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## Quantitative Chemical Exchange Saturation Transfer MRI of Intervertebral Disc in a Porcine Model

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### Abstract

**Purpose**—Previous studies have associated low pH in intervertebral discs (IVDs) with discogenic back pain. The purpose of this study was to determine whether quantitative CEST (qCEST) MRI can be used to detect pH changes in IVDs in vivo.

**Methods**—The exchange rate  $k_{sw}$  between glycosaminoglycan (GAG) protons and water protons was determined from qCEST analysis. Its dependence on pH value was investigated in GAG phantoms with varying pH and concentrations. The relationship between  $k_{sw}$  and pH was studied further in vivo in a porcine model on a 3T MR scanner and validated using a pH meter. Sodium lactate was injected into the IVDs to induce various pH values within the discs ranging from 5 to 7.

**Results**—Phantom and animal results revealed that  $k_{sw}$  measured using qCEST MRI is highly correlated with pH level. In the animal studies, the relationship can be described as  $k_{sw} = 9.2 \times 10^6 \times 10^{-pH} + 196.9$ ,  $R^2 = 0.7883$ .

**Conclusion**—The exchange rate between GAG and water protons determined from qCEST MRI is closely correlated with pH value. This technique has the potential to noninvasively measure pH in the IVDs of patients with discogenic pain.

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## Keywords

intervertebral disc; low back pain; CEST; quantitative CEST; gagCEST; pH

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## INTRODUCTION

Lower back pain is a major medical condition estimated to affect up to 85% of the United States population (1). Intervertebral disc (IVD) degeneration is often associated with back pain. Although degenerate discs can be identified using MRI, they do not always cause pain. Therefore, if a patient with lower back pain has several degenerate discs, further examination is required to determine which disc is the source of the pain, prior to a decision of surgical intervention. Standard procedures include discography, during which the suspected discs are pressurized in order to provoke pain. This is a painful procedure that is also known to further accelerate disc degeneration, disc herniation, and loss of disc height and affect the adjacent endplates (2). It is also subjective to variations of the placement of the needle, pressure exerted, and anesthesia. Recent studies have associated low pH with discogenic pain (3,4). pH could potentially serve as a new metabolic biomarker for disco-genic back pain (5).

Chemical exchange saturation transfer (CEST) is an emerging MR technique to measure pH-dependent signal changes (6–9). This technique exploits the constant chemical exchange, which is pH-sensitive, between water protons and solute protons in certain molecules. The chemical exchange rate is dependent on pH values. The solute protons are first magnetization-saturated with a series of frequency-selective radiofrequency (RF) pulses, and after exchanging with water protons, the saturation is indirectly detected in the water signal (10,11).

Glycosaminoglycan (GAG) is a critical component to support the function in the IVD. It has been reported that GAG can be detected by CEST imaging because of its exchangeable hydroxyl protons (12,13). Previous studies have applied gagCEST to detect pH change in the IVD in animal models and patients with degenerative disc disease (14,15). However, CEST contrast is a rather complicated effect. It involves multiple confounding factors, including but not limited to 1) exchange rate between water and GAG protons, which is dependent on the pH; 2) labile proton ratio, which is linearly correlated with GAG concentration; 3) water relaxation parameters  $T_1$  and  $T_2$ ; and 4) the RF irradiation power of the CEST saturation module.

Multiple studies have focused on separating the exchange rate or the labile proton ratio from other confounding factors in the CEST experiments (16–21). Among these methods, quantitative CEST (qCEST) allows for simultaneous measurements of the exchange rate and labile proton ratio. It was developed based on the observation that the CEST effect can be represented as a linear function of  $1/B_1^2$  (22). Multiple CEST experiments were performed with varying  $B_1$  amplitudes for omega plot analysis (22).

Simultaneous measurements of pH value and concentration using qCEST have been shown in creatine phantom studies (18,20,21). Creatine protons have a slow to intermediate exchange rate with water protons (23). However, for GAG protons, which undergo relatively

faster chemical exchange, whether this technique can detect pH changes has not been investigated. In addition, most of the studies were performed on a preclinical scanner using continuous-wave saturation pulse (18,21). No in vivo validation has been performed and potential clinical application is not yet clear.

In this study, we propose an in vivo pH-dependent imaging technique in the IVD using qCEST on a 3T clinical scanner. We tested the technique in GAG phantoms and validated it in vivo in a porcine model using measurement from the tissue pH probe as reference.

## THEORY

Previous investigators have studied the two-pool exchange model using Bloch-McConnell equations, describing the proton exchange between pool “w” (water pool) and pool “s” (solute pool). In this two-pool system,  $f_r$  refers to the labile proton ratio  $M_{0s}/M_{0w}$  and  $k_{sw}$  the exchange rate between solute pool and water pool.  $R_{1w}$ ,  $R_{2w}$ ,  $R_{1s}$ , and  $R_{2s}$  are longitudinal and transverse relaxation rates for water protons and solute protons, respectively.

The conventional CEST asymmetry analysis takes direct difference between the label scan (at the resonant frequency of the solute pool) and reference scan (at the opposite frequency with respect to water). It can be defined as  $CESTR = Z_{label} - Z_{ref}$ , where  $Z_{label}$  and  $Z_{ref}$  are the normalized signal intensity or Z-spectrum for the label scan and reference scan. However, this analysis has its limits with regard to quantitative imaging, because its expression is rather complicated and involves multiple confounding factors.

In recent studies, the inverse CEST difference (CES-TR<sub>ind</sub>) was proposed because of its simplified expression (16,18)

$$\frac{1}{CESTR_{ind}} = \frac{1}{Z_{label}} - \frac{1}{Z_{ref}} \approx \frac{R_{1w}}{f_r \cdot k_{sw}} + \frac{k_{sw} \cdot (R_{2s} + k_{sw}) \cdot R_{1w}}{f_r \cdot k_{sw}} \frac{1}{\omega_1^2}, \quad [1]$$

where  $\omega_1$  is the RF irradiation amplitude.

Equation [1] is only valid for continuous-wave CEST saturation. When pulsed saturation is applied in CEST experiments, Eq. [1] can be written as (20)

$$\frac{1}{CESTR_{ind}} \approx \frac{R_{1w}}{DC \cdot f_r \cdot k_{sw} \cdot c_1} + \frac{k_{sw} \cdot (R_{2s} + k_{sw}) \cdot R_{1w} \cdot c_2^2}{DC \cdot f_r \cdot k_{sw} \cdot c_1} \frac{1}{\omega_1^2}, \quad [2]$$

where DC represents the duty cycle and  $c_1$  and  $c_2$  represent the shape of Gaussian saturation pulses ( $c_1 = \sigma \sqrt{2\pi}/t_p$ ,  $c_2 = c_1 \sqrt{\sqrt{2}}$ ;  $\sigma$  and  $t_p$  are the width and length of the Gaussian pulse). Note that  $\omega_1$  here is defined as the average RF irradiation amplitude of one Gaussian pulse (i.e.,  $\omega_1 = \text{flip angle}/\text{pulse duration}$ ). This equation is equivalent to Equation 18 by Meissner et al. (20).

In this expression,  $1/\text{CESTR}_{\text{ind}}$  is described as a linear function of  $1/\omega_1^2$ . By measuring  $\text{CESTR}_{\text{ind}}$  with different RF irradiation amplitude, we can calculate the slope  $m$  and intercept  $n$  and eventually estimate  $k_{sw}$  and  $f_r$ :

$$k_{sw} = \frac{\sqrt{R_{2s}^2 + \frac{4m}{n \cdot c_2^2}} - R_{2s}}{2} \quad [3]$$

$$f_r = \frac{R_{1w}}{k_{sw} \cdot n \cdot c_1 \cdot DC} \quad [4]$$

$R_{1w}$  can be measured using  $T_1$  mapping techniques.  $R_{2s}$  of GAG is approximately  $200 \text{ s}^{-1}$  (24).

Note that Eq. [1] is a simplified expression that describes the steady state signal of CEST imaging. When performing qCEST experiments, RF saturation pulses need to be long enough to ensure the steady state is reached. The simplification only holds for dilute CEST agents undergoing slow and intermediate chemical exchange.

## METHODS

### Phantom

Two sets of phantoms containing GAG prepared from chondroitin sulphate A (Sigma-Aldrich, St. Louis, Missouri, USA) and phosphate-buffered saline with varying pH values and concentrations were prepared. For the pH set, the GAG concentration was fixed at 60 mM and pH was titrated to 5.8, 6.1, 6.4, 6.7, and 7.0. For the concentration phantom, we used various GAG concentrations (100, 80, 60, 40, and 20 mM) and titrated the pH to 7.0. The solution was then transferred to 15 mL tubes. These 10 tubes were put in a phantom holder filled with water.

### In Vitro MRI Experiments

Imaging experiments were performed at room temperature on a 3T clinical scanner (Magnetom Verio; Siemens Healthcare, Erlangen, Germany). All images were acquired with a slice thickness of 8 mm, field of view of  $160 \times 160 \text{ mm}^2$ , and imaging matrix of  $128 \times 128$ .

CEST MRI was performed with pulsed RF saturation turbo spin echo (TSE) sequence (pulse repetition time [TR]/echo time [TE] = 16,000/12 ms; two averages). The CEST saturation module consisted of 39 Gaussian-shaped pulses, with a duration  $t_p = 80 \text{ ms}$  for each pulse and an interpulse delay  $t_d = 80 \text{ ms}$  (duty cycle = 50%, total saturation time  $T_{\text{sat}} = 6240 \text{ ms}$ ) at saturation flip angle  $900^\circ$ ,  $1500^\circ$ ,  $2100^\circ$ , and  $3000^\circ$  [ $B_1$  amplitudes = flip angle/ $(\gamma t_p)$  = 0.73, 1.22, 1.71, and 2.45  $\mu\text{T}$ ; Gaussian saturation pulse parameters  $c_1 = 0.50$ ,  $c_2 = 0.59$ ]. Z-spectrum was acquired with 10 different saturation frequencies at  $\pm 1.6$ ,  $\pm 1.3$ ,  $\pm 1.0$ ,  $\pm 0.7$ , and

$\pm 0.4$  ppm. The  $B_0$  field was corrected using a water saturation shift referencing (WASSR) map (25).

$T_1$ -weighted MR images were acquired by an inversion recovery TSE sequence with 10 different inversion delays (inversion time [TI] = 50, 150, 350, 700, 1050, 1400, 2000, 2500, 3000, and 4000 ms; TR/TE=6000/12 ms).  $T_2$ -weighted MR images were acquired using a TSE sequence with varying echo delays (TE = 12, 24, 48, 97, 205, and 399 ms; TR = 6000 ms).

### Animal Preparation

All animal-related procedures were approved by the Institutional Animal Care and Use Committee at Cedars-Sinai Medical Center. Four female Yucatan minipigs (S&S Farms, Ramona, California, USA) were used. Following an 18-hour preoperative fast, the pigs were sedated with intramuscular drugs (acepromazine 0.25 mg/kg, ketamine 20 mg/kg, and atropine 0.02–0.05 mg/kg) and were then injected intravenously with propofol (2 mg/kg) to induce full anesthesia. The trachea was intubated and anesthesia was maintained using 1%–3.5% isoflurane inhaled via the tracheal tube for the duration of the procedure. Under fluoroscopic guidance, three MR-compatible 14G coaxial needles (Invivo, Gainesville, Florida, USA) were inserted into the mid-substance of lumbar discs L1/L2, L3/L4, and L5/L6. These lumbar discs were injected with different concentrations of sodium lactate (Sigma-Aldrich) in order to induce a gradient of pH values within the discs ranging from 5 to 7, as described by Melkus et al. (15) and in accordance with pH values measured within patients' pathological discs (26). After intradiscal injection, exact pH values inside the discs were measured using a custom-made needle-shaped tissue pH probe (Warner Instruments, Hamden, Connecticut, USA) which was inserted through the MR-compatible needle shortly before the MR scan. Lumbar disc L2/L3 was also scanned as the control disc. Its pH value was measured immediately after the animal was euthanized.

### In Vivo MRI Experiments

Imaging experiments were performed on a 3T clinical scanner (Magnetom Verio; Siemens Healthcare, Erlangen, Germany). Animals were placed in the right decubitus position with body array coils centered on the posterior aspect spinous process. Throughout the imaging procedures, anesthesia was maintained with isoflurane (1%–3.5%).

CEST MRI was performed using a two-dimensional reduced field of view TSE CEST sequence (TR/TE = 10,500/10 ms, two averages, single shot). Reduced field of view can effectively suppress bowel motion artifacts and increase scan efficiency (27). For each IVD, images were acquired in the axial plane with a slice thickness of 3 mm, field of view of  $100 \times 40$  mm<sup>2</sup>, and spatial resolution of  $0.8 \times 0.8$  mm<sup>2</sup>. CEST saturation module consists of 39 Gaussian-shaped pulses, with a duration  $t_p = 80$  ms for each pulse and an interpulse delay  $t_d = 80$  ms (duty cycle = 50%, total saturation duration  $T_s = 6240$  ms) at saturation flip angle  $90^\circ$ ,  $1500^\circ$ ,  $2100^\circ$ , and  $3000^\circ$  [ $B_1$  amplitudes = flip angle/ $(\gamma t_p) = 0.73, 1.22, 1.71, \text{ and } 2.45$   $\mu\text{T}$ ; Gaussian saturation pulse parameters  $c_1 = 0.50, c_2 = 0.59$ ]. Z-spectrum was acquired with 10 different saturation frequencies at  $\pm 1.6, \pm 1.3, \pm 1.0, \pm 0.7, \text{ and } \pm 0.4$  ppm. The scan

time of the CEST experiment for each RF irradiation amplitude was approximately 6 min. The  $B_0$  field was corrected using WASSR.

$T_1$ -weighted MR images were acquired using an inversion recovery TSE sequence with seven varying TI (50, 150, 350, 700, 1050, 1400, and 2000 ms). Images were acquired at the same slice position as the CEST MRI sequence (TR/TE = 6000/12 ms; 1 average; FOV =  $200 \times 200 \text{ mm}^2$ ; spatial resolution =  $0.8 \times 0.8 \times 3 \text{ mm}^3$ ; scan time =  $\sim 2.5 \text{ min}$ ).

## Data Analysis

Postprocessing was performed with custom-written programs in MATLAB (MathWorks, Natick, Massachusetts, USA).  $\text{CESTR}_{\text{ind}}$  was calculated according to Eq. [1] after  $B_0$  correction at 1.0 ppm [ $Z_{\text{lab}} = Z(+1.0 \text{ ppm})$ ,  $Z_{\text{ref}} = Z(-1.0 \text{ ppm})$ ]. Linear regression was used to perform  $\Omega$ -plot analysis between  $1/\text{CESTR}_{\text{ind}}$  and  $1/\omega_1^2$  to obtain the slope and intercept. The exchange rate  $k_{\text{sw}}$  and labile proton ratio  $f_r$  were calculated afterward following Eqs. [3] and [4]. These calculations were performed pixel-by-pixel and by region of interest (ROI). The  $T_1$  maps were obtained by pixel-by-pixel least-squares fitting of the signal equation  $I = I_0[1 - (1 + \eta) \cdot \exp(-\text{TI}/T_1)]$ , where  $I$  is the signal intensity, TI is the inversion time, and  $\eta$  is the inversion efficiency. The  $T_2$  maps were obtained by fitting the signal equation  $I = I_0 \cdot \exp(-\text{TE}/T_2)$ , where  $I$  is the signal intensity and TE is the echo time.

## RESULTS

### Phantom

In Figure 1, we evaluated the relationship between  $1/\text{CES-TR}_{\text{ind}}$  and  $1/\omega_1^2$  in tubes with varying GAG concentration and pH values.  $1/\text{CESTR}_{\text{ind}}$  is the average signal within the region-of-interest (ROI) of each tube. In all tubes,  $1/\text{CESTR}_{\text{ind}}$  can be represented as a linear function of  $1/\omega_1^2$ . This experimental finding is consistent with Eq. [2].

In addition, pixel-wise mapping of chemical exchange rate  $k_{\text{sw}}$  and labile proton ratio  $f_r$  were reconstructed (Fig. 2). One can appreciate the dependence of chemical exchange rate on pH (Fig. 2a) and labile proton ratio on GAG concentrations (Fig. 2b). Quantitatively, the chemical exchange rate can be described as  $k_{\text{sw}} = 1.5 \times 10^8 \times 10^{-\text{pH}} + 252.0$ ,  $R^2 = 0.9508$  (Fig. 2c). This follows an acid catalyzed chemical exchange formula (28). The labile proton ratio is linearly correlated with GAG concentration (Fig. 2d). It can be represented as  $f_r = 4.6 \times 10^{-5} [\text{GAG}] - 4.4 \times 10^{-5}$  ( $R^2 = 0.9869$ ), where [GAG] is the GAG concentration in mM. The error bars in Figure 2c and 2d represent the standard deviation of all the pixels within the ROI of each tube for  $k_{\text{sw}}$  and  $f_r$ , respectively. These experimental results encouraged in vivo application of qCEST technique.

### Animal Studies

Sixteen IVDs were investigated in this study, three of which were excluded because the needle went through both sides of the IVD and caused morphological damage. The pH values of the studied IVDs after sodium lactate injection ranged from 5.0 to 7.2.

Figure 3 shows the anatomical images of one representative pig's lumbar IVDs and the corresponding exchange rate maps. As shown in the figure, the exchange rate was higher in the IVDs with lower pH values. Within each disc, there was some inhomogeneity in the exchange rate map. This is partially because at the current SNR, we cannot guarantee accurate measurement for a signal pixel. However, the average value within the ROI of each IVD will provide more reliable measurement. This is because the SNR will increase after averaging all pixels that are in the similar pH environment.

As shown in Figure 4a, we evaluated the relationship between  $1/\text{CESTR}_{\text{ind}}$  and  $1/\omega_1^2$  in representative IVDs with different pH values (5.0, 5.8, and 6.7). Similar to the phantom studies,  $1/\text{CESTR}_{\text{ind}}$  can be represented as a linear function of  $1/\omega_1^2$ . In Figure 4b, we took the average exchange rate of each disc and evaluated its relationship with the corresponding pH value, which was obtained by directly measuring the intradiscal pH value using a pH probe. The exchange rate can be described by an acid catalyzed chemical exchange formula  $k_{sw} = 9.2 \times 10^6 \times 10^{-\text{pH}} + 196.9$ ,  $R^2 = 0.7883$ . However, because of the difficulty in determining the location of the pH probe, there could be some uncertainty of the pH values measured by the tissue pH probe.

## DISCUSSION

In this study, we investigated the feasibility of qCEST technique to detect pH changes in IVDs in vivo on a 3T MR scanner. Phantom studies showed that the approximations used in qCEST analysis hold true for GAG and that the exchange rate determined from qCEST analysis is dependent on pH levels of GAG solutions. The relationship between the exchange rate and pH was further studied in the porcine spine studies. The results showed the exchange rate can be described as a function of pH using acid catalyzed proton chemical exchange formula. To our knowledge, this is the first in vivo study to show the validity of qCEST analysis using tissue pH meter as reference.

Previous studies have investigated the pH dependence of gagCEST. Even though the GAG concentration can be corrected using  $T_{1p}$ , water relaxation parameters  $T_1$  and  $T_2$  still contribute to the gagCEST signal (14). qCEST analysis, on the other hand, has been shown to detect pH changes independent of  $T_1$ ,  $T_2$ , and concentration in numerical simulations and in phantom studies (18,20). It is a more reliable approach to measure pH changes in the IVD, because  $T_1$  and  $T_2$  change significantly after disc degeneration (29). In the present in vivo study, we established a relationship between exchange rates and pH levels, which can be potentially applied in future studies to translate exchange rates to pH levels.

Pulsed CEST saturation pulses were used because this study was performed on a 3T clinical MR scanner. Pulsed qCEST analysis is even more complicated because of the constantly changing RF irradiation amplitude. Pulsed CEST experiments normally report the irradiation power as the equivalent continuous-wave  $B_1$  field strength. However, the proton exchange in pulsed CEST experiments is rather complicated. Simply integrating the equivalent continuous-wave  $B_1$  field strength will cause errors in estimating the exchange rate and labile proton ratio. Meissner et al. (20) developed an analytical solution for pulsed CEST experiment. This enables more accurate quantitative results of pulsed CEST experiments.

It should be noted that because of the simplification of the in vivo situation, there could be some potential systemic error in estimating the exchange rate. One error source is magnetization transfer effects are not considered. This will lead to the underestimation of CEST effects, especially when the RF irradiation amplitude is higher, which means the slope can be also underestimated. Therefore, there could be underestimation of exchange rate when magnetization transfer effects are not considered. Another error source is the approximation of the Gaussian-shaped saturation pulses. Even though we have made some corrections as discussed above, the performance of the Gaussian-shaped pulses is not fully simulated, especially in the case of intermediate to fast chemical exchange.

In this study, we explored the relationship between exchange rates and pH levels in both phantom studies and in vivo animal studies. However, the results are not exactly the same. One reason is these two studies were performed at different temperatures ( $\sim 20^{\circ}\text{C}$  for phantom studies and  $\sim 38^{\circ}\text{C}$  for animal studies). Another possible reason is the GAG protons in the IVD experience a more complicated environment. In addition to CEST effects, magnetization transfer effects are also present in the IVD from semisolid components such as macromolecules, which could affect the qCEST analysis (16). As discussed above, because we did not consider magnetization transfer effects in our model, there is systemic error of underestimating the exchange rate in the in vivo studies, which explains the discrepancy between phantom and animal studies.

A potential limitation of this study is the long scan time (30–40 min for one IVD). Regular CEST experiments are relatively slow because of long TR, multiple averages, and so forth. In addition, qCEST analysis requires 1) long RF saturation time (6 s in our study) to achieve the steady state and 2) multiple CEST experiments with varying RF irradiation amplitudes to perform the  $\Omega$ -plots. Compressed sensing and parallel imaging techniques can be used to accelerate qCEST experiments (30).

It is known that CEST imaging is prone to  $B_1$  inhomogeneity. Not knowing the exact  $B_1$  field may cause errors in the estimation of exchange rate  $k_{sw}$ . We performed a pilot study and found that the  $B_1$  field is relatively homogenous within the small ROI (nucleus pulposus) for all discs; this is why we did not acquire a  $B_1$  map for every IVD. It should be noted though, that  $B_1$  inhomogeneity issues need to be carefully considered in qCEST imaging to avoid potential errors.

The manipulation of pH levels in the IVDs by injecting sodium lactate mimics the degeneration condition only to a limited extent. In addition to pH change, disc degeneration is also correlated with a loss of GAG and water content in the nucleus pulposus (31). GAG loss will significantly lower the CEST values (32), and the dehydration process will cause a change of MR relaxation parameters (33,34). In order to better simulate the degeneration situation, qCEST experiments could be performed in the disc degenerative porcine model (35) and in patients with discogenic pain (14).



## CONCLUSIONS

Our work demonstrates the feasibility of in vivo qCEST analysis of GAG in IVDs. The validation study shows that the exchange rate determined from qCEST analysis is closely correlated with pH value, and can be used to noninvasively measure pH in IVDs. qCEST technique has the potential to provide additional information on IVD physiology and help gain insight into the pathogenesis of low back pain and its underlying degenerative processes.

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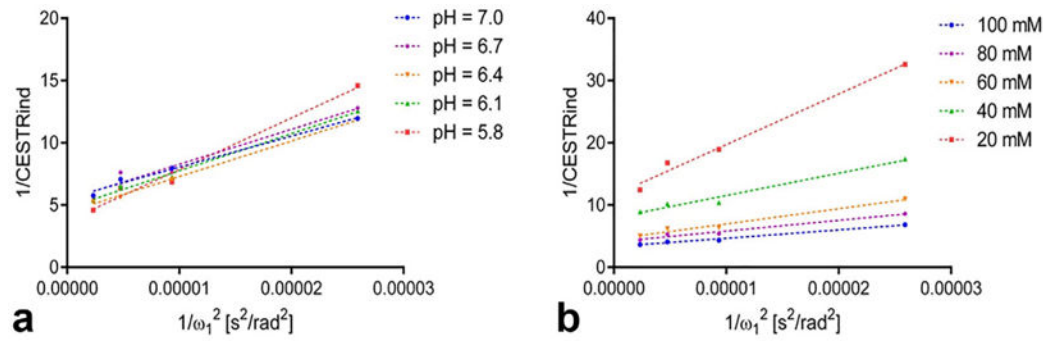
Zulma Gazit, Gadi Pelled, and Dan Gazit are founders and equity holders in TavorStem Therapeutics. TavorStem Therapeutics provided no funding for this study.

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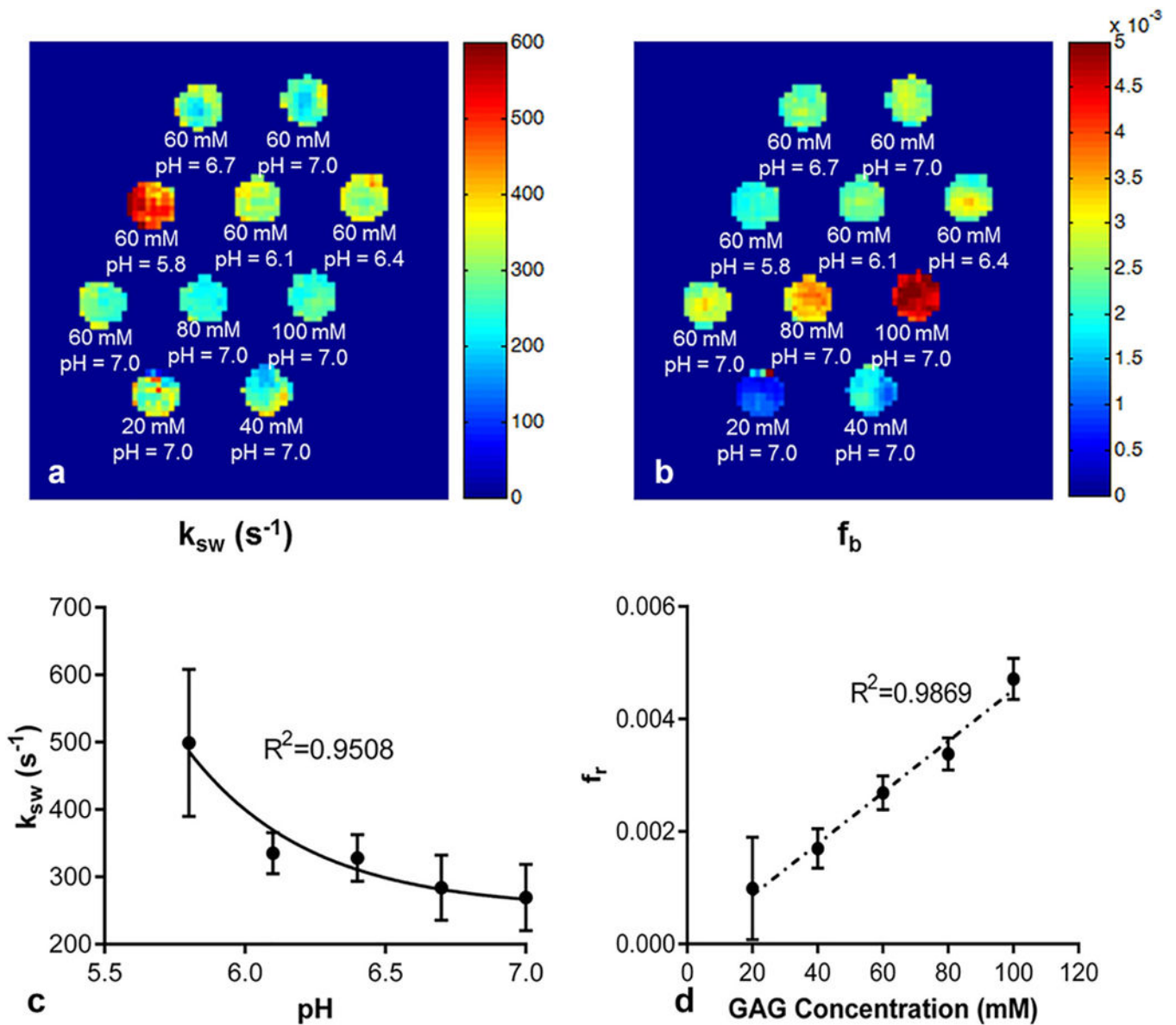
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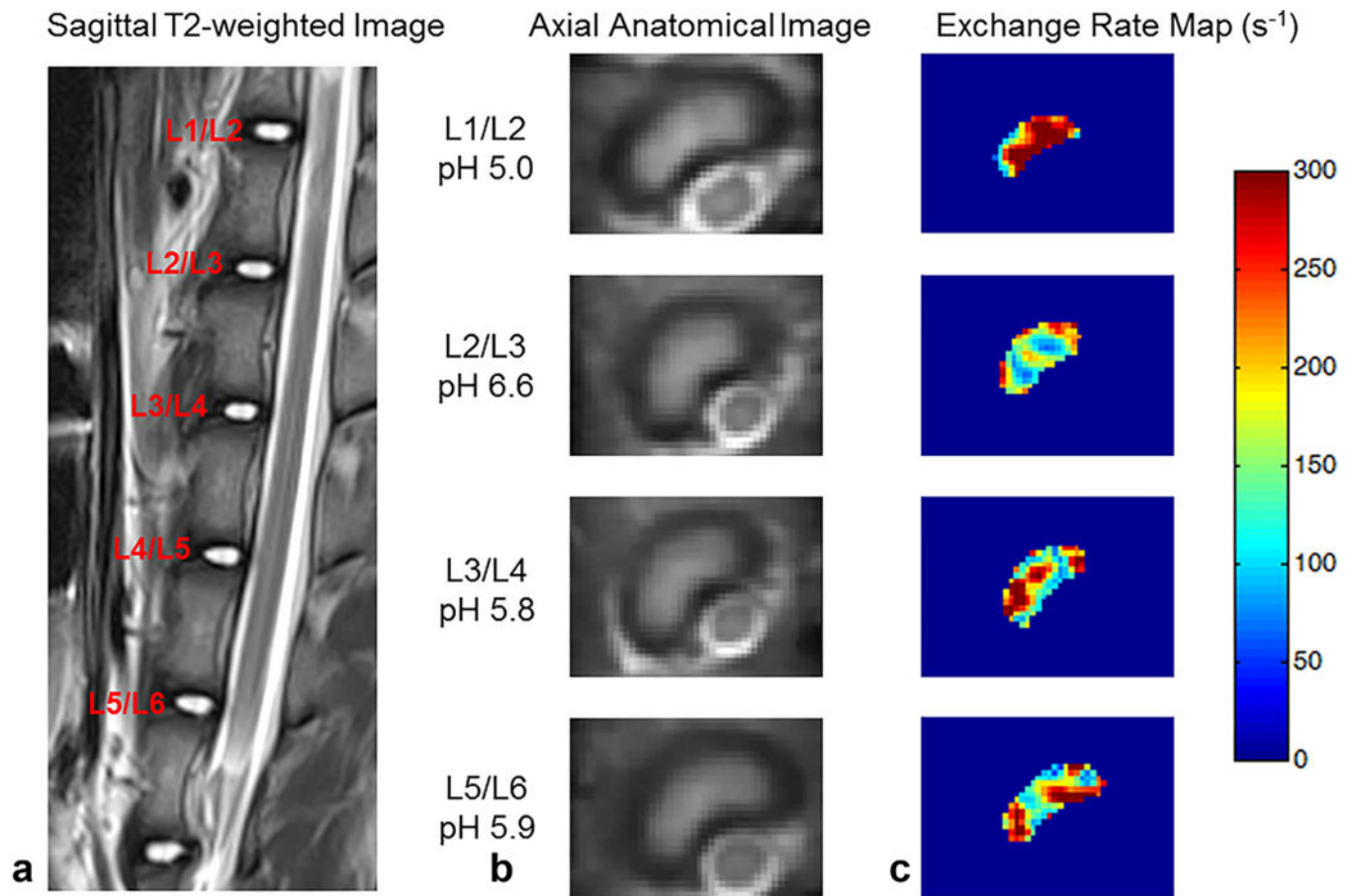
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**FIG. 1.**

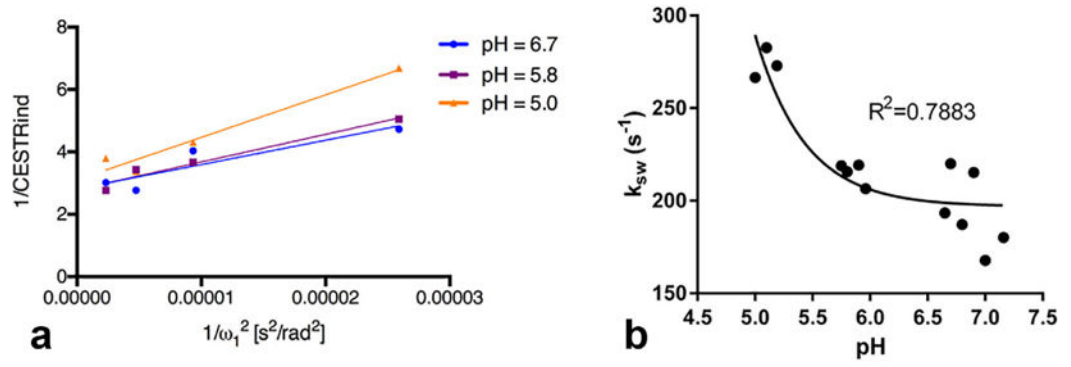
$\Omega$ -plots analysis of (a) phantoms with the same concentration (60 mM) but varying pH values (5.8, 6.1, 6.4, 6.7, and 7.0) and (b) phantoms with the same pH value (7.0) but varying GAG concentrations (20, 40, 60, 80, and 100 mM).



**FIG. 2.** Quantitative results of the phantom study. (a) Pixel-wise mapping of labile proton exchange rate. (b) Pixel-wise mapping of labile proton ratio. (c) Chemical exchange rate as a function of pH. (d) Labile proton ratio as a function of GAG concentration. The error bars in panels (c) and (d) represent the standard deviation of all the pixels within the ROI of each tube for  $k_{sw}$  and  $f_r$ , respectively.



**FIG. 3.** Representative images of IVDs and corresponding exchange rate maps in one pig. (a) T<sub>2</sub>-weighted image in the sagittal plane. (b) Axial anatomical images of corresponding IVDs. (c) Exchange rate maps of corresponding IVDs. The IVDs with lower pH tend to have a higher exchange rate.



**FIG. 4.** (a)  $\Omega$ -plot analysis of representative IVDs with varying pH values (5.0, 5.8, and 6.7). (b) Chemical exchange rate as a function of pH in the animal studies.