

Capturing LTA₄ hydrolase in action: Insights to the chemistry and dynamics of chemotactic LTB₄ synthesis

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Human leukotriene (LT) A₄ hydrolase/aminopeptidase (LTA₄H) is a bifunctional enzyme that converts the highly unstable epoxide intermediate LTA₄ into LTB₄, a potent leukocyte activating agent, while the aminopeptidase activity cleaves and inactivates the chemotactic tripeptide Pro-Gly-Pro. Here, we describe high-resolution crystal structures of LTA₄H complexed with LTA₄, providing the structural underpinnings of the enzyme's unique epoxide hydrolase (EH) activity, involving Zn²⁺, Y383, E271, D375, and two catalytic waters. The structures reveal that a single catalytic water is involved in both catalytic activities of LTA₄H, alternating between epoxide ring opening and peptide bond hydrolysis, assisted by E271 and E296, respectively. Moreover, we have found two conformations of LTA₄H, uncovering significant domain movements. The resulting structural alterations indicate that LTA₄ entrance into the active site is a dynamic process that includes rearrangement of three moving domains to provide fast and efficient alignment and processing of the substrate. Thus, the movement of one dynamic domain widens the active site entrance, while another domain acts like a lid, opening and closing access to the hydrophobic tunnel, which accommodates the aliphatic tale of LTA₄ during EH reaction. The enzyme-LTA₄ complex structures and dynamic domain movements provide critical insights for development of drugs targeting LTA₄H.

 LTA_4 hydrolase | LTB_4 | innate immunity | drug design | X-ray crystallography

eukotriene A₄ hydrolase/aminopeptidase (LTA₄H; EC 3.3.2.6) is a 69-kDa cytosolic bifunctional zinc metalloenzyme that converts the transient epoxide LTA4 (5S-5,6-oxido-7,9-trans-11,14cis-eicosatetraenoic acid) into the dihydroxy acid LTB₄ (5S,12Rdihydroxy-6,14-cis-8,10-trans-eicosatetraenoic acid) via a unique epoxide hydrolase (EH) reaction (1). LTB₄, a potent chemotactic and immune-activating agent, is implicated in acute and chronic inflammatory diseases (2, 3), cardiovascular disorders (4, 5), and carcinogenesis (6). In addition to its proinflammatory EH activity, LTA₄H may also counteract inflammation by its aminopeptidase activity, which inactivates by cleavage another neutrophil attractant, the tripeptide Pro-Gly-Pro (PGP) (7). In addition, the fatty acid epoxide LTA₄ is also the key precursor for biosynthesis of lipoxins, a family of anti-inflammatory, proresolving lipid mediators (8). Hence, LTA₄H is a checkpoint in lipid mediator biosynthesis that regulates both the initiation and resolution phases of inflammation.

 LTA_4H consists of N-terminal, catalytic, and C-terminal domains whose interface delimits a deep cleft harboring an L-shaped pocket, with a wide polar arm that binds PGP and a narrow hydrophobic arm that is believed to accommodate LTA_4 (9). The elbow of the pocket includes H295, H299, and E318, which coordinate a Zn^{2+} ion essential for both LTA_4H activities.

 LTA_4H is the only EH reported to form a nonvicinal distant 5,12 diol (10), requiring an atypical catalytic mechanism, the details of which have remained elusive in the absence of a crystal structure with bound substrate (11). Acquirement of an enzyme-substrate complex is complicated by the fact that LTA_4 is a highly labile allylic epoxide with a half-life of about 10 s at neutral pH, physical properties that until now have precluded determination of its binding at the active site of LTA_4H . In the

present report, we describe successful trapping and structure determination of LTA_4H - LTA_4 complexes providing insights to the EH mechanism. Moreover, we show that LTA_4H undergoes domain movements, with structural alterations indicating gated substrate entrance into the active site followed by induced fit to achieve optimal alignment and catalysis.

Results and Discussion

In this study, we describe six different structures of LTA₄H from five distinct crystal forms. These visualize two conformational states of the enzyme (Tables S1 and S2). The mutants we investigated include variants E271A (12), D375N (13), and R563A (11), which lack EH activity. An orthorhombic structure of cocrystallized D375N, as well as orthorhombic and trigonal structures of soaked E271A, were obtained in a closed conformational state with captured LTA₄ in the active site. In contrast, trigonal structures of both cocrystallized and apo D375N, as well as an orthorhombic structure of R563A, show an open conformation of LTA₄H without bound ligand.

LTA₄ Binding and Turnover in the Active Site of LTA₄H. Structures of D375N and E271A showed difference density for LTA₄ in the active site cavity (Table S1). The ligand, which consists of a hydrophobic tail, epoxytriene, and polar carboxylate group (Fig. S14), was modeled in two conformations in the structures of D375N and trigonal E271A, and in three conformations in the structure of orthorhombic E271A. The conformations differ mostly in the position of the C1 carboxylate that is either bound by the guanidine group of R563 or unbound facing toward the active site

Significance

Leukotriene B₄ (LTB₄) is a one of the most potent chemotactic agents known to date and participates in leukocyte recruitment during the innate immune response. Leukotriene A₄ hydrolase/ aminopeptidase (LTA₄H) catalyzes the committed step in LTB₄ biosynthesis. Here we report high-resolution crystal structures of LTA₄H in complex with its highly labile substrate LTA₄, which reveal the structural basis for the enzyme's unique epoxide hydrolase mechanism. Moreover, we show that LTA₄H undergoes domain movements, which gates the hydrophobic cavity for entrance of LTA₄ followed by induced fit. Our results provide new insights to the mechanism of LTA₄H and structure-based drug design.

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Data deposition: The atomic coordinates and structure factors have been deposited in the Protein Data Bank, www.wwpdb.org (PDB ID codes 5NI2, 5NI4, 5NI6, 5NIA, 5NID, and 5NIE).

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Fig. 1. Binding of LTA₄ in the active site of LTA₄H. (*A* and *B*) Different views of the active site of orthorhombic D375N (residues are in green) complexed with LTA₄ (orange) and acetate (ACT, violet). Yellow and red spheres represent Zn^{2+} and catalytic waters, respectively. mF_o-DF_c polder omit map contoured at 3.0 σ is shown in gray mesh around LTA₄ and ACT. (*C* and *D*) Different views of the active site of orthorhombic E271A-LTA₄ complex (residues are in cyan). (*E*) LTA₄ bound with its C1 carboxylate anchored to R563 in the active sites of D375N and (*F*) E271A. mF_o-DF_c omit map for Zn²⁺ and catalytic waters is shown in green mash at 3.0 σ . Distances are labeled in Å.

entrance (Fig. 1 *A*–*D*). The positions of the hydrophobic tail and epoxytriene motif (C5–C12) of LTA₄ are similar in all LTA₄-bound LTA₄H structures. The aliphatic tail of the substrate occupies the hydrophobic cavity with its ω -end abutting three structural waters at the bottom of the cavity. The epoxide oxygen is coordinated by Zn²⁺ and the hydroxyl group of Y383 (Fig. 1 *E* and *F*). Y383 locates opposite to one of the catalytic waters (W1), which is coordinated by Zn^{2+} and E271, and shields the unstable epoxide ring from nucleophilic attack by nearby water molecules. Residues surrounding the conjugated triene system of LTA₄ (Y378, Y267, M270, and Q136) also protect the ligand from nonenzymatic hydrolysis. The second catalytic water (W2) is found close to C12 of LTA₄ and



Fig. 2. A catalytic water has the same position in structures of LTA₄H complexed with different ligands. Superposition of D375N-LTA₄ (residues are in green; LTA₄, catalytic water and Zn^{2+} are in orange) onto (*A*) LTA₄H-OPB-Pro (PDB ID code: 4MS6) (residues are in gray; OPB-Pro, catalytic water and Zn^{2+} are in light cyan), onto (*B*) structure of LTA₄H-captopril (PDB ID code: 4DPR) (residues are in beige; captopril, catalytic water and Zn^{2+} are in magenta), and onto (*C*) E271A-LTA₄ (residues are in cyan; LTA₄, catalytic water and Zn^{2+} are in yellow).



Fig. 3. Dynamic domain motion in the structure of LTA_4H . (A) Closed conformation of LTA_4H with its FD in light blue and MD1 and MD2 in blue and teal, respectively. The same regions are colored pink, magenta, and hot pink for the open conformation. The hinge regions are in yellow. Zn^{2+} is in blue in the closed conformation and in magenta in the open conformation. (B) Y378 seals off the hydrophobic tunnel. Superposition of the active site and region of MD2 of closed D375N with bound LTA_4 (residues and Zn^{2+} are in blue, LTA_4 is in orange) onto open D375N (in pink) with blocked tunnel, and onto (C) open D375N (in magenta) with open tunnel.

coordinated by asparagine in D375N-LTA₄ (Fig. 1*E*), and by aspartic acid in E271A-LTA₄ (Fig. 1*F*).

Summarizing the structural information, we conclude that the EH reaction strictly depends on the correct alignment of LTA₄ in the active site cavity. As the ω -end enters the hydrophobic cavity, the C1 carboxylate group anchors to R563, while Zn²⁺ and Y383 coordinate the epoxide oxygen. At the same time, W1, polarized by Zn^{2+} and E271, donates a proton to promote S_N1 acid-induced opening of the epoxide ring. A carbonium ion is formed, whose positive charge is delocalized over the triene system with its C6-C7 bond in a procis configuration stabilized by the L-shape of the tight pocket (Fig. S1B). At the final step of the reaction, W2, perfectly poised to control stereospecificity, is activated by D375 and attacks C12 to generate a hydroxyl group in R configuration. In this way, LTA_4H converts LTA_4 into a dihydroxy acid containing a Δ^6 -*cis*- $\Delta^{8,10}$ -*trans* geometry of the conjugated triene system and a 12R hydroxyl group, two key structural features of LTB_4 that confer bioactivity (Fig. S1C). It is noteworthy that our conclusions reported here rationalize previous functional experiments with site-directed mutagenesis (11-13).

Comparison of the structures of LTA₄H-LTA₄ and a previously reported structure of LTA₄H complexed with a PGP analog (OPB-Pro) (14) shows that a single catalytic water, W1, is involved in both activities of LTA₄H (Fig. 24). This water is coordinated to both E271 and E296, creating hydrogen bonds to their carboxylate groups. Depending on activity, W1 is activated by E296 for peptide bond hydrolysis or by E271 for epoxide ring opening. This water molecule is also present at the same position in a structure of LTA₄H in complex with captopril (Fig. 2*B*). In complexed E271A-LTA₄, in the absence of E271, W1 remains hydrogen bonded only to the side chain of E296 (Fig. 2*C*).

Dynamic Gating of LTA₄ in the Active Site of LTA₄H. Structures of the two mutants D375N and R563A reveal for the first time an

"open" domain conformation of LTA4H distinct from all previously reported structures of the enzyme (Table S2). Comparison of "closed" and open conformations of our present structures (Tables S1 and S2) delineated three dynamic domains: the fixed domain (FD), the first moving (relative to FD) domain (MD1), and the second moving domain (MD2), with bending regions between them (Fig. 3A and Movie S1). The FD and MD1 can rotate relative to each other by $\sim 16-22^{\circ}$, governing the width of the entrance to the active site cavity. During dynamic domain motion, the residues V260, T465, and N466 act as a hinge, and MD2 moves by 3-4 Å and acts as a lid separating the exterior hydrophilic part of the active site from the interior hydrophobic tunnel. In the open conformation, Y378 from the lid seals off the entrance to the hydrophobic cavity, which accommodates the tail of LTA₄ during EH reaction (Figs. 3B and 4A and Movie S2). In this capacity, Y378 becomes a molecular gate for entry of LTA₄ into the hydrophobic tunnel and, simultaneously, prevents access of water molecules that could otherwise interfere with subsequent binding of the lipid substrate, as previously reported for the soluble bovine β -lactoglobulin (15). Of note, we could not find any structural obstacles for peptide substrate binding and turnover in the active site during lid displacement (Fig. S2).

The structure of open D375N cocrystallized with LTA₄ shows no density for the ligand, but the internal lid of MD2 has changed position, uncovering the hydrophobic cavity for substrate entrance (Figs. 3*C* and 4*B*). Moreover, superposition of open conformation of D375N and closed conformation of D375N and E271A suggests that in the open conformation of LTA₄H, Y383 may anchor the polar group of the substrate during LTA₄ entrance (Fig. S3 *A* and *B*). This hypothesis explains the presence of unbound carboxylate of LTA₄ observed in complex structures of D375N-LTA₄ and E271A-LTA₄ (Fig. 1 *A–D*). After domain closure, the carboxylate group of LTA₄ switches from the hydroxyl group of Y383 to the guanidinium



Fig. 4. Tunnel formations (*Upper*; the structures and corresponding Zn^{2+} are colored as in Fig. 3 *B* and *C*) and proposed mechanism of substrate binding (*Lower*) during dynamic domain motion between open and closed D375N. (*A*) In the open conformation (polar part of binding pocket is shown as tiny blue spot; wide space between domains at the entrance of the active site is indicated as an oval in shaded light blue) Y378 blocks the entrance of the hydrophobic cavity. (*B*) During substrate binding, lid region moves and opens the hydrophobic transle (red surface representation) for LTA₄ entrance. When LTA₄ slides into the hydrophobic pocket, its carboxylate group is guided by Y383. (C) During domain motion, the ω -end reaches the bottom of the cavity, the epoxide binds to Zn^{2+} , and the C1 carboxylate switches from Y383 to R563. The closed conformation affords productive substrate alignment and allows epoxide activation and subsequent hydrolysis according to an S_N 1 mechanism. The polar arm of the binding pocket that is formed as a result of convergence of domains is shown as blue surface.

group of R563. In a concerted manner, Y383 changes conformation and forms hydrogen bonds to the hydroxyl group of Y378 and the epoxide oxygen, which are molecular events of a substrate-mediated induced fit to achieve optimal alignment of LTA₄ in the active site cavity (Fig. 4*C*).

Changing of Coordination Mode of E318 in the Zn^{2+}-Binding Site. The open and closed conformations of LTA₄H display different coordination modes of the carboxylate group of the Zn^{2+} ligand E318. In closed structures, Zn^{2+} is coordinated by five bonds to surrounding ligands (Fig. S4*A*), whereas in open structures, we observe six Zn^{2+} -ligand interactions arranged in octahedral geometry, involving bidentate binding to the carboxylate of E318 (Fig. S4 *B* and *C*). Thus, during movement between open and closed conformations, we observe a carboxylate shift (16) from bidentate to monodentate coordination (Fig. S4*D*). This shift can occur against a relatively low-energy barrier and may reflect reduction of Zn^{2+} -positive charge during binding of the negatively charged epoxide oxygen of LTA₄ (17).

The Side Chain of E318 Helps the Y378 Gate Stay Open. Interestingly, analysis of the open structures also reveals two conformational states of the side chain of E318, one of which appears to prevent the Y378 gate from closing. Thus, in one state, the carboxylate group of E318 makes two bonds to Zn^{2+} , as described earlier. In the second state, the side chain of E318 is slightly rotated and has changed position, retaining only one bond to the metal ion, which instead binds an additional water molecule to complete the octahedral geometry (Fig. S4 B and C). In the open structure with blocked hydrophobic tunnel, Y378 is not within interacting distance to the E318 carboxylate, which leaves room for alternative position of the side chain of Y383 (Fig. S4B). In contrast, when the hydrophobic tunnel is open, the hydroxyl group of Y378 interacts with the E318 carboxylate in its second, rotated conformation, a hydrogen bond that can help stabilize the Y378 gate in open position and promote entrance of LTA₄ into the hydrophobic tunnel. Also, when the Y378 gate is open, Y383 has only space to allow one position distant from Y378 (Fig. S4C).



Fig. 5. Residue S379 changes position during lid movement. Superposition of open D375N (residues and Zn^{2+} are in pink) with blocked hydrophobic tunnel onto closed LTA₄H (residues and Zn^{2+} are in blue) in complex with LTA₄ (orange).

Implications for Rational Design of LTA₄H Inhibitors. As a result of the role of LTB₄ in inflammation, LTA₄H is an important drug target. The enzyme's substrate LTA₄ is also a precursor for the lipoxins, a family of proresolving lipid mediators, and thus inhibition of LTA4H may enhance lipoxin synthesis by shunting of LTA₄. Indeed, inhibitors of LTA₄H have been developed in the past but have not reached the clinic, perhaps because they have all blocked the enzyme's aminopeptidase activity and details of LTA₄ binding and enzyme dynamics have not been known. Notably, the results of the present study define the chemical landscape for drug design. Thus, small molecules should take advantage of the hydrophobic tunnel, reach beyond the Y378 lid, and anchor polar residues such as D375 or the carbonyl of W311, D312, F362, K364, or L365 without compromising the aminopeptidase active site. While exploiting these residues, it is important to exclude S379 from the list of possible contacts for drug molecule stabilization. S379 is a polar residue of the hydrophobic pocket but is juxtaposed to Y378 on the mobile lid and completely changes its position during lid displacement (Fig. 5). A successful LTA₄H inhibitor may have three modes of action: it will dampen the early inflammatory response by inhibition of LTB₄ synthesis and promote resolution of inflammation by preservation of PGP inactivation and shunting of LTA₄ into lipoxin generation.

Concluding Remarks. LTA₄H catalyzes hydrolysis of the unstable epoxide intermediate LTA₄ into LTB₄, which is a potent chemotactic factor for human neutrophils and plays a role in the mounting of an inflammatory response. In the present report, we describe the structural characterization of LTA₄H–LTA₄ complexes providing the structural basis for LTB₄ synthesis. We also show that LTA₄H undergoes domain movements, which gate the hydrophobic cavity for entrance of LTA₄. Our results give insights to general enzymology, biochemistry of innate immunity, and structure-based design of anti-inflammatory drugs.

Materials and Methods

DNA Constructs. LTA₄H mutant E271A was generated as described (12); mutation D375N was introduced with a QuikChange Site-Directed Mutagenesis Kit (Agilent Technologies). Oligonucleotides were from Plasmid TAG Copenhagen A/S, and constructs were verified by DNA sequencing (SeqLab). A codon-optimized gene for N-terminally 6His-tagged LTA₄H mutant R563A was synthesized by GenScript.

Protein Expression and Purification. All mutants were expressed in Escherichia coli JM101. Cells grew in LB medium with ampicillin (0.1 mg·mL⁻¹) at 37 °C to an optical density (λ = 600) of 0.6. Protein expression was induced by addition of 0.1 mM isopropyl β-D-1-thiogalactopyranoside. After overnight cultivation at 20 °C, the cells were harvested by 10 min centrifugation at $3,000 \times q$. The cells were disrupted by sonication, and the lysate was centrifuged at $60,000 \times g$ for 18 min. To perform immobilized ion affinity chromatography, the crude extract was applied onto a disposable chromatography column (BioRad) with affinity resin charged with Ni²⁺ and equilibrated with a standard buffer consisting of 25 mM Tris-HCl at pH 7.8, 200 mM NaCl, and 10% (vol/vol) glycerol. Elution of unbound and weakly bound fractions was performed in two steps with standard buffers containing 20 and 40 mM imidazole. C-terminally 6His-tagged proteins were eluted with standard buffer containing 250 mM imidazole. After immobilized ion affinity chromatography, the proteins were dialyzed against 25 mM Tris-HCl buffer at pH 7.8 and were purified by ion-exchange chromatography followed by size-exclusion chromatography. Ion-exchange chromatography was performed on MonoQ HR5/50 GL column (GE Healthcare) that was equilibrated with 25 mM Tris-HCl at pH 7.8, and LTA₄H mutants were eluted with a linear gradient of KCl (0-0.4 mM) in 25 mM Tris HCl at pH 7.8. Size-exclusion chromatography was performed using HiLoad 16/600 Superdex 200 pg column (GE Healthcare) equilibrated with gelfiltration buffer (25 mM Tris·HCl at pH 7.8 and 50 mM NaCl). All eluted fractions were analyzed with SDS/PAGE on a PhastSystem (GE Healthcare), using 10–15% gradient gel. Fractions containing proteins were concentrated in Amicon Ultra units (Millipore).

Saponification of LTA₄ Methyl Ester. To obtain LTA₄, LTA₄ methyl ester (Med Chem 101) in tetrahydrofuran was saponified with 1 M LiOH [6% (vol/vol)] for 48 h at 4 °C.

Crystallization and Data Collection. Mutants E271A and D375N were concentrated to 9 and 2.5 mg/mL, respectively, and crystallized by liquid/liquid diffusion in capillaries (Vitrex Medical A/S) at 4 °C, essentially as described for wild-type LTA₄H (9).

Rhomboid-shaped orthorhombic crystals of E271A appeared in 3 d in 22% (wt/vol) PEG 8000, 100 mM imidazole at pH 6.7, 5 mM YbCl₃. After 15 s soaking with 1 mM LTA₄, specimens were flash-cooled in liquid nitrogen and used for X-ray diffraction experiments under cryogenic conditions (100 K) at beamline ID23-1 (18) of the European Synchrotron Radiation Facility (Grenoble, France). To preserve LTA₄ from nonenzymatic hydrolysis, just before soaking, the ligand was exposed to a stream of nitrogen gas to evaporate tetrahydrofuran, and subsequently dissolved in 100 mM borate buffer pH 10 mixed with 20% (vol/vol) glycerol.

To cocrystallize mutant D375N with LTA₄, protein was preincubated with 0.4 mM ligand for 30 min on ice. Plate-shaped orthorhombic crystals of the D375N-LTA₄ complex could only be obtained in the presence of sodium acetate, which was added at 100 mM concentration to 22% (wt/vol) PEG 8000, 100 mM imidazole at pH 7.0, 5 mM YbCl₃. Crystals grew over the course of 3 mo and were cryoprotected with 25% (vol/vol) glycerol in crystallization solution before flash-cooling in liquid nitrogen. For X-ray experiments, crystals were placed into a nitrogen gas stream at 100 K, and data were collected at a wavelength of 13.5 KeV (0.91841 Å) at beamline BL14.1 (19) operated by the Helmholtz-Zentrum Berlin at the BESSY II electron storage ring.

The mutants E271A, D375N and R563A were used for crystallization experiments by vapor-diffusion method, as well. All experiments were performed at room temperature with a ratio of 1 μ L protein to an equal amount reservoir solution.

Optimized crystallization conditions of Morpheus screen (20) turned out to be successful for crystallization of mutants E271A (8 mg/mL) and D375N (7 mg/mL) by hanging drop technique. Rhomboid-shaped trigonal crystals of E271A grew in 5 d in the drops containing 8% (wt/vol) PEG 20 000, 15% (wt/vol) PEG MME 550, 100 mM Mes/imidazole at pH 6.9, and 20 mM of each alcohol [1,6-hexanediol, 1-butanol, (R,S)-1,2-propanediol, 2-propanol, 1,4butanediol, 1,3-propanediol], and were used for soaking experiments with 1 mM LTA₄ as described earlier. Trigonal crystals of mutant D375N and cocrystallized D375N were found in 5 d in drops consisting of 8% (wt/vol) PEG 20 000, 15% (wt/vol) PEG MME 550, 100 mM Hepes/Mops at pH 7.7–8.0, 15 mM CaCl₂, and 15 mM MgCl₂. Before flash-cooling in liquid nitrogen, crystals of D375N were cryo-protected with crystallization solution containing 15% (vol/vol) glycerol. Crystals of both mutants E271A and D375N were transferred into a nitrogen gas stream at 100 K and X-ray exposed at a wavelength of 12.7 KeV (0.97625 Å) at beamline ID-29 (21) (the European Synchrotron Radiation Facility).

Rod-shaped orthorhombic crystals of mutant R563A (4.1 mg/mL) appeared in 1 wk in the sitting drop plate in optimized conditions [20% (wt/vol) PEG MME 5000, 100 mM Mes monohydrate at pH 6.0 and 150 mM ammonium sulfate] of Crystal Screen HT (Hampton Research). Crystals were cryoprotected with 10% (vol/vol) glycerol in crystallization solution and flash-cooled in liquid nitrogen. X-ray data were collected under cryogenic conditions (100 K) at a wavelength of 14.2 KeV (0.8729 Å) at beamline ID23-2 (22) at the European Synchrotron Radiation Facility.

Data Processing and Structure Determination. All datasets were integrated and scaled using XDS (23).

To obtain unbiased electron density for the ligand-binding site of LTA₄H, the orthorhombic structure of the E271A mutant was determined de novo by single-wavelength anomalous diffraction at 12.7 KeV (0.97625 Å). AutoSHARP (24) located 2 Yb³⁺ ions and 1 Zn²⁺ ion, which were used to calculate experimental phases (phasing power, 1.014; overall figure of merit, 0.26) that were improved with SOLOMON (25), using an optimized solvent content of 49.4%. An initial model was built by ARP/wARP (26), which placed and docked in sequence 583 residues (R = 26.0%; $R_{free} = 29.8\%$). After rebuilding in Coot (27), the structure was refined with the phenix.refine module of the PHENIX package (28), using experimental phase restraints (maximum likelihood Hendrickson–Lattman target). Analysis of the resulting maps showed clear positive difference density in the active site cavity of the enzyme for the LTA₄ ligand, which was placed using model and restraint files generated by PRODRG (29). Protein geometry was validated using MolProbity (30).

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The structures of D375N and R563A, as well as the trigonal structure of E271A from a crystal grown by vapor-diffusion, were solved by molecular replacement with PHASER (31), using as search model the partially refined orthorhombic structure of E271A (before addition of the LTA₄ ligand model). Rebuilding and refinement were performed as described earlier, using an ML target.

To prevent the bulk solvent mask from penetrating the active site cavity of the enzyme, omit mF_o-DF_c difference map for LTA₄ in the active site of mutants E271A and D375N was generated using program phenix.polder (32). Omit map for catalytic waters and Zn²⁺ was calculated with the FFT program (33) from CCP4 program suite (34).

Domain motion in LTA₄H was analyzed with DynDom (35), which identified three dynamic domains, corresponding to residues 1–260, 269–372, and 380–465 (FD); residues 261–268 and 466–610 (MD1); and residues 373– 377 (MD2). The bending regions include residues 260–270 and 465–466 for MD1 and residues 370–373 and 378–381 for MD2.

Least squares fits of structures were performed using the McLachlan algorithm (36), as implemented in the program ProFit (www.bioinf.org.uk/ software/profit/). Animated visualization of the trajectory from open to closed conformation of LTA₄H was performed with Morph (37) PyMOL feature. The tunnels were calculated and visualized with CAVER (38) PyMOL plugin. Figures were made with PyMOL (Schrödinger, LLC) and ChemDraw Prime 15.1 (PerkinElmer).

Data collection and refinement statistics are presented in Table S3 (orthorhombic and trigonal structures of E271A, orthorhombic structure of D375N) and Table S4 (trigonal structures of D375N and orthorhombic structure of R563A).

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