

REPLY TO BREUER ET AL.:

# Molecular dynamics simulations do not provide functionally relevant values of redox potential in MtrF

Hiroshi C. Watanabe<sup>a,b</sup>, Yuki Yamashita<sup>a</sup>, and Hiroshi Ishikita<sup>a,b,1</sup>

We previously reported that Asp631 decreased the redox potential ( $E_m$ ) for heme 9 by 136 mV [1,362 mV according to Breuer et al. (1)] in native MtrF, whereas Asp631 protonation (corresponding to Asn631 mutation) did not significantly affect  $E_m$  for heme 9, because deprotonation of other residues compensated for the  $E_m$  shift (2). Breuer et al. (3) cannot observe the corresponding effect, since they fixed the protonation states of titratable residues (1).

According to Breuer et al. (1), residues make unusually large contributions to  $E_m$ ; for example, Asp228 decreases  $E_m$  for heme 2 by  $-2,280$  mV [ $-61$  mV in our calculations (2)]. In their letter, Breuer et al. (3) state that “it is only meaningful to compare the sign but not the magnitude of the single-residue contributions”; however, in their original report, they focus on not only the sign but also the magnitude of the single-residue contributions, stating, for example, “we find that every charged residue in the environment of a cofactor contributes several tenths of volts” (1). The contribution of each residue to  $E_m$  must be comparable to  $E_m$  shifts experimentally measured upon mutations.

To verify  $E_m$  obtained using their molecular dynamic (MD) simulation-based thermodynamic integration (TI) approach, we employed the slow-growth TI approach (4), as widely used for free energy calculations (e.g., ref. 5), wherein the redox states gradually transit from  $\text{Fe}^{3+}$  to  $\text{Fe}^{2+}$  (2). Breuer et al. (3) argue against the time scale of our TI simulations (10 ns), but they justify the time scale of their TI simulations (11 ns) assuming the structural “stiffness of the decaheme motif” (however, see below). We emphasize that any MD-based approaches are not appropriate in this case, irrespective of simulation time due to the following reasons.

For the structural fluctuations (Fig. 1), we have pointed out that the side-chain orientations are incorrect in the  $\beta$ -barrel motif of domain I of the original MtrF crystal structure (e.g., hydrophobic residues are

oriented toward the bulk solvent), resulting in remarkably high calculated B-factors (2). Nevertheless, Breuer et al. (1) have used the original side-chain orientations in the MtrF crystal structure. Heme 2 is closest to domain I (heme 7 to domain III). Notably, only  $E_m$  values of the heme 2 and heme 7 pair differ significantly (130 mV) in their perfectly symmetrical  $E_m$  profile (1). Moreover, even heme-binding domain IV is also unstable (Fig. 1). Since their simulations were still in the slow-decay process of domains I and IV, being far from equilibrium,  $E_m$  profiles strongly depend on the MD-starting structure (e.g., obtained after equilibration for 0, 100, and 1,000 ns) as demonstrated in our test MD calculations (2). From the statement, “the considerable stiffness of the decaheme motif” in Breuer et al.’s letter (3), they have missed this point while calculating  $E_m$  (1).

Finally, the  $E_m$  for hemes 2 ( $-57$  mV) and 7 (74 mV) reported by Breuer et al. (1) seem unlikely to support a role of bound flavin [ $-150$  mV (6)] serving as the terminal electron acceptor (7, 8). Their MD simulation-based TI approach, using their geometry and fixing the protonation states of titratable residues and heme-propionic groups, is unlikely to provide functionally relevant values of  $E_m$  in MtrF.

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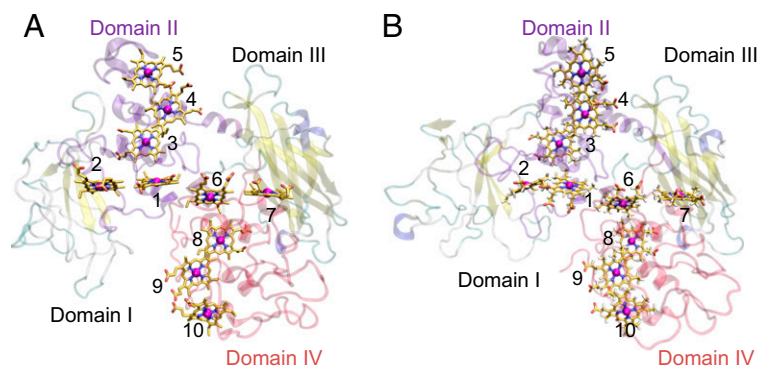
<sup>a</sup>Department of Applied Chemistry, Graduate School of Engineering, The University of Tokyo, Tokyo 113-8654, Japan; and <sup>b</sup>Research Center for Advanced Science and Technology, The University of Tokyo, Tokyo 153-8904, Japan

Author contributions: H.I. designed research; H.C.W., Y.Y., and H.I. performed research; H.C.W., Y.Y., and H.I. analyzed data; and H.C.W. and H.I. wrote the paper.

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<sup>1</sup>To whom correspondence should be addressed. Email: hiro@appchem.t.u-tokyo.ac.jp.



**Fig. 1. (A)** Crystal structure of MtrF (9). Hemes 2 and 7 have been proposed to be located near the flavin-binding site (9, 10). **(B)** MtrF structure obtained after equilibration for 1.2  $\mu$ s. Domains I and IV show remarkably large structural deviations from the original MtrF crystal structure. For comparison with their calculation, we also used the original MtrF crystal structure in MD-based TI calculations.

- 1 Breuer M, Zarzycki P, Blumberger J, Rosso KM (2012) Thermodynamics of electron flow in the bacterial deca-heme cytochrome MtrF. *J Am Chem Soc* 134:9868–9871.
- 2 Watanabe HC, Yamashita Y, Ishikita H (2017) Electron transfer pathways in a multiheme cytochrome MtrF. *Proc Natl Acad Sci USA* 114:2916–2921.
- 3 Breuer M, Rosso KM, Blumberger J (2017) Redox potentials in the decaheme cytochrome MtrF: Poisson–Boltzmann vs. molecular dynamics simulations. *Proc Natl Acad Sci USA* 114:E10028.
- 4 Postma JPM, Berendsen HJC, Haak JR (1982) Thermodynamics of cavity formation in water. A molecular dynamics study. *Faraday Symp Chem Soc* 17:55–67.
- 5 Bash PA, Singh UC, Langridge R, Kollman PA (1987) Free energy calculations by computer simulation. *Science* 236:564–568.
- 6 Okamoto A, Hashimoto K, Nealon KH, Nakamura R (2013) Rate enhancement of bacterial extracellular electron transport involves bound flavin semiquinones. *Proc Natl Acad Sci USA* 110:7856–7861.
- 7 Edwards MJ, et al. (2015) Redox linked flavin sites in extracellular decaheme proteins involved in microbe–mineral electron transfer. *Sci Rep* 5:11677.
- 8 Xu S, Jangir Y, El-Naggar MY (2016) Disentangling the roles of free and cytochrome-bound flavins in extracellular electron transport from *Shewanella oneidensis* MR-1. *Electrochim Acta* 198:49–55.
- 9 Clarke TA, et al. (2011) Structure of a bacterial cell surface decaheme electron conduit. *Proc Natl Acad Sci USA* 108:9384–9389.
- 10 Brutinel ED, Gralnick JA (2012) Shuttling happens: Soluble flavin mediators of extracellular electron transfer in *Shewanella*. *Appl Microbiol Biotechnol* 93:41–48.