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CrossTalk opposing view: proton transfer in Hv1 utilizes a water wire, and does not require transient protonation of a conserved aspartate in the S1 transmembrane helix

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Genes encoding Hv1 voltage-gated proton (H^+) channels were first identified in 2006 (Ramsey et al. 2006; Sasaki et al. 2006), but the mechanism of proton transfer (PT) is still debated. Two basic hypotheses have emerged: (a) explicit ionization of a highly conserved Asp (D112/D1.51) carboxyl group mediates PT by a proton 'shuttle' conduction (G_{SH}) mechanism (Musset et al. 2011; Dudev et al. 2015), and (b) protein-associated water molecules support PT via an 'aqueous' conductance (G_{AO}) that does not require titration of D1.51 but instead utilizes a water wire for Grotthuss-type PT (Ramsey et al. 2010; Randolph et al. 2016). Here we summarize evidence supporting the latter hypothesis, which is depicted in cartoon form in Fig. 1A.

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 G_{AQ} and G_{SH} share the need for hydrogen bonds; one question is whether the PT pathway is constituted purely by water molecules (G_{AQ}) or whether the D^{1.51} carboxylate is a required component of the pathway (G_{SH}). Early biophysical studies of native voltage-gated H⁺ conductances (G_{vH}) provided important insights into H⁺ channel mechanisms (see DeCoursey, 2003), and the striking similarity to G_{vH} of currents mediated by expressed Hv1 channels (Ramsey *et al.* 2006; Sasaki *et al.* 2006; Musset *et al.* 2008) suggests that they utilize the same PT mechanism. Native G_{vH} is decreased ~1.9-fold by deuterium isotope substitution, exhibits a shallow dependence on pH_I, and is faster than H₃O⁺ diffusion in water (see DeCoursey, 2003). Together, the data suggest that the hydronium ion (H₃O⁺) is not the permeant species and that the PT mechanism in Hv1 is not identical to the water-filled gramicidin A channel (DeCoursey, 2003). Importantly, biophysical observations reported prior to the cloning of Hv1 are compatible with a G_{AQ} mechanism in which waters with restricted mobility mediate PT.

The distances and orientations of hydrogen bond donor and acceptor atoms are critical for PT, raising the possibility that PT via a $G_{\rm SH}$ mechanism could be disrupted by mutations that cause even modest changes in protein structure. In the first direct test of the G_{SH} hypothesis, we found that Hv1 channels containing neutralizing mutations at all ionizable residues in the voltage sensor (VS) domain express robust currents (Ramsey et al. 2010); channel function has now been reported in over 50 different mutant constructs (Ramsey et al. 2010; DeCoursey et al. 2016; Randolph et al. 2016). Some mutations produce dramatic shifts in the voltage at which channels open (V_{THR}) : for example, V_{THR} ranges from -135 mV in D174N to +135 mV in R205A-R208A (Ramsey et al. 2010). Hv1 is therefore highly tolerant of mutationinduced structural perturbations, implying that the PT mechanism is either surprisingly malleable or that the architecture of the PT pathway is nearly identical in all mutants. One interpretation of the data is that Hv1 utilizes a GAQ mechanism, and PT in both WT and mutant channels is supported by resident water molecules in the VS central crevice that can adopt multiple, distinct electronic structures, at least some of which are sufficient for PT (Ramsey et al. 2010).

A report demonstrating that mutation of D112V (D^{1.51}V) abolishes H⁺ current but reintroduction of Asp at V116 (V^{1.55}) is sufficient to restore function (Morgan *et al.* 2013) further demonstrates the inherent plasticity of the PT mechanism in Hv1, and is in good agreement with the G_{AQ} hypothesis.

Current reversal potential (E_{REV}) shifts in solutions of varying anion concentration gradients demonstrate that the normally exquisite selectivity for H⁺ is eroded in D112 (D^{1.51}) mutant channels, and Cl⁻ is permeant (Musset et al. 2011). Because the dehydration energy for Cl- is large, Clpermeance strongly argues that the pathway is well hydrated (Hille, 2001). Cl- (and $CH_3SO_3^-$ or OH^-) permeation is therefore diffusive, and fundamentally different from PT. Although D^{1.51}-mutant channels appear to exhibit a preference for anions, the relative permeance of H+ vs. OH- cannot be discriminated from EREV shifts (Musset et al. 2011). Assuming $P_{\rm H} = 0$, then D^{1.51} mutant channels are $\sim 10^6$ -fold selective for Cl⁻ over H⁺, but if OH⁻ is impermeant ($P_{OH} =$ 0), the mutants are $\sim 10,000$ times selective for H⁺ over Cl⁻ (Musset et al. 2011). We consider it more likely that D1.51 mutant channels are permeable to both H⁺ and OH⁻, and each ion carries a fraction of the total current. R211 (R^{4.53} or R3) mutations also erode ion selectivity in Hv1 evidently without eliminating PT (Berger & Isacoff, 2011). The effects of mutations in Hv1 can therefore experimentally dissociate effects on PT vs. ion selectivity.

We propose a unifying hypothesis (Ramsey *et al.* 2010; Randolph *et al.* 2016) for selectivity and PT in WT and mutant Hv1 channels that is compatible with all the available experimental data. As reported previously, charged side chains function to repel solution anions ($D^{1.51}$) or cations ($R^{4.53}$) from the hydrated VS central crevice,

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Figure 1. Cartoon and atomic models of hydrated central crevice in Hv1

A, cartoon of Hv1 VS domain in resting- (left), intermediate- (centre) and activated-state (right) conformations. Solvent-accessible vestibules extend toward the centre of the membrane from the extracellular (out) and intracellular (in) sides of the protein, creating an hourglass-shaped central crevice (white area) that is bounded by protein (grey area). The S4 helix is represented as a dark grey cylinder, and the positions of conserved Arg gating charge residues are indicated by shaded blue spheres; the shaded teal sphere (left panel) represents the R205H (R1H) mutation and the side chain is shown in stick representation; the shaded green sphere (right panel) represents N214 (N^{4.46}; N4). Water molecules are schematically shown in coloured space-filling representation (oxygen, red; hydrogen, white). At large negative potentials (i.e. < -80 mV) the VS adopts a resting-state conformation (left panel) in which R1 is located midway across the membrane where the electrical field is most highly focused and within the hydrated central crevice; in this location, hydrogen bonds formed between imidazole/imidizolium nitrogen(s) and intra- or extracellular water molecules catalyse PT via the G_{SH} mechanism (dashed orange arrows). At intermediate voltages (i.e. -50 to -30 mV), R1H is moved outward to a position where the R1H side chain is no longer positioned to catalyse transmembrane PT via G_{SH} (centre panel). In resting- and intermediate-state conformations, one or more S4 guanidinium⁺ gating charges block the central crevice, and PT via G_{AO} is prevented. At more depolarized potentials (i.e. > 0 mV), S4 has moved sufficiently far that waters in the central crevice are accessible to intra- and extracellular H⁺, and Grotthuss-type PT (blue arrows) in a network of hydrogen-bonded waters (shaded blue oval schematically represents the Zundel cation, $H_5O_2^+$) and G_{AO} is therefore open (right panel). B, snapshots taken from MD simulations of Hv1 B (Ramsey et al. 2010) activated state (centre and right panels) and Hv1 E (Randolph et al. 2016) R1H resting state (centre and left panels) systems show magnified side views of Hv1 VS domain model structures in isolation and in overlay (MultiSeg STAMP structural alignment, VMD 1.9.3; http://www.ks.uiuc.edu). Helical segments are represented as coloured ribbons (S1, yellow; S2, green; S3, blue; S4, red), except where omitted for clarity (centre and right panels). Side chains of selected residues are shown in stick representation (colour-coded by atom type: carbon, cyan; nitrogen, blue; oxygen, red; hydrogens, white; H atoms are omitted for clarity in left and centre panels), and indicated by labels (red, D112/D^{1.51}; cyan, R205H/R^{4.47}H/R1H; blue, R211/R^{4.53}/R3; green, N214/N^{4.56}/N4; grey, F150/F^{2.50}); residue numbering is as described previously (Randolph et al. 2016). Water molecules in the Hv1 E system are shown in CPK representation coloured by atom type (red, oxygen; white, hydrogen); waters are omitted for clarity in the centre and right panels. Solid cyan lines (right panel) indicate average water density measured during a 50 ns MD simulation and dashed lines and labels indicate distances between the indicated atoms measured at the end the trajectory, illustrating that D^{1.51} and R3 engage in multiple close-range interactions with nearby side chains and water molecules and are unlikely to exclusively form a bidentate pair, as depicted in a previous study (Dudev et al. 2015).

preventing permeation by ions other than H⁺ (Berger & Isacoff, 2011; Musset et al. 2011; Chamberlin et al. 2015). In the absence of contaminating ions, the hydrated crevice mediates a 'pure' H+ current via the water-based G_{AO} PT mechanism. D^{1.51} or R^{4.53} mutations enhance the diffusive leak of solution ions through the central crevice water network and likely disrupt PT. During intervals between solution ion occupancy, resident waters mediate rapid G_{AO} -type PT in both WT and mutant Hv1 channels. Selectivity is determined by the relative rates of PT vs. diffusive ion leak, and is expected to be proportional to the macroscopic permeability ratios. A corollary is that in WT Hv1, solution ions also occasionally leak into the central crevice, and selectivity is therefore not perfect (DeCoursey, 2003). Although there is a paucity of evidence demonstrating permeability to ions other than H+ in WT Hv1, experimental imprecision and variability in whole-cell seal resistance limit our ability to resolve small E_{REV} shifts, and the magnitude of H⁺ selectivity remains unknown.

Indirect support for the idea that the central crevice mediates a finite, non-proton leakage is provided by a widely accepted model of voltage sensor activation. The central crevice forms the permeation pathway for S4 Arg guanidinium group 'gating charges', which behave like tethered permeant ions and prevent solution ions from leaking through the VS 'gating pore' (Ramsey et al. 2010, Randolph et al. 2016). Resident waters in the VS central crevice may be a general feature of VS domains that facilitate gating charge translocation through the otherwise hydrophobic environment formed by conserved aliphatic and aromatic side chains (Tao et al. 2010; Lacroix et al. 2014; Ramsey et al. 2010; Randolph et al. 2016). Molecular dynamics simulations of VS domain model structures agree that waters are an integral feature of the central crevice (Fig. 1B), and differences in hydration and hydrogen bond networks may help explain why Hv1 is evidently unique among known VS domains in mediating activated-state PT (Ramsey et al. 2010; Randolph et al. 2016).

Additional experimental, computational and structural studies of Hv1 and other VS domains are needed to refine atomic and electronic models of gating and PT mechanisms. We lack high resolution openchannel Hv1 structures (Li *et al.* 2014; Takeshita *et al.* 2014) and careful vetting of new models can be painstaking (Randolph

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et al. 2016), so progress toward understanding PT has been limited. Nonetheless, structural validation is a *sine qua non* for reliable interpretation of results from computational studies. For example, the inclusion of only one water/hydronium molecule in a simplified Hv1-like system (Dudev *et al.* 2015) precludes PT in a water wire *a priori*. Furthermore, D^{1.51} interacts with multiple partners in MD simulations that are also absent from the QM study supporting the *G*_{SH} mechanism (Dudev *et al.* 2015).

In summary, we currently lack sufficient evidence to unambiguously determine the mechanism of PT in Hv1. Although the G_{AQ} mechanism (Ramsey *et al.* 2010) is compatible with a wider variety of the available experimental data, additional experimental and computational studies are necessary to discriminate between H⁺ conduction mechanisms in voltage-gated proton channels.

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Additional information

Competing interests

None declared.

Author contributions

Both authors have approved the final version of the manuscript and agree to be accountable for all aspects of the work. All persons designated as authors qualify for authorship, and all those who qualify for authorship are listed.