



Structural and biochemical analyses reveal insights into covalent flavinylation of the *Escherichia coli* Complex II homolog quinol:fumarate reductase

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The *Escherichia coli* Complex II homolog quinol:fumarate reductase (QFR, FrdABCD) catalyzes the interconversion of fumarate and succinate at a covalently attached FAD within the FrdA subunit. The SdhE assembly factor enhances covalent flavinylation of Complex II homologs, but the mechanisms underlying the covalent attachment of FAD remain to be fully elucidated. Here, we explored the mechanisms of covalent flavinylation of the *E. coli* QFR FrdA subunit. Using a Δ *sdhE* *E. coli* strain, we show that the requirement for the assembly factor depends on the cellular redox environment. We next identified residues important for the covalent attachment and selected the FrdA^{E245} residue, which contributes to proton shuttling during fumarate reduction, for detailed biophysical and structural characterization. We found that QFR complexes containing FrdA^{E245Q} have a structure similar to that of the WT flavoprotein, but lack detectable substrate binding and turnover. In the context of the isolated FrdA subunit, the anticipated assembly intermediate during covalent flavinylation, FrdA^{E245} variants had stability similar to that of WT FrdA, contained noncovalent FAD, and displayed a reduced capacity to interact with SdhE. However, small-angle X-ray scattering (SAXS) analysis of WT FrdA cross-linked to SdhE suggested that the FrdA^{E245} residue is unlikely to contribute directly to the FrdA-SdhE protein-protein interface. We also found that no auxiliary factor is absolutely required for flavinylation, indicating that the covalent flavinylation is autocatalytic. We propose that multiple factors, including the SdhE assembly factor and bound dicarboxylates, stimulate covalent flavinylation by preorganizing the active site to stabilize the quinone-methide intermediate.

The isoalloxazine-conjugated ring system of enzyme-associated flavins can support oxidoreduction and group transfer reactions. Recent surveys of the literature identify that this flavin cofactor is covalently attached to ~10% of flavoenzymes (1, 2). The covalent bond has been proposed to have many roles including preventing the loss of the cofactor, increasing the stability of the flavoenzyme, or raising the redox-potential of the cofactor (3–5).

One model system that has been used to investigate covalent flavin attachment is the integral-membrane heterotetramer quinol:fumarate reductase (QFR).⁴ QFR is a member of the Complex II superfamily and is composed of FrdABCD subunits. The FrdA and FrdB subunits are soluble and attached to the membrane through interaction with the membrane-embedded FrdC and FrdD subunits. In *Escherichia coli* QFR, the flavin is covalently attached to the FrdA subunit via an α linkage to the *N*- ϵ of His⁴⁴. This places the flavin isoalloxazine group at the interface of two domains, termed the flavin-binding domain and the capping domain (6). These domains can move with respect to each other, which may allow the catalytic subunit to close over the substrate-bound active site and protect the reaction intermediates from solvent (7).

Physiologically, QFR catalyzes the reduction of fumarate ($k_{\text{cat}} \sim 250/\text{s}$) at this flavin-based active site during anaerobic respiration with fumarate as the terminal electron acceptor (8). In this process, the two electrons for fumarate reduction derive from the oxidation of menaquinol in the membrane and the two protons are likely transferred from solvent via a proton shuttle pathway consisting of the FrdA^{E245}, FrdA^{R248}, and FrdA^{R287} side chains located on the capping domain (9–12). QFR can also catalyze the reverse reaction, succinate oxidation, albeit with slower kinetics ($k_{\text{cat}} \sim 30/\text{s}$) and poorer catalytic efficiency (8). Other Complex II superfamily members, including bacterial succinate:quinone oxidoreductase (SQR, SdhABCD) and mammalian mitochondrial Complex II (also termed SdhABCD), kinetically favor the succinate oxidation reaction

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The atomic coordinates and structure factors (code 5VPN) have been deposited in the Protein Data Bank (<http://www.pdb.org/>).

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⁴ The abbreviations used are: QFR, quinol:fumarate reductase; SQR, succinate:ubiquinone oxidoreductase; SAXS, small angle X-ray scattering; FAD, flavin adenine dinucleotide; pBpF, *para*-benzoyl phenylalanine; PDB, Protein Data Bank; BisTris, 2-[bis(2-hydroxyethyl)amino]-2-(hydroxymethyl)propane-1,3-diol.

Covalent flavinylation of Complex II

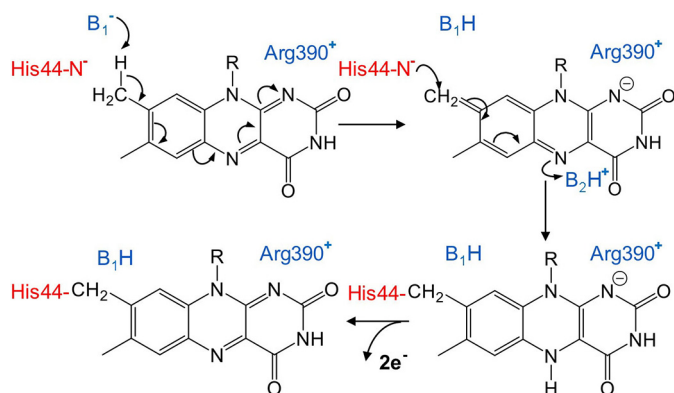


Figure 1. Proposed chemical mechanism of covalent flavinylation via a quinone-methide intermediate. Catalytic bases are shown in *blue*. The identities of the residues involved in proton abstraction and intermediate stabilization in QFR have not been unambiguously determined, but likely possibilities are shown here. The FrdA^{H44} side chain (*red*) is the nucleophile that forms a covalent bond, and may also be the base for proton abstraction.

over the fumarate reduction reaction. As in QFR, the FAD cofactor is attached via a 8α -N3-histidyl linkage in the SdhA subunit (13); this linkage forms at SdhA His⁴⁵ in the *E. coli* SQR.

An early study of covalent flavinylation performed mutagenesis of QFR-FrdA^{H44}, which forms the histidyl linkage to FAD (3). When FrdA^{H44} was substituted with serine, cysteine, or tyrosine the intact QFR (FrdABCD) enzyme retained tightly associated non-covalent flavin at stoichiometric levels. Furthermore, kinetic characterization identified that QFR-FrdA^{H44S}, QFR-FrdA^{H44C}, and QFR-FrdA^{H44Y} lacked detectable succinate oxidation activity but retained fumarate reduction at 25–50% of wild-type activity. This result suggested that the increase in flavin potential imparted by the covalent linkage is important for the succinate oxidation reaction catalyzed by Complex II enzymes. From this, it was proposed that one reason for the inability of the variants to oxidize succinate was because of a lowered redox potential of the non-covalent FAD (3). This was later supported by findings where similar site-directed variants of *E. coli* SQR result in an ~ 90 mV reduction in the $E_{m,7}$ of the non-covalent flavin compared with wild-type enzyme (14) and in *E. coli* QFR where there was a -100 mV drop in redox potential of the FAD compared with the wild-type enzyme (15).

A number of years ago, Walsh (16) proposed that the mechanism of covalent flavin bond formation was fully autocatalytic and involved a quinone-methide intermediate (Fig. 1). This mechanism only requires the FAD and a specific site of attachment; electrons and protons can be shuttled to this location to activate both the isoalloxazine ring and the amino acid where the covalent bond forms. More recently, the field has focused on the role of auxiliary proteins in covalent flavinylation. One protein of intense interest, termed SdhE (or YgfY) in bacteria, Sdh5 in yeast, and SdhAF2 in humans (17, 18), is a ~ 10 -kDa protein conserved throughout all kingdoms (19) that can support covalent flavinylation of human SdhA in an *in vitro* reconstituted system (20). Using bacterial QFR and SQR as models, the SdhE assembly factor has been shown to bind directly to the FrdA or SdhA subunits prior to assembly into the intact complex (21). The interaction involves a surface of SdhE/Sdh5/SdhAF2 containing a conserved RGXXE motif (22, 23)

and a surface of FrdA/SdhA that interacts with the FrdB/SdhB subunit in the assembled heterotetramer (21). Moreover, the FrdA/SdhA subunit must be pre-associated with FAD for robust interaction with the SdhE assembly factor (20, 21).

Despite these advances, how SdhE/Sdh5/SdhAF2 promotes covalent flavinylation is not clear. We previously proposed a hypothesis where closure of the flavoprotein-capping domain over bound dicarboxylate (the term dicarboxylate is used in the general sense to refer to molecules containing two carboxylates, such as the substrate fumarate, the product succinate, the natural inhibitors oxaloacetate and malonate, etc.) at the active site promotes alignment of the FrdA/SdhA active site side chains near the bound flavin (21). This scenario could allow the FrdA/SdhA protein to adopt a conformation that stabilizes the quinone-methide tautomer at the C(8) α position on the isoalloxazine ring of the flavin. Tight closure of the flavoprotein-capping domain would be followed by nucleophilic attack by the nearby histidyl residue to form the covalent bond at the C(8) position. An alternative proposal could be that residues of SdhE/Sdh5/SdhAF2 directly participate in the chemistry of covalent bond formation. This alternative is not favored because of recent evidence that the SdhAF2 assembly factor enhances covalent flavinylation, but is not essential (24, 25). For both scenarios, the observation that SdhE and FrdB bind to the same surface of FrdA indicates that the flavoprotein-assembly factor complex must disassociate prior to assembly of FrdA/SdhA into the intact QFR/SQR heterotetramer (21).

In this study, we demonstrate that in *E. coli* QFR and SQR, the influence of deleting SdhE on covalent flavinylation depends upon the redox environment of the cell. To investigate SdhE-independent mechanisms of covalent flavinylation, we next identified variants of the QFR FrdA subunit with levels of covalent flavinylation below those observed in the Δ *sdhE* strains. We selected the FrdA^{E245Q} variant for in-depth study; this side chain has previously been suggested as a part of the proton delivery pathway during fumarate reduction (7, 11), but has not been suggested as participating in covalent flavinylation. We identify that the QFR-FrdA^{E245Q} variant retains a structure similar to the wild-type enzyme, but exhibits impaired dicarboxylate binding and catalytic activity. Because covalent flavinylation is normally anticipated to occur in the isolated FrdA subunit prior to assembly into the QFR (FrdABCD) complex, we additionally investigated the effects of this variant in the context of isolated FrdA subunit. We found similar stability of wild-type FrdA and FrdA^{E245} variants, but impaired interactions with the SdhE assembly factor. SAXS data indicate that this residue is unlikely to interact directly with SdhE. Because no single auxiliary factor is essential for the formation of the covalent flavin linkage, the most likely conclusion from these data is that covalent flavinylation is autocatalytic. One interpretation of much of the data is that factors stimulating covalent flavinylation would pre-organize the active site, suggesting that the autocatalytically formed quinone-methide tautomer originally proposed by Walsh (16) remains the most likely pathway to covalent flavinylation.

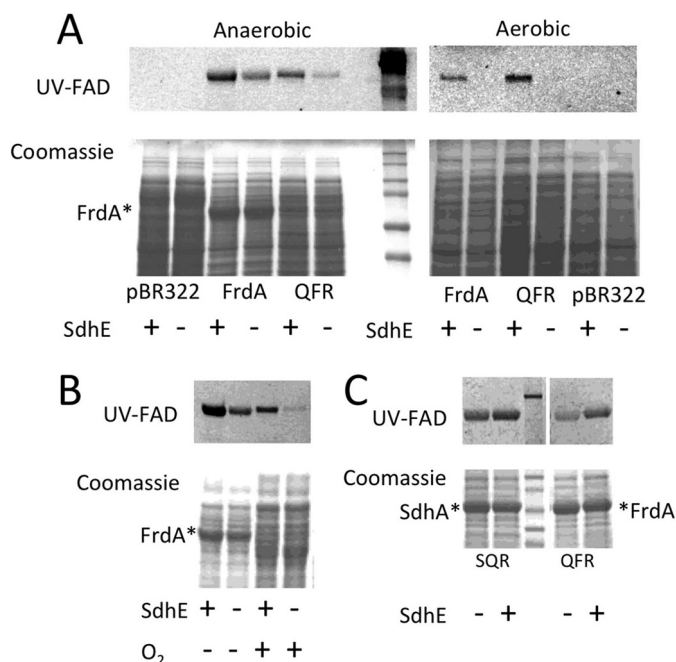


Figure 2. Effect of aerobic and anaerobic cell growth on flavinylation.

Expression of FrdA from pFrdA and pH3 plasmids was performed in *E. coli* RP-2 (Δ frdABCD, Δ sdhCDAB) and RP-3 (Δ frdABCD, Δ sdhCDAB, Δ sdhE) under aerobic and anaerobic conditions. **A**, SDS-PAGE analysis of whole cell samples (20 μ g of protein per lane) shows attenuated, but not abrogated covalent flavinylation under anaerobic conditions, but substantial loss of covalent flavinylation under aerobic growth conditions. The *top panels* indicate UV fluorescence from the covalently-bound FAD cofactor and the *bottom panels* are Coomassie Blue G staining of the cell extracts. *M_r* markers are shown on the *right side* of the gel for the anaerobically grown cells. Note, some of the pre-stained MW markers (Bio-Rad, number 1610374) show inherent fluorescence when exposed to a UV light. The *asterisk* indicates the position of FrdA in the Coomassie-stained samples. **B**, analysis of isolated membrane fractions from anaerobically and aerobically grown cells. Note, there is reduced expression of FrdA in the aerobically grown cells because expression is driven by the natural FRD (*i.e.* anaerobic) promoter. Nevertheless, there is reduced covalent flavinylation in the absence of SdhE for cells grown aerobically as compared with anaerobically. **C**, effect of SdhE on flavinylation in anaerobically expressed SQR and QFR. Comparison of membrane fractions isolated from RP-3 (Δ sdhE) and RP-2 (+sdhE) cells indicates that SdhE is not essential for covalent flavinylation of either SdhA or FrdA. The *top panel* shows UV fluorescence of the covalent FAD. The *middle lane* of both panels shows prestained *M_r* markers. The *bottom panel* shows Coomassie Blue G staining of proteins in the membrane fraction. The *asterisk* indicates the location of SdhA or FrdA, respectively. Note, both SQR and FRD are expressed anaerobically from the FRD promoter as previously described (8).

Results

SdhE influence on covalent flavin attachment

Clinical studies and yeast genetics were instrumental in the discovery of the assembly factor SdhAF2 (17, 26), and homologs have been identified in yeast (Sdh5 (17)) and bacteria (SdhE (18)). We focused on the *E. coli* system and used Δ sdhE strains to evaluate the impact of SdhE on covalent flavinylation of QFR and SQR complexes. We find that anaerobic expression of QFR (FrdABCD) in Δ sdhE strains results in retention of covalent flavinylation, albeit at a reduced level compared with when SdhE is present (Fig. 2, **A** and **B**). In contrast, QFR shows significant reduction of covalent FAD when expressed aerobically (Fig. 2, **A** and **B**). We then tested to see whether the SQR homolog (SdhABCD), which is kinetically optimized to function under aerobic conditions, would retain covalent FAD when expressed anaerobically (Fig. 2**C**). Anaerobically, we observe levels of SdhA flavinylation similar to those found for FrdA when

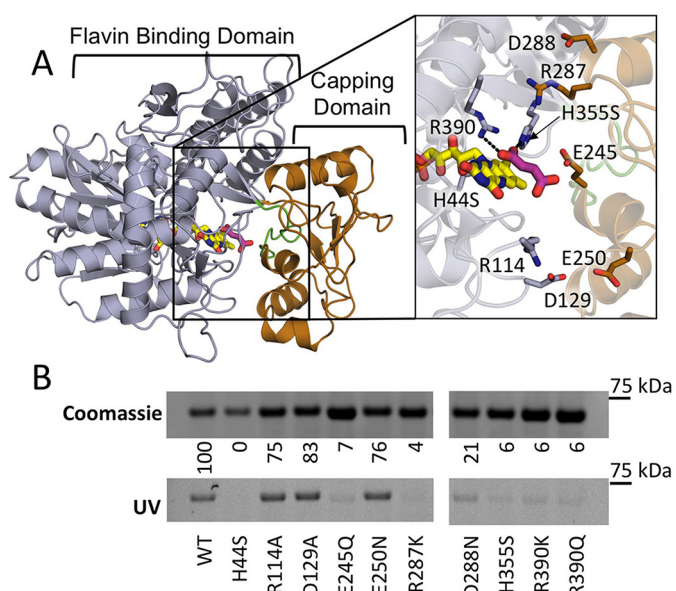


Figure 3. Covalent flavinylation of QFR mutants. **A**, location of substitutions in the FrdA subunit of QFR on the flavin-binding (gray) and -capping domains (copper) (PDB entry 3P4P (38)). Flavin (yellow) and fumarate (pink) ligands are modeled and the active site loop flanking FrdA^{E245E} is colored green (residues 234–244 in QFR). Zoomed window shows a close up of the location of the residues that were altered to evaluate their effect on covalent flavinylation. They are colored according to their location on either the flavin-binding or -capping domain. **B**, quantitation of covalent FAD levels in wild-type and variant QFR complexes. Coomassie-stained SDS-PAGE gels are normalized to the total amount of protein to assess differences in expression levels of variant QFR complexes. The same gel was illuminated with UV light to measure covalent flavinylation by flavin fluorescence. The ratio between the Coomassie signal and the flavin fluorescence was quantified. Wild-type covalent flavinylation was set to 100%, whereas QFR-FrdA^{H44S} was set to 0%, and these were used as standards to quantify the percentage of covalent FAD in the remaining variants.

the full enzyme complexes are expressed. This suggests that the enhancement of flavinylation imparted by SdhE under aerobic conditions may be partially compensated for by anaerobiosis, when fumarate is likely to bind to the flavoprotein.

QFR Mutants deficient in covalent flavinylation

We generated a library of site-directed variants within the FrdA subunit of the *E. coli* QFR complex (Fig. 3**A**). We selected residues for mutagenesis because they exhibited one or more of the following properties: (i) they directly interact with bound substrate (FrdA^{R287}, FrdA^{H355S}, and FrdA^{R390} (6)); (ii) they are either previously proposed to contribute to the shuttling of protons during catalysis (FrdA^{E245} and FrdA^{R287} (10, 11)) or could contribute to the stability of the substrate-binding site and effect the properties of other catalytic residues (FrdA^{D288}); (iii) they may stabilize the interdomain orientations or motions important for catalysis (FrdA^{E245} and FrdA^{E250} (11)); (iv) they are in regions that may be allosterically connected to the active site (FrdA^{E250}); or (v) they are near the proposed binding surface for the assembly factor SdhE (FrdA^{D129} (21)). QFR containing wild-type FrdA or FrdA^{R114}, which is a charged residue near the FAD but without any proposed mechanistic role were used as positive controls; the previously described FrdA^{H44S} variant of QFR, which alters the normal histidyl linkage to the FAD and retains stoichiometrically bound non-covalent FAD, was used as a negative control (3).

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We evaluated the effect of these substitutions on the levels of covalent flavinylation in the intact FrdABCD enzyme produced under anaerobic conditions. To do this, we separated QFR-enriched membranes on SDS-PAGE and monitored the level of FAD fluorescence associated with the FrdA subunit (Fig. 3B). Wild-type QFR, the QFR-FrdA^{R114} variant, and the QFR-FrdA^{H44S} variant were used as comparators. Five of our designed variants (FrdA^{E245Q}, FrdA^{R287K}, FrdA^{H355S}, FrdA^{R390K}, and FrdA^{R390Q}) had almost a complete absence of covalent flavinylation, consistent with studies of the SQR protein (14). The FrdA^{D288N} variant showed minimal residual fluorescence, suggesting covalent flavinylation was severely compromised. Each of these variants had levels of covalent flavin attachment below that observed for QFR expressed in the Δ *sdhE* strain that was similarly grown anaerobically (Fig. 2, A and B). This suggests that these side chains have a role in flavinylation that is independent of assembly factor function. It should be noted that with the exception of FrdA^{D288}, each of the amino acids associated with loss of flavinylation is known to be involved in substrate binding and/or the catalytic mechanism of fumarate-succinate interconversion (6, 12, 27). Accordingly, one possibility is that these side chains also directly participate in the mechanism of flavinylation.

Of these residues, we selected QFR containing the FrdA^{E245Q} variant for more detailed study. FrdA^{E245} exhibits loss of detectable flavinylation but has not previously been implicated in the covalent flavinylation process. It has been proposed as a proton shuttle during fumarate reduction within the context of both the *E. coli* QFR (7) and the *Shewanella* soluble fumarate reductase homolog (11), but is not directly involved in substrate binding. This residue is adjacent to FrdA^{T244}, a residue implicated in the catalytic efficiency of the enzyme (7) and associated with stoichiometric covalent FAD. FrdA^{T244} is part of an 11-amino acid loop (FrdA^{T234} to FrdA^{T244} in *E. coli* QFR) that connects the flavin-binding domain and the capping domain of the flavoprotein (Fig. 3A). The location of the FrdA^{E245} side chain at the domain interface of the flavoprotein suggests that this side chain could have an additional role in stabilizing the orientations of the flavin-binding and capping domains of FrdA.

Catalytic activity of QFR complexes lacking covalent FAD

We began by investigating how the FrdA^{E245} mutation affected function within the context of the assembled QFR complex. Previous findings have shown that the QFR-FrdA^{H44S} variant lacks covalent FAD, however, it can bind stoichiometric amounts of non-covalent FAD (3). Catalytically, the QFR-FrdA^{H44S} variant retained ~50% of its ability to reduce fumarate, whereas it only retains 1% of its ability to oxidize succinate (3). In the current study we found that the QFR-FrdA^{E245Q} variant lacked detectable catalytic activity for both fumarate reduction and succinate oxidation (data not shown). This finding is consistent with the previously reported reduction of detectable fumarate reduction activity to <1% of wild-type levels upon mutation of the analogous residue in the soluble FrdA homolog from *Shewanella frigidimarina*, which naturally contains non-covalent FAD as its redox cofactor (22). We antici-

pate that the FAD cofactor remains non-covalently bound in the analogous QFR-FrdA^{E245Q} variant. Indeed, prior studies demonstrate that in variant forms of QFR containing non-covalent FAD, the cofactor cannot be removed without KBr treatment, which is proposed to induce reversible unfolding (3).

To distinguish whether the QFR-FrdA^{E245Q} variant lacks activity due to altered ligand binding at the bound, non-covalent FAD, we analyzed the QFR interaction with two dicarboxylates: fumarate and oxaloacetate. We evaluated the binding of fumarate because it is the natural substrate. We also evaluated oxaloacetate because oxaloacetate is a natural tight-binding inhibitor of Complex II enzymes, with nanomolar affinity for most homologs. In addition, oxaloacetate forms a charge transfer band with the isoalloxazine ring of the flavin, which results in a characteristic spectrum for Complex II proteins that is easier to follow than fumarate/succinate. For these two reasons, measuring oxaloacetate binding can be particularly useful for assessing altered dicarboxylate binding capacity in variants.

For this measurement, we used difference spectra obtained by subtraction of a ligand-free spectrum from that of a ligand-bound wild-type QFR spectrum. This experiment monitors characteristic spectral changes in the flavin spectrum that occur upon binding of dicarboxylates. These spectral changes include an increase in absorption at 500 nm indicative upon fumarate binding near the FAD cofactor and a charge-transfer band in the 600-nm region upon oxaloacetate binding near the FAD. Both of these features are observed for the wild-type QFR (Fig. 4A). The QFR-FrdA^{H44S} variant containing noncovalent FAD shows spectral changes similar to wild-type QFR, indicating that dicarboxylate binding is similar to the wild-type enzyme (Fig. 4B). The difference spectra of the QFR-FrdA^{E245Q} variant, however, show no distinct features (Fig. 4C). This observed lack of spectral changes from the bound FAD upon the addition of dicarboxylates is likely due to significant changes in binding capability for these ligands.

Crystal structure of the QFR-FrdA^{E245Q} mutant

To identify whether the lack of dicarboxylate binding is the result of an altered fold of the enzyme, or loss of the non-covalent FAD cofactor we determined the crystal structure. Crystallization trials of QFR-FrdA^{E245Q} using conditions that were previously successful for wild-type and variant enzymes did not result in the appearance of crystals in this case. As a result, we screened for distinct chemical conditions to support crystallization of this variant. These crystals diffracted to low resolution, with the best data set merging to 4.25-Å resolution (Fig. 5A, Table 1). At this resolution, side chain conformations cannot be unambiguously assigned. Thus, we are not able to identify whether the FrdA^{E245Q} substitution has allosterically altered the positions of the dicarboxylate binding residues within the active site, which would be one explanation for why dicarboxylate binding is lost. However, the capping domain is in a similar position as it is found in the wild-type enzyme. As a result, side chains from the flavin-binding and -capping domains could conceivably be presented to the active site with similar geometries as is observed in the wild-type enzyme. Indeed, the structure of the QFR-FrdA^{E245Q} complex showed

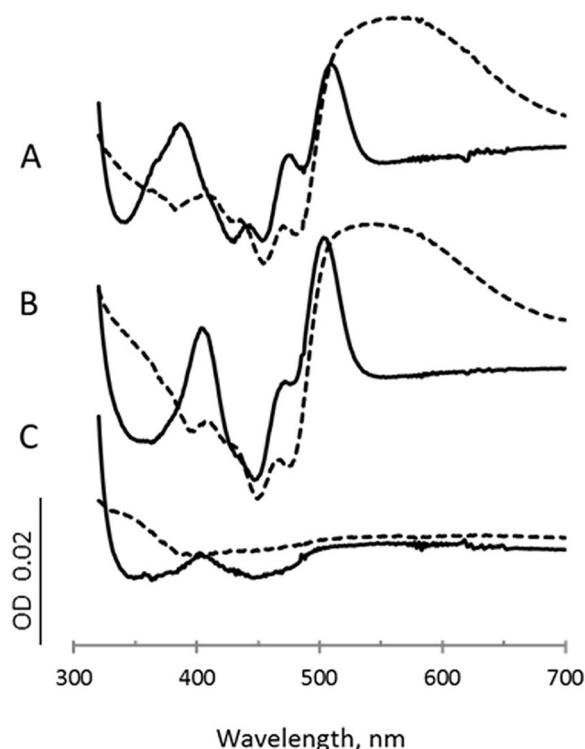


Figure 4. Dicarboxylate scans. Spectra are the difference between the spectrum of the enzyme after addition of each ligand minus the spectrum of the oxidized enzyme. Difference spectra for fumarate (10 mM) is shown as *solid lines*, and difference spectra for oxaloacetate (0.2 mM) are shown as *dashed lines*. A, wild-type QFR; B, QFR-FrdA^{H44S}; C, QFR-FrdA^{E245Q}.

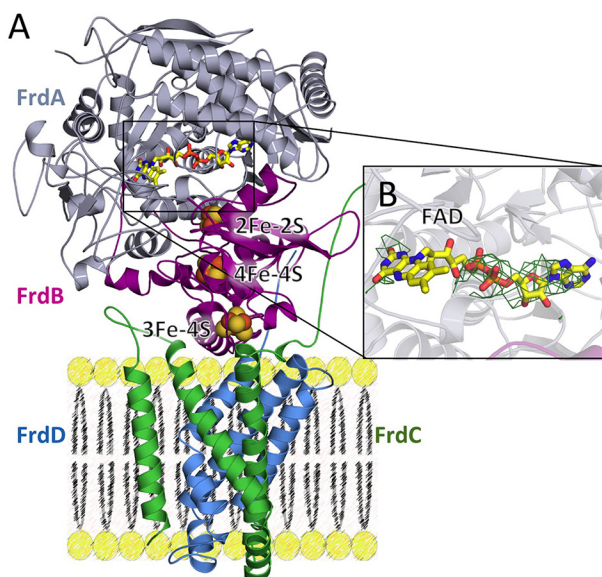


Figure 5. Structure of QFR containing the FrdA^{E245Q} mutation. A, structure of QFR-FrdA^{E245Q}. The flavoprotein (FrdA) is shown in *gray*, the iron-sulfur protein (FrdB) is shown in *magenta*, and the membrane subunits (FrdC and FrdD) are shown in *green* and *blue*, respectively. The locations of the cofactors (FAD as *yellow sticks*, and iron-sulfur clusters as spheres) are also shown. B, $|F_o| - |F_c|$ electron density maps contoured at 3σ were calculated in phenix.refine after the removal of FAD from the input PDB file. The quality of the electron density is consistent with the resolution for the adenine dinucleotide, but there is no interpretable density for the isoalloxazine ring.

close similarity to the wild-type protein, with a root mean square deviation of 0.33 Å for all atoms of the flavoprotein. This is consistent with previous work on the *Shewanella* soluble fumarate

Table 1
Crystallographic data collection and refinement statistics

Values in parentheses are for the highest resolution shell.

Data statistics	
Data collection	
PDB entry	5VPN
Beamline	APS 21-IDG
Wavelength	0.97856 Å
Resolution ^a	4.25 Å
Space group	P2 ₁ 2 ₁ 2 ₁
Unit cell	<i>a</i> = 133.6 Å <i>b</i> = 138.1 Å <i>c</i> = 220.1 Å
<i>R</i> _{sym}	0.166 (0.899) ^b
<i>R</i> _{int}	0.075 (0.448)
<i>CC</i> _{1/2}	0.919
<i>I</i> / σ ^a	10.6 (1.4)
Completeness	96.8% (98.2%)
Redundancy	5.5 (4.6)
Refinement	
<i>R</i> _{cryst}	0.215
<i>R</i> _{free}	0.270
Root mean square deviation	
Bond lengths	0.003 Å
Bond angles	0.614°
Ramachandran	
Most favored	93.6%
Allowed	6.1%
Outliers	0.3%

^a The data have an *I*/ σ > 2 in the outer shell at 4.4-Å resolution.

^b Values in parentheses are statistics for the highest resolution shell, 4.4–4.25-Å resolution.

reductase, where mutation of the analogous residue to aspartate (E378D) resulted in little structural perturbation (11).

Notably, there was a loss of strong electron density for much of the isoalloxazine ring of the FAD, despite retention of electron density for the adenine dinucleotide (Fig. 5B). This loss of electron density could arise for several reasons. One possibility is increased mobility of the isoalloxazine ring in the absence of the tethering covalent linkage. The relative structural stability of the isoalloxazine in flavoprotein homologs lacking the covalent linkage would argue against, but would not preclude, this interpretation. An alternative is that the low resolution of the data has reduced the quality of the electron density in this region. Nevertheless, it is clear from the crystal structure that the QFR complex is folded and assembled with non-covalent FAD in the presence of the FrdA^{E245Q} mutation.

Folding and thermal stability in wild-type and variant FrdA subunits

We next assessed the function of the FrdA^{E245} side chain in isolated FrdA subunits. The rationale for additionally assessing the role of FrdA^{E245} in the isolated FrdA subunits is that FrdA containing non-covalent FAD represents the assembly intermediate, *i.e.* covalent flavinylation is believed to occur within the isolated subunit. We first ensured that the variant FrdA subunits were properly folded. As we were unable to identify crystallization conditions that support the growth of isolated FrdA subunits, we used CD spectroscopy and thermal denaturation to measure folding and stability. Changes in stability of the isolated FrdA subunit in the presence of mutations at the FrdA^{E245} could suggest whether stability affects the covalent flavinylation and maturation process, as has previously been suggested for the human complex (20). For these studies, we compared wild-type FrdA, FrdA^{H44S} (lacking the histidine link-

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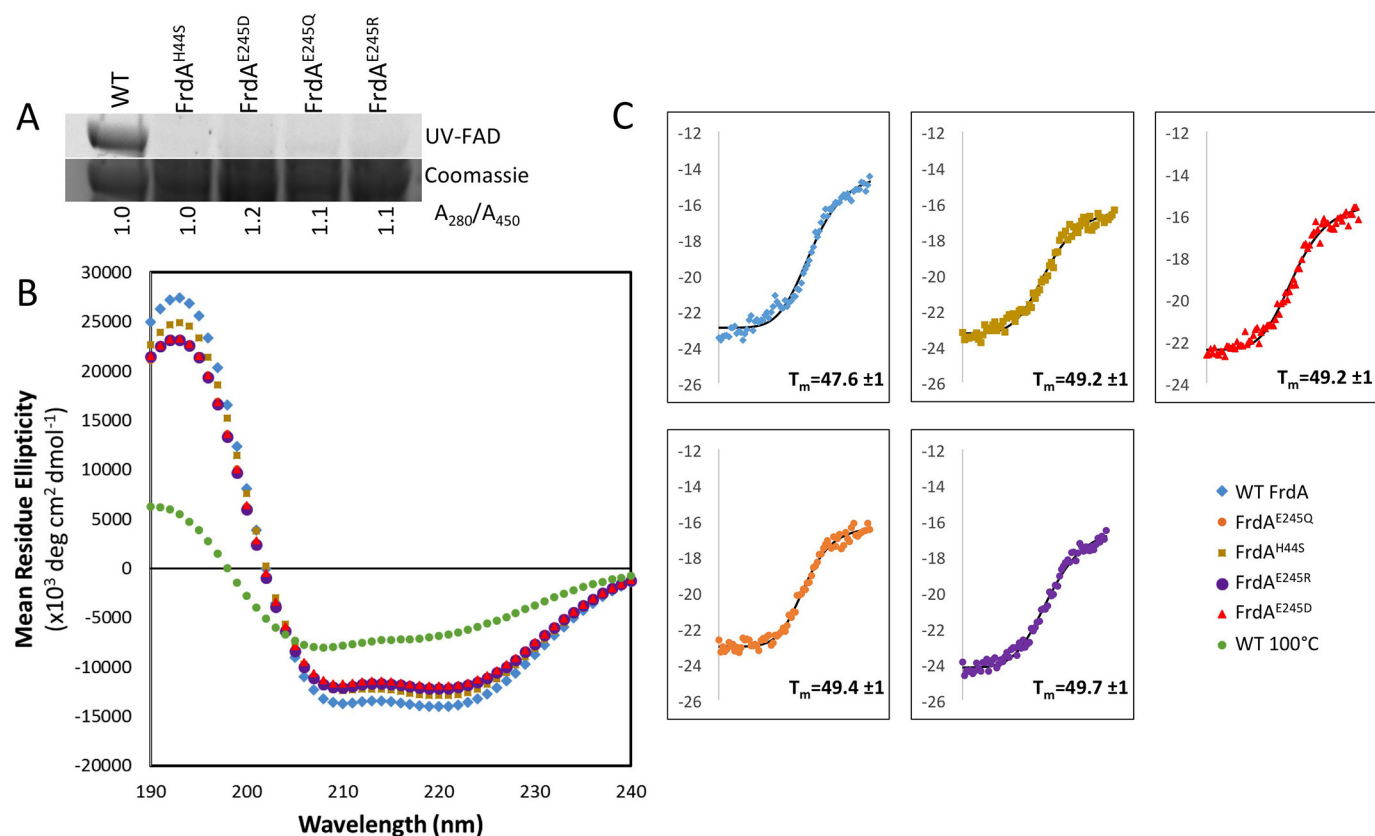


Figure 6. Circular dichroism spectroscopy of wild-type and variant FrdA subunits. *A*, measurement of flavin in wild-type and variant FrdA subunits (~67 kDa). The quantity of flavin covalently associated with FrdA was measured by separating lysates on SDS-PAGE and measuring FAD fluorescence following illumination with UV light. The relative quantity of FAD associated with wild-type and variant FrdA subunits was assessed by measuring the A_{280}/A_{450} ratio of purified proteins. This value was normalized to 1.0 for wild-type protein. The ratio of A_{280}/A_{450} was similar in wild-type and variant subunits, indicating that these had a similar amount of flavin associated with each subunit. As *in gel*/UV detection indicates that flavin is not covalently associated with the subunits, the tightly associated FAD must be non-covalently bound, likely via specific binding to the active site. *B*, far-UV CD spectrum of wild-type FrdA compared with spectra for the FrdA^{H44S} variant, the FrdA^{E245} variants, and FrdA incubated at 100 °C for 2 h. *C*, representative traces of thermal CD of wild-type FrdA, FrdA^{H44S}, FrdA^{E245D}, FrdA^{E245Q}, and FrdA^{E245R}. The average T_m values are indicated.

age to the FAD), FrdA^{E245Q}, and two additional variants of the FrdA^{E245} side chain: FrdA^{E245R} and FrdA^{E245D}. Each of these isolated FrdA^{E245} variants is associated with stoichiometric non-covalent flavin (Fig. 6A), as assessed by comparing the total amount of bound FAD to the wild-type FrdA subunits and verifying that these variants lack measurable covalent FAD.

We measured the far-UV CD spectrum for wild-type FrdA, which is consistent with a predominantly helical fold (Fig. 6B). The spectra for all variants tested were similar to that of wild-type FrdA, indicating that each variant FrdA subunit is properly folded and there are no major changes in the secondary structure. We next monitored changes in the ellipticity at 220 nm as a function of temperature in thermal denaturation CD spectroscopy. There were no significant differences in the melting temperature (T_m) or midpoint of the unfolding transition for the variants as compared with wild-type FrdA (Fig. 6C). This indicates that neither the amino acid substitutions nor the loss of the covalent bond to the bound flavin significantly impacts FrdA subunit stability.

Cross-linking with the assembly factor, SdhE

To identify whether the FrdA^{E245} side chain influences the interaction of FrdA with the assembly factor SdhE, we monitored the ability of FrdA variants to cross-link with SdhE incor-

porating the artificial amino acid *para*-benzoyl phenylalanine (pBpF) at position 8. As previously described, this variant of SdhE, termed SdhE-R8BpF, readily cross-links with wild-type flavoprotein when illuminated with UV light (21), suggesting that this method can be used to report upon whether FrdA variants retain the ability to cross-link with SdhE.

We first measured cross-linking between wild-type FrdA and SdhE-R8BpF over the course of a 4-h illumination with UV light (Fig. 7A) and observed a time-dependent increase in FrdA-SdhE association. We selected the 3-h time point as suitable for measuring cross-linking to variants, and measured cross-linking between SdhE-R8pBpF and the FrdA^{E245D/Q/R} and FrdA^{H44S} variants. Using this assay, we observed robust cross-linking between SdhE and wild-type FrdA. Indeed, the cross-linking could be observed on a Coomassie-stained SDS-PAGE gel, as previously reported (21). In contrast, cross-linking to each of the variants was substantially attenuated, and we could no longer observe the formation of a stable complex when monitoring by Coomassie-stained SDS-PAGE. To assess whether a low level of cross-linking occurred, we evaluated the formation of the FrdA-SdhE cross-linked complex using Western blotting analysis, probing with an antibody that recognizes that His₆-affinity tag of SdhE. Even with Western blotting analysis, titra-

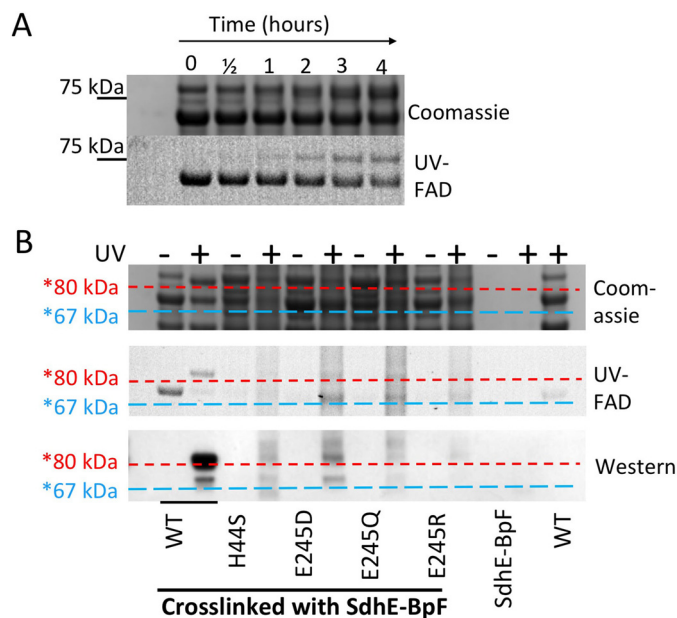


Figure 7. Cross-linking between wild-type and variant FrdA subunits and SdhE-R8BpF. Purified SdhE-R8BpF was mixed with lysate containing FrdA. Following UV exposure, samples were immobilized on Ni^{2+} resin, washed, and eluted with SDS-PAGE loading buffer prior to separation on an SDS-PAGE gel. *A*, time course of cross-linking of SdhE-R8BpF with wild-type flavoprotein over 4 h after exposure to UV light. The position of the 75-kDa molecular mass marker is highlighted. *B*, comparison of flavin fluorescence for detection of covalent FAD, Coomassie-stained SDS-PAGE and anti-His₆ Western blot analysis of cross-linked samples. Horizontal lines were drawn after aligning the Coomassie, UV, and Western blot analyses to assist with identifying the molecular weights of the bands. The blue dashed lines are drawn just under the location of the FrdA subunit (~67 kDa) and red dashed lines are drawn just under the approximate location of the FrdA-SdhE complex (~80 kDa).

tion of the antibody concentrations was required before the signal was observed (Fig. 7B).

A reduction in binding between FrdA and the SdhE assembly factor is somewhat anticipated for the FrdA^{H44S} variant, as the FrdA^{H44} side chain is proposed to be part of the FrdA-SdhE binding interface (21). However, the location of the FrdA^{E245} side chain is on the capping domain (6). In the assembled (Frd-ABCD) QFR heterotetramer, this side chain is buried, but mediates contacts between the FAD-binding domain and capping domain of the FrdA subunit. One possibility for the observed loss of interaction is that the cross-linking assay appears to be exquisitely sensitive to small structural changes introduced by mutations, as was shown in prior studies (21). Thus even minor structural changes accompanying these mutations could result in a loss of cross-linking. Alternatively, it is possible that the FrdA subunit undergoes conformational changes during SdhE binding that are altered in the context of FrdA^{E245} variants. Because FrdA^{E245} is positioned on the capping domain of FrdA adjacent to a loop that links the two domains, and because the capping domain of FrdA likely rotates upon the interaction with SdhE, this rotation could be altered or prohibited in the context of FrdA^{E245Q}.

SAXS of the FrdA-SdhE-R8BpF cross-linked complex

Although there is currently not an experimental structure of the flavoprotein-SdhE complex, a computational model has been developed using restraints from site-specific cross-linking

studies combined with mass spectrometry (21). This model proposed that the FrdA-SdhE interaction requires a rotation of the FrdA-capping domain, but did not suggest how the capping domain might reorient. To investigate a potential role of FrdA^{E245} in conformational changes in the FrdA subunit during SdhE binding, we performed SAXS analysis of the wild-type *E. coli* FrdA-SdhE-R8BpF cross-linked complex at protein concentrations of 1.2 and 2.4 mg/ml. Scattering intensity profiles and linear Guinier plots indicated that the protein lacks aggregation and is monodisperse (Fig. 8A). Furthermore, a Guinier approximation exhibited good linearity at both protein concentrations and yielded R_g values of 28.11 ± 1.68 and 27.79 ± 0.92 Å with an average of $R_g \sim 28$ Å. Fourier transformation of SAXS data in the same range gave D_{max} of 82–86 Å and R_g of 27–28 Å (Fig. 8B). We then developed several sets of coordinates for docking into the SAXS envelope. These coordinates began with the Rosetta-minimized computational model of the FrdA-SdhE complex that lacked the FrdA-capping domain (21). We manually assessed several plausible alternatives for capping domain positions, with the capping domain being moved as a rigid body. Each alternative that was tested reduced, but did not completely eliminate, steric clash with either SdhE or other parts of the FrdA. This procedure identified one model exhibiting a reasonable correlation between the theoretical scattering curve and the experimental scattering curves at both protein concentrations, computing χ^2 values of 2.13 and 2.75, respectively (Fig. 8C). The selected computational model of the FrdA-SdhE cross-linked complex was then superimposed on the SAXS envelope by automated alignment of inertial axes (Fig. 8D). In the best model, the capping domain containing FrdA^{E245} is rotated so that the FrdA^{E245} side chain approaches the proposed SdhE-binding site, but this side chain does not appear to interact directly with SdhE. Although the resolution of SAXS data is not sufficient to place secondary structural elements or side chains, this large rotation is in contrast to small but significant structural changes observed in the flavoprotein *p*-cresol methyl hydroxylase upon interaction with a cytochrome subunit that promotes covalent flavinylation in that case (4). If the FrdA-SdhE model is correct, the most likely interpretation for the loss of cross-linking between FrdA^{E245Q} and SdhE-R8BpF is that the mutation either prevents FrdA from adopting a conformation that can bind SdhE or the mutation causes small surface changes that reduce SdhE binding without being a direct part of the binding site.

Discussion

Although we know a great deal about the mechanism of succinate and quinone oxidoreduction by Complex II superfamily members, comparatively little is known about Complex II assembly. Indeed, our understanding of protein assembly in general is underdeveloped as compared with our understanding of how individual proteins function. In the case of Complex II, one critical aspect of assembly is the insertion and attachment of its cofactors. For well over 30 years, this process was considered fully autocatalytic, potentially proceeding via a quinone-methide intermediate. Only within the last 10 years were any specific assembly factors discovered, initiating a debate of the mechanisms of cofactor attachment.

Covalent flavinylation of Complex II

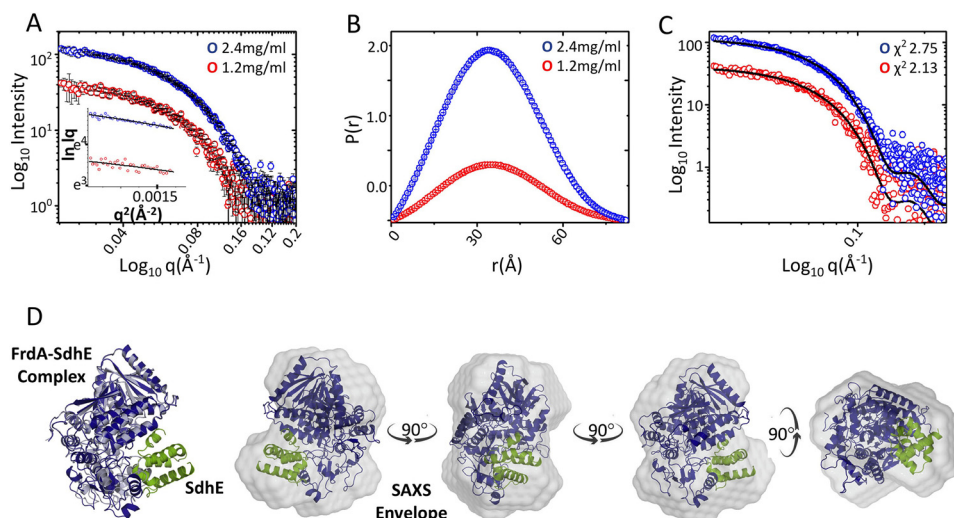


Figure 8. SAXS of the cross-linked FrdA-SdhE-R9pBpF complex. *A*, intensity profiles in log scale of the cross-linked FrdA-SdhE complex acquired at protein concentrations of 1.2 (red) and 2.4 mg/ml (blue). The insets show the linear fit to the Guinier region of the respective protein concentrations. *B*, $P(r)$ plot depicting the frequency distribution of interatomic vectors in the predominant scattering species after indirect Fourier transformation of the SAXS data provided an R_g and D_{max} of 27.25 ± 0.32 , 28.02 ± 0.13 , and 82.83 and 86.81 Å, respectively, at lower and higher protein concentration. *C*, comparison of theoretical scattering curve (black line) of the computational model of the cross-linked protein with experimental scattering curves at different protein concentrations give χ^2 values of 2.13 and 2.75, respectively. *D*, conceptual model of FrdA (blue) and SdhE (green) complex superimposed on wild-type FrdA (gray, PDB entry 3P4P (38)) and the SAXS envelope by automated alignment of inertial axes. Rotating views of the FrdA-SdhE-R8BpF model docked into the SAXS envelope.

SdhAF2 was identified in patients who presented clinical Complex II deficiency, but who did not have mutations in Complex II genes (17). Phylogenetic analysis of SdhAF2 (~11 kDa) indicates that this protein evolved prior to the development of mitochondria, a proposal consistent with its strong conservation in all kingdoms of life (18). After its discovery, SdhAF2 was initially believed to be essential for covalent flavinylation of human Complex II. Indeed, its mutation recapitulates diseases associated with Complex II deficiency (17). However, recent studies of Crisper-CAS9 Δ *sdhAF2* breast cancer cells suggest that covalent flavinylation can proceed in the absence of SdhAF2 (24). In addition, there are examples of thermophilic bacteria and archaea that lack SdhE genes, but still produce functional, covalently flavinylated SQR (25). Our finding that the requirement for SdhE is more important during aerobic respiration adds complexity to this debate, but further supports SdhE as a stimulating, non-essential factor for covalent flavinylation.

Integratingly, SdhE is not the only factor that can enhance the efficiency of covalent flavinylation. It has also been shown that dicarboxylates, including citric acid cycle intermediates, stimulate covalent flavinylation of Complex II isoforms (28). Moreover, the deletion of the iron-sulfur protein (FrdB or SdhB) reduces, but does not eliminate covalent flavinylation, identifying this subunit as modestly supporting the formation of the covalent linkage (29, 30). Given that neither SdhE (24), the iron-sulfur protein (29), nor dicarboxylates (28) are absolutely required for covalent flavinylation of FrdA/SdhA, but that SdhE can enhance flavinylation of human SdhA in a purified system (20), the most logical conclusion is that the chemistry of covalent attachment is supported by the flavoprotein itself. But if covalent flavinylation does proceed autocatalytically, what are the chemical requirements, and how do SdhE, dicarboxylates, or the iron-sulfur protein enhance this?

Taken in aggregate, our findings and those from the literature support a model where autocatalytic covalent attachment

via the quinone-methide intermediate requires a pre-organized active site with the capping domain closed tightly. A tightly closed structure would also be anticipated to be more stable and less sensitive to proteolysis, as has been proposed (20). This finding is consistent with observations where Sdh5 binding to yeast SdhA stabilizes Sdh5 against LON-mediated proteolysis (31). Indeed, binding of the flavoprotein to the assembly factor probably protects both from proteolysis. The tightly closed conformation may align active site residues into orientations that optimally support the chemistry of covalent flavinylation. Certainly the FrdA^{H44} residue must be positioned closely enough to the isoalloxazine C(8) α to allow formation of the covalent bond. In addition, other active site side chains may impact the covalent flavinylation chemistry directly, for example, FrdA^{R390} is positioned close to the N1-C2 of the isoalloxazine ring in the assembled complex.

Overall, the data suggest that there may be more than one way to promote active site pre-organization. The previously reported findings that bound dicarboxylates (28), the iron-sulfur protein (29, 30), and SdhE (17) can each enhance covalent flavinylation are consistent with this proposal, as each of these could function to pre-organize the active site. Bound dicarboxylates could directly interact with and orient active site residues, which would help to organize the active site and promote tight closure of the capping domain. The role of dicarboxylates in covalent flavinylation is consistent with our finding that mutation of residues that impact dicarboxylate binding directly or indirectly reduces covalent flavinylation to levels just above detection (Fig. 3), and substantially below those observed in Δ *sdhE* strains grown under the same conditions (Fig. 2). However, substrate turnover might not be necessary. Indeed, prior studies of other catalytically compromised QFR and SQR variants indicate that covalent flavinylation is usually retained (7, 8).

The binding of FrdA to SdhE or the iron-sulfur protein may similarly promote the organization of active site residues. A

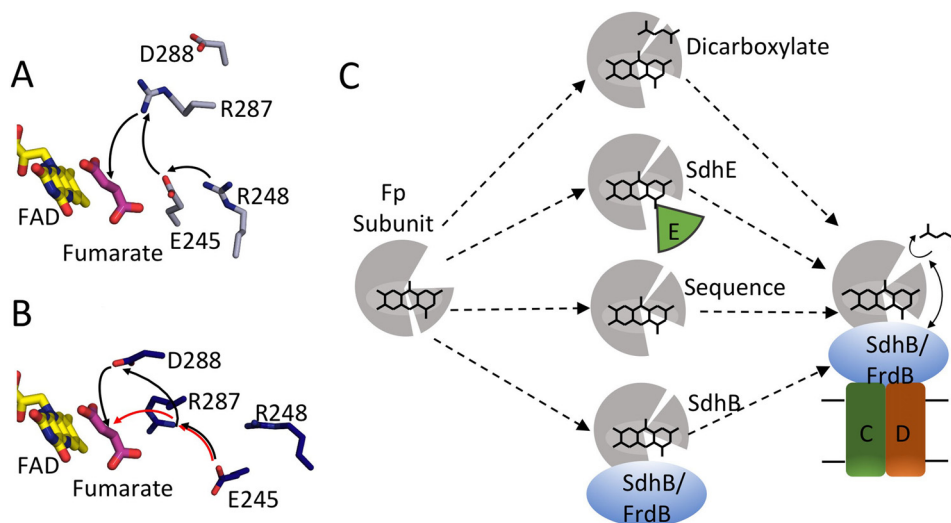


Figure 9. Model for side chains in covalent flavinylation. *A*, active site organization in wild-type QFR with fumarate bound (gray, PDB entry 3P4P (38)). FAD (yellow) and fumarate (pink), and side chains of FrdA^{E245}, FrdA^{R248}, FrdA^{R287}, and FrdA^{D288} are shown as sticks. The black arrows indicate the direction of proton transfer during fumarate reduction. *B*, the same view for the FrdA-SdhE-R8BpF model (blue). The black and red arrows indicate two potential alternate proton transfer pathways after reorganization of the active site as a result of domain rotation in the model. *C*, scheme for organization of the flavoprotein (gray) active site to promote autocatalysis of the covalent flavin bond. In this model, closure of the capping domain after binding of flavin (represented as stick model of the isoalloxazine ring) and proper orientation of the attachment site can be facilitated in several ways: (i) binding of dicarboxylate; (ii) binding of the assembly factor; (iii) enhanced stability and preorganization by sequence in thermophilic bacteria; (iv) binding of the iron-sulfur subunit. After this organization, covalent flavinylation can occur (indicated by additional bond) followed by complex assembly.

computational model developed in the context of experimental distance restraints strongly suggests that this FrdA-SdhE interaction promotes a conformational change in FrdA that closes the capping domain of the flavoprotein tightly over bound dicarboxylate (21). Similarly, the position of the iron-sulfur protein in the assembled complex may restrict the motions of the capping domain. SdhE may be particularly important during aerobic respiration with succinate. It is plausible that succinate may have lower affinity for an active site containing non-covalent FAD, the capping domain may not close as tightly prior to the formation of the covalent linkage. As a result, an auxiliary factor such as SdhE may increase the efficiency of covalent flavinylation under aerobic conditions (Fig. 2). An additional intriguing interpretation of a difference in flavinylation under aerobic *versus* anaerobic conditions is that SdhE may have evolved during the transition and adaption of cells to an aerobic environment. It may also be consistent with the idea that SQR formed from a QFR at the transition from anaerobic to aerobic life, with the covalent bond fulfilling the need for higher redox potential to oxidize succinate *versus* reduce fumarate.

Covalent flavinylation via a mechanism that uses a quinone-methide intermediate would also require a proton shuttle. Although the FrdA^{E245} side chain is proposed as part of a proton delivery pathway during fumarate reduction, it is not clear whether it plays the same role during covalent flavinylation. Pre-organization of the active site would have the impact of aligning the proton shuttling pathway consisting of FrdA^{R287}, FrdA^{E245}, and FrdA^{R248} (Fig. 9A). However, FrdA^{R248} variants retain covalent flavin (32). One interpretation of this finding is that this proton delivery pathway is used during fumarate reduction but not during covalent flavinylation. An alternative possibility could be considered in light of the rotation of the capping domain when FrdA is in complex with SdhE. The FrdA-SdhE model would suggest that this rotated capping

domain could allow the FrdA^{E245} side chain to be directly exposed to solvent, such that the FrdA^{R248} side chain that normally participates in proton delivery could be bypassed during covalent flavin attachment (Fig. 9B). The amino acid residue equivalent to FrdA^{D288} is conserved in the Complex II family of enzymes (*E. coli* SdhA^{D287} and in the mature (*i.e.* 42-amino acid transit peptide processed) human SdhA^{D299}). In the available X-ray structures of bacterial, yeast, and mammalian Complex II the aspartate residue is hydrogen bonded to a conserved Asn residue (FrdA^{N389}, *E. coli* SdhA^{N398}, and human SdhA^{N408}), which is adjacent to FrdA^{R390} (*E. coli* SdhA^{R399} and human SdhA^{R409}). FrdA^{R390} is known to be essential for covalent flavinylation in bacteria (Fig. 3B) and mutation of the equivalent residue in humans is associated with disease (33). FrdA^{R390} is thought to stabilize the negative charge at the N1-C2 position of the isoalloxazine that develops in the quinone-methide tautomer and thus is essential for development of the tautomer. It is reasonable to suggest that mutation of FrdA^{D288} to a neutral amino acid such as Asn impacts the architecture of the active site. This would be expected to affect dicarboxylate binding and the environment around FrdA^{R390} such that it is unable to sufficiently stabilize the negative charge that is required near the isoalloxazine N1-C2 position during the autocatalytic formation of the covalent flavin linkage. It is also possible, given the potential reorganization of the active site upon SdhE binding, that FrdA^{D288} has a role in proton shuttling in covalent flavinylation (Fig. 9B). It is likely that SdhE helps impact the active site architecture by stabilizing the environment near the flavin to help align the active site residues during the time necessary for formation of the covalent flavin bond. Further insight into the role of SdhE in covalent flavinylation requires the determination of a high-resolution structure of the FrdA-SdhE complex.

Covalent flavinylation of Complex II

Conclusions

Together, the data argues in favor of multiple complementary mechanisms that promote the pre-organization of the Complex II active site during covalent flavinylation. One or more molecular players (bound dicarboxylate, SdhE, iron-sulfur protein, or potentially others) could facilitate the correct orientation of active site residues and promote tight domain closure over the isoalloxazine (Fig. 9C). The findings support the original proposal of Walsh (16) that covalent attachment is self-catalytic and argue against a direct role for SdhE in the chemistry of covalent flavinylation. Rather the role of SdhE or its homologous molecular chaperones (SdhAF2/Sdh5) is to stabilize the flavoprotein in a tightly closed conformation to organize the active site residues and to allow time for the covalent flavinylation to occur.

Experimental procedures

Expression and purification of the QFR complex (FrdABCD)

Wild-type and variant QFR (FrdABCD) were expressed as described (21) from plasmid pH3, which encodes the *frdA*⁺*B*⁺*C*⁺*D*⁺ operon under the control of the fumarate reductase (FRD) promoter. QFR was expressed as described in *E. coli* strain DW35, where both the *frd* and *sdh* operons are disrupted via the insertion of a kanamycin gene (34). Expression is induced under micro-aerophilic conditions, which is achieved in Terrific Broth medium by increasing the culture volume in the flasks (1.6 liters in a 2-liter growth flask), and reducing the shaking to 160 rpm (8). Cells were harvested by centrifugation at 9200 × *g* and disrupted by sonication. Membranes were isolated by ultracentrifugation. QFR comprises ~50% of the total membrane protein when expressed using this protocol, and covalent flavinylation was assessed on this membrane-embedded QFR without further purification.

For crystallization, the QFR-FrdA^{E245Q} variant was purified as described (35). Membranes were resuspended in 25 mM Tris, pH 7.4, 0.1 mM EDTA containing complete protease inhibitor tablets (Roche Applied Science) and solubilized in 2% thesitol detergent (C₁₂E₉, Anatrace). The insoluble fraction was removed by centrifugation at 34,000 × *g*, and the supernatant containing membranes was filtered and applied to a Q-Sepharose column. The protein was eluted with a linear NaCl gradient from 50 mM to 1 M. The NaCl concentration was reduced via three rounds of 1:10 dilution and reconcentration using a 30-kDa molecular mass cut-off filter (Amicon). QFR was then further purified using a Poros 50HQ column, eluting with a linear NaCl gradient from 50 mM to 1 M. The protein was then concentrated as above and injected onto a Superdex 200 Increase 10/300 column. Protein concentration was determined using the Bradford assay.

Expression and purification of isolated FrdA subunits

Isolated wild-type and variant FrdA subunits were expressed in *E. coli* strain RP-2 (Δ *frdABCD*, Δ *sdhCDAB*), a plasmid that inserted an additional stop codon after the *frdA* gene in the pH3 plasmid, as described (21). Cells were grown as described above for the intact complex. Cells were harvested by centrifugation and stored at –20 °C. The cells were then resuspended in 20

mM potassium phosphate, pH 7.4, with complete protease inhibitors (Roche Applied Science). Cells were lysed by three cycles of vortexing (3 min) with protein extraction beads (diameter <1 mm) followed by a freeze-thaw at –20 °C. Lysate was cleared by centrifugation at 34,000 × *g* and the supernatant was stored at –20 °C. For photocross-linking experiments, no additional purification was performed. For circular dichroism (CD) spectroscopy experiments, isolated FrdA subunits were purified using an anion-exchange Poros 50HQ column in buffer containing 25 mM Tris-HCl, pH 7.4, and a linear NaCl gradient from 50 mM to 1 M NaCl. This was followed by gel filtration in the same buffer on a Superdex 200 Increase 10/300 column. Elution was monitored by following *A*₂₈₀. Following this procedure, the FrdA subunits were estimated to be 95% pure based on SDS-PAGE.

Expression and purification of SdhE containing the artificial amino acid pBpF

The *sdhE* gene with an amber codon substituting R8 was cotransformed with the pEVOL-pBpF to express pBpF-containing SdhE, as described (21). SdhE-R8BpF was grown in LB medium containing 0.5 mM pBpF. After 1 h, the temperature was adjusted to 30 °C. SdhE-R8BpF was induced by the addition of 0.1 mM isopropyl β -D-1-thiogalactopyranoside and the tRNA synthase/tRNA for pBpF was induced with 0.2 mM arabinose. Cells were harvested by centrifugation and lysed by sonication. The lysate was cleared of cellular debris by centrifugation for 45 min at 34,000 × *g*. SdhE-R8pBpF was purified by nickel affinity chromatography, as described (21).

Quantitation of covalent flavinylation under aerobic and anaerobic growth conditions

LB supplemented with ampicillin (100 μ g/ml) (1.5 ml in a 14-ml round bottom Falcon tube) was inoculated with a single colony. The cells were grown overnight at 37 °C with 270 rpm shaking, then divided into aerobic or anaerobic cultures. For aerobic growth, 25 μ l of the overnight culture was used to inoculate 1 ml of LB supplemented with ampicillin (100 μ g/ml) and cells were grown for 7 h at 37 °C with shaking. For anaerobic growth, 1 ml of the overnight culture inoculated 10 ml of LB supplemented with ampicillin (100 μ g/ml) in a 15-ml Falcon tube; the cells were grown at 37 °C in the Isoterm rotisserie incubator (Fisher Scientific) for 17 h with slow rotation. Prior to analysis by SDS-PAGE, the *A*₆₀₀ was measured to normalize protein load. Cells (0.3 to 0.9 ml) were collected by centrifugation for 1 min, and the pellets resuspended in 2× SDS loading buffer (Bio-Rad). Each lane on the gel contains an equivalent amount of the cell culture (*A*₆₀₀ = 1.0).

Quantification of covalent and non-covalent flavinylation

Covalent flavinylation of wild-type and variant QFR was detected as described (36). To analyze expression levels of the enzyme, 15 μ g of isolated membranes were separated by SDS-PAGE on a 4–12% BisTris gel (NuPAGE Novex). FAD was quantified by UV fluorescence (Gel Doc EZ imager) using ImageJ software. The gel was then stained with SimplyBlue Safe Stain (ThermoFisher) and the protein was quantified by densitometry. The ratio between the UV and Coomassie was used to

assess the level of covalent flavinylation, with wild-type QFR covalent flavinylation set to 100% and QFR-FrdA^{H44S} set to 0%.

To determine the level of total FAD present in the purified FrdA variants, A_{280}/A_{450} values were measured using a Nano-Drop1000 spectrophotometer. The A_{280}/A_{450} absorbance ratio was calculated for wild-type FrdA containing covalent FAD, and this value was set to 100%. The A_{280}/A_{450} absorbance ratio for the FrdA variants was compared with the ratio from wild-type. Because FAD was tightly associated with the variants but not covalently bound, all measured FAD absorbance was assumed to reflect non-covalent FAD specifically bound at the active site in the variants.

Measurement of enzyme activity

Enzymatic activity was determined in bacterial membranes containing wild-type and FrdA^{E245Q} QFR variants. Succinate dehydrogenase and menaquinol-1 fumarate reductase reactions were assayed as previously described (37).

Spectroscopic evaluation of ligand binding

Ligand binding was performed using a modification of a described protocol (38). Purified wild-type and variant QFR enzymes were incubated with 3 mM malonate (to remove tightly bound oxaloacetate) in 50 mM potassium phosphate, pH 7.0, 0.01% Thesit for 20 min at room temperature and passed through a PD-10 gel filtration column in the same buffer minus malonate. The resultant enzymes were stored on ice. Optical spectra were recorded using an Agilent 8453 diode array spectrophotometer in 50 mM potassium phosphate, pH 7.0, 0.01% Thesit at 25 °C. Difference spectra were obtained by subtraction of a ligand-free spectrum from the corresponding spectrum of the enzyme in the presence of either 0.2 mM oxaloacetate or 3 mM fumarate.

Crystallization and structure determination

Crystals of the QFR-FrdA^{E245Q} mutant were grown using the hanging-drop vapor diffusion method in droplets containing 1 μ l of protein solution (15 mg/ml of QFR FrdA^{E245Q} in 25 mM Tris-HCl, pH 7.4, 1 mM EDTA, 0.02% C₁₂E₉) and 1 μ l of the reservoir solution (275 mM sodium malonate, 19% polyethylene glycol 6000, 100 mM sodium citrate, pH 4.0, 1 mM EDTA, and 0.001% dithiothreitol). Droplets were equilibrated over 1 ml of the reservoir solution at 20 °C. Crystals were cryo-protected with a solution that was 80% reservoir solution and 20% of a 1:1 mixture of glycerol and ethylene glycol prior to flash cooling with liquid nitrogen.

X-ray diffraction data were collected at the Advanced Photon Source (APS) beamline 21-ID-G at -173 °C using a wavelength of 0.9798 Å and a MarMosaic225 CCD detector. Data were processed and scaled using HKL2000 (39). The structure was determined using molecular replacement in the program Phaser (40) through the Phenix interface (41), and wild-type QFR (PDB entry 1LOV (6)) as the search model. For refinement, the capping domain was separated from the flavoprotein and the positions of each subunit plus the isolated capping domain were optimized using rigid body refinement in Phenix (41). The capping domain was then reconnected to the flavoprotein and additional refinement was performed using standard xyz

refinement in Phenix (41) after constraining the secondary structural elements. As assessed by monitoring the *R*-factors and geometry, real space refinement decreased the quality of the model and was not used. The refined model was analyzed in Coot (42), but significant manual model building was not performed given the resolution. Data collection and refinement statistics are summarized in Table 1.

Far-UV CD spectroscopy

Spectra were collected for wild-type FrdA subunits and FrdA variants on a Jasco J-810 CD spectropolarimeter at 20 °C using a 0.1-cm path length quartz cuvette. Measurements were taken from 190 to 260 nm in 1-nm intervals. Secondary structure was analyzed using the DichroWeb analysis server (43). Data are the average of three runs and have been corrected by subtraction of the buffer spectra.

Thermal denaturation CD spectroscopy

Variable temperature CD spectra were taken by monitoring the change in the CD signal (mDeg) at 220 nm from 5 to 85 °C in 1 °C increments with a 2-min equilibration time between data points and a 4-s response. Data are the average of three runs and were corrected by subtraction of buffer spectra.

Using the *lsqcurvefit* function in Matlab R2015a, the CD data were fitted as a function of temperature to a modified version of the Hill equation.

$$CD = \frac{AT^n}{B^n + T^n} + C \quad (\text{Eq. 1})$$

Where *A* is the maximum CD signal minus the minimum CD signal in mDeg, *T* is temperature in °C, *B* is the midpoint of the plot of CD signal versus temperature, *n* is the Hill exponent, and *C* is the minimum CD signal. To determine the melting point, the derivative of this function,

$$\frac{d}{dx} \left[\frac{AT^n}{B^n + T^n} + C \right] = \frac{AB^n n T^{n-1}}{(T^n + B^n)^2} \quad (\text{Eq. 2})$$

was evaluated at temperature values ranging from the minimum observed temperature to the maximum observed temperature, in steps of 0.01 °C. The temperature corresponding to the maximum of the derivative was taken as the melting point.

Cross-linking of FrdA subunits to pBpF-incorporated SdhE

Purified His₆-SdhE-R8BpF was mixed with *E. coli* strain RP-2 lysate containing overexpressed and untagged FrdA in a Corning 6-well plate on ice. Cross-linking was induced by illumination with 365-nm light (Black-Ray 100 Watt UV lamp). The SdhE-FrdA cross-linked complex was purified by Ni²⁺ affinity chromatography and analyzed by SDS-PAGE, where a characteristic shift in molecular weight was observed. The presence of SdhE in the shifted band was verified by Western blot analysis with a primary antibody against the His₆ tag; the presence of FrdA in the shifted band was confirmed by monitoring UV fluorescence. For the time course experiment, FrdA and SdhE-R8BpF samples were mixed in a 96-well plate on ice and exposed to UV light. Samples were collected at the indicated time points, and quenched with SDS loading buffer.

Covalent flavinylation of Complex II

SAXS data collection and analysis

SAXS data were collected at the SIBYLS beamline at the Advanced Light Source in Berkeley, CA, as described (44). Measurements of the SdhE-FrdA cross-linked complex, purified as described above and dialyzed into 50 mM MES, pH 6.0, with 5% glycerol, were carried out at 283 K. Data were collected at two different concentrations (1.2 and 2.4 mg/ml) to analyze concentration-dependent effects. Samples were placed 1.5 m from a MAR165 CCD detector arranged coaxial with the 11 keV monochromatic beam; 10^{12} photons/s were impinging on the sample. A series of 0.3-s exposures were collected for 10 s, resulting in 32 frames per condition. Buffer subtraction and raw image data were integrated by beamline software (45).

Guinier analysis and indirect Fourier transformation were performed in the PRIMUS (46) and GNOM (47) programs, respectively. The *ab initio* shape determination was carried out using DAMMIF (48). The resulting SAXS-based model was then superimposed on models of the FrdA-SdhE-R8BpF cross-linked complex by their inertial axes alignment using the SUPCOMB program (49). The experimental scattering data were further validated with the scattering profile of the computational model of cross-linked complex using CRY SOL (50).

Author contributions—C. A. S. analyzed covalent and non-covalent flavinylation in wild-type and variant QFR complexes and FrdA subunits, performed CD spectroscopy of FrdA subunits, purified, crystallized, and determined the structure of the QFR-FrdAE245Q variant with the assistance of S. Q. H., prepared samples for SAXS analysis, and analyzed the SAXS data under the direction of P. S. E. M. measured covalent flavinylation of QFR/SQR in *E. coli* strains grown under aerobic and anaerobic conditions, measured dicarboxylate binding to wild-type and variant QFR complexes, and performed kinetic analyses. T. M. I. and G. C. developed the study. T. M. I., G. C., and C. A. S. wrote the paper with input from all authors.

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