Identification of a nitrogenase FeMo cofactor precursor on NifEN complex

Yilin Hu, Aaron W. Fay, and Markus W. Ribbe*

Department of Molecular Biology and Biochemistry, University of California, Irvine, CA 92697-3900

Edited by Douglas C. Rees, California Institute of Technology, Pasadena, CA, and approved January 25, 2005 (received for review December 9, 2004)

The biosynthesis of the FeMo cofactor (FeMoco) of Azotobacter vinelandii nitrogenase presumably starts with the production of its Fe/S core by NifB (the *nifB* gene product). This core is subsequently processed on the $\alpha_2\beta_2$ tetrameric NifEN complex (formed by the *nifE* and *nifN* gene products). In this article, we identify a NifENbound FeMoco precursor form that can be converted to fully assembled FeMoco in a so-called FeMoco-maturation assay containing only purified components. We also establish that only molybdate, homocitrate, MgATP, and Fe protein are essential for FeMoco maturation. The FeMoco-maturation assay described here will further address the remaining questions related to the assembly mechanism of the ever-intriguing FeMoco.

Fe protein | FeMoco | MoFe protein

The biochemical machinery for the reduction of dinitrogen to ammonia is provided by the metalloenzyme nitrogenase (for recent reviews, see refs. 1–7). This enzyme is composed of two proteins, the Fe and the MoFe protein. The homodimeric Fe protein couples ATP hydrolysis to interprotein electron transfer, serving as an obligate electron donor to the catalytically active component, the MoFe protein. The $\alpha_2\beta_2$ tetrameric MoFe protein contains two copies of unique metal clusters, designated the P-cluster and the FeMoco, respectively. Whereas the [8Fe—7S] P-cluster (8) likely participates in interprotein electron transfer, the FeMoco serves as the active site of substrate binding and reduction.

There is a vast amount of interest in elucidating the mechanism by which the metalloclusters of the MoFe protein are synthesized *in vivo* because of their importance in N₂ fixation and because they are biologically and chemically unprecedented. In particular, the structure and assembly of FeMoco, which provides the site of substrate binding and reduction, has attracted considerable attention for more than a decade. FeMoco is a heterometallic double cubane consisting of one [4Fe—3S] and one [Mo—3Fe—3S] partial cubane that are bridged by three sulfides and share a μ_6 -central atom of which the identity is unknown but is considered to be C, O, or N (9). Situated entirely in the α -subunit, FeMoco is attached to the protein by only two ligands: a Cys that is bound to the Fe at one end of the cluster and a His that is bound to the Mo at the opposite end of the cluster. The Mo is also coordinated by homocitrate.

Progress has been made in understanding the biosynthesis of FeMoco in *Azotobacter vinelandii*, which starts with the production of an Fe/S core of the FeMoco (designated NifB-co) by NifB,[†] the *nifB* gene product (11, 12). NifB-co, which probably contains all of the Fe and S that ends up in the FeMoco, is then transferred to the $\alpha_2\beta_2$ tetrameric scaffold NifEN protein (encoded by the *nifE* and *nifN* gene products) (13, 14). Subsequently, NifB-co is further processed on NifEN by an unknown mechanism and forms the completed FeMoco. The process of FeMoco formation also requires the Fe protein and MgATP (15–17). The completed FeMoco is then presumably transferred to a *nafY*-encoded protein, called γ (18, 19), which delivers the FeMoco to a FeMoco-deficient form of MoFe protein. However, whether γ is essential for MoFe protein assembly process has been questioned recently (19).

Sequence similarity of nifE and nifN to nifD and nifK (encoding the α - and β -subunits of the MoFe protein, respectively) has led to the hypothesis that NifE and NifN form a complex (20) that is structurally homologous to the MoFe protein (21). Thus, by analogy to the MoFe protein, it has been speculated that NifEN could also contain two types of metallocluster sites. One such site within the NifEN complex could be analogous to the MoFe protein P-cluster site, whereas the other site may provide a place for FeMoco assembly (13). Although the proposed P-cluster analogue has been identified as a [4Fe-4S] cluster that is likely coordinated at the NifE–NifN interface by NifE–Cys-37, NifE-Cys-62, NifE-Cys-124, and NifN-Cys-44 (13), identification of the speculated FeMoco precursor has proven to be elusive so far, hampering the possibility to answer such questions as follow. (i) How and when is Mo and homocitrate incorporated into the cluster? (ii) What are the functions of Fe protein and MgATP in the process? (*iii*) What is the role of the nonessential γ protein in the MoFe protein assembly? There was evidence that such a FeMoco precursor was attached to the NifEN complex in the early stages of its purification from a $\Delta nifHDK$ background (14). However, the NifEN complexes isolated from either a $\Delta nifB$ or a $\Delta nifHDK$ background appeared to be identical and "FeMoco precursor-less" (14, 22). In other words, the FeMoco precursor was lost from the NifEN complex during its time-consuming purification (13).

In recent years, we have successfully improved the one-step purification procedure of the His-tagged, fragile MoFe protein variants, and the results of such improvement include increased yield (in some cases, >10-fold) and improved integrity of the metalloproteins in terms of their metal clusters [a recent example being the identification of a P-cluster precursor, which like the proposed FeMoco precursor on NifEN, was easily "lost" during the conventional protein purification (23–25)]. Here, we apply this procedure to the purification of His-tagged NifEN and $\Delta nifB$ NifEN from A. vinelandii strains DJ1041 and YM9A, respectively. We report a large-scale isolation of NifEN that contains a FeMoco precursor. Also, we show that through a so-called FeMoco-maturation assay that requires Fe protein and, likely, the hydrolysis of MgATP, this NifEN-bound FeMoco precursor can be converted to mature FeMoco with the addition of molybdate and homocitrate. We believe that this assay can be used as a test system that allows us to further address questions related to the assembly mechanism of FeMoco.

Materials and Methods

Unless noted otherwise, all chemicals and reagents were obtained from Fisher Scientific, Baxter Scientific Products (McGaw Park, IL), or Sigma.

This paper was submitted directly (Track II) to the PNAS office.

Abbreviations: FeMoco, FeMo cofactor; IDS, indigo disulfonate.

^{*}To whom correspondence should be addressed. E-mail: mribbe@uci.edu.

[†]The C-terminal region of NifB bears significant sequence similarity to NifY, NifX, and γ , all of which are involved in the biosynthesis of the FeMoco (10). The N-terminal region of NifB shows sequence similarity to members of the radical S-adenosylmethionine-dependent enzyme superfamily (10).

^{© 2005} by The National Academy of Sciences of the USA

Construction of Variant A. vinelandii Strains. Construction of the A. vinelandii nifB-deletion strain DJ1143 (AvDJ1143) producing a His-tagged MoFe protein (designated $\Delta nifB$ MoFe protein) and nifHDKTY-deletion strain DJ1041 (AvDJ1041) producing a Histagged NifEN (designated NifEN) has been described earlier (13, 26). By following the methods described in ref. 26, pHR18, a pGemT(Easy)-originated plasmid carrying a fragment of nifB gene was constructed. This fragment has an internal portion of nifB (flanked by SphI sites) removed and replaced by a 1.3-kb kanamycin-resistance cartridge. The plasmid pHR18 was then transformed into A. vinelandii DJ1041, and the resulting kanamycin-resistant strain is designated A. vinelandii YM9A (AvYM9A), which produces a His-tagged NifEN (designated $\Delta nifB$ NifEN) and has a deletion/kanamycin resistance cartridge insertion in nifB. Construction of A. vinelandii strains producing A157S, A157G, and M156C Fe proteins has been published (27-29).

Cell Growth and Protein Purification. All A. vinelandii strains were grown in 180-liter batches in a 200-liter New Brunswick fermentor in Burke's minimal medium supplemented with 2 mM ammonium acetate. The growth rate was measured by cell density at 436 nm by using a Spectronic 20 Genesys (Spectronic, Westbury, NY). After the consumption of the ammonia, the cells were derepressed for 3 h, followed by harvesting using a flowthrough centrifugal harvester (Cepa, Lahr/Schwarzwald, Germany). The cell paste was washed with 50 mM Tris-HCl (pH 8.0). Published methods were used for the purification of all Fe proteins (28), WT MoFe protein (30), His-tagged $\Delta nifB$ MoFe protein (26), His-tagged NifEN, and His-tagged $\Delta nifB$ NifEN (23). The purification procedure of all His-tagged proteins was improved by (i) adding 10% glycerol to all buffers; (ii) limiting the purification process to <15 h; and (iii) performing cell rupture at <10,000 psi (1 psi = 6.89 kPa).

EPR Spectroscopy. All EPR samples were prepared in a Vacuum Atmospheres (Hawthorne, CA) dry box with an O_2 level of <4ppm. All dithionite-reduced samples were in 25 mM Tris·HCl (pH 8.0), 10% glycerol, and 2 mM Na₂S₂O₄. Indigo disulfonate (IDS)-oxidized samples were prepared as described (28). Samples were either used as they were or concentrated in a Centricon-30 (Amicon) in anaerobic centrifuge tubes outside of the dry box. All EPR spectra were recorded by using an ESP 300 E_z spectrophotometer (Bruker, Billerica, MA), interfaced with an ESR-9002 liquid helium continuous-flow cryostat (Oxford Instruments). Except for power- and temperature-dependent EPR experiments, all spectra were recorded at 13 K by using a microwave power of 50 mW, a gain of 5×10^4 , a modulation frequency of 100 kHz, and a modulation amplitude of 5 G. A microwave frequency of 9.43 GHz was used to record 10 scans for each sample. Spin quantitation of EPR signals was carried out under nonsaturating conditions as described (24).

FeMoco Maturation and Insertion Assays. The assays designed to determine the maximum possible FeMoco maturation contained (total volume, 0.8 ml) 25 mM Tris·HCl (pH 8.0), 20 mM Na₂S₂O₄, 0.5 mg of purified FeMoco-deficient $\Delta nifB$ MoFe protein from strain AvDJ1143 (26), 1.4 mg of Fe protein, 0.3 mM homocitrate, 0.3 mM sodium molybdate, 0.8 mM ATP, 1.6 mM MgCl₂, 10 mM creatine phosphate, and 8 units of creatine phosphokinase. The FeMoco maturation was initiated with the addition of 0.04–4 mg of isolated NifEN to the mixture mentioned above. Such reaction mixtures were incubated at 30°C for 30 min and stopped by the addition of 40 nmol of (NH₄)₂MoS₄ (31, 32), and the enzymatic activity was then determined as described (27, 30, 33). (NH₄)₂MoS₄ is known to block FeMoco insertion into the FeMoco site (33). Homocitrate lactone (Sigma)

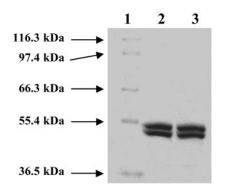


Fig. 1. Coomassie blue stained 7.5% SDS/PAGE of purified NifEN and $\Delta nifB$ NifEN. Lane 1, 5 μ g of protein standard; lane 2, 10 μ g of purified NifEN; lane 3, 10 μ g of purified $\Delta nifB$ NifEN.

containing an undefined mixture of stereochemical configurations was converted to the free acid as described in ref. 34.

Experiments designed to determine the minimum requirements for FeMoco maturation were carried out as described above, except that the reconstitution was initiated with the addition of 2 mg of isolated NifEN. FeMoco-maturation assays evaluating the function of Fe protein variants had the same composition as described above but contained 0.14 mg of Fe protein. The activities of the reconstitution assays were determined as described in ref. 30. FeMoco-maturation assays evaluating the function of various nucleotides contained, in the same buffer as described above, 0.5 mg of $\Delta nifB$ MoFe protein, 1.4 mg of Fe protein (\approx 33 μ M), 0.3 mM homocitrate, 0.3 mM sodium molybdate, 0.6-6.6 mM MgCl₂, and 0.3-3.3 mM of the following nucleotides: ATP, ADP, ATPyS [adenosine 5'-O-(3-thiotriphosphate)], or AMPPNP (5'-adenylylimidodiphosphate). The FeMoco maturation was initiated with the addition of 2 mg of isolated NifEN and stopped as described above. Subsequently, the activities of the maturation assays were determined as described in ref. 30, except that the ATP concentration was increased to a 25-fold molar excess relative to the concentrations of various nucleotides tested in the maturation assay.

The EPR sample of reconstituted $\Delta nifB$ MoFe protein contained all components of the FeMoco-maturation assay as described above, except that the concentrations of all ingredients were up-scaled by the same factor to yield a final MoFe protein concentration of 15 mg/ml. The EPR spectrum described in *Results* was corrected by subtracting the spectrum of an identical assay containing $\Delta nifB$ NifEN instead of NifEN.

The reaction products H_2 and C_2H_4 were analyzed as described in ref. 27, whereas ammonium was determined by using an HPLC fluorescence method (35).

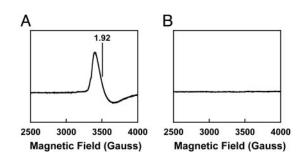


Fig. 2. EPR spectra of IDS-oxidized NifEN (A) and $\Delta nifB$ NifEN (B). The spectra were measured at a protein concentration of 10 mg/ml, as described in *Materials and Methods*. The g value is indicated.

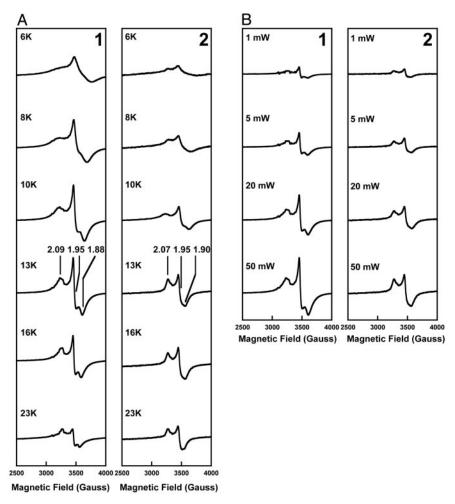


Fig. 3. Temperature (*A*) and power (*B*) dependence of the EPR signal exhibited by dithionite-reduced NifEN (1) and ∆*nifB* NifEN (2). The spectra were measured at a protein concentration of 10 mg/ml between 6 and 23 K as well as between 1 and 50 mW at 13 K, as described in *Materials and Methods*. The *g* values are indicated.

Metal Analysis. Mo (36) and Fe (37) were determined according to published protocols.

Results

By using the methods described in refs. 23–25, up to 750 mg of the His-tagged NifEN was purified from 250 g of cells of AvDJ1041, whereas up to 300 mg of the His-tagged $\Delta nifB$ NifEN was purified from 250 g of cells of AvYM9A. As shown in Fig. 1 (lanes 2 and 3) both NifEN and $\Delta nifB$ NifEN are composed of α (\approx 52 kDa)- and β (\approx 49 kDa)-subunits. The molecular masses of both proteins are \approx 200 kDa based on their elution profiles on gel filtration Sephacryl S-200 HR column (data not shown), indicating that both proteins are $\alpha_2\beta_2$ tetramers. In contrast to what has been reported (14), at the same protein concentration, purified NifEN is much darker in color than $\Delta nifB$ NifEN, indicating that NifEN contains an additional cluster (likely the FeMoco precursor), which is missing from $\Delta nifB$ NifEN because of the absence of nifB.[‡]

Consistent with the possible presence of an additional cluster on NifEN, the EPR spectroscopic features of NifEN clearly differ from those of $\Delta nifB$ NifEN. IDS-oxidized NifEN shows a g = 1.92 EPR signal (Fig. 2A), which is not present in the case of IDS-oxidized $\Delta nifB$ NifEN (Fig. 2B). This g = 1.92 signal is discernible only at temperatures of <30 K, with maximum intensity observed at 13 K. The presence of two [4Fe-4S] clusters (one at each NifE-NifN interface) has been reported for an earlier preparation of NifEN (13). These clusters exhibit a midpoint potential of -350 mV (13) and can be oxidized to an EPR silent state by dyes such as IDS ($E^{o'} = -125 \text{ mV}$). Therefore, it is likely that the $\Delta nifB$ NifEN in this study resembles the previously purified NifEN (13), which contains two permanent [4Fe-4S] clusters that become EPR silent upon IDS oxidation (Fig. 2*B*). Meanwhile, the g = 1.92 EPR signal of IDS-oxidized NifEN (Fig. 2A) likely originates from an additional cluster of unknown structure, which is bound to NifEN but absent from $\Delta nifB$ NifEN. This cluster may have been lost in the earlier preparation of NifEN (13), rendering it in a state identical or similar to the $\Delta nifB$ NifEN in this study.

At 10 K, the dithionite-reduced NifEN and $\Delta nifB$ NifEN show similar S = 1/2 EPR signals of slightly rhombic line shape in the g = 2 region (Fig. 3A, 1 and 2). However, at temperatures other than 10 K (and, in particular, temperatures of >10 K), the S =1/2 EPR signals exhibited by NifEN and $\Delta nifB$ NifEN appear to be different from each other (Fig. 3A, 1 and 2). The S = 1/2signal of NifEN adopts a more rhombic line shape at >10 K, with an additional feature between g values of 1.95 and 1.88 becoming

[‡]Consistent with previous studies (13), in the presence of dithionite, both NifEN and $\Delta nifB$ NifEN exhibit broad, nearly featureless UV-visible spectra. However, the intensity of the NifEN spectrum is significantly stronger than that of the $\Delta nifB$ NifEN at the same protein concentration (data not shown), indicating the possible presence of an additional metal cluster on NifEN.

Table 1. Metal contents of purified NifEN and $\Delta nifB$ NifEN

	Metal		
Protein	Fe	Мо	
MoFe protein	29.5 ± 2.0	1.8 ± 0.1	
NifEN	16.1 ± 2.4	< 0.01	
$\Delta nifB$ NifEN	8.5 ± 1.0	<0.01	

Data are expressed as mol of metal per mol of protein.

more evident (Fig. 3A, 1), whereas the S = 1/2 signal of $\Delta nifB$ NifEN maintains nearly the same line shape at >10 K, and the additional feature of NifEN at these temperatures is not apparent in this case (Fig. 3A, 2). The same difference between the line shapes of the S = 1/2 signals of NifEN and $\Delta nifB$ NifEN can be observed in spectra collected at 13 K and various microwave powers (ranging 1-50 mW) (Fig. 3B, 1 and 2). In contrast to the S = 1/2 signal of $\Delta nifB$ NifEN, which saturates slightly at 50 mW (Fig. 3B, 2), the signal of NifEN is not easily saturated within this power range (Fig. 3B, 1). These results again indicate a different cluster composition of NifEN than that of $\Delta nifB$ NifEN. Furthermore, our spin integration under nonsaturating conditions indicates the presence of approximately two spins per $\Delta nifB$ NifEN and approximately four spins per NifEN. Therefore, it is likely that in addition to the previously reported two [4Fe-4S] clusters that are present in both $\Delta nifB$ NifEN and NifEN, there is an additional metal cluster bound on NifEN. Consistent with this observation, the Fe content of NifEN (16.1 \pm 2.4 Fe atoms per molecule) is nearly twice as much as that of $\Delta nifB$ NifEN $(8.5 \pm 1.0 \text{ Fe atoms per molecule})$ (Table 1). If two [4Fe—4S] clusters (eight Fe atoms) are assigned to each molecule of $\Delta nifB$ NifEN or NifEN, then the additional Fe atoms per molecule of NifEN could be explained by the possible presence of a FeMoco precursor on NifEN.

It has been reported that the FeMoco-deficient, His-tagged $\Delta nifB$ MoFe protein forms fully active holo-MoFe protein upon the addition of isolated FeMoco (26, 38). Here, we show that the His-tagged $\Delta nifB$ MoFe protein [in crude extract of AvDJ1143 (data not shown) or purified form (Fig. 4)] can also be activated in a so-called FeMoco-maturation assay containing purified NifEN and other factors known to be required for FeMoco maturation, such as the Fe protein, MgATP, homocitrate, and Mo. Like the WT MoFe protein (Fig. 5A), $\Delta nifB$ MoFe protein activated by the maturation assay shows a well characterized S = 3/2 EPR signal (Fig. 5B) that arises from the FeMoco center of the protein (1). These results strongly point to the presence of a FeMoco and inserted subsequently into the $\Delta nifB$ MoFe protein, resulting in the formation of the active, holo-MoFe protein.

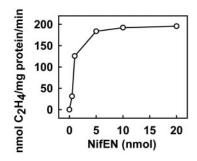


Fig. 4. FeMoco-maturation assay based on the reconstitution of purified FeMoco-deficient $\Delta nifB$ MoFe protein. The assays also contained Fe protein, homocitrate, molybdate, MgATP, and increasing amounts of NifEN, as described in *Materials and Methods*.

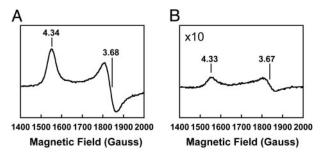


Fig. 5. EPR spectra of WT MoFe protein (*A*) and reconstituted $\Delta nifB$ MoFe protein (*B*) in the FeMoco-maturation assay. The samples (15 mg of protein per ml) were prepared and measured as described in *Materials and Methods*. The *g* values are indicated.

Upon the addition of increasing amounts of NifEN, a maximum activity of 217 nmol of C_2H_4 formation per mg of $\Delta nifB$ MoFe protein per min can be observed in the assay containing only purified proteins (Fig. 4). Activated MoFe protein in this assay shows not only C₂H₄-formation but also H₂-formation and N_2 -fixation activities (Table 2), which is consistent with the presence of a normal, catalytically active FeMoco in the protein. No activities can be observed if NifEN in the assay is replaced by equimolar amounts of $\Delta nifB$ NifEN or if any of the components, such as homocitrate, Mo, Fe protein, MgATP, NifEN, or $\Delta nifB$ MoFe protein, are omitted (Table 2). Several conclusions can be drawn from these results. (i) Accumulation of the FeMoco precursor on NifEN requires the presence of the *nifB* gene product. This result is consistent with the model that NifB is essential for FeMoco biosynthesis and represents the starting point of the pathway of FeMoco formation (10, 12, 13). (ii) The identified FeMoco precursor on NifEN does not contain homocitrate and Mo. Therefore, the addition of both components occurs either on the NifEN scaffold during the FeMocomaturation process or in a later step. The requirement of Mo for FeMoco maturation in NifEN is also consistent with the absence of Mo from purified NifEN (Table 1). (iii) Conversion of the FeMoco precursor to the mature FeMoco requires Fe protein and MgATP because activities can be observed only when both components are present (Table 2).

Table 3 shows that various Fe protein variants, which are able to bind MgATP but are defective in MgATP hydrolysis, are not able to support full FeMoco maturation in our FeMocomaturation assays. The extent of FeMoco maturation by these Fe protein variants is <10% compared with that by the WT Fe protein, suggesting that MgATP hydrolysis is likely required for FeMoco maturation. Consistent with this hypothesis, neither ADP nor nonhydrolyzable analogues of ATP, such as ATP γ S [adenosine 5'-O-(3-thiotriphosphate)] or AMPPNP (5'adenylylimidodiphosphate), are able to support FeMoco maturation in our FeMoco-maturation assays (Table 3).^{§¶}

Discussion

Consistent with what has been proposed (14), we identified a FeMoco precursor synthesized in the presence of nifB gene product through this study. This NifEN-bound precursor con-

[§]It needs to be noted that the conformational rearrangement of the Fe protein induced by such nonhydrolyzable analogues of ATP as ATP-5 [adenosine 5'-O-(3-thiotriphosphate)] cannot be distinguished from that induced by ATP based on the properties and/or behavior of the Fe protein in EPR spectroscopy and chelation experiments (data not shown).

¹The requirement of Fe protein concomitant with ATP hydrolysis suggests that there is/are electron(s) transferred to the emerging FeMoco. Consistent with this observation, preliminary data indicate the requirement of dithionite in the FeMoco-maturation assay (data not shown).

Table 2. Determination of factors required for FeMoco maturation

	Activities*			
Assay condition	C_2H_4 formation under C_2H_2/Ar	H ₂ formation under Ar	NH_3 formation under N_2	H_2 formation under N_2
Complete [†]	191 ± 26 (100)	316 ± 11 (100)	96 ± 6 (100)	65 ± 13 (100)
Complete plus (NH ₄) ₂ MoS ₄ [‡]	0 (0)	2 ± 0.04 (<1)	0 (0)	2 ± 0.2 (3)
Complete minus MgATP	0 (0)	0 (0)	0 (0)	0 (0)
Complete minus Fe protein	0 (0)	0 (0)	0 (0)	0 (0)
Complete minus homocitrate	0 (0)	0 (0)	0 (0)	0 (0)
Complete minus molybdate [§]	10 ± 1 (5)	15 ± 3 (5)	0 (0)	0 (0)
Complete minus NifEN	0 (0)	0 (0)	0 (0)	0 (0)
Complete minus $\Delta nifB$ MoFe protein	0 (0)	0 (0)	0 (0)	0 (0)
Complete minus NifEN, plus ∆ <i>nifB</i> NifEN [¶]	0 (0)	1 ± 0.1 (<1)	0 (0)	0 (0)
$\Delta nifB$ MoFe protein alone	0 (0)	0 (0)	0 (0)	0 (0)
NifEN alone	0 (0)	0 (0)	0 (0)	0 (0)
$\Delta nifB$ NifEN alone	0 (0)	1 ± 0.1 (<1)	0 (0)	0 (0)

Data are expressed as nmol per min per mg of protein. Percentages are given in parentheses.

*The lower detection limits were 0.01, 0.02, 0.001, and 0.02 nmol per min per mg of protein for C_2H_4 formation under C_2H_2/Ar , H_2 formation under Ar, NH₃ formation under N₂, and H₂ formation under N₂, respectively.

[†]The complete assay contains purified NifEN, purified $\Delta nifB$ MoFe protein, purified Fe protein, molybdate, homocitrate, and MgATP at concentrations described in *Materials and Methods*.

[‡]Insertion of FeMoco into $\Delta nifB$ MoFe protein was inhibited by the addition of $(NH_4)_2MoS_4$ at the beginning of the experiment, as described in *Materials and Methods*.

[§]The minor activities were likely caused by molybdenum contamination in the assays.

[¶]Assay contained the same components as described in \dagger , except that NifEN was replaced by equimolar amounts of $\Delta nifB$ NifEN.

tains Fe as the only metal, and Mo and homocitrate are added while the cluster is still bound to the NifEN complex or at a later step. The addition of Mo and homocitrate to the precursor and the final maturation of the holo-MoFe protein require only the participation of Fe protein and MgATP, leading to the following

Table 3. Effect of Fe protein variants and various nucleotides on FeMoco maturation

Factors		Activities*
	Fe protein	
No Fe protein		0 (0)
WT Fe protein		196 ± 6 (100)
A157S Fe protein ⁺⁺		18 ± 4 (9)
A157G Fe protein ^{†§}		11 ± 1 (6)
M156C Fe protein ⁺¹		6 ± 2 (3)
	Nucleotide	
No nucleotide		0 (0)
ATP		186 ± 10 (100)
ADP**		0 (0)
ATPγS**		0 (0)
AMPPNP**		0 (0)

Data are expressed as nmol of C_2H_4 evolution per min per mg of protein. Percentages are given in parentheses. ATP γ S, adenosine 5'-O-(3-thiotriphosphate; AMPPNP, 5'-adenylylimidodiphosphate.

*The lower detection limit was 0.01 nmol of C_2H_4 evolution per min per mg of protein.

⁺All Fe protein variants are able to bind MgATP (27–29).

[‡]A157S Fe protein is unable to undergo a MgATP-induced conformational change and does not support MgATP hydrolysis (27).

[§]A157G Fe protein undergoes a delayed conformational change upon MgATP binding, resulting in a reduced substrate-reduction activity (28).

¹M156C Fe protein undergoes a MgATP-induced conformational change that differs from WT Fe protein, resulting in the loss of substrate-reduction activity (29).

Identical results have been obtained by using 10-, 50-, and 100-fold molar excess of nucleotides relative to Fe protein in the FeMoco-maturation assay.

**Note that, with the addition of excess MgATP as described in *Materials and Methods*, these nucleotides do not inhibit substrate-reduction activity of the WT MoFe protein.

conclusions. (*i*) FeMoco carrier protein(s) such as γ (18, 19) is not essential for FeMoco maturation; and (*ii*) Fe protein facilitates Mo and homocitrate insertion into the FeMoco, likely upon MgATP hydolysis. Interestingly, it has been reported that Mo accumulated on the Fe protein during FeMoco biosynthesis (39), an observation in line with our results. Also, the first published x-ray structure of the Fe protein of *A. vinelandii* contained ADP bound in an unusual location across the subunit– subunit interface and an adjacent Mo located in a position that could correspond to the γ phosphate of ATP (40). It has been speculated that the binding mode in this structure could be involved in the initial entry of the nucleotide into the Fe protein (40). In light of our results, it is also possible that this ADP/Mo binding mode is related to the process of FeMoco maturation.

Although the dual requirement of Fe protein and MgATP for FeMoco maturation has been well documented in the past years (16, 17, 41, 42), previous studies suggest that Fe protein only needs to bind but does not need to hydrolyze MgATP to carry out its function in this process (27-29). Ref. 43 showed that a truncated form of Fe protein was unable to support substrate reduction but was active in FeMoco biosynthesis. This effect could be explained in two ways. (i) Catalytically active Fe protein capable of MgATP hydrolysis may not be required for the FeMoco biosynthesis. (ii) The truncated Fe protein may have lost its ability to interact with the MoFe protein during substrate reduction yet retained its capacity to hydrolyze MgATP, a feature required for FeMoco biosynthesis. Based on our observations that (i) the Fe protein variants defective in MgATP hydrolysis are inactive in FeMoco maturation (Table 3), (ii) nonhydrolyzable analogues of ATP are unable to support FeMoco maturation (Table 3), and (iii) the Fe protein is the only known nucleotide binding protein in our maturation assay, it appears to be likely that the Fe protein carries out its function in FeMoco biosynthesis through MgATP hydrolysis, and the effects of the truncated Fe protein form could be accounted for by the second explanation.

The FeMoco-maturation assay developed in this study is an improvement of earlier FeMoco biosynthesis assays (27-29, 44-46) in that (*i*) it allows observation of much higher specific

activity of reconstituted MoFe protein (≈100-fold) than the reported values of earlier FeMoco biosynthesis assays; (ii) it contains all proteins in purified forms and, therefore, avoids the complication caused by other factors in crude extracts often used in earlier FeMoco biosynthesis assays; and (iii) it uses $\Delta nifB$ MoFe protein as the target protein for FeMoco insertion instead of $\Delta nifH$ -type MoFe protein used in previous FeMoco biosynthesis assays. Recently, we reported that $\Delta nifH$ MoFe protein was not only FeMoco-deficient but also contained a P-cluster precursor (23, 25). Therefore, previous FeMoco biosynthesis assays based on $\Delta nifH$ -type MoFe protein were in fact combined assays of FeMoco and P-cluster maturation and consequently, it was difficult to interpret the results accurately. The FeMocomaturation assay in this study uses $\Delta nifB$ MoFe protein (which has fully assembled P-clusters) as the "receptor" for FeMoco synthesized during the assay, and as a result, the sole effect of FeMoco maturation can be observed without interference of P-cluster formation.

Note that despite the fact that our FeMoco-maturation assay yields much higher activity of reconstituted $\Delta nifB$ MoFe protein

- 1. Burgess, B. K. & Lowe, D. J. (1996) Chem. Rev. 96, 2983-3011.
- 2. Howard, J. B. & Rees, D. C. (1996) Chem. Rev. 96, 2965-2982.
- 3. Smith, B. E. (1999) Adv. Inorg. Chem. 47, 159-218.
- 4. Rees, D. C. & Howard, J. B. (2000) Curr. Opin. Chem. Biol. 4, 559-566.
- 5. Christiansen, J., Dean, D. R. & Seefeldt, L. C. (2001) Annu. Rev. Plant Physiol.
- Plant Mol. Biol. 52, 269–295.
 G. Igarashi, R. Y. & Seefeldt, L. C. (2003) Crit. Rev. Biochem. Mol. Biol. 38, 351–384.
- 7. Seefeldt, L. C., Dance, I. G. & Dean, D. R. (2004) *Biochemistry* 43, 1401–1409.
- Peters, J. W., Stowell, M. H., Soltis, S. M., Finnegan, M. G., Johnson, M. K. & Rees, D. C. (1997) *Biochemistry* 36, 1181–1187.
- Einsle, O., Tezcan, F. A., Andrade, S. L. A., Schmid, B., Yoshida, M., Howard, J. B. & Rees, D. C. (2002) *Science* 297, 1696–1700.
- Dos-Santos, P. C., Dean D. R., Hu, Y. & Ribbe, M. W. (2004) Chem. Rev. 104, 1159–1174.
- Shah, V. K., Allen, J. R., Spangler, N. J. & Ludden, P. W. (1994) J. Biol. Chem. 269, 1154–1158.
- Allen, R. M., Chatterjee, R., Ludden, P. W. & Shah, V. K. (1995) J. Biol. Chem. 270, 26890–26896.
- Goodwin, P. J., Agar, J. N., Roll, J. T., Roberts, G. P., Johnson, M. K. & Dean, D. R. (1998) *Biochemistry* 37, 10420–10428.
- Roll, J. T., Shah, V. K., Dean, D. R. & Roberts, G. P. (1995) J. Biol. Chem. 270, 4432–4437.
- Allen, R. M., Chatterjee, R. Ludden, P. W. & Shah, V. K. (1996) J. Biol. Chem. 271, 4256–4260.
- Chatterjee, R., Allen, R. M., Shah, V. K. & Ludden, P. W. (1994) J. Bacteriol. 176, 2747–2750.
- Robinson, A. C., Dean, D. R. & Burgess, B. K. (1987) J. Biol. Chem. 262, 14327–14332.
- Homer, M. J., Dean, D. R. & Roberts, G. P. (1995) J. Biol. Chem. 270, 24745–24752.
- Rubio, L. M., Rangaraj, P., Homer, M. J., Roberts, G. P. & Ludden, P. W. (2002) J. Biol. Chem. 277, 14299–14305.
- 20. Roberts, G. P. & Brill, W. J. (1980) J. Biol. Chem. 144, 210-216.
- Brigle, K. E., Weiss, M. C., Newton, W. E. & Dean, D. R. (1987) J. Bacteriol. 169, 1547–1553.
- Paustian, T. D., Shah, V. K. & Roberts, G. P. (1989) Proc. Natl. Acad. Sci. USA 86, 6082–6086.
- Ribbe, M. W., Hu, Y., Guo, M., Schmid, B. & Burgess, B. K. (2002) J. Biol. Chem. 277, 23469–23476.
- Hu, Y., Fay, A. W., Dos Santos, P. C., Naderi, F. & Ribbe, M. W. (2004) J. Biol. Chem., 279, 54963–54971.

than those of the previously reported FeMoco biosynthesis assays, the maximum activity of reconstitution is only $\approx 10\%$ of that of the WT holo-MoFe protein (23, 33, 44). This observation indicates that additional factors, albeit not essential for MoFe protein maturation, are required to optimize this process *in vivo*. Such factors could be GroEL, NifY, or γ , all of which have been implicated in the process of MoFe protein assembly (18, 19, 33, 47, 48).

In summary, we were able to unravel some of the baffling issues regarding FeMoco maturation in this work. However, other key questions such as the structure of the FeMoco precursor on NifEN or the mechanism of Mo insertion into the FeMoco remain unanswered. Also, the fashion by which other factors, such as GroEL, NifY, or γ , facilitate FeMoco insertion and final holo-MoFe protein maturation awaits further investigation.

We thank Prof. Dennis Dean (Virginia Polytechnic Institute and State University, Blacksburg) for kindly providing the *A. vinelandii* strains DJ1143 and DJ1041. This work was supported by National Institutes of Health Grant GM-67626 (to M.W.R.).

- Corbett, M. C., Hu, Y., Naderi, F., Ribbe, M. W., Hedman, B. & Hodgson, K. O. (2004) J. Biol. Chem. 279, 28276–28282.
- Christiansen, J., Goodwin, P. J., Lanzilotta, W. N., Seefeldt, L. C. & Dean, D.R. (1998) *Biochemistry* 37, 12611–12623.
- 27. Gavini, N. & Burgess, B. K. (1992) J. Biol. Chem. 267, 21179-21186.
- 28. Bursey, E. H. & Burgess, B. K. (1998) J. Biol. Chem. 273, 16927-16934.
- 29. Bursey, E. H. & Burgess, B. K. (1998) J. Biol. Chem. 273, 29678-29685.
- Burgess, B. K., Jacobs, D. B. & Stiefel, E. I. (1980) Biochim. Biophys. Acta 614, 196–209.
- Rangaraj, P., Shah, V. K. & Ludden, P. W. (1997) Proc. Natl. Acad. Sci. USA 94, 11250–11255.
- Shah, V. K., Ugalde, R. A., Imperial, J. & Brill, W. J. (1985) J. Biol. Chem. 260, 3891–3894.
- 33. Ribbe, M. W. & Burgess, B. K. (2001) Proc. Natl. Acad. Sci. USA 98, 5521-5525.
- 34. Madden, M. S., Paustian, T. D., Ludden, P. W. & Shah, V. K. (1991) J. Bacteriol. 173, 5403–5405.
- 35. Corbin, J. L. (1984) Appl. Environ. Microbiol. 47, 1027-1030.
- 36. Clark, L. J. & Axley, J. H. (1955) Anal. Biochem. 27, 2000.
- 37. Van de Bogart, M. & Beinert, H. (1967) Anal. Biochem. 20, 325-334.
- Schmid, B., Ribbe, M. W., Einsle, O., Yoshida, M., Thomas, L. M., Dean, D. R., Rees, D. C. & Burgess, B. K. (2002) *Science* 296, 352–356.
- 39. Rangaraj, P. & Ludden, P. W. (2002) J. Biol. Chem. 277, 40106-40111.
- Georgiadis, M. M., Komiya, H., Chakrabarti, P., Woo, D., Kornuc, J. J. & Rees, D. C. (1992) *Science* 257, 1653–1659.
- Robinson, A. C., Chun, T. W., Li, J.-G. & Burgess, B. K. (1989) J. Biol. Chem. 264, 10088–10095.
- Rangaraj, P., Ryle, M. J., Lanzilotta, W. N., Ludden, P. W. & Shah, V. K. (1999) J. Bio. Chem. 274, 19778–19784.
- 43. Filler, W. A., Kemp, R. M., NG, J. C., Hawkes, T. R., Dixon, R. A. & Smith, B. E. (1986) *Eur. J. Biochem.* 160, 371–377.
- 44. Ribbe, M. W., Bursey, E. H. & Burgess, B. K. (2000) J. Biol. Chem. 275, 17631–17638.
- Shah, V. K., Imperial, J., Ugalde, R. A., Ludden, P. W. & Brill, W. J. (1986) Proc. Natl. Acad. Sci. USA 83, 1636–1640.
- Imperial, J., Shah, V. K., Ugalde, R. A., Ludden, P. W. & Brill, W. J. (1987) J. Bacteriol. 169, 1784–1786.
- Rangaraj, P., Ruttimann-Johnson, C., Shah, V. K. & Ludden, P. W. (2001) J. Biol. Chem. 276, 15968–15974.
- Shah, V. K., Rangaraj, P., Chatterjee, R., Allen, R. M., Roll, J. T., Roberts, G. P. & Ludden, P. W. (1999) *J. Bacteriol.* 181, 2797–2801.