



Published in final edited form as:

*Angew Chem Int Ed Engl.* 2017 August 21; 56(35): 10530–10534. doi:10.1002/anie.201704755.

## Practical Alkoxythiocarbonyl Auxiliaries for Ir(I)-Catalyzed C–H Alkylation of Azacycles

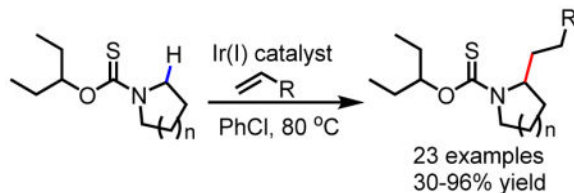
Dr. Anh T. Tran and Prof. Dr. Jin-Quan Yu

Department of Chemistry, The Scripps Research Institute (TSRI), 10550 North Torrey Pines Road, La Jolla, CA 92037 (USA)

### Abstract

The development of new and practical 3-pentoxythiocarbonyl auxiliaries for Ir(I)-catalyzed C–H alkylation of azacycles is described. This method allows for the  $\alpha$ -C–H alkylation of a variety of substituted pyrrolidines, piperidine and tetrahydroisoquinoline through alkylation with alkenes. While the practicality of these simple carbamate type auxiliaries is underscored in the ease of installation and removal, the method's novel reactivity is demonstrated in its ability to functionalize biologically relevant (L)-proline and (L)-trans-hydroxyproline, delivering unique 2,5-dialkylated amino acid analogues that are not accessible by other C–H functionalization methods.

### Graphical Abstract



The development of a carbamate type 3-pentoxythiocarbonyl auxiliary for Ir(I)-catalyzed C–H alkylation of azacycles is described. This method allows for the  $\alpha$ -C–H alkylation of a variety of substituted pyrrolidines, piperidine and tetrahydroisoquinoline through alkylation with alkenes. This method can also be extended to biologically relevant (L)-proline and (L)-trans-hydroxyproline delivering unique 2,5-dialkylated amino acid analogues.

### Keywords

C–H activation; alkylation; cationic Ir(I); pentoxythiocarbonyl; pyrrolidine

Alkyl amines, as in acyclic amines or azacycles, are ubiquitous building blocks in biologically and medically relevant compounds<sup>[1]</sup>. Indeed, the myriad of  $\alpha$ -functionalization methods is a testament of the importance of these fragments<sup>[2]</sup> with

Correspondence to: Jin-Quan Yu.

Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/anie.201xxxxx>.



While the *tert*-butoxy thiocarbonyl directing group proved to be unstable to our reaction conditions, we were pleased to find that its more acid-stable variant, namely the menthoxythiocarbonyl directing group, which was synthesized in two steps from (–)-menthol via intermediate **1** (Scheme 2), gave a promising 45% yield of the desired hydroalkylated product in the presence of cationic [Ir(cod)<sub>2</sub>]BARF catalyst and ethyl acrylate coupling partner. The switch from *tert*-butoxythiocarbonyl to menthoxythiocarbonyl directing groups was based on two main considerations in mind. First, we hypothesized that the sterically imposing nature of the menthoxy group will enhance proximity-inducing effect<sup>[18]</sup> on the initial C–H activation step by Ir(I) catalyst. Second, the menthoxythiocarbonyl directing group can be synthesized in one step from (–)-menthol, a cheap and readily available chiral source with the potential for diastereoselective α-C–H alkylation. Encouraged by the initial hit with cationic [Ir(cod)<sub>2</sub>]BARF, we subsequently undertook screening for the most suitable non-coordinating counterions (BF<sub>4</sub><sup>–</sup>, PF<sub>6</sub><sup>–</sup>, OTf<sup>–</sup>), which were previously found to have a profound impact on catalyst's reactivity<sup>[13]</sup>. To our delight, we found that triflate was the best counteranion, delivering the desired product in 80% yield (Table 1).

With this result in hand, we were initially interested in investigating the coupling partner scope. While initial results were promising with electronically neutral terminal alkenes, it immediately became apparent that the (–)-menthol derived chiral auxiliary only gave rise to slight diastereoselectivity (1.5:1 dr). Addition of chiral bidentate phosphine ligands resulted in a reduction in efficiency without any improvement in diastereoselectivity (see Supporting Information).

Furthermore, with the menthoxythiocarbonyl auxiliary, extension of the current α-C–H alkylation methodology to piperidine was not successful with only trace product being obtained. In light of these limitations, we decided to opt for a simpler, achiral alkoxythiocarbonyl directing group derived from 3-pentanol. We hypothesized that, due to decreased steric bulk, this directing group might be more accommodating to the binding of more conformationally flexible substrates such as piperidines to Ir(I) center. We were delighted to observe that the 3-pentoxythiocarbonyl auxiliary enabled α-C–H alkylation of pyrrolidine with ethyl acrylate in an encouraging 65% yield. Further tuning of reaction conditions revealed the superiority of chlorobenzene as solvent over 1,2-dimethoxyethane due to improved material balance. Furthermore, reduction of the reaction time was key to limiting desulfurization of the thiocarbonyl directing group, improving the yield of the desired alkylated product **4a** to 68% yield (Table 2).

With these optimized conditions in hand, the coupling partner scope was evaluated. A variety of terminal alkenes such as ethyl acrylate (**4a**), styrenes (**4c** and **4d**), simple alkenes (**4e–4g**), allylbenzenes (**4h–4k**) are compatible coupling partners delivering the desired alkylated products in synthetically useful to excellent yields (Table 2). Interestingly, while ethyl acrylate is a competent coupling partner, acrylonitrile was not compatible with the current protocol. We were pleased to find that allyl nitrile, however, was a suitable coupling partner (**4l**) along with range of other alkenes with pendant functionalities such as 4-bromobenzene (**4k**), ketone (**4m**), sulfone (**4n**) and phthalimide (**4o**) delivering products that can be subjected to further synthetic manipulations. For a number of coupling partners, such

as allylbenzenes (**4h–4k**) and alkenes with pendant functional groups (**4l–4o**), only di-alkylated products were observed. Reducing the loading of the alkene coupling partners or reaction time did not change the exclusive formation of di-alkylated products.

Subsequently, the substrate scope for Ir(I)-catalyzed C–H alkylation was evaluated. Various 2- and 3-substituted pyrrolidines (**6a–6c**) gave the desired alkylated products in synthetically useful yields (30–76%). We were pleased to observe that under our optimized reaction conditions, sterically incumbent and medicinally relevant bicyclic (**6d**) and spirocyclic pyrrolidines (**6e**) are suitable substrates. Furthermore, this method can also be extended to (L)-Proline (**6f**) and (L)-Hydroxyproline (**6g**), which feature frequently in investigational antiviral agents<sup>[19]</sup>, delivering for the first time (L)-Proline and (L)-Hydroxyproline analogues with unique 2,5-substitution pattern with moderate diastereoselectivity. These unnatural amino acids will allow for investigation of new chemical space in peptide-based drug discovery which is currently undergoing a renaissance<sup>[20]</sup>.

Finally, we were also able to demonstrate that this method could be used to alkylate piperidine (**6h**) and tetrahydroisoquinoline (**6i**) albeit in lower yields, in the presence of catalytic HBF<sub>4</sub>·Et<sub>2</sub>O as an additive. We also demonstrated the successful  $\alpha$ -C–H alkylation of pyrrolidine **3** with allylacetone on large scale (1.0 mmol) delivering **4m** with the same efficiency (68%, Scheme 3). The directing group could readily be removed by treatment with an aqueous solution of TFA (75% vol/vol) at 65 °C. The intermediate amine was immediately protected as an Fmoc-carbamate delivering **7** in 68% yield over two steps.

In conclusion, we have developed a novel and practical 3-penthoxythiocarbonyl directing group that allows for the  $\alpha$ -C–H alkylation of a range of pyrrolidines, piperidine and tetrahydroisoquinoline. The synthetic utility of this simple carbamate derived directing group was further demonstrated in the  $\alpha$ -C–H alkylation of important medicinally relevant motifs such as (L)-Proline and (L)-Hydroxyproline delivering, for the first time, unique 2,5-di-alkylated proline analogues. Future efforts in our laboratories will focus on the development of enantioselective  $\alpha$ -C–H alkylation utilizing the current 3-penthoxythiocarbonyl directing group in the presence of chiral catalysts.

## Experimental Section

### General procedure for C–H alkylation

A 2-dram vial was charged with substrate **3** (20.2 mg, 0.1 mmol) and the vial was evacuated by passing through alternative cycles of vacuum/Argon. Degassed PhCl (0.5 mL) was then added to the above vial. In a separate 2-dram vial, under an Argon atmosphere (glovebox), [Ir(cod)<sub>2</sub>]OTf (5.7 mg, 0.01 mmol) was added followed by a solution of substrate **3** in degassed PhCl (0.5 mL). Subsequently, alkene (0.8 mmol) was added to the above reaction mixture and the reaction was allowed to stir at 85 °C, under an Argon atmosphere, for 6 or 24 h. Upon completion, the reaction was filtered over Celite® and the filter cake was thoroughly washed with EtOAc: MeOH (9:1 v/v). Solvent was removed *in vacuo* to give a crude residue that was purified by preparative TLC. NB: addition of degassed PhCl and alkene was performed outside of glovebox.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

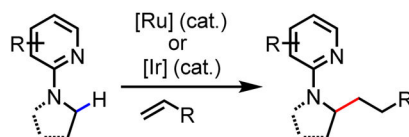
## Acknowledgments

We gratefully acknowledge The Scripps Research Institute and the NIH (NIGMS, 2R01GM084019) for financial support. We thank the University of Sydney (postdoctoral fellowship to A. T. Tran), Dr. Lara Malins for assistance with preparative HPLC purification and Mr. Ruyi Zhu for obtaining HRMS data.

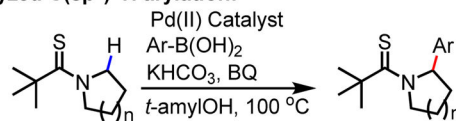
## References

1. Vitaku E, Smith DT, Njardarson JT. *J Med Chem.* 2014; 57:10257. [PubMed: 25255204]
2. a Campos KR. *Chem Soc Rev.* 2007; 36:1069. [PubMed: 17576475] b Mitchell EA, Peschiulli A, Lefevre N, Meerpoel L, Maes BUW. *Chem -Eur J.* 2012; 18:10092. [PubMed: 22829434] c Maes, J., Maes, BUW. *Advances in Heterocyclic Chemistry.* Eric, FVS., Christopher, AR., editors. Vol. 120. Academic Press; 2016. p. 137
3. a Meyers AI, Edwards PD, Rieker WF, Bailey TR. *J Am Chem Soc.* 1984; 106:3270. b Cordier CJ, Lundgren RJ, Fu GC. *J Am Chem Soc.* 2013; 135:10946. [PubMed: 23869442] c Liniger M, Estermann K, Altmann KH. *J Org Chem.* 2013; 78:11066. [PubMed: 24102677]
4. a Asami R, Fuchigami T, Atobe M. *Org Biomol Chem.* 2008; 6:1938–1943. [PubMed: 18480907] b Honzawa S, Uchida M, Sugihara T. *Heterocycles.* 2014; 88:975.
5. a Suga S, Okajima M, Fujiwara K, Yoshida J-i. *J Am Chem Soc.* 2001; 123:7941. [PubMed: 11493082] b Murahashi SI, Komiya N, Terai H, Nakae T. *J Am Chem Soc.* 2003; 125:15312. [PubMed: 14664574] c Wei C, Li CJ. *J Am Chem Soc.* 2002; 124:5638. [PubMed: 12010027] d Wei C, Mague JT, Li CJ. *Proc Natl Acad Sci USA.* 2004; 101:5749. [PubMed: 15067132] e Kumaraswamy G, Murthy AN, Pitchaiah A. *J Org Chem.* 2010; 75:3916. [PubMed: 20446705]
6. a Booth SE, Benneche T, Undheim K. *Tetrahedron.* 1995; 51:3665. b Bertrand S, Glapski C, Hoffmann N, Pete JP. *Tetrahedron Lett.* 1999; 40:3169. c Yoshikai N, Mieczkowski A, Matsumoto A, Ilies L, Nakamura E. *J Am Chem Soc.* 2010; 132:5568. [PubMed: 20361736] d Noble A, MacMillan DWC. *J Am Chem Soc.* 2014; 136:11602. [PubMed: 25026314] e Johnston CP, Smith RT, Allmendinger S, MacMillan DWC. *Nature.* 2016; 536:322. [PubMed: 27535536] f Qin T, Cornella J, Li C, Malins LR, Edwards JT, Kawamura S, Maxwell BD, Eastgate MD, Baran PS. *Science.* 2016; 352:801. [PubMed: 27103669] g Xie J, Yu J, Rudolph M, Rominger F, Hashmi ASK. *Angew Chem, Int Ed.* 2016; 55:9416. h Joe CL, Doyle AG. *Angew Chem, Int Ed.* 2016; 55:4040.
7. a Pastine SJ, Gribkov DV, Sames D. *J Am Chem Soc.* 2006; 128:14220. [PubMed: 17076471] b Prokopcova, Bergman SD, Aelvoet K, Smout V, Herrebout W, Van der Veken B, Meerpoel L, Maes BUW. *Chem -Eur J.* 2010; 16:13063. [PubMed: 20981669] c Peschiulli A, Smout V, Storr TE, Mitchell EA, Elias Z, Herrebout W, Berthelot D, Meerpoel L, Maes BUW. *Chem -Eur J.* 2013; 19:10378. [PubMed: 23780756] d Dastbaravardeh N, Schnürch M, Mihovilovic MD. *Org Lett.* 2012; 14:1930. [PubMed: 22449256] e Kumar NYP, Jeyachandran R, Ackermann L. *J Org Chem.* 2013; 78:4145. [PubMed: 23574572]
8. a Spangler JE, Kobayashi Y, Verma P, Wang DH, Yu JQ. *J Am Chem Soc.* 2015; 137:11876. [PubMed: 26322957] b Jain P, Verma P, Xia G, Yu JQ. *Nat Chem.* 2017; 9:140. [PubMed: 28282045]
9. a Herzon SB, Hartwig JF. *J Am Chem Soc.* 2007; 129:6690. [PubMed: 17474747] b Herzon SB, Hartwig JF. *J Am Chem Soc.* 2008; 130:14940. [PubMed: 18937477] c Eisenberger P, Ayinla RO, Lauzon JMP, Schafer LL. *Angew Chem, Int Ed.* 2009; 48:8361–8365. d Garcia P, Lau YY, Perry MR, Schafer LL. *Angew Chem, Int Ed.* 2013; 52:9144. e Chong E, Brandt JW, Schafer LL. *J Am Chem Soc.* 2014; 136:10898. [PubMed: 25041474]
10. Kubiak R, Prochnow I, Doye S. *Angew Chem, Int Ed.* 2009; 48:1153.
11. Jovel I, Prateptongkum S, Jackstell R, Vogl N, Weckbecker C, Beller M. *Chem Commun.* 2010; 46:1956.

12. a Chatani N, Asaumi T, Yorimitsu S, Ikeda T, Kakiuchi F, Murai S. *J Am Chem Soc.* 2001; 123:10935. [PubMed: 11686697] b Schinkel M, Wang L, Bielefeld K, Ackermann L. *Org Lett.* 2014; 16:1876. [PubMed: 24635222]
13. Tsuchikama K, Kasagawa M, Endo K, Shibata T. *Org Lett.* 2009; 11:1821. [PubMed: 19354323]
14. Kawamorita S, Miyazaki T, Iwai T, Ohmiya H, Sawamura M. *J Am Chem Soc.* 2012; 134:12924. [PubMed: 22816772]
15. a Pan S, Endo K, Shibata T. *Org Lett.* 2011; 13:4692–4695. [PubMed: 21812393] b Pan S, Matsuo Y, Endo K, Shibata T. *Tetrahedron.* 2012; 68:9009.
16. Lahm G, Opatz T. *Org Lett.* 2014; 16:4201–4203. [PubMed: 25057785]
17. a Hodgson DM, Kloesges J. *Angew Chem, Int Ed.* 2010; 49:2900. b Hodgson DM, Mortimer CL, McKenna JM. *Org Lett.* 2015; 17:330. [PubMed: 25535850]
18. Beak P, Meyers AI. *Acc Chem Res.* 1986; 19:356.
19. Stevenazzi A, Marchini M, Sandrone G, Vergani B, Lattanzio M. *Bioorg Med Chem Lett.* 2014; 24:5349. [PubMed: 25455481]
20. Fosgerau K, Hoffmann T. *Drug Disc Today.* 2015; 20:122–128.

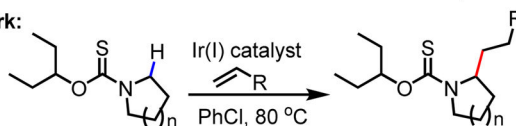
**A. Ru- and Ir-catalyzed C(sp<sup>3</sup>)-H alkylation (Murai, Shibata and Ackermann):**

- Harsh conditions for removal of directing groups
- Limited synthetic utility

**B. Pd(II)-catalyzed C(sp<sup>3</sup>)-H arylation:**

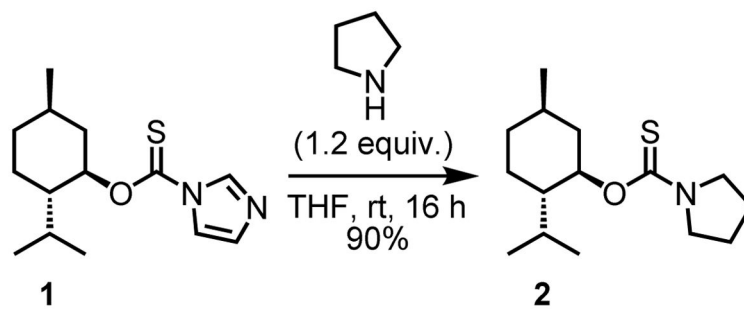
53 examples  
51-99% yield

- Restricted to arylation

**C. This work:**

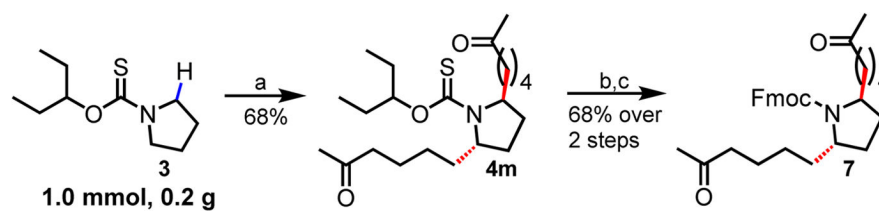
- Novel directing group
- Easy installation and removal
- Unique reactivity

**Scheme 1.**  
α-C-H Functionalizations of Azacycles

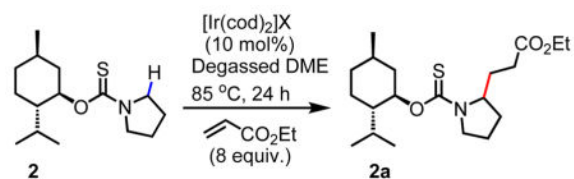


**Scheme 2.**  
Installation of the menthoxythiocarbonyl directing group.



**Scheme 3.**

Large scale  $\alpha$ -C-H alkylation of pyrrolidine and removal of 3-pentoxythiocarbonyl directing group. Reaction conditions: a. **3** (1.0 mmol),  $[\text{Ir}(\text{cod})_2]\text{OTf}$  (10 mol%), allylacetone (8 equiv.), degassed PhCl, Ar (1 atm), 85 °C, 6 h, 68%; b. 75% TFA in  $\text{H}_2\text{O}$ , 65 °C, 2 h; c. Fmoc-OSu (2 equiv.), Dioxane: saturated aqueous  $\text{NaHCO}_3$  solution (1:1 v/v), 16 h, rt, 68% over 2 steps.

**Table 1**Screening of non-coordinating counteranions<sup>[a,b]</sup>

Entry	Counteranion (X)	Yield 2a (%)
1	BARF	45%
2	BF <sub>4</sub> <sup>-</sup>	64%
3	PF <sub>6</sub> <sup>-</sup>	74%
4	OTf <sup>-</sup>	80%

<sup>[a]</sup> Conditions: **2** (0.1 mmol), ethyl acrylate (8.0 equiv), [Ir(cod)<sub>2</sub>]X (10 mol%), degassed 1,2-dimethoxyethane (DME), 80 °C, Ar, 24 h.

<sup>[b]</sup> The yields were determined by <sup>1</sup>H NMR using mesitylene as internal standard.

Table 2

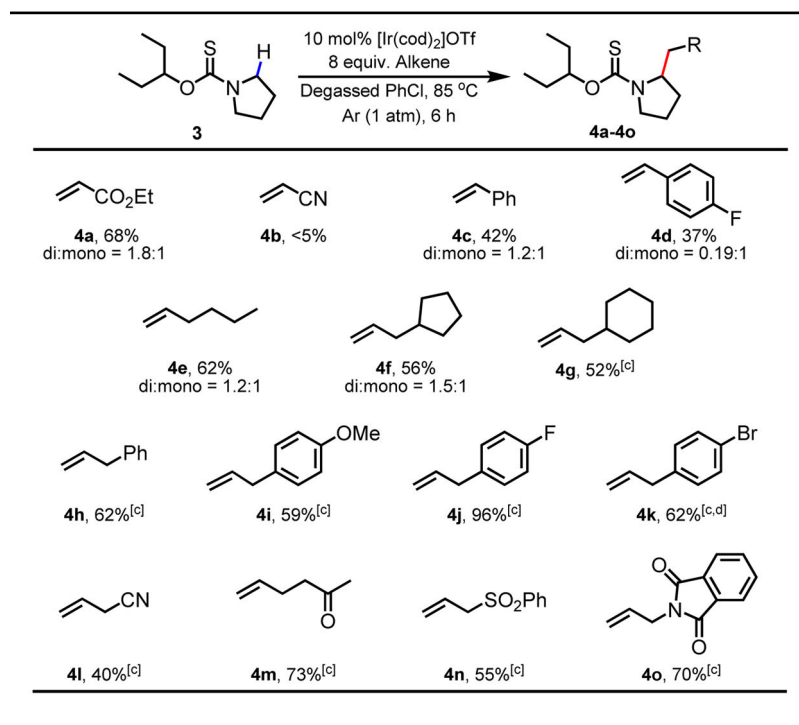
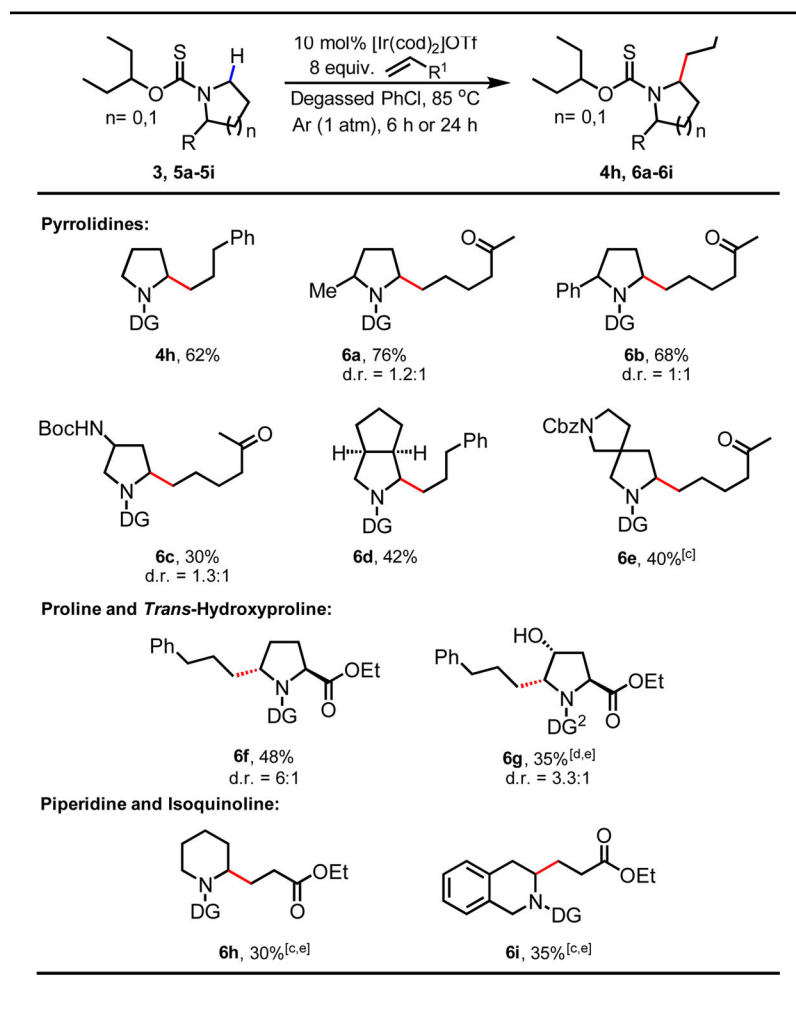
Alkene coupling partner scope for Ir(I)-catalyzed C–H alkylation<sup>[a,b]</sup>.<sup>[a]</sup> Conditions: **2** (0.1 mmol), alkene (8.0 equiv), [Ir(cod)<sub>2</sub>]OTf (10 mol%), degassed PhCl, 80 °C, Ar, 6 h.<sup>[b]</sup> Isolated yields are given here.<sup>[c]</sup> Only di-alkylated products (2,6-disubstituted) were obtained for these substrates.<sup>[d]</sup> Isolated yield given here was the yield of product obtained after removal of directing group.

Table 3

Substrate scope for Ir(I)-catalyzed C–H alkylation<sup>[a,b]</sup>

<sup>[a]</sup> Conditions: **2** (0.1 mmol), alkene (8.0 equiv), [Ir(cod)<sub>2</sub>]OTf (10 mol%), degassed PhCl, 80 °C, Ar, 6 h.

<sup>[b]</sup> Isolated yields are given here.

<sup>[c]</sup> HBF<sub>4</sub>·Et<sub>2</sub>O (10 mol%) was used as an additive.

<sup>[d]</sup> Methoxythiocarbonyl directing group was used.

<sup>[e]</sup> Reaction was run for 24 h.