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Clues to NaCN formation *

G. Quintana-Lacaci¹, J. Cernicharo¹, L. Velilla Prieto¹, M. Agúndez¹, A. Castro-Carrizo², J.P. Fonfría¹, S. Massalkhi¹, and J.R. Pardo¹

¹Instituto de Ciencia de Materiales de Madrid, Sor Juana Inés de la Cruz, 3, Cantoblanco, 28049 Madrid, Spain

²Institut de RadioAstronomie Millimétrique, 300 rue de la Piscine, 38406 Saint Martin dHéres, France

Abstract

Context—ALMA is providing us essential information on where certain molecules form. Observing where these molecules emission arises from, the physical conditions of the gas, and how this relates with the presence of other species allows us to understand the formation of many species, and to significantly improve our knowledge of the chemistry that occurs in the space.

Aims—We studied the molecular distribution of NaCN around IRC +10216, a molecule detected previously, but whose origin is not clear. High angular resolution maps allow us to model the abundance distribution of this molecule and check suggested formation paths.

Methods—We modeled the emission of NaCN assuming local thermal equilibrium (LTE) conditions. These profiles were fitted to azimuthal averaged intensity profiles to obtain an abundance distribution of NaCN.

Results—We found that the presence of NaCN seems compatible with the presence of CN, probably as a result of the photodissociation of HCN, in the inner layers of the ejecta of IRC +10216. However, similar as for CH₃CN, current photochemical models fail to reproduce this CN reservoir. We also found that the abundance peak of NaCN appears at a radius of 3×10^{15} cm, approximately where the abundance of NaCl, suggested to be the parent species, starts to decay. However, the abundance ratio shows that the NaCl abundance is lower than that obtained for NaCN. We expect that the LTE assumption might result in NaCN abundances higher than the real ones. Updated photochemical models, collisional rates, and reaction rates are essential to determine the possible paths of the NaCN formation.

Keywords

astrochemistry; stars: AGB and post-AGB; circumstellar matter; stars: individual (IRC +10216)

g.quintana@icmm.csic.es.

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1 Introduction

During its few cycles of operation, the ALMA interferometer has proved to be an incomparable tool for the study of molecule formation in space. A particular well-suited source for such studies is the C-rich AGB star IRC +10216, one of the closest evolved stars, located at a distance of ~ 123 pc from us (Groenewegen et al. 2012). While this object has been intensively studied, in particular in the field of the astrochemistry (see, e.g., Cernicharo et al. 2000, 2013; Agúndez et al. 2012, etc), essential information on the exact location of the different molecular reservoirs has remained inaccessible for years because of the limited angular resolution of the instruments operating at (sub-)millimeter (mm) wavelengths.

Only recently has the scientific exploitation of ALMA allowed us to understand with unprecedented detail the structure and kinematics of the circumstellar envelope of IRC +10216 (Cernicharo et al. 2015; Decin et al. 2015; Quintana-Lacaci et al. 2016; Guélin et al. in press) as well as the exact regions where the different species are formed (Agúndez et al. 2015, 2017; Velilla Prieto et al. 2015).

The location of the different molecules in the ejecta is crucial to understand the chemical processes that are at work in IRC +10216, in particular, and in circumstellar envelopes in general. In this work we study the distribution of the metal-bearing molecule NaCN around IRC +10216 and explore its probable origin and the limitations of current chemical models. This work highlights once again the deep importance of the synergy between high-spatial resolution observations and laboratory and theoretical work to obtain collisional rates and chemical reaction rate constants.

2 Observations and NaCN distribution

We carried out a λ 3 mm spectral survey of IRC +10216 with ALMA band 3 during Cycle 2, covering the frequency range 84.0-115.5 GHz. In addition, we recently obtained a mosaic with ALMA covering selected frequencies at 2 mm. These observation are described in detail in Appendix A.

We obtained interferometric maps of the NaCN transitions presented in Table 1, both merging ALMA compact configuration data and the on-the-fly (OTF) maps obtained with the IRAM 30m telescope, and the ALMA data in the compact and extended configurations with the OTF single-dish maps for the observations of Cycle 2. This resulted in two sets of maps, one with a typical angular resolution of $4'' \times 3''$, and another with a higher angular resolution of $0'' 8 \times 0'' 7$.

The sensitivity of the individual maps of each transition is not high enough to have a clear view of the distribution of the NaCN emitting gas. The brightness distribution that is visible in the velocity channels suggests a spherical hollow shell-like distribution. In order to increase the signal-to-noise ratio (S/N) and to confirm this brightness distribution, we stacked the emission from the NaCN lines presented in Table. 1. In the case of the low angular resolution maps, this stacking already allowed us to confirm the hollow shell gas distribution suggested by the single maps (see Fig.A.1). However, in the case of the high spatial resolution maps, the stacking alone was not enough. We therefore smoothed the

spectral resolution to 5km s⁻¹ to increase the S/N. This high spatial – low spectral resolution map is presented in Fig. A.2.

In addition, we merged the visibilities of the seven fields for Cycle 4 NaCN data and obtained cleaned maps with an angular 1". These maps presented negative fluxes and a smaller extent than the 3 mm NaCN data, confirming the flux loss. Similar as for the 3 mm data, we reduced the spatial resolution to increase the S/N. In particular, we imposed in the cleaning algorithm the synthezised beam to obtain the same spatial resolution as in the map presented in Fig. A.1. These maps are presented in Figs. A.4 and A.5. The NaCN brightness distribution of the stacked map of the 3 mm lines and from the two NaCN transitions observed at 2 mm show the very same structure, confirming the presence of the hollow shell mentioned above (Fig. 1).

3 NaCN modeling

In order to fit the observed gas distribution, we assumed the density and temperature profiles adopted by Quintana-Lacaci et al. (2016), slightly modified to follow the temperature law deduced by Guélin et al. (in press). Similarly to what was done in that work, the abundance profile was adjusted to the observation using as starting point the abundance profiles derived by Agúndez et al. (2012). We only aimed at fitting the 3 mm data because for these transitions all the flux has been recovered by merging the ALMA visibilities with the short-spacing data.

Since there are no available collisional rates for NaCN, we used a local thermodynamical equilibrium (LTE) multi-shell approach carried out with the MADEX code (Cernicharo 2012) to reproduce the observations. In this approach, we solve the level populations at the different radii, assuming LTE conditions. When this is solved, the synthetic profiles are obtained by solving the ray tracing and convolving with the beam of the telescope.

As for fitting the abundance profile, we assumed that the emission distribution is mainly spherical. Therefore, we obtained the azimuthal averaged emission of the NaCN stacked map for both the high- and the low-resolution maps. We aimed at fitting both emission profiles at the same time, taking into account that the S/N difference of the low-resolution and high-resolution maps is significant. A good fitting of the high S/N emission profile was mandatory, while a reasonable fit of the low S/N profile was enough. These fits are shown in Fig.3.

4 Discussion

The location of the different metal-bearing cyanides can be separated into two main groups. Some of them, such as MgNC (Guélin et al. 1993) and HMgNC (Cabezas et al. 2013), are found to appear far from the star's photosphere, forming a shell-like distribution with a radius of ~ 15''. This structure has been confirmed in MgNC by our present ALMA observations (see Fig.A.3). In contrast, species such as NaCN were found to present compact emission when observed with a spatial resolution of ~3'' (Guélin et al. 1993, 1997). These latter results suggested that the formation of NaCN takes place in chemical equilibrium conditions in the vicinity of the stellar photosphere. However, the maps we

present here reveal a small inner hole with a radius ~ 1.5'' in the brightness distribution of NaCN, suggesting that it is formed in regions where the gas has left the chemical equilibrium regime.

These two-fold distribution of the metal cyanides has been theoretically explained by Petrie (1996) based on the physical characteristics of the parent species that combine with CN. In particular, metal-bearing species such as AlCl, NaCl, or KCl are closed-shell molecules that form in the inner hot regions of the envelope, while other species containing Mg, Fe, or Si are open-shell radicals that could react with other neutral molecules in regions with low temperatures. Reactions involving CN and the former group of species would therefore result in metal cyanides in the innermost regions of the ejecta, while those involving the latter group will result in cyanides forming extended shell-like structures like those cited above. A comparison of the extent of these different cyanides is shown in Fig.2.

In the particular case of NaCN, Petrie (1996) suggested the following pathway:

$$NaCl+CN \rightarrow NaCN+Cl.$$
 (1)

Since NaCl is a closed-shell species that is abundant in the innermost regions of the envelope (Quintana-Lacaci et al. 2016), this reaction could take place when a significant amount of CN is available. The distribution of the NaCl J = 7 - 6 emission, with an energy of the upper level (E_{up}) similar to the NaCN transitions we analyzed here and obtained within the same ALMA Cycle 2 project (see Appendix A and Table 1 for details), showed that its brightness distribution is complementary to that of NaCN, that is, when NaCl emission fades, NaCN emission rises. This is shown in Fig. 4 by comparing the azimuth-averaged emission of the central velocity channel for the different transitions. This seems to support the assumption that NaCl is a parent molecule for the formation of NaCN.

On the other hand, the models suggest that CN is present only near the photosphere and in the outer layers of the CSE as a result of the photodissociation of HCN (Lucas et al. 1995). Recent ALMA CN maps confirm that the CN emission appears at typical radii of ~ 15'' (Agúndez et al. 2017).

Recently, Agúndez et al. (2015) showed that the CH₃CN spatial distribution was unexpectedly located in an inner hollow shell. To study the possible relation of the formation of NaCN with that of CH₃CN, we have compared the brightness distribution of this molecule with that of the stacked NaCN map. While CH₃CN 6_0 - 5_0 has a excitation temperature similar to that of the NaCN transitions we present here, it is blended with the 6_1 - 5_1 . We therefore used the unblended transition 6_3 - 5_3 for the comparison. The two azimuth-averaged maps present very similar distributions, suggesting that the parent molecule responsible for both molecules might be similar. In particular, an injection of CN at a radii of $\sim 1^{"}_{..}5 - 2^{"}$ could explain the distribution of both CH₃CN and NaCN. Such an injection would also affect other species, in particular, HC₃N. However, since the rate constant of reaction (1) is unknown, we cannot estimate the balance between the different reactions involving CN.

Furthermore, Agúndez et al. (2015) showed that the model presented by Agúndez et al. (2010), which takes into account the effect of penetrating UV photons in the innermost layers of the circumstellar envelope (CSE) around IRC +10216 fails to explain the CH₃CN distribution. Therefore, to understand the CN source and reaction distribution, a detailed new model is mandatory. This model will be developed in a forthcoming paper.

The results shown in Fig. 4 clearly suggest that the reaction proposed by Petrie (1996) is plausible and that a certain amount of cyanide is freed by photodissociation of HCN or other CN-bearing species with weaker bounds than NCCN, not detected in space so far, and that it reacts rapidly to form NaCN and CH₃CN.

5 Results

The abundance profile obtained for NaCN is presented in Fig. 5 As suggested by the NaCN emission maps (see Fig. A.1 and A.2), we found that NaCN arises at a radius of 3×10^{15} cm.

To check the abundance relation between NaCN and NaCl, we compared the abundance profile obtained here with that derived by Quintana-Lacaci et al. (2016) for NaCl (dashed line in Fig. 5). This comparison shows that the tentative precursor has a lower abundance than the resulting species. This might indicate that reaction (1) is not the main formation path of NaCN. However, we have to keep in mind that while the NaCl abundance profile was accurately derived by solving the level population in non-LTE conditions, that of NaCN has been derived assuming LTE.

At the regions where NaCN abundance rises $(3 \times 10^{15} \text{ cm}) T_{\text{K}} \sim 160 \text{ K}$ (Guélin et al. in press). This means that because LTE assumes $T_{\text{ex}} = T_{\text{K}}$, the high-excitation lines are favored over low-excitation lines such as we studied here. The regions where, in LTE, these low-excitation transitions are expected to dominate lie at radii ~ 2.5×10^{16} cm. However, as shown by Agúndez et al. (2012), metal-bearing species studied by these authors leave the LTE regime at the regions where NaCN emission arises. Furthermore, these authors showed that the line intensity ratios from the NaCN lines observed are not compatible with an LTE regime. Therefore, we might expect non-LTE modeling to derive lower values of T_{ex} and therefore higher level populations, higher intensities, and lower abundances for the transitions observed. This lower abundance would then conciliate the NaCl and NaCN abundance ratio confirming reaction (1) as the main formation path for NaCN.

Another factor that might affect the estimate of the NaCN abundance is the effect of the IR pumping on the NaCN excitation. This effect has been found to be important for other species such as NaCl (Quintana-Lacaci et al. 2016). However, there is no information available in the literature about the IR ro-vibrational spectrum of NaCN that might help to estimate the effect of the IR pumping.

6 Conclusions

We have obtained interferometric maps with high and intermediate angular resolution of the metal-bearing molecule NaCN. As shown by Guélin et al. (1997) and Petrie (1996), this molecule emission arises in the inner regions of the envelope of IRC +10216. New maps

have shown that this emission presents an inner hole that has previously not been detected. Furthermore, emission from NaCl, as well as its abundance, suggest that when NaCl declines, NaCN rises. However, two problems prevent us from confirming reaction (1) as the main formation path of NaCN.

First, the origin of CN is not clear. Agúndez et al. (2015) did not succeed to model the abundance of this species taking into account the penetration of UV photons into inner layers as a source of CN. A new photochemical model is required to simultaneously explain the source of the CN and its impact on CH₃CN, NaCN, and HC₃N formation. Obtaining a reaction rate for (1) is essential to solve the competition for CN for the different chemical paths.

Second, the derived abundance of NaCN seems to be higher than that of NaCl. This might be a sign of different parent species, or, more probably, an artifact derived from the LTE assumption. Obtaining collisional rates for NaCN would allow us to solve this problem.

Appendix

Refer to Web version on PubMed Central for supplementary material.

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Fig. 1.

Comparison of the central channel of the low spatial resolution 3mm stacked map with maps of the ALMA Cycle 4 observations at 2 mm. The lowest contour corresponds to 10% of the peak flux, and the rest of contours are equally spaced in jumps of 10% with respect to the first contour. For absolute intensities see Figs.A.1,A.4, and A.5.



Fig. 2.

Comparison of the extent of emission between NaCN (red contours, see Fig A.1) and MgNC (color scale, see Fig.A.3).



Fig. 3.

Top: Model fitting (red line) of the azimuthally averaged emission of the stacked NaCN lines presented in Table 1 for the low spatial resolution maps (Fig. A.1). *Bottom:* Model fitting (red line) of the azimuthally averaged emission of the stacked NaCN lines presented in Table 1 for the high spatial resolution maps (Fig. A.2).

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Fig. 4.

Comparison of the azimuthally averaged intensity profiles of the NaCN maps (black line), the NaCl 7-6 emission map (red line), and the CH₃CN $6_3 - 5_3$ (blue line).



Fig. 5.

Solid line: NaCN abundance profile obtained assuming LTE conditions. *Dashed line:* NaCl abundance profile obtained by Quintana-Lacaci et al. (2016).

Table 1

Parameters of the observed lines.

					Low spatial resolution		High spatial resolution	
Molecule	Trans.	Freq (MHz)	E _{up} (K)	S _{ij}	Beam	P.A. (°)	Beam	P.A. (°)
NaCN	6 _{1,6} -5 _{1,5}	90394.38	17.6	5.83237	$3''_{.9} \times 2''_{.7}$	89.9	$0''_{.}8 \times 0''_{.}6$	39.2
NaCN	6 _{0,6} -5 _{0,5}	93206.09	15.7	5.99388	4″3×2″7	72.1	$0''_{.}6 \times 0''_{.}6$	15.2
NaCN	6 _{2,4} -5 _{2,3}	94334.80	25.4	5.33293	4″3×2″6	72.4	$0''_{.}6 \times 0''_{.}6$	40.4
NaCN	6 _{1,5} -5 _{1,4}	96959.81	18.7	5.83223	3."9 × 2."6	92.7	$0''_{.}8 \times 0''_{.}7$	20.0
NaCN	9 _{1,8} -8 _{1,7}	145075.57	37.3	8.88452	$1''_{\cdot}4 \times 1''_{\cdot}1^*$	52.4		
NaCN	10 _{2,8} –9 _{2,7}	158616.77	51.2	9.59903	$1''_{.}3 \times 1''_{.}0^{*}$	65.1		
NaCl	7 – 6	91169.88261	17.5	7.0	_	-	$0''_{.}8 \times 0''_{.}6$	38.8
CH ₃ CN	$6_3 - 5_3$	110364.35314	82.8	9.0	—	_	$0''_{.}8 \times 0''_{.}7$	39.1
CH ₃ CN	$6_0 - 5_0$	110383.49871	18.5	6.0	-	-	$0''_{.}8 \times 0''_{.}7$	38.9

High spatial resolution: short-spacing and ALMA compact data merged. Low spatial resolution: short-spacing and ALMA compact and extended data merged.

* only ALMA visibilities.