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New ursane triterpenoids from *Ficus pandurata* and their binding affinity for human cannabinoid and opioid receptors

Amgad I. M. Khedr¹, Sabrin R. M. Ibrahim^{2,3}, Gamal A. Mohamed^{4,5}, Hany E. A. Ahmed^{2,6}, Amany S. Ahmad³, Mahmoud A. Ramadan³, Atef E. Abd El-Baky⁷, Koji Yamada⁸, and Samir A. Ross⁹

¹Department of Pharmacognosy, Faculty of Pharmacy, Port Said University, Port Said 42526, Egypt

²Department of Pharmacognosy and Pharmaceutical Chemistry, College of Pharmacy, Taibah University, Al Madinah Al Munawarah 30078, Saudi Arabia

³Department of Pharmacognosy, Faculty of Pharmacy, Assuit University, Assuit 71526, Egypt

⁴Department of Natural Products and Alternative Medicine, Faculty of Pharmacy, King Abdulaziz University, Jeddah 21589, Saudi Arabia

⁵Department of Pharmacognosy, Faculty of Pharmacy, Al-Azhar University, Assuit Branch, Assuit 71524, Egypt

⁶Pharmaceutical Organic Chemistry Department, Faculty of Pharmacy, Al-Azhar University, Cairo, Egypt

⁷Department of Biochemistry, Faculty of Pharmacy, Port Said University, Port Said 42526, Egypt

⁸Garden for Medicinal Plants, Graduate School of Biomedical Sciences, Nagasaki University, Bunkyo-machi 1-14, Nagasaki 852-8521, Japan

⁹National Center for Natural Products Research, and Department of Pharmacognosy, School of Pharmacy, University of Mississippi, University, MS 38677, USA

Abstract

Phytochemical investigation of *Ficus pandurata* Hance (Moraceae) fruits has led to the isolation of two new triterpenoids, ficupanduratin A [1 β -hydroxy-3 β -acetoxy-11a-methoxy-urs-12-ene] (11) and ficupanduratin B [21a-hydroxy-3 β -acetoxy-11a-methoxy-urs-12-ene] (17), along with 20 known compounds: a-amyrin acetate (1), a-amyrin (2), 3 β -acetoxy-20-taraxasten-22-one (3), 3 β -acetoxy-11a-methoxy-olean-12-ene (4), 3 β -acetoxy-11a-methoxy-12-ursene (5), 11-oxo-a-amyrin acetate (6), 11-oxo- β -amyrin acetate (7), palmitic acid (8), stigmast-4,22-diene-3,6-dione (9), stigmast-4-ene-3,6-dione (10), stigmasterol (12), β -sitosterol (13), stigmast-22-ene-3,6-dione (14), stigmastane-3,6-dione (15), 3 β ,21 β -dihydroxy-11a-methoxy-olean-12-ene (16), 3 β -hydroxy-11a-methoxy-urs-12-ene (20), 11a,21a-dihydroxy-3 β -acetoxy-urs-12-ene (21), and β -

Correspondence to: Sabrin R. M. Ibrahim.

Compliance with ethical standards

Conflict of interest The authors report no conflicts of interest with respect to this work.

sitosterol-3-*O*- β -D-glucopyranoside (**22**). Compound **21** is reported for the first time from a natural source. The structures of the **20** compounds were elucidated on the basis of IR, 1D (¹H and ¹³C), 2D (¹H–¹H COSY, HSQC, HMBC and NOESY) NMR and MS spectroscopic data, in addition to comparison with literature data. The isolated compounds were evaluated for their antimicrobial, anti-malarial, anti-leishmanial, and cytotoxic activities. In addition, their radioligand displacement affinity on opioid and cannabinoid receptors was assessed. Compounds **4**, **11**, and **15** exhibited good affinity towards the CB2 receptor, with displacement values of 69.7, 62.5 and 86.5 %, respectively. Furthermore, the binding mode of the active compounds in the active site of the CB2 cannabinoid receptors was investigated through molecular modelling.

Keywords

Ficus pandurata; Triterpenes; Cannabinoid receptors; Opioid receptors; Anti-malarial; Anti-leishmanial

Introduction

Medicinal plants have been used extensively as a source of numerous active constituents for treating human diseases and have a high therapeutic value (Nostro et al. 2000). Ficus (Moraceae), which comprises more than 800 species, is a large genus of trees or shrubs cultivated for their ornamental leaves and edible fruits (Metcalf and Chalk 1950; Baily 1963; Lin and Wunder 1997). Plants of the Ficus genus are distributed in tropical and subtropical regions (Mbosso et al. 2012) and known to be rich sources of triterpenes and sterols (Tuyen et al. 1998; Kuo and Chiang 1999; Ragasa et al. 1999; Chiang et al. 2001; Chiang and Kuo 2001, 2002; Kuo and Lin 2004; Chiang et al. 2005; Sisy and Abeba 2005; Parveen et al. 2009). In Ayurvedic and traditional Chinese medicine (TCM), Ficus species are widely used for the treatment of various diseases such as inflammation, diabetes, tumour and malaria (Singh et al. 2011). In Egypt, many Ficus species are found in streets, gardens and parks, as well as outside canal banks (Harborne 1973; Riffle 1998). The fruits of *E carica* L. and *E*. sycomorus L. are two of the most popular fruits eaten by Egyptians. In traditional Egyptian medicine, Ficus species are used for anti-diabetic, hypotensive and anti-cough applications, as well as in the treatment of respiratory disorders and certain skin diseases (Chopra et al. 1950; Edlin and Nimmo 1978; Mousa et al. 1994). Large quantities of latex, which is a source of rubber, have been found in the wood of Ficus species, representing one of the largest economical uses of Ficus in Egypt (Edlin and Nimmo 1978). F. pandurata Hance, known as Xiao xianggou in Lishui District (Zhejiang, China), can be used for removing dampness and strengthening the spleen. It is also used in TCM to treat gout, arthritis, hyperuricemia, and indigestion (Chen et al. 2005; Lei and Li 2007). To the best of our knowledge, there are few reports about the phytochemical composition and biological activities of *F. pandurata* (Jeong and Lachance 2001; Basudan et al. 2005; Vaya and Mahmood 2006; Zhang et al. 2015). Our previous phytochemical study of *F. pandurata* led to the isolation of sterols and triterpenes (Ramadan et al. 2009; Ahmed 2010; Khedr et al. 2015). As a continuation of our investigation, we report herein the isolation and structural elucidation of two new triterpenes, ficupanduratins A (11) and B (17), along with 20 known compounds from the fruit of F. pandurata H (Fig. 1). The structures of these compounds

were verified by various spectroscopic methods. The anti-microbial, anti-malarial, antileishmanial, and cytotoxic activities of the isolated compounds were evaluated. In addition, their radioligand displacement affinity on opioid and cannabinoid receptors has been tested.

Materials and methods

General experimental procedures

Optical rotations were measured with a JASCO DIP-370 digital polarimeter. Melting points were determined using an Electrothermal 9100 Digital Melting Point apparatus (Electrothermal Engineering Ltd., Essex, England). IR spectra were obtained using a JASCO FT/IR-410 spectrophotometer. HRFABMS were recorded on a JMS DX-303 spectrometer (JEOL Ltd., Japan) using *m*-nitrobenzyl alcohol or Magic Bullet as a matrix. HRESIMS were measured on a Finnigan MAT 95 mass spectrometer. ESIMS were recorded on a Finnigan MAT TSQ-7000 triple stage quadrupole mass spectrometer. EIMS were recorded on a JEOL JMS-SX/SX 102A mass spectrometer. 1D and 2D NMR spectra were recorded with a Unity Plus 500 spectrometer (Varian Inc., USA) operating at 500 MHz for ¹H and 125 MHz for ¹³C. ¹H NMR chemical shifts are expressed in δ values referring to the solvent peak of $\delta_{\rm H}$ 7.26 for CDCl₃ and $\delta_{\rm H}$ 7.19, 7.55 and 8.71 for C₅D₅N; coupling constants are expressed in Hz. ¹³C NMR chemical shifts are expressed in δ values referring to the solvent peaks δ_C 77.23 for CDCl_3 and δ_C 123.5, 135.5 and 149.9 for C_5D_5N. Sephadex LH-20 (0.25–0.1 mm, Pharmacia Fine Chemical Co. Ltd, Piscataway, NJ), Wakogel C-300 (Wako Pure Chemical Industries, Ltd, Japan, 45–75 µm) and Cosmosil 5C18-140 PREP (Nacalai Tesque, No. 379-34) were used for column chromatography. Pre-coated Kieselgel 60 F254 silica gel plates (0.25 mm, Merck, Darmstadt, Germany) and RP-18 F_{254s} plates (0.04–0.063 mm, Merck, Darmstadt, Germany) were used for thin-layer chromatographic analysis. Semipreparative HPLC was performed on a Develosil C30-UG-5 column (250×4.6 mm, Nomura Chemical Co., Aichi, Japan) and a Wakosil-II 5sil-100 column (150×4.6 mm 'B', Wako Pure Chemical Industries Ltd., Japan) at a flow rate of 1.5 mL/min, equipped with a TOSOH RI-8020 detector and a JASCO BIP-I HPLC pump.

Plant material

Fruits of *F. pandurata* were collected in May 2013 from authorized trees growing in the garden of the Faculty of Pharmacy, Assiut University. The plant was taxonomically identified by Prof. Dr. Salah EL-Nagar (Professor of Botany, Department of Botany, Faculty of Science, Assiut University, Egypt). A voucher specimen (registration code FPF-2013) was deposited at the Department of Pharmacognosy, Faculty of Pharmacy, Al-Azhar University, Assiut, Egypt.

Extraction and isolation

The air-dried and powdered fruits (1.1 kg) were exhaustively extracted by cold percolation with 70 % MeOH (4 × 4 L) at room temperature. The alcoholic extract was concentrated under vacuum to afford a dark brown residue (95 g). The residue was mixed with 500 mL distilled H₂O and subjected to solvent fractionation using EtOAc and *n*-BuOH; the fractions were separately concentrated to yield EtOAc (37 g), *n*-BuOH (21 g) and aqueous (28 g) extracts. The EtOAc extract was subjected to vacuum liquid chromatography (VLC) using *n*-

hexane:EtOAc and EtOAc:MeOH gradient to give 20 fractions (F1-F20). Fraction F1 (3.4 g, 5 % EtOAc/n-hexane) was subjected to silica gel column chromatography (n-hexane:EtOAc 10:0 to 9:1), yielding eight sub-fractions (A–H). Sub-fraction F1-B (360 mg) was subjected to semi-preparative HPLC (Wakosil-II 5sil-100) using 100 % MeOH to give compounds 1 $(R_t 55 \text{ min}, 125 \text{ mg}, \text{white needles})$ and 2 $(R_t 43 \text{ min}, 160 \text{ mg}, \text{white amorphous powder})$. Semi-preparative HPLC (Wakosil-II 5sil-100) using 100 % MeOH of sub-fraction F1-F (295 mg) gave 3 (R_t 14 min, 20 mg, white needles), 4 (R_t 23 min, 50 mg, white amorphous solid) and 5 (R_t 34 min, 78 mg, white amorphous solid). Sub-fraction F1-H (95 mg) was chromatographed by semi-preparative HPLC (Develosil C-30-UG-5) using 100 % MeOH to yield 6 (R_t 22 min, 12 mg, colourless amorphous solid) and 7 (R_t 38 min, 8 mg, colourless amorphous solid). Fraction F3 (1.6 g) eluted by 15 % EtOAc/n-hexane was chromatographed over silica gel CC using n-hexane:EtOAc gradient to yield four subfractions (A–D). Sub-fraction F3-A (348 mg) was subjected to ODS CC (95 % MeOH:H₂O) to give compound 8 (127 mg, white amorphous solid) and three sub-fractions (A1–A3). Subfraction F3-A3 (53 mg) was subjected to semi-preparative HPLC (Develosil C-30-UG-5) using MeOH:H₂O (95:5) to afford 9 (R_t 35 min, 16 mg, pale yellow amorphous solid) and 10 (R_t 31 min, 31 mg, pale yellow amorphous solid). Sub-fraction F3-B (479 mg) was chromatographed on silica gel CC using *n*-hexane:EtOAc gradient to yield four sub-fractions (B1-B4). Sub-fraction F3-B2 (300 mg) was subjected to ODS CC (95 % MeOH:H₂O) to yield three sub-fractions (B2a-B2c). Sub-fraction F3-B2a (43 mg) was subjected to semipreparative HPLC (Develosil C-30-UG-5) using MeOH:H₂O (95:5) to give 11 (R_{t} 48 min, 5 mg, white amorphous powder). Sub-fraction F3-B2c (250 mg) was subjected to semipreparative HPLC (Wakosil-II 5sil-100) using 100 % MeOH to obtain 12 (R_t 54 min, 232 mg, colourless amorphous solid) and 13 (R_t 60 min, 27 mg, colourless amorphous solid). ODS column chromatography (90 % MeOH:H₂O) of sub-fraction F3-D (300 mg) yielded four sub-fractions (D1–D4). Semi-preparative HPLC (Wakosil-II 5sil-100) using 95 % MeOH of sub-fractions F3–D3 (27 mg) and F3–D4 (53 mg) gave **14** (*R*t 48 min, 13 mg, colourless amorphous solid, F3–D3), 15 (Rt 52 min, 5.8 mg, colourless amorphous solid, F3 to D3), 16 (R_t 58 min, 19 mg, colourless amorphous solid, F3–D4), 17 (R_t 65 min, 5 mg, white amorphous powder, F3–D4) and 18 (R_t 73 min, 7 mg, colourless amorphous solid, F3–D4). Fraction F5 (0.78 g, 30 % EtOAc/n-hexane) was chromatographed over silica gel CC using *n*-hexane:EtOAc gradient to yield four sub-fractions (A–D). Sub-fraction F5-B (300 mg) was chromatographed on Sephadex LH-20 (CHCl₃:MeOH; 1:1) to yield three subfractions (B1-B3). Sub-fraction F5-B2 (23 mg) was subjected to semi-preparative HPLC (Wakosil-II 5sil-100) using 100 % MeOH to give 19 (R_t 83 min, 9.8 mg, colourless amorphous solid) and 20 (Rt 87 min, 7.8 mg, colourless amorphous solid). Sub-fraction F5-C (80 mg) was crystallized from MeOH to obtain impure 21, which was further purified on Sephadex LH-20 (CHCl₃:-MeOH; 1:1) to yield **21** (7 mg, white needles). Compound **22** (350 mg, white amorphous powder) was crystallized from fraction F9 (0.95 g, nhexane:EtOAc 95:5) by MeOH.

Ficupanduratin A (11): White amorphous powder (5 mg). m.p. 139-141 °C (dec.)

 $[\alpha]_D^{25}$ +51.3 (*c* 0.87 CHCl₃). IR (KBr) ν_{max} 3458, 2943, 1732, 1618, 1046 cm⁻¹. NMR data: see Tables 1 and 2. HRESIMS *m*/*z* 537.3916 (calcd for C₃₃H₅₄NaO₄, 537.3920 [M + Na]⁺) and 1051.7939 (calcd for C₆₆H₁₀₈NaO₈, 1051.7942 [2M + Na]⁺).

Ficupanduratin B (17): White amorphous powder (5 mg). 143–145 C (dec.) $[\alpha]_D^{25}$ +66.4 (*c* 0.15, CHCl₃). IR (KBr) ν_{max} 3478, 2955, 1725, 1454, 1032, 665 cm⁻¹. NMR data: see Tables 1 and 2. HRFABMS *m*/*z* 515.4106 (calcd for C₃₃H₅₅O₄, 515.4100 [M + H]⁺), 483.3873 [M + H-OCH₃]⁺.

11α,21α-dihydroxy-3β-acetoxy-urs-12-ene (21): White needles (7.0 mg). m.p. 261–263 °C.

 $[\alpha]_D^{25}$ +45.7 (*c* 0.87, CHCl₃); IR (KBr) ν_{max} 3425, 2950, 1725, 1630, 1450, 1320, 1015 cm⁻¹; NMR data: see Tables 1 and 2; HRFABMS *m*/*z* 501.3946 (calcd for C₃₂H₅₃O₄, [M + H]⁺, 501.3944).

Radioligand displacement for cannabinoid and opioid receptor subtypes

Compounds 1–22 were evaluated in competition binding with cannabinoid receptor subtypes, cannabinoid receptor 1 (CB1) and cannabinoid receptor 2 (CB2), as previously described (Thomas et al. 2005; León et al. 2013). Also, these compounds were tested against the opioid receptor subtypes (δ , κ and μ) as previously described (León et al. 2013).

Assays

Anti-microbial assay—All the isolated compounds were tested for their anti-microbial activity at concentrations of 20-0.8 µg/mL against *Candida albicans* ATCC 90028, *Candida glabrata* ATCC90030, *Candida krusei* ATCC 6258, *Asperigillus fumigates* ATCC 90906, methicillin-resistant *Staphylococcus aureus* ATCC 33591, *Cryptococcus neoformans* ATCC 90113, *Staphylococcus aureus* ATCC 2921, *Escherichia coli* ATCC 35218, *Klebsiella pneumonia* ATCC 13883, *Pseudomonus aeruginosa* ATCC 27853 and *Mycobacterium intracellulare* ATCC 23068 using modified versions of the CLSI/NCCLS methods (Ibrahim et al. 2012a, 2015a, b).

Anti-malarial assay—The isolated compounds were tested at concentrations of 4760– 528.9 ng/mL on chloroquine sensitive (D6, Sierra Leone) and resistant (W2, Indochina) strains of *Plasmodium falciparum*. Standard anti-malarial agents, chloroquine and artemisinin, were used as positive controls, whereas DMSO was used as a negative control (El-Shanawany et al. 2011; Ibrahim et al. 2016; Elkhayat et al. 2016).

Anti-leishmanial assay—The anti-leishmanial activity of the isolated metabolites was tested in vitro against *L. donovani* promastigotes as previously described (Ibrahim et al. 2012a, 2015a, b). Pentamidine and amphoterecin B were used as positive controls.

Cytotoxicity assay—The in vitro cytotoxic activity of the compounds was determined against noncancerous kidney cell lines (VERO) at concentrations of 4760-528.9 ng/mL. The cell line was obtained from the American Type Culture Collection (ATCC, Rockville, MD). The cell viability was determined using Neutral Red dye according to a previous method with modification (Borenfreund et al. 1990; Mohamed et al. 2013; Ibrahim et al. 2015a, b). Doxorubicin was used as a positive control, whereas DMSO was used as the negative control.

Molecular modelling and docking studies

Receptor modelling—The Homology Modelling module implemented in the program Molecular Operation Environment (MOE, version 2010.11, Chemical Computing Group Inc., Montreal, Canada) was employed to build the CB2 model (MOE 2012). For this purpose, the sequence of human CB2 was subjected to a protein database (PDB) search (Munro et al. 1993; Liu et al. 2007) of the protein structure database implemented in MOE 2010.11 with the Z cutoff set to 2.0 while leaving the other parameters at their default values. The search suggested the human adenosine receptor A2a (hAA2R) as a template for the comparative modelling procedure on the basis of high similarity, and the hAA2R structure [PDB:3EML] was chosen as the reference for the sequence alignment. After downloading the 3D structure of [PDB:3QAK] from the RCSB database (Liu et al. 2007), the structure was superimposed on the crystal structure of hAA2R. After removal of all lipid molecules, ions and water molecules, the residues between 209 and 222, which belong to lysozyme T4, were removed from the structure. In order to improve the modelling accuracy, the co-crystallized agonist UK-432097 of [PDB:3QAK] was retained in the system during the modelling procedure. Ten independently generated models were generated employing the default parameters of the homology modelling module, and only the consensus model was computed. An energy-minimization step was applied for the optimization of the new protein model; this aided the removal of residue clashes using the AMBER89 force field with default values. The quality of the model was verified by running the Protein Report module of MOE. After removal of the template ligand UK-432097, the primary structure receptor was energy minimized to convergence, giving the final hCB2 model structure.

Compounds docking studies—Small compounds were built in MOE with subsequent geometry optimization based on the MMFF force field incorporated in the program package, with a distance-dependent dielectric constant. The residues of the proposed binding pocket were defined based on the Site Finder module of MOE. The same binding site was proposed in the literature (Abood 2005). For the molecular docking studies, the module dock in MOE with parameters set to their default values was employed. All single bonds of the small ligands were allowed to rotate freely during the docking simulation. The side chains of the residues composing the binding pocket were treated as flexible, and the atoms of the protein backbone were maintained at fixed positions. The predicted poses were analysed according to their low energy states and were selected for presentation.

Results

The dried fruits of *Ficus pandurata* were extracted with MeOH. The concentrated MeOH extract was mixed with H₂O and partitioned between EtOAc and *n*-BuOH. The EtOAc extract was subjected to VLC, silica gel, Sephadex LH-20, ODS, and semi-preparative HPLC column chromatography to yield two new (**11** and **17**) and 20 known compounds (**1–10, 15, 16**, and **18–22**).

Compound **11** was obtained as white amorphous powder. It gave a positive Liebermann-Burchard reaction (Gołembiewska et al. 2013; Al Musayeib et al. 2014), indicating its triterpenoidal nature. Its molecular formula was determined as $C_{33}H_{54}O_4$ based on the

HRESIMS pseudo-molecular ion peaks at m/z 537.3916 (calcd for C₃₃H₅₄NaO₄, 537.3920 [M+Na]⁺) and 1051.7939 (calcd for C₆₆H₁₀₈NaO₈, 1051.7942 [2M+Na]⁺), requiring 7° of unsaturation. These degrees of unsaturation can be accounted for by five ring systems, one olefinic double bond, and one acetoxy carbonyl group. The HRESIMS showed characteristic fragment ion peaks at m/z 467.4360 [M+H–(OCH₃+OH)]⁺ and 424.3731 [M+H–(OCH₃+– OH+COCH₃)]⁺. Its IR spectrum showed characteristic absorption bands for hydroxyl (3458 cm⁻¹), acetoxy (1732 cm⁻¹) and double bonds (1618 cm⁻¹). The ¹³C and HSQC spectra showed the presence of 33 carbons, comprising 10 methyls, 7 methylenes, 9 methines (3 for oxymethines at $\delta_{\rm C}$ 76.8 (C-1), 76.2 (C-3) and 73.0 (C-11) and 1 for a *tri*-substituted olefinic double bond at $\delta_{\rm C}$ 122.0 (C-12)), and 7 quaternary carbons, including one for a carbonyl carbon at $\delta_{\rm C}$ 170.7 (C-31). Detailed 1D and 2D NMR analysis of **11** suggested that it is an ursane-type pentacyclic triterpene (Ibrahim et al. 2012b). The ¹H NMR spectrum of **11** exhibited six singlet methyls at δ_H 0.95 (H₃-23), 0.88 (H₃-24), 1.07 (H₃-25), 1.02 (H₃-26), 1.16 (H₃-27) and 0.82 (H₃-28) and two doublet methyl groups at $\delta_{\rm H}$ 0.84 (d, J = 6.5 Hz, H₃-29) and 0.97 (d, J = 6.5 Hz, H₃-30), supporting the ursane-type carbon framework of 11 (Mahato and Kundu 1994; Ibrahim et al. 2012b). They correlated to the carbons resonating at δ_{C} 27.9, 16.3, 18.1, 18.0, 22.6, 28.9, 17.4 and 21.2, respectively, in the HSQC spectrum. In the HMBC spectrum, H₃-23 and H₃-24 showed cross peaks to C-3, C-4 and C-5; H₃-25 to C-5, C-9 and C-10; H₃-26 to C-7, C-8, C-9 and C-14; H₃-27 to C-8, C-13, C-14 and C-15; H₃-28 to C-16, C-17 and C-18; H₃-29 to C-18, C-19 and C-20; and H₃-30 to C-19, C-20 and C-21, confirming their positions. Moreover, the ¹H and ¹³C NMR spectra of **11** revealed signals at δ_C 5.45 (d, J = 4.0 Hz, H-12) and δ_C 122.0 (C-12) and 145.4 (C-13), indicating the presence of a tri-substituted olefinic double bond. The double bond was established at C-12/ C-13 from the HMBC correlations of H₃-27 to C-13 and H-9 and H-18 to C-12 (Fig. 2). Three oxymethine groups were observed at $\delta_{\rm H}$ 3.53 (dd, J = 11.5, 5.0 Hz, H-1), 5.06 (dd, J =12.0, 4.5 Hz, H-3) and 3.74 (dd, J = 8.5, 4.0 Hz, H-11), correlating to the carbon signals at $\delta_{\rm C}$ 76.8 (C-1), 76.2 (C-3) and 73.0 (C-11) in the HSQC spectrum. This assignment was secured by the ${}^{1}\text{H}{-}^{1}\text{H}$ COSY correlations of 1-OH (δ_{H} 4.35) and H-2 with H-1 and H-3, as well as H-9 and H-12 with H-11; it was also confirmed by the HMBC cross peaks of 1-OH, H-3, H-9 and H₃-25 to C-1; H-5, H₃-23 and H₃-24 to C-3; and H-9 and H-12 to C-11 (Fig. 2). Additionally, the signals at $\delta_{\rm H}$ 2.03 (H₃-32)/ $\delta_{\rm C}$ 21.5 (C-32) and 170.7 (C-31) indicated the presence of an acetoxy group in **11**. This was further confirmed by the IR absorption band at 1732 cm⁻¹ and further secured by the ESIMS fragment ion peak at m/z 473 [M+H– COCH₃]⁺. Its connectivity at C-3 was established on the basis of the HMBC correlation of H-3 to C-31 and confirmed by the downfield shift of H-3 (δ_H 5.06). Furthermore, a methoxy group was observed at δ_H 3.37 (11-OCH₃)/ δ_C 55.8 (11-OCH₃). Its HMBC cross peak to C-11 established its placement at C-11. The relative stereochemistry at C-1, C-3 and C-11 was assigned based on the comparison of the ¹H and ¹³C chemical shifts as well as the coupling constant values with those of related triterpenes; it was confirmed by the NOESY experiment (Topçu and Ulubelen 1999; Mathias et al. 2000; Chiang and Kuo 2001; Mallavadhani et al. 2006; Mohamed 2014). The NOESY cross peaks of H-1 to H-3, H-5 and H-9 as well as H-9 to 11-OCH₃, H₃-27 and H₃-30 indicated that these protons were present on the same side of the molecule. Furthermore, the correlations of H-11 with H-18, H₃-25 and H₃-26 and H-20 with H-18 and H₃-28 suggested that these protons are on the other side of the molecule. On the basis of these findings and by comparison with the literature

(Mahato and Kundu 1994; Mathias et al. 2000), the structure of **11** was assigned as 1β -hydroxy- 3β -ace-toxy-11a-methoxy-urs-12-ene. To the best of our knowledge, **11** is a new triterpenoid, to which the name ficupanduratin A was given.

Compound **17** was obtained as a white amorphous powder. The HRFABMS showed a pseudo-molecular ion peak at m/z 515.4106 (calcd for C₃₃H₅₅O₄, 515.4100 [M+H]⁺), corresponding to a molecular formula of C₃₃H₅₄O₄. The IR, ¹H and ¹³C NMR spectral data of **17** were similar to those of **11**, with the exception that the signals for the oxymethine group at $\delta_{\rm H}$ 3.53 (dd, J = 11.5, 5.0 Hz, H-1)/ $\delta_{\rm C}$ 76.8 (C-1) and the methylene groups at $\delta_{\rm H}$ 2.10 (m, H-2A) and 1.95 (m, H-2B)/ $\delta_{\rm C}$ 32.2 (C-1) were missing. Instead, a new signal at $\delta_{\rm H}$ 3.45 (ddd, J = 10.0, 3.0, 3.0 Hz, H-21) correlated to the methine proton at $\delta_{\rm H}$ 1.34 (m, H-20) and the methylene proton at $\delta_{\rm H}$ 1.43 (m, H-22) in the ¹H–¹H COSY spectrum, indicating the presence of an oxymethine group at C-21. This was confirmed by the HMBC cross peaks of H-30 [$\delta_{\rm H}$ 0.88 (d, J = 6.5 Hz)] and H-19 to C-21 (Topçu et al. 2003). The β -configuration of H-21 was assigned based on the coupling constant value and was confirmed by the observed NOESY correlations of H-21 with H-18 and H-28 (Topçu et al. 2003). Based on the data obtained from the COSY, HSQC, HMBC and NOESY experiments, the structure of **17** was identified as 21*a*-hydroxy-3 β -acetoxy-11*a*-methoxy-urs-12-ene; this compound was named ficupanduratin B.

Compound **21** was obtained as white needles. The molecular formula was $C_{32}H_{52}O_4$, confirmed by the HRFABMS pseudo-molecular ion peak at $m/z 501.3946 [M + H]^+$ (calcd for $C_{32}H_{53}O_4$, 501.3944). Compound **21** was 15 mass units less than **17**, suggesting that **21** had one methyl group less than **17**. The ¹H and ¹³C NMR spectra (Tables 1, 2) of **21** were almost identical to those of **17**; however, the signals associated with the methoxy group at $\delta_H 3.36 (11-OCH_3)/\delta_C 53.5 (11-OCH_3)$ in **17** were absent. The NMR data of **21** were identical to those of 11a,21a-dihydroxy- 3β -acetoxy-urs-12-ene, which was previously obtained synthetically (Topçu et al. 2003). This is the first report of its isolation from a natural source. On the basis of the coupling constant values and NOESY correlations, the configuration of **21** was assigned to be the same as that of **17** and of 11a,21a-dihydroxy- 3β -acetoxy-urs-12-ene.

The known compounds were identified by analysing the spectroscopic data (1D and 2D NMR) and comparing their data with those in the literature to be *a*-amyrin acetate (1) (Virgilio et al. 2015), *a*-amyrin (2) (Virgilio et al. 2015), 3 β -acetoxy-20-taraxasten-22-one (3) (Ramadan et al. 2009; Ahmed 2010), 3 β -acetoxy-11*a*-methoxy-olean-12-ene (4) (Kuo and Chiang 2000; Mallavadhani et al. 2006; Barbosa et al. 2010), 3 β -acetoxy-11*a*-methoxy-olean-12-ene (4) (Kuo and Chiang 2000; Mallavadhani et al. 2000; Barbosa et al. 2010), 11-oxo-*a*-amyrin acetate (6) (Fingolo et al. 2013), 11-oxo- β -amyrin acetate (7) (Fingolo et al. 2013), palmitic acid (8) (Ajoku et al. 2015), stigmast-4,22-diene-3,6-dione (9) (Itokawa et al. 1973), stigmast-4-ene-3,6-dione (10), stigmasterol (12) (Mohamed and Ibrahim 2007; Kamboj and Salujai 2011; Chaturvedula and Prakash 2012), β -sitosterol (13) (Al-Musayeib et al. 2013), stigmast-22-ene-3,6-dione (14) (Wei et al. 2004), stigmastane-3,6-dione (15) (Wei et al. 2004; Lim et al. 2005), 3β ,21 β -dihydroxy-11*a*-methoxy-olean-12-ene (16) (Cáceres-Castillo et al. 2008), 3β -hydroxy-11*a*-methoxyurs-12-ene (18) (Fujita et al. 2000), 6-

hydroxystigmast-4,22-diene-3-one (**19**) (Kontiza et al. 2006; Georges et al. 2006), 6hydroxystigmast-4-ene-3-one (**20**) (Kontiza et al. 2006; Georges et al. 2006) and β sitosterol-3-*O*- β -D-glucopyranoside (**22**) (Khatun et al. 2012).

The isolated compounds **1–22** were evaluated for their anti-microbial activity against *C. albicans, C. glabrata, C. krusei, A. fumigates,* methicillin-resistant *S. aureus* (MRSA), *C. neoformans, S. aureus, E. coli, P. aeruginosa* and *M. intracellulare*; their anti-leishmanial activity against *L. donovani* promastigotes; and their anti-malarial activity against chloroquine-sensitive (D6, Sierra Leone) and resistant (W2, Indochina) strains of *Plasmodium falciparum.* Furthermore, they were tested for cytotoxicity against the VERO cell line. The results showed that none of the tested compounds had activity in these assays.

In addition, these compounds were evaluated using in vitro radioligand binding affinity assays of cannabinoid receptors (CB1 and CB2) and opioid receptors (subtypes δ , κ and μ) as described previously (León et al. 2013; Thomas et al. 2005). It is noteworthy that compounds 4, 11, and 15 (conc. 10 µM) selectively inhibited 69.7, 62.5 and 86.6 %, respectively, of the specific binding of $[^{3}H]$ -CP-55,940 to human embryonic kidney (HEK) cell membranes expressing human CB2 (Fig. 3; Table 3). Meanwhile, compound 6 exhibited low radioligand binding displacement at 10 µM, with an affinity of 22.2 % for the CB2 receptor. Moreover, compounds 4, 8, 13, 14, 15, 16, 19, and 22 showed moderate inhibition, with affinity values of 36.9, 39.1, 24.9, 44.5, 44.4, 22.4, 22.8 and 32.2 %, respectively, of the specific binding of [³H]-CP-55,940 to HEK cell membranes expressing human CB1. Interestingly, only compounds 6 and 22 inhibited 22.8 and 21.2 % of the specific binding $[^{3}H]$ -DPDPE to HEKhDOR cell membranes expressing δ -opioid receptors at 10 μ M, respectively. Compounds 8, 12, 13, 17, and 21 displaced the U-69,593 radioligand to HEKhKOR cell membranes expressing κ -opioid receptors by 22.1, 36.1, 28.1, 30.9 and 26.5 %, respectively. However, compounds 6 and 20 inhibited 23 and 28.5 % of the specific binding of [³H]-DAMGO to HEKhMOR cell membranes expressing μ -opioid receptors, respectively.

Molecular modelling studies were performed to obtain a more detailed structure-based analysis of the binding mode of the most active compounds towards CB2. For the CB receptors, no experimentally determined crystal structures are yet available. Hence, a homology modelling protocol was used to build a three-dimensional (3D) structure model of this receptor. Sequence similarity searches were performed to establish this protocol, and the results obtained identified the human beta1-adrenergic receptor (bAR1) [PDB:2Y00] (Warne et al. 2011) as the closest structure with the highest sequence similarity to the hCB2 sequence. However, the X-ray structure of the human A2A adenosine receptor ([PDB: 3QAK]) was selected as the template for comparative modelling (Xu et al. 2011). The newly constructed 3D model of human CB2 with the proposed binding pocket is shown in Fig. 4. Moreover, a mutagenesis protocol was applied in the binding pocket residues of reference ligand CP-55,940 (a CB1/CB2 cannabinoid receptor agonist) so that these residues are accessible in the new proposed model (Poso and Huffman 2008). This model was in agreement with published work (Rühl et al. 2012).

In order to investigate the productivity as well as the characteristics of the binding site of our models and to facilitate structure-activity relationship (SAR) analysis of the active cannabinoid receptor ligands 4, 11 and 15, docking analysis was performed on the reference CP-550,940 to adjust the structures of the residues (Fig. 5). Analysis of the receptor-ligand complex models generated after successful ligand docking was based on the analysis of hydrogen bonding interactions, aromatic and hydrophobic interactions, and binding energy (Table 4). The placement of ligands in the proposed pocket revealed different types of interactions, including three hydrogen bonding interactions with Lys109, Ser285 and Ala282 residues with distances ranging between 1.9 and 2.5 Å through the compound side chains. In addition, the alkyl side chain exhibited tolerably stable hydrophobic aromatic interactions with corresponding Val251, Met265 and Leu254 amino acids. However, 4, 11, and 15 showed mostly similar binding modes of interaction to the reference ligand, mainly with Ser285, Lys279 and Ala282, through the formation of stable hydrogen bonds. The most prominent feature in these compounds is their greater rigidity, which orients the structures to fit in the CB2 binding pocket. Also, molecular descriptors were calculated for all compounds, including descriptions of their polarity, lipophilicity and solubility, as structural indicators for topological and surface analyses (Table 5).

Discussion

Both natural and synthetic triterpenoids have a wide range of unique and potentially biological activities, including anti-microbial, antioxidant, anti-inflammatory, hepato-and cardioprotective, anticancer, anti-HIV, antiulcerogenic, antiplasmodial, and analgesic activities (Dzubak et al. 2006; Laszczyk 2009; Vrao and Patlolla 2012; Zhang et al. 2013; Han and Bakovic 2015). Vitor et al. (2009) and Okoye et al. (2014) reported that α -amyrin acetate (**1**) and α -amyrin (**2**) exhibited pronounced anti-inflammatory effects and suppressed the levels of inflammatory cytokines and COX-2 levels via inhibition of NF- κ B activity and of signalling pathways involving phospho-cyclic AMP response element binding protein (Vitor et al. 2009; Okoye et al. 2014). Moreover, **2** inhibited platelet aggregation induced by adenosine 5'-diphosphate, collagen and arachidonic acid (Aragão et al. 2007). Qi et al. 2008 mentioned that stigmast-4,22-diene-3,6-dione (**9**) exhibited anti-larval activity. Compounds **12** and **13** possessed anti-inflammatory activity through inhibition of cytokines, prostaglandin E2, lipoxgenase-5 and cyclooxygenase-2 (Prieto et al. 2006; Al-Attas et al. 2015). Also, **12** and **13** showed effective anti-genotoxic activity (Lim et al. 2005).

To the best of our knowledge, this is the first report of the evaluation of the affinity of this class of compounds towards human cannabinoid and opioid receptors. The cannabinoid and opioid receptors are G-protein coupled receptors and have long been known to modulate pain (Bushlin et al. 2010). Natural compounds with selective affinity for specific opioid and cannabinoid receptors could provide novel drug leads for neuropathic pain (Gao et al. 2013). Special interest has been focused in CB2 as a target for the treatment of neuropathic pain, inflammatory conditions and a variety of other pathologies (Grimsey et al. 2011). It is noteworthy that compounds **4**, **11**, and **15** showed good selective CB2 receptor binding affinity. The results confirmed that these compounds can be considered as lead structures for novel CB2 selective ligands. This is one of the possible mechanisms of the anti-inflammatory effect of triterpenoids. Hence, **4**, **11**, and **15** could represent potential anti-

inflammatory compounds. This supports the rationale behind the traditional use of this plant in the treatment of inflammation and its related symptoms.

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Fig. 1. Structure of isolated compounds 1-22

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Fig. 2. $^{1}\mathrm{H}\mathrm{-}^{1}\mathrm{H}$ COSY and HMBC correlations of 11, 17 and 21







Fig. 4.

Molecular surface analysis of CB2 model. The protein is shown with the *N*-terminus at the top and the residues forming the binding pocket are shown in space-filling mode among the transmembrane helices and isolated figure as a surface with colour-coded features: H bonding (*magenta*), lipophilic (*green*), mild polar (*blue*)



Fig. 5.

Hypothetical binding modes of the docked compounds **a 4**, **b 11**, **c 15**, and **d** CP-55,940 ligand. The figures exhibit the residues in the proposed binding pocket of CB2 model with corresponding hydrogen bonding interactions and hydrophobic contact

Table 1

1H NMR data of compounds 11, 17, and 21 (500 MHz, CDCl3)

Position	11	17	21
	δ H, m, (J in Hz)	δ H, m, (<i>J</i> in Hz)	δ H, m, (<i>J</i> in Hz)
1	3.53 dd (11.5, 5.0)	1.86 m	1.48 m
		1.31 m	1.28 m
2	2.10 m	2.05 m	2.03 brdd (11.6, 4.5)
	1.95 m	2.00 m	1.98 m
3	5.06 dd (12.0, 4.5)	4.56 dd (12.0, 4.2)	4.56 dd (11.6, 4.5)
4	-	-	-
5	1.42 m	0.72 dd (10.0, 2.0)	0.73 brd (11.3)
6	1.65 m	1.38 m	1.38 m
	1.41 m	1.01 m	1.00 m
7	1.28 m	1.53 m	1.48 m
		0.87 m	1.31 m
8	_	-	-
9	2.27 d (8.5)	1.81 d (8.9)	1.68 d (9.2)
10	_	-	-
11	3.74 dd (8.5, 4.0)	4.13 dd (8.9, 3.5)	4.33 dd (9.4, 4.5)
12	5.45 d (4.0)	5.35 d (3.5)	5.23 d (4.0)
13	_	_	_
14	_	-	-
15	1.61 m	1.44 m	1.85 m
	1.21 m	0.89 m	1.78 m
16	1.52 m	1.41 m	1.38 m
		0.93 m	0.90 m
17	_	_	_
18	1.41 m	1.42 d (10.0)	1.41 m
19	1.36 m	0.95 m	0.87 m
20	0.96 m	1.34 m	1.34 m
21	1.32 m	3.45 ddd (10.0, 3.0, 3.0)	3.58 brdd (11.2, 2.5)
22	1.46 m	1.43 m	1.44 m
	1.31 m	1.28 m	1.30 m
23	0.95 s	0.86 s	0.87 s
24	0.88 s	1.05 s	1.10 s
25	1.07 s	0.87 s	0.87 s
26	1.02 s	1.08 s	1.04 s
27	1.16 s	1.15 s	1.16 s
28	0.82 s	0.80 s	0.80 s
29	0.84 d (6.5)	0.87 d (6.5)	0.88 d (6.4)
30	0.97 d (6.5)	0.88 d (6.5)	0.93 d (6.5)
1-OH	4.35 brs	_	_

Position	11	17	21
	$\delta H, m, (J in Hz)$	δ H, m, (<i>J</i> in Hz)	δ H, m, (J in Hz)
21-OH	-	6.13 d (3.0)	_
11-OCH3	3.37 s	3.36 s	-
31	-	-	-
32	2.03 s	2.06 s	2.06 s

Table 2

13C NMR and HMBC data of compounds 11, 17, and 21 (500 and 125 MHz, CDCl3)

Position	11		17		21	
	8C, m	HMBC	8C, m	HMBC	8C, m	HMBC
1	76.8 CH	3, 10	31.9 CH ₂	3, 10, 25	31.0 CH ₂	2, 3, 25
2	32.2 CH ₂	3, 4, 10	27.8 CH ₂	3, 4, 10	21.4 CH ₂	4, 10
3	76.2 CH	1, 5, 23, 24, 31	77.3 CH	1, 5, 23, 24	77.6 CH	1, 2, 5, 23, 24
4	38.3 C	I	44.6 C	1	44.2 C	I
5	47.9 CH	3, 7, 10, 25	52.3 CH	3, 7, 23, 24, 25	52.3 CH	3, 10, 24, 25
9	18.3 CH ₂	4, 6, 8	17.5 CH ₂	7, 8	17.6 CH ₂	5, 7, 8
7	31.3 CH ₂	5, 9	33.3 CH ₂	5, 6, 26	33.0 CH ₂	5, 9, 26
8	42.0 C	1	43.8 C	1	43.7 C	I
6	45.9 CH	1, 7, 11, 12, 25, 26	51.8 CH	7, 11, 12, 25, 26	56.5 CH	7, 11, 12, 25, 26
10	43.0 C	Ι	38.0 C	I	37.8 C	I
11	73.0 CH	8, 9, 10, 12, 13, 11-OCH ₃	77.0 CH	8, 10, 12, 13, 11-OCH ₃	67.5 CH	8, 10, 12, 13, 11-OCH ₃
12	122.0 CH	9, 11, 13, 14, 18, 27	122.5 CH	9, 11, 14, 18	126.5 CH	9, 11, 14, 18
13	145.4 C	Ι	146.2 C	I	144.4 C	I
14	42.5 C	I	41.9 C	I	41.9 C	I
15	27.1 CH ₂	I	31.0 CH ₂	8, 17	32.1 CH ₂	8, 17
16	28.0 CH ₂	1	27.1 CH2	14, 17, 28	26.9 CH_2	14, 17, 28
17	33.6 C	1	33.6 C	1	33.7 C	1
18	58.5 CH	Ι	58.6 CH	12, 16, 19, 29	57.8 CH	12, 16, 19, 29
19	39.5 CH	I	39.2 CH	13, 17, 21, 29	39.3 CH	13, 17, 21, 29
20	39.7 CH		39.4 CH	18, 22, 30	39.2 C	18, 22, 30
21	32.1 CH ₂	1	76.7 CH	17, 19, 22, 30	76.7 CH	17, 19, 22, 30
22	41.5 CH ₂	1	41.3 CH ₂	17, 18, 20	41.2 CH ₂	17, 18, 20
23	27.9 CH ₃	3, 4, 5, 24	28.0 CH ₃	3, 4, 5, 24	28.6 CH ₃	3, 4, 5, 24
24	16.3 CH ₃	3, 4, 5, 23	13.1 CH ₃	3, 4, 5, 23	13.3 CH ₃	3, 4, 5, 23
25	18.1 CH ₃	1, 5, 9, 10	16.3 CH ₃	1, 5, 9, 10	16.2 CH ₃	1, 5, 9, 10
26	18.0 CH ₃	7, 8, 9, 14	18.2 CH ₃	8, 9, 14	17.9 CH ₃	8, 9, 14

Position	11		17		21	
	8C, <i>m</i>	HMBC	8C, <i>m</i>	HMBC	8C, m	HMBC
27	22.6 CH ₃	8, 13, 14, 15	22.4 CH ₃	8, 13, 14, 15	23.1 CH ₃	8, 13, 14, 15
28	28.9 CH ₃	16, 17, 18	28.6 CH ₃	16, 17, 18, 22	29.7 CH ₃	16, 17, 18, 22
29	17.4 CH ₃	18, 19, 20	17.7 CH ₃	18, 19, 20	17.7 CH ₃	18, 19, 20
30	21.2 CH3	19, 20, 21	21.4 CH ₃	19, 20, 21	21.3 CH ₃	19, 20, 21
1-OH	I	1, 2	I	I		I
21-OH			I	21, 22		21, 22
11-OCH3	55.8 CH ₃	11	53.5 CH ₃	11		11
31	170.7 C	I	170.8 C	Ι	170.8 C	I
32	21.5 CH ₃	31	21.3 CH ₃	31	21.4 CH ₃	31

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Table 3

Results of binding affinity assay of compounds 1–22 for human cannabinoid (subtypes CB1 and CB2) and opioid receptors (subtypes δ , κ , and μ)

Compounds	CB1 % Displacement	CB2 % Displacement	S % Displacement	κ % Displacement	μ % Displacement
1	I	I	I	5.3	I
2	I	I	I	I	I
3	15.9	I	I	6.9	I
4	36.9	69.7	I	11.7	9.1
S.	I	I	I	7.7	I
9	2.9	22.2	3.1	16.7	23
7	1.7	14.8	I	12.9	I
8	39.1	0.4	22.8	22.1	I
6	9.1	I	7.3	7.6	I
10	15.8	I	9.2	7.5	I
11	11.3	62.5	I	7	11.5
12	11.8	5.5	I	36.1	Ι
13	24.9	17.5	I	28.1	Ι
14	44.5	I	13.2	16.1	I
15	44.4	86.6	I	2.7	1.9
16	22.4	17.1	I	16.7	5.8
17	I	I	4.7	30.9	5.2
18	Ι	Ι	3.4	10.7	7.1
19	22.8	Ι	2.6	I	I
20	7	I	I	I	28.5
21	Ι	3.4	I	26.5	8.9
22	32.2	Ι	21.2	I	Ι
CP-55,940	91.49	103.34	I	I	Ι
Naloxone	I	I	99.11	101.43	101.92

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The interaction data of compounds 4, 11, and 15 compared to CP-55,940

Compounds	Energy	Binding I	oocket resid	nes (location	u)			
		Lys109 TM3	Lys279 TM3	Ser285 TM7	Ser268 TM6	Ala282 TM7	Asp189 Loop2nd-Ex	
		Hydrogeı	n bonding in	teraction (c	listance Å)			Hydrophobic interaction
CP-55,940	12	2.2	I	1.9	I	2.5	I	Val251, Met265, Leu254
4	12.5	I	3.1	3.4	I	I	I	Phe197, Val251, Phe87, Ala282
11	11.9	I	I	3.5	3.5	1.85	I	Val251, Phe197
15	13.5	2.7	3.1	I	I	I	3	Phe91, Phe283, Ala282

All interaction data are presented in the table showing the contact between the essential compounds and the residues in pocket compared to reference ligand (CP-55,940). These data includes binding energy, distance in case of hydrogen bonding interaction and amino acid residues responsible in hydrophobic contact with showing of transmembrane positions

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Molecular properties of the active compounds

Compounds	Molecular weight	Log S	TPSA	Log P(o/w)	VSA
4	499	-10.46	35.53	8.41	523.7
11	515	-9.74	55.76	7.34	533.3
15	429	-10.53	34.14	7.17	500.4
CP-55,940	377	-6.67	60.69	6.34	461.2

Different molecular descriptors were calculated including their molecular weight, solubility, polar surface area, partition coefficient, and Van Der Waals surface area