine-2-

carboxamide (8)—Synthesized using a similar procedure used to prepare **1** except using **33e**. Crude was purified by silica gel chromatography (hexanes/EtOAc) to give **8** (32 mg, 38%) as a light yellow solid. Purity 96%. ¹H NMR (500 MHz, DMSO- d_6) δ 9.20 (s, 1H), 8.36 (d, $J = 8.3$ Hz, 1H), 7.77 (d, $J = 7.8$ Hz, 1H), 7.71 (t, $J = 8.0$ Hz, 1H), 7.57 (d, $J = 7.9$ Hz, 1H), 7.50 (t, $J = 7.6$ Hz, 1H), 7.34 (d, $J = 8.3$ Hz, 1H), 7.26 (s, 2H), 2.60 (s, 3H). ¹³C NMR (125 MHz, DMSO-d₆) δ 164.60, 159.64, 158.42, 146.95, 135.82, 132.89, 131.09, 130.77, 126.92, 126.40 (q, $J = 4.9$ Hz), 125.90 (q, $J = 29.2$ Hz), 123.74, 123.71 (q, $J = 273.3$ Hz), 119.56, 95.28, 24.29. ESI HRMS: 352.07291 (M+H)⁺.

3-Amino-6-methyl-N-(3-(trifluoromethyl)benzyl)thieno[2,3-b]pyridine-2-

carboxamide (9)—Synthesized using a similar procedure used to prepare **1** except using **33f**. Crude was purified by silica gel chromatography (hexanes/EtOAc) to give **9** (106 mg, 60%) as a light yellow solid. Purity 100%. ¹H NMR (500 MHz, DMSO- d_6) δ 8.35 – 8.30 $(m, 2H)$, 7.66 (s, 1H), 7.64 – 7.62 $(m, 1H)$, 7.61 – 7.55 $(m, 2H)$, 7.31 (d, J = 8.3 Hz, 1H), 7.18 (s, 2H), 4.48 (d, J = 5.9 Hz, 2H), 2.58 (s, 3H). ¹³C NMR (125 MHz, DMSO- d_6) δ 165.27, 159.50, 158.22, 145.89, 141.75, 131.65, 131.07, 129.53, 129.11 (q, $J = 31.5$ Hz), 124.47 (q, $J = 272.2$ Hz), 124.06, 123.90 (q, $J = 3.7$ Hz), 123.63 (q, $J = 3.8$ Hz), 119.69, 95.79, 42.03, 24.35. ESI HRMS: 366.08889 (M+H)+.

3-Amino-6-isopropyl-N-(3-(trifluoromethyl)phenyl)thieno[2,3-b]pyridine-2-

carboxamide (10)—Synthesized using a similar procedure used to prepare **1** except using **31c**. Crude was purified by flash column chromatography (hexanes/EtOAc) on silica gel to give **10** (44 mg, 78%) as a light yellow oil which solidified upon standing. Purity 93%. 1H NMR (500 MHz, DMSO- d_6) δ 9.68 (s, 1H), 8.44 (d, $J = 8.4$ Hz, 1H), 8.22 (s, 1H), 7.99 (d, J $= 8.2$ Hz, 1H), 7.56 (t, $J = 7.9$ Hz, 1H), 7.47 (s, 2H), 7.43 – 7.38 (m, 2H), 3.15 (p, $J = 6.9$ Hz, 1H), 1.29 (d, $J = 6.9$ Hz, 6H). ¹³C NMR (125 MHz, DMSO- d_6) δ 168.50, 164.23, 158.51, 147.78, 139.97, 131.47, 129.58, 129.15 (g, $J = 31.7$ Hz), 124.27, 124.23 (g, $J =$ 272.8 Hz), 124.03, 119.43 (q, $J = 3.8$ Hz), 117.37, 116.86 (q, $J = 4.0$ Hz), 94.92, 39.85, 35.91, 22.41. ESI HRMS: 380.10396 (M+H)+.

6-Methyl-N-(3-(trifluoromethyl)phenyl)thieno[2,3-b]pyridine-2-carboxamide (11)

—Synthesized using reported procedure with modification.25 To a solution of **1** (56 mg, 0.16 mmol) in MeOH (6 mL) was added a solution of 50% H3PO2 (259 $> L$, 2.4 mmol) at 0 °C. To the reaction mixture was added conc. H_2SO_4 (655 >L), Cu₂O (14 mg, 0.10 mmol),1 M NaNO₂ aqueous solution (239 >L) and the reaction mixture was stirred for 30 min at 0 °C. After stirring for 2 h and 30 min at room temperature, the reaction mixture was diluted with H2O (30 mL) and the aqueous layer was extracted with EtOAc (20 mL –). The combined organic layer was washed with brine, dried over Na2SO4, filtered and concentrated in vacuo.

Crude was purified by flash column chromatography (hexanes/EtOAc) on silica gel to give **11** (12 mg, 22%) as a white solid. Purity 98% .¹H NMR (500 MHz, DMSO- d_6) δ 10.83 (s, 1H), $8.36 - 8.32$ (m, 2H), 8.21 (s, 1H), 8.05 (d, $J = 8.2$ Hz, 1H), 7.63 (t, $J = 8.0$ Hz, 1H), 7.49 (d, $J = 7.8$ Hz, 1H), 7.41 (d, $J = 8.2$ Hz, 1H), 2.63 (s, 3H).¹³C NMR (125 MHz, DMSO d_6) δ 161.21, 160.54, 158.36, 139.31, 137.39, 133.71, 130.21, 130.09, 129.46 (q, $J = 31.7$ Hz), 124.53, 124.09 (q, $J = 272.3$ Hz), 123.78, 120.96, 120.34 (q, $J = 3.7$ Hz), 116.35 (q, $J =$ 4.0 Hz), 24.29. ESI HRMS: 337.06096 (M+H)+.

7-Methyl-3-(3-(trifluoromethyl)phenyl)pyrido[3′**,2**′**:4,5]thieno[3,2-**

d]pyrimidine-2,4(1H,3H)-dione (12)—Synthesized using reported procedure with modification.26 Triphosgene (13 mg, 0.043 mmol) was added to a stirred solution of **1** (30 mg, 0.084 mmol) in THF (2.5 mL) at room temperature. The resulting mixture was heated under reflux for 1 h. The mixture was then cooled to room temperature and diluted with THF (5 mL). The solid was collected by filtration and rinsed with cold THF to give a white solid (28 mg, 90%). Purity 98%. ¹H NMR (500 MHz, DMSO- d_6) δ 12.81 (s, 1H), 8.65 (d, J $= 8.4$ Hz, 1H), 7.87 (s, 1H), 7.83 (d, J = 7.4 Hz, 1H), 7.78 – 7.72 (m, 2H), 7.55 (d, J = 8.4 Hz, 1H), 2.67 (s, 3H). ¹³C NMR (125 MHz, DMSO- d_6) δ 161.09, 160.73, 159.08, 151.33, 138.79, 136.39, 133.62, 131.96, 130.16, 129.73 (q, $J = 32.1$ Hz), 126.28 (q, $J = 3.9$ Hz), 125.24 (q, J = 3.7 Hz), 124.98, 122.82, 121.07, 120.37, 107.68, 24.46.. ESI HRMS: $378.05243 (M+H)⁺$.

Representative procedure for thienopyridine sulfonamide preparation

3-Amino-6-methyl-N-(3-(trifluoromethyl)phenyl)thieno[2,3-b]pyridine-2 sulfonamide (13)—Synthesized using reported procedure with modification.²⁷ To a solution of **25** (32 mg, 0.21 mmol) in anhydrous DMF (1.5 mL) was added **35** (63 mg, 0.22 mmol) and sodium ethoxide (33 mg, 0.46 mmol). The mixture was heated to 130 °C for 1 h. The mixture was diluted with H₂O (10 mL) and extracted with EtOAc (10 mL \times 2) and finally washed with brine (10 mL). The organic layer was dried (Na_2SO_4) , filtered, and concentrated in vacuo. Crude was purified by silica gel chromatography (hexanes/EtOAc) to give 13 (27 mg, 33%) as a light yellow solid. Purity 96%.¹H NMR (400 MHz, DMSO- d_6) δ 10.75 (s, 1H), 8.35 (d, $J = 8.3$ Hz, 1H), 7.49 (t, $J = 7.9$ Hz, 1H), 7.45 – 7.38 (m, 2H), 7.37 – 7.30 (m, 2H), 6.83 (s, 2H), 2.55 (s, 3H).13C NMR (126 MHz, DMSO) δ 160.74, 158.96, 144.69, 139.28, 132.25, 130.93, 130.25 (q, J = 31.5 Hz), 125.34, 123.57, 123.16, 120.51, 115.43, 97.64, 24.68. ESI HRMS: 388.04044 (M+H)+.

3-Amino-N-(3-fluorophenyl)-6-methylthieno[2,3-b]pyridine-2-sulfonamide (14)

—Synthesized using a similar procedure used to prepare **13** except using **36**. Crude was purified by silica gel chromatography (hexanes/EtOAc) to give **14** (96 mg, 72%) as an oil which was solidified to a beige solid upon standing. Purity 100% . ¹H NMR (400 MHz, DMSO- d_6) δ 10.60 (s, 1H), 8.33 (d, J = 8.3 Hz, 1H), 7.33 (d, J = 8.4 Hz, 1H), 7.26 (td, J = 8.4, 6.9 Hz, 1H), 6.97 – 6.88 (m, 2H), 6.85 – 6.80 (m, 1H), 6.80 – 6.76 (m, 2H), 2.54 (s, 3H).¹³C NMR (100 MHz, DMSO- d_6) δ 162.21 (d, J = 242.8 Hz), 160.13, 158.48, 143.89, 139.73 (d, $J = 10.7$ Hz), 131.72, 130.82 (d, $J = 9.5$ Hz), 123.13, 119.98, 114.88 (d, $J = 2.7$ Hz), 110.08 (d, $J = 21.0$ Hz), 105.70 (d, $J = 25.4$ Hz), 97.72, 24.17. ESI HRMS: 338.04340 $(M+H)^+$.

Representative procedure for oxadiazolyl thienopyridine preparation

6-Methyl-2-(5-(3-(trifluoromethyl)phenyl)-1,3,4-oxadiazol-2-yl)thieno[2,3-

b]pyridin-3-amine (15)—To a stirring solution of **25** (42 mg, 0.28 mmol) in anhydrous DMF (2 mL) was added commercially available 5-(chloromethyl)-3-[3- (trifluoromethyl)phenyl]-1,3,4-oxadiazole, **37I**, (66 mg, 0.24 mmol) and sodium ethoxide (50 mg, 0.63 mmol). The mixture was heated to 130 °C for 1 h. The mixture was then cooled to room temperature and solid precipitated at the bottom of the flask. The solid was collected by filtration and rinsed with cold MeOH and dried in vacuo to give a yellow solid (28 mg, 31% yield). Purity 96%. ¹H NMR (500 MHz, DMSO- d_6) δ 8.50 (d, $J = 8.3$ Hz, 1H), 8.40 (d, $J = 7.9$ Hz, 1H), 8.35 (s, 1H), 8.01 (d, $J = 7.9$ Hz, 1H), 7.87 (t, $J = 7.8$ Hz, 1H), 7.39 (d, $J = 8.3$ Hz, 1H), 7.26 (s, 2H), 2.61 (s, 3H). ¹³C NMR (125 MHz, DMSO- d_6) δ 162.36, 160.47, 159.33, 159.28, 143.73, 131.25, 130.80, 130.50, 130.18, 129.92, 128.06 (q, $J = 3.3$ Hz), 124.80, 124.45, 123.38, 122.77 (q, $J = 3.7$ Hz), 122.63, 119.88, 85.96, 24.21. ESI HRMS: 377.06832 (M+H)+.

6-Methyl-2-(3-(3-(trifluoromethyl)phenyl)-1,2,4-oxadiazol-5-yl)thieno[2,3-

b]pyridin-3-amine (16)—Synthesized using a similar procedure used to prepare **15** except using commercially available 5-(chloromethyl)-3-[3-(trifluoromethyl)phenyl]-1,3,4 oxadiazole, **37m**. Precipitated solid was filtered, rinsed with cold MeOH, and dried in vacuo to give the title compound (31 mg, 67% yield) as a red solid. Purity 99%. ¹H NMR (500 MHz, DMSO- d_6) δ 8.55 (d, $J = 8.3$ Hz, 1H), 8.48 (d, $J = 7.7$ Hz, 1H), 8.43 (s, 1H), 8.00 (d, J $= 7.6$ Hz, 1H), 7.84 (t, $J = 7.7$ Hz, 1H), 7.63 (s, 2H), 7.41 (d, $J = 8.3$ Hz, 1H), 2.61 (s, 3H). 13C NMR (125 MHz, DMSO-^d6) δ 172.46, 166.38, 160.44, 159.78, 146.38, 131.88, 131.35, 130.51, 129.92 (q, $J = 32.1$ Hz), 128.09 (d, $J = 3.3$ Hz), 127.46, 123.64 (q, $J = 3.7$ Hz), 122.93, 122.76 (q, $J = 272$ Hz), 119.99, 85.79, 24.28.ESI HRMS: 377.06809 (M+H)⁺.

6-Methyl-2-(5-phenyl-1,3,4-oxadiazol-2-yl)thieno[2,3-b]pyridin-3-amine (17)—

Synthesized using a similar procedure used to prepare **15** except using **25** and commercially available 2-(chloromethyl)-5-phenyl-1,3,4-oxadiazole, **37n**. Precipitated solid was filtered, rinsed with cold MeOH, and dried in vacuo to give the title compound (24 mg, 65% yield) as a yellow solid. Purity 96%. ¹H NMR (500 MHz, DMSO d_6) δ 8.46 (d, J = 8.3 Hz, 1H), 8.11 -8.08 (m, 2H), $7.65 - 7.61$ (m, 3H), 7.40 (d, $J = 8.3$ Hz, 1H), 7.18 (s, 2H), 2.61 (s, 3H). ¹³C NMR (125 MHz, DMSO-d₆) δ 161.93, 161.45, 159.19, 159.17, 143.37, 131.70, 131.12, 129.41, 126.50, 123.40, 123.28, 119.87, 86.23, 24.21.ESI HRMS: 309.08079 (M+H)+.

2-(5-(3-Chlorophenyl)-1,3,4-oxadiazol-2-yl)-6-methylthieno[2,3-b]pyridin-3-

amine (18)—Synthesized using a similar procedure used to prepare **15** except using **25** and commercially available 2-(chloromethyl)-5-(3-chlorophenyl)-1,3,4-oxadiazole, **37o**. Precipitated solid was filtered, rinsed with cold MeOH, and dried *in vacuo* to give the title compound (26 mg, 42% yield) as a yellow solid. Purity 96%. ¹H NMR (500 MHz, DMSO d_6) δ 8.46 (d, J = 8.3 Hz, 1H), 8.11 (t, J = 1.7 Hz, 1H), 8.06 (dt, J = 7.6, 1.2 Hz, 1H), 7.70 (ddd, $J = 8.1, 2.0, 1.1$ Hz, 1H), 7.65 (t, $J = 7.9$ Hz, 1H), 7.39 (d, $J = 8.3$ Hz, 1H), 7.20 (s, 2H), 2.61 (s, 3H). 13C NMR (125 MHz, DMSO-^d6) δ 162.22, 160.43, 159.29, 159.27, 143.58, 134.03, 131.44, 131.41, 131.16, 125.93, 125.25, 125.20, 123.38, 119.87, 86.08, 24.21. ESI HRMS: 343.04216 (M+H)+.

2-(5-(3-(Trifluoromethyl)phenyl)-1,3,4-oxadiazol-2-yl)thieno[2,3-b]pyridin-3 amine (19)—Synthesized using a similar procedure used to prepare **15** except using commercially available 2-Mercaptopyridine-3-carbonitrile, **38**, and **37m**. Precipitated solid was filtered, rinsed with cold MeOH, and dried *in vacuo* to give the title compound (55 mg, 76% yield) as a yellow solid. Purity 99%. ¹H NMR (500 MHz, DMSO- d_6) δ 8.69 (dd, $J =$ 4.6, 1.6 Hz, 1H), 8.61 (dd, $J = 8.2$, 1.6 Hz, 1H), 8.41 (d, $J = 7.9$ Hz, 1H), 8.36 (s, 1H), 8.02 (d, $J = 7.9$ Hz, 1H), 7.88 (t, $J = 7.8$ Hz, 1H), 7.54 (dd, $J = 8.1$, 4.6 Hz, 1H), 7.29 (s, 2H). ¹³C NMR (125 MHz, DMSO-d₆) δ 162.22, 160.66, 159.56, 149.93, 143.49, 131.17, 130.80, 130.55, 129.94 (q, $J = 32$ Hz), 128.14 (q, $J = 3.3$ Hz), 125.69, 124.79, 124.41, 122.84 (q, $J =$ 3.9 Hz), 122.62, 119.86, 87.02. ESI HRMS: 363.05232 (M+H)+.

Representative procedure for 2-mercaptopyridine preparation

2-Mercapto-6-methylnicotinonitrile (25)—Synthesized using reported procedure with modification.20 A mixture of 2-chloro-6-methylnicotinonitrile, **24**, (222 mg, 1.44 mmol) and thiourea (348 mg, 4.57 mmol) was heated at reflux (118 °C) in n-butanol for 3 h. A yellow suspension was obtained. After cooling to room temperature, the yellow solid was precipitated. The solids were collected by filtration, rinsed with cold n-butanol, and dried in *vacuo* to give the title compound (217 mg, quantitative) as a light yellow solid. ¹H NMR $(500 \text{ MHz}, \text{ DMSO-}d_6)$ δ 14.07 (s, 1H), 7.99 (d, J = 7.7 Hz, 1H), 6.73 (d, J = 7.7 Hz, 1H), 2.39 (s, 3H). ¹³C NMR (125 MHz, DMSO-d₆) δ 177.70, 155.00, 145.03, 117.20, 113.40, 112.75, 19.14. ESI MS: m/z 149.1 (M-H)−.

Representative procedure for acetamide preparation

2-Bromo-N-(3-(trifluoromethyl)phenyl)acetamide (27)—Synthesized using reported procedure with modification.²¹ A stirred solution of 3-(trifluoromethyl)aniline, **26**, (170 $>$ L, 1.36 mmol) and 4-(dimethylamino)pyridine (70 mg, 42 mol%) in CH₂Cl₂ (8 mL) was treated dropwise with bromoacetyl bromide (140 $> L$, 1.61 mmol) at 0 °C. The reaction mixture was allowed to warm up to room temperature with good stirring overnight. The mixture was diluted with H₂O (30 mL) and extracted with EtOAc (20 mL \times 2). The organic layer was dried ($MgSO₄$), filtered, and concentrated *in vacuo*. Crude was purified by flash column chromatography (hexanes/EtOAc) on silica gel to give **27** (339 mg, 88%) as a light brown oil which solidified upon standing. ¹H NMR (500 MHz, CDCl₃) δ 8.23 (s, 1H), 7.83 $(s, 1H)$, 7.75 (d, J = 8.1 Hz, 1H), 7.49 (t, J = 7.9 Hz, 1H), 7.45 – 7.41 (m, 1H), 4.05 (s, 2H).¹³C NMR (125 MHz, CDCl₃) δ 163.70, 137.55, 131.74 (q, J = 32.7 Hz), 129.87, 123.83 (q, $J = 272.4$ Hz), 123.18, 121.95 (q, $J = 3.8$ Hz), 116.88 (q, $J = 3.9$ Hz), 29.40. ESI MS: m/z 281.55, 283.50 (M+H)⁺.

Representative procedure for pyridinium acetamido bromide preparation

4-(2-Bromoacetamido)pyridin-1-ium bromide (29g)—Synthesized using reported procedure with modification.23 To a stirred solution of 4-aminopyridine, **28g**, (201 mg, 2.14 mmol) in dry CH₂Cl₂ (5 mL) at 0 °C was added bromoacetyl bromide (195 >L, 2.24 mmol) dropwise. As soon as bromoacetyl bromide was added, the reaction mixture turned cloudy and a white precipitate formed. The mixture was allowed to warm to room temperature with good stirring overnight. Mixture was filtered and the solid was washed with cold CH_2Cl_2 to

give **29g** as a white solid which was used in the next step without further purification. ESI MS: m/z 214.45, 216.45 (M+H)⁺.

3-(2-Bromoacetamido)pyridin-1-ium bromide (29h)—Synthesized using a similar procedure used to prepare **29g** using 3-aminopyridine, **28h**. The mixture was allowed to warm to room temperature with good stirring over 3 h. Mixture was filtered and the solid was washed with cold CH₂Cl₂ to give the title compound (609 mg, 95%) as a beige solid which was used in the next step without further purification. ${}^{1}H$ NMR (400 MHz, DMSO d_6) δ 11.42 (s, 1H), 9.20 (d, $J = 2.3$ Hz, 1H), 8.65 (d, $J = 5.4$ Hz, 1H), 8.50 (d, $J = 8.6$ Hz, 1H), 7.99 (dd, $J = 8.5$, 5.5 Hz, 1H), 4.19 (s, 2H).¹³C NMR (100 MHz, DMSO- d_6) δ 166.22, 137.81, 137.58, 134.35, 132.52, 127.48, 29.53. ESI MS: m/z 214.45, 216.40 (M+H)+.

2-Bromo-N-(pyridin-2-yl)acetamide hydrobromide (29i)—Synthesized using a similar procedure used to prepare **29g** except using 2-aminopyridine, **28i**. The mixture was allowed to warm to room temperature with good stirring over 3 h. Mixture was filtered and the solid was washed with cold CH_2Cl_2 to give the title compound (342 mg, 55%) as a light purple solid which was used in the next step without further purification. 1 H NMR (500) MHz, CDCl₃) δ 8.81 (s, 1H), 8.32 (d, $J = 3.7$ Hz, 1H), 8.17 (s, 1H), 7.74 (t, $J = 7.6$ Hz, 1H), 7.13 – 7.06 (m, 1H), 4.01 (s, 2H).¹³C NMR (125 MHz, CDCl₃) δ 164.23, 150.72, 148.15, 138.66, 120.72, 114.02, 29.20. ESI MS: m/z 214.45, 216.45 (M+H)+.

2-Mercapto-4-methylnicotinonitrile (31a)—Synthesized using a similar procedure used to prepare **25** except using 2-chloro-4-methylnicotinonitrile, **30a**. Precipitated solid was filtered, rinsed with cold n-butanol, and dried *in vacuo* to give the title compound (237 mg, quantitative) as a light yellow solid. ¹H NMR (400 MHz, DMSO- d_6) δ 7.81 (d, J = 6.4 Hz, 1H), 6.82 (d, J = 6.4 Hz, 1H), 2.40 (s, 3H).¹³C NMR (100 MHz, DMSO- d_6) δ 177.05, 157.57, 140.65, 116.31, 115.93, 114.61, 20.99. ESI MS: m/z 149.1 (MH) −.

2-Mercapto-5-methylnicotinonitrile (31b)—Synthesized using a similar procedure used to prepare **25** except using 2-chloro-5-methylnicotinonitrile, **30b**. Precipitated solid was filtered, rinsed with cold n-butanol, and dried *in vacuo* to give the title compound (121 mg) , quantitative) as a light yellow solid. ¹H NMR (500 MHz, DMSO- d_6) δ 14.10 (s, 1H), 8.04 (d, $J = 1.9$ Hz, 1H), 7.83 (s, 1H), 2.10 (s, 3H). ¹³C NMR (126 MHz, DMSO) δ 174.77, 147.03, 141.45, 122.24, 116.90, 116.08, 16.06. ESI MS: m/z 149.1 (M-H) −.

6-Isopropyl-2-mercaptonicotinonitrile (31c)—Synthesized using reported procedure with modification.²⁴ A suspension of 2-chloro-6-isopropylnicotinonitrile, **30c**, (53 mg, 0.29) mmol) and Lawesson's reagent (135 mg, >90% purity, 0.30 mmol) in toluene (2.5 mL) was refluxed at 110 °C for 4 h 30 min. The reaction mixture was cooled down to room temperature and filtered through a Celite plug to remove any unreacted Lawesson's reagent. The solvent was evaporated and crude was purified by silica gel chromatography (hexanes/ EtOAc) to give $31c$ (22 mg, 42% yield) as a yellow sticky solid. ¹H NMR (400 MHz, DMSO- d_6) δ 14.04 (s, 1H), 8.04 (d, J = 7.8 Hz, 1H), 6.78 (d, J = 7.8 Hz, 1H), 3.06 (p, J = 6.9 Hz, 1H), 1.20 (d, $J = 6.9$ Hz, 6H). ¹³C NMR (100 MHz, DMSO- d_6) δ 177.78, 163.81, 145.44, 117.06, 113.98, 109.04, 31.42, 21.17. ESI MS: m/z 179.1 (M+H)+, 177.1 (M-H) −.

2-Bromo-N-(4-(trifluoromethyl)phenyl)acetamide (33d)—Synthesized using a similar procedure used to prepare **27** except using 4-(trifluoromethyl)aniline, **32d**. The reaction mixture was allowed to warm up to room temperature with good stirring over 8 h. The mixture was diluted with H₂O (30 mL) and extracted with EtOAc (20 mL \times 2). The organic layer was dried $(MgSO_4)$, filtered, and concentrated *in vacuo*. Crude (355 mg, 94%), a white solid, was used in the next step without further purification. ¹H NMR (500 MHz, CDCl₃) δ 8.24 (s, 1H), 7.68 (d, $J = 8.5$ Hz, 2H), 7.62 (d, $J = 8.5$ Hz, 2H), 4.05 (s, 2H).¹³C NMR (125 MHz, CDCl₃) δ 163.69, 140.05, 127.18 (q, $J = 33.0$ Hz), 126.56 (q, $J = 3.8$ Hz), 124.04 (g, $J = 271.8$ Hz), 119.73, 29.44. ESI MS: m/z 281.50, 283.55 (M+H)⁺.

2-Bromo-N-(2-(trifluoromethyl)phenyl)acetamide (33e)—Synthesized using a similar procedure used to prepare **27** except using 2-(trifluoromethyl)aniline, **32e**, and letting the reaction to stir overnight at room temperature. Crude (397 mg, 89%), a beige solid, was used in the next step without further purification. ¹H NMR (500 MHz, CDCl₃) δ 8.60 (s, 1H), 8.19 (d, $J = 8.3$ Hz, 1H), 7.65 (d, $J = 7.9$ Hz, 1H), 7.59 (t, $J = 7.9$ Hz, 1H), 7.29 (t, $J =$ 7.7 Hz, 1H), 4.07 (s, 2H). ¹³C NMR (125 MHz, CDCl₃) δ 163.89, 134.59, 133.11, 126.38 $(q, J = 5.3 \text{ Hz})$, 125.40, 124.17, 123.98 $(q, J = 273.1 \text{ Hz})$, 120.79 $(q, J = 29.9 \text{ Hz})$, 29.43. ESI MS: m/z 281.50, 283.50 (M+H)+.

2-Bromo-N-(3-(trifluoromethyl)benzyl)acetamide (33f)—Synthesized using a similar procedure used to prepare **27** except using 3-(trifluoromethyl)benzylamine, **32f**, and letting the reaction to stir for 6h at room temperature. Crude was purified by silica gel chromatography (hexanes/EtOAc) to give $33f(316 \text{ mg}, 62\%)$ as a white solid. ¹H NMR (500) MHz, CDCl₃) δ $7.58 - 7.55$ (m, 1H), 7.54 (s, 1H), $7.51 - 7.45$ (m, 2H), 6.85 (s, 1H), 4.54 $(d, J = 6.0 \text{ Hz}, 2\text{H})$. ¹³C NMR (125 MHz, CDCl₃) δ 165.68, 138.54, 131.31 (q, $J = 33.5 \text{ Hz}$), 131.18, 129.49, 124.79, 124.54, 124.06 (g, $J = 272.1$ Hz), 43.80, 29.16. ESI MS: m/z 295.50, 297.55 (M+H)+.

Representative procedure for chloromethanesulfonamide preparation

1-Chloro-N-(3-(trifluoromethyl)phenyl)methanesulfonamide (35)—Synthesized using reported procedure with modification.²⁷ To a stirring solution of 3-(trifluoromethyl)aniline, 26 , $(388 \times L, 3.11 \text{ mmol})$ in CH₂Cl₂ (6 mL) was added Et₃N (865) $>$ L, 6.20 mmol) at room temperature. The reaction mixture was cool down to 0 $^{\circ}$ C and chloromethanesulfonyl chloride (626 > L, 6.20 mmol) in CH₂Cl₂ (3 mL) was added at 0 °C. The reaction mixture was stirred at room temperature overnight. The reaction mixture was concentrated on vacuo then 2 N NaOH (15 mL) was added and the mixture was stirred at room temperature for 2 h 30 min. The reaction mixture was neutralized by 1N HCl. The mixture was diluted with H₂O (10 mL) and extracted with EtOAc (20 mL \times 2) and finally washed with brine (10 mL). The organic layer was dried (Na_2SO_4) , filtered, and concentrated in vacuo. Crude was purified by silica gel chromatography (hexanes/EtOAc) to give 35 (614 mg, 72%) as a beige solid. ¹H NMR (400 MHz, DMSO- d_6) δ 10.77 (s, 1H), 7.62 – 7.56 (m, 1H), 7.56 – 7.51 (m, 2H), 7.49 (d, $J = 7.6$ Hz, 1H), 5.13 (s, 2H). ¹³C NMR $(100 MHz, DMSO-d₆)$ δ 138.25, 130.64, 129.96 (q, $J = 31.8$ Hz), 125.17, 123.64, 122.46, 120.71 (q, $J = 3.6$ Hz), 116.08 (q, $J = 4.1$ Hz), 54.87. ESI MS: m/z 272.0 (M-H) ⁻.

1-Chloro-N-(3-fluorophenyl)methanesulfonamide (36)—Synthesized using a similar procedure used to prepare **35** except using 3-fluoroaniline, **34**. Crude was purified by silica gel column chromatography (hexanes/EtOAc) to give **36** (347 mg, 60%) as a yellow oil which was solidified upon standing. ¹H NMR (400 MHz, CDCl₃) δ 7.35 (td, *J* = 8.2, 6.3 Hz, 1H), $7.11 - 7.03$ (m, 2H), 6.97 (tdd, $J = 8.3, 2.4, 0.8$ Hz, 1H), 6.90 (s, 1H), 4.51 (s, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 163.33 (d, $J = 248.3$ Hz), 136.96 (d, $J = 10.2$ Hz), 131.22 (d, $J =$ 9.3 Hz), 117.68 (d, $J = 3.2$ Hz), 113.62 (d, $J = 21.2$ Hz), 109.71 (d, $J = 25.2$ Hz), 53.58. ESI MS: m/z 222.00 (M-H) [−].

Supplementary Material

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Figure 1.

(A) Secondary structure of 29nt TAR hairpin with the sites of 2AP replacement highlighted. Constructs were designed with either U23 or U25 replaced with 2-AP. Binding curves from (B) 25-2AP and (C) FRET assays for **2** and **15**.

Figure 2.

(A) Ligand-detected 1H NMR spectra of aromatic region of **2** (150 >M) and B) **14** (150 >M) in the presence (red spectra) and absence of TAR hairpin (10 >M) (black spectra).

Figure 3.

Overlaid 2D TOCSY NMR spectra of pyrimidine residues of TAR (A) with increasing concentrations of **2**. Indicated are free TAR (black) and TAR:**2** ratio of 1:1 (green), 1:2 (light blue), 1:3 (dark blue), and 1:5 (red). Arrows indicate the direction of chemical shift changes upon addition of **2**. (B) Secondary structure of 29nt TAR hairpin with bases perturbed in the presence of **2** highlighted with red circles.

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Figure 4.

SHAPE analysis of the HIV-1 5′ UTR in the presence of (A) **2** and (B) **15**, respectively. top: nucleotides are color-coded according to normalized SHAPE reactivity values. 1M7 reactivity of the extreme 5′ and 3′ terminal nucleotides (represented in white) are not assigned due to limited resolution in the immediate vicinity of the primer and the final extension product. Nucleotide 1 is vector derived. bottom: step plots for quantitative comparison of 1M7 reactivity values obtained from the HIV-1 RNA 5′ UTR probed in the presence (black) or absence (red) of tested analog. Boxed region corresponds to nucleotides involved in formation of the 5′ TAR hairpin (nts G2–C58).

Scheme 1.

Synthetic route to **1** a

^a Reagents and conditions: (a) thiourea, n-BuOH, reflux, 3 h, quantitative; (b) bromoacetyl bromide, DMAP, DCM, 0 °C to rt, 12 h, 88%; (c) NaOEt, EtOH, reflux, 1 h, 58%.

Scheme 2.

Synthetic routes to (A) pyridinyl analogs **2**–**4** (B) **5**–**10**^a

a Reagents and conditions: (a) bromoacetylbromide, DCM, rt, 3–12 h, 55%-95%; (b) **25**, NaOEt, EtOH, reflux, 12 h, 32%-46%; (c) thiourea, n-BuOH, reflux, 3 h, quantitative; (d) Lawesson's reagent, toluene, reflux, 4 h 30 min, 42%; (e) bromoacetyl bromide, DMAP, DCM, 0 °C to rt, 6–12 h, 62%-94%; (f) NaOEt, EtOH, reflux, 1 h, 23%-78%.

Scheme 3.

Synthetic route to **11**–**14**^a

^a Reagents and conditions: (a) 50% aq. H₃PO₂, conc. H₂SO₄, Cu₂O, 1M aq. NaNO₂, MeOH, 0 °C to rt, 2 h 30 min, 22%; (b) triphosgene, THF, rt, 1 h, 98%; (c) chloromethylsulfonyl chloride, Et₃N, DCM, 0° C to rt, 12 h then 2N NaOH, rt, 2 h 30 min, 60%-72%; (d) 25, NaOEt, DMF, 130 °C, 1 h, 33%-72%.

Scheme 4.

Synthetic route to oxadiazole analogs **15**–**19**^a

a Reagents and conditions: (a) **25** or 2-mercaptopyridine-3-carbonitrile (**38**), NaOEt, DMF, 130 °C, 1 h, 31%-76%.

Table 1

Affinity measurement and Tat/TAR displacement assay for **1** and analogs

 a^a Highest concentration evaluated was 400 µM unless otherwise indicated. Values represent the average of three replicates, \pm standard deviation.