# **Kinetic Analysis of Phospholipase C from Catharanthus roseus Transformed Roots Using Different Assays<sup>1</sup>**

# S.M. Teresa Hernández-Sotomayor\*, César De Los Santos-Briones, J. Armando Muñoz-Sánchez, **and Victor M. Loyola-Vargas**

Unidad de Biología Experimental, Centro de Investigación Científica de Yucatán, Apartado Postal 87 Cordemex 97310, Merida, Yucatan, Mexico

**The properties of phospholipase C (PLC) partially purified from Catharanthus roseus transformed roots were analyzed using substrate lipids dispersed in phospholipid vesicles, phospholipiddetergent mixed micelles, and phospholipid monolayers spread at an air-water interface. Using [33P]phosphatidylinositol 4,5** bisphosphate (PIP<sub>2</sub>) of high specific radioactivity, PLC activity was **monitored directly by measuring the loss of radioactivity from monolayers as a result of the release of inositol phosphate and its subsequent dissolution on quenching in the subphase. PLC activity was markedly affected by the surface pressure of the monolayer, with reduced activity at extremes of initial pressure. The optimum surface pressure for PIP2 hydrolysis was 20 mN/m. Depletion of PLC** from solution by incubation with sucrose-loaded PIP<sub>2</sub> vesicles fol**lowed by ultracentrifugation demonstrated stable attachment of PLC to the vesicles. A mixed micellar system was established to assay PLC activity using deoxycholate. Kinetic analyses were performed to determine whether PLC activity was dependent on both** bulk PIP<sub>2</sub> and PIP<sub>2</sub> surface concentrations in the micelles. The **interfacial Michaelis constant was calculated to be 0.0518 mol fraction, and the equilibrium dissociation constant of PLC for the** lipid was 45.5  $\mu$ *M*. These findings will add to our understanding of **the mechanisms of regulation of plant PLC.**

 $Ca<sup>2+</sup>$  is the principal second messenger in plants. One of the mechanisms that regulates the levels of this messenger involves the enzyme PLC. PLC catalyzes the hydrolysis of  $PIP<sub>2</sub>$  to generate two second messengers:  $IP<sub>3</sub>$  and 1,2diacylglycerol. PLC is a family of isoenzymes that has been classified into three groups:  $\beta$ ,  $\gamma$ , and  $\delta$  (Rhee and Bae, 1997). In plants several studies have reported the biochemical presence of this enzyme (McMurray and Irvine, 1988; Tate et al., 1989; Melin et al., 1992; Pical et al., 1992; Yotsushima et al., 1992, 1993; Huang et al., 1995; De Los Santos-Briones et al., 1997). Different genes for PLC have also been cloned (Hirayama et al., 1995, 1997; Shi et al., 1995; Yamamoto et al., 1995; Kopka et al., 1998), all of them resembling the  $\delta$  type. However, the molecular basis for activation of plant PLC isoforms is not clear, nor is it completely understood how PLCs interact with their membrane substrates.

Data concerning the manner in which phospholipases interact with lipid substrates are accumulating, and show general similarities but also some remarkable differences between the different isoforms. A variety of assay procedures have been used to measure lipase activity. Most involve presentation of enzymes with pre-aggregated lipids, including phospholipid vesicles, phospholipiddetergent mixed micelles, pre-immobilized and crosslinked lipids, and phospholipid monolayers. The activity of all PLC isoforms studied to date is affected by the surface pressure of monolayer substrates (Rebecchi et al., 1992; James et al., 1994). PLCs act on substrates that form a lipid-water interface, an arrangement that complicates kinetic analyses of the enzyme catalytic activity. To our knowledge, no investigation has been reported thus far on the kinetics of plant PLCs using different assays. Such studies would provide information if PLC binds the substrate in a noncatalytic manner as a prerequisite to anchoring into the membrane to start catalysis, or if the PLC activity from plant sources is affected when it is measured in monolayers. We now report kinetic analyses, vesiclebinding studies, and monolayer assays using a semipurified enzyme from *C. roseus* transformed roots.

#### **MATERIALS AND METHODS**

#### **Tissue Culture**

Hairy root line J1 of *Cantharanthus roseus* was obtained by infection of leaves with *Agrobacterium rhizogenes* (Ciau-Uitz et al., 1994) and maintained in B5 medium (Gamborg et al., 1968) supplemented with 30 g/L Suc. The pH was adjusted to 5.7 prior to autoclaving of the medium with 0.1 m KOH/HCl. One hundred milliliters of medium was placed in a 250-mL Erlenmeyer flask and autoclaved for 20 min at 15 p.s.i. Flasks were inoculated with 0.5 g (fresh mass) of hairy roots. Roots were subcultured every 14 d. Cultures were grown in darkness at 25°C on a rotary

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<sup>\*</sup> Corresponding author; e-mail ths@cicy.mx; fax 99–81–3900.

Abbreviations: IP<sub>3</sub>, inositol 1,4,5-trisphosphate; PC, dioleoylphosphatidylcholine; PIP, phosphatidylinositol monophosphate; PIP<sub>2</sub>, phosphatidylinositol 4,5-bisphosphate; PLC, phospholipase C; PS, phosphatidylserine.

# **Preparation of Tissue and Cell Extracts**

Roots were quickly frozen with liquid nitrogen and homogenized with a polytron in buffer A (1 g of tissue in 2.5 mL of 50 mm NaCl, 1 mm EGTA, 50 mm Tris-HCl, pH 7.4, 250 mm Suc, 10% [v/v] glycerol, 1 mm PMSF, 10 mm sodium pyrophosphate, 0.2 mm orthovanadate, and 1 mm b-mercaptoethanol). Extracts were passed through gauze, and tissue debris were removed by centrifugation at 14,000*g* for 30 min at 4°C. The supernatant was further centrifuged at 100,000*g* for 45 min. The supernatants (protein: 3.5–5.0 mg/mL) was recovered as the soluble fraction. The pellet was resuspended in the same buffer A (protein: 0.5–1.2 mg/mL), and was used as a crude membrane fraction. All steps during the extraction were performed at 4°C. The cell extracts were quickly frozen in liquid nitrogen and stored at  $-75^{\circ}$ C. Protein concentrations of samples were measured with bicinchoninic acid protein assay reagent using BSA as a standard (Smith et al., 1985).

## **Partial Purification of C. roseus PLC**

A method has been developed for the rapid partial purification of *C. roseus* membrane-associated PLC, and has been used on at least six different occasions with similar results. Twenty grams of fresh roots from the 6th d of culture were extracted as mentioned above. The microsomal crude fraction was resuspended in buffer A in the presence of 2 m KCl. This suspension was sonicated for 3 min. After this, the suspension was centrifuged at 100,000*g* for 45 min at 4°C. The clarified solution was desalted in a Sephadex G-25 column (Pharmacia) pre-equilibrated with buffer B (20 mm Tris, pH 7.4, 1 mm EDTA, 1 mm DTT, 0.6 mm PMSF, and 2  $\mu$ g/mL leupeptine) at a flow rate of 2 mL/min. The fractions with PLC activity were applied to a heparin-Sepharose CL-6B column (Pharmacia;  $1.6 \times 17$  cm) pre-equilibrated in buffer C (20 mm  $K_2HPO_4$ , pH 7.3, 1 mm EDTA, 1 mm DTT, 0.6 mm PMSF, and 2  $\mu$ g/mL leupeptine). The column was developed with a 250-mL linear gradient of 0 to 1.5 m KCl in buffer C and 5-mL aliquots were collected. PLC activity was eluted at 0.6 m KCl. PLC was stored in 20% (v/v) glycerol at  $-70^{\circ}$ C.

## **PLC Assay**

The hydrolysis of  $[^{3}H]$ PIP<sub>2</sub> was measured as described by Hernández-Sotomayor and Carpenter (1993) and De Los Santos-Briones et al. (1997) in a reaction mixture (50  $\mu$ L) that contained 35 mm  $\text{NaH}_2\text{PO}_4$ , pH 6.8, 70 mm KCl, 0.8 mm EGTA, 0.8 mm CaCl<sub>2</sub> (final Ca<sup>2+</sup> concentration, 25  $\mu$ m), 200  $\mu$ M PIP<sub>2</sub> (approximately 20,000 cpm), and 0.08% deoxycholate. The reaction was stopped with 100  $\mu$ L of 1% (w/v) BSA and 250  $\mu$ L of 10% (w/v) TCA. Precipitates were removed by centrifugation (13,000*g* for 10 min) and the supernatants were collected for quantification of  $[^3H]IP_3$  released by liquid scintillation counting using Aquasol (DuPont-New England Nuclear, Boston).

# **Monolayer Assays**

Monolayer assays were performed in a monolayer trough (16-mL subphase volume) supplied by Nima Technology (Coventry, UK). Monolayer surface pressure was measured continuously using filter paper Wihelmy plates suspended from an electronic microbalance. Surface radioactivity was monitored continuously using a remote detector (model FC-006, Bioscan, Washington, DC) suspended 0.5 cm above the monolayer, coupled to computer software (LabChrom v2.10, Lab Logic, Sheffield, UK) that recorded the data. The subphase buffer was composed of 35 mm NaH<sub>2</sub>PO<sub>4</sub> (pH 6.8), 70 mm KCl, 0.8 mm EGTA, and 0.8 mm CaCl<sub>2</sub> (final Ca<sup>2+</sup> concentration, 25  $\mu$ M, when added). This buffer was stirred with a Teflon-coated stirrer, and composite lipid monolayers (70% [mol] phosphatidylcoline: 27% PS:3%  $[33P]$ PIP<sub>2</sub>) were spread at the surface. An aliquot of 0.7 mL of the subphase was replaced with 0.7 mL of enzyme preparation (20–30  $\mu$ g of protein) only after a stable monolayer with a constant pressure had formed. After 5 min,  $Ca^{2+}$  ions were added to the subphase to start catalysis and the reaction was either monitored continuously  $(^{33}P$ -labeled monolayer) or 0.5-mL aliquots were taken at different times  $(^{3}H$ -labeled lipids).

# **Preparation of**  $[^{33}P]$ **PIP<sub>2</sub>**

[<sup>33</sup>P]PIP<sub>2</sub> was prepared using partially purified PIP kinase (James et al., 1994). PIP (200  $\mu$ M) was sonicated in PIP kinase buffer (200 mm Hepes, pH 7.4, 40 mm  $MgCl<sub>2</sub>$ , 4 mm EGTA, 4 mm DTT, and 400 mm NaCl) and  $[^{33}P]$ ATP (specific radioactivity  $> 1,000$  Ci/mmol) with no unlabeled ATP was added to this substrate suspension. The reaction was initiated with the PIP-kinase preparation and incubated at 37°C for 20 min. The reaction was terminated with 750  $\mu$ L of chloroform:methanol:concentrated HCl (40:80:1, v/v). Radiolabeled  $[^{33}P]PIP_2$  (specific activity > 1,000 Ci/ mmol) was purified by HPLC using an amino-cyano analytical column pre-equilibrated with chloroform:methanol: water (20:9:1, v/v).

#### **Vesicle Binding**

Large unilamellar vesicles were produced by extrusion through 100-nm polycarbonate membranes using a phospholipid extruder (Lipex Biomembranes, Vancouver) according to the manufacturer's instructions. Fifteen milligrams of lipid (PC:PS:PIP<sub>2</sub> [70:27:3 by molarity]) was dried to a film, resuspended by vortexing in 10 mm Hepes, pH 7.4, 200 mm Suc, 3.4 mm EDTA, and 20 mm KCl, and treated with repeated freeze-thawing in a liquid nitrogen-40°C water bath. Lipids were extruded with at least 10 passes through the polycarbonate filters, and large unilamellar vesicles were stored at 4°C. For PLC-binding studies, vesicles were diluted to 100  $\mu$ m with respect to PIP<sub>2</sub> in buffer D (10 mm Hepes, pH 7.4, 3.4 mm EDTA, and 150 mm NaCl), and used as a stock for all lower concentrations required in the binding studies.

Binding was performed in 100  $\mu$ L of buffer D with 10  $\mu$ g of *C. roseus* PLC per tube. Ca<sup>2+</sup> was omitted to eliminate PLC-catalyzed PIP<sub>2</sub> hydrolysis. Enzyme was incubated with vesicles for 10 min on ice followed by ultracentrifugation at 60,000 rpm for 30 min at 4°C (TLA rotor and TL100 centrifuge, Beckman), and PLC activity remaining in the supernatant was assayed against 100  $\mu$ M PIP<sub>2</sub> as described above.

### **Data Presentation**

All experiments were repeated at least three times using extracts prepared on separate occasions, and all gave similar results. Each figure contains data from a single, representative experiment assayed in duplicate and the errors varied by less than 10%.

#### **Analysis of Kinetic Data**

The kinetic analysis of the PLC activity from *C. roseus* was based on the conditions previously described for secretory PLA2 (phospholipase A2; Hendrickson and Dennis, 1984). The binding of the interface was dependent on the concentration of the substrate (bulk  $\text{PIP}_2$  concentration), while the binding into the interface was dependent on the substrate mole fraction (case II). All the data were adjusted to the Hill equation (Eq. 1) using the GraFit program (Erithacus Software, Middlesex, UK).

$$
v = V_{\text{max}} \times [S]^n / K + [S]^n \tag{1}
$$

Case I

For the determination of the  $K<sub>s</sub>$  (binding to the interface), PLC activity was measured increasing concentrations of  $PIP<sub>2</sub>$  (bulk concentration) with a constant mole fraction of 0.052. To achieve this the concentrations of  $PIP<sub>2</sub>$  and deoxycholate were varied proportionally, keeping constant the mole fraction.

Equation 1 was reduced to the equation of Henri Michaelis-Menten (Eq. 2) in which the values of  $K_s$  were determined.

$$
v = V_{\text{max}} \times [S]/K_s + [S]
$$
 (2)

Case II

To determine the  $K_{\text{m}}$ , PLC activity was assayed keeping the concentration (100  $\mu$ m) constant and increasing the PIP<sub>2</sub> mole fraction by varying the concentrations of deoxycholate. A double-reciprocal Lineweaver-Burk approach was used. The interfacial  $K<sub>m</sub>$  was determined with the intercept to the ordinate  $([-1/K<sub>m</sub>]<sup>n</sup>)$ .

## Case III

To determine if there are several binding sites due to interactions with the lipid interface and the subsequent binding of the substrate to the interface, PLC activity was assayed using a fixed concentration of deoxycholate (1.92 mm) and increasing concentrations of PIP<sub>2</sub>.

### **Materials**

[<sup>3</sup>H]PIP<sub>2</sub> and Aquasol were purchased from DuPont-New England Nuclear, [<sup>33</sup>P]ATP was purchased from Amersham, and unlabeled  $PIP_2$  was purified from Folch (fraction I) extract of brain lipid (Sigma) by a neomycin affinity column as described in Waldo et al. (1994). The bicinchoninic acid protein assay reagent was supplied by Pierce, and B5 medium PC, PS, and neomycin were obtained from Sigma.

# **RESULTS**

#### **Monolayer Surface Pressure and PLC Activity**

To determine the molecular interaction of PLC with lipid interfaces and the reciprocal influences that they have on each other, PLC activity as a function of time was measured in monolayer assays (Fig. 1A). The rate of substrate hydro-



**Figure 1.** Surface pressure-PLC activity relationship for PIP<sub>2</sub>containing monolayers. A, Monolayers formed at an initial surface pressure of 20 mN/m and 20  $\mu$ g of partially purified PLC from C. roseus transformed roots. Ca<sup>2+</sup> (25  $\mu$ M free Ca<sup>2+</sup>) was added to the subphase after 5 min. Results show a smooth trace, representative of five experiments performed at that pressure. B, PLC activity determined against monolayers formed at increasing initial surface pressures. Reaction time was 20 min after the addition of  $Ca^{2+}$  and data are expressed as percentage of lipid hydrolyzed per minute. Data are means of at least five experiments for each pressure and varied by less than 5%.

lysis was followed for 20 min and  $PIP<sub>2</sub>$  hydrolysis was extended in monolayer assays. A typical time course for PLC-catalyzed hydrolysis of PC-PS-PIP<sub>2</sub> monolayers at an initial pressure of 20 mN/m and loss of radioactivity from monolayers due to  $\text{PIP}_2$  hydrolysis is shown in Figure 1A. Hydrolysis of PIP<sub>2</sub>-containing monolayers was absolutely dependent on the  $Ca^{2+}$  concentration of the subphase and all data presented were obtained using 25  $\mu$ M Ca<sup>2+</sup> (final concentration), which sustained the maximum rate of  $PIP<sub>2</sub>$ hydrolysis. Variations of the initial surface pressure of the monolayer resulted in markedly different rates of loss of radioactivity into the subphase.  $PIP<sub>2</sub>$  hydrolysis in monolayers was transformed into the percentage of  $\text{PIP}_2$  hydrolyzed per unit time and expressed against the initial surface pressure of the monolayer (Fig. 1B). The percentage of PIP<sub>2</sub> hydrolyzed per unit of time was calculated by measuring radioactivities remaining in the monolayer and present in the subphase after 20 min. In the early portion of the pressure-activity curve, PLC activity increased as surface pressure increased (Fig. 1B). A peak in PLC activity was seen at 20 mN/m but as pressure increased beyond this point, PLC activity was markedly reduced.

## **PLC-Lipid Binding Effect**

Lipid-metabolizing enzymes are considered to bind to and sometimes penetrate lipid interfaces, with subsequent further substrate binding within the interface as component parts of their catalytic mechanism. Therefore, the relationship between bulk substrate concentration and PLC binding was investigated using a noncatalytic vesicle binding assay as described previously for PLC  $\delta$  (Rebecchi et al., 1992). PLC was incubated with Suc-loaded large unilamellar vesicles with a phospholipid composition of  $PC:PS:PIP<sub>2</sub>$ (70:27:3 by molarity), and the activity of PLC in the supernatant was determined by assay after ultracentrifugation as described in "Materials and Methods." When incubated



**Figure 2.** Binding of C. roseus PLC to Suc-loaded vesicles. Ten micrograms of PLC was incubated with a range of concentrations of PIP<sub>2</sub>-containing vesicles (PIP<sub>2</sub>:PC:PS, 3:70:27 mol/mol), followed by ultracentrifugation as described in "Materials and Methods." PLC activity remaining in the supernatant was assayed against  $\text{PIP}_2$  and compared with vesicle-free controls in which the activity was 10 pmol/min.



Figure 3. PIP<sub>2</sub> hydrolysis by PLC from C. roseus as a function of both bulk concentration and the mole fraction of substrate. The concentration of deoxycholate was held constant at 1.92 mM, and the bulk PIP<sub>2</sub> concentration was increased up to 400  $\mu$ M. Assays were as stated in "Materials and Methods." Data are representative of four experiments with similar results.

with the vesicles composed of  $PC-PS-PIP_2$ , PLC was depleted from the supernatant and bound to the large unilamellar vesicle pellet in a manner that was dependent on the total vesicle  $\text{PIP}_2$  concentration (Fig. 2).

# **PLC Activity and Dependency on Bulk and Surface** Concentrations of PIP<sub>2</sub>

PIP2 hydrolysis by *C. roseus* PLC was analyzed according to the method of Hendrickson and Dennis (1984), as has been proposed for animal-PLC  $\delta$  and  $\gamma$  (Wahl et al., 1992; Cifuentes et al., 1993; Rebecchi et al., 1993). The activity was examined using the three-case kinetic analysis described in detail in "Materials and Methods," in which bulk and surface  $\text{PIP}_2$  concentrations were varied independently and concurrently as established for phospholipase A (Hendrickson and Dennis, 1984). Enzyme activity was measured



Figure 4. PLC activity toward PIP<sub>2</sub>-deoxycholate mixed micelles as a function of bulk  $\text{PIP}_2$  concentration. The  $\text{PIP}_2$  mole fraction was held constant 0.052 and bulk  $PIP<sub>2</sub>$  plus deoxycholate concentrations varied proportionally. Assays were performed as described in "Materials and Methods." Data are representative of four experiments.



Figure 5. PLC activity toward PIP<sub>2</sub>-deoxycholate mixed micelles as a function of PIP<sub>2</sub> surface concentration. Bulk PIP<sub>2</sub> concentration was held constant at 100  $\mu$ M with various deoxycholate concentrations to vary PIP<sub>2</sub> mole fraction. Other indications are as in Figures 3 and 4.

using PIP<sub>2</sub>-deoxycholate mixed micelles as substrate. Under the conditions used, deoxycholate behaved as a neutral diluent for  $\text{PIP}_2$ . To investigate whether PLC activity involves multiple binding events (a previous different binding site from the catalytic site) due to interactions with the bulk lipid interface and subsequent substrate binding within the interface, the bulk concentration and mole fraction of  $PIP_2$  were increased simultaneously and the  $IP_3$ production was measured with assays using a single concentration of deoxycholate and increasing concentrations of  $PIP<sub>2</sub>$ . The relationship between PLC activity and bulk and surface concentrations of substrate is presented in Figure 3. Interesting, it did not follow a sigmoidal relationship as described for mammalian PLC  $\beta$  (James et al., 1995). In our model, the behavior of the enzyme was Michaelian, giving a Hill coefficient of 0.968, and thus indicating a single binding site.

The  $K_s$  for interface binding by PLC was determined by assaying  $PIP<sub>2</sub>$  hydrolysis as a function of bulk concentration at a fixed mole fraction of  $\text{PIP}_2$  (0.052). This was achieved by varying both bulk  $\text{PIP}_2$  and deoxycholate concentrations proportionately, while maintaining the mole fraction constant. PLC showed a hyperbolic relationship with  $\text{PIP}_2$  bulk concentration (Fig. 4).

PLC catalytic activity was also measured at a constant bulk concentration of  $PIP<sub>2</sub>$  and varied mole fraction. This was achieved by varying the concentration of deoxycholate alone. Experiments were performed at a bulk  $\text{PIP}_2$  concentration of 100  $\mu$ m. The relationship between enzyme velocity and substrate mole fraction appeared sigmoidal between 0 and 0.08 (Fig. 5). The constants calculated from this kinetic analysis are:  $K_{\rm s}$ , 45.5  $\mu$ m, interfacial  $K_{\rm m}$ , 0.05, and *V*<sub>max</sub>, 137.2 pmol/min.

## **DISCUSSION**

We previously reported that PLC activity is present in *C. roseus* transformed roots (De Los Santos-Briones et al.,

1997). We report here a kinetic study of the partially purified, membrane-associated enzyme using different approaches. Few reports regarding the biochemical characterization of plant PLC are available (Tate et al., 1989; Melin et al., 1992; Pical et al., 1992; Yotsushima et al., 1992, 1993; Hirayama et al., 1995; Huang et al., 1995; Shi et al., 1995; Kopka et al., 1998). To our knowledge, this is the first report in which different assays were used to characterize this enzyme in plants. It is not known if plant PLCs follow a similar mechanism of activation as that reported for PLC from other sources (Hendrickson and Dennis, 1984; Rebecchi et al., 1992, 1993; Wahl et al., 1992; James et al., 1994, 1995, 1997). Most of the PLC enzymatic activity from plant sources is found in the cytosol, but since the substrate of the enzyme is membrane associated, there is the possibility that the cytosolic enzyme under certain physiological circumstances may suffer a redistribution from the cytosol to the membrane.

One of the most characteristic and intriguing features of lipolytic enzymes is their activation by interfaces. To explain the pathways of lipolysis, several investigators have proposed a reversible enzyme adsorption to, or penetration into, the interfaces (Hendrickson and Dennis, 1984; Rebecchi et al., 1993; James et al., 1997). However, such a mechanism has not been demonstrated for plant PLCs. To address this question, we describe the activity of PLC using lipid substrates dispersed in phospholipid vesicles, phospholipid-detergent mixed micelles, and phospholipid monolayers spread at an air-water interface.

We used <sup>33</sup>P-labeled substrates to measure PLC activity directly, which showed that the rate and extent of  $PIP_2$ hydrolysis in a PC-PS composite monolayer are surface pressure dependent. Although increasing phospholipid mass in the monolayer increased surface pressure, this was not accompanied by a simple increase in enzyme activity throughout the pressure range investigated (Fig. 1B). The reduction in PLC activity, as initial monolayer pressures were increased above the optimum pressure  $(20 \text{ mN/m}$ , Fig. 1B), was presumably a result of a decrease in the ability of the enzyme to bind the substrate, since this phenomenon has to be very specific and is probably regulated by the PIP<sub>2</sub> concentration or another unknown mechanism. However, there were no pressure-induced changes in the rate of catalysis during the course of the experiments (Fig. 1A), which indicates that it was the initial surface pressure that was crucial in determining the subsequent penetration of PLC into the monolayers. It also indicates that the changes in activity were not due to changes in the  $PIP<sub>2</sub>$  composition of the monolayer after PLC started catalysis. The lower rate of catalysis of lipids at a lower initial surface pressure, which would be expected to permit relatively easy penetration of PLC, may be due to enzyme denaturation by unfolding at the monolayer.

PLC activity may be affected by the composition of the subphase in these assays. The subphase buffer was design to be a simplified intracellular-type buffer solution composed of KCl and NaH<sub>2</sub>PO<sub>4</sub>. The resultant free Ca<sup>2+</sup> concentration was determined in part by the ionic strength and the ionic composition of the solution, which was characterized in previous experiments in which the  $Ca^{2+}$  requirements of the enzyme were studied (De Los Santos-Briones et al., 1997). The pressure-activity relationship (low activity at pressure below 20 mN/m, reaching a maximum at 20 mN/m, and decreasing above 20 mN/m) for PLC activity against PIP<sub>2</sub>-containing monolayers presented here (see Fig. 1B) contrasts with that previously reported for PLC  $\delta$ (Rebecchi et al., 1992, 1993).

Our results (Fig. 1B) resemble those of PLC  $\beta$  (James et al., 1995, 1997), which is surprising since all of the genes cloned to date for plant PLC are of the  $\delta$  type. For the  $\delta$ isoform (Rebecchi et al., 1992), it was shown that PLC activity decreased linearly with increasing monolayer surface pressure, with maximum activity being observed at the lowest pressures investigated  $(15 \text{ mN/m})$ . The basis for the differences between PLC from *C. roseus* and PLC δ activity in monolayers is not clear, but it establishes the phenomenon that different isoforms are affected differently by the quality of the interfaces with which they interact. The data presented here with plant PLC, as well as data from other studies (Rebecchi et al., 1992, 1993; James et al., 1995, 1997), clearly show that  $\text{PIP}_2$  hydrolysis in monolayers is surface pressure dependent, which is consistent with some element of penetration of lipid interfaces by this family of enzymes.

Our results, together with previous studies using monolayer substrates in which PLC activity was inhibited as the surface pressure increased, suggest that PLCs must penetrate lipid aggregates in order to bind and hydrolyze their substrates. Such a model may seem unnecessary given that the phosphodiester bond in  $\text{PIP}_2$  is likely to be exposed in the aqueous environment at the surface of the membrane. We propose that this mode of action may facilitate catalysis by restricting diffusion of PLCs into the two-dimensional membrane. Indeed, this may be why a dual substrate mechanism is apparently conserved among a wide range of lipid-metabolizing enzymes.

The data regarding the vesicle binding assay imply association of plant PLC with membrane interfaces through the substrate,  $PIP_2$  (Fig. 2). For efficient PLC-catalyzed production of second messengers, PLC may bind to membrane interfaces in a PIP<sub>2</sub>-specific noncatalytic manner, and subsequent bindings or rearrangements occur within the interface that may help to form a stable anchorage of PLC at the membrane surface. This mechanism may also lead to a series of catalysis reactions whereby PLC could hydrolyze multiple  $PIP_2$  molecules before detaching from the interface. The binding of a  $\text{PIP}_2$  molecule to at least one site in PLC other than the active site is inherent in the above proposal. When PLC from *C. roseus* was incubated with Suc-loaded PC-PS vesicles lacking  $\text{PIP}_2$ , no measurable binding of the enzyme to the vesicles was exhibited (data not shown). These data strongly support a multisubstrate mechanism in which binding at the interface is a specific process requiring the presence of lipid substrate.

Another unexpected result is shown in Figure 3, in which the kinetic data were analyzed to see if there were multiple binding sites onto the interface and a subsequent binding to the lipid substrate inside the interface. Surprisingly, the curve was not sigmoidal as reported for most mammalian PLCs (Wahl et al., 1992; Rebecchi et al., 1993; James et al.,

1995); instead, the curve followed a Michaelis-Menten curve with a Hill coefficient close to 1, probably due to a single binding site. This suggests that plants are regulated in a completely different way from animals. The pleckstrin homology domain, which is found in a broad array of signaling proteins (including all animal PLC isoenzymes), has been suggested to be a sequence that associates proteins with membranes to function (Musacchio et al., 1993), based on evidence for the interaction of PLC  $\delta$ -pleckstrin homology domains with  $\text{PIP}_2$ . In animal PLC  $\delta$  enzyme, the amino-terminal region containing the pleckstrin homology domain was necessary for binding to phospholipid vesicles containing  $\text{PIP}_2$  (Cifuentes et al., 1993). These results suggest that  $PIP_2$  might be important for localizing proteins containing pleckstrin homology domains at the membrane surface. However, the pleckstrin homology domain has not to our knowledge been reported in the gene products cloned to date for plant PLC. Perhaps plant PLC first has to bind noncatalytically to  $PIP<sub>2</sub>$  at the same site where catalysis occurs.

PLCs, which are involved in signal transduction responses to cellular stimuli, are members of a diverse family of enzymes whose mode of interaction with lipid substrates is complex and only partly defined. In summary, we have described the establishment of a controlled monolayer system for studying the family of PLCs, which will permit further investigations into different aspects of the interaction of plant PLCs with their substrates and their regulation. In this study, we have examined the kinetic characteristics of PIP<sub>2</sub> hydrolysis by partially purified plant PLC in the absence of their physiological activators. Although it has been proposed that  $Ca^{2+}$  may act as a regulator for plant PLCs, this has never been demonstrated. The data presented here help to establish a basic understanding of how this enzyme behaves toward lipid-water interfaces from which physiologically relevant mechanisms of regulation may eventually be discernible.

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