

# Reflections on O<sub>2</sub> as a Biosignature in Exoplanetary Atmospheres

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## Abstract

Oxygenic photosynthesis is Earth's dominant metabolism, having evolved to harvest the largest expected energy source at the surface of most terrestrial habitable zone planets. Using CO<sub>2</sub> and H<sub>2</sub>O—molecules that are expected to be abundant and widespread on habitable terrestrial planets—oxygenic photosynthesis is plausible as a significant planetary process with a global impact. Photosynthetic O<sub>2</sub> has long been considered particularly robust as a sign of life on a habitable exoplanet, due to the lack of known “false positives”—geological or photochemical processes that could also produce large quantities of stable O<sub>2</sub>. O<sub>2</sub> has other advantages as a biosignature, including its high abundance and uniform distribution throughout the atmospheric column and its distinct, strong absorption in the visible and near-infrared. However, recent modeling work has shown that false positives for abundant oxygen or ozone could be produced by abiotic mechanisms, including photochemistry and atmospheric escape. Environmental factors for abiotic O<sub>2</sub> have been identified and will improve our ability to choose optimal targets and measurements to guard against false positives. Most of these false-positive mechanisms are dependent on properties of the host star and are often strongest for planets orbiting M dwarfs. In particular, selecting planets found within the conservative habitable zone and those orbiting host stars more massive than 0.4 M<sub>☉</sub> (M3V and earlier) may help avoid planets with abundant abiotic O<sub>2</sub> generated by water loss. Searching for O<sub>4</sub> or CO in the planetary spectrum, or the lack of H<sub>2</sub>O or CH<sub>4</sub>, could help discriminate between abiotic and biological sources of O<sub>2</sub> or O<sub>3</sub>. In advance of the next generation of telescopes, thorough evaluation of potential biosignatures—including likely environmental context and factors that could produce false positives—ultimately works to increase our confidence in life detection. Key Words: Biosignatures—Exoplanets—Oxygen—Photosynthesis—Planetary spectra. *Astrobiology* 17, 1022–1052.

## 1. Introduction

**P**OWERED BY OUR PARENT STAR, oxygenic photosynthesis is the dominant metabolism on our planet and is arguably the easiest to detect over interstellar distances. In addition to fueling the organisms that use it, oxygenic photosynthesis also creates a highly useful environmental energy gradient between oxygen and reduced organic carbon, the recombination of which produces sufficient energy to support multicellular, differentiated organisms (*e.g.*, Hedges *et al.*, 2004; Catling *et al.*, 2005; Falkowski and Godfrey, 2008; Reinhard *et al.*, 2016). Photosynthetic organisms, including cyanobacteria and vegetation, have left a global mark on Earth's current environment—significantly modifying its atmosphere, surface, geochemical cycling, ecological structure, and sea-

sonal appearance (Meadows, 2008; Meadows and Seager, 2010). The strongest and most widely recognized sign of life on our planet is photosynthetically generated O<sub>2</sub>, which is evenly mixed throughout the atmospheric column and composes 21% of our atmosphere. O<sub>2</sub> is also spectrally active at visible and near-IR (NIR) wavelengths; consequently, abundant oxygen will likely be the first sign of life that we search for on extrasolar planets with next-generation telescopes (Des Marais *et al.*, 2002; Brandt and Spiegel, 2014; Misra *et al.*, 2014a).

Phototrophs can also produce surface reflectance and seasonally dependent biosignatures, which could be sought as a secondary confirmation of a biospheric source for abundant O<sub>2</sub>. The vegetation red edge is currently Earth's dominant surface biosignature, producing a sharp rise in reflectivity longward of chlorophyll absorption at 688 nm (Gates *et al.*,

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1965). This effect produces a 2% variation in Earth's disk-averaged spectrum, as vegetated hemispheres of Earth replace ocean-dominated views (Arnold *et al.*, 2002; Montañés-Rodríguez *et al.*, 2006; Tinetti *et al.*, 2006b; Arnold, 2008). Many photosynthetic microorganisms, including algae and bacterial mats, produce similar characteristic reflectivity rises but at different wavelengths (Kiang *et al.*, 2007a; Hegde *et al.*, 2015). Time-dependent signs of photosynthetic life include the seasonal cycles of surface vegetation coverage and the annual cycle of global atmospheric CO<sub>2</sub> abundance. The latter occurs as phototrophs draw down CO<sub>2</sub> in the spring and summer during their productive growth phase and release it back to the atmosphere when vegetation dies and decays in autumn and winter. This cycle produces a 1–2% change in the CO<sub>2</sub> abundance over the year (Keeling *et al.*, 1976, 1996; Bacastow *et al.*, 1985), and its amplitude in the global average is a function of our land mass distribution and obliquity.

Although it requires complicated molecular machinery (Blankenship, 2010; Hohmann-Mariott and Blankenship, 2012), some form of photosynthesis is highly plausible as an eventually significant life process on extrasolar habitable planets. In particular, the environmental resources required to fuel and drive photosynthesis are likely common on a terrestrial planet in the surface liquid water habitable zone—that region around a star in which an Earth-like planet could support liquid water on its surface (Kasting *et al.*, 1993; Kopparapu *et al.*, 2013). Oxygenic photosynthesis uses light from the parent star—the ubiquitous and likely dominant energy source for a planet in the habitable zone; carbon dioxide—a common molecule in Solar System terrestrial planet atmospheres; and liquid water—which is expected to be abundant on a classically defined habitable planet. By using these globally available and energetically rich components of the habitable terrestrial environment, oxygenic photosynthesis would have a significant evolutionary advantage, allowing it to dominate its planetary surface environment (Kiang *et al.*, 2007b; Léger *et al.*, 2011; Kiang, 2014), which in turn would make it more detectable on a planetary scale. On our own planet, oxygenic phototrophs have been able to colonize a significant fraction of Earth's surface, populating marine, freshwater, and terrestrial habitats, including extreme environments such as hot springs (Ward and Castenholz, 2000), hypersaline settings (Oren, 2015), and polar regions (Painter *et al.*, 2001; Vincent, 2002; Williams *et al.*, 2003).

Earth's photosynthesis is also an ancient metabolic process that, despite its apparent complexity, developed relatively rapidly—appearing within a billion years after the Moon-forming impact. Multiple geological and geochemical lines of evidence suggest that anoxygenic photosynthesizers were present on Earth by at least 3.3 Ga (Westall *et al.*, 2011) and possibly as far back as 3.5 Ga (Buick *et al.*, 1981) or 3.7 Ga (Nutman *et al.*, 2016). These early anoxygenic phototrophs likely used strong reductants such as hydrogen or hydrogen sulfide—rather than the water used by oxygenic phototrophs—to donate electrons to a single photosystem that harvested light (Olson and Pierson, 1986). These reductants were produced by geological processes such as volcanism and would have been present in sufficient concentration to drive photosynthesis in only a relatively limited number of environments. Consequently, anoxygenic photosynthesizers would have been challenging to detect remotely in the global average, as they were spatially restricted

to these sources of strong reductants and are not known to produce gaseous by-products. The best way to detect these anoxygenic phototrophic organisms remotely may have been via their surface reflectivity signal (*e.g.*, Sanromá *et al.*, 2013, 2014). Nonphotosynthetic pigments developed for other purposes by phototrophs—including environmental survival (Dartnell, 2011)—may also produce surface reflectivity biosignatures prior to the rise of oxygen—both for the early Earth and on other planets (Schwieterman *et al.*, 2015a).

The limited environments suitable for early anoxygenic photosynthesizers provided evolutionary pressure for the development of a second photosystem that, when linked to the other, could use weaker—but abundant and globally widespread—reductants such as water, and produce oxygen as a waste product (Blankenship, 2010). The timing of the origin of oxygenic photosynthesis is not known but can be constrained by its subsequent impact on the environment. Geological evidence for oxidative weathering of continents suggests that O<sub>2</sub> may have temporarily risen to appreciable levels in our atmosphere as far back as 2.5 Ga (Anbar *et al.*, 2007; Kaufman *et al.*, 2007; Buick, 2008; Lyons *et al.*, 2014). This evidence for early O<sub>2</sub> production was subsequently confirmed by nitrogen (Garvin *et al.*, 2009) and selenium (Stüeken *et al.*, 2015a) isotope studies. Biomarkers in fluid inclusions also suggest that oxygenic photosynthesis was active at 2.45 Ga (Dutkiewicz *et al.*, 2006; George *et al.*, 2008, 2009). However, more recent isotope measurements point to transient low levels of O<sub>2</sub> that may have existed at even earlier times, potentially pushing the environmental impact of oxygenic photosynthesis back to 3.0–2.65 Ga (Czaja *et al.*, 2012; Crowe *et al.*, 2013; Planavsky *et al.*, 2014a; Riding *et al.*, 2014). Yet the irreversible, global accumulation of O<sub>2</sub> in the atmosphere—likely mediated by burial and removal of organic carbon from Earth's surface environment (Kasting, 2001; Lyons *et al.*, 2014)—evidently occurred somewhat later, between 2.45 and 2.2 Ga (Farquhar *et al.*, 2000; Bekker *et al.*, 2004; Canfield, 2005), with recent sulfur isotope studies (Luo *et al.*, 2016) placing it at 2.33 Ga. Recent evidence suggests that the rise of oxygen to its modern high levels in Earth's atmosphere may have been delayed even further, to 0.8 Ga (Planavsky *et al.*, 2014b), long after the advent of oxygenic photosynthesis.

## 2. O<sub>2</sub> as a Biosignature

For an atmospheric gas to audition for the role of planetary biosignature, it must be produced predominantly by life, build up to detectable levels in the atmosphere by resisting destruction or sequestration in the planetary environment, and exhibit strong spectral signatures that are within the wavelength range of planned astronomical instrumentation (*e.g.*, Domagal-Goldman *et al.*, 2011; Seager *et al.*, 2012). Photosynthetic O<sub>2</sub> excels in meeting all three criteria. It is the volatile by-product of the metabolism driven by the dominant source of energy on our planet's surface. Over time O<sub>2</sub> has risen to be the second most abundant gas in our atmosphere, after the dominant N<sub>2</sub>—which may also be biologically mediated through biological fixation, mineral absorption, and sequestration into the mantle (Catling and Kasting, 2007; Johnson and Goldblatt, 2015). The less abundant O<sub>2</sub>, however, still wins in spectral detectability. N<sub>2</sub> is a homonuclear molecule with no permanent

dipole, and it absorbs strongly only in the ultraviolet (80–100 nm)—a region that is crowded with UV bands from many species that may be common in terrestrial planet atmospheres, including H<sub>2</sub>O, CO<sub>2</sub>, CO, O<sub>2</sub>, CH<sub>4</sub>, and other hydrocarbons (Rothman *et al.*, 2013). The spectral region below 100 nm is also not accessible by ground-based telescopes due to strong absorption by N<sub>2</sub> in Earth's atmosphere. It is also challenging to observe from space, primarily due to sharp decreases in mirror reflectivity, even for UV-optimized mirror coatings (*e.g.*, Hennessy *et al.*, 2016). Reflected light observations of the planet in this wavelength region are also hampered by the relatively low UV flux from the star, and therefore from the planet, when compared to visible light observations.

In comparison, abundant atmospheric oxygen produces several strong and distinct features throughout the UV to mid-IR (MIR) from either O<sub>2</sub> itself, from its photochemical by-product, O<sub>3</sub>, or from collisional production of O<sub>4</sub> molecules (Richard *et al.*, 2012). O<sub>2</sub> has strong features at UV wavelengths <0.2 μm and in the visible/NIR. The latter include the γ band at 0.628 μm, B-band at 0.688 μm, A-band at 0.762 μm, and the *a*<sup>1</sup>Δ<sub>g</sub> band at 1.269 μm; the A-band is the strongest of these (Rothman *et al.*, 2013). O<sub>2</sub> collisional pairs (O<sub>4</sub>) form preferentially in high-density, high-O<sub>2</sub> atmospheres, although O<sub>4</sub> can also be seen in Earth's disk-averaged and transmission spectra (Tinetti *et al.*, 2006a; Pallé *et al.*, 2009). O<sub>4</sub> exhibits several features from 0.3–0.7 μm (Hermans *et al.*, 1999; Thalman and Volkamer, 2013), as well as at 1.06 (Greenblatt *et al.*, 1990) and 1.269 μm (Maté *et al.*, 1999; Misra *et al.*, 2014a; Schwieterman *et al.*, 2016a, 2016b). This latter O<sub>4</sub> band coincides with the O<sub>2</sub> *a*<sup>1</sup>Δ<sub>g</sub> band and O<sub>2</sub>-N<sub>2</sub> collisionally induced absorption (Smith and Newnham, 2000; Pallé *et al.*, 2009), which can complicate its detailed interpretation, although all these features clearly indicate the presence of O<sub>2</sub>. The presence of O<sub>2</sub> may also be inferred from its photochemical by-product, O<sub>3</sub> (Ratner and Walker, 1972), which has strong bands in the UV (0.2–0.3 μm), visible (0.5–0.7 μm) and MIR (9.6 μm) (Rothman *et al.*, 2013). In the MIR, O<sub>3</sub> is considered a more detectable proxy for the presence of oxygen (Segura *et al.*, 2003, 2005), as in this wavelength range O<sub>2</sub> has only extremely weak absorption at wavelengths within the strong 6.3 μm water band. However, O<sub>3</sub> can also be produced by the photolysis of other oxygen-bearing compounds, such as CO<sub>2</sub> and H<sub>2</sub>O. The superior detectability of O<sub>2</sub> and its photochemical by-products has contributed to O<sub>2</sub> and/or O<sub>3</sub> being a biosignature focus for missions that are being planned to search for evidence of life on exoplanets, including most recently the Exo-Coronagraphy (Stapelfeldt *et al.*, 2015), Exo-Starshade (Seager *et al.*, 2015), and High-Definition Space Telescope concepts (*e.g.*, Dalcanton *et al.*, 2015; Rioux *et al.*, 2015).

In addition to its strong absorption features in the visible and NIR, oxygen has another significant advantage as a biosignature because it is more likely to be evenly mixed throughout the atmospheric column, making it accessible to observation by transit spectroscopy. On Earth our abundant atmospheric O<sub>2</sub> maintains a near-constant mixing ratio throughout the troposphere, stratosphere, and much of the upper atmosphere. This is due to relatively low volcanic outgassing rates, a lack of strong surface sinks from our oxidized crust, and O<sub>2</sub>'s relative robustness to photolytic destruction in the stratosphere. As transit spectroscopy

cannot probe as deeply into an atmosphere as direct imaging due to either the presence of high clouds or hazes (Kreidberg *et al.*, 2014; Charnay *et al.*, 2015; Arney *et al.*, 2016, 2017) or the effects of refraction (García Muñoz *et al.*, 2012; Bétrémieux and Kaltenecker, 2013, 2014; Misra *et al.*, 2014b), this technique is generally most sensitive to the upper tropospheres and stratospheres of terrestrial planets, where our O<sub>2</sub> is still highly abundant. This gives O<sub>2</sub> an advantage over other proposed, often more complex biosignature molecules, including methanethiol, dimethyl disulfide (*e.g.*, Pilcher, 2003; Domagal-Goldman *et al.*, 2011; Seager and Bains, 2015), which are more susceptible to photolysis by UV radiation and so are confined largely to the lower troposphere, with significantly smaller concentrations in the stratosphere, which are less likely to be detectable (Domagal-Goldman *et al.*, 2011). Some organic molecules that could act as alternative biosignatures, such as ethane, may also resist photolytic destruction and be evenly mixed throughout the atmosphere (Domagal-Goldman *et al.*, 2011). Other more speculative biosignatures, such as NH<sub>3</sub> in oxidizing atmospheres as a disequilibrium biosignature (Lovelock, 1975) and NH<sub>3</sub> in H<sub>2</sub>-dominated atmospheres (Seager *et al.*, 2013a, 2013b), may have to contend with NH<sub>3</sub> being susceptible to being dissolved in an ocean or destroyed by photolysis (*e.g.*, Pavlov *et al.*, 2001). However, in H<sub>2</sub>-dominated atmospheres NH<sub>3</sub> may be more detectable in transmission due to the larger scale-height (Seager *et al.*, 2013b). Both ethane and ammonia have their strongest features in the MIR (near 13 and 9 μm, respectively; Rothman *et al.*, 2013), and sufficiently sensitive spectra of terrestrial planet atmospheres may be difficult to obtain at these wavelengths using transmission with the James Webb Space Telescope (JWST; Arney *et al.*, 2017). The MIR wavelength region is also not currently under consideration for future spaceborne exoplanet direct imaging telescopes (Dalcanton *et al.*, 2015), although it may be accessible to ground-based telescopes for very nearby planetary systems (Males *et al.*, 2014; Snellen *et al.*, 2015).

Although a single atmospheric species like O<sub>2</sub> can be considered a good biosignature, a combination of gases in chemical thermodynamic disequilibrium is considered even more robust (*e.g.*, Lederberg, 1965; Lovelock, 1965, 1975; Hitchcock and Lovelock, 1967). In chemical thermodynamic equilibrium, a mixture of gases is stable and will not react further. If atmospheric gases are seen in chemical disequilibrium, then this implies either an active planetary—possibly biological—source of the reactant gases, or the action of photochemistry. Once a chemical disequilibrium is identified, it is important to first rule out abiotic planetary processes as a source. The classic disequilibrium signature of life has been the simultaneous presence of both oxygen and methane in Earth's atmosphere, with a methane abundance that is many orders of magnitude above the equilibrium value (Lovelock, 1975; Sagan *et al.*, 1993). In chemical equilibrium these molecules would react, via formation of OH molecules from H<sub>2</sub>O and O<sub>3</sub> photolysis, to form CO<sub>2</sub> and H<sub>2</sub>O (*e.g.*, Segura *et al.*, 2005). Their disequilibrium implies active sources of both gases, which on Earth include photosynthetically generated O<sub>2</sub>, and CH<sub>4</sub> from methanogenic bacteria and archaea, which outproduce geologically generated methane from serpentinization (Kelley *et al.*, 2005; Guzmán-Marmolejo *et al.*, 2013) and other water/rock reactions by at least a factor of

25 (Kasting and Catling, 2003; Kharecha *et al.*, 2005; Etiope and Sherwood Lollar, 2013). Another proposed disequilibrium biosignature is the combination of O<sub>2</sub> from photosynthesis and N<sub>2</sub>O from bacterial denitrification (Lovelock, 1975). Like CH<sub>4</sub>, biological sources of N<sub>2</sub>O on Earth outproduce possible abiotic sources (Stein and Yung, 2003), of which several have been identified (Zhu-Barker *et al.*, 2015). These include chemodenitrification via brine/rock reactions in hypersaline environments, as observed in the Antarctic Dry Valleys (Samarkin *et al.*, 2010; Peters *et al.*, 2014); photolysis of nitrate ions in soils, an abiotic effect that may also show a seasonal signal (Rubasinghege *et al.*, 2011); or photochemical oxidation of atmospheric NH<sub>3</sub> (Stein and Yung, 2003). It has also recently been postulated that coronal mass ejections from the early Sun may have allowed energetic particles to drive the formation of N<sub>2</sub>O on the early Earth via destruction of N<sub>2</sub> and CO<sub>2</sub> in the upper atmosphere (Airapetian *et al.*, 2016). However, the O<sub>2</sub>/CH<sub>4</sub> and O<sub>2</sub>/N<sub>2</sub>O disequilibrium biosignatures were identified by considering only the composition of Earth's atmosphere, without considering the significant fraction of atmospheric gases that are dissolved in the ocean and undergo aqueous reactions. When the atmosphere/ocean system is considered, Earth's strongest disequilibrium biosignature is the simultaneous presence of abundant atmospheric O<sub>2</sub> and N<sub>2</sub> with an ocean, rather than a nitrate-rich ocean—which would be the equilibrium state (Krissansen-Totton *et al.*, 2016).

These disequilibrium biosignatures are potentially observable, although the O<sub>2</sub>/N<sub>2</sub>/ocean is perhaps the most challenging. O<sub>2</sub> and CH<sub>4</sub> may be detectable in a visible to NIR transmission or direct imaging spectrum of an exoplanet's atmosphere, or as the O<sub>3</sub> (9.6 μm)/CH<sub>4</sub> (7.7 μm) pair in the MIR. N<sub>2</sub>O has strong features in the MIR, so the O<sub>3</sub> (9.6 μm)/N<sub>2</sub>O (8.5 μm) pair would be the most diagnostic, although N<sub>2</sub>O also absorbs, albeit more weakly, in the NIR at 2.11 and 2.25 μm (Rothman *et al.*, 2013). For modern Earth, both the N<sub>2</sub>O and CH<sub>4</sub> abundances are low, at 0.3 and 1.8 ppm, respectively. Consequently, these disequilibria may have been more detectable during Earth's Proterozoic period, when the lower O<sub>2</sub> abundance could still generate potentially detectable O<sub>3</sub> yet also allow CH<sub>4</sub> to exist at higher atmospheric abundances than is currently possible (Segura *et al.*, 2003; Meadows, 2005; Kaltenecker *et al.*, 2007). Similarly, an N<sub>2</sub>O-rich atmosphere may have formed during the Proterozoic due to the biological unavailability of Cu in a strongly sulfidic ocean. The dearth of Cu would have inhibited the final step of the denitrification process and preferentially released N<sub>2</sub>O—rather than N<sub>2</sub>—into the planet's atmosphere (Buick, 2007).

These disequilibria could also become more detectable for terrestrial planets orbiting M dwarfs, when compared to Earth-Sun analogues. For an M dwarf host, the slope of the UV spectrum of the star can alter photochemistry to extend the atmospheric lifetimes, and therefore the abundance, of both CH<sub>4</sub> and N<sub>2</sub>O in the planetary atmosphere, given an Earth-like surface flux of these gases (Segura *et al.*, 2005; Grenfell *et al.*, 2012, 2014; Rugheimer *et al.*, 2015). In the case of CH<sub>4</sub>, the UV spectrum of the star is less effective at photolyzing ozone to produce O(<sup>1</sup>D), thereby slowing the production of the OH radical from H<sub>2</sub>O and inhibiting the destruction of CH<sub>4</sub> (Segura *et al.*, 2005). For N<sub>2</sub>O, atmospheric buildup occurs because the M dwarf produces less radiation that would directly photolyze N<sub>2</sub>O (<220 nm), when compared with a G dwarf (Segura *et al.*, 2005).

For the O<sub>2</sub>/N<sub>2</sub>/ocean disequilibrium biosignature, a surface ocean might be sought by using reflective “glint” from the ocean in the planet's phase-dependent photometry at visible to NIR wavelengths (Williams and Gaidos, 2008; Robinson *et al.*, 2010, 2014; Zugger *et al.*, 2011; Cowan *et al.*, 2012) while the oxygen could be sought as either O<sub>2</sub> or O<sub>4</sub> in the visible or NIR (Misra *et al.*, 2014a; Schwieterman *et al.*, 2016a, 2016b). N<sub>2</sub> could be searched for in the 80–100 nm region of the UV, or via N<sub>4</sub> absorption near 4.1 μm (Schwieterman *et al.*, 2015b). Uniquely identifying N<sub>2</sub> in the UV may be difficult without both an EUV-capable telescope and a relatively high spectral resolution—as the UV contains broad, overlapping bands from many atmospheric species. Alternatively, atmospheric N<sub>2</sub> may be sought near 4.1 μm via N<sub>4</sub> (also written as N<sub>2</sub>-N<sub>2</sub>) collisional absorption from pairs of N<sub>2</sub> molecules (Lafferty *et al.*, 1996), which overlap, but can be distinguished from, the 4.2 μm CO<sub>2</sub> absorption band in Earth's atmosphere (Schwieterman *et al.*, 2015b). However, the longer wavelength of the N<sub>4</sub> absorption will make these observations more challenging than the detection of Earth-like levels of O<sub>2</sub> for first-generation exoplanet characterization missions that will observe in the visible and/or NIR. To detect N<sub>4</sub>, future direct imaging missions would need a significantly more expensive, cooled telescope to observe wavelengths as long as 4.2 μm. Similarly, JWST has less sensitivity for exoplanet transmission spectra at these longer wavelengths, due to increasing thermal background and the drop in stellar brightness.

### 3. Oxygen Biosignature False Positives

O<sub>2</sub> has long been considered the most robust biosignature possible, because there were no known “false positives.” In this context, a false positive is an abiotic planetary process—such as volcanism or photochemistry—that could produce large, Earth-like quantities of O<sub>2</sub> on a habitable planet (*e.g.*, Rosenqvist and Chassefière, 1995). On Earth, the abiotic production of O<sub>2</sub>, principally by photolysis of water vapor, is at least a million times less than that produced by photosynthesis (Walker, 1977; Harman *et al.*, 2015). Initial modeling results attempting to identify false-positive mechanisms concentrated on photolysis of CO<sub>2</sub> as a possible source of abiotic O<sub>2</sub>. Rosenqvist and Chassefière (1995) used a steady-state photochemical model, and neglected surface sinks, to show that carbon dioxide-dominated abiotic atmospheres would produce an O<sub>2</sub> partial pressure of no more than 5 mbar (<600 ppm for the atmospheres modeled). Selsis *et al.* (2002) showed that for warm, ocean-bearing planets with 1 bar CO<sub>2</sub> atmospheres, O<sub>2</sub> could build up to only 1/10 of Earth's 21% present atmospheric level (PAL). However, even this relatively small amount was only possible if strong sinks for O<sub>2</sub> were neglected. These sinks include volcanic outgassing of reduced compounds, redox balance at the surface, and rainout of oxidized species, like H<sub>2</sub>O<sub>2</sub>. The latter mechanism acts as an O<sub>2</sub> sink because oxidized gases can react with reduced species in the crust or seawater to produce a net flux of H<sub>2</sub>, which reacts with and removes the O<sub>2</sub>. Segura *et al.* (2007) revisited the possible photolytic production of abiotic O<sub>2</sub> in CO<sub>2</sub>-rich atmospheres with the three O<sub>2</sub> sinks discussed above included. They explored whether a CO<sub>2</sub>-rich (up to 2 bar) atmosphere irradiated with strong incident UV from a

young Sun-like star (EK Dra; G1.5V)—a best-case scenario for CO<sub>2</sub> photolysis—could generate detectable levels of O<sub>2</sub>. However, with the sinks in place, their models returned O<sub>2</sub> abundances that were typically less than 1 ppb in the troposphere. Later, Hu *et al.* (2012) explored abiotic generation of O<sub>2</sub> on potentially habitable planets assuming no volcanic outgassing, but with rainout of oxidized species included, and achieved an abundance of only  $5 \times 10^{-3}$  PAL (0.1% abundance), which is also highly unlikely to be detectable. When more realistic assumptions of volcanic outgassing were used, their O<sub>2</sub> abundances dropped to a few parts per million. As another possible abiotic mechanism for O<sub>2</sub> buildup, Léger *et al.* (2011) discussed the efficacy of known catalysts for abiotic photogeneration of O<sub>2</sub> from water-splitting on a planetary surface but concluded that it was highly unlikely to occur on a habitable planet, although a subsequent study argued that this may be possible, if significantly large areas of the planet supported a catalyst (Narita *et al.*, 2015). Another possible mechanism, O<sub>2</sub> derived from the release of H<sub>2</sub>O<sub>2</sub> during the melt of a pole-to-pole ice-covered Snowball Earth (Liang *et al.*, 2006), also only produces abundances of a few parts per million of O<sub>2</sub> in the lower atmosphere.

### 3.1. False positives for oxygen and ozone on uninhabitable planets

Although buildup of significant amounts of abiotic O<sub>2</sub> or O<sub>3</sub> on habitable planets was initially precluded, it was thought more likely to occur on some uninhabitable planets. Mars and Venus have only small quantities of O<sub>3</sub> in their atmospheres, due to photolytic reactions with CO<sub>2</sub> (Blamont and Chassefière, 1993; Fast *et al.*, 2009; Montmessin *et al.*, 2011; Villanueva *et al.*, 2013). However, a terrestrial planet undergoing a runaway greenhouse or a frozen Mars-like planet beyond the outer edge of the habitable zone could both build up large amounts of O<sub>2</sub>, and possibly O<sub>3</sub>, in their atmospheres (Schindler and Kasting, 2000). The runaway greenhouse state would evaporate the planetary ocean and modify the atmospheric temperature structure to allow water to permeate the planet's stratosphere, above any UV-protective ozone layer. Photolysis of the unprotected H<sub>2</sub>O and the subsequent loss of hydrogen to space would allow atmospheric O<sub>2</sub> to accumulate (Ingersoll, 1969; Kasting, 1988). Subsequent buildup of stratospheric O<sub>3</sub> would be delayed until the odd H from water photolysis had been lost to space (Schindler and Kasting, 2000; Leconte *et al.*, 2013). For a dry terrestrial planet beyond the outer limit of the habitable zone that was large enough to retain heavier gases, but too small or too old to maintain volcanic outgassing, the available sinks for O<sub>2</sub> would be small. Surface ice would preclude reactions with a reduced surface, and the lack of reduced gases from volcanic outgassing would allow O<sub>2</sub> from photolysis of H<sub>2</sub>O or CO<sub>2</sub> to accumulate in the planetary atmosphere. The martian atmosphere today contains 0.1% O<sub>2</sub>, and Mars may have built up more O<sub>2</sub> if it had been more massive and therefore resistant to O<sub>2</sub>'s atmospheric loss (McElroy, 1972).

Hot sub-Neptune planets may also form O<sub>2</sub>-rich atmospheres when they are closer to the star than the habitable zone. Simulations of thermochemical and photochemical equilibrium states for hot (radiative equilibrium temperatures from 500 to 2000 K) super-Earth/mini-Neptune planets as a function of atmospheric hydrogen fraction and carbon-

to-oxygen ratio do generate O<sub>2</sub>-rich atmospheres in some cases (Hu and Seager, 2014). Specifically, these simulations show that planets with hydrogen fractions below about 50%, and with C/O ratios <0.1, form O<sub>2</sub>-rich atmospheres of at least 20% O<sub>2</sub> (Hu and Seager, 2014, their Fig. 7). However, these hot sub-Neptunes would also display very high fractions of water vapor (from 1% to 60%) that would persist to high altitude (Hu and Seager, 2014) and potentially be visible to transit spectroscopy. This would help discriminate them from habitable Earth-like planets, which have water in the troposphere but relatively dry stratospheres, that would produce weak or no water vapor features in transmission.

Consequently, identifying known false positives for oxygen on planets not within the habitable zone may be relatively straightforward if the planet's orbit—and possibly size or mass—is known and the spectral range encompasses absorption from H<sub>2</sub>O as well as O<sub>2</sub> and O<sub>3</sub>. Such a planet will be outside or very close to the conservative habitable zone boundaries for its mass and atmospheric characteristics (Kopparapu *et al.*, 2013, 2014) and, if showing the abundant O<sub>2</sub> of a recent runaway or a hot Neptune, will also exhibit strong stratospheric H<sub>2</sub>O, but no O<sub>3</sub> in either transmission or direct detection spectra. In the case of the planet beyond the outer edge of the habitable zone with abundant O<sub>2</sub>, and for an older runaway greenhouse atmosphere close to the star that also exhibits both O<sub>2</sub> and a large O<sub>3</sub> component, water vapor will instead be weak or nonexistent in transmission or direct imaging spectra (Schindler and Kasting, 2000).

### 3.2. New false positives for oxygen and ozone on habitable zone planets

The recent identification of multiple mechanisms for abiotic generation of atmospheric oxygen or ozone for planets in the habitable zone presents a far more serious challenge to the robustness of oxygen as a biosignature. In many cases, water vapor, an extremely common molecule—and one hoped to be abundant on a habitable planet—serves as the key source of abiotic oxygen. Several of these false-positive mechanisms are more likely on planets orbiting in the habitable zones of M dwarf stars (Domagal-Goldman and Meadows, 2010; Domagal-Goldman *et al.*, 2014; Tian *et al.*, 2014; Gao *et al.*, 2015; Harman *et al.*, 2015; Luger and Barnes, 2015), but at least one mechanism could operate for planets orbiting in the habitable zones of stars of any spectral type, including G dwarfs (Wordsworth and Pierrehumbert, 2014). Photochemistry plays a significant role in all of these mechanisms and can leave telltale signs of its role in forming abiotic oxygen or ozone in the planetary atmosphere. By exploring these mechanisms and identifying the conditions under which they are most likely to occur, we will improve our interpretation of the spectra of habitable planets and be able to increase our confidence in life detection on exoplanets. In particular, we can improve target selection for the search for life on exoplanets by determining the stellar or planetary characteristics that increase the likelihood of abiotic O<sub>2</sub> or O<sub>3</sub> production, and identify the specific measurements required to discriminate between biological and abiotic production of O<sub>2</sub> and O<sub>3</sub>.

3.2.1. Low noncondensable gas inventories. The possible abiotic O<sub>2</sub> generation mechanism that will likely be the

most difficult to recognize involves photolysis of water—and subsequent hydrogen loss—from terrestrial atmospheres that are depleted in noncondensable gases (Wordsworth and Pierrehumbert, 2014). Because this mechanism relies primarily on a planetary property, the lack of atmospheric gases that will not condense at typical terrestrial planetary temperatures and pressures (*e.g.*, N<sub>2</sub> or O<sub>2</sub>), this mechanism may work for planets orbiting stars of any spectral type, including Sun-like stars. In this proposed mechanism, the “cold trap”—the rapid reduction in temperature with altitude that causes rising water vapor on Earth to condense and thereby remain trapped in the troposphere—is no longer effective. This can occur either when the atmospheric temperature is high or when the inventory of noncondensable gases—for example, N<sub>2</sub> or Ar—is low. In these cases, water rises into the stratosphere, where it is more vulnerable to photolysis by incident stellar UV radiation. The resultant H atoms escape, leaving oxygen to build up abiotically in the atmosphere. This continues until O<sub>2</sub>, itself a noncondensable gas, reaches a sufficiently high abundance that it is able to establish a cold trap and halt the loss of water vapor.

Wordsworth and Pierrehumbert (2014) describe the relationship between surface temperature and surface partial pressure of noncondensable gas that determines whether a planet is more likely to have a moist upper atmosphere or one that remains dry. For 1 bar of N<sub>2</sub> in an N<sub>2</sub>-H<sub>2</sub>O atmosphere, a surface temperature >340 K would be required to produce a moist upper atmosphere. Conversely, for Earth’s current surface temperature, a reduction in our current N<sub>2</sub> partial pressure by a factor of ~10 would be enough to move a planet into the moist upper atmosphere regime, where it would be susceptible to water loss and abiotic O<sub>2</sub> buildup (Wordsworth and Pierrehumbert, 2014). Interestingly, even though the Archean Earth’s troposphere likely contained very little oxygen, there is recent evidence from preserved micrometeorites that the stratosphere of the Archean contained oxygen in abundances similar to modern Earth’s, as might possibly have been formed by a water-rich stratosphere that was undergoing photolysis (Tomkins *et al.*, 2016).

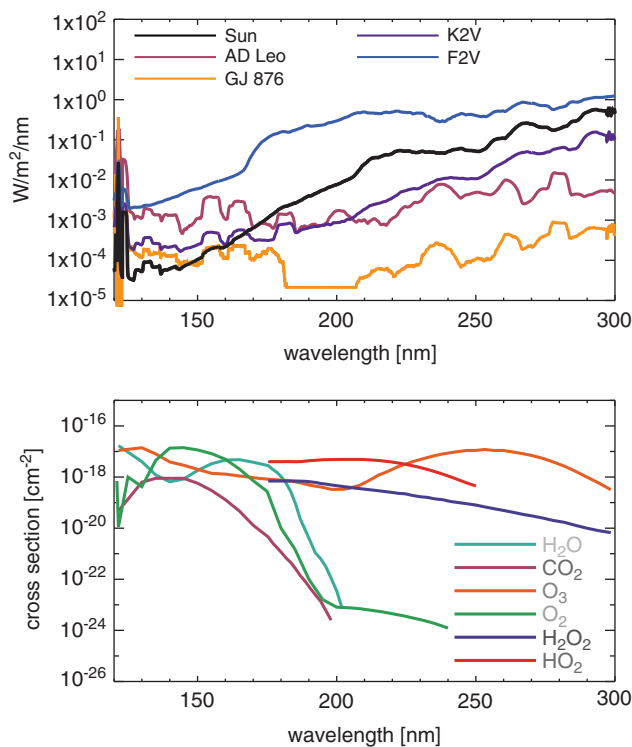
Terrestrial exoplanets with different N<sub>2</sub> inventories are entirely plausible, and the delivery and loss of volatiles from a terrestrial planet are dependent on a number of planetary and exogenous processes that will be difficult to know *a priori*. In our own solar system, Mars and Venus have smaller ( $2 \times 10^{-4}$ ) and larger (3.3 times) atmospheric nitrogen abundances than Earth (Mancinelli and Banin, 2003; Hunten, 1993), with Venus’ water loss and subsequent oxidation of the mantle potentially enhancing the outgassing of N<sub>2</sub> (Wordsworth, 2016). Hydrodynamic escape during the pre-main sequence phase of M dwarfs may also lead to atmospheric N<sub>2</sub> loss. Recent modeling estimates for Proxima Centauri b, a possibly terrestrial planet with a minimum mass of  $1.3 M_{\oplus}$  orbiting an M5.5V star (Anglada-Escudé *et al.*, 2016), suggest that less than a bar of N<sub>2</sub> may be lost this way in the first 100 million years (Ribas *et al.*, 2016).

Earth may have also exhibited different N<sub>2</sub> partial pressures throughout its history, as its atmosphere evolved. Paleobarometry measurements of 2.7 Ga fossilized raindrop imprints indicate that Earth’s atmospheric pressure was no more than twice the modern day’s but potentially as low as 0.5 bar (Som *et al.*, 2012). Subsequent analyses of nitrogen and argon isotopes in 3.5 Ga rock (Marty *et al.*, 2013) showed a similar range

of 1.1–0.5 bar for the partial pressure of N<sub>2</sub> in the ancient atmosphere. Som *et al.* (2016) used the distribution of sizes in gas vesicles in 2.7 Ga volcanic rocks to derive an atmospheric pressure of  $0.23 \pm 0.23$  bar, which when combined with the previous studies suggested that 0.5 bar was a likely upper limit to the atmospheric pressure in Earth’s Archean.

For an abiotic planet that is largely Earth-like except for the N<sub>2</sub> inventory, this proposed mechanism could result in a planet that nonetheless exhibited a very Earth-like O<sub>2</sub> partial pressure, while losing significant amounts of water. A pure water vapor atmosphere would initially be susceptible to water loss and O<sub>2</sub> buildup, until the (noncondensable) O<sub>2</sub> constituted a large enough fraction of the atmosphere to produce a cold trap for the water vapor. Under the assumption of redox balance—with loss of hydrogen to space balanced by the surface removal rate of oxidizing material for an Earth-like (288 K surface temperature) planet—the resultant atmospheric fraction of O<sub>2</sub> was found to be 0.15 bar (Wordsworth and Pierrehumbert, 2014). However, this amount of O<sub>2</sub> could be higher or lower, depending on surface and tropopause temperatures. In the case of a more Earth-like atmosphere containing N<sub>2</sub>-H<sub>2</sub>O and sufficient CO<sub>2</sub> to produce our current surface temperature, once the N<sub>2</sub> content drops significantly, to below a few percent of Earth’s current levels, the upper atmosphere becomes rich in H<sub>2</sub>O and susceptible to hydrogen loss and O<sub>2</sub> buildup. If this process persists for 4 billion years around a Sun-like star—and the resultant O<sub>2</sub> is absorbed by the planet’s surface and interior, instead of building up in the atmosphere and shutting off the water loss—then as much as a third of an Earth ocean could ultimately be lost from the planetary environment. Even higher water loss rates are expected for stellar XUV fluxes higher than the Sun’s current flux (Wordsworth and Pierrehumbert, 2014), as might be expected for planets orbiting F dwarfs and active M dwarfs (*e.g.*, France *et al.*, 2016; Fig. 1).

**3.2.2. Enhanced pre-main sequence stellar luminosity.** Another recently proposed mechanism to generate high levels of abiotic oxygen, first proposed by Luger and Barnes (2015), focuses on the effects of the pre-main sequence, superluminous phase of young stars on terrestrial planet environments. Before settling in to their main sequence hydrogen-burning phase, young stars can be significantly more luminous due to additional energy production from their extended contraction phase (*e.g.*, Baraffe *et al.*, 1998), so planets that form in what will become the main sequence habitable zone are subjected to very high levels of radiation early on (Lissauer, 2007). This superluminous phase is most pronounced—and of longer duration—for smaller-mass M dwarfs than for any other stellar spectral type, where it can extend for up to 1 billion years (Baraffe *et al.*, 1998). Modeling suggests that during this superluminous phase a terrestrial planet that forms within what will become the main sequence habitable zone around an M dwarf star may lose up to several Earth ocean equivalents of water due to evaporation and hydrodynamic escape, and this can lead to generation of large amounts of abiotic O<sub>2</sub> (Luger and Barnes, 2015; Tian, 2015). This superluminous phase may persist for 200 million years or more for M dwarf stars of less than  $0.4 M_{\odot}$  (M3V and later; Baraffe *et al.*, 1998). The resultant water-rich atmosphere will be susceptible to



**FIG. 1.** Comparison of stellar UV fluxes and molecular absorption cross sections. The upper panel shows the UV fluxes for a range of stellar types, including the two M dwarf stars AD Leo (M3.5V) and GJ 876 (M4V). The lower panel shows the UV absorption cross sections for oxygen and ozone, as well as several key molecules involved in the abiotic production of oxygen and ozone. Because the stellar UV spectra have different slopes, and the molecules have different regions of the UV spectrum in which they absorb radiation and can potentially be photolyzed, stellar spectral slopes can drive the formation or destruction of abiotic oxygen and ozone. (Credit: G. Arney)

photolysis and hydrogen escape, thereby producing potentially hundreds or thousands of bar of  $O_2$  (Luger and Barnes, 2015) as photolysis of an Earth ocean of water can result in  $\sim 240$  bar of  $O_2$  (Kasting, 1997). The amount of  $O_2$  generated is a strong function of several factors, including stellar spectral type, XUV flux, original water inventory, planetary mass, and the position of the planet in the habitable zone. Planets in the outer regions of the habitable zone for M0–M3V stars are the least likely to generate significant amounts of abiotic  $O_2$  (Luger and Barnes, 2015), along with M1–M3V dwarfs with lower stellar XUV, as slower water loss may preclude  $O_2$  generation and buildup (Tian, 2015). However, for planets orbiting later type M dwarfs (M4V and later), depending on the planet’s original water inventory, mass, stellar parameters, and the strength of surface sinks (Rosenqvist and Chassefière, 1995; Schaefer *et al.*, 2016), several Earth ocean equivalents of water, or up to several hundreds of bar of photolytically produced  $O_2$ , could plausibly build up in the atmospheres of these planets (Luger and Barnes, 2015). For the recently discovered Proxima Centauri b, orbiting an M5.5V star, estimates for maximum ocean loss during the pre-main sequence phase span less than one ocean, assuming a constant XUV flux for the host star for the first 3 billion years (Ribas *et al.*, 2016),

to 3–10 oceans lost, assuming that the XUV scales with the decreasing bolometric luminosity (Barnes *et al.*, 2016).

Note also that this mechanism for  $O_2$  production and buildup relies on water loss being shut down within 1 billion years of planet formation, when the host star joins the main sequence and dims in luminosity and the habitable zone sweeps in to encompass the planet. For planets that form much closer to the star than the main sequence habitable zone, and do not enter the habitable zone during the star’s lifetime, hydrodynamic escape and subsequent loss of photolytically produced  $O_2$  can be enhanced. Schaefer *et al.* (2016) modeled this scenario for GJ 1132b, a 5-billion-year-old 1.2 Earth radii planet that is currently still closer to its star than the habitable zone (Berta-Thompson *et al.*, 2015). Schaefer *et al.*’s XUV Model A is comparable to that used by Luger and Barnes (2015), yet their simulations showed that as the planet’s total amount of water decreased over 5 billion years, the ratio of  $O_2$  escape to  $O_2$  production from water photolysis increased, driving the planet toward states with no more than a few bar of  $O_2$  and  $H_2O$  remaining (Schaefer *et al.*, 2016). For these “hot Earths,” they concluded that hundreds to thousands of bar of  $O_2$  could persist only for planets with very large initial water abundances of  $>5\%$  wt %, which over 5 billion years would never be sufficiently depleted in water to get to the  $O_2$  escape “runaway” scenario. However, as Luger and Barnes (2015) and Barnes *et al.* (2016) show, a planet that enters the habitable zone prior to 1 billion years can also shut down hydrodynamic escape and avoid massive  $O_2$  loss and could do so with water abundances significantly lower than 5% wt %.

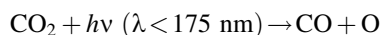
Once they enter the habitable zone, these M dwarf planets with  $O_2$ -rich atmospheres may be completely desiccated, or could still support a surface ocean, especially if the superluminous phase ends before the runaway process is complete (Luger and Barnes, 2015; Barnes *et al.*, 2016). Water may also be delivered after the superluminous phase is complete, although in the case of M dwarfs the billion-year superluminous phase will significantly exceed the up to several million-year timescales for planet formation, and the majority of volatile delivery (Raymond *et al.*, 2008). Volatile outgassing from the mantle may be another way to form a surface ocean. However, in the case of Earth, noble gas analysis suggests that the vast majority of our volatiles (80–95%) outgassed within the first 500 million years (Turner, 1989). An ocean of water may have remained in the mantle after this initial loss (Albarède, 2009; Sleep *et al.*, 2012), and it is being more slowly outgassed. If a similar process works for terrestrial planets orbiting M dwarfs, then these planets would be susceptible to significant loss of mantle-outgassed water during the pre-main sequence phase. However, under favorable conditions a terrestrial planet could perhaps, over billions of years, slowly accumulate a surface ocean from outgassed water, after the M dwarf has settled into its more benign main sequence phase. Conversely, if the mantle is desiccated during the extended pre-main sequence phase, then the development of a later ocean may be stymied. The lack of surface water and the inhibition of subduction, due to a lack of hydrated minerals, will severely limit the recycling of water back into the mantle for later release, as was possibly the case for the now largely desiccated Venus (Grinspoon, 1993; Hamano *et al.*, 2013).

**3.2.3. Stellar spectrum-driven photochemical production.** A perhaps more subtle mechanism for formation of abiotic

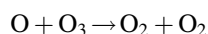
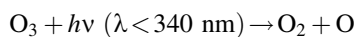
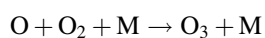
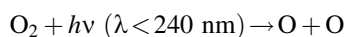
O<sub>2</sub> and its proxy, O<sub>3</sub>, is planetary photochemistry. In this context, photochemistry is primarily an interaction between molecules in a planetary atmosphere and the incident UV spectrum of the parent star, which can split molecules and drive reactions that change the composition of the atmosphere. Each photochemical reaction requires photons that exceed a threshold energy level—or equivalently, photons at wavelengths shorter than a threshold wavelength—to be absorbed by the molecule undergoing photolysis. Planetary photochemistry is therefore strongly sensitive to the UV spectral energy distribution of the parent star (*e.g.*, Segura *et al.*, 2003, 2005; Grenfell *et al.*, 2007, 2014; Rugheimer *et al.*, 2013, 2015), and in particular the ratio of shorter- to longer-wavelength UV radiation (Fig. 1).

In addition to the stellar UV radiation, the entire planetary environment can be involved in providing sources and sinks for the photochemical reactions. For example, photochemical abiotic O<sub>2</sub> production will depend strongly on the source of O atoms, which is primarily controlled by the photolysis rates of H<sub>2</sub>O, CO<sub>2</sub>, SO<sub>2</sub>, and other O-bearing gases. Similarly, the atmospheric sink will depend on the availability of H atoms in the atmosphere from H<sub>2</sub> and H<sub>2</sub>S. While water vapor can be sourced from a planetary ocean, the availability of other gases, such as CO<sub>2</sub>, SO<sub>2</sub>, and reducing (H-bearing) gases, is governed by the planet's volcanic outgassing rates, which can be sustainable on long geological timescales. CO<sub>2</sub> in particular is likely a common atmospheric gas on terrestrial planets. In our own solar system, CO<sub>2</sub> dominates the atmospheric composition on Mars and Venus and was likely a significant component of the early Earth's atmosphere as well (Kasting, 1993; Sheldon, 2006; Sleep, 2010; Driese *et al.*, 2011).

The photochemical production of O<sub>2</sub> is governed primarily by a few key photochemical reactions. CO<sub>2</sub> is photolyzed to produce CO and O atoms, by wavelengths of light shorter than 175 nm:



O released from this reaction or from similar reaction for other O-bearing gases may recombine to form O<sub>2</sub> via three-body reactions, and eventually O<sub>3</sub>. The distribution of those O atoms between O<sub>2</sub> and O<sub>3</sub> is critical to the concentration of either species and is primarily controlled by four reactions known as the Chapman mechanism (Chapman, 1930):



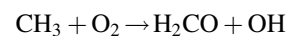
Here,  $h\nu$  represents the photon energy for photons with wavelengths in the indicated range, and the M denotes a third molecule that does not participate in the reaction but only carries off excess energy. Note that although the Chapman mechanism dominates ozone production, other reactions, especially with oxygen-bearing species such as HO<sub>x</sub> (*e.g.*, H<sub>2</sub>O, OH), NO<sub>x</sub> (*e.g.*, N<sub>2</sub>O, NO, NO<sub>2</sub>) and chlorine-bearing compounds can also affect O<sub>2</sub> and O<sub>3</sub>

abundances (*e.g.*, Stolarski and Cicerone, 1974; Johnston, 1975; Thornton *et al.*, 2002). In CO<sub>2</sub>- and O<sub>2</sub>-rich atmospheres, these molecules can reduce O<sub>2</sub> and O<sub>3</sub> abundances by catalyzing CO<sub>2</sub> recombination (Domagal-Goldman *et al.*, 2014), although these molecules can react to generate O<sub>2</sub> in anoxic environments such as the early Earth, especially for planets orbiting stars with relatively high far-UV (FUV) fluxes (Arney *et al.*, 2017). The effects of these catalytic species on O<sub>2</sub> and O<sub>3</sub> abundances in planetary atmospheres are discussed in more detail below.

Concentrating on the direct photolytic effects described in the Chapman mechanism, the photolysis of CO<sub>2</sub> and O<sub>2</sub> requires FUV photons at wavelengths < 200 nm, and the process provides O atoms to help form O<sub>2</sub> and O<sub>3</sub>. Balancing O<sub>3</sub> production, the photolytic destruction of O<sub>3</sub> is primarily driven by photons in the mid-UV (MUV: 200 nm < λ < 300 nm) but can also be achieved by photons in the near-UV (NUV: 300 nm < λ < 440 nm) and visible (~440 to 800 nm) (Sander *et al.*, 2006). Since CO<sub>2</sub>-photolyzing FUV photons are produced by stellar activity (Pace and Pasquini, 2004) and O<sub>3</sub>-photolyzing MUV-NUV and visible photons are primarily due to a star's blackbody radiation, then the spectrum and activity levels of the host star will strongly affect the balance between production and destruction of oxygen and ozone. Consequently, to interpret the atmospheric composition of terrestrial exoplanets, it will be extremely important to understand the UV spectrum of the host star, especially for the high-UV-output F and M dwarfs (*e.g.*, Hawley *et al.*, 2003; Walkowicz *et al.*, 2008; France *et al.*, 2012, 2016; Loyd *et al.*, 2016; Youngblood *et al.*, 2016). Because Ly α emission dominates M dwarf UV output and can photolyze CH<sub>4</sub>, CO<sub>2</sub>, and H<sub>2</sub>O in the upper planetary atmosphere (>60 km in Earth's atmosphere), it may also be important to reconstruct the star's intrinsic Ly α (112 nm) profile by correcting the observed Ly α line for the effects of interstellar absorption (Wood *et al.*, 2005; Linsky *et al.*, 2013; Shkolnik *et al.*, 2014; Youngblood *et al.*, 2016).

The sinks for O<sub>2</sub> and O<sub>3</sub> are primarily controlled by the chemical context of the atmosphere and oceans, including the redox state of the atmosphere. As discussed above, these sinks include the destruction of O<sub>2</sub> and O<sub>3</sub> via direct photolysis but also include reactions with catalytic agents that can work to drive the recombination of O atoms with the parent molecule. On Earth, these include reactions with Cl and other halogens.

In anoxic atmospheres, the greatest sinks for O<sub>2</sub> and O<sub>3</sub> are reactions with reduced radicals in the atmosphere, such as

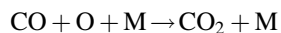


When the concentration of reduced species such as CH<sub>4</sub> increases in the atmosphere, so do the concentrations of photolytically produced radicals such as CH<sub>3</sub>, and these can react with O<sub>2</sub> and O<sub>3</sub>, keeping their concentrations low. Major abiotic sources of reduced species include volcanic outgassing of H<sub>2</sub> and submarine production of CH<sub>4</sub> (Kelley *et al.*, 2005; Guzmán-Marmolejo *et al.*, 2013), and their sinks are primarily determined by the redox state of the atmosphere and oceans (Kharecha *et al.*, 2005; Domagal-Goldman *et al.*, 2014; Harman *et al.*, 2015).

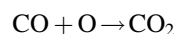
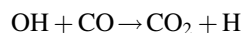
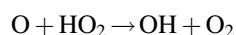
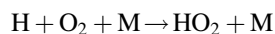
In the absence of reduced radicals, the main counter to O<sub>2</sub>/O<sub>3</sub> production by CO<sub>2</sub> photolysis would be recombination of



CO and O to form CO<sub>2</sub>. However, the direct three-body recombination reaction for CO<sub>2</sub>, which would remove O from the atmosphere,



is spin forbidden and proceeds extremely slowly. Consequently, catalysts are much more efficient at stabilizing CO<sub>2</sub> against photolysis in a planetary atmosphere. In particular, photolysis of water vapor ( $\text{H}_2\text{O} + h\nu \rightarrow \text{H} + \text{OH}$ ) and HO<sub>x</sub> chemistry can facilitate the recombination of CO<sub>2</sub> via the following reaction sequence:



Conversely, an atmosphere with low amounts of water vapor can accumulate higher amounts of O<sub>2</sub> and O<sub>3</sub>—due to the slowdown of the catalytic recombination of CO<sub>2</sub>.

In addition to reactions in the atmosphere, there is also a possible role for aqueous reactions in the ocean as sinks to remove atmospheric oxygen. For example, the aqueous reaction of dissolved CO and O<sub>2</sub> to reform CO<sub>2</sub> and draw down atmospheric oxygen is poorly understood but is crucial to understanding the final balance of O<sub>2</sub> in the atmosphere (Harman *et al.*, 2015). Similarly, weathering of surface crust (*e.g.*, Anbar *et al.*, 2007) and the sequestration of O<sub>2</sub> into the planetary mantle (*e.g.*, Schaefer *et al.*, 2016) are key processes that control O<sub>2</sub> drawdown and could result in abiotic O<sub>2</sub> buildup if weak. Conversely, if these processes are aggressive enough, they could instead result in a false negative for biologically produced O<sub>2</sub> by scrubbing photochemically generated O<sub>2</sub> from a planetary atmosphere. This may have been the case on our planet, where the evolution of oxygenic photosynthesis may have predated the rise of atmospheric oxygen by hundreds of millions to billions of years (Catling *et al.*, 2001; Catling and Kasting, 2007; Lyons *et al.*, 2014; Planavsky *et al.*, 2014a).

Several groups have explored these complicated, interacting factors and have found multiple mechanisms by which photochemical processes may generate abiotic O<sub>2</sub> and O<sub>3</sub> on terrestrial planets in the habitable zone (Domagal-Goldman and Meadows, 2010; Hu *et al.*, 2012; Domagal-Goldman *et al.*, 2014; Tian *et al.*, 2014; Gao *et al.*, 2015; Harman *et al.*, 2015). However, the amounts of abiotic O<sub>2</sub> and O<sub>3</sub> generated in these simulations differ, in part due to different assumptions about the ability of CO and O<sub>2</sub> to react in seawater.

Domagal-Goldman *et al.* (2014) used an altitude-dependent (1-D) photochemical model to perform an extensive parameter sweep through a wide variety of volcanic gas fluxes and stellar energy distributions to explore those conditions that could lead to abiotic O<sub>2</sub> and O<sub>3</sub> production. The approach used was conservative with respect to generation of O<sub>2</sub> and O<sub>3</sub> and included the enhanced boundary

conditions needed for redox balance in the atmosphere and ocean. They assumed that a redox imbalance at the atmosphere-ocean interface would work to increase the ocean concentration of the species contributing to the redox imbalance. This allowed the aqueous CO and O<sub>2</sub> reaction to occur, serving as a strong sink for O<sub>2</sub>. However, the increased concentration would ultimately slow down the rate the species dissolved into the ocean—as that rate is dependent on concentration—until an equilibrium was reached. Even with this conservative approach, Domagal-Goldman *et al.* (2014) still found that it was possible to generate abiotic ozone, primarily from the photolysis of CO<sub>2</sub>, that was only a factor of 10 less than Earth's current ozone abundance. Small amounts of abiotic O<sub>2</sub>, of the order of 40 ppm, significantly less than Earth's current 21% O<sub>2</sub> abundance, were also generated on these planets. This mechanism for abiotic O<sub>3</sub> generation was most likely to occur for habitable zone planets orbiting stars that have high FUV fluxes, as might be expected for F dwarf and active M dwarf stars, and for atmospheres that have high CO<sub>2</sub> concentrations and low H<sub>2</sub> concentrations (Domagal-Goldman *et al.*, 2014). This work demonstrated that O<sub>3</sub> could build up to potentially detectable levels, even if CO and O<sub>2</sub> can react in seawater. It also identified the environmental conditions associated with this false positive: high FUV fluxes from the host star, low CH<sub>4</sub> in the atmosphere, and high amounts of volcanic CO<sub>2</sub>. Given that stellar UV output and volcanic sources of CO<sub>2</sub> and H<sub>2</sub> could be sustained for billions of years on a geologically active planet, this abiotic source of O<sub>3</sub> could be sustainable on similar timescales. Earlier work by Hu *et al.* (2012) had also been able to generate high levels of O<sub>3</sub> in CO<sub>2</sub>-dominated atmospheres orbiting Sun-like stars using a similar mechanism, but only if the assumption of no volcanic outgassing was used. In that case, O<sub>3</sub> column abundances were only a factor of 3 lower than that on present-day Earth. When more realistic volcanic outgassing rates were used, however, the abundance dropped by 4 orders of magnitude.

Tian *et al.* (2014) used a 1-D photochemical model for a smaller array of cases, varying surface temperature and CO<sub>2</sub> mixing ratio (from 1% to 10%), to show that abiotic O<sub>2</sub> and O<sub>3</sub> could be generated around M dwarf stars that have FUV/NUV ratios significantly larger than the solar value. These researchers used HST observations of six planet-hosting M dwarfs (France *et al.*, 2012, 2013), showing that five of them are strong emitters at FUV wavelengths but weak emitters in the NUV. The remaining M dwarf emitted too little UV radiation to get a definitive measurement. The results of the photochemistry model showed buildup of O<sub>2</sub> in the atmosphere of a hypothetical planet orbiting GJ 876 to a mixing ratio of 0.2% (10<sup>-3</sup> PAL) below 70 km. The authors argued that this is primarily due to the M dwarf's significantly weaker NUV compared to FUV flux, which slows the production of OH radicals (from photolysis of H<sub>2</sub>O<sub>2</sub> and HO<sub>2</sub>) that catalyze the recombination of CO<sub>2</sub> and the removal of atmospheric O<sub>2</sub>. However, the absorption cross sections for the H<sub>2</sub>O<sub>2</sub> and HO<sub>2</sub> molecules are not measured in the FUV (Fig. 1) and are unavailable to be included in photochemical models, so it is not clear what the action of the more abundant FUV radiation would be on the production of OH radicals from these molecules. Harman *et al.* (2015) instead argued that it is in fact decreased water vapor photolysis,

which slows by a factor of 400 for the GJ 876 case when compared to the solar case, coupled with the longer lifetimes of HO<sub>2</sub> and H<sub>2</sub>O<sub>2</sub> that ultimately limits the availability of OH for the recombination of CO and O<sub>2</sub> in these atmospheres.

The lower boundary conditions may also lead to this O<sub>2</sub> buildup. In the Tian *et al.* (2014) model, surface CO and O<sub>2</sub> deposition velocities were set equal to each other, as the two molecules have similar diffusivities and solubilities in water. However, O<sub>2</sub> reacts much more strongly than CO with other constituents in the ocean, and O<sub>2</sub> would be preferentially removed from ocean water. This would increase the deposition rate of O<sub>2</sub> compared to CO and generate a net sink for O<sub>2</sub> that is not taken into account in this study. Also, although the O<sub>2</sub> may be generated by photochemistry in the stratosphere, it is the deposition rates at the lower boundary that control the vertical profile and column depth of the gas in the atmosphere (*e.g.*, see Fig. 3 in Harman *et al.*, 2015). Additionally, the highest O<sub>2</sub> and O<sub>3</sub> values modeled were associated with an assumption that CO and O<sub>2</sub> cannot recombine in seawater, which therefore assumes no aqueous sink for CO, as well. CO would also tend to accumulate in an atmosphere that could generate abiotic O<sub>2</sub>, and act as an indicator for this photochemical mechanism.

Following on from Domagal-Goldman *et al.* (2014) and Tian *et al.* (2014), Harman *et al.* (2015) used a 1-D photochemical model to perform a comparison of abiotic O<sub>2</sub> from the photolysis of CO<sub>2</sub> on habitable planets with water oceans orbiting stars of different spectral types. Like Domagal-Goldman *et al.* (2014), they modeled systems with global redox balance between the ocean-atmosphere system, although they neglected oxygen sinks such as surface weathering. However, they simulated boundary condition limits for the deposition of CO and O<sub>2</sub> into the ocean (*i.e.*, insoluble and extremely soluble), to account for the uncertainty in the reaction rates for CO and O<sub>2</sub> in seawater. Harman *et al.* (2015) also defined an oxygen false positive not as something comparable to Earth's PAL but instead as being equivalent to the lower limit on the abundance of O<sub>2</sub> during the Proterozoic era of Earth's history, namely 10<sup>-3</sup> PAL (Planavsky *et al.*, 2014b), or a mixing ratio of 0.02% O<sub>2</sub>. They argued that this is the lower bound for a biologically produced O<sub>2</sub> concentration, although estimates of the Proterozoic O<sub>2</sub> abundance vary widely (*e.g.*, Kump, 2008; Lyons *et al.*, 2014). Harman *et al.* (2015) found that O<sub>2</sub> derived from CO<sub>2</sub> photolysis was significantly lower than 0.02% on planets around F- and G-type stars, although presumably O<sub>2</sub> concentrations in the atmosphere could be higher if background N<sub>2</sub> concentrations were unusually small and the planets had compromised cold traps (Wordsworth and Pierrehumbert, 2014). For planets orbiting a Sun-like star, the surface O<sub>2</sub> mixing ratio found by Harman *et al.* (2015) never rose above 10<sup>-13</sup>, so they were not able to reproduce the relatively high surface O<sub>2</sub> concentrations (0.2%) obtained by Hu *et al.* (2012), even when using the unrealistic boundary condition of no H<sub>2</sub> outgassing. However, Harman *et al.* (2015) found that K dwarf and especially M dwarf planets may build up more significant amounts of abiotic O<sub>2</sub> because of the higher FUV/NUV flux ratio from their parent stars, similar to the results of Tian *et al.* (2014). This abiotic O<sub>2</sub> would be enhanced if recombination of dissolved CO and O<sub>2</sub> in the oceans is slow

and if other O<sub>2</sub> sinks (*e.g.*, reduced volcanic gases or dissolved ferrous iron) are small. They posited a "worst-case scenario" where for each planet modeled the ocean was saturated with O<sub>2</sub>, implying that both the deposition velocity and the rainout rate for O<sub>2</sub> were zero, and sink reactions could not occur in the ocean. For this scenario, their model generated an atmosphere with 6% O<sub>2</sub> (0.3 PAL) for an abiotic planet orbiting M dwarf GJ 876. When adopting more realistic boundary conditions, and also assuming that CO and O<sub>2</sub> react rapidly in solution, the column abundance of abiotic O<sub>2</sub> was a thousand times less. In keeping with Domagal-Goldman *et al.* (2014) and Hu *et al.* (2012), they also found that O<sub>3</sub> would be built up to potentially detectable levels at UV wavelengths for a much broader range of boundary conditions and stellar types.

Gao *et al.* (2015) also used a 1-D photochemical model to investigate whether CO<sub>2</sub>-dominated atmospheres could remain chemically stable on terrestrial planets orbiting M dwarfs. Their mechanism relies on the planet being desiccated, so that water is not available to catalyze the recombination of CO<sub>2</sub> and the balance between CO<sub>2</sub> photolysis and recombination would instead be shifted toward production of large amounts of atmospheric O<sub>2</sub>. In this case, the O<sub>2</sub> buildup for this habitable zone planet is also due to suppression of the catalytic recombination of CO+O<sub>2</sub> to form CO<sub>2</sub>, but instead of the stellar FUV/NUV ratio driving the suppression of this cycle as proposed in Tian *et al.* (2014), it is inhibited by lack of atmospheric water vapor. We do not see this phenomenon on Earth, as our relatively high abundance of water vapor drives the catalytic recombination of CO<sub>2</sub> after photolysis. However, planets orbiting in the habitable zones of M dwarf stars may be depleted in water due to M dwarfs' prolonged, high-luminosity pre-main sequence phases (*e.g.*, Luger and Barnes, 2015) or may have formed volatile-poor (Lissauer, 2007; Raymond *et al.*, 2007). For these desiccated planets with atmospheric H mixing ratios <1 ppm, Gao *et al.* (2015) found that ~40% of the atmospheric CO<sub>2</sub> was converted to CO and O<sub>2</sub> on a timescale of 1 million years, resulting in an atmosphere dominated by CO<sub>2</sub>, CO, and O<sub>2</sub>. However, the process did not destroy all the CO<sub>2</sub>, as the increased O<sub>2</sub> abundance also led to high O<sub>3</sub> concentrations, and O<sub>3</sub> photolytic products form another CO<sub>2</sub>-regenerating catalytic cycle. This catalytic cycle feedback places an upper limit of ~50% on the CO<sub>2</sub> that can be destroyed via photolysis, but this is still enough to generate Earth-like abundances of abiotic O<sub>2</sub> and O<sub>3</sub>. Specifically, a 90% CO<sub>2</sub> atmosphere with a total hydrogen mixing ratio of a few parts per million equilibrated at a concentration of 50% CO<sub>2</sub>, 30% CO, and 15% O<sub>2</sub> (with the remainder being N<sub>2</sub>), producing a similar oxygen fraction to that seen in Earth's current atmosphere (Gao *et al.*, 2015). If desiccated, CO<sub>2</sub>-rich planets are common in the habitable zones of M dwarfs, then the abiotic conditions that lead to such high oxygen levels could be common also.

**3.2.4. Summary and discussion of abiotic generation of O<sub>2</sub> and O<sub>3</sub>.** In aggregate, this research indicates that there are several mechanisms that could produce abiotic O<sub>2</sub> and O<sub>3</sub> in a planet's atmosphere, with each presenting a potential false positive to different degrees. A summary of this information is presented in Table 1. Two of the mechanisms allow water to enter a planet's stratosphere where it is

TABLE 1. SUMMARY OF KNOWN ABIOTIC O<sub>2</sub> GENERATION MECHANISMS

<i>Mechanism</i>	<i>Action</i>	<i>Targets affected</i>	<i>Potential O<sub>2</sub> produced</i>	<i>Potential O<sub>3</sub> produced</i>	<i>Spectral discriminant</i>	<i>References</i>
O <sub>2</sub> runaway from a superluminous pre-main sequence star	Massive H <sub>2</sub> O evaporation and photolysis during the host star's superluminous pre-main sequence phase	HZ planets orbiting late-type M dwarfs	100s of bar, depending on initial water inventory	Possible, after complete loss of H <sub>2</sub> O	O <sub>4</sub> in transmission (NIR) and direct imaging (visible+NIR)	Luger and Barnes, 2015; Tian, 2015
Lack of noncondensable gases	Lack of cold trap allows water to enter stratosphere and be photolyzed	HZ planets orbiting any stellar type	15% O <sub>2</sub>	Not calculated	Quantification of O <sub>2</sub> and N <sub>2</sub> abundance via the N <sub>2</sub> -N <sub>2</sub> collisional pair at 4.1 μm	Wordsworth and Pierrehumbert, 2014
Desiccated planets	Lack of water inhibits catalytic recombination of CO <sub>2</sub>	HZ planets orbiting late-type M dwarfs, also volatile-poor planets	15% O <sub>2</sub>	0.2 times Earth's for M dwarfs	Absence of H <sub>2</sub> O absorption in direct imaging. O <sub>3</sub> looks similar to Earth's.	Gao <i>et al.</i> , 2015
Photochemical production from CO <sub>2</sub> photolysis	High stellar FUV/NUV, reduction of O <sub>2</sub> sinks	HZ planets orbiting K and M dwarfs	<0.02% for F and G star planets <6% for M dwarf planets with O <sub>2</sub> -saturated oceans	0.15–0.01 times Earth's for M dwarf planets depending on sinks	Presence of CO, CO <sub>2</sub> , M dwarf spectral host	Harman <i>et al.</i> , 2015
Photochemical production on CO <sub>2</sub> -rich, H-poor planets	High stellar FUV/MUV photolysis CO <sub>2</sub> and produces O <sub>3</sub>	HZ planets orbiting F dwarfs and some M dwarfs	40 ppm	0.1 times Earth's for M dwarfs	Presence of CO, absence of H-bearing gases such as CH <sub>4</sub>	Domagal-Goldman <i>et al.</i> , 2014
Photochemical production from CO <sub>2</sub> photolysis and stellar inhibition of recombination	High stellar FUV/NUV destroys HO <sub>x</sub> species and inhibits CO <sub>2</sub> recombination	HZ planets orbiting M dwarfs, CO <sub>2</sub> -rich (<10%) atmospheres	0.2% for M dwarfs with high FUV/NUV ratios	0.06 times Earth's for M dwarfs	Presence of CO, CO <sub>2</sub> , high FUV/NUV ratio for the parent star with low absolute NUV	Tian <i>et al.</i> , 2014
Photochemical production from CO <sub>2</sub> photolysis	CO <sub>2</sub> photolysis and no CO <sub>2</sub> or CH <sub>4</sub> surface flux	1 bar CO <sub>2</sub> -rich (90%) atmospheres orbiting a G2V	0.1% O <sub>2</sub>	0.3 times Earth's for G dwarfs	Presence of CO, CO <sub>2</sub>	Hu <i>et al.</i> , 2012

HZ = habitable zone.

photolyzed, and the H atoms lost to space, resulting in O<sub>2</sub> buildup in the planet's upper atmosphere. Water entering the stratosphere is either enabled by loss of an ocean in a runaway greenhouse process (Luger and Barnes, 2015)—a mechanism that is most effective for late-type (*i.e.*, less massive) M dwarfs—or via lack of noncondensable gases in the planetary atmosphere, which could affect planets orbiting stars of any spectral type (Wordsworth and Pierrehumbert, 2014). The runaway mechanism could produce an O<sub>2</sub>-dominated atmosphere of hundreds of bar, and the lack of noncondensable gases could potentially result in atmospheres that are ~15% O<sub>2</sub>. The other major class of processes that build up abiotic O<sub>2</sub> relies on the photolysis of CO<sub>2</sub> and circumstances that inhibit CO<sub>2</sub> recombination from CO and O<sub>2</sub> (Hu *et al.*, 2012; Tian *et al.*, 2014; Gao *et al.*, 2015; Harman *et al.*, 2015). In the mechanism that could potentially produce the largest signal from CO<sub>2</sub> photolysis, a desiccated, cold, hydrogen-poor atmosphere inhibits the CO<sub>2</sub> recombination reaction due to lack of photolytic generation of the OH catalyst from water (Gao *et al.*, 2015). A catalytic cycle feedback with ozone formation may result in stable atmospheric fractions of O<sub>2</sub> near 15%. It has also been proposed that recombination of photolyzed CO<sub>2</sub> can be slowed by a parent star with a higher FUV ( $\lambda < 200$  nm) to MUV and NUV ( $200 \text{ nm} < \lambda < 440$  nm) ratio when compared to the Sun, as the higher FUV photolyzes CO<sub>2</sub>, but the lower MUV-NUV radiation inhibits the photolysis of water and other HO<sub>x</sub> chemistry that would drive recombination (Tian *et al.*, 2014; Harman *et al.*, 2015). For the cases considered for this mechanism, O<sub>2</sub> abundances as high as 0.2% to 6% are predicted, with higher values corresponding to little or no O<sub>2</sub> sinks in the planetary environment. When more realistic sinks are included, abiotic O<sub>2</sub> abundances are reduced by many orders of magnitude (*e.g.*, Harman *et al.*, 2015). Abiotic O<sub>3</sub> buildup can occur via photolysis of abiotic O<sub>2</sub>, with calculated values ranging from 1% to 30% of Earth's current ozone abundance. However, O<sub>3</sub> could also be photochemically generated as a result of the spectral slope of the UV radiation of the star, even without buildup of O<sub>2</sub>. FUV radiation can favor the generation of ozone via photolysis of CO<sub>2</sub> (and O<sub>2</sub>), whereas MUV or NUV radiation photolytically destroys O<sub>3</sub>. Consequently, abiotic O<sub>3</sub> could accumulate—even without significant generation of O<sub>2</sub>—for stars with the highest FUV to MUV ratios. O<sub>3</sub> was produced at 10% of Earth's current O<sub>3</sub> column abundance in the simulations of Domagal-Goldman *et al.* (2014) for M dwarf planets, without appreciable buildup of abiotic O<sub>2</sub>. Although these mechanisms to generate abiotic O<sub>2</sub> and O<sub>3</sub> are driven primarily by the interaction of the incident stellar spectrum with the planetary atmosphere, they can be balanced by the destruction or sequestration of O<sub>2</sub> and O<sub>3</sub> in the planetary environment. These losses could be via photolysis, catalytic recombination into CO<sub>2</sub>, or interaction with the planetary surface and ocean, if present. The net atmospheric accumulation of O<sub>2</sub> and O<sub>3</sub> is highly sensitive to these boundary conditions, which include weathering rates, and aqueous sinks for CO, all of which are currently poorly understood.

Venus, for example, appears to stand as a sobering counter example for abiotic O<sub>2</sub> production, as its postrunaway, mostly desiccated, high-CO<sub>2</sub> atmosphere is currently undergoing rapid photolysis—yet Venus does not exhibit a

high atmospheric abundance of abiotic O<sub>2</sub> or O<sub>3</sub>. The modern venusian atmosphere is extremely hydrogen poor, with only about 30 ppm of water vapor below its cloud deck (Meadows and Crisp 1996; Arney *et al.*, 2014), and about 2–5 ppm of H<sub>2</sub>O above it (Cottini *et al.*, 2012), with the other principal H-bearing gases, HF and HCl, found at abundances much less than 1 ppm (Krasnopolsky, 2010). Venus may have been wetter in the past but likely underwent a catastrophic runaway greenhouse, and the observed enhancement in its atmospheric D/H ratio suggests that Venus lost a global ocean of at least 3 m in depth (De Bergh *et al.*, 1991). This is a lower limit, and Venus' ocean could have been significantly larger—as the D/H ratio may not record hydrodynamic escape, where heavier molecules are removed along with the lighter hydrogen (Chassefière, 1996b). Taking into account hydrodynamic H and O loss, along with likely surface and mantle sinks for O<sub>2</sub>, Chassefière (1996a) calculated that Venus could only have avoided developing an O<sub>2</sub>-rich atmosphere if it had a global ocean no larger than 0.45 Earth oceans. However, the efficiencies of surface and mantle sequestration processes are very poorly understood for all terrestrial planets (Rosenqvist and Chassefière, 1995). Hamano *et al.* (2013) also suggested that early Venus may have generated a steam atmosphere that slowed planetary cooling and maintained a magma ocean for an extended period after formation; this magma ocean could have also acted as a massive sink for O<sub>2</sub>.

On modern-day Venus, the photolytic dissociation rate of CO<sub>2</sub> is high (McElroy *et al.*, 1973), and the atmosphere is largely desiccated, suggesting that the photochemical production of abiotic oxygen or ozone may be possible. Ozone has been detected on the venusian nightside, although with a column abundance  $1.5 \times 10^{-4}$  times less than seen on Earth (Montmessin *et al.*, 2011). Excited O<sub>2</sub>—produced by the recombination of O atoms released from CO<sub>2</sub> photolysis—is also detected via its airglow on the venusian nightside. The excited, newly formed O<sub>2</sub> relaxes to the ground state, emitting airglow photons near 1.27  $\mu\text{m}$  (Connes *et al.*, 1979; Allen *et al.*, 1992; Crisp *et al.*, 1996). However, ground-state O<sub>2</sub> has never been detected on Venus, and observational upper limits suggest uniform mixing ratios of <2 ppm above 60 km (Trauger and Lunine, 1983; Mills, 1999). This implies that shortly after its formation, the O<sub>2</sub> is efficiently scrubbed from the atmosphere, possibly by a chlorine-containing catalyst that accelerates the recombination of O with CO to reform CO<sub>2</sub> (Yung and DeMore, 1982). This mechanism has been verified in laboratory experiments (Pernice *et al.*, 2004). Venus' O<sub>3</sub> abundance and spatial distribution also appear to support this mechanism (Montmessin *et al.*, 2011), but the intermediate species have not yet been detected in the venusian atmosphere. Another possible mechanism involves heterogeneous chemistry on aerosols or cloud particles (Mills and Allen, 2007). Little is actually known about these catalytic CO<sub>2</sub> recombination processes; consequently, these mechanisms are not generally included in photochemical codes for terrestrial planets, including the models that were used in the photochemical false-positive studies mentioned above (Hu *et al.*, 2012; Domagal-Goldman *et al.*, 2014; Tian *et al.*, 2014; Gao *et al.*, 2015; Harman *et al.*, 2015). Therefore, determining the sinks for O<sub>2</sub> on Venus is a critical piece of the search for life on exoplanets, as it would place additional constraints on which environments can

accumulate O<sub>2</sub> and allow improvements in photochemical models with regard to O<sub>2</sub>/O<sub>3</sub> formation and destruction in terrestrial planet atmospheres. Ultimately, if the mechanisms that drive efficient recombination of CO<sub>2</sub> on Venus also occur on exoplanets, it may make false positives for O<sub>2</sub> more difficult to generate.

#### 4. Implications for the Search for Life beyond Our Solar System

Exploring and understanding the possible mechanisms for abiotic O<sub>2</sub>/O<sub>3</sub> production enable identification of planets that are less susceptible to abiotic generation and allow us to recognize planetary and stellar characteristics that will help discriminate between abiotic and biological sources. In the search for life, detecting a candidate biosignature molecule is necessary but not sufficient. To claim a robust detection, environmental abiotic sources for biosignature molecules must also be identified, sought, and excluded. For the first generation of exoplanet characterization missions, spectroscopic observations will be extremely challenging, and biosignatures—and their abiotically generated false positives—will be difficult to detect. Here, we discuss potentially observable characteristics of the planet and planetary system that may allow us to choose the most promising targets for life detection or help identify abiotic sources for O<sub>2</sub> and O<sub>3</sub>.

##### 4.1. Target selection considerations

Determining whether a promising planet lies within the star's habitable zone is the first-order assessment performed when selecting planetary targets for spectroscopy. However, in addition to denoting a higher probability for liquid water, the habitable zone could also be used to help screen out those targets that are more likely to exhibit false positives. This latter application will be most effective when conservative habitable zone limits are used (*e.g.*, Kopparapu *et al.*, 2013<sup>1</sup>). Significantly more permissive limits for the habitable zone have been advocated by Seager (2013), for example, for a Sun-like star, an inner limit near the orbit of Mercury [as proposed for low-humidity planets by Zsom *et al.* (2013); see also the discussion by Kasting *et al.* (2014) on the feasibility of these limits] and an outer limit exterior to the orbit of Saturn [as discussed by Pierrehumbert and Gaidos (2011), for H<sub>2</sub>-dominated atmospheres]. While these broader limits may include planets with a rare combination of fortuitous characteristics that could support habitability, near their boundaries they are also far more likely to include a higher fraction of planets that could exhibit false positives or false negatives for O<sub>2</sub>. At a more permissive inner edge, while some small fraction of terrestrial planets may be habitable, many will be uninhabitable due to higher levels of stellar radiation enhancing Venus-like runaway and water vapor loss processes (Kane *et al.*, 2014). These planets could exhibit abiotic false positives for O<sub>2</sub> or O<sub>3</sub>, without requiring an extended higher-luminosity pre-main sequence phase (Schaefer *et al.*, 2016). At a more permissive outer edge, far from the star and with the dense, reducing H<sub>2</sub> atmosphere required to warm the surface, false positives are photochemically unlikely, but false negatives for O<sub>2</sub> may be

prevalent, precluding detection of oxygenic photosynthesis, even if present.

While the habitable zone is the first-order means of down-selecting a pool of potentially habitable planetary targets, if there are multiple habitable zone targets, then we will need more comprehensive methods to rank these planets for observation. This ranking could include using observations of the star, planet, and planetary system, along with modeled evolution and environmental constraints to determine the likelihood of both habitability (*e.g.*, Schulze-Makuch *et al.*, 2011; Barnes *et al.*, 2015) and biosignature false-positive generation within the habitable zone. For example, most habitable zone O<sub>2</sub> and O<sub>3</sub> false-positive mechanisms preferentially affect planets orbiting M dwarf stars, which have extended superluminous pre-main sequence phases and UV spectral slopes that are the most favorable for abiotic generation.

Consequently, since these false positives are most likely to affect planets orbiting M dwarfs, they will impact searches for biosignatures using both space-based (Deming and Seager, 2009; Hedelt *et al.*, 2013; Misra *et al.*, 2014b) and ground-based transmission spectroscopy (Snellen *et al.*, 2013; Rodler and López-Morales, 2014), as well as ground-based direct imaging observations, for which Earth-sized planets orbiting M dwarfs are the favored targets (Riaud and Schneider, 2007; Snellen *et al.*, 2015; Crossfield, 2016; Lovis *et al.*, 2016). These smaller, terrestrial planets are easier to detect in transit around lower mass, later-type M dwarfs. This is due to multiple factors [*cf.* Shields *et al.* (2016) for a more detailed discussion], including the more favorable star/planet size ratio, which increases the depth of the transit; the proximity of the habitable zone to the star, which increases the probability of transit; and the concomitant shorter orbital period, which enables more transits to be coadded to increase detectability within a given observing period or telescope lifetime. Later-type M dwarf stars are also favored as likely targets for transmission observations simply because they are the most populous type of star in the local solar neighborhood—with M4V stars (0.2 M<sub>☉</sub>, 0.25 R<sub>☉</sub>) being the most numerous (Henry *et al.*, 2016)—increasing the likelihood that a nearby planet will be found transiting a member of this population. Indeed, the closest known transiting, and likely terrestrial (with planetary radii ≤1.2 R<sub>⊕</sub>), exoplanets orbit GJ 1132, an M3.5V star (Berta-Thompson *et al.*, 2015) and TRAPPIST-1, an M8V star (Gillon *et al.*, 2016, 2017).

4.1.1. Target selection for transmission. The existing studies suggest that false positives for O<sub>2</sub> and O<sub>3</sub> in transit spectroscopy can be mitigated by selecting planets in the habitable zones of earlier-type M dwarfs—although this is clearly at odds with the other target selection criteria described above. To avoid those planets most susceptible to generating large amounts of false O<sub>2</sub>, the work of Luger and Barnes (2015, their Fig. 13) suggests that we should target planets orbiting stars more massive than 0.4 M<sub>☉</sub> (M3V and earlier) and especially favor those planets found toward the outer edge of the habitable zone around these stars. These earlier-type, brighter stars would backlight a planet's atmosphere more strongly and so would produce stronger signals than later-type M dwarfs at a given distance from the observer. However, there may be fewer of these targets available, as these earlier-type M dwarfs are less common, and their habitable zone planets, which orbit farther from their

<sup>1</sup><http://depts.washington.edu/naivpl/sites/default/files/hz.shtml>

star, would be less likely to be found in transit. Clearly, the selection of targets in the habitable zone for transmission spectroscopy will not be straightforward and will need to be made on a case-by-case basis. It will depend on the number of suitable targets available, the brightness of the host star and other observability considerations, the likelihood that the planet is habitable (*e.g.*, Barnes *et al.*, 2015), and the probability that the planet and star combination is susceptible to the abiotic generation of biosignatures such as O<sub>2</sub> and O<sub>3</sub>.

**4.1.2. Target selection for direct imaging.** For direct imaging considerations, target selection to avoid the likelihood of abiotic O<sub>2</sub> or O<sub>3</sub> generation depends primarily on whether the planets are to be imaged with ground-based or space-based telescopes. Future large (30–40 m diameter) ground-based telescopes planned for 2020 and beyond (*e.g.*, Quanz *et al.*, 2015) will have the capability to directly image about 5–10 terrestrial planets orbiting in the habitable zones of nearby bright M dwarfs (Crossfield, 2016). Space-based telescopes for direct imaging—anticipated to launch on the 2020–2040 timeframe—will necessarily have smaller mirror diameters and will find it more challenging to observe M dwarf habitable zone planets, instead focusing primarily on F, G, and K dwarfs (Stark *et al.*, 2014).

For ground-based telescopes, the focus on directly imaging M dwarf habitable zone planets means that these telescopes will encounter similar target selection issues as transmission observations, namely prioritizing earlier-type M dwarfs and avoiding late-type M dwarfs as planet hosts, or at least considering these targets with caution. Although M dwarf habitable zones are much closer to the star—and so have a relatively small angular separation between star and planet—the large diameter of future ground-based telescopes (30–40 m) will provide adequate angular resolution to observe nearby M dwarf systems in the visible and NIR (*e.g.*, Kawahara *et al.*, 2012; Snellen *et al.*, 2015). It may even be possible to observe Proxima Centauri b out to 0.78  $\mu\text{m}$  by upgrading instrumentation on the existing 8.2 m diameter Very Large Telescope (Lovis *et al.*, 2016). The main advantage M dwarf habitable zone planets provide for ground-based direct imaging is that these planets will have the most favorable contrast between planet and star at NIR and IR wavelengths, where the use of adaptive optics to remove the effects of turbulence in Earth's atmosphere has the best performance (Males *et al.*, 2014; Snellen *et al.*, 2015).

Spaceborne direct imaging missions such as the Wide Field Infrared Survey Telescope (WFIRST; Spergel *et al.*, 2015) and mission concepts currently being considered for a High Definition Space Telescope (HDST; Dalcanton *et al.*, 2015) such as the Habitable Exoplanet Imaging Mission (HabEx; Mennesson *et al.*, 2016) and the Large Ultraviolet Optical Infrared Survey Telescope (LUVOIR; Bolcar *et al.*, 2015) will complement transmission spectroscopy and ground-based direct imaging by focusing primarily on planets orbiting F, G, and K dwarfs (Dalcanton *et al.*, 2015), with M dwarfs being less than 10% of their anticipated targets (Stark *et al.*, 2014, 2015). Even the most ambitious space-based direct imaging missions will have smaller mirror diameters than their ground-based counterparts, making it challenging to separate habitable zone planets from their late-type M dwarf (*e.g.*, M3V and later) hosts, even for very nearby targets (*e.g.*, Meadows *et al.*, 2016).

Consequently, the bulk of their survey samples will consist of planets that will be less susceptible to many of the known false-positive mechanisms that affect later-type M dwarfs. However, F dwarf planets may still be affected by high levels of abiotic O<sub>3</sub> generation from CO<sub>2</sub> photolysis (Domagal-Goldman *et al.*, 2014), and planets orbiting F-K dwarfs may also be susceptible to abiotic O<sub>2</sub> generation if their atmospheres have a lower abundance of noncondensable gases (Wordsworth and Pierrehumbert, 2014). The degree to which a planet has an atmosphere rich in CO<sub>2</sub>, or a noncondensable atmospheric component, is dependent primarily on planetary and planetary system properties, including planet formation and evolution, and the delivery of volatiles from the protoplanetary disk. Planetary atmospheres that favor abiotic O<sub>2</sub> or O<sub>3</sub> generation are therefore a fundamental planetary property and largely independent of the characteristics of the host star, which are more readily observable. In the absence of information about the planet's atmosphere, it will be difficult to identify in advance planets susceptible to these mechanisms of abiotic O<sub>2</sub> or O<sub>3</sub> generation. Target selection considerations for space-based direct imaging will therefore focus more practically on the suite of planetary and stellar characteristics that increase the likelihood of planetary habitability. Indications that O<sub>2</sub> or O<sub>3</sub> is being abiotically generated will come from spectroscopic observations of the planetary environment.

#### 4.2. Recognizing abiotic O<sub>2</sub> and O<sub>3</sub>

Consequently, since target selection can mitigate, but not preclude, the possibility of false positives in transit and direct imaging observations, it will be important to also understand which measurements will help directly identify an increased likelihood of abiotic generation of O<sub>2</sub> or O<sub>3</sub>. These measurements can either be made via transit transmission or secondary eclipse, especially for M dwarfs (*e.g.*, Rauer *et al.*, 2011; Arnold *et al.*, 2014), or via direct imaging for all spectral types (*e.g.*, Meadows *et al.*, 2016; Schwieterman *et al.*, 2016a,b). For ground-based telescopes, additional techniques must be used to mitigate the effect of absorption in Earth's atmosphere, which contains similar molecules to those being sought in terrestrial exoplanet atmospheres. These techniques include the use of high-resolution spectroscopy ( $R \sim 100,000$ ) to observe spectral features that shift in wavelength with the radial component of the orbital motion of the exoplanet. This behavior can then be used to separate the planet's absorption from the relatively stationary telluric and stellar lines (Snellen, 2014). This technique can be used for ground-based observations of transmission spectra from exoplanets (*e.g.*, Snellen *et al.*, 2013; Rodler and López-Morales, 2014), although many transits will need to be coadded to obtain enough signal on the planet. With the planet only available at favorable geometries for  $\sim 10$  h per year, the entire observation could take several to 10 years to complete (Crossfield, 2016; Lovis *et al.*, 2016). Recently, several research groups have explored combining ground-based high-angular-resolution direct imaging—including starlight suppression—with high-spectral-resolution spectroscopy to further suppress light from the star and observe molecules in the exoplanet atmosphere (Kawahara *et al.*, 2012; Snellen *et al.*, 2015; Lovis *et al.*, 2016). Brogi *et al.* (2016) have also pioneered a technique that combines low-resolution

spacecraft observations of a given exoplanet target with high-resolution ground-based observations to enhance the information retrieved from both data sets, allowing retrieval of molecular abundances from the spectra.

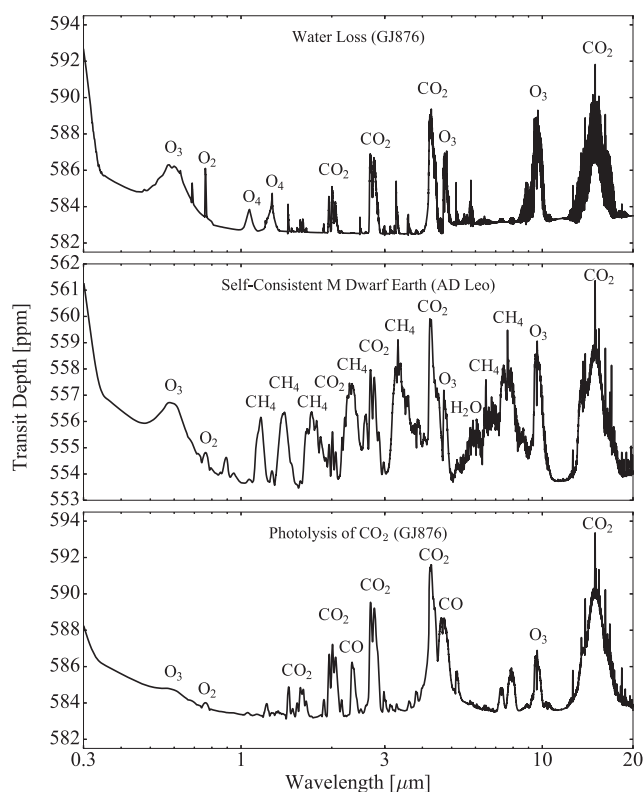
Habitable zone planets orbiting M dwarfs have several intrinsic advantages for characterization via transmission, compared to planets orbiting other stellar types. They are less affected by refraction in the planetary atmosphere, and the favorable star/planet size ratio produces deeper spectral features for atmospheric gases. Refraction can limit how deep into an aerosol-free planetary atmosphere a transmission spectrum can probe, precluding sampling below  $\sim 13$  km in the planetary atmosphere for an Earth-Sun analogue (García Muñoz *et al.*, 2012; Bétrémieux and Kaltenegger, 2013, 2014; Misra *et al.*, 2014b). However, due to a more favorable geometry, refraction is not the limiting factor in probing deeper altitudes for planets orbiting later-type M dwarfs, and accessing the lowest atmospheric layer is theoretically possible (Bétrémieux and Kaltenegger, 2014; Misra *et al.*, 2014b). However, in practice, the wavelength-dependent opacity of a terrestrial planetary atmosphere will limit how deep in the atmosphere a transmission spectrum is able to probe. This opacity will be a function of the atmosphere's composition and mass, and includes the effects of Rayleigh scattering. A minimum altitude of 5 km above the surface near  $1 \mu\text{m}$  has been simulated for modern-Earth-twin planets orbiting M dwarfs of M5V and later (*e.g.*, Bétrémieux and Kaltenegger, 2014). However, simulated transmission spectra for planets with different atmospheric compositions orbiting M dwarfs, including  $\text{CO}_2$ -dominated planets, and photochemically self-consistent Earth-like and early Earth-like planets (with higher  $\text{CH}_4$  and  $\text{CO}_2$  fractions than modern Earth), show minimum altitudes closer to 10 km and above near  $1 \mu\text{m}$  (Meadows *et al.*, 2016; Arney *et al.*, 2017). A 10 km altitude is within the troposphere for an Earth-like planet but well above the majority of tropospheric water vapor, and Earth's clouds. This is important for biosignature detection and false-positive assessment, as the planet's troposphere usually contains a significant fraction of the atmospheric mass (80% in the case of Earth), and the near-surface environment is most likely to record the presence of surface and interior processes such as life, volcanism, and a hydrological cycle.

Direct imaging observations measure reflected or emitted light from the planet, and this method is potentially useful for studies of planets orbiting all stellar types, as long as the star and habitable zone planet can be angularly resolved, either by using a large-enough mirror diameter (*e.g.*, Snellen, 2014; Stark *et al.*, 2015) or by improving how close to the star its light can be suppressed to a sufficient level for the planet to be observed (the "inner working angle," or IWA). The IWA can be improved with various starlight suppression techniques including coronagraphy for ground- and space-based telescopes (*e.g.*, Beuzit *et al.*, 2008; Stapelfeldt *et al.*, 2015; Robinson *et al.*, 2016) or the use of a starshade spacecraft that flies in front of a space telescope to block light from the host star (Cash, 2006; Seager *et al.*, 2015). Direct imaging observations can be used to obtain a disk-averaged photometric or spectroscopic measurement that captures light from the top of the planetary atmosphere down to the emitting or reflecting layer—which could be a haze, cloud, a level of the atmosphere, or the actual plane-

tary surface. Because the path taken by the direct imaging observation through the atmosphere is more direct than the limb-skimming path of a transmission observation, direct imaging observations complement transmission observations because they can see much deeper into a planetary atmosphere, and possibly all the way to the surface—and are less sensitive to the attenuating effects of hazes and other upper atmospheric absorbers (Arney *et al.*, 2016). Below we discuss how these two techniques—transit transmission and direct imaging—could be used to help discriminate between abiotic and biological sources of  $\text{O}_2$ .

4.2.1. Discriminating the effects of enhanced pre-main sequence stellar luminosity. The buildup of thousands of bar of  $\text{O}_2$  due to catastrophic loss of water vapor during an M dwarf's pre-main sequence phase is the strongest false positive for  $\text{O}_2$  currently suggested (Luger and Barnes, 2015; Tian, 2015). Identifying the presence of a massive, and therefore likely abiotic, oxygen atmosphere could be done by detecting absorption from  $\text{O}_2$  collisional pairs ( $\text{O}_4$ ) in the spectrum of the planet for both transmission and direct imaging observations (Schwieterman *et al.*, 2016a, 2016b). In contrast to  $\text{O}_2$  absorption, which is directly proportional to the number density of the gas, the collisionally induced absorption of  $\text{O}_4$  has a density-squared dependence, so it increases in strength—relative to  $\text{O}_2$ —as the number density of the gas increases (*e.g.*, Misra *et al.*, 2014a). In transit transmission, for modest atmospheric pressures (up to 10 bar) and  $\text{O}_2$  fractions ( $<20\%$ ),  $\text{O}_4$  features are potentially detectable—although extremely challenging—near  $1.06$  and  $1.27 \mu\text{m}$  (Misra *et al.*, 2014a). However, for  $\text{O}_2$ -dominated atmospheres of at least a bar, simulated spectra of self-consistent climate-photochemistry models show significantly stronger NIR features at  $1.06$  and  $1.27 \mu\text{m}$  in transmission spectra, but the  $\text{O}_4$  features shortward of  $1.06 \mu\text{m}$  are not detectable due to the masking effects of Rayleigh scattering induced by the long transmission path lengths through these dense atmospheres (Fig. 2a; Schwieterman *et al.*, 2016a). Consequently, a wavelength range of at least  $0.6$ – $1.3 \mu\text{m}$  would be needed to search for  $\text{O}_4$  and thereby identify likely abiotic  $\text{O}_2$  in transmission observations. Interestingly, Schwieterman *et al.* (2016a) showed that for an example 100 bar  $\text{O}_2$  atmosphere (equivalent to the loss of 40% of an Earth ocean) the NIR  $\text{O}_4$  features are stronger and more detectable than the  $\text{O}_2$  features in transmission spectra. In this case, the discriminator of the abiotic source of the  $\text{O}_2$  would be detected in transmission spectra *before* the false positive itself. Once the telltale  $\text{O}_4$  was detected, the decision could be made to either continue the observations or move on to a more promising target.

For direct imaging spectra,  $\text{O}_4$  features in both the visible and NIR could be detected. The  $\text{O}_4$  features can be stronger for direct imaging for the same atmospheric composition when compared to transmission, as the shorter-pathlength, downward-looking observations probe deeper, denser regions of the atmosphere where the density-dependent  $\text{O}_4$  molecules are more likely to form. This shorter pathlength also reduces masking by Rayleigh scattering, and even at 1 bar of  $\text{O}_2$  the corresponding  $\text{O}_4$  features in the visible from  $0.34$  to  $0.65 \mu\text{m}$  become significant (see Fig. 1 in Schwieterman *et al.*, 2016b)—in addition to the NIR features at  $1.06$  and  $1.27 \mu\text{m}$ . Consequently, observations at  $0.4$ – $1.0 \mu\text{m}$  may be all that is needed to search for  $\text{O}_4$  to discriminate



**FIG. 2.** Transmission spectra of three simulated terrestrial planets orbiting M dwarfs. (Top) The transmission spectrum of a 1 bar, 95% O<sub>2</sub> atmosphere generated by loss of water from a pre-main sequence M dwarf (Luger and Barnes, 2015; Schwieterman *et al.*, 2016a,b). Although the abiotic O<sub>2</sub> absorption at 0.76 μm is quite strong, the much stronger collisionally induced O<sub>4</sub> bands in the planet’s spectrum would indicate the presence of a more massive O<sub>2</sub> atmosphere. (Middle) For comparison, the transmission spectrum of a planet with Earth-like surface fluxes, including O<sub>2</sub> fluxes from photosynthesis, that has photochemically and climatically interacted with the spectrum of its parent star. Note the prevalence of not only O<sub>2</sub> and O<sub>3</sub> but strong CH<sub>4</sub> as well. (Bottom) The transmission spectrum of a planet with abiotic O<sub>2</sub> at an abundance of 6%, generated by photolysis of CO<sub>2</sub> (Harman *et al.*, 2015; Schwieterman *et al.*, 2016a,b). The O<sub>2</sub> is weak and unlikely to be detectable; however, both CO and CO<sub>2</sub> are prominent in this spectrum and would serve as beacons to indicate that the O<sub>2</sub> observed may be due to CO<sub>2</sub> photolysis. (Credit: J. Lustig-Yaeger, E. Schwieterman)

abiotic O<sub>2</sub> in direct imaging observations. These visible wavelength O<sub>4</sub> features are well within the wavelength range proposed for NASA direct imaging mission concepts such as WFIRST (Spergel *et al.*, 2015), Exo-C/S (Seager *et al.*, 2015; Stapelfeldt *et al.*, 2015), and the HabEx/LUVOIR concepts in support of a HDST (Dalcanton *et al.*, 2015) and can be taken simultaneously or near-simultaneously with the O<sub>2</sub> measurement itself. However, as discussed, M dwarf habitable zones are challenging to observe with direct imaging, and the first generation of direct imagers (*e.g.*, WFIRST) will struggle to spectrally characterize habitable zone super-Earths. The best targets will also be orbiting hotter F and G dwarf stars, where catastrophic loss of H<sub>2</sub>O in the main sequence habitable zone, and the corresponding O<sub>2</sub> false positive, is less likely. In these cases,

massive O<sub>2</sub> atmospheres may only be seen for close-in planets with significantly higher insolation levels than Earth. It is also important to note that O<sub>4</sub> collisionally induced absorption is a broad continuum absorption, and high-resolution ground-based spectroscopy will find it extremely challenging to separate the planet’s O<sub>4</sub> absorption from that in Earth’s atmosphere.

It is important to stress that obtaining near-term spectra of “hot Earths” or “exo-Venuses”—terrestrial exoplanets that orbit closer to the star than the currently defined habitable zone limits—will support our future interpretation of O<sub>2</sub> as a biosignature by providing a crucial observational test of the evolution of terrestrial planet atmospheres (Schaefer *et al.*, 2016) and the proposed ocean-loss false-positive mechanism (Luger and Barnes, 2015). Hot Earths orbiting M dwarfs would have too high an insolation level to be habitable, and the detection of O<sub>2</sub> or O<sub>4</sub> in these atmospheres would confirm the efficacy and longevity of this potential false positive. Conversely, if more desiccated, CO<sub>2</sub>-rich, O<sub>2</sub>-poor Venus-like atmospheres are detected, especially for older stars, this may indicate that close-in terrestrial planets are able to effectively lose, or internally sequester, O<sub>2</sub> from a lost ocean (*e.g.*, Schaefer *et al.*, 2016), or that these planets form dry. Even in advance of the first direct imagers, JWST transmission spectra or ground-based observations of the atmosphere of transiting “super-Venus” M dwarf planets may prove interesting as potential tests of this abiotic O<sub>2</sub> mechanism, and indeed of the calculated limits of the habitable zone itself. A potential candidate for these observational tests includes GJ 1132b (Berta-Thompson *et al.*, 2015) which receives 19 times Earth’s insolation ( $\sim 3$  times Mercury’s insolation) and orbits an M3.5V, at least 5-billion-year-old star. Similarly, the recently discovered TRAPPIST-1b and TRAPPIST-1c planets (Gillon *et al.*, 2016, 2017), which orbit an M8V dwarf believed to be older than 500 million years, both receive insolation in excess of Venus’, at 4.3 and 2.3 times Earth’s insolation, and would have likely experienced an extended pre-main sequence super-luminous phase from their parent star (Baraffe *et al.*, 1998).

4.2.2. Discriminating the effects of low noncondensable gas inventories. Observations that set limits on a terrestrial exoplanet’s noncondensable gas abundances—and in particular the O<sub>2</sub>/N<sub>2</sub> ratio in the planetary atmosphere—could be used to identify the likely generation of abiotic O<sub>2</sub> via weak cold-trapping of water vapor in atmospheres with small noncondensable gas inventories. Although the O<sub>2</sub> is formed from water photolysis, these atmospheres will not necessarily be depleted in water vapor, so detection of water vapor would not conclusively rule out abiotic O<sub>2</sub> production in this case. Rayleigh scattering could be used to attempt to determine overall atmospheric pressure and therefore deduce an “invisible” component that may be N<sub>2</sub> or another gas that is otherwise not spectrally active in the observed wavelength range. For an Earth-like planet seen in direct imaging, the observed Rayleigh scattering will be influenced by the different atmospheric columns probed over clear sky and cloudy scenes, returning a disk-averaged value for atmospheric pressure that will be systematically lower than the actual surface pressure. Additionally, this technique for pressure determination can be highly inaccurate in the presence of planetwide clouds and hazes that limit the



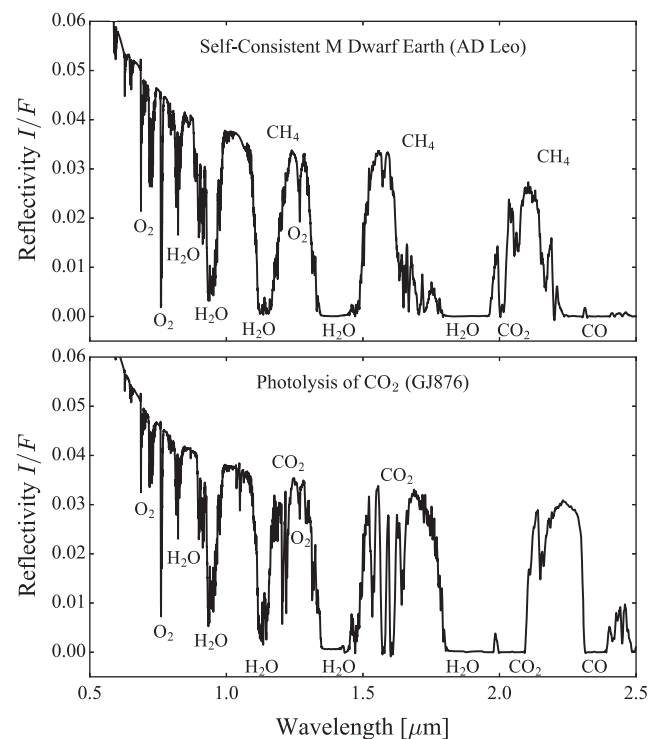
observable atmospheric column. This is the case for Venus, where high sulfuric acid hazes truncate the observable atmosphere at 30 mbar at visible wavelengths. This technique will also have low accuracy if Rayleigh scattering is countered or masked by a blue absorbing component on the planetary surface or in its atmosphere, as would be the case for iron oxides in the martian surface or hydrocarbon hazes on an early Earth (Arney *et al.*, 2016). However, all of these mechanisms tend to reduce the observed Rayleigh scattering effect, so if a strong Rayleigh scattering slope is still observed, it may be a good indicator of higher atmospheric pressure and a subsequently lower likelihood of a lower noncondensable gas fraction. Similarly, pressure-dependent effects, such as the presence of collisionally induced absorption of any kind (Richard *et al.*, 2012), condensates, or broad molecular absorption bands, may also point to more massive atmospheres and could possibly be used to rule out a lower-pressure atmosphere. It has also been suggested that an averaged  $N_2$  column density might be determined from measurements of  $O_2$ - $N_2$  collisionally induced absorption near  $1.27 \mu\text{m}$  along with other  $O_2$  bands (Pallé *et al.*, 2009). However, it may prove challenging to disentangle the effects of the  $O_2$ - $N_2$  absorption from  $O_4$  and  $O_2$  absorption and emission, which also occurs near  $1.27 \mu\text{m}$ .

The most unambiguous discriminant, but potentially the most challenging to observe, would be detection and quantification of pressure-dependent  $O_4$  and  $N_4$  collisionally induced absorption. Determination of the oxygen partial pressure for a terrestrial planet may be achieved by observing  $O_2$  collisional pairs in either transmission—challenging for JWST (Misra *et al.*, 2014a)—or direct imaging, as discussed above. For  $N_2$ , absorption from collisional pairs ( $N_4$ ) could be sought near  $4.1 \mu\text{m}$ . This absorption is broader at high abundance and could help constrain  $N_2$  fraction (Schwieterman *et al.*, 2015b). Consequently, to discriminate this abiotic  $O_2$  generation mechanism with the least ambiguity, observations of  $O_2$  and  $N_2$  collisional pairs, over a wavelength range from  $0.6$  to  $4.5 \mu\text{m}$  for transmission and  $0.4$  to  $4.5 \mu\text{m}$  for direct imaging, would be desirable. These longer wavelengths may be accessible, but challenging, for transmission, although transmission cannot probe to the planet's surface, making the resulting pressures difficult to interpret. For space-based direct imaging telescopes, for which planets orbiting F, G, and K stars will be the majority of the targets, telescope mirror diameter and temperature would likely mean that only a handful of habitable zone planets would be observable at the longer wavelengths needed to detect  $N_4$ . Ground-based high-spectral-resolution observations may also struggle to observe the broad continuum of the  $N_4$  collisionally induced absorption and separate it from Earth's. These challenges suggest that alternative means for discriminating this false positive should also be explored and identified.

**4.2.3. Discriminating the effects of  $CO_2$  photolysis in a desiccated atmosphere.** M dwarf planets may also produce another significant false positive for  $O_2$ —for the high  $CO_2$ , low H inventory atmospheres that do not catalyze  $CO_2$  recombination. These atmospheres may produce stable  $O_2$  abundances of 10–15% for total hydrogen mixing ratios less than 1 ppm (Gao *et al.*, 2015). However, in this mechanism, both  $CO_2$  and CO would constitute large fractions of the

atmosphere and would be more detectable in transit spectra than the abiotic  $O_2$  (Schwieterman *et al.*, 2016a), requiring a wavelength range of  $0.6$ – $2.5 \mu\text{m}$  in transmission to discriminate. For direct imaging,  $CO_2$  and CO may also be present but are likely to be more challenging to observe than in transmission, due to weaker signals in the shorter path-lengths of direct imaging (Fig. 3b; Schwieterman *et al.*, 2016a, 2016b; Wang *et al.*, 2016). However, the extreme desiccation required for this mechanism to work would manifest as little or no water vapor absorption at either reflected visible to NIR or emitted MIR wavelengths (Gao *et al.*, 2015), and strong  $O_2$  without corresponding water absorption bands in direct imaging could be used to discriminate this process (over a wavelength range of only  $0.6$ – $1.0 \mu\text{m}$ ). For ground-based high-resolution spectroscopy, observations in the visible and NIR of  $O_2$ ,  $CO_2$ , CO, and an unsuccessful search for  $H_2O$  in either transmission or direct imaging, may help discriminate this false positive.

**4.2.4. Discriminating the effects of  $CO_2$  photolysis in a habitable atmosphere.** Finally, for  $N_2$ - and  $H_2O$ -rich habitable zone planets, abiotic generation of  $O_2$  or  $O_3$  by photolysis is still possible, but in these cases the models predict



**FIG. 3.** Simulated reflectivity spectra of  $CO_2$  photolysis atmospheres and a self-consistent M dwarf Earth. (Top) The reflectivity spectrum for the planet in Fig. 2 with Earth-like surface fluxes, that has photochemically and climatically interacted with the spectral energy distribution of its parent M3.5V dwarf star. (Bottom) Reflectivity spectrum for the  $CO_2$ -rich planetary atmosphere with 6% abiotic  $O_2$  plotted in transmission in Fig. 2. This comparison shows that the presence of  $CH_4$  for the Earth-like planet and the strong absorption from  $CO_2$  and CO for the planet orbiting GJ 876 can be used to discriminate between the biological and abiotic source of the  $O_2$  seen in these spectra. (Credit J. Lustig-Yaeger, E. Schwieterman)

relatively small amounts, and for the O<sub>2</sub> in particular, these will be extremely challenging to detect with next-generation telescopes. In Domagal-Goldman *et al.* (2014), the highest abiotic O<sub>3</sub> produced, 0.1 PAL, occurred for planets orbiting stars with relatively high FUV fluxes, such as F dwarfs. This abiotic abundance can produce a Hartley O<sub>3</sub> absorption band (0.2–0.3 μm) that is comparable to that of modern Earth. However, the lower, 0.1 PAL O<sub>3</sub> abundance from this abiotic mechanism becomes apparent when looking at other O<sub>3</sub> features in the visible—which are much weaker than Earth’s—and could be used to help to quantify the total O<sub>3</sub> column abundance (Domagal-Goldman *et al.*, 2014). The high-FUV M3.5 dwarf AD Leo also produced a relatively large absorption feature at UV wavelengths but with negligible O<sub>2</sub> or O<sub>3</sub> absorption features elsewhere in the spectrum. In this case then, the false positive is confined to observations at UV wavelengths, which will not be observable for habitable zone exoplanets with JWST, WFIRST, or ground-based extremely large telescopes and must wait for HDST-class telescopes in the 2030+ time frame.

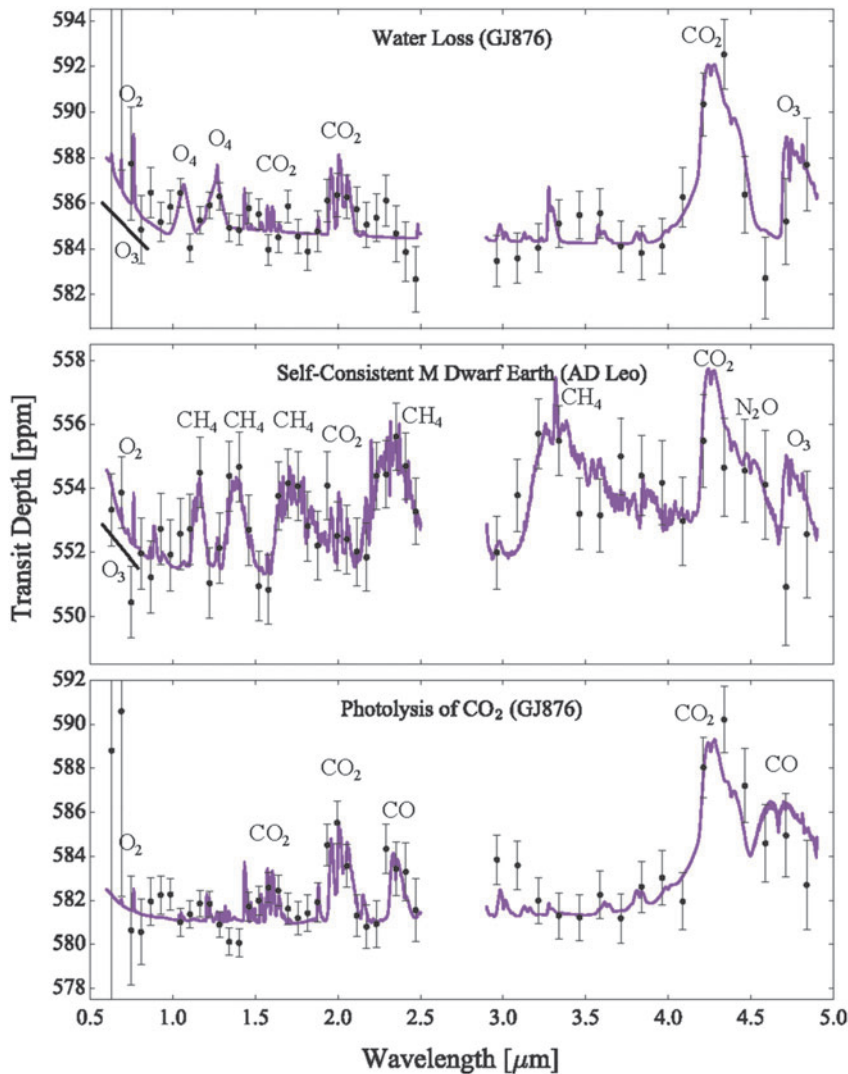
Domagal-Goldman *et al.* (2014) also showed that generally, production of abiotic O<sub>3</sub> by CO<sub>2</sub> photolysis only occurred in atmospheres with high CO<sub>2</sub> concentrations and low CH<sub>4</sub> concentrations. The CO<sub>2</sub>-rich atmospheres required could produce multiple strong CO<sub>2</sub> features between 1 and 2 μm, which could be sought as discriminants. Observation of strong CH<sub>4</sub> could also be used as a proxy for the reducing power of the atmosphere, as high CH<sub>4</sub> abundances make abiotic production of O<sub>3</sub> far less likely. However, quantifying the CH<sub>4</sub> abundance to help determine the redox state of the atmosphere will likely be challenging for direct imaging observations of M dwarf planets in the UV-visible-NIR, as the strongest CH<sub>4</sub> features are in the NIR. Due to the wavelength dependence of the IWA constraint—that angular distance from the star where its light is first suppressed sufficiently to measure the planet—these longer wavelengths may not be accessible for all but the closest M dwarf planetary systems, due to the small angular separation of the M dwarf star and its habitable zone planet. However, Lovis *et al.* (2016) suggested that a search for O<sub>2</sub>, H<sub>2</sub>O, and CH<sub>4</sub> could be performed for Proxima Cen b using coupled high-resolution imaging and spectroscopy on ground-based telescopes over the 0.6–0.78 μm visible range.

In Tian *et al.* (2014) and Harman *et al.* (2015), the models and boundary conditions generated more abiotic O<sub>2</sub> than seen in Domagal-Goldman *et al.* (2014). However, with reasonable sinks included, the amount of O<sub>2</sub> generated was still extremely small when compared to modern-day Earth values. In addition, abiotic O<sub>2</sub> buildup was again most abundant for planets orbiting M dwarfs, which will be difficult to observe with direct imaging. Tian *et al.* (2014) did not generate spectra of their model atmospheres, and their detectability arguments for the 0.2% abiotic O<sub>2</sub> (a hundredth of Earth’s current O<sub>2</sub> abundance) were made by comparing to previously generated synthetic spectra for similar O<sub>2</sub> and O<sub>3</sub> abundances, from Segura *et al.* (2003). Since the abundance of O<sub>3</sub> generated from O<sub>2</sub> is extremely nonlinear (Kasting and Donahue, 1980), a factor of 100 drop in O<sub>2</sub> abundance can result in only a 60% reduction in the ozone column depth for a planet orbiting a Sun-like star (Segura *et al.*, 2003). O<sub>2</sub> absorption, however, does drop in much closer correspondence to the abundance drop, and the

equivalent width of the O<sub>2</sub> 0.76 μm A-band for 0.2% O<sub>2</sub> will be only 1% of that of modern-day Earth’s. This will likely not be detectable by the first or even second generation of terrestrial exoplanet direct imaging missions (WFIRST and HDST-class), even if there is an accessible M dwarf planetary candidate. JWST would be more likely to observe such a planet, and JWST transit spectra will require several hundred hours to observe present-day Earth’s abundance of O<sub>2</sub> in a terrestrial planet atmosphere (Fig. 4; Cowan *et al.*, 2015; Schwieterman *et al.*, 2016a), so observing a signal that is 100 times weaker will be infeasible during the mission lifetime.

Similarly, the O<sub>2</sub> false positive defined by Harman *et al.* (2015) as anything in excess of the Planavsky *et al.* (2014b) estimate of Proterozoic O<sub>2</sub>, namely 10<sup>-3</sup> PAL, is highly unlikely to be detectable by near-future telescopes operating in the visible. The resultant O<sub>3</sub> features in the UV and MIR will be stronger, but in the MIR, detection of O<sub>3</sub> at 9.6 μm can be complicated by abundant atmospheric CO<sub>2</sub>, which produces strong and broad 9.4 μm absorption from CO<sub>2</sub> hot bands (Selsis *et al.*, 2002). Observations at these wavelengths in transmission will also be even more challenging due to the strong falloff of the host star’s blackbody at these longer wavelengths. The highest abiotic O<sub>2</sub> abundance of 6% for a planet orbiting GJ 876 assumed no sinks for O<sub>2</sub> in the planetary atmosphere, and this amount dropped by a factor of 1000 when more realistic boundary conditions were used (Harman *et al.*, 2015). Six percent O<sub>2</sub> may be detectable by future larger-aperture direct imaging telescopes on the ground or in space, for a nearby M dwarf with an accessible habitable zone (*i.e.*, at a larger angular separation than the IWA). However, this amount will still be extremely challenging for JWST to detect. Even with 6% O<sub>2</sub>, the signal is at the 1 ppm level in the simulated JWST transmission spectrum (Fig. 2), which is likely not achievable. Also, as was the case for the desiccated CO<sub>2</sub> atmospheres, in both the Harman *et al.* (2015) and Tian *et al.* (2014) atmospheres, the abundant CO<sub>2</sub>, and CO—the photolytic by-product of abiotic O<sub>2</sub> production—would be more readily detectable than the false O<sub>2</sub> signal itself (Fig. 2), alerting observers to the likely abiotic origin for the O<sub>2</sub> (see Schwieterman *et al.*, 2016a, their Fig. 2). Consequently, it is highly unlikely that the photochemical methods of abiotic O<sub>2</sub>/O<sub>3</sub> generation are going to significantly impact observations made with JWST or HDST (HabEx/LUVOIR). Ground-based observations would also have the option to search for CO as well as O<sub>2</sub> in the spectrum (Snellen, 2014), which could help discriminate this false positive.

However, although JWST is unlikely to detect abiotically generated O<sub>2</sub> from CO<sub>2</sub> photolysis, it is generally true that transmission observations are more sensitive than direct imaging to photochemically generated O<sub>2</sub> or O<sub>3</sub>. Abiotic O<sub>2</sub> generation from the photochemical destruction of CO<sub>2</sub> and H<sub>2</sub>O is most effective in the stratosphere, where UV photons are not attenuated and where geometry and relative atmospheric transparency maximize the transmission signal for trace gases. And while production occurs high in the atmosphere, sinks for O<sub>2</sub> or O<sub>3</sub> from either volcanic outgassing of reduced gases or reactions in seawater are concentrated in the near-surface troposphere, resulting in a profile where abiotically generated O<sub>2</sub> may be less abundant near the surface (see, *e.g.*, Harman *et al.*, 2015, their Fig. 8).



**FIG. 4.** Simulated JWST observations of the three simulated planets from Fig. 2. Synthetic JWST/NIRISS and JWST/NIR-SPEC observations (simulations performed by Drake Deming; Schwieterman *et al.*, 2016a,b) of an O<sub>2</sub>-rich water loss atmosphere (Top), a self-consistent M dwarf Earth (Middle), and the CO<sub>2</sub>-rich atmosphere (Bottom). These results are for 10 transits, corresponding to about 65 h of JWST time for each object and instrument. The observed points are in black, and to improve the S/N the wavelength resolution has been degraded by coadding the 2048 resolution elements of the detector down to 32 columns. Photon-limited noise was used in the simulations, assuming on-orbit instrument performance and exoplanet-tailored observing techniques will ultimately be able to achieve this. The intensity of the star backlighting the planet's atmosphere begins to drop at longer wavelengths, so the photon-limited S/N will decrease, resulting in larger error bars. Note that the features seen here with the assumed photon-limited noise are of the order of 2–8 ppm, which is below the instrument systematic errors assumed prelaunch (Greene *et al.*, 2016). These observations will therefore be extremely challenging and will depend ultimately on on-orbit performance of the JWST science instruments and excellent knowledge of instrument systematics and how to ameliorate them. (Credit: J. Lustig-Yaeger, D. Deming, E. Schwieterman)

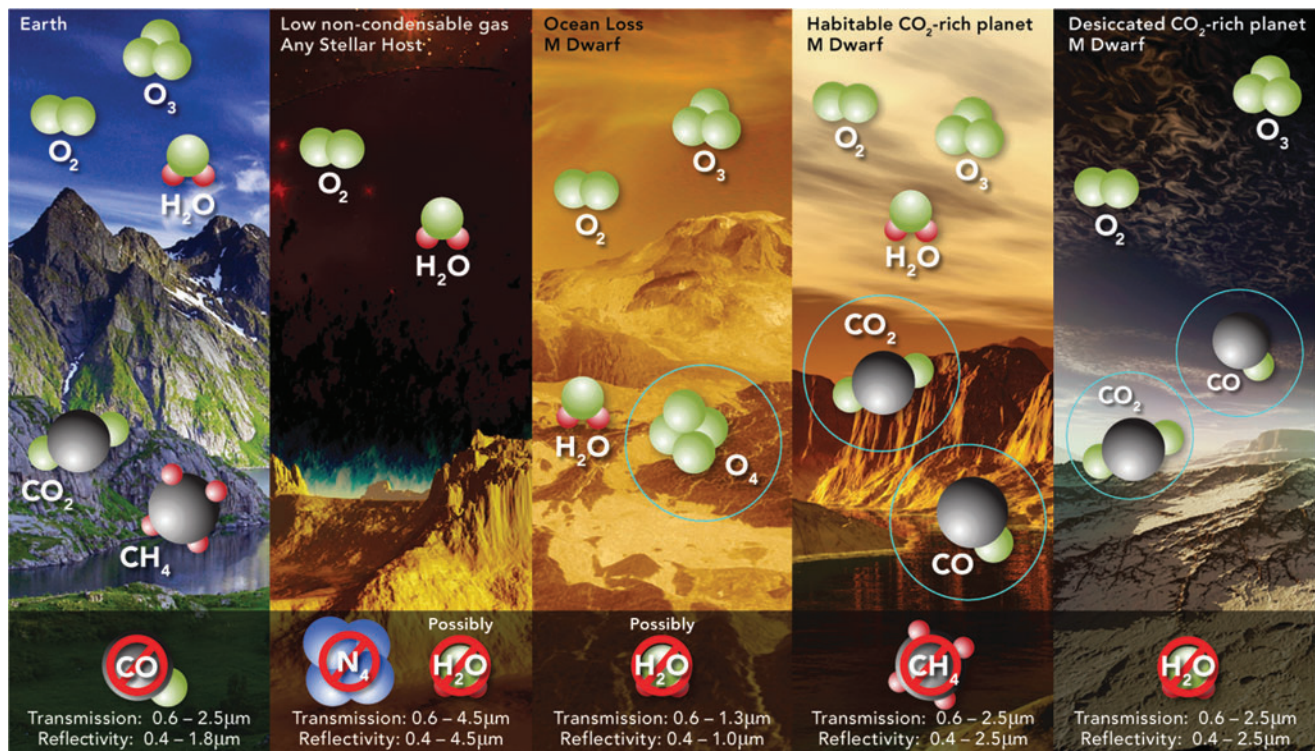
In contrast, biological O<sub>2</sub> fluxes are likely concentrated near the surface, resulting in higher near-surface abundances, or if the O<sub>2</sub> is abundant, an O<sub>2</sub> profile that is evenly mixed throughout the atmospheric column.

Looking to the future, larger-aperture, more sensitive telescopes may be able to measure this altitude-dependent disparity in abiotic O<sub>2</sub> abundance, if they are able to take sensitive, time-dependent observations of a planet as it transits. This may be possible using the effects of refraction on observations of transiting planets, as prior to ingress (and post egress), when the planet is not yet on the stellar disk, refracted light can probe deep into a planet's atmosphere, whereas observations of the planet on the star's disk probe higher in the atmosphere (Misra *et al.*, 2014b). These observations would require relatively short exposures to yield sufficient temporal resolution that different levels of the planetary atmosphere can be probed during the ingress and egress phases. An advanced telescope—well beyond JWST's capabilities—would be needed to make such a measurement. This could be a possible complementary technique for HDST-class or ground-based extremely large telescopes that are built for direct imaging of exoplanets but still have the aperture size to make sensitive transmission measurements, if a suitable target is available.

## 5. Synthesis and Future Work

In summary, recent research has improved our understanding of potential false positives for O<sub>2</sub> and shown that each false-positive mechanism may reveal itself via observable properties of the planet or star. The recently identified abiotic sources of large atmospheric fractions (>6%) of O<sub>2</sub> for habitable zone planets are as follows: early stellar evolution and planetary atmospheric hydrogen loss (Luger and Barnes, 2015; Tian, 2015), lack of noncondensable species (Wordsworth and Pierrehumbert, 2014), photochemical production in atmospheres with no aqueous CO sinks (Harman *et al.*, 2015), and desiccated, CO<sub>2</sub>-dominated atmospheres (Gao *et al.*, 2015). These mechanisms all have additional environmental and spectroscopic features that would help distinguish an abiotic source from a photosynthetic biosphere, whether that be stellar UV spectrum, atmospheric composition, or the presence of liquid water (Fig. 5).

However, these discriminants have varying degrees of observability relative to the false O<sub>2</sub> signal itself. Observation of collisional pairs of O<sub>2</sub> molecules, O<sub>4</sub>, could discriminate O<sub>2</sub> generated by atmospheric loss, and the O<sub>4</sub> molecule may be more detectable than the O<sub>2</sub> molecule in both direct imaging and transit spectra (Misra *et al.*, 2014a;



**FIG. 5.** Discriminating false positives for different environments. For a photosynthetic biosphere (Left) the simultaneous presence of O<sub>2</sub>, O<sub>3</sub>, CH<sub>4</sub>, and the absence of a large abundance of CO points to a photosynthetic origin for Earth's O<sub>2</sub>. In the remaining panels, we show the predominant molecules for various false-positive scenarios, the molecular discriminants whose presence indicates a false-positive mechanism (circled), and the missing molecules (in the bottom region) whose absence indicates a likely false-positive mechanism. The nominal wavelength ranges needed to observe these discriminants or check for their absence are given at the bottom of each panel. From left to right (after Earth), abiotic O<sub>2</sub> from water photolysis due to a low noncondensable gas inventory; a massive O<sub>2</sub> atmosphere from water loss in which H<sub>2</sub>O may or may not still be present; a habitable world—in which a combination of M dwarf stellar spectrum and reduced atmospheric and surface sinks for O<sub>2</sub> has resulted in the buildup of abiotic O<sub>2</sub> from CO<sub>2</sub> photolysis, or the generation of abiotic O<sub>3</sub>; and a desiccated, H-poor environment in which O<sub>2</sub> is a stable photolytic by-product of CO<sub>2</sub>. (Credit R. Hasler, V. Meadows, S. Domagal-Goldman)

Schwieterman *et al.*, 2016a, 2016b). The presence of strong CO and CO<sub>2</sub> in transmission spectra would also point to an abiotic origin from CO<sub>2</sub> photolysis for observed O<sub>2</sub> (Wang *et al.*, 2016) and would also likely be more detectable than the O<sub>2</sub> itself (Schwieterman *et al.*, 2016a,b). Determining the abundance of noncondensable species in a planetary atmosphere will be far more difficult, however, as atmospheric pressure measurements for terrestrial planets are challenging and in the presence of clouds can be difficult to interpret. While searching for N<sub>2</sub> in the planet's spectrum is desirable, at non-UV wavelengths N<sub>2</sub> is only accessible through its collisional pair at 4.1 μm, and this is a challenging observation to make (Schwieterman *et al.*, 2015b). Consequently, more work is needed to find alternative ways to identify a low noncondensable gas atmosphere.

Other proposed mechanisms that rely on the UV spectrum of the star to preferentially build up abiotic O<sub>2</sub> or O<sub>3</sub> have so far not predicted false positives of comparable strength to the modern Earth's photosynthetic biosphere. These mechanisms work predominantly for planets orbiting M dwarf stars, which are more amenable to transmission spectroscopy and which could also be observed using current and future ground-based telescopes with direct imaging and/or high-resolution spectroscopy capabilities (*e.g.*, Snellen *et al.*, 2013; Males *et al.*, 2014; Rodler and López-Morales,

2014; Quanz *et al.*, 2015; Crossfield, 2016; Lovis *et al.*, 2016). For the upcoming JWST mission, these false positives will either be too small to be detected with transmission spectroscopy, or telltale signs, such as abundant CO and CO<sub>2</sub>, will likely be seen in the spectrum of the planet before the false-positive O<sub>2</sub> signal is detected (Schwieterman *et al.*, 2016a). The amount of atmospheric CH<sub>4</sub>, or its absence, will also help identify these false positives (Domagal-Goldman *et al.*, 2014) but will likely be more challenging to detect or set limits on, especially in direct imaging.

Abiotic generation of strong O<sub>3</sub> may also occur for planets orbiting F dwarfs (Domagal-Goldman *et al.*, 2014). This may be of concern to direct imaging missions that observe in the UV only, but if visible measurements of O<sub>3</sub> and O<sub>2</sub> are also attempted, it may be possible to quantify the abiotically generated O<sub>3</sub> and discriminate it from the larger fraction expected from a productive photosynthetic biosphere.

Looking to the future, biosignature science must address two significant questions: "How do we discover new potential biosignatures—especially those with higher probabilities of detection?" and "How do we increase our confidence in the interpretation of the more detectable candidates we do have?" There are two main ways to address the first question, namely by looking for additional disequilibrium or other anomalous or unexpected planetary

processes (*e.g.*, Hitchcock and Lovelock, 1967; Krissansen-Totton *et al.*, 2016) or by systematically studying gases likely to be produced by life to understand their longevity in the planetary environment and whether or not they are likely to be detectable (*e.g.*, Segura *et al.*, 2003, 2005; Domagal-Goldman *et al.*, 2011; Grenfell *et al.*, 2012, 2014; Rugh-eimer *et al.*, 2013, 2015; Seager *et al.*, 2013a, 2013b). The first approach could be very powerful, as it will be largely independent of known metabolisms and will instead focus on identifying life as an anomalous planetary process. However, it has the disadvantage that the environment being studied would need to be observed and understood extremely well so that an anomalous process could be identified.

In the second approach, we must first choose a candidate biosignature gas to study, and there are several means to do this, including exploring Earth's current biosignatures. This initial approach has the advantages that we know these characteristics can be produced by life, that they are observed in the relevant environment, and that the survivability of the gas is already proven. The disadvantage is that these biosignatures are drawn from the environment of one planet and may not represent the diversity of biological processes and planetary environments on other worlds.

To expand the possible dominant metabolisms and habitable environments, while maintaining high levels of plausibility, we can also search the biological and geological records of the early Earth, which for this purpose serves as a suite of habitable, and yet alien, environments. Specifically, the early Earth provides geochemical evidence that different metabolisms originated, and were dominant, in different time periods and environments (*e.g.*, Stüeken *et al.*, 2015b), and we can work to understand their likely biosignatures from constraining the properties of these ancient environments (*e.g.*, Arney *et al.*, 2016) and understanding the organisms that remain today. The disadvantage again is that these biosignatures, although potentially more varied than those for modern Earth, are still "Earth-centric."

Another approach is to remain agnostic to biological source or planetary context and survey a very large array of chemically stable, relatively small molecules that are likely to be volatile in a planetary atmosphere (*e.g.*, Seager and Bains, 2015). The advantage of this approach is that it can initially be non-metabolism-specific and so may consider molecules that are not currently, or in the past, produced by life on Earth. The disadvantage is that the extremely large number of candidate gases produced by chemistry must still be winnowed down to the most likely to be produced by life, to survive in the environment, and to be observed. This selection process to choose the best candidates will still require the application of some level of Earth-centric knowledge. Understanding and identifying false positives for potential biosignature gases identified without biological or environmental context will also be extremely challenging.

Once a candidate gas is chosen as being likely to be produced by life using one of the above methods, it must then pass the gauntlets of survivability and detectability in the planetary environment. Future research in this area must be able to answer the following questions: Is the gas able to avoid the normal sinks in a planetary environment, including destruction by photochemistry, reaction with volcanic gases, reaction with the surface, or being dissolved into an

ocean? Finally, and perhaps most importantly, is the candidate biosignature likely to modify the global environment of an extrasolar planet and so produce a sufficiently large signal to be detectable via transmission, secondary eclipse, phase curves, or direct imaging spectroscopy? Is it spectrally active in accessible wavelength regions, and is it clear of overlap with other common planetary species? An example of the power of this approach in generating a new biosignature was given by Domagal-Goldman *et al.* (2011), where a photochemical model was used to understand the buildup of gases in an anoxic, early-Earth-like atmosphere due to surface fluxes from a putative sulfur-dominated biosphere, irradiated by stars of different spectral type. The resulting biosignature was not, as expected, a biogenic sulfur gas such as dimethyl sulfide, but rather ethane, which was produced as a by-product of the photolytic destruction of the biogenic sulfur gases. Future work should use similar methods to explore the survivability and detectability of biosignature gases in a range of atmospheres and under different stellar illumination.

To answer the second significant question—on increasing our confidence in the interpretation of the biosignatures we do have—the depth to which the environmental context for O<sub>2</sub> has now been studied sets a new standard for the field of biosignatures that should also guide future work. While O<sub>2</sub> clearly has many advantages as a highly detectable biosignature, some may consider it no longer "robust," because we now know of potential abiotic sources. However, the existence of abiotic sources is likely true, or should at least be assumed, for all proposed biosignatures, given the complexity and probable diversity of planetary environments. Understanding possible abiotic generation mechanisms puts us in a much stronger position when planning observations or interpreting exoplanet data. Following the example being developed for O<sub>2</sub>, a thorough study of environmental lifetime, context, potential false positives, and the detectability of both biosignature and false-positive discriminators should be undertaken for all newly proposed biosignature candidates. These studies will enhance the robustness of proposed biosignatures and provide knowledge and observing strategies to increase confidence in biosignature detection.

For the specific example of biogenic O<sub>2</sub>, it is now clear that the robustness of the detection will increase as potential false-positive mechanisms are ruled out by observations of other atmospheric gases or planetary environmental characteristics, and measurements of more of these can be obtained as the accessible wavelength range increases (Table 2). Discriminating the false positive due to water loss and the formation of massive O<sub>2</sub> atmospheres can be done with observations of O<sub>4</sub> absorption in the visible and NIR, requiring wavelength ranges from 0.4 to 1.0 μm for direct imaging and 0.6 to 1.3 μm for transmission. To search for the CH<sub>4</sub>, CO, and CO<sub>2</sub>, which will help discriminate several other possible abiotic mechanisms, observations out to 2.5 μm will be required, whereas observations out to 4.5 μm will be desirable to identify abiotic O<sub>2</sub> generated by water loss from a low inventory of noncondensable gases and to also pick up stronger CO<sub>2</sub> and CO bands to strengthen interpretation of some of the other false-positive mechanisms. If MIR observations are available, such as for ground-based direct imaging of M dwarf planets, then absorption from

TABLE 2. SPECTRAL FEATURES AS A FUNCTION OF WAVELENGTH RANGE THAT COULD BE SOUGHT FOR IDENTIFICATION OF AN OXYGENIC PHOTOSYNTHETIC BIOSPHERE

<i>Molecules/Feature</i>	<i>0.2–1.8 μm</i>	<i>1.8–2.5 μm</i>	<i>2.5–5.0 μm</i>	<i>5.0–20 μm</i>	<i>Notes</i>
O <sub>2</sub>	0.2, 0.69, 0.76, 1.27			—	Biosignature sought—also disequilibrium pair with CH <sub>4</sub> , N <sub>2</sub> O
O <sub>3</sub>	0.2–0.3 (strong) 0.38–0.65			9.6	Biosignature sought
O <sub>4</sub> (O <sub>2</sub> -O <sub>2</sub> CIA) <sup>a</sup>	0.45, 0.48, 0.53, 0.57, 0.63, 1.06, 1.27 (strong)			—	False-positive indicator—dense O <sub>2</sub> from ocean runaway. Schwieterman <i>et al.</i> , 2016a, 2016b
CH <sub>4</sub>	0.79, 0.89, 1.0, 1.1, 1.4, 1.7	2.31 (strong)	3.3 (strong)	7.7	Biosignature sought—disequilibrium pair with O <sub>2</sub> . Indicates presence of O <sub>2</sub> sink. Domagal-Goldman <i>et al.</i> , 2014
CO <sub>2</sub>	1.05, 1.21, 1.6	2.01	4.2 (strong)	9.4, 10.4, 15	False positive indicator, esp. in combination with CO—ongoing CO <sub>2</sub> photolysis. Schwieterman <i>et al.</i> , 2016a; Wang <i>et al.</i> , 2016
CO	1.6	2.35			False-positive indicator, esp. in combination with CO <sub>2</sub> —ongoing CO <sub>2</sub> photolysis. Schwieterman <i>et al.</i> , 2016a; Wang <i>et al.</i> , 2016
N <sub>4</sub> (N <sub>2</sub> -N <sub>2</sub> CIA) <sup>b</sup>			4.1		False-positive discriminant—helps quantify noncondensable gas fraction, disequilibrium biosignature when paired with N <sub>2</sub> /O <sub>2</sub> . Schwieterman <i>et al.</i> , 2015a; Krissansen-Totton <i>et al.</i> , 2016
N <sub>2</sub> O		2.11, 2.25	2.6, 2.67, 2.97, 3.6, 3.9, 4.3, 4.5	7.9, 17.0	Biosignature sought—disequilibrium pair with O <sub>2</sub>
H <sub>2</sub>	0.64–0.66, 0.8–0.85				Possible bulk atmospheric constituent
H <sub>2</sub> O	0.65, 0.72, 0.82, 0.94, 1.12, 1.4	1.85, 2.5		6.3	Habitability indicator, false-positive discriminant—could show ocean loss or presence of catalyst for CO <sub>2</sub> recombination. Tian <i>et al.</i> , 2014; Gao <i>et al.</i> , 2015; Schwieterman <i>et al.</i> , 2016a,b
Ocean glint <sup>c</sup>	0.8–0.9 (optimal)				Habitability indicator, false-positive discriminant, disequilibrium biosignature when paired with O <sub>2</sub> /N <sub>2</sub> . Robinson <i>et al.</i> , 2010; Zügger <i>et al.</i> , 2011; Krissansen-Totton <i>et al.</i> , 2016
Vegetation red edge	0.6 (halophile) <sup>d</sup> 0.7 (photosynthesis—G dwarf) <sup>e</sup>				Biosignature sought. Kiang <i>et al.</i> , 2007b; Arnold, 2008; Schwieterman <i>et al.</i> , 2015a
Seasonal variability	CO <sub>2</sub> (1.6), CH <sub>4</sub> (1.1 and 1.4)			CO <sub>2</sub> (15)	Biosignature sought—seasonal variability in biomass building and metabolic output. Meadows, 2008

All values in the table are given in microns (μm).

<sup>a</sup>Hermans *et al.* (1999); Thalman and Volkamer (2013); Greenblatt *et al.* (1990); Maté *et al.* (1999).

<sup>b</sup>Lafferty *et al.* (1996).

<sup>c</sup>Robinson *et al.* (2010).

<sup>d</sup>Schwieterman *et al.* (2015a).

<sup>e</sup>Gates *et al.* (1965).

All other molecular band wavelengths in Table 2 derived from HITRAN; Rothman *et al.* (2013).

water vapor at 6.3  $\mu\text{m}$ ,  $\text{CH}_4$  at 7.7  $\mu\text{m}$ ,  $\text{O}_3$  at 9.6  $\mu\text{m}$ , and  $\text{CO}_2$  at 15  $\mu\text{m}$  will also greatly help the interpretation of  $\text{O}_2$  or  $\text{O}_3$  as biosignatures (e.g., Selsis *et al.*, 2002).

Finally, the studies to date have shown that potential false positives for life are inextricably tied to the planetary environment and will depend on planetary evolution and properties, the incoming stellar spectrum, and the planetary atmosphere's interaction with it. These studies have therefore also highlighted the importance of understanding the spectral characteristics of the host star and especially its UV spectrum and activity—work that needs to be done in the near-term before we lose UV-capable telescopes like the Hubble Space Telescope. Additional laboratory and field work to understand the origin and evolution of metabolisms on Earth, their output gases, and rates for key reactions that can serve as sources or sinks for abiotic oxygen, such as the rate of the aqueous recombination of CO and O in the oceans, is also needed to inform the search for life on exoplanets. To improve our understanding of potential sinks for abiotic  $\text{O}_2$ , we also need to better understand the photochemical cycles on Venus and incorporate catalytic reactions with chlorine into photochemical models of terrestrial exoplanets. Finally, to more accurately predict false-positive mechanisms and observable features for terrestrial exoplanets, our models of the interaction of the planetary atmosphere with stellar radiation will require improvements in gas molecular opacities, especially in the presence of other broadening gases, improved databases for collisionally induced absorption, and laboratory studies of haze and condensate formation and optical properties (cf. Fortney *et al.*, 2016).

## 6. Conclusions

Phototrophs have developed a metabolism that can capitalize on plentiful sunlight, abundant water, and the carbon dioxide in our atmosphere to dominate our planet's surface biosphere and significantly modify our planet's atmosphere. The atmospheric oxygen that photosynthesizers produce has many advantages as a biosignature for first-generation terrestrial planet characterization missions, including its relative abundance and uniform distribution throughout the atmospheric column. This allows  $\text{O}_2$  to be visible in the presence of planetwide clouds and to be searched for in transmission observations that, due to refraction, may be limited to probing planetary stratospheres for signs of life. However, we now know that abundant oxygen or ozone in a planetary atmosphere could be produced by several abiotic mechanisms, including photolysis and atmospheric escape. These studies have advanced the field of biosignatures by identifying possible false positives and allowing us to explore how to avoid or identify an abiotic source of  $\text{O}_2$ —thereby increasing our confidence in biosignature detection. These mechanisms inform our choices of optimal targets and allow us to design better measurements to guard against false positives. In general, a better understanding of the environmental context of the detection is critical, including properties of the chemistry of the planet itself as well as the radiation from its host star.

Most of these mechanisms affect planets orbiting M dwarf stars and so will be relevant to missions such as JWST and to extremely large telescopes that will observe M dwarf exoplanets in transmission and direct imaging. Target se-

lection of host stars more massive than  $0.4 M_\odot$  (M3V and earlier) and planets within the conservative habitable zone may help avoid planets with abundant abiotic  $\text{O}_2$  due to water loss. Additionally, searching for the signs of strong  $\text{O}_4$  or CO in the planetary spectrum, or the lack of  $\text{H}_2\text{O}$  or  $\text{CH}_4$ , could help discriminate between biological and abiotic sources of  $\text{O}_2$  or  $\text{O}_3$ . Most abiotic mechanisms that produce  $\text{O}_2$  or  $\text{O}_3$  via photochemistry driven by the M dwarf host star spectrum are unlikely to produce signals strong enough to be readily detectable by first-generation exoplanet characterization missions either by transmission or direct imaging spectroscopy. For direct imaging observations, photochemical production of ozone for planets orbiting F dwarf stars, atmospheric loss near the inner edge of the habitable zone, or production of  $\text{O}_2$  via photolysis in planetary atmospheres with low noncondensable gas fractions could produce the strongest false positives. In these cases, observations of  $\text{O}_3$  and  $\text{O}_4$  at visible wavelengths and quantification of  $\text{O}_4$  or  $\text{N}_4$  at NIR wavelengths could help rule out these scenarios.

In advance of telescopic observations, the future of biosignature research will focus on two key areas related to the identification of new biosignatures for study and research to enhance our confidence in biosignature interpretation through the vetting of these biosignatures for environmental lifetime, context, potential false positives, and detectability. Supporting laboratory and field work on biosignature gas sources and sinks, along with observations of the full spectrum of the host star, including the UV portion, will improve our ability to predict planetary photochemistry and discriminate between abiotic and biological sources for atmospheric gases. Near-term observations of “hot Earths,” planets too close to their star to be habitable, may help illuminate terrestrial planet evolution and some of the proposed false-positive mechanisms. The recent detailed study of  $\text{O}_2$  as a biosignature has shown us that to search for life on a distant world it is not enough to just detect  $\text{O}_2$  or  $\text{O}_3$  in the planetary atmosphere—other molecules, including  $\text{O}_4$ , CO,  $\text{CO}_2$ ,  $\text{CH}_4$ ,  $\text{H}_2\text{O}$ , and  $\text{N}_4$  should also be sought, encouraging us to obtain the broadest wavelength range possible when characterizing potentially habitable extrasolar planets. Ultimately we have to not only observe gases in the spectrum of a distant planet but also recognize their significance and context in the planetary environment.

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#### Abbreviations Used

- FUV = far-UV  
 HabEx = Habitable Exoplanet Imaging Mission  
 HDST = High Definition Space Telescope  
 IWA = inner working angle  
 JWST = James Webb Space Telescope  
 LUVOR = Large Ultraviolet Optical Infrared  
 Survey Telescope  
 MIR = mid-IR  
 MUV = mid-UV  
 NIR = near-IR  
 NUV = near-UV  
 PAL = present atmospheric level  
 WFIRST = Wide Field Infrared Survey Telescope