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## **Plant protein-based hydrophobic fine and ultrafine carrier particles in drug delivery systems**

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## **Abstract**

For thousands of years, plants and their products have been used as the mainstay of medicinal therapy. In recent years, besides attempts to isolate the active ingredients of medicinal plants, other new applications of plant products, such as their use to prepare drug delivery vehicles have been discovered. Nanobiotechnology is a branch of pharmacology that can provide new approaches for drug delivery by the preparation of biocompatible carrier nanoparticles (NPs). In this article we review recent studies on four important plant proteins that have been used as carriers for targeted delivery of drugs and genes. Zein is a water-insoluble protein from maize; Gliadin is a 70% alcohol-soluble protein from wheat and corn; Legumin is a casein-like protein from leguminous seeds such as peas; Lectins are glycoproteins naturally occurring in many plants that recognize specific carbohydrate residues. NPs formed from these proteins show good biocompatibility, possess the ability to enhance solubility, and provide sustained release of drugs and reduce their

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toxicity and side effects. The effects of preparation methods on the size and loading capacity of these NPs are also described in this review.

#### **Keywords**

Plant proteins; Zein; Gliadin; Legumin; Lectin; Drug delivery; Nanocarrier; Sustained release

## **1-Introduction**

In recent years, biotechnology and biomedical science have played a major role in devising drug delivery systems and especially, progress in incorporation of nanotechnology for biomedical investigations has offered innovative materials and approaches for drug delivery applications, which provide many advantages over traditional routes of administering pharmaceutical products (1–8). Many macromolecules (proteins, polymers, carbohydrates, etc) have been used to prepare NPs and other forms of nanocarriers for drug delivery. The term "nanoparticles" describes solid, colloidal particles consisting of macromolecular substances that vary in size from 10 nm to 1000 nm, however, particles >200 nm are less often studied and nanomedicine usually refers to particles <200 nm (9–11).

The application of proteins derived from animal/insect sources such as gelatin(12), collagen(13), albumin (14), casein (15) and silk (16) have been widely investigated for drug delivery purposes as nanocarriers. However, some of these carriers may pose a risk of infectious pathogens due to contamination with animal tissues. On the other hand, there are significant numbers of individuals who avoid consuming animal-based proteins as food due to religious beliefs or personal preferences. Plant based-proteins derived from abundant natural plant resources have been employed as an alternative source of material to overcome the above-mentioned limitations. Also, plant proteins are often less expensive than animal proteins and possess functional groups which can be easily employed either to adsorb or to covalently couple molecules capable of modifying the targeting properties of NPs (17). Meanwhile, in recent years, some other biologically-derived elements, such as viruses and bacteria have been adapted to act as targeted drug delivery vehicles (18–20). These biological elements include viral vectors, virus-like particles, virosomes, recombinant bacteria, "microbots", bacterial ghosts (21), as well as various kinds of living cells (22) that have been employed to either to transport a therapeutic cargo, or to produce it in situ. However, despite promising potential for efficient in-vivo targeted delivery, there remain worrisome biosafety issues related to immunogenicity caused by human viruses, virus-like particles and virosomes etc. (23). Additionally, well-documented proof will be needed to convince health authorities and dispel concerns of the biomedical community about the safety of employing bacteria for medical treatments (24).

Several plant-derived polysaccharides, such as cellulose, hemicelluloses, pectin, inulin, alginates, carageenans, rosin, gums and musilages in various forms have been used to prepare implants, films, beads, microparticles, and NPs, and these formulations have also been investigated for drug delivery purposes (25).

Reddy et al.(26) published a review paper describing the potential of plant proteins for medical applications, and recently some review papers have been published concerned with general classes of protein-based nanocarriers, and these have also included some plantprotein derived nanocarriers (17, 27–29). In this review, the drug delivery applications of plant-derived proteins with particulate sizes that fall into the range of 10 to 100nm (ultrafine) and 100 to 1000nm (fine) as drug carriers along with recent studies and advances are summarized.

Zein is a water-insoluble plant protein derived from maize and owing to its ability to maintain humidity levels in prepared goods, has found many applications in the packaging and food industries. It contains many nonpolar amino acids such as alanine, phenylalanine, leucine and proline, but because of its protein structure and its hydrophobic properties, can be phagocytosed by macrophages and can therefore stimulate the immune system (30–32). Currently, commercial zein, is provided in two forms, yellow and white zein (33). According to its solubility and molecular weight, zein is divided into three types: "alpha" 21–25 kDa; "beta" 17–18 kDa; and "gamma" 27 kDa. These types comprise 75–87%, 15%, and 5–10% of total natural zein protein respectively. Alpha zein has an isoelectric pH point at 6.8 but the surface charge of zein can vary according to the environmental conditions. Due to the special structure of zein proteins, they can form globules with a diameter of 150 to 550 nm having somewhat different shapes in vivo (17). Zein therefore has the potential to be loaded with other cargo molecules and can avoid interaction with other hydrophilic polymer systems in the body (34). Gamma zein is able to deliver drugs through the cell membrane due to the interaction of its N-terminus with the cell membrane (35, 36). The N-terminus sequence of zein can be described as a member of the proline-rich class of cell penetrating peptides (CPPs), and is therefore used in intracellular drug delivery systems (37). In recent years, there have been intensive ongoing investigations aimed at developing zein nanoparticles for drug delivery applications along with in vivo pharmacokinetic preclinical studies on animal models for evaluating oral bioavailability and toxicology of zein-based drug delivery, which have produced considerably enhanced oral bioavailability of drugs encapsulated by these nanocarriers (38–42).

Another common plant protein is gluten which is the main storage protein in the seeds of wheat and corn, and is obtained from the industrial processing of starch (31). Based on its solubility, gluten is divided into two constituents; monomeric gliadin that is soluble in 70% alcohol with a molecular weight of 25–100 kDa, and polymeric glutenin that is rather insoluble due to disulfide bond cross-linking with a molecular weight of 106 kDa (43, 44). Glutenin nanoparticles have demonstrated excellent stability, which eliminates the need for usage of potentially cytotoxic crosslinking agents. Additionally, recent in-vivo preclinical and in-vitro studies have revealed the high potential of glutenin nanoparticles for therapeutic drug delivery (45).

The amino acid composition analysis shows the presence of large amounts of non-polar neutral amino acids in gliadin and only small amounts of charged polar amino acids are present in its structure (46, 47). The high proline content in gliadin causes it to interact with dermal and epidermal keratin protein and allows it to mediate controlled drug release (48).

Legumin is an albumin-like protein which plays a storage role in pea seeds (49). The main storage proteins in pea seeds are called globulins (50), and the two most abundant fractions are legumin and vicilin. These proteins are water-soluble depending on the pH and ionic strength. Legumin NPs have been utilized in pharmaceutical applications such as cutaneous or transdermal administration of drugs (51). These particles have been used as interesting pharmaceutical carriers for the delivery of antitumor drugs in local cancer treatment for an improved cytostatic effect (52).

Lectins are another group of naturally occurring proteins or glycoprotein's, which were first discovered in plants. However, they are also present in other living organisms. They are distinguished by their ability to bind non-covalently and specifically to carbohydrate residues attached to biomolecules such as proteins and lipids (53–55). The first study on lectin proteins was conducted by Hermann Stillmark in 1888 who reported that ricin protein extracted and partially purified from castor beans, possessed an agglutinating property (56). In the 1950s, lectins were named for preparations derived from plants, which could recognize and distinguish blood groups via the different sugar residues expressed (57). Later in 1972, Sharon et al. listed a large number of different lectin proteins which had up to then been purified (58). In recent decades, there has been a substantial body of knowledge accumulated on plant lectins, including their biochemical properties, biological functions and specific carbohydrate binding affinities (59–61). A variety of lectin-NP conjugates have been prepared in the last decade and proposed for drug delivery applications.

## **2- Zein**

#### **2-1 Protein structure and properties**

The tertiary structure of zein is ribbon-like, with dimensions of 16 nm long, 4.6 nm wide and 1.2 nm thick with distinct hydrophobic and hydrophilic segments (62). Based on small-angle X-ray scattering, the structure of zein consists of 10 successive helical sequences, arranged in an anti-parallel fashion and stabilized by hydrogen bonds between the turns. At each end there is a final glutamine-rich helix (63) (figure 1). The specific structure of zein makes it behave as an amphiphilic polymer, and when it interacts with molecules such as drugs, this interaction can influence its tertiary structure. Hydrophilic drugs open the coiled structure of zein and allow formation of smooth films, whereas lipophilic compounds enhance the curvature of the zein particles and form smaller spheres. Amphiphilic compounds allow the formation of interconnecting channels, thereby decreasing the curvature and forming a sponge (64). As compared to animal proteins such as silk and collagen, zein has more negative charge that makes it suitable for carrying the positive charged drug into the body (26). Although zein has an overall amphiphilic structure, it can have a better interaction with hydrophobic drugs in sustained release delivery systems, followed by hydrophilic and amphiphilic compounds respectively (65). Zein, having a high proportion of apolar amino acid residue (>50%), is an alcohol soluble protein. The variation in concentration of ethanol or isopropanol in the aqueous solution along with thermal treatment (at  $60^{\circ}$ C for 10 min) has been studied to evaluate the change in physical properties of the suspension of zein particles. It was found that the particle size was increased by raising the ethanol concentration in the ethanol/water solvent mixture initially from 70% (13.4nm) to 85% (29 nm) and the

In table 1 various protein NP preparation methods, generated particle size along with encapsulation efficiency of some therapeutics with zein NPs are listed. In the following section, the latest studies and applications of zein in drug delivery systems are covered.

#### **2-2 Application of zein NPs in drug delivery systems**

**2-2-1 Zein NPs generated by various phase separation methods—**Zein NPs are the most intensively investigated group of plant proteins for drug delivery applications either individually or in the form of hybrids with other materials to improve stability as well as other properties. Mesoporous silica SBA-15 and recombinant human bone morphogenetic protein-2 (rhBMP-2) was loaded onto a zein-based scaffold and hydroxypropyltrimethyl ammonium chloride derivatized chitosan (HACC) was incorporated into it as an antiinfective agent. The developed zein-HACC-S20 displayed good antibacterial activity against E. coli and S. aureus lasting for 5 days and was also able to release rhBMP-2 at a low controlled rate for up to 27 days (67). Luo et al. (68) proposed a nanohybrid formed by a complex between chitosan and zein for encapsulation of alpha-tocopherol (TOC, the active compound in vitamin E) to protect it from destruction in the stomach. The average particle size after drug loading varied from 200 to 800 nm and there was a burst release of the drug followed by a slower sustained release.

To control reactive oxygen species (ROS), one of the important pathogenic elements in rheumatoid arthritis (RA), the oral administration of antioxidant enzymes are desirable. Zein NPs were generated using a phase separation method and folate-conjugated enzymes, superoxide dismutase (SOD) and catalase, were loaded onto zein NPs to protect them from gastrointestinal degradation after oral administration and also to target them to activated macrophages via folate receptors. The particle size in this system was 255.21 nm with 30– 50 % loading efficiency. The use of zein increased the therapeutic activity of the proteins from 5–10% with free catalase and SOD, to 40–60% with the zein-loaded enzymes, and also protected the enzymes from pepsin as well as the acidic conditions of the stomach (69).

Coacervation or desolvation (sometimes referred to as anti-solvent precipitation, drawingout precipitation or solvent displacement) is based on the variable solubility of proteins in different solvents which is a function of pH, ionic strength and electrolytes (70, 71). Desolvation is a technology applied for preparing NPs from a wide range of polymeric macromolecules, and is based on desolvation by changes in charge and pH, or by addition of a desolvating agent (ethanol or concentrated inorganic salt solutions). The main advantage is that this process does not require an increase in temperature and, therefore is applicable to heat-sensitive drugs (48).

Zou et al. (41) generated zein nanoparticles (ZN) and emulsified these particles with TPGS1000 (TPGS) to form TZN for oral administration of daidzin. The ZN and TZN were prepared by a modified antisolvent method. The results indicated that the average size was around 200 nm, and the presence of TPGS (a food additive) enhanced drug encapsulation

from 53% in ZN to 63% in TZN, and moreover the cellular uptake and drug release was significantly higher in intestinal Caco-2 cells. These authors carried out in vivo investigations on bioavailability and pharmacokinetics performed by analyzing mice plasma for daidzin metabolites after oral delivery of daidzin loaded TZN. These studies confirmed improved oral bioavailability of diadzin by the TZN drug delivery system.

In a separate study, Luo et al. (72) used caseinate (CAS) to stabilize ZN to enhance the cellular uptake and transportation in Caco-2 cells. The best uptake occurred with a 2:1 ratio of CAS:zein at 1 mg/ml where 40% was taken up by the cells after 4h. The size of these particles was 120–140 nm and the uptake occurred through an energy-dependent endocytosis mechanism. Moreover, Li et al. (73) reported a two-phase sodium caseinate (SC)/ZN preparation by a simple antisolvent procedure via direct pouring of SC into a ZN solution. The average size of the NPs was around 300 nm and after freeze drying, they were able to be re-dispersed in deionized water. The carrier showed good antimicrobial activity against E. coli and Salmonella spp and could be a possible preservative against microbial contamination in the food industry.

The liquid-liquid dispersions method has been widely utilized both for generation of protein NPs and encapsulation of some compounds within zein NPs. The process generally involves, preparation of a zein solution in 60–90% aqueous ethanol, shearing the stock solution by pouring into bulk deionized water which causes emulsification of the solution into small droplets, separation of alcohol from the droplets and partition into the water solution due to its immiscibility with water, precipitation of the undissolved zein NPs and encapsulation of the co-dissolved nonpolar compound (74). Wu et al. (75) used a liquid-liquid dispersion method to encapsulate two antimicrobial essential oils, thymol and carvacrol, into ZN with an average size of <800nm. The antioxidant compounds extracted from cranberries known as procyanidins (CPs) were also able to be loaded into ZN by employing liquid-liquid dispersion owing to the formation of both hydrogen bonds and hydrophobic interactions between zein and the CPs. The average particle size was 392 to 447 nm depending on the CPs/zein ratio. This CPs zein reduced the cytotoxicity of procyanidins in pro-myelocytic leukemia L-60 cells in comparison with CPs-solution (76). The combination of ZN and betalactoglobulin is proposed to be a good carrier of lipophilic drugs. The tangeretinencapsulated lactoglobulin-coated ZN NPs were produced via liquid-liquid dispersion and used for delivery of water-insoluble tangeretin (a polymethoxyflavone). First, zein and tangeretin were dissolved in ethanol solution  $(90\%, v/v)$ , then this organic phase was introduced to aqueous solution of beta-lactoglobulin in saline to provide an amphiphilic surface coating of ZN. The yielded core-shell colloidal NPs were stable at low salt concentrations and pH values outside the isoelectric point, but aggregated at the isoelectric pH and at high salt concentrations (77). In a similar fashion, ZN coated with an alginate shell was proposed as nano-delivery system. Zein alginate core/shell NPs were generated via an antisolvent precipitation to form the zein core and, electrostatic deposition to form the alginate shell. The core diameter was 80 nm, surrounded by a 40 nm thick shell and having an overall diameter of 160 nm. The ZN showed good thermal stability at pH 7, without change in size after heating at 90° C for 2 h (78). In another study, ZNs were generated with food-grade nonionic surfactant, Tween 80 coating as a food stabilizer via antisolvent precipitation. The developed NPs had a 76 nm diameter zein core and a Tween 80 sheath

with 6 nm thickness and the obtained coreshell nanostructures could be employed as food grade delivery systems for encapsulating, protecting, and releasing bioactive molecules (79). Regier et al. (36) applied the coacervation technique to prepare ZN, then loaded hydrophilic plasmid DNA (pDNA) into the nanospheres with a loading efficiency that depended on the zein-pDNA ratio, with a maximum of  $65.3 \pm 1.9\%$  and a density of  $6.1 \pm 0.2$  mg DNA/g zein. The encapsulation protected the pDNA from DNAase I degradation and could provide sustained release for gene therapy.

Luo et al. (80). encapsulated two drugs (indol-3-carbinol (I3C) and 3,3′ diindolylmethane (DIM)), into zein/carboxymethyl chitosan (CMCS) by combining liquid-liquid phase separation and ionic gelation methods. These nanospheres could provide good controlled release of both I3C and DIM. In another study, Lai et al. (81) encapsulated 5-fluorouracil (5- FU) into ZN using a phase separation procedure for liver targeting. The results showed higher accumulation of drug in the liver after intra-vascular administration when the zein:5-FU ratio was 3:1 v/v. It could remain in the bloodstream for 24h. The average size of these particles was 11.4 nm, the encapsulation efficiency was 60.7%, and 4°C was the best storage temperature.

Another technology, which closely resembles anti-solvent precipitation, is modification of the (anti-)solvent system by employing supercritical fluids, such as  $CO<sub>2</sub>$  as the anti-solvent. This system has advantages of the absence of residual solvent, mild operating temperatures, and a narrow particle size distribution and has been used as a tool for encapsulation (71). Hu et al. (82) used the "solution-enhanced dispersion by supercritical fluids" (SEDS) technique (figure 2) to load the poorly water-soluble compound, lutein, into ZN. The study showed that the application of the SEDS technique could influence the morphology, drug loading, and size of the NPs. A ratio of lutein/zein of 1:18 w/w, and a pressure of 10 Mpa coupled with a solution flow rate of 10ml/min at 45°C could produce lutein-loaded ZN in a narrow sizerange with controlled drug release.

**2-2-2 Zein nanofibers generated by electrospinning—**The incorporation of various therapeutics, growth factors, vitamins etc into polymeric nanofibers prepared from natural materials has been investigated for numerous delivery applications via oral, pulmonary, transdermal and ocular administration routes (83). The electrospinning method is the most popular method employed for preparation of nanofibers from natural or synthetic materials for drug delivery purposes. The method is generally based on applying high voltage (1 to 30Kv) to polymer solutions or melts which generates highly electrified droplets with charge evenly distributed over the surface. When subjected to the higher potential, the droplet acquires a deformed shape named as "Taylor cone" and by further increasing the voltage beyond the threshold value, the electric field overcomes the surface tension and the jetting of charged solution and solvent evaporation occurs leading to the transition between liquid and solid and formation of a nonwoven fabric mat (84).

Aceclofenac and pantoprazole (members of two different classes of drugs) were simultaneously transported by composite nano-fibers made by electrospinning zein/eudragit mixtures. Aceclofenac could be loaded onto zein, whereas pantoprazole, an acid labile proton pump inhibitor, was carried by eudragit S 100 to inhibit gastric acid secretion. The

nanocarrier was prepared via single nozzle electrospinning process (figure 3A) then the drug was loaded in nanofibers. The sustained release of the drug was demonstrated for up to 8h both in vitro and in vivo. This study showed less gastrointestinal toxicity by controlled release of NSAIDs (85). Jian et al. used coaxial electrospining to provide a biphasic drug release (BDR) system (fig 3B). They loaded ketoprofen (KET), a poorly-water soluble NSAID, onto core-sheath nano-fibers. The sheath part consisted of hydrophilic polyvinylpyrrolidone (PVP) as a filament-forming matrix covering a core matrix of zein. The average size of the nano-fibers was  $730 \pm 190$  nm and they had good compatibility with KET. In this system 42.3% of loaded drug was released immediately and the remaining drug was released over 10h (86). In a similar study Yang et al. used acetic acid as the sheath liquid in a coaxial electrospining process to coat zein and ferulic acid (FA) as the core fluid. The results obtained from The proposed modified coaxial electrospinning process indicated that the generated nanofibers possessed higher quality and improved functional performance, such as sustained-release profiles (87). Another modified coaxial electrospinnig technique was used to prepare ibuprofen (IBU) loaded zein nanofibers using dimethylformamide as the sheath fluid. Zein exhibited good compatibility with IBU due to the formation of hydrogen bonds and the drug sustained release over 10 h in vitro was observed (88). Streptomycin sulfate was incorporated into electrospun nanofibers made from a mixture of polyurethane (PU) with added zein and cellulose acetate (CA). The presence of zein in the polymer mixture, improved the overall hydrophobicity and the ability to mediate cell attachment (89). Curcumin is a natural occurring compound with various therapeutic applications and desirable features of free radical scavenging and stimulating cell proliferation. The electrospinning method has been adapted to prepare a zein-curcumin fibrous scaffold. The i<sup>n</sup> vitro curcumin release from electrospun fibers was studied and the initial burst release of curcumin from the scaffold was observed, followed by sustained release without any toxicity towards mouse fibroblasts (90).

Other encapsulation techniques have also been investigated for efficient delivery of curcumin. To prevent the possibility of macrophage uptake and consequent immunogenicity which can be due to the protein origin and hydrophobicity of zein nano/microparticles, PEGylation of zein has been proposed. The hydrophobic anticancer drug, curcumin was encapsulated into a nanocarrier consisted of ZN conjugated with methoxy poly(ethylene glycol) (mPEG) to form micelle structures with an external hydrophilic shell. The results displayed higher drug uptake in drug resistant NCI/ADR-RES cancer cells in vitro (91). In another study, Gomez-Estaca et al. (92) encapsulated curcumin into ZN by an electrohydrodynamic atomization technique with an encapsulation efficiency of around 85– 90%. It was observed that the ZN size could be increased from 175 to 900 nm by increasing the zein concentration from 2.5 to 15% (w/w).

Zein nanocarriers are also promising for gene therapy applications. Karthikeyan et al. (93) developed siRNA loaded zein nanofibers for the first time and the potential of electrospun zein nanofibers for safe delivery of siRNA and sustained release was demonstrated. In vitro study revealed gene silencing efficiency of 26%, which was obtained by total 35% released siRNA.

These studies demonstrate the potential of zein protein as a promising nanocarrier for clinical applications which should be further investigated for sustained and biphasic drug release and gene delivery systems.

## **3- Gliadin**

#### **3-1 Protein structure and properties**

The gliadin structure consists of two main parts: a hydrophobic central domain (CD) containing highly repetitive amino acid sequences such as glutamine and proline, surrounded by hydrophobic terminal domains (TD). These two domains give amphiphilic properties to this molecule (94). According to the size of the TD, three type of gliadin are recognized;  $\alpha/\beta$ gliadin, γ gliadin, and ω gliadin in which  $\alpha/\beta$  and γ gliadins have a long carboxy-TD chain and a short amino-TD chain with high amphiphilic properties, whereas in ω gliadin, both carboxyl and amino TD chains are short and the CD comprises 90–95% of the protein structure (95). However, use of gliadin may cause immune response in celiac disease. In this disease, glutamine and proline containing proteins get ingested in the gut stimulate Th1 and Th17 cells, leading to malabsorption and villous atrophy (96). Antibodies against gliadin have been detected in patients who suffer from celiac disease, therefore it would be important to take this into account if considering administration of gliadin nano-carriers in those patients (97).

### **3-2 Applications of gliadin NPs in drug delivery systems**

The liquid anti-solvent precipitation (desolvation) is the most widely employed strategy for preparation of gliadin NPs and encapsulation, which is based on the lowering of the solvent power during dissolution of bioactive compounds. As mentioned before, a mixture of the bio-component and the excipient are dissolved in a mixture of a suitable organic solvent and water, then an anti-solvent is added to this mixture (71). The size of the generated gliadin NPs, (which is a determinant characteristic for medical purposes), depends on the organic medium and can be optimized by varying the solubility parameter δ of the protein solvent phase, so that, when  $\delta$  is closer to the protein solubility parameter, the size of the NPs is smaller (150nm when  $\delta_{Mixture}$  corresponds to the gliadin solubility parameter  $\delta_G$ ). Besides considering the solubility parameter for achieving the desired particle size, the environmental acceptability issue should also be taken into account before choosing a suitable organic solvent: generally an ethanol and water mixture is always preferred (98), (99).

The large surface area of the GNP provides the ability for good interaction with biological surfaces (100). Since gliadin is rich in neutral and hydrophobic amino acids, hydrogen bonding and hydrophobic interaction can link it to the mucosal surface and lipid-rich tissue. Carbazole-incorporated gliadin NPs have been prepared via a desolvation procedure and utilized for delivery of lipophilic drug carbazole for oral administration (101). Thereby, gliadin is a suitable material to produce various mucoadhesive NPs that can stick to mucosal surfaces, and be used for oral drug delivery of lipophilic molecules (102). Some unique properties of GNPs such as its affinity for the upper gastrointestinal tract, the small size of GNPs that increase penetration into the gastric mucosa and its ability to protect antibiotics,

make it suitable in treatment of *Helicobacter pylori* infection (103). Taking advantage of the bioadhesive ability of gliadin, it was applied for delivery of clarithromycin and omeprazole to the stomach (the site of H. pylori infection). The drug loaded NPs were prepared via desolvation and the estimated size of these GNPs were 400 to 650 nm with around 75% loading efficiency and with sustained release behavior at pH 1.2 and 37°C (104). Similarly, a long-lasting mucoadhesive preparation of amoxicillin-loaded GNPs was more effective against  $H.$  pylori infection than free amoxicillin, due to its extended gastrointestinal residence time (105). Moreover, Ramteke et al. (106) produced a triple antibiotic combination therapy consisting of amoxicillin, clarithromycin and omeprazole entrapped in GNP to increase the effectiveness of H. pylori therapy.

In another study, owing to the mucoadhesive properties of GNPs, the oral administration of gliadin-encapsulated superoxide dismutase was studied to treat feline immunodeficiency virus (FIV) infection (107).

In a separate study, Fajardo et al. (108) used cinnamaldehyde as cross-linker to bind gliadin to lysozyme, a naturally occurring antimicrobial agent that is used against Gram-positive microorganisms. In this research, the effect of the cinnamaldehyde on the release of the lysozyme from the gliadin was studied against Listeria innocua at pH 6.2. As expected, the results showed that increasing the cross-linker led to a decrease in lysozyme release and lower antimicrobial activity.

Kim et al. (109) utilized the amine groups of the gliadin protein as a polymerization initiator to produce an amphiphilic combination of gliadin-ethylene cyanoacrylate (ECA) and increase the water solubility of GNP. Gliadin plays a hydrophilic role whereas poly ECA provides the hydrophobic part on the surface. Duclairoir et al. (110) studied the interaction of GNPs with three different types of drug; vitamin E (VE) as a hydrophobic compound, linalool/linalyl acetate mixture (LLA) as a slightly polar compound mixture, and benzalkonium chloride (BZC) as an amphiphilic cationic drug. The results of this study showed the average size of these NPs was around 450–475 nm and the entrapment of VE and LLA was higher than BZC because of the non-polarity of the carrier, whereas the sustained release of the VE was more than LLA and BZC. LLA-GNPs and BZC-GNPs exhibited burst release of the drugs that confirmed the better interaction of gliadin with nonpolar compounds in comparison with either polar or amphiphilic bioactive molecules.

The encapsulation of lipophilic retinoic acid into GNPs was reported by Duclairoir et al. (111). Gliadin NPs were prepared by a desolvation method and they exhibited satisfying carrier characteristics for all-*trans*-retinoic acid (RA) with good entrapment efficiency: about 75% of added drug at 60 μg• (mg gliadin)<sup>-1</sup> and a payload of 76.4 μg• (mg gliadin)<sup>-1</sup> NPs. A rapid release of 20% of the drug after 15 min in sink conditions was observed followed by a slower diffusion process as a second step.

In a different strategy, Gulfam et al. used electrospray deposition to generate gliadin and gliadin-gelatin composite NPs for delivery of anticancer drugs, such as cyclophosphamide, against breast cancer cells. The average size of the GNPs varied from 220–450 nm according to the percentage of gliadin. The non-hybrid gliadin NPs gradually released

cyclophosphamide over 48 h while gliadin-gelatin hybrid NPs provided a more rapid drug release. Moreover, GNPs containing 7% loading of drug induced apoptosis in vitro within 24 h (112).

## **4- Legume proteins**

#### **4-1 Legumin and Vicilin**

**4-1-1 Protein structure and properties—**Legumin and vicilin are known as storage proteins in leguminous seeds, since they are laid down at an early stage of the life cycle of seed development for future use at a metabolically more active stage. Many different proteins exist in seeds and the storage proteins should be distinguished from those proteins which have metabolic or structural roles (113).

Generally, pea proteins comprise 20–30% by weight of pea seeds, mainly consisting of globulins (65–80%), glutelins and albumin. Globulins contain three different components including legumin, vicilin and convicilin. The 11S globulin components represented by legumin have molecular weights of 350 and 400 kDa whereas the 7S globulin component has a molecular weight of 150 kDa consisting of vicilin and convicilin (50). According to work by Bailey et al. (114) who created a model for legumin with ten polypeptide chains, a molecular weight of 320kDa is preferred. There are some differences and also similarities between legumin and vicilin. Their shape and the size of the subunit domains in an aqueous solution are similar, but the molecular weight of legumin (360 kDa) is higher than that of vicilin (200kDa) (115). Moreover legumin is resistant to coagulation at high temperatures (116). In both proteins, the carboxy-terminus of the subunits exhibits a conserved core region that is predominantly hydrophobic with most of the residues in a beta-sheet conformation. The amino-terminus regions of the subunits from the two different proteins cannot be aligned, although they typically contain a high proportion of residues in common (117, 118), predicted to be in a helical conformation i.e. leucine, glycine, and threonine (114). There is an inserted polypeptide span in the legumin proteins with a high proportion of acidic amino acids (between the central and carboxy-terminus) (117). These residues are predicted to be in a helical conformation (118).

**4-1-2 Application of legumin and vicilin in drug delivery—**Legumin and vicilin NPs can be prepared by a simple controlled desolvation (pH-coacervation) method (119, 120). These coacervates are not stable enough and must be made rigid with physical or chemical cross linking to prevent the formation of aggregates. Glutaraldehyde is used as the most common cross-linking agent (121). A limitation in particle growth, and hence a decrease in the NP size results by increasing the anti-solvent/solvent ratio (122). However agglomeration can result from the use of high salt concentration due to its neutralizing effect on surface charge (123). Because legumin is hydrophobic, when solvent ionic strength is increased, the protein solubility would be increased, resulting in smaller NPs (124). The cross-linking agent, glutaraldehyde is known as a toxic material, so it is necessary to remove it as completely as possible (125). Legumin proteins need the presence of a surfactant such as Poloxamer in their fabrication process to stabilize them against aggregation. It was found that an increase in the product yield could result from an increase in the surfactant

concentration even with no detectable change in the particle size (29). The legumin NP preparation process is shown below in Figure 4 with the first chromatographic procedure done in order to purify legumin, and followed by the production of legumin coacervates by stirring with sodium phosphate citrate buffer. The coacervation method is the most frequently applied method for the synthesis of legumin NPs. By varying the surfactant content, pH, and ionic strength, the percentage yield of legumin coacervates can be optimized. The optimum pH to obtain coacervates in submicron size was near to neutrality and the results showed an opalescent appearance.

Irache et al. (124) synthesized legumin NPs with a size around 250 nm by crosslinking with glutaraldehyde using a pH-coacervation method (near neutral pH). The NPs at neutral pH were more stable, whereas in an acidic environment the NPs suffered rapid degradation.

Surface adsorption or drug entrapment inside the particles are the most common methods of drug loading with legumin (126).

Vicilin NPs have been prepared and used as drug carriers in a complex system in which these particles were conjugated with *UIex europaeus* lectin molecules due to their sugarbinding property via a two-stage carbodiimide method. The NPs had an average diameter of approximately 660 nm and the proposed delivery system was used in-vitro studies which revealed high potential of the prepared conjugate toward bovine submaxillary gland mucin (BSM) glycoprotein as a biological model (127).

## **4-2 Soy protein Isolate**

It is known that soybeans which are classified as legumes contain proteins consisting of albumins and globulins, and as stated previously, legumes also include legumin and vicilin (113). The protein known as soybean isolate (SPI) consists of four prominent protein sub types, 2S, 7S, 11S and 15S, among which glycinin (11S protein) and β-conglycinin (7S protein) comprise more than 80% (128). Soybeans are an abundant natural protein source, which offer satisfactory water solubility, and SPI has been employed to develop different kinds of nano-sized drug delivery vehicles, owing to its good encapsulation ability. Since SPI is degraded in the fluid of the gastrointestinal tract, this limits its routes of delivery to injection-based delivery (parenteral) methods. Many recent studies have therefore been SPI for oral delivery, by fabricating hybrid and complex nanocarriers in conducted to adapt combination with other natural products (129–131). In a study conducted by Teng et al. (132), SPI nanoparticles were developed based on a desolvation approach with ethanol as desolvating agent, and curcumin was used as a model drug. The preparation process included different steps of incorporation, cross-linking, dispersion, evaporation and desolvation. The average size of the nanoparticles was found to range from 220.1 to 286.7 nm. The efficiency of encapsulation and loading of the drug were 97.2% and 2.7%, respectively. Basak et al. (133) prepared hybrid composite particles from S PI and sodium alginate with addition of the clay mineral, cloisite 30B in an aqueous solution. Different amounts of vincristine sulfate were added to the SPI-cloisite 30 B solution followed by stirring for 1h, and then the polymer-drug complexes were dried at room temperature. Drug release kinetics were analyzed by plotting the cumulative release data versus time, that was fitted to an exponential equation which indicated a non-Fickian type of kinetics. The release

of the drug was studied under va rying pH conditions and it was found that the drug release depended on the pH and the composition of the matrix.

## **5- Lectin**

## **5-1 Protein structure and properties**

The lectins are another class of glycoproteins isolated from legumes consisting of two or four identical, or almost identical, subunits (or protomers) of 25–30 kDa, and each contains a single carbohydrate binding site with the same sugar specificity. The subunits of the legume lectins all consist of single polypeptide chains of about 250 amino acids which may carry one or two N-linked oligosaccharides. In some lectins (e.g., those from pea and lentil) the polypeptides are divided into a light  $(a)$  and heavy  $(\beta)$  chain. Legume lectins exhibit significant sequence homologies, along with about 20% of invariant amino acids, and close to 20% of similar ones (134).

Plant lectins are divided into several groups according to the molecular structure including merolectins (possessing one carbohydrate-recognition/binding domain e.g. havein), hololectins (having two or more carbohydrate- recognition/binding domains with homological structure which bind to structurally similar sugars e.g. concanavalin A, ConA), chimerolectins (fusion proteins consisting of additional non-related domains and/or tandemly arrayed-binding domains e.g. ricin) and superlectins (lectins with at least two different carbohydrate binding domains e.g. TXCL-1)(54, 135).

Plant lectins based on their carbohydrate binding specificities are classified into polyspecific or monospecific. Lectins reversibly bind through their non-catalytic domain to specific sugar residues such as monosaccharides or oligosaccharides including galactose, glucose and glucose derivatives, mannose and mannose-containing glycans (55, 136, 137). These carbohydrate residues are involved in triggering various cell recognition and adhesion processes. Many vesicular transport systems such as endocytosis in cells are accomplished through recognition of and binding to carbohydrate residues on the cell surface (138–142).

Various studies have investigated the stability of lectins exposed to different conditions. For example, it is known that *Solanum tuberosum* lectin (STL), a glycoprotein composed of two identical subunits, remained stable throughout a pH range of 4–10 and its thermal stability did not change below 50°C. In addition, the endocytic capacity of lectins is also a critical characteristic. In this regard, STL shows a high endocytic capacity in rabbit conjunctival epithelial cells (143, 144).

#### **5-2 The function of lectin-conjugates in drug delivery systems**

For many years, lectins have captured the intense interest of many researchers with diverse applications in the medical, biological and agricultural fields (145, 146). Some of the most studied areas are anti-insect activity (147, 148), antimicrobial activity (149–151), antiviral activity (152–155), and use as biomarkers for disease diagnosis (156–158). They have been widely studied in anti-cancer research (154, 156, 159 161), and anti-cancer activities have been found for several plant lectins including wheat germ agglutinin (WGA) (162) and ricin

(163), and autophagic effects caused by  $P$ . cyrtonema lectin (PCL) and ConA in tumor cells (164).

On the other hand, lectins have long shown great potential for drug targeting applications (142, 165). Therefore, lectin-mediated drug delivery systems have become a major field of study. Lectin-mediated drug delivery systems were first proposed in the 1980s (166). In recent years, drug delivery applications of lectins have also attracted a large number of studies (143, 167–170). These drug delivery applications of lectins include epithelial delivery (171–174), pulmonary delivery (173), ocular delivery (155), mucosal delivery (141, 148, 169, 175, 176), oral antigen delivery (154, 177), and protein and peptide drug delivery systems (173, 177).

The targeting ability of lectins can be explained by a mechanism in which lectins bind to carbohydrates on the exterior surface of specific cells followed by a cellular uptake process, and when internalized lectins show a particularly strong binding to nuclear pore membranes (54, 171). The majority of cell surface proteins and lipids in cell membranes are glycosylated by forming glycan arrays, which offer suitable binding sites for lectins. Different combinations of various sugars allow the existence of the immense range of glycan arrays in nature and the subsequent enormous variety in the chemical signatures of different cells. Therefore, the lectins attached to the drug and the glycosylated surface of cells can interact with each other (141, 174). In this manner, a specific lectin protein can target different cells and tissues. Therefore, the great potential of lectins has been exploited in targeted drug delivery systems (142, 171). Two major mechanisms are employed for preparing lectin mediated drug delivery systems: 1) direct lectin based targeting in which the carbohydrate groups bound to the delivery system are recognized by endogenous lectins on the cell surface. These endogenous lectins have been found in many mammalian cells and, 2) reverse lectin based targeting wherein the exogenous lectins attached to the delivery system recognize the carbohydrate groups located on glycoproteins and glycolipids on the cell surface (178). However, some drawbacks have been observed for lectin-based drug delivery systems. Owing to the plant origin of lectins, they have a considerable potential for immune-stimulatory effects (179). This recognition by the immune system could have adverse effects on the immune function and could cause cytotoxicity which can cause harmful side-effects (180, 181).

In table 2, some recent researches based on lectin-polymeric NP conjugates as drug delivery systems are summarized.

Mucosal targeted drug delivery systems have been the goal of several of these studies. In recent decades, lectins have been exploited to enhance targeting to lymphoid tissue (187). Lectins have been proved to increase the adhesion of NPs to the intestinal epithelium resulting in enhanced drug penetration (54, 171). According to these studies, lectins including WGA, odorranalectin, and UEA have been conjugated with NPs resulting in improved bioadhesive delivery systems (139, 141, 174–176, 188).

It has been shown that targeting ligands can enhance particle uptake via binding to specific receptors on the cell surfaces. A common approach is to coat the NPs by the lectin-ligands

(143, 167, 168, 171, 174). Lectin proteins can bind with high specificity to carbohydrate moieties of the glycocalyx of intestinal enterocytes and also to the mucus layer. In this case, lectin conjugated NPs can be exploited as a mucoadhesive system to target mucosal surfaces (149, 189).

The thin layer of mucus adhering to epithelial cells contains highly glycosylated proteins named mucins which provide a great potential for lectin-mediated drug delivery systems (167). Furthermore, it is proposed that plant lectins are ideal reagents for direct delivery of drugs to the gut epithelium since they are stable at low pH conditions and they have specificity for binding to epithelial cells (155, 167). Figure 5 shows the process consisting of three steps of binding, internalization and intracellular transport of lectins or lectin conjugates into a specific cell in an epithelial barrier.

Uptake of the Lectin-polymeric NP conjugates can be improved via the increased interactions with mucus (143, 170, 190, 191). Ying Liu et al. (141) suggested a mechanism for intestinal transport of wheat germ agglutinin (WGA)-grafted lipid NPs for oral delivery of bufalin to target intestinal mucosal tissue. Fig. 6 shows the transport mechanism of the drugs from WGA-grafted lipid NPs to intestinal mucosal cells. According to this mechanism, glycosylated structures are recognized by WGA and causing bioadhesion. This is followed by the triggering of endocytosis via cell signaling cascades.

Recently, several efforts based on novel drug delivery systems using lectins have been conducted for treatment of inflammatory bowel disease (IBD). Moulari et al. (183) exploited a lectin-mediated drug delivery system loaded with glucocorticoids for active targeting and selective bioadhesion to the inflamed tissue in an experimental murine colitis model. Peanut agglutinin (PNA) and WGA lectins were attached to the surface of the NPs via covalent binding. The results of quantitative adhesion analysis confirmed a higher binding and selective bioadhesion for lectin mediated NPs compared to unconjugated NPs which led to an enhanced anti-inflammatory therapeutic efficacy for the glucocorticoid-loaded lectinconjugated delivery systems. WGA lectin conjugates strongly bound to healthy intestinal mucosa. However, PNA lectin conjugated NPs revealed much-improved bioadhesion and subsequent therapeutic effects compared to those of WGA. Moreover, the resistance of lectin moieties to degradation by digestive processes was evaluated in simulated gastro-intestinal fluids suggesting a high stability of lectin-conjugates.

Many studies for treatment of central nervous system (CNS) disorders and diseases have been investigated using lectins (182, 192). Some studies have investigated whether therapeutic drugs can be delivered to the CNS via intranasal administration (193–195). Chen et al. (169) studied the in vitro and in vivo brain delivery after nasal adsorption of NPs using Solanum tuberosum lectin (STL) conjugated biodegradable MePEG-PLGA NPs. STL was exploited as a surface biorecognition ligand through covalent coupling. STL-conjugated NPs showed good mucoadhesion, followed by enhanced internalization of NPs in Calu 3 cells. Brain delivery of the STL-mediated conjugates via nasal administration was improved. The safety and biocompatibility of STL-mediated delivery was also confirmed by low cytotoxicity and only minor irritation of the cilia. Among CNS disorders, schizophrenia has been also been studied (193, 195). In a study conducted by Piazza et al. (182) intranasal

administration of haloperidol-loaded therapeutic NPs to increase transport to the brain was investigated. STL lectin-functionalized, polyethylene glycol-block-poly (D,L)-lactic-coglycolic acid (PEG-PLGA) NPs were loaded with haloperidol. The in vitro release of haloperidol demonstrated showed minimal drug leakage, and they found efficient drug transport through the nasal epithelium to target brain tissue. Furthermore, this study revealed that a lower amount of drug dose was needed in order to cause a therapeutic effect for treatment of schizophrenia.

Bladder cancer is considered a challenging neoplastic malignancy in oncology around the world (196). Many efforts have been conducted to improve intravesical delivery using anti cancer drugs. Neutsch et al. (197) studied the potential of lectin conjugates as delivery systems to urothelial cells. Multivalent protein bioconjugates were formed via covalent coupling of WGA as a targeting ligand, to fluorescent-labeled bovine serum albumin (fBSA). Their study showed that fBSA payload protein conjugated to WGA resulted in improved binding to cancer cells. This approach could be used for preferential internalization of macromolecules to bladder cancer as well as triggering efficient endogenous uptake via lectin-mediated endocytosis. Terminal accumulation in degradative lysosomal compartments was seen using WGA-induced internalization. Therefore, lectin bioconjugates are promising for delivery of small molecule drugs and also complex biomolecules in intravesical therapy for bladder cancer.

## **6- Conclusion**

The use of plant proteins as drug carriers can provide controlled release possibilities and can also enhance the distribution and compatibility of the drugs in the biological environment. Plant proteins such as zein, gliadin, legumin and lectins possess favorable properties such as low toxicity, ready availability, easily modifiable structures with free functional groups, and good stability. These plant proteins are basically part of natural foodstuffs and their biological compatibility has been established by millennia of evolution. Different methods of preparation to produce various types of NPs can be utilized. The binding of cargo to the carrier can be achieved either by covalent conjugation using a cleavable linker or by noncovalent encapsulation. Zein has probably been used the most to prepare drug-carrying protein NPs due to its low cost, easy availability, "generally regarded as safe" status, together with its high drug-loading capacity. There have been a few recent attempts toward developing novel stimuli-responsive (smart) nanocarriers based on zein NPs that incorporate fluorescent markers for cell imaging, and also dual-responsive drug delivery systems for targeted and controlled delivery of therapeutics (190, 191). These can be regarded as only the first steps toward developing smart plant-derived drug delivery vehicles. Apart from these reports, several recent preclinical in-vivo studies, have been performed to evaluate the possibility of employing plant protein-based nanocarriers in living animals. These studies have important implications for the future possible widespread use of these natural proteins for medical applications.

The studies using gliadin and legumin as drug carriers are less common and only a relatively few reports have been published, but since these efforts show acceptable results, these two proteins could be good candidates for further researches. Lectins show great potential for

drug delivery via interaction with carbohydrates on mucosal surfaces. However, rather than individual lectin NPs, there have been a large number of lectin-NP conjugates reported so far as drug carriers. Meanwhile, their ability to be recognized by the immune system means that this possibility should be taken into account when designing in vivo studies. Further progress in the application of plant proteins as drug delivery systems requires deeper investigation into the nature of protein-drug interactions, studying the binding kinetics of protein nanocarriers to the cell receptors for targeted therapy and optimizing several related parameters.

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**Figure 1.**  Structure of a single unit of zein



#### **Figure 2.**

Schematic diagram for the apparatus and process of the solution enhanced dispersion by supercritical fluids (SEDS) technique; the apparatus consists of three major systems: a CO2 supply system, an organic solution delivery system, and a high pressure vessel. The components are: 1) CO2 cylinder, 2) cooling system, 3) piston pump, 4) heat exchanger, 5) HPLC pump, 6) solution, 7) high pressure Wessel, 8) back pressure regulator, 9) coaxial nozzle, 10) pressure meter, 11) gas bath.



#### **Figure 3.**

Electrospinning setup consists of a spinneret (typically a hypodermic syringe needle) connectedto a high-voltage, direct current power supply, a syringe pump, and a grounded collector. A polymer solution, is loaded into the syringe and this liquid is extruded from the needle tip at a constant rate by a syringe pump A) single electrospinning B) coaxial electrospinning (A coaxial setup uses a multiple solution feed system which allows for the injection of one solution into another at the tip of the) spinneret.



## **Figure 4.**

**Production of Legumin NPs by the coacervation method**: First, the extracted plant legumin is purified by ion exchange chromatography; then, phosphate buffer as desolvating agent is added to the purified protein and finally, the generated protein NP coacervates are stabilized through crosslinking by glutaraldehyde.



## **Figure 5.**

Three main steps to target specific cells at epithelial barriers by lectins including: Binding, Internalization and Intracellular transport of lectins or lectin conjugates



## **Figure 6.**

Schematic for intestinal drug transport using WGA-conjugated lipid NPs targeted to intestinal mucosal cells: (a,c) fluid phase endocytosis, (b) binding of WGA to glycoproteins followed by receptor-mediated endocytosis, (d) transcellular transport, (e) endocytosis might be blocked by absence of receptors or unsuitable conditions

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# **Table 1**







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**Table 2**

Recent researches about lectin-NP conjugates in drug delivery systems Recent researches about lectin-NP conjugates in drug delivery systems



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