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## 4*S*-Hydroxylation of insulin at ProB28 accelerates hexamer dissociation and delays fibrillation

Seth A. Lieblich<sup>†,‡</sup>, Katharine Y. Fang<sup>†,‡</sup>, Jackson K. B. Cahn<sup>†</sup>, Jeffrey Rawson<sup>§,||</sup>, Jeanne LeBon<sup>§</sup>, H. Teresa Ku<sup>§,||,¶</sup>, and David A. Tirrell<sup>†,\*</sup>

<sup>†</sup>Division of Chemistry and Chemical Engineering, California Institute of Technology, Pasadena, CA 91125, USA

<sup>§</sup>Department of Translational Research and Cellular Therapeutics, Diabetes and Metabolism Research Institute, City of Hope, Duarte, CA 91010, USA

<sup>||</sup>Beckman Research Institute of City of Hope, Duarte, CA 91010, USA

<sup>¶</sup>Irell & Manella Graduate School of Biological Sciences, City of Hope, Duarte, CA 91010, USA

### Abstract

Daily injections of insulin provide lifesaving benefits to millions of diabetics. But currently available prandial insulins are suboptimal: The onset of action is delayed by slow dissociation of the insulin hexamer in the subcutaneous space, and insulin forms amyloid fibrils upon storage in solution. Here we show, through the use of non-canonical amino acid mutagenesis, that replacement of the proline residue at position 28 of the insulin B-chain (ProB28) by (4*S*)-hydroxyproline (Hzp) yields an active form of insulin that dissociates more rapidly, and fibrillates more slowly, than the wild-type protein. Crystal structures of dimeric and hexameric insulin preparations suggest that a hydrogen bond between the hydroxyl group of Hzp and a backbone amide carbonyl positioned across the dimer interface may be responsible for the altered behavior. The effects of hydroxylation are stereospecific; replacement of ProB28 by (4*R*)-hydroxyproline (Hyp) causes little change in the rates of fibrillation and hexamer disassociation. These results demonstrate a new approach that fuses the concepts of medicinal chemistry and protein design, and paves the way to further engineering of insulin and other therapeutic proteins.

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Blood glucose levels are tightly controlled in mammals through a sensitive regulatory system mediated by insulin, a 51-amino acid endocrine hormone composed of two disulfide-linked polypeptide chains (designated A and B). Upon binding to its receptor, insulin initiates a signaling cascade that accelerates glucose uptake and glycogen production. In diabetic patients, this system malfunctions, and glucose levels must be controlled through subcutaneous injections of insulin<sup>1</sup>. The *C*-terminus of the B-chain is important in mediating dimerization of the hormone<sup>2–3</sup>, and the flexibility of the B-chain *C*-terminus is believed to

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\*Corresponding author. tirrell@caltech.edu.

<sup>‡</sup>Authors contributed equally to this work.

### Supporting Information

The following files are available free of charge via the internet at <http://pubs.acs.org>. Experimental procedures and methods, supporting figures and tables, additional references.

contribute to aggregation through formation of amyloid fibrils<sup>4–6</sup>. Pharmaceutical formulations of insulin are stabilized with respect to fibrillation by addition of zinc and phenolic preservatives, which drive assembly of the R<sub>6</sub> hexamer (Fig. 1a)<sup>7–9</sup>. The R<sub>6</sub> form of insulin is inactive and dissociates slowly to its active monomeric form after subcutaneous injection; the lag time for dissociation delays the onset of action<sup>10</sup>. Mutation of ProB28 yields rapid-acting insulins (RAIs) by disrupting contacts that are critical for dimer formation<sup>8</sup>, but replacement of Pro through conventional mutagenesis also increases the flexibility and perturbs the trajectory of the protein backbone (Fig. 1B). We sought a means to disrupt the dimer interface without releasing the conformational constraints characteristic of proline by using non-canonical amino acid (ncAA) mutagenesis<sup>11–13</sup>. Specifically, we introduced hydroxyl groups at the 4-position of ProB28 (Fig. 1B, C) by replacing Pro with Hzp or Hyp. In addition to introducing a polar functional group and the capacity for hydrogen-bonding (including transannular hydrogen bonding), hydroxylation at the 4-position is known to alter the *endo/exo* preference of the pyrrolidine ring and the *cis/trans* equilibrium of the backbone amide bond<sup>14–16</sup>.

We expressed modified proinsulins (PIs) in the proline-auxotrophic *E. coli* strain CAG18515 in M9 minimal media supplemented with Hyp or Hzp. The extent of replacement of Pro by either Hyp or Hzp was approximately 90%<sup>17–18</sup> as determined by matrix-assisted laser desorption/ionization mass spectrometry (MALDI-MS; fig. S1). Denatured PIs were purified by Ni-NTA affinity chromatography in yields of 32 mg/L for Hzp-PI and 29 mg/L for Hyp-PI (table S1) from the inclusion body fraction. The PIs were refolded and cleaved with trypsin and carboxypeptidase B<sup>19</sup>. The resulting mature insulins were purified by reversed phase HPLC<sup>18–19</sup>, and proper proteolytic processing of each variant was verified by MALDI-MS (table S1). Wild-type insulin (ProI) and RAI Aspart (AspI, in which ProB28 is replaced by aspartic acid) were produced similarly. All of the variants caused similar reductions in blood glucose upon subcutaneous injection into diabetic mice (Fig. 1D)<sup>2, 20–22</sup>. RAIs cannot be distinguished from ProI in rodent models<sup>23</sup>.

In the absence of Zn<sup>2+</sup> and phenolic preservatives, insulins dimerize with a dissociation constant (K<sub>D</sub>) of approximately 10 μM. In contrast, K<sub>D</sub> for RAIs is typically >500 μM, and it is believed that destabilization of the dimer interface causes the accelerated onset of insulin action after subcutaneous injection<sup>20, 24–25</sup>. Monomeric forms of insulin give rise to characteristic circular dichroism (CD) spectra with distinct minima at 208 and 222 nm<sup>26–27</sup> (e.g., AspI; Fig. 2A). Dimerization causes a loss of negative ellipticity at 208 nm<sup>27</sup> (e.g., ProI; Fig. 2A). At a concentration of 60 μM, HypI appears to be monomeric (with a CD spectrum nearly identical to that of AspI; Fig. 2A) while the spectrum of HzpI suggests a dimeric insulin (Fig. 2A). Sedimentation velocity (SV) and sedimentation equilibrium (SE) experiments were consistent with the results of the CD analysis (fig. S3). SE data were fitted to a model of monomer-dimer-hexamer self-association (SEDPHAT)<sup>28–29</sup>, and yielded monomer-dimer dissociation constants (K<sub>D</sub>) of >200 μM and 25 μM for HypI and HzpI, respectively.

Previous studies of RAIs have shown that destabilization of the dimer interface correlates with accelerated dissociation of the hexamer and rapid onset of insulin action<sup>8, 13</sup>. Triggered dissociation of Zn<sup>2+</sup>-hexamers by addition of terpyridine<sup>30</sup> revealed nearly identical rates of

dissociation for HypI and ProI, ( $\tau_{1/2} = 87.0 \pm 10$  s and  $90.4 \pm 4.2$  s, respectively; Fig. 2B and fig. S4) while HzpI exhibited kinetics similar to those of AspI ( $\tau_{1/2} = 53.6 \pm 3.7$  s and  $42.7 \pm 4.3$  s, respectively; Fig. 2B and fig. S4). We found these results surprising – replacement of Pro by Hyp destabilizes the dimer but has essentially no effect on hexamer dissociation, while introduction of Hzp causes little change in dimer stability but a substantial increase in the rate of hexamer disassembly.

Each of the insulin variants was subjected to fibrillation lag time analysis (Fig. 2C)<sup>31</sup>. We found similar times to onset of fibrillation for HypI, ProI and AspI; in contrast, HzpI is markedly more resistant to aggregation, with a mean time to onset more than three-fold longer than that observed for ProI. The behavior of HzpI is especially striking, in that it combines fast hexamer dissociation with enhanced stability toward fibrillation.

Each subunit in the insulin hexamer adopts one of two conformational states (T or R), depending on the concentration of phenolic ligand (Fig. 1A)<sup>13</sup>. Pharmaceutical formulations are prepared in the more stable R<sub>6</sub> form, whereas the T-state is observed in the absence of phenolic ligands, most commonly in the form of T<sub>2</sub>-dimers<sup>32</sup>. To elucidate the molecular origins of the dissociation and fibrillation behavior of HypI and HzpI, we examined crystal structures of both states.

Hydroxylation at ProB28 does not cause substantial perturbation of the overall insulin structure (Fig. 3, fig. S5). In comparison with ProI, the backbone RMSD values of HypI and HzpI are 0.31 Å (T<sub>2</sub>-HypI), 0.44 Å (T<sub>2</sub>-HypI), 0.38 Å (R<sub>6</sub>-HypI) and 0.69 Å (R<sub>6</sub>-HypI)<sup>33</sup>. The most notable feature of the HzpI structures is the proximity of the hydroxyl group of Hzp to the backbone carbonyl oxygen atom of GluB21', which lies across the dimer interface (denoted by prime; Fig. 3B, E). The inter-oxygen distances (2.8 Å in the T<sub>2</sub> structure, 2.7 Å in R<sub>6</sub>), are consistent with the formation of strong hydrogen bonds between the hydroxyl group of HzpB28 and the backbone carbonyl of GluB21' in both structures. An analogous hydrogen bond has been observed in a structure (PDB ID: 1ZEH) of R<sub>6</sub>-AspI co-crystallized with *m*-cresol<sup>34</sup>; here the phenolic ligand serves as the hydrogen-bond donor (fig. S6). Although the significance of this hydrogen bond has not been discussed in the literature, we suggest that it may play an important role in determining the relative stabilities of the insulin species involved in dissociation and fibrillation. In contrast to the (4*S*)-hydroxyl group of Hzp, the (4*R*)-hydroxyl of HypB28 does not contact any crystallographically resolved hydrogen bond acceptor in the T<sub>2</sub>-structure (Fig. 3C), and appears to bond to an ordered water molecule in the R<sub>6</sub>-hexamer (Fig. 3F). The absence of new hydrogen-bonding interactions is consistent with the unaltered dissociation and fibrillation kinetics of HypI.

Taken together, our results show that replacement of Pro by Hzp at position 28 of the insulin B-chain introduces a new hydrogen bond across the inter-subunit interface, accelerates hexamer dissociation and delays the onset of fibrillation (Table 1). We suggest that the hydrogen bond between Hzp and Glu21' may stabilize the dimer relative to the hexamer, or perhaps reduce the energy of the transition state for the conformational change from the R-state to T-state, and thereby speed disassembly. The delayed onset of fibrillation may reflect changes in the structure and dynamics of the HzpI monomer or in the kinetics of fibril

nucleation. Subtle conformational effects caused by *4S*-substitution on the pyrrolidine ring may also contribute to the observed behavior.<sup>35</sup> Whether or not these hypotheses are correct, the results described here demonstrate the power of ncAA mutagenesis to control functionally relevant biophysical properties of therapeutic proteins. We anticipate that this approach will find increasing application in the design of antibody-drug conjugates, bispecific antibodies, and other novel protein therapeutics.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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## Abbreviations

<b>Hzp</b>	( <i>4S</i> )-hydroxyproline
<b>Hyp</b>	( <i>4R</i> )-hydroxyproline
<b>RAI</b>	Rapid Acting Insulin
<b>ncAA</b>	non-canonical amino acid
<b>PI</b>	proinsulin
<b>HypI</b>	Hyp-Insulin
<b>HzpI</b>	Hzp-Insulin
<b>ProI</b>	Wild-type human insulin
<b>AspI</b>	Insulin Aspart
<b>CD</b>	circular dichroism
<b>SV</b>	sedimentation velocity
<b>SE</b>	sedimentation equilibrium

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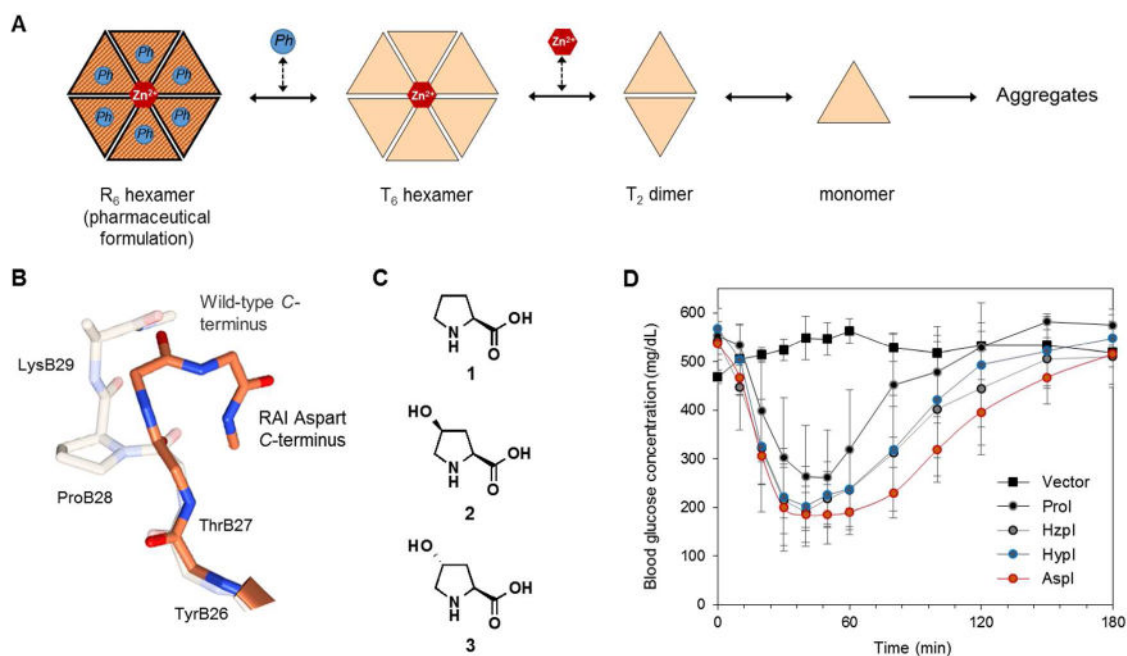
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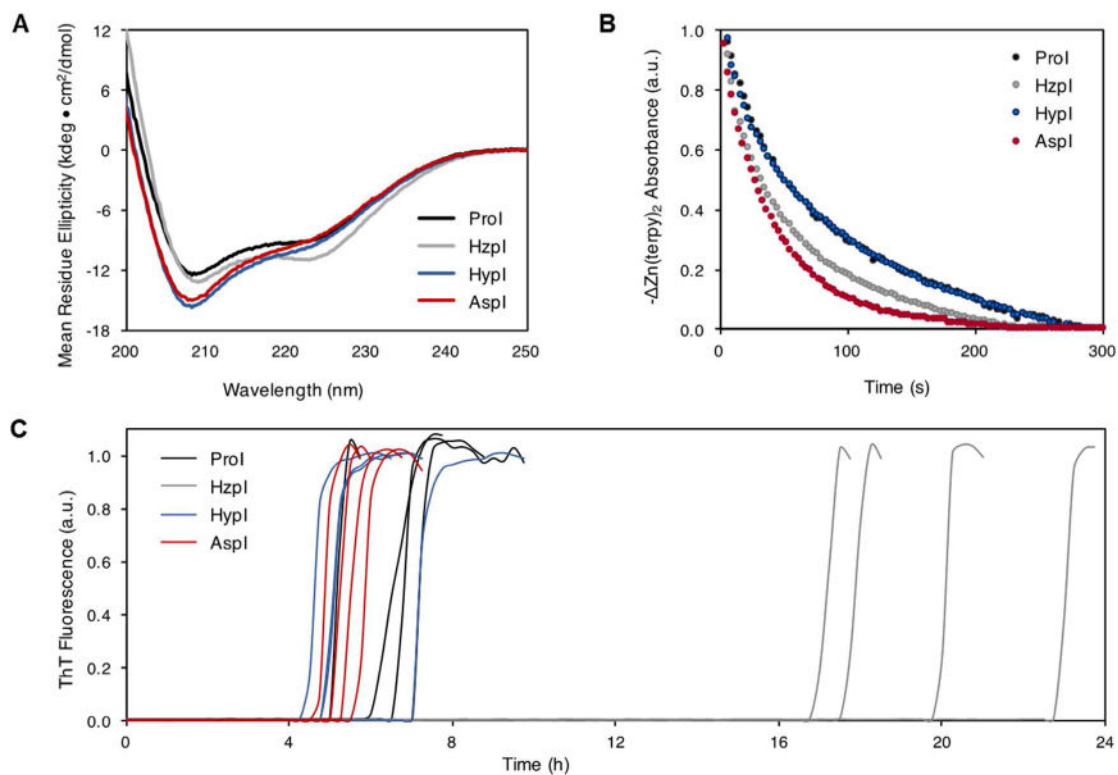
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**Figure 1. Hydroxyinsulins retain activity**

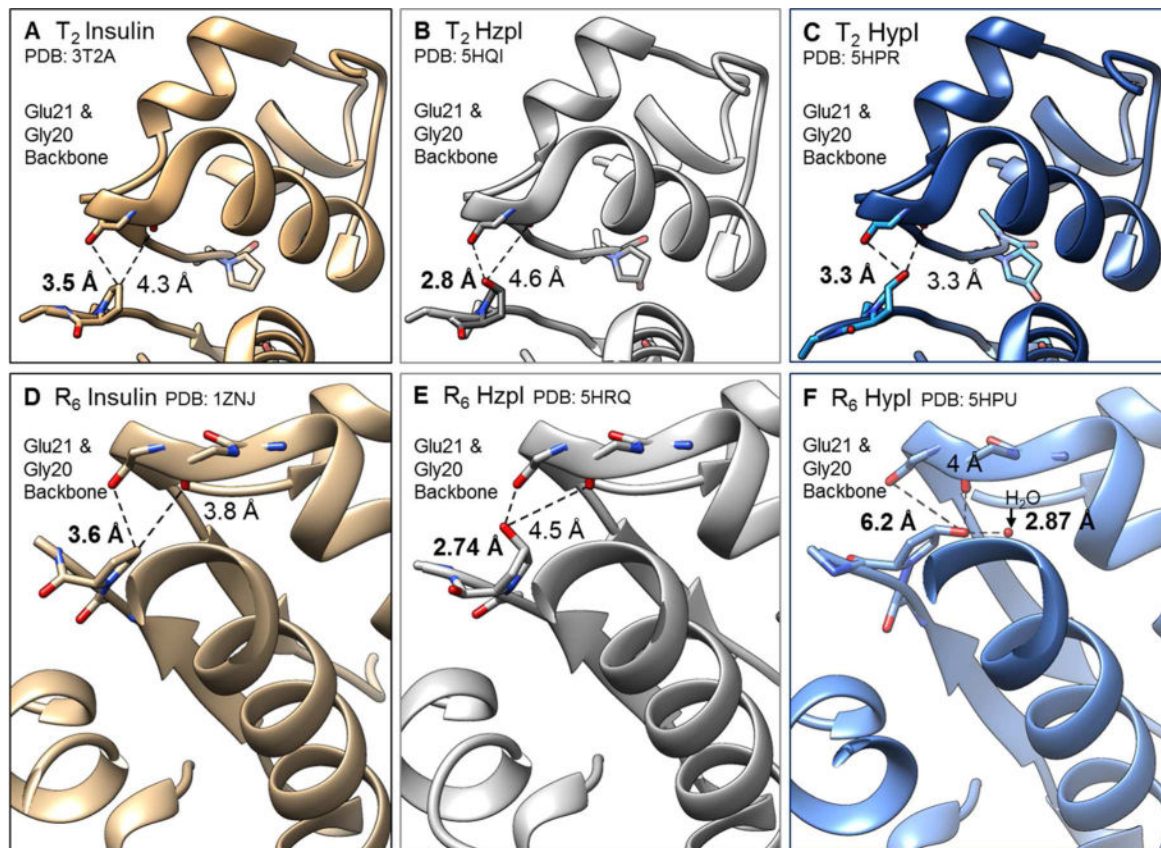
(A) Schematic of hexamer disassembly (adapted from mechanism previously described<sup>8</sup>). Phenolic ligand (Ph), zinc ion ( $Zn^{2+}$ ), insulin monomer (triangle). Darker shading indicates the R-state of the hexamer. (B) Structures of the B-chain C-termini of wild-type insulin (ProI) and RAI Aspart (AspI). (C) Chemical structures of L-proline (1), (2*S*,4*S*)-hydroxyproline (Hzp; 2), (2*S*,4*R*)-hydroxyproline (Hyp; 3). (D) Reduction of blood glucose following subcutaneous injection of 35  $\mu$ g/kg insulins into streptozotocin-induced diabetic mice. Glucose levels were measured post-injection via tail vein sampling. ProI, AspI, HzpI, HypI or vector were formulated as described<sup>22</sup>. Error bars denote one standard deviation ( $n = 3$ ).



**Figure 2. Hydroxylation at ProB28 modulates insulin dimerization, dissociation kinetics, and stability**

(A) Far UV CD spectra collected on 60 μM insulins in 10 mM phosphate buffer, pH 8.0 at 25°C. (B) Insulin hexamer dissociation following sequestration of Zn<sup>2+</sup> by terpyridine. Zn<sup>2+</sup>-(terpy) signal was monitored at 334 nm and fitted to a mono-exponential decay. Hzpl and HypI contain 10% ProI. The curves for HypI and ProI are indistinguishable in this plot. (C) Representative fibrillation curves for 60 μM insulins (37°C, 960 RPM; n=4). Insulin fibrils were detected by the rise in Thioflavin T (ThT) fluorescence that accompanies binding to fibrillar aggregates.





**Figure 3. Crystal structures of HzpI and HypI**

In tan (left), wt insulins from PDB (3T2A, 1ZNJ) highlighting the distance between the carbon atom at the 4<sup>th</sup> position of ProB28 and its closest neighbors, backbone carbonyl oxygen atoms of GlyB20' and GluB21 in the T<sub>2</sub> dimer (**A**) and R<sub>6</sub> hexamer (**D**) forms. In grey (middle), HzpI in the T<sub>2</sub> dimer (**B**) and R<sub>6</sub> hexamer (**E**) forms. In blue (right), HypI in the T<sub>2</sub> dimer (**C**) and R<sub>6</sub> hexamer (**F**) forms.

**Table 1**

## Biophysical Characteristics of Insulin Variants

Protein	Hexamer $t_{1/2}$ (s)	Fibrillation lag time(h)	$K_D$ dimer ( $\mu$ M)
Prol	90.4 $\pm$ 4.2	5.1 $\pm$ 1.5	9 $\mu$ M
Hzipl *	53.6 $\pm$ 3.7	19.6 $\pm$ 2.6	~25 $\mu$ M
HypI *	87.0 $\pm$ 10	5.5 $\pm$ 1.2	>200 $\mu$ M
Aspl	42.7 $\pm$ 4.3	5.3 $\pm$ 1.0	>500 $\mu$ M

Errors are given as one standard deviation ( $n > 4$ ).

\* 10% Prol present.

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