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Bioactive compounds and antioxidant activities of some cereal milling by-products

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Abstract The present study was performed to evaluate the phytochemicals profiles of some cereal milling by-products such as wheat (bran, germ and shorts), rice (bran, germ and husk) and corn (bran, germ and germ meal) to assess their potentiality as bioactive compounds sources. Distilled water, ethanol, methanol, and acetone separately were used as solvents for the extraction of phytochemicals compounds. The antioxidant activity (AOA), total phenolics content (TPC), and total flavonoids content (TFC) of the extracts were investigated using various in vitro assays. The results showed that tannins content was ranged from 113.4 to 389.5 (mg/100 g sample). The study revealed that TPC and TFC of cereal by-products extracts were significantly different for various solvents. TPC content varied from 366.1 to 1924.9 mg/100 g and TFC content varied from 139.3 to 681.6 mg/100 g. High carotenoids content was observed for corn germ meal and minimum for wheat bran. Distilled water, ethanol and methanol extracts showed significantly different antioxidant activity. Significant variations were observed with regard to AOA of

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² Department of Food Science, Design and Consumer Behaviour, Faculty of Science, University of Copenhagen, Rolighedsvej 30, 1958, Frederiksberg, Denmark different cereal by-products by using various solvents. The ethanol and methanol were observed to be the best solvents to extract phenolic compounds and antioxidant activity, while acetone extract showed less efficiency. Also, the cereal milling by-products were rich in bioactive compounds and could be used as a value added products.

Keywords Cereal milling by-products · Phytochemicals · Antioxidant activity · Carotenoids · TPC

Abbreviations

AOA	Antioxidant activity
TPC	Total phenolic content
TFC	Total flavonoids content

Introduction

Cereals constitute a major part of human nutrition, being the important source of proteins and energy, particularly in developing countries. Cereals and their products are rich in antioxidant phytochemicals that make them ideal for developing functional foods and ingredients (Serafini et al. 2002). Cereals and its by-products like bran have antioxidative, antimutagenic, and anticarcinogenic activities (Nam et al. 2005).

Cereals such as wheat and rice represent more than half of the world's grain production, accounts for a chief source of wastes in several countries (Zhang et al. 2011). These wastes are also a good source of dietary fibers (Crittenden et al. 2002) and bioactive compounds like phenolics, pigments, flavonoids, tannins and vitamins. Hence, the production of value added products such as food additives and supplements from food processing wastes have gained worldwide attention (Wang and Chen 2010). The valorizations of wastes are not only economical but also environmentally benign (Djilas et al. 2009).

The phenolics compound in cereals could be grouped into benzoic acid or cinnamic acid derivatives, where ferulic acid is the major phenolic acid in cereals (Abdel-Aal et al. 2001). The phytochemicals are concentrated in the outer layers compared to the internal layers of the cereals. Thus, the cereals by-products have a high content of bioactive, particularly antioxidant compounds such as phenolics compound, tannins, and carotenoids (Heinio et al. 2008). The antioxidant potential of these compounds could be attributed to several mechanisms including free radical scavenging, minimizing the formation of peroxides, metal ion chelation and enzyme activation of defense system (Zhou and Yu 2004a).

The different solvent can yield extracts with variable antioxidant activity, consequently, the capabilities of extracts might be variable to prevent lipid oxidation. In the last decade there has been a great demand from consumers for whole grains because of their high nutritional value, antioxidant compounds and their health benefits against various diseases such as cardiovascular diseases, cancer (breast and prostate cancer), reduce the risk of diabetes (Broekaert et al. 2011).

Considering the nutritional value and associated health benefits of consuming cereal by-products the present study was planned to evaluate the phytochemicals compounds in different cereal by-products. In addition, four different solvents, distilled water, ethanol, methanol and acetone separately were used for the extraction of the bioactive compounds and its antioxidant activity. The concentrations of total phenolic compounds (TPC), total flavonoids content (TFC) and the capabilities of antioxidant activity; scavenging free radicals and ferric reducing power of cereal by-products were studied and compared with different solvent extracts separately.

Materials and methods

Raw materials

Cereals by-products such as wheat milling by-products (bran, shorts and germ) obtained from South Cairo & Giza Flour Mills, Al Haram, Giza, Egypt, Corn wet milling byproducts (bran, germ and germ meal) obtained from National Company for Maize Products, El Asher men Ramadan city, Sharqiyah, Egypt., Rice milling by-products (bran, husk and germ) obtained from Rice mills company, Sharqiyah, Egypt were used.

Chemicals and solvents

2,2-Diphenyl-1-picrylhydrazyl (DPPH), sodium carbonate, gallic acid, catechin, β -carotene, 2,4,6-tri(2-pyridyl)-s-triazine (TPTZ), 2,2'-azinobis(3-ethylbenothiazoline-6-sulfonic acid) diammonium salt (ABTS+), Folin–Ciocalteu's reagent, aluminium chloride, FeSO₄, FeCL₃·6H₂O, sodium nitrite were purchased from Sigma–Aldrich Chemical Co., Denmark. All other chemicals and solvents used were of analytical grade.

Samples preparation

Cereals by-products were ground using a coffee grinder then sieved through a 50-mesh sieve; to remove broken pieces of cereal by-products and impurities. Finally, mixed homogenously, stored in tight polyethylene bags and kept at -20 °C until analysis.

Determination of phytochemicals profiles

Extraction and determination of total phenolics content (*TPC*) *and total flavonoids content* (*TFC*)

Methods of extraction Distilled water, ethanol, methanol and acetone separately were used to extract bioactive phytochemicals from cereal by-products according to Dent et al. (2013) with some modification. Recovered extracts were evaluated in terms of total phenolic compounds, total flavonoids content, and antioxidant properties. In brief, an amount of 5 g of cereal by-products (in triplicate) was extracted with 100 ml of distilled water, ethanol, methanol and acetone, respectively, in separate Erlenmeyer flasks and kept for extraction in a water bath at 50 °C. After 60 min, the supernatants were separated by filtration using Whatman filter paper no. 1.

Total phenolic content (TPC) Total phenolic compounds of various solvent extracts were determined colorimetrically using Folin–Ciocalteau reagent according to the method described by Adom et al. (2005); using Spectramax i3x multi-mode microplate reader system (Molecular Devices, Wokingham, UK) spectrophotometer at 760 nm against the blank. Gallic acid was used as a standard for quantification; expressed as milligram (mg) gallic acid equivalents (GAE) per 100 g of cereal milling by-products.

Total flavonoids content (TFC) The determination of total flavonoid content of various extracts was carried out using Liu et al. (2002) method. Total flavonoid content was calculated as milligram (mg) of catechin equivalent (CE) per 100 g of sample against a standard curve of catechin.

Determination of carotenoids

The carotenoids were assessed on cereal by-products using AACC-approved method 14–50 (AACC 2000). Briefly, A saturated mixture of *n*-butanol and distilled water (8:2 ratio) was used for carotenoids extraction. Ten mL of water-saturated butyl alcohol was added to 1 g of different cereal milling by-products, shaken and extracted for 16 h. Extracts were then filtered through Whatman No. 1 filter paper, and absorbance measured at 440 nm using a Spectramax i3x (Molecular Devices, Wokingham, UK) spectrophotometer. A calibration curve was made from pure β -carotene. The carotenoids content was expressed as $\mu g \beta$ -carotene/g samples.

Determination of tannins content

Quantitative determination of tannins were carried out as described by Price et al. (1978) followed with minor modification. The formed color was measured at 500 nm by using spectrophotometer (UV-1280 spectrophotometer, Japan). Catechin was used to prepare the standard curve. Tannins were calculated as mg/100 g on a dry weight basis.

Determination of antioxidant activity

2,2-Diphenyl-1-picrylhydrazyl (DPPH) assay

The antioxidant activity was determined using the method described by (Yu et al. 2002). The absorbance of the various extracts and control was measured at 515 nm. The lower absorbance of the reaction mixture indicated higher free radical scavenging activity.

Antioxidant activity was calculated as a percentage of radical scavenging activity (% RSA) using the following equation:

 $\text{\%RSA} = [(A_0 - A_1 / A_0)] \times 100$

 ${A_0 = \text{the absorbance of the control reaction}}$ (containing all reagents except the test compounds)

and A_1 = the absorbance in the presence of

the tested extracts after 30 min}.

$ABTS \bullet^+$ scavenging assay

The scavenging activity using 2,2-asino-bis (3-ethylbenzothiazoline-6-sulphonic acid) (ABTS \bullet^+) was measured by the method described by Li et al. (2015). The radical scavenging activity (% RSA) of the samples was calculated as follows:- $RSA\% = (1-A/A_0) \times 100.$ (A₀ is the absorbance at 734 nm of the negative control; A is the absorbance at 734 nm of the mixture with the sample).

Ferric-reducing antioxidant power (FRAP) assay

FRAP assay was carried out according to the literature (Benzie and strain 1996). The absorbance of the reaction mixture was detected at 593 nm. The standard curve was constructed using FeSO₄ solution, and the results were expressed as mmol L^{-1} FeSO₄ g^{-1} dry weight of cereal milling by-products.

Statistical analysis

All the analyses were carried out in triplicates and represented as mean \pm SD. The statistical analysis was performed using SPSS software (version 24.0, USA). One way ANOVA was applied following Tukey's post hoc test to compare the mean and differences were considered as significant when P < 0.05.

Results and discussion

Total phenolics content (TPC)

Total phenolics content (TPC) of cereal milling by-products extracted with various solvents are shown in Table 1.

Results of the assays for phenolics indicated a wide variation in TPC in the different extracts of cereal milling by-products. The results in Table 1 showed that the highest amount of TPC was obtained by using distillate water for wheat milling by-products (germ, bran and shorts 1861.9, 844.7 and 698.2 mg GAE/100 g, respectively) and corn bran (1925 mg GAE/100 g) followed by methanol, ethanol and acetone. These results were similar to those obtained by Liu et al. (2017) who stated that total phenolics content of defatted wheat germ was 10.55 mg GAE/g. On the other hand, the lowest amount of TPC was achieved by using acetone. In this respect, Zhao et al. (2017) reported that total phenolics content of wheat bran as a gallic acid was 4206.16 µg/g. Ciccoritti et al. (2017) reported different wheat bran fractions had total phenolics ranged from 5698.8 to 6820.3 mg Ferulic acid/kg dw. Also, Thakur et al. (2017) reported that corn grit containing a high amount of bound phenolics such as gallic acid, ferulic acid, and sinapic acid would be beneficial for human health.

The results revealed that TPC of rice milling by-products (husk, germ and bran) and corn milling by-products with ethanol were significantly high compared to methanol,

Table 1Total phenolicscontent (TPC) of cereal millingby-products (dw)

Cereal milling by-products	IPC (mg GAE/100 g dw)						
	Distilled water	Ethanol	Methanol	Acetone			
Wheat milling by-products							
Bran	$844.7\pm5.7^{\rm a}$	$566.6\pm8.5^{\rm c}$	630.0 ± 15.1^{b}	430.6 ± 4.2^{d}			
Germ	1861.9 ± 22.9^{a}	$1351.2 \pm 42.5^{\circ}$	1704.5 ± 12.2^{b}	987.7 ± 15.4^{d}			
Shorts	698.2 ± 11.5^a	423.1 ± 10.4^{c}	500.5 ± 16.3^{b}	366.1 ± 29.3^{d}			
Rice milling by-products							
Bran	1084.8 ± 12.2^{c}	1335.9 ± 36.5^{a}	1176.9 ± 57.8^{b}	953.6 ± 7.4^{d}			
Germ	$652.7 \pm 23.81^{\circ}$	1012.2 ± 59.4^{a}	$865.4 \pm 15.8^{\text{b}}$	555.5 ± 18.6^{d}			
Husk	$845.7\pm20.0^{\rm c}$	1060.0 ± 26.5^{a}	933.4 ± 25.6^{b}	776.9 ± 22.8^{d}			
Corn milling by-products							
Bran	1925.0 ± 16.3^{a}	1779.5 ± 17.43^{b}	1814.0 ± 7.8^{b}	$1538.0 \pm 82.8^{\circ}$			
Germ	691.1 ± 6.2^{c}	1235.1 ± 21.5^a	1182.7 ± 3.9^{b}	676.6 ± 14.6^{d}			
Germ meal	$932.8 \pm 11.2^{\circ}$	1356.4 ± 11.9^{a}	1158.0 ± 12.5^{b}	638.1 ± 7.1^{d}			

a . . .

TPC total phenolics content, dw dry weight, GAE gallic acid equivalent

ma

Means in the same row with different letters are significantly different (P < 0.05)

distillate water and acetone, respectively. The polyphenols extraction from plants have received great interest due their chemical diversity and interaction with other food molecules (Kosar et al. 2005). Therefore, studies have been conducted to evaluate the efficacy of various solvents such as methanol and acetone in the extraction of polyphenols (Złotek et al. 2016). Sun et al. (2006) mentioned that methanol was the best solvent for extracting the phenolic compound from oat bran. They have also reported that the amount of phenols extracted by methanol was higher than those extracted with acetone and hexane (three and four folds, respectively). In this regards Do et al. (2014) reported that ethanol more efficiency in extraction of TPC may be caused by some phenolic compounds (have higher molecular weights) are soluble in ethanol than the phenolics in the water. Indeed, the choice of solvents depends mainly on the polarity of target compounds, safety and cost (Yu et al. 2002).

Total flavonoids content (TFC)

Total flavonoids content (TFC) of different cereal milling by-products were extracted with various solvents and the results are presented in Table 2.

The data in Table 2 revealed that the TFC of cereal milling by-products extracted by using different solvents (P < 0.05). The results also showed that TFC of cereal milling by-products (wheat milling by-products, rice milling by-products and corn milling by-products) extracted by acetone had the highest values followed by ethanol and methanol. On the other hand, distilled water had the less efficiency for TFC extraction. The results also showed that TFC of different cereal milling by-products extracted by

distillate water was ranged from 139.3 to 316.8 mg/100 g, while the TFC of acetone extracts was varied from 262.4 to 681.6 mg/100 g.

Results showed in Table 2 were similar to those reported by Iloki-Assanga et al. (2015) who suggested that acetone was more efficient in extracting of TFC than methanol and ethanol from *Bucid abuceras*. In this respect, Abozed et al. (2014) noticed that TFC of *Gemiza-9* bran had no significant difference between 70% methanol and 50% acetone extracts. Nevertheless, the other extract (Beni-suef-3bran) showed that 50% acetone containing a higher level of total flavonoid content than methanol.

Carotenoids

The total Carotenoids of cereal milling by-products was extracted by *n*-butanol and expressed as β -carotene and the results are presented in Fig. 1.

The present study shows that there was statistically significant different (P < 0.05) between cereal milling byproducts in carotenoids. The results also revealed that corn germ meal showed the highest content of carotenoids (57.9 µg/g) followed by corn bran (32.0 µg/g). On the other hand, wheat bran showed the lowest value of carotenoids (4.2 µg/g). Hentschel et al. (2002) stated that pigments (yellow color compounds) of wheat varieties were concentrated in the outer layers than in the inner layers. In this respect, Ndolo and Beta (2013) reported that the total carotenoids content of wheat germ (9.11 µg/g) significantly higher than endosperm (1.92 µg/g) and bran (0.74 µg/g).

The obtained results are also was higher than those obtained by Al-Okbi et al. (2014) who found that the β -

Table 2 Total flavonoidscontent (TFC) of cereal milling

by-products (dw)

Cereal milling by-products	TFC (mg CE/100 g dw)						
	Distilled water Ethanol		Methanol	Acetone			
Wheat milling by-products							
Bran	$163.6\pm8.6^{\rm c}$	276.7 ± 12.4^{b}	$160.6 \pm 7.5^{\circ}$	330.7 ± 12.7^{a}			
Germ	$316.8\pm7.9^{\rm d}$	500.5 ± 3.6^{b}	$338.6 \pm 12.4^{\circ}$	681.6 ± 6.6^a			
Shorts	$139.3 \pm 1.2^{\circ}$	257.3 ± 9.5^a	154.4 ± 8.4^{b}	262.4 ± 2.6^a			
Rice milling by-products							
Bran	$272.4 \pm 17.6^{\circ}$	530.1 ± 12.4^{b}	$310.4 \pm 13.9^{\circ}$	673.5 ± 13.4^{a}			
Germ	180.3 ± 16.6^{d}	386.4 ± 8.0^{b}	$250.8 \pm 13.4^{\circ}$	673.5 ± 13.4^{a}			
Husk	143.3 ± 1.5^{d}	$241.0 \pm 11.6^{\mathrm{b}}$	$217.2 \pm 1.5^{\circ}$	307.3 ± 15.5^{a}			
Corn milling by-products							
Bran	228.0 ± 5.8^{d}	539.0 ± 18.2^{b}	$451.0 \pm 17.4^{\circ}$	624.1 ± 14.2^{a}			
Germ	$257.6\pm3.8^{\rm c}$	$444.9 \pm 20.4^{\rm b}$	437.3 ± 22.5^{b}	546.4 ± 9.0^a			
Germ meal	237.9 ± 12.3^{d}	$466.9 \pm 9.6^{\rm b}$	$378.3 \pm 10.1^{\circ}$	549.8 ± 6.1^{a}			

TFC total flavonoids content, dw dry weight, CE catechin equivalent

Means in the same row with different letters are significantly different (P < 0.05)



Fig. 1 Carotenoids content as β -Carotene of cereal milling byproducts (dw). The vertical bars represent the SD (n = 3), and values marked by the same letter are not significantly different (P < 0.05); *dw* dry weight

carotene of whole rice bran extracted by hexane was 101.0 μ g/100 g. Zhou et al. (2004) suggested that bran samples might significantly differ in their carotenoid profiles due to the wheat variety and growing condition. Panfili et al. (2004) reported that wheat germ contained total carotenoids about 5.5 μ g/g.

Tannins content

Tannins contents of different cereals milling by-products are presented in Fig. 2.

Dykes and Rooney (2006) reported that tannins have an antioxidant effect as well as other therapeutic properties such as lowering cholesterol, anticarcinogenic and cardiovascular.

The results in Fig. 2 revealed that corn germ meal contained the highest value (389.5 mg CE/100 g) followed





Fig. 2 Tannins content of cereal milling by-products (dw). The vertical bars represent the SD (n = 3), and values marked by the same letter are not significantly different (P < 0.05). dw dry weight, CE catechin equivalent

by corn germ and wheat bran (315.3 and 295.5 mg CE/ 100 g, respectively) when compared with than those of the rest of the tested cereal by-products. On the other hand, rice bran contained the lowest values (113.4 mg CE/100 g).

Antioxidant activity (AOA)

DPPH, ABTS+ radical scavenging activity and ferric reducing power of the various cereal milling extracts are illustrated in Table 3. The extracts of each cereal milling by-products were evaluated for their free radical scavenging properties, which are indicative of antioxidant capacities.

The results indicated that cereal milling by-products contained the high amounts of some phytochemicals (bioactive compounds). These bioactive compounds would have antioxidant activity as indicated in Table 3.

The data in Table 3, showed that antioxidant activity of different cereal milling by-products which determined by

Tal	ble	3	Antioxidant	activity	(AOA) of	cereal	milling	by-products
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Cereal by-products	Antioxidant activity	Distilled water	Ethanol	Methanol	Acetone
Wheat by-products					
Bran	AOA DPPH (% RSA)	39.3 ± 0.1^{b}	49.0 ± 0.5^{a}	$50.9\pm1.1^{\rm a}$	37.5 ± 1.7^{b}
	AOA ABTS (% RSA)	$32.4 \pm 1.5^{\circ}$	39.3 ± 0.7^a	40.6 ± 0.7^a	$23.3\pm0.7^{\rm d}$
	FRAP (mmol L ⁻¹ ferrous sulfate/g sample)	21.5 ± 0.5^{d}	$31.3\pm0.3^{\rm b}$	36.1 ± 0.4^{a}	$25.9\pm0.1^{\rm c}$
Germ	AOA DPPH (% RSA)	$43.9\pm0.8^{\rm c}$	$55.4\pm0.6^{\rm b}$	69.2 ± 3.4^{a}	40.6 ± 1.2^{c}
	AOA ABTS (% RSA)	88.3 ± 0.3^{a}	$53.1\pm1.6^{\rm b}$	88.5 ± 2.1^a	$43.6 \pm 1.1^{\circ}$
	FRAP (mmol L ⁻¹ ferrous sulfate/g sample)	$28.8\pm0.5^{\rm d}$	39.2 ± 0.1^{a}	$37.8\pm0.1^{\text{b}}$	$29.7\pm0.7^{\rm c}$
Shorts	AOA DPPH (% RSA)	$20.5\pm0.7^{\rm c}$	$23.7\pm0.6^{\rm b}$	27.5 ± 1.7^a	17.0 ± 0.6^{d}
	AOA ABTS (% RSA)	34.0 ± 1.3^{a}	$24.8\pm1.4^{\rm c}$	30.0 ± 1.9^{b}	$18.8\pm0.5^{\rm d}$
	FRAP (mmol L ⁻¹ ferrous sulfate/g sample)	$13.7 \pm 0.1^{\mathrm{b}}$	$24.2\pm0.4^{\rm b}$	28.4 ± 0.3^a	$21.8\pm0.3^{\rm c}$
Rice by-products					
Bran	AOA DPPH (% RSA)	63.7 ± 0.9^{a}	$59.0\pm2.0^{\rm b}$	65.1 ± 2.2^{a}	47.3 ± 1.9^{c}
	AOA ABTS (% RSA)	70.6 ± 2.0^{a}	43.6 ± 3.6^{b}	74.3 ± 3.1^a	$35.4 \pm 1.5^{\rm c}$
	FRAP (mmol L ⁻¹ ferrous sulfate/g sample)	$20.6\pm0.5^{\rm d}$	38.6 ± 0.2^{b}	40.3 ± 0.4^a	$35.0 \pm 0.1^{\circ}$
Germ	AOA DPPH (% RSA)	$43.9 \pm 1.02^{\circ}$	47.5 ± 0.5^{b}	48.7 ± 0.2^{a}	33.6 ± 0.2^{d}
	AOA ABTS (% RSA)	$69.0\pm0.7^{\rm a}$	$35.2\pm1.2^{\rm c}$	$55.7\pm2.1^{\rm b}$	32.6 ± 0.7^d
	FRAP (mmol L ⁻¹ ferrous sulfate/g sample)	19.1 ± 0.2^{d}	38.8 ± 0.3^{a}	36.1 ± 0.2^{b}	$34.0 \pm 0.8^{\circ}$
Husk	AOA DPPH (% RSA)	$50.2 \pm 0.3^{\mathrm{b}}$	$50.4\pm0.8^{\rm b}$	54.5 ± 0.2^a	34.3 ± 1.4^{c}
	AOA ABTS (% RSA)	$55.4\pm0.2^{\rm b}$	$43.5\pm1.1^{\rm c}$	62.6 ± 0.7^a	29.7 ± 1.4^{d}
	FRAP (mmol L ⁻¹ ferrous sulfate/g sample)	$23.5\pm0.3^{\rm c}$	$37.8\pm0.1^{\rm b}$	39.0 ± 1.1^{a}	$33.7\pm0.5^{\rm d}$
Corn milling by-prod	ucts				
Bran	AOA DPPH (% RSA)	60.5 ± 2.2^{b}	$70.0\pm0.9^{\rm a}$	$71.3\pm.7^a$	40.3 ± 1.4^{c}
	AOA ABTS (% RSA)	48.1 ± 2.1^{a}	32.8 ± 1.8^{b}	51.7 ± 2.7^{a}	26.1 ± 0.7^{c}
	FRAP (mmol L ⁻¹ ferrous sulfate/g sample)	$12.6\pm0.3^{\rm c}$	34.7 ± 0.2^a	$33.7\pm0.3^{\text{b}}$	$29.3\pm0.2^{\rm d}$
Germ	AOA DPPH (% RSA)	$37.4 \pm 1.0^{\circ}$	$49.8\pm0.8^{\rm b}$	52.1 ± 0.4^{a}	$29.8\pm0.8^{\rm d}$
	AOA ABTS (% RSA)	$30.1 \pm 0.8^{\rm b,c}$	$32.7\pm2.0^{\rm b}$	42.8 ± 1.8^a	$27.6\pm0.6^{\rm c}$
	FRAP (mmol L ⁻¹ ferrous sulfate/g sample)	$16.3 \pm 0.1^{\circ}$	37.3 ± 0.1^{a}	37.1 ± 0.2^{a}	$31.9 \pm 2.0^{\mathrm{b}}$
Germ Meal	AOA DPPH (% RSA)	$41.2 \pm 1.3^{\circ}$	$53.2\pm0.5^{\rm b}$	$59.1\pm0.6^{\rm a}$	31.8 ± 0.8^{d}
	AOA ABTS (% RSA)	$28.9 \pm 1.3^{\rm c}$	$45.4\pm1.8^{\rm b}$	59.5 ± 2.0^a	$25.7\pm0.8^{\rm d}$
	FRAP (mmol L ⁻¹ ferrous sulfate/g sample)	15.2 ± 0.1^{d}	37.1 ± 0.3^{b}	38.2 ± 0.2^a	$32.6 \pm 1.2^{\rm c}$

AOA antioxidant activity, DPPH 2,2-Diphenyl-1-picrylhydrazyl, ABTS 2,2'-Azino-bis(3-ethylbenzthiazoline-6-sulphonic acid), FRAP ferric reducing antioxidant power, RSA radical scavenging activity

Means in the same row with different letters are significantly different (P < 0.05)

different Antioxidant Activity (AOA) methods such as DPPH, ABTS+ and FRAP were significantly different according to different solvent extracts (P < 0.05).

Wheat by-products results showed that germ had higher AOA compared to bran and shorts. It was also found that AOA of wheat germ ranged from 40.6 to 69.2% RSA as DPPH assay, 43.6 to 88.5% RSA as ABTS assay and 28.8 to 39.2 mmol L^{-1} sulfate/g samples as FRAP assay. These results are in the line of Li et al. (2015) who mentioned that the FRAP assay results of some wheat flour varieties were ranged from 31.7 to 42.3 mmol L^{-1} FeSO4 g⁻¹. Liyana-Pathirana and Shahidi (2007) reported that the biologically active substances and antioxidant activity compounds were

concentrated in the outer layers of grains, such as bran and shorts, compared to the inner layers (endosperm).

The results in Table 3 showed that AOA of wheat bran was ranged from 37.5 to 50.9% RSA as DPPH assay, 23.3 to 40.6% RSA as ABTS assay and 21.5 to 36.1 mmol L⁻¹ sulfate/g samples as FRAP assay. These results are closed to those obtained by Abozed et al. (2014) who reported that both bran varieties of wheat Gemiza-9 and Beni-suef-3 showed strong DPPH free radical scavenging activities (28.1–31.8 and 33.8–36.8% RSA, respectively). Previous studies have used solvents of different polarity and mixture of water such as acetone 50% (Zhou and Yu 2004b); methanol 80% (Kim et al. 2006) and concluded that polar solvents are more effective in extracting of hydrophilic antioxidants compounds.

The results revealed that rice bran had the highest value of TPC and TFC and had antioxidant activity ranged from 47.3 to 65.1% RSA as DPPH assay, 35.4 to 74.3% RSA as ABTS+ assay and 20.6 to 40.3 mmol L^{-1} ferrous sulfate/g as FRAP assay followed by rice husk which had AOA 34.3 to 54.5% RSA, 29.7 to 62.6% RSA and 23.5 to 39.0 mmol L^{-1} ferrous sulfate/g as DPPH, ABTS+ and FRAP, respectively. In a food product, antioxidant plays an important role in extending the shelf life and improving the nutritional quality. This means the significant influence of the polarity solvent on antioxidant activity observed for different rice milling by-products.

Concerning corn wet milling by-products; i.e. bran, germ and germ meal, the results revealed that AOA extracts by various solvents were ranged from 37.4 to 60.5, 49.8 to 70.0, 52.1 to 71.3 and 29.8 to 40.3% RSA as DPPH assay for distillate water, ethanol, methanol and acetone, respectively. On the other hand, FRAP assay for each extract was ranged from 12.6 to 16.3, 34.7 to 37.3, 33.7 to 38.2 and 29.3 to 32.6 mmol L^{-1} ferrous sulfate/g for distillate water, ethanol, methanol and acetone, respectively. This indicated that ethanol and methanol were more efficient in the extraction of bioactive compounds of high antioxidant activity. The phytochemical profile and antioxidant activity of wheat and wheat milling by-products were different owing to the different type of solvent used. The low molecular weight phenolic compounds from cereal milling by-products were rapidly extracted by methanol and ethanol (Sun et al. 2006).

In general, distilled water, ethanol and methanol extracts were significantly different in antioxidant activity. The DPPH radical scavenging activity for the different cereal by-products extracts was varied from 17.0 to 71.3% RSA, while cereal by-products extracts showed ABTS⁺ scavenging activities varied from 18.8 to 88.5% inhibition. These data suggested that ethanol followed by methanol and distillate water were the best solvents to extract phenolic compounds and antioxidant activity. Therefore, ethanol and methanol might be a better choice for extracting antioxidants compounds from cereal milling byproducts samples. Due to the polyphenol's chemical diversity and its interaction with other matrix constituents, extraction from different plant sources are complex (Masisi et al. 2016). Since antioxidant capacity is mainly linked to the concentration of these compounds, it could also be affected as a property of the final extract mixture (Stefanello et al. 2018).

These outcomes demonstrated that high polarity solvents (ethanol and methanol) were more effective for extraction of antioxidants with efficient free radical scavenging properties versus an intermediate-polarity solvent (acetone). In the same line, Osman et al. (2009) reported that DPPH and ABTS activities varied with high levels of total phenols and total flavonoids and exhibited high free radical scavenging activity. In this respect, Saharan et al. (2017) revealed that a positive correlation was obtained between total phenol and flavonoid content with antioxidant activity. In this study, significant differences in free radical scavenging ability of extracts of different cereal milling by-products were observed depending on the kind of extraction solvent used.

Conclusion

In this study, antioxidant activities, total phenolics, flavonoids, tannins and carotenoids contents of wheat, rice and corn milling by-products were quantified, aiming to evaluate the different fractions. The research was tried to establish the relationship between the amount of bioactive compounds and the different solvents used for the extraction. In addition, the results indicate that TPC and antioxidant activities of wheat, rice and corn milling by-products achieved by ethanol were significantly high compared to methanol then distillate water. Also, the lowest amount of TPC was achieved by using acetone. The study, showing that corn milling by-products to be a rich source of carotenoids compared with wheat and rice milling by-products. These results could provide useful information for further studies and applications of cereal milling by-products for functional food or related products.

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Compliance with ethical standards

Conflict of interest The authors declare no conflict of interest.

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