

Allosteric histidine switch for regulation of intracellular zinc(II) fluctuation

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Metalloregulators allosterically control transcriptional activity through metal binding-induced reorganization of ligand residues and/or hydrogen bonding networks, while the coordination atoms on the same ligand residues remain seldom changed. Here we show that the MarR-type zinc transcriptional regulator ZitR switches one of its histidine nitrogen atoms for zinc coordination during the allosteric control of DNA binding. The Zn(II)-coordination nitrogen on histidine 42 within ZitR's high-affinity zinc site (site 1) switches from Ne2 to Nδ1 upon Zn(II) binding to its low-affinity zinc site (site 2), which facilitates ZitR's conversion from the nonoptimal to the optimal DNA-binding conformation. This histidine switch-mediated cooperation between site 1 and site 2 enables ZitR to adjust its DNA-binding affinity in response to a broad range of zinc fluctuation, which may allow the fine tuning of transcriptional regulation.

histidine switch | zinc homeostasis | allostery | transcription factors | X-ray crystallography

The zinc transcriptional regulator ZitR in *Lactococcus lactis*, along with its close homolog AdcR in *Streptococcus pneumo*niae, are among the first metal-responsive members of the widely distributed multiple antibiotic resistance regulator (MarR) family of transcriptional factors (1–5). Most MarR family members serve as transcriptional repressors that dissociate from their promoter DNA upon the stimulation from cognate effector molecules, thus causing the derepression of their downstream genes (6–10). In contrast, ZitR and AdcR have been shown to undergo a unique corepression mechanism in which excess Zn(II) ions significantly increase their DNA-binding affinity, thus leading to the repression of the downstream zinc-responsive genes such as zitSQP for ZitR and *adcCBA*, *phtABDE*, and *adcAII* for AdcR (5, 11-13). Both ZitR and AdcR contain two closely located Zn(II)-coordination sites per monomer, with site 1 possessing a Zn(II)-binding affinity $(K_d \sim 10^{-13}$ M) that is four orders of magnitude higher than site 2 $(K_d \sim 10⁻⁹ M)(14, 15)$. Recent studies on AdcR indicate that site 1 is the primary zinc regulatory site, while site 2's functional roles remain elusive $(14, 15)$. Herein we show that $Zn(II)$ binding in site 2 triggers the histidine 42 residue (H42) in site 1 to switch its Zn(II) coordination atom from Ne2 to Nδ1, which acts as a key allosteric regulation mechanism to convert ZitR from the non–DNA-binding conformation to the DNA-binding conformation that is optimal for transcriptional corepression. This histidine switch-mediated site-1 and site-2 cooperation represents an example for coupling the metal ion coordination with allosteric control of metalloregulator conformation and DNA-binding affinity.

Results

Di-Zn(II)-ZitR^{WT} Adopts an Optimal DNA-Binding Conformation. MarR family transcriptional regulators usually undertake conformational changes upon interaction with substrates (e.g., small molecules) that will affect their DNA binding (6–10). We speculate that a similar mechanism may be employed by ZitR and we adopted a structural biology approach to elucidate the mechanism underlying

ZitR-mediated transcriptional regulation. We started by obtaining the crystal structures of di-Zn(II)-ZitR^{WT} (wild-type ZitR with both zinc sites occupied) in the presence and absence of DNA. The structure of di- $Zn(II)$ -Zit R^{WT} was solved by molecular replacement and refined to 2.4-Å resolution (Fig. 1A and SI Appendix, Fig. S1A [and Tables S1 and S2](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)), which revealed a symmetrical dimer with each monomer, consisting of six α-helices, two antiparallel β-strands, one long loop (loop 1) connecting α1 and α2 (residues $20-37$, as well as two coordinated $Zn(II)$ ions in site 1 and site 2, respectively (Fig. 1A and *[SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)*, Fig. S1 A and *B*). This structure is highly similar to the previously reported triangleshaped di-Zn(II)-Adc R^{WT} structure (15). For example, ZitR's dimerization domain (α 1 and α 6) is connected by the winged helixturn-helix (wHTH) DNA-binding domain (α2, α3, α4, β1, and β2) through loop 1 and helix α 5 ([SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. S1 C and D). The $Zn(II)$ ion in site 1 is coordinated by four residues (E24 Oe1, H42 Nδ1, H108 Ne2, and H112 Ne2) while the Zn(II) ion in site 2 is coordinated by residues C30 Sγ, E41 Oe1, E107 Oe1, and a water molecule (14) ([SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. S1E). The DNA-binding helices α4 and α4' (Cα atoms of A71 and A71') are separated by 30.1 Å in di-Zn(II)-ZitR^{WT} (Fig. 1B and [SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. S1D and [Table S2\)](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), representing a suitable conformation ready for DNA binding. To confirm that what we observed is indeed the optimal

Significance

Metal homeostasis is critical to numerous biological processes, and metalloregulators play key roles in its regulation. In transcriptional regulation, which is allosterically controlled by metalloregulators, reorganization of their metal-binding residues and/ or related hydrogen bonding networks is usually utilized, while the coordination atoms on the same metal-binding residues remain seldom changed. Our study shows an example whereby the zinc-induced transcriptional regulator ZitR switches one of its histidine nitrogen atoms for zinc coordination in response to zinc fluctuation. This histidine-switch process facilitates conformational change of ZitR protein, allowing allosteric and fine-tuned control of DNA binding and transcriptional regulation.

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Data deposition: The atomic coordinates and structure factors have been deposited in the
Protein Data Bank, www.wwpdb.org (PDB ID codes [5YHX](http://www.rcsb.org/pdb/explore/explore.do?structureId=5YHX) for ZitR^{wr}, 5Y12 for ZitR^{WT}-DNA,
<mark>[5YHY](http://www.rcsb.org/pdb/explore/explore.do?structureId=5YHY) for ZitR^{C305}, 5Y13 for ZitR^{C305}-DN</mark> for apo-ZitR^{C30AH42A}, respectively).

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Fig. 1. Zn(II) coordination in site 2 alters ZitR's conformation. (A) Superposition of crystal structures of di-Zn(II)-ZitRWT protein in the presence (purple) and absence (green) of DNA. Closeup view of the dual Zn(II)-binding pocket is on the Right. Zn(II) ion at site 1 is coordinated by residues E24, H42, H108, and H112, while site 2 contains residues C30, E41, E107, and a water molecule for Zn(II) binding. Residues E24 from site 1 and C30 from site 2 reside on loop 1, while the rest of the residues reside on helix α 2 (E41 and H42) and helix α5 (E107, H108, and H112), respectively. (B) Structural superposition of dimeric di-Zn(II)-ZitR^{WT} (green) and Zn(II)^{site-1}-ZitR^{C30S} (yellow) proteins. The change of distance and rotation angles on helices α4s, the tip
of the wings as well as the whole wHTH domains between Zn(II)^{site-1}-ZitR^{C30S} and di-Zn(II)-ZitR^{WT} are indicated by arrows.

DNA-binding conformation, the structure of di- $Zn(II)$ - $ZitR^{WT}$ in complex with its operator DNA was solved by molecular replacement and refined to 2.6-Å resolution (*[SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), SI Text*, Fig. [S2, and Tables S1 and S2](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)). The overall structure of di-Zn(II)- $ZitR^{WT}$ including the two zinc-binding sites are highly similar with and without DNA (rmsd = 0.85 Å for 275 C α atoms) (Fig. 1A and [SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. $S2B$), indicating that di-Zn(II)-ZitR^{WT} prelocks an optimal DNA-binding conformation even in the absence of DNA.

Lack of Zn(II) Coordination in Site 2 Results in a Nonoptimal Conformation for DNA Binding. Whereas site 1 has been shown as the primary zinc regulatory site, the role of site 2 remains elusive. We next set out to elucidate the structural and functional roles of site 2 in ZitR. To obtain a site 2-deprived ZitR protein, we mutated one of the Zn(II)-coordination residues in site 2 (C30S). Inductively coupled plasma MS analysis confirmed that this mono-Zn(II)-ZitR variant binds to only one equivalent of $Zn(II)$ per monomer [termed as $Zn(II)$ ^{site-1}-Zit R^{C30S}] ([SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Table S2). The crystal structure was solved by molecular replacement and refined to 1.65-Å resolution, which confirmed that Zn(II) was coordinated at site 1 but not site 2 (SI Appendix, Fig. S3A [and Tables S1 and S2](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)). Despite the high structural similarity between di-Zn(II)-
ZitR^{WT} and Zn(II)^{site-1}-ZitR^{C30S} when comparing their dimerization domains or DNA-binding domains separately, significant movements were observed regarding the relative positions of these domains (Fig. 1B and [SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. S3B). The winged-HTH DNA-binding domain swings ∼30° with the tip of the wing ($C\alpha$ atom of N88) translocated 15.3 Å in the $Zn(II)$ ^{site-1}-ZitR^{C30S} structure compared with di-Zn(II)-ZitR^{WT}. In addition, helix α4 rotates about 24.1° and the distance between $α4$ and $α4'$ expands to 41.9 Å, generating a "widened" conformation not suitable for DNA binding (Fig. $1B$ and SI Appendix, Table $S2$). We further mutated one of the $Zn(II)$ coordination residues in site 1 (H42A) and the resulting apo form of the double mutant protein apo-ZitRC30AH42A shared a similar structure as that of $Zn(II)$ ^{site-1}-ZitR^{C30S} ([SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf),

SI Text[, Fig. S3](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) C–G[, and Tables S1 and S2](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)). Therefore, in contrast to the optimal DNA-binding conformation observed in di- $Zn(II)$ -ZitR^{WT}, both $Zn(II)$ ^{site-1}-ZitR^{C30S} and apo-ZitR^{C30AH42A} adopt a nonoptimal DNA-binding conformation.

Functional Role of Site 2 in ZitR's Zn(II) Binding and DNA Binding. The apparent differences in ZitR structures with and without the second coordinated Zn(II) prompted us to examine the potential contribution of site 2 to site 1's Zn(II) affinity as well as to ZitR's DNA-binding affinity. Site 1's Zn(II) affinities were determined with competition against the Zn(II) chelator quin-2 $[K_{Zn(II)} =$ 3.7×10^{-12} M] and indo-1 $[K_{Zn(II)} = 1.6 \times 10^{-10}$ M] (14, 16). Analysis of the titration data confirmed that $ZitR^{WT}$ has a subpicomolar Zn(II) affinity at pH 8.0 and subnanomolar Zn(II) affinity at pH 6.0 in site 1 $[K_{Zn(II)} = 7.6 \pm 0.5 \times 10^{-13}$ M at pH 8.0 and $K_{Zn(II)} = 3.1 \pm 0.5 \times 10^{-10}$ M at pH 6.0] (Fig. 2A and [SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), SI Text and Fig. S4A), which is similar to that of AdcR $[K_{Zn(II)} = 7.1 \times 10^{-13}$ M at pH 8.0 and $K_{Zn(II)} < 1 \times 10^{-9}$ M at pH 6.0] (14). Mutation in site 2 caused a ∼10-fold reduction of $Zn(II)$ affinity in site 1 $[K_{Zn(II)} = 8.4 \pm 0.6 \times 10^{-12}$ M] (Fig. 2B). The DNA-binding affinity was measured by fluorescence anisotropy (FA) with the fluorescently labeled *zit* promoter DNA.
In comparison with $ZitR^{WT}$, which possesses an apparent DNA affinity of $2.6 \pm 0.4 \times 10^{-9}$ M, the site-2 mutant ZitR^{C30S} showed a near 10-fold decrease of apparent DNA affinity (apparent $K_{\text{DNA}} = 2.0 \pm 0.2 \times 10^{-8}$ M for ZitR^{C30S}) (Fig. 2 C and D and [SI](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) [Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. S4B). As expected, additional mutations in site 1 caused a more dramatic reduction of ZitR's zinc-binding and DNA-binding affinities $[K_{Zn(II)} = 2.3 \pm 0.2 \times 10^{-9}$ M and apparent $K_{\text{DNA}} = 1.2 \pm 0.2 \times 10^{-4}$ M for ZitR^{C30AH42A}, which is ~300-fold decreased for site 1's Zn(II) affinity and ~10⁴-fold decreased for DNA affinity in comparison with $ZitR^{C30S}$] ([SI](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) Appendix[, Fig. S3](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) H and I). Therefore, with site 1 being the primary site for regulating ZitR's DNA binding, we found that Zn(II) coordination in site 2 further enhanced its DNA-binding affinity, which may fine tune the transcriptional regulation under $Zn(II)$ stress conditions [e.g., a transient increase of free $Zn(II)$ level to the nanomolar range as a $Zn(II)$ pulse] (17).

Since FA only offers affinity constants under the steady state, we next conducted surface plasmon resonance (SPR) experiment to determine ZitR's DNA-binding kinetics that are affected by Zn(II) coordination. Under the nonequilibrium condition during SPR analysis, $\text{Zn}(II)$ ^{site-1}-ZitR^{C30S} exhibited a near threefold increase of apparent DNA-dissociation rate constant (k_{off}) over di-Zn(II)- $ZitR^{WT}$, while its apparent DNA-association rate constant (k_{on}) was nearly threefold lower than that of di-Zn(II)-ZitR^{WT} (Fig. $2 E$ and F and *[SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)*, Fig. S4 C and D). These results suggest that $Zn(II)^{\text{site-1}}$ -ZitR^{C30S} needs to readjust its conformation to fit α 4 and α 4′ helices into the DNA major grooves, while α 4 and α 4′ helices of di-Zn(II)-ZitR^{WT} can simultaneously bind both DNA major grooves in the prelocked optimal DNA-binding conformation. Taken together, these results indicate that the lack of Zn(II) in site 2 can cause ZitR to undergo the induced-fit rearrangement for DNA binding with a lower binding affinity. This is similar to the induced-fit mechanism observed in many nonmetal-responsive MarR family proteins (7, 9, 18–20). The subsequent Zn(II) coordination in site 2 then favors ZitR to adopt the more optimal conformation with enhanced DNA-binding affinity.

Histidine Switch as an Allosteric Mechanism for Controlling ZitR Conformation. To understand molecular details underlying the conformational change between the dual-Zn(II)-occupied and the mono-Zn(II)-occupied forms of ZitR, we closely inspected and compared the structures between di-Zn(II)-Zit R^{WT} and $Zn(II)$ ^{site-1}-ZitR^{C30S}. A dramatic conformational change was observed on loop 1 with and without the second zinc coordination: loop 1 is positioned toward the central region of ZitR dimer without $Zn(II)$ ion in site 2, whereas the presence of $Zn(II)$ ion in

Fig. 2. DNA-binding affinity and kinetics of ZitR with and without the second coordinated Zn(II). (A and B) Representative binding isotherms in measuring site-1 affinity by titrating Zn(II) into a mixture of ZitR protein and quin-2. The titration utilized (A) 45.0 μM ZitR^{WT} monomer and 21.7 μM quin-2 and (B) 36.2 μM ZitR^{C305} monomer and 19.9 μM quin-2. The red solid lines are fitting to a model for 1:1 stoichiometric binding, while the black solid and dashed lines are simulated curves with Zn(II) affinities of ZitR proteins 10-fold tighter and weaker than the fitted value, respectively. (C and D) Binding isotherms of titrating ZitR^{WT} (C) and ZitR^{C30S} (D) into the fluorescently labeled zit promoter DNA. The x axes are plotted on logarithmic scales. The red solid lines are simulated curves of the mean apparent K_{DNA} values in a 1:1 ZitR dimer:DNA binding model. Indicated affinity values are mean \pm SEM of three independent experiments (different symbol shapes). (*E* and *F*) SPR sensorgrams
of di-Zn(II)-ZitR^{WT} (*E*, 10–6,000 nM) and Zn(II)^{site-1}-ZitR^{C30S} (*F*, 50–30,000 nM) proteins binding to their cognate DNA. Fitted curves are shown as red lines. Indicated rate constants and affinity values are mean \pm SEM of $n =$ 3 independent experiments.

site 2 brings loop 1 into close proximity with helix α 5 (Fig. 3 A and B). As residues surrounding site 2 are heavily negatively charged (e.g., E31, D33, E41, and E107), we speculate that electrostatic repulsion may contribute to the conformational change with and without $Zn(II)$ binding in site 2. Indeed, four of the total seven charged residues in loop 1 showed large movement upon Zn(II) binding at site 2: the side chains of E31 and D33 are moved away from the helix α 5 in Zn(II)^{site-1}-ZitR^{C30S}, while the side chains of K29 and K35 are positioned within the space between helix α 5 and loop 1. In contrast, upon Zn(II) binding in site 2, the side chains of E31 and D33 were moved toward helix α 5, while the side chains of K29 and K35 were flipped away, which is consistent with our assumptions (SI *Appendix*, Fig. S5 *A* and *B*). Further electrostatic analysis also verified electrostatic repulsion between loop 1 and helix α 5 when both zinc sites are occupied (Fig. 3C and [SI Ap](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)*pendix*[, Fig. S5](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)C). Notably, when contoured at 1.0 σ , the $2F_0-F_c$ density map of loop 1 is clear in di-Zn(II)-ZitR^{WT} on the amino acid side chains, but is missing at several amino acid side chains and even a portion of the main chain in the $Zn(II)^{site-1}$ - $ZitR^{C30S}$ structure. This indicates that the flexibility of loop 1 is lower in di- $Zn(II)$ -ZitR^{WT} than that of $Zn(II)$ ^{site-1}-ZitR^{C30S} ([SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. S5 D [and](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) E). Therefore, $Zn(II)$ coordination in site 2 forces loop 1 to adopt a more rigid, higher energy conformation in close proximity with helix α 5, which further causes a rearrangement of helix α 2 that ultimately propagated to the DNA-binding sites ([SI](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) [Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. S5 F and G).

We found that $Zn(II)$ -binding residue H42 in site 1 on helix α2 uses different nitrogen atoms from its imidazole ring for zinc coordination: the $Zn(I)$ -binding atom switches from Nδ1 in di-
 $Zn(II)$ -ZitR^{WT} to Nε2 in $Zn(I)$ ^{site-1}-ZitR^{C30S} (Fig. 3 D and E and \overline{SI} Appendix, Fig. S6 A and B). This switch of Zn(II)coordinating nitrogen on H42 (termed "histidine switch") is triggered by the movement of loop 1 and helix α 2 upon Zn(II) binding in site 2. The distance between Zn(II) and the Cβ atom of H42 decreases from 5.5 Å to 3.6 Å and causes a 16.4° rotation of helix α 2, which ultimately drives the swing of wHTH DNA-binding motif ([SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Figs. S5G and S6C). To further verify the structural and biochemical differences between $Zn(II)$ ^{site-1}-ZitR^{C30S} and di-Zn(II)-ZitR^{WT} that are dependent on Zn(II) coordination at site 2, we mutated E41, another $Zn(II)$ -binding residue in site 2, and solved the crystal structure of this ZitR variant ($ZitR^{E41A}$) by molecular replacement. The structure was refined to 1.9-Å resolution and also exhibited the non–DNA-binding conformation with the distance between α4 and α ²' at 36.0 Å (*[SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)*, Fig. S7 *A* and *B* and Tables [S1 and S2](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)). Notably, similar to Zn(II)^{site-1}-ZitR^{C30S}, the Zn(II)coordination atom on H42 was also flipped to Ne2 in this ZitR variant (*[SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), SI Text* and Fig. S7C). Meanwhile, simi-
lar to Zn(II)^{site-1}-ZitR^{C30S}, ZitR^{E41A} also exhibited decreased apparent DNA affinity (apparent $K_{\text{DNA}} = 5.4 \pm 0.1 \times 10^{-8}$ M, 20-fold lower than Zit R^{WT} , while its Zn(II) affinity was still in the picomolar range $[K_{Zn(II)} = 3.1 \pm 0.8 \times 10^{-12} \text{ M}, 4 \text{-fold lower}]$ than $ZitR^{WT}$ (*[SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)*, Fig. S7 D and E). We thus speculate that the switch of $Zn(II)$ -coordination nitrogen on H42 may serve as an allosteric mechanism that regulates ZitR between the nonoptimal and optimal DNA-binding conformations.

To further verify the switchable feature of H42, we solved the crystal structure of the $Zn(\text{II})^{\text{site-1}}$ -Zit R^{C30S} protein in complex with DNA by molecular replacement and refined to 2.9-Å resolution (SI Appendix, Fig. S8A [and Tables S1 and S2\)](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf). Zn(II)^{site-1}-ZitR^{C30S} protein within this complex exhibits an optimal DNA-binding conformation that is highly similar to $ZitR^{WT}$ from the di- $Zn(II)$ -Zit R^{WT} -DNA complex ([SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. S8 B–[D](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf)). In particular, the coordination environment of its site 1 resembles that of di-Zn(II)-ZitR^{WT}, with the Zn(II)-coordination atom on H42 now switched back to the N δ 1 atom (Fig. 3 D and F and [SI](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) Appendix[, Fig. S8](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) E and F). Therefore, whereas the mono- $Zn(II)$ -occupied ZitR adopts a non–DNA-binding conformation with $Zn(II)$ coordinated to H42's Nε2 atom in site 1, this ZitR variant can be reversed to adopt the DNA-binding conformation via the induced-fit mechanism, which switched H42's Zn(II) coordination atom back to Nδ1 upon DNA binding. Interestingly, we found no hydrogen bonding networks within these crystal structures that may contribute to ZitR's allosteric control of DNA

Fig. 3. A histidine-switch mediated site-1 and site-2 cooperation for optimal DNA binding. (A) Superposition of crystal structures of monomers from di-Zn(II)- ZitR^{WT} (green) and Zn(II)^{site-1}-ZitR^{C30S} (yellow) proteins. Coordinated Zn(II) ions are shown as spheres. (B) Closeup view of A shows that coordination of Zn(II) at site 2 brings loop 1 to close proximity with helix α5. (C) Electrostatic potential surface presentation of site 2 in the di-Zn(II)-ZitR^{WT} structure displayed at the +3 kT/e (blue) and −3 kT/e (red) levels. Zn(II)-occupied site 2 is surrounded by negatively charged residues. Side chains of residues from site 2 (C30, E41, and E107) and the four charged residues (K29, E31, D33, and K35) from loop 1 are labeled and shown as sticks. (D and E) Closeup view of the site-1 Zn(II)-binding pockets of
Zn(II)^{site-1}-ZitR^{C30S} (D) and di-Zn(II)-ZitR^{WT} (E). Diffe absence of Zn(II) at site 2. (F) Crystal structure of the Zn(II)^{site-1}-ZitR^{C30S}-DNA complex, with the distances between α 4 and α ^{4'} labeled. Closeup view of the site-1 Zn(II)-binding pocket is shown below. The Zn(II)-coordination atom on H42 is now switched from the Ne2 atom back to the Nδ1 atom in the presence of DNA. All electron density maps ($2F_o$ - F_o) are contoured at 1.0 σ .

recognition, which is different from the previously reported AdcR protein (15) ([SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), SI Text and Fig. S1C). Therefore, it remains to be verified whether AdcR uses a similar mechanism for allosteric control of its DNA binding to ZitR presented here.

Genetic Incorporation of a Nonswitchable Histidine Analog to Verify the Histidine-Switch Mechanism. To further probe the distinct switchable feature of H42 in allosteric control of ZitR, we used a nonswitchable unnatural histidine analog to site-specifically replace H42 on ZitR. We envisioned that the blockage of Nδ1 on H42 may have neglected effects on Zn(II)-binding affinity in site 1 since ZitR can utilize H42's Ne2 for Zn(II) binding and adopt a non–DNA-binding conformation as shown by $Zn(II)^\text{site-1}$ - $ZitR^C$ However, because the "switching back" of H42's Zn(II)-binding atom to Nδ1 is essential for ZitR to adopt the optimal DNAbinding conformation, blockage of this Nδ1 atom will significantly disrupt ZitR's DNA binding. Previous work has incorporated Ne2 methylated histidine into metalloregulatory proteins via native chemical ligation to investigate their allostery (21, 22). Here we introduced the Nδ1-blocked histidine, termed δ-methyl-histidine (δmeH) (Fig. 4A), as an unnatural histidine analog to replace H42 on ZitR via the genetic code expansion strategy (23, 24). The pyrrolysyl-based aminoacyl-tRNA synthetase (PylRS)/tRNA pair from the archaea species Methanosarcina barkeri and Methanosarcina mazei has recently been employed to genetically incorporate δmeH into proteins at an in-frame amber codon (25, 26). By using a mutant PylRS and its cognate tRNA pair, we sitespecifically incorporated δmeH at residue 42 on ZitR^{WT}, which was verified by mass spectrometry (Fig. 4B). As expected, the generated ZitR variant ZitR^{H42δmeH} has a Zn(II)-binding affinity of $7.2 \pm 0.5 \times 10^{-12}$ M in site 1, which is only approximately eightfold

lower than that of $ZitR^{WT}$ and is at a similar level as that of $ZitR^{C30S}$ (Figs. $2 \text{ } A$ and B and $4C$). This result suggests that replacing H42 with δmeH has limited effects on ZitR's metal coordination property in site 1 because Ne2 on H42δmeH can be used as the Zn(II)-coordination atom under this condition. In contrast, when H42 becomes nonswitchable, dramatically decreased DNA-binding affinity (apparent $K_{\text{DNA}} = 7.9 \pm 0.8 \times 10^{-6}$ M) was observed on $ZitR^{H42\delta m eH}$, which is three magnitudes lower than that of $ZitR^{WT}$ (Figs. $2C$ and $4D$). This result verified our speculation that $Zn(II)$ coordination with H42's Nδ1 atom is essential for ZitR to adopt the optimal DNA-binding conformation. Therefore, the switchable feature of H42 is critical for controlling the transition between ZitR's DNA-binding and non–DNA-binding conformations.

Investigating the Site 2-Mediated ZitR Response to Zn(II) Stress Inside Cells. Finally, we directly tested the site 2-mediated ZitR response to $Zn(II)$ stress inside cells. Since $Zn(II)$ coordination at this nanomolar-affinity site would enhance ZitR's DNA-binding affinity, we reasoned that $Zn(II)$ binding at site 2 may facilitate ZitR's transcription regulation under the nanomolar Zn(II) stress conditions. Previous work has shown that the addition of a large excess of Zn(II) ions (micromolar level) to bacterial culture can rapidly increase the intracellular Zn(II) to the nanomolar range (within 10 min) (17). We investigated whether site 2's coordination would enhance ZitR's transcriptional repression upon such a Zn(II) pulse treatment. To develop a cell-based assay to monitor the transcriptional regulation of ZitR, we engineered a ZitR-inducible GFP reporter that encodes the gfp gene under the control of zit promoter (zitR-gfp reporter, [SI](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) [Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Fig. S9). The reporter plasmid was cotransformed with a ZitR-expressing plasmid (pBAD-ZitR) into the Escherichia coli

Fig. 4. Allosteric control of the broad-range transcriptional regulator ZitR via a histidine-switch mechanism. (A–D) Verifying the histidine-switch mechanism by a genetically encoded histidine analog δ-methyl-histidine (δmeH, A). δmeH was incorporated at residue H42 in ZitRWT by the MbPylRStRNA^{pyl} pair. The efficiency and fidelity of this incorporation on the resulting protein ZitR^{H42δmeH} was confirmed by mass spectrometry (Β). (C) Representative binding isotherms of titrating Zn(II) into a mixture of 23.0 μM $ZitR^{H42\delta meH}$ monomer and 20.7 μ M quin-2. Data are presented in the same way as in Fig. 2 A and B. (D) Binding isotherms of titrating ZitR^{H42δmeH} into the fluorescently labeled zit promoter fragment. The x axis is plotted on a logarithmic scale. Data are presented in the same way as in Fig. 2 C and D. Indicated affinity values are mean \pm SEM of three independent experiments.

BW25113 strain. This allowed us to quantify the ZitR-induced transcriptional repression via flow cytometric analysis of GFP expression. Upon the addition of 100 μM Zn(II), the ZitRWT- expressing bacterial strain harboring the zitR-gfp reporter showed a decreased fluorescence compared to without Zn(II) treatment, indicating that the transcription of GFP was further repressed under this $Zn(II)$ stress condition (Fig. 5A). The $ZitR^{C30S}$ expressing strain harboring the same z *it* R -gf p reporter was used as a control. Because ZitR^{C30S} loses $Zn(II)$ binding at site 2, this bacterial strain showed negligible change with and without 100 μM Zn(II) (Fig. 5B). Together, this cell-based analysis agrees with our aforementioned in vitro results that Zn(II) coordination at site 2 may enhance and fine tune ZitR's DNA-binding affinity and thus transcriptional repression when cells are under Zn(II) stress conditions [e.g., nanomolar Zn(II) pulse].

Discussion

We report here a unique histidine switch-mediated allosteric regulation mechanism for the zinc transcriptional factor ZitR (Fig. 5C). Metalloproteins are specialized allosteric proteins in which the metal binding-induced conformational changes are frequently employed for allosteric control of binding with substrates, cofactors, partner proteins, as well as additional metal ions (27–32). In particular, additional unique features of metal ions and/or ligands have been harnessed by metalloproteins for allosteric regulation. The former example includes hemoglobin, which uses the spin-state alteration between the heme-bound Fe(II) and Fe(III) ions to control oxygen binding (33, 34). For the latter case, the two nitrogen atoms on the same histidine side chain of Mycobacterium tuberculosis CsoR are engaged in Cu(I) binding and hydrogen-binding network formation, respectively, which permits the coupling of metal coordination with the allosteric control of DNA binding (21, 35). The unique histidineswitch mechanism we observed here represents an unprecedented example in which the heteroaromatic feature of histidine's imidazole ring is exploited to coordinate two neighboring metal-binding sites for allosteric control and fine tuning of metalloregulator's DNA-binding affinity (Fig. 5C). The multifunctional properties of

Fig. 5. ZitR fine tunes cell's transcriptional response to Zn(II) fluctuation. (A and B) Site 2-mediated ZitR response to Zn(II) stress inside cells. Flow cytometric analysis of the zitR-gfp reporter-harbored E. coli cells expressing ZitR^{WT} (A) or ZitR^{C30S} (B) proteins with and without the Zn(II) pulse treatment. Since previous work has shown that the addition of the micromolar level of Zn(II) ions can increase the intracellular free Zn(II) to the nanomolar range within 10 min, the bacteria were treated with (red lines) and without (black lines) 100 μM Zn(II) ions for 1 h before the fluorescence intensity of expressed GFP was recorded. (C) The proposed working model for histidine switch-mediated allosteric control of ZitR for the broad-range and fine-tuned transcriptional response to Zn(II) fluctuation. Zn(II) binding at site 1 under steady state [subpicomolar range of Zn(II)] allows ZitR to undergo an induced-fit mechanism for DNA binding, which can be used to control the steady-state zinc homeostasis. Elevated Zn(II) levels would produce a dual-Zn(II) form of ZitR with both sites occupied, which prelocks an optimal DNA-binding conformation for strong transcriptional response under zinc stress conditions [e.g., nanomolar Zn(II) pulse]. A histidine-switch mechanism on site 1's Zn(II) coordination residue H42 mediates the transition between these two ZitR forms to fine tune the cell's transcriptional response to intracellular Zn(II) fluctuation.

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histidine may be employed as a general allosteric strategy to couple metal ion coordination with protein conformational change.

Instead of being gratuitous, our study in vitro and inside bacterial cells suggests that $Zn(II)$ binding at site 2 is involved in the allosteric regulation of ZitR's transcriptional corepression. Interestingly, the zinc affinities in site 1 (subpicomolar) and site 2 (nanomolar) fall in the range of the intracellular Zn(II) concentrations under steady-state and stress-state conditions, respectively (17, 30, 36–38). The large reduction ($\sim 10^4$ -fold) of the DNA affinity induced by site-1 mutation on ZitR protein shows that the primary role of ZitR is regulating steady-state zinc homeostasis via Zn(II) coordination in site 1. Meanwhile, the higher DNA-binding affinity (∼10-fold higher) observed on dual-Zn(II)-occupied ZitR seems to indicate that ZitR may mount an enhanced transcriptional response (e.g., repression of zinc uptake genes) through site 2 upon a nanomolar Zn(II) pulse (Fig. 5C). When lacking Zn(II) coordination in site 2, ZitR adopts a non–DNA-binding conformation that may undergo the inducedfit mechanism for DNA binding as well as the control of steadystate zinc homeostasis with a weaker repression. Together, a single metalloregulator ZitR may maintain broad as well as finetuned responses to steady-state and stress-state Zn(II) fluctuations via an allosteric histidine switch-mediated dual Zn(II)-site cooperation. Previous work on S. pneumoniae bacteria harboring different AdcR variants showed that although site 1 plays a major role in AdcR function in vivo, bacterial strains harboring a site-2 mutant AdcR (e.g., adcR-C30A) indeed showed a slightly higher but statistically significant amount of cellular zinc than the

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parent $adcR^+$ strain (14). Similarly, this small difference may indeed account for the physiological role of site 2 in ZitR, which is expected to be a fine-tuned response. In addition, another Zn(II) transcription regulator Zur contains multiple Zn(II) binding sites and is previously reported to mount a stepwise repression of different promoters with altered Zn(II) levels (39–41). Considering the fact that both ZitR and Zur proteins have enhanced DNA affinity upon additional Zn(II) coordination besides their primary sites, ZitR might also adopt such a "stepwise" regulation. For example, a subset of promoters might be more dependent on site 2 and undergo enhanced repression at highly elevated Zn(II) levels [e.g., nanomolar Zn(II) pulse] to cope with transient Zn(II) stress. This speculation requires further investigation, particularly under in vivo settings (14).

Methods

Detailed materials and methods are available in [SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf). See SI Appendix, [Table S1](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) for data collection and refinement statistics of ZitR variants and [SI](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) [Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), Table S2 for the structural information of ZitR variants. [SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), [Table S3](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) shows conditions for crystallization of ZitR variants. [SI Appendix](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf), [Table S4](http://www.pnas.org/lookup/suppl/doi:10.1073/pnas.1708563115/-/DCSupplemental/pnas.1708563115.sapp.pdf) reports the oligonucleotides used in this study.

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