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Pharmacokinetics of [¹⁴C]-Benzo[a]pyrene (BaP) in Humans: Impact of Co-Administration of Smoked Salmon and BaP Dietary Restriction

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Abstract

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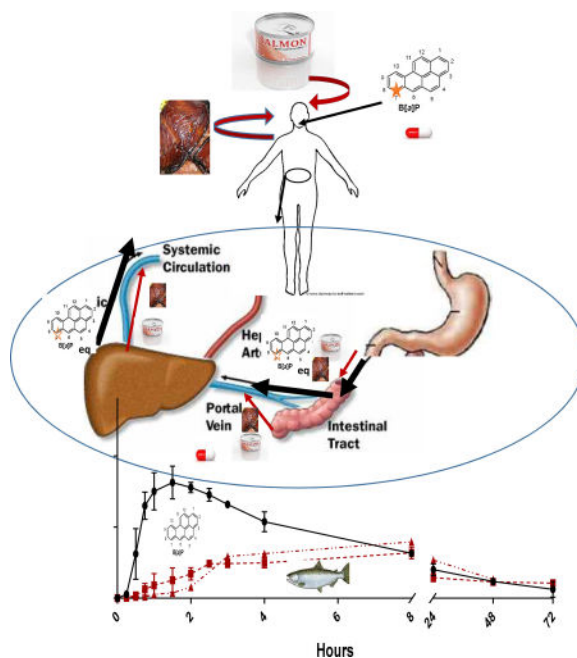
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Authors' contributions

JMH, EPM, KAA, KWT, JNS, SCT, WMB and DEW all contributed to the conceptual design of this study. LKS, SLU, SKK and DEW were responsible for all certifications and approvals associated with human use. JMH, EPM, LKS, TJM, KAA, TJO, SH, and JNS all contributed to sample collections, preparations and analytical assessments. Data analysis was primarily conducted by JMH, EPM, KAA, GB, JNS and SCT. The initial drafts of this manuscript were done by JMH, EPM and DEW. All coauthors contributed to revisions of drafts and approval of the final submission.

Benzo[a]pyrene (BaP), a polycyclic aromatic hydrocarbon (PAH), is a known human carcinogen. In non-smoking adults greater than 95% of BaP exposure is through diet. The carcinogenicity of BaP is utilized by the U.S. EPA to assess relative potency of complex PAH mixtures. PAH relative potency factors (RPFs, BaP=1) are determined from high dose animal data. We employed accelerator mass spectrometry (AMS) to determine pharmacokinetics of [^{14}C]-BaP in humans following dosing with 46 ng (an order of magnitude lower than human dietary daily exposure and million-fold lower than animal cancer models). To assess the impact of co-administration of food with a complex PAH mixture, humans were dosed with 46 ng of [^{14}C]-BaP with or without smoked salmon. Subjects were asked to avoid high BaP-containing diets and a 3-day dietary questionnaire given to assess dietary exposure prior to dosing and three days post-dosing with [^{14}C]-BaP. Co-administration of smoked salmon, containing a complex mixture of PAHs with an RPF of 460 ng BaP_{eq}, reduced and delayed absorption. Administration of canned commercial salmon, containing very low amounts of PAHs, showed the impacts on pharmacokinetics were not due to high amounts of PAHs but rather a food matrix effect.

Graphical abstract



Keywords

Pharmacokinetics; Benzo[a]pyrene; Accelerator Mass Spectrometry; Dietary Polycyclic Aromatic Hydrocarbons

1. Introduction

Polycyclic aromatic hydrocarbons (PAHs), a major human health concern, are formed by the incomplete combustion or volatilization of carbon (e.g., coal, petroleum, wood, tobacco, coal tar-based sealcoat products, automobile tires) and human exposures are associated with

multiple diseases including atherosclerosis, asthma, and cancers in a number of target organs including lung (IARC, 2010; Sadiktsis et al., 2012; Titaley et al., 2016). The Agency for Toxic Substances and Disease Registry (ATSDR) lists PAHs as 3 of the top 10 chemicals of concern at priority pollutant sites (ATSDR, 2013). The higher molecular weight PAHs (4 rings and higher) tend to be more carcinogenic and the majority of daily human exposure is through diet (Bansal and Kim, 2015; Dieziel et al., 2011; Domingo and Nadal, 2015; Kazerouni et al., 2001). Benzo[a]pyrene (BaP) is the prototypical carcinogenic PAH, 8th on the ATSDR list (ATSDR, 2013), classified by the International Agency for Research on Cancer (IARC) as a known human carcinogen (IARC, 2010), and used as the standard by agencies such as U.S. Environmental Protection Agency (EPA) in determining relative carcinogenic potency for complex PAH mixtures (the Relative Potency Factor or RPF) (U.S. EPA, 2010) to which we are daily exposed.

Physiologically Based Pharmacokinetic (PBPK) models and cancer studies used for human risk assessment are done primarily in rodent models at doses 5–6 orders of magnitude higher than the average daily BaP exposure, estimated at 270–700 ng (non-smoking adult in the U.S.) (U.S. EPA, 2017). Accelerator mass spectrometry (AMS), with its high sensitivity (low attomole) (Forsgard et al., 2010) allows for study of the pharmacokinetics of human carcinogens at doses that represent a *de minimus* risk to subjects (Cupid et al., 2004; Garner et al., 1999; Jubert et al., 2009; Lightfoot et al., 2000; Madeen et al., 2015; 2016; Malfatti et al., 2016; Turteltaub et al., 1997). Previously, our laboratory determined the pharmacokinetics of aflatoxin B₁ (AFB₁) (Jubert et al., 2009), and dibenzo[*def,p*]chrysene (DBC) (Madeen et al., 2015; 2016) in humans at doses below the LDAL and [¹⁴C] amounts (5 nCi) that are orders of magnitude lower than previously used in diagnostic procedures (Atherton and Spiller, 1994) or clinical trials (Ottaviani et al., 2016).

The pharmacokinetic studies of AFB₁ (Jubert et al., 2009) and DBC (Madeen et al., 2015; 2016) were performed with individuals that had fasted overnight and with cellulose capsules containing only the chemical under test. A more realistic scenario would be co-administration with food. The use of smoked salmon from the Confederated Tribes of the Umatilla Indian Reservation (CTUIR) provided a food matrix with a complex PAH mixture formed during the wood-smoking of Chinook salmon caught by the tribe from the Columbia River (Forsberg et al., 2011; 2012; Motorykin et al., 2015). Co-administration of this complex PAH mixture allows for a test of assumptions critical to the use of the RPF approach to risk assessment for PAH mixtures, e.g., that a PAH congener does not interfere with the ADME of another PAH (in this case [¹⁴C]-BaP) and risk assessment can be done by summing the RPF for each congener multiplied by the percent composition in the mixture of that congener. Thus, this additivity depends upon no inhibition or induction of enzymes involved in ADME or alteration of transport (Jarvis et al., 2014).

BaP and total PAHs are found in almost all foods and are especially high in smoked meats and cheeses and charcoal-broiled foods (Bansal and Kim, 2015; Dieziel et al., 2011; Domingo and Nadal, 2015; Kazerouni et al., 2001; Zelinkova and Wenzl, 2015). In an attempt to reduce the impact on study subjects from even the small BaP dose used in this study, subjects were asked to follow a diet restricted in levels of BaP prior to and during the study.

2. Materials and Methods

2.1. Enrollment Criteria and Demographics of Subjects

Note- This study was conducted under an FDA IND (#117175), Oregon State University IRB approval (#5644) and LLNL-approved IRB Protocol 2017-008.

Volunteers had to meet the following entry criteria: age 21–65; healthy; nonsmoking household; not using medications that can affect gut motility; no history of gastrointestinal surgeries; kidney or liver disease; GI diseases such as Crohn's, ulcerative colitis, or gastritis. Women volunteers had to be post-menopausal or have had surgical sterilization to eliminate any possibility for fetal exposure as the fetus is expected to have a greater sensitivity to PAH toxicity. Volunteers with potentially high occupational PAH exposures (roofers, asphalt pavers, fire-fighters, etc.) were excluded. A recent routine medical examination was performed (within 4 weeks) by a licensed physician to ensure good general health. The screening assessment included obtaining a menopausal history and performance of a urine pregnancy test for all women. We did not require exclusion based on PAH exposure in air or diet, however, during the screening we provided volunteers with a list of foods containing high levels of PAHs and asked they avoid these foods during the trial. The weight (mean 86.7, range 55.6–131.8 Kg), height (mean 176, range 157–190 cm), BMI (mean 27.9, range 18.5–36.5) gender (3 males and 2 females), age (mean 46, range 30–63 years), and race (all Caucasian, not Hispanic) were recorded under an assigned code number for enrolled subjects. As the research involves dosing human participants with radiolabeled compounds, the Radiation Safety Committee at OSU reviewed and approved the protocol.

2.2. Food Diaries

Each volunteer was requested to avoid intake, for 1 week prior to dosing and during the study, of foods high in PAHs such as smoked meats and cheeses or food broiled over charcoal. Volunteers were required to keep a 3-day food diary prior to dosing and for the 3-day study period (see Table 1 for example). The results were used to estimate dietary intake of BaP over the study period with the latest data from the Joint FAO/WHO Expert Committee on Food Additives (JECFA, 2012).

2.3. Radiochemical Purity of [7-¹⁴C]-BaP and Preparation of [¹⁴C]-BaP Capsule

The FDA IND (#117175) and OSU Radiation Safety and IRB committees required quarterly reports of the radiochemical purity of the [7-¹⁴C]-BaP (26.7 $\mu\text{Ci}/\mu\text{mol}$) stock used for dosing. If not 98%, the material had to be purified. We utilized a slight modification of Ramsauer *et al.* (2011), with reverse-phase LC and UV detection. Fractions were collected and counted via liquid scintillation. The stock solution (toluene) was stored in amber vials under argon at -80°C . For preparation of dosing capsules, a portion of the stock was blown to dryness under argon and re-dissolved in food grade ethanol (25 $\mu\text{L}/5$ nCi) prior to adding to food-grade cellulose capsules (ethanol evaporated prior to sealing capsule). At least 3 capsules were counted by liquid scintillation to ensure the proper dose/capsule. If the total counts varied (5 nCi = 11,500 dpm) by more than $\pm 10\%$, a new batch was made. Each batch was stored in the dark at 4°C and discarded if not used within 1 week.

2.4. Acquisition of Smoked Salmon from the CTUIR, Analysis of PAH Composition and Estimated BaP_{eq} Based on Known RPFs

Salmon were smoked in tipi or shed structures over apple or alder wood fire (Forsberg, 2012). The smoking process preserves the salmon as well as cooks it. The salmon was obtained from Tribal fishing grounds on the Columbia River. There were no advisories in these areas with respect to safety. The smoked salmon was placed in sealed food-grade containers in a second container with dry ice and transported to OSU. The quantity of a number of PAHs was determined using the method of Forsberg et al. (2011; 2012), and the results for two batches are shown in Table 2 along with the RPF values used to calculate ng BaP_{eq}/g smoked salmon. That RPF was driven entirely by fluoranthene in the 2015 batch; the 2014 batch also had a small (3.4%) contribution from benzo[*g,h,i*]perylene (Table 2). Once the salmon was analyzed and the PAH levels determined, it was portioned out, weighed, vacuum-sealed and stored at -20°C in a locked container devoted solely to samples intended for human use. The facility was registered with FDA (#11833682472) and licensed by the Oregon Department of Agriculture (AG-L0077970FP). Subjects were given a portion of smoked salmon containing either 46 or 460 BaP_{eq} corresponding to 22.55 g (2014 batch RPF=2.04) or 125.02 g (2015 batch RPF=3.68), respectively. These two levels corresponded to a 1:1 and 10:1 ratio, respectively of BaP_{eq}:[¹⁴C]-BaP. Canned salmon (Ocean Beauty Seafoods, LLC, Icy Point lot #035DR720C) was purchased from Bi Mart (Corvallis, OR). The canned salmon contained a significant amount of water; therefore dry weights of canned and smoked salmon were compared in order to administer equivalent dry weights of each. Liquid was drained from canned salmon and any bones carefully removed from portions of both smoked and canned salmon. Six portions of each were placed in aluminum trays, weighed and placed in a drying oven preheated to 90°C. Salmon was dried for 24 hours, cooled in a desiccator and reweighed. Salmon was then heated in the same oven for an additional 2 hours to ensure all moisture had been removed. Cooling and weighing was repeated for validation. The average percent dry weights of canned and smoked salmon were 35.63 and 51.25%, respectively. Dry weight estimates of 1:1 BaP_{eq} smoked salmon portions were 11.56 g and 64.07 g for both 10:1 BaP_{eq} smoked and canned salmon portions. Canned salmon was not compared at the 1:1 BaP_{eq} ratio. Volunteers consumed 179.74 g wet weight of canned salmon, the equivalent of 125.02 g smoked salmon. Canned salmon was portioned, weighed, and chilled for volunteers 24 hours before serving in order to make it more palatable.

2.5. Dosing of Subjects with [¹⁴C]-BaP and Salmon and Collection of Blood, DNA Isolation from PBMCs and Determination of Covalent Adduction with [¹⁴C]-BaP

Fasted (overnight) subjects were administered a cellulose capsule containing [¹⁴C]-BaP (46 ng, 5 nCi) swallowed with 100 mL water. Canned commercial salmon, or smoked salmon from the CTUIR, consumption was prior to dosing with [¹⁴C]-BaP. Normal eating and drinking resumed 2 hours after dosing with [¹⁴C]-BaP. Blood (up to 10 mL) was drawn by a registered nurse at 0, 0.25, 0.5, 0.75, 1.0, 1.5, 2.0, 2.5, 3, 4, 8, 24, 48 and 72 hours (the 0.25, 2.5 and 72 hour blood draws were not done with the canned salmon cycle) following [¹⁴C]-BaP dosing. An indwelling IV catheter was used for blood collection over the first 4 hours and the remainder with straight needle sticks. A minimum 2 week washout between dosing with [¹⁴C]-BaP was adopted. Peripheral blood mononuclear cells (PBMCs) were isolated

from the buffy coat and used to assess covalent binding of [^{14}C]-BaP to DNA. Briefly, DNA was isolated from PBMCs using a commercial kit for mammalian blood (Roche Diagnostics, Indianapolis, IN). The DNA was precipitated with 100% ethanol and stored at -20°C . Subsequently, the DNA was washed with 100%, precipitated by centrifugation and washed twice with 70% ethanol. The final DNA pellet was solubilized in nuclease-free water at 65°C 30–60 minutes and then stored at 4°C for 49–72 hours. The DNA concentration and purity was determined by using a Nanodrop ND-1000 (Thermo Scientific, Wilmington, DE). The 260/280 nm ratio was 1.7–1.9. Samples were dissolved in water to a concentration of 200 ng/ μL and 50 μg placed in amber glass vials and stored at -80°C for subsequent AMS analysis. Twenty μg (100 μL) was used for the AMS analysis.

2.6. Extraction of [^{14}C]-BaP_{eq} from Plasma and Analysis of [^{14}C]-BaP_{eq} by Accelerator Mass Spectrometry

Aliquots of plasma (0.75 mL) from each time point were extracted within two hours after collection according to the method of Crowell et al. (2011) as modified by Madeen et al. (2015; 2016). The samples were acidified with H_2SO_4 , vortexed and extracted with ethyl acetate. Combined extracts were evaporated under nitrogen to dryness in glass vials with PTFE cap liner and stored at -80°C until shipped on dry ice to the Center for Accelerator Mass Spectrometry at LLNL where they were stored at -80°C until processing. As PAHs can adsorb to plastic, care was taken to use glass containers and vials flushed with argon or nitrogen prior to capping to prevent oxidation of samples. Plasma extracts were reconstituted with 50 μL ethyl acetate and converted to graphite by the method of Ognibene et al. (2004) as described in our previous study with [^{14}C]-DBC (Madeen et al., 2015). Briefly, samples were evaporated and flame-sealed in a quartz tube containing Cu(II) and combusted to 900°C , producing CO_2 . The CO_2 was then transferred to a septa sealed glass tube containing Zn and Co and heated to 525°C , producing graphite on the Co catalyst. The graphite was then loaded into an aluminum sample holder for AMS analysis.

AMS analysis was conducted on the 250 kV Single Stage AMS at the Center for Accelerator Mass Spectrometry at LLNL. AMS operating conditions were optimized to determine the ratio of $^{14}\text{C}:\text{C}$ with a precision of 3% and sensitivity of 0.7 attomole ^{14}C per mg of total carbon (Ognibene et al., 2018). Solid sample standards containing a $^{14}\text{C}:\text{C}$ content of $1.5 \times$ modern are measured intermittently throughout the analysis to assess the ionization and counting efficiency of the AMS. The biochemical samples are measured 4–10 times with the collection of at least 10,000 ^{14}C counts or for 30 seconds for each replicate (Ognibene et al., 2018).

2.7. Pharmacokinetic and Statistical Analysis

The pharmacokinetics of [^{14}C]-BaP_{eq} ([^{14}C]-BaP and metabolites) were evaluated using non-compartmental analysis and a two-compartment model after oral administration of 46 ng [^{14}C]-BaP capsule in one of four treatments: (1) neat or co-administration with (2) 22.55 g smoked salmon (wet weight, 46 ng BaP_{eq} of PAHs), (3) 125.02 g smoked salmon (wet weight, 460 ng BaP_{eq} of PAHs), or (4) 179.74 g commercial canned salmon (wet weight, 460 ng BaP_{eq} of PAHs) as previously detailed in Madeen et al., (2018). AUCs were calculated using the trapezoidal rule. A two-compartment model was used to evaluate the

amount (fg) of [^{14}C]-BaP_{eq} in the absorption, central, and peripheral compartments (first-order rate constants ka , ke , $k12/k21$, respectively). A maximum log likelihood objective and Nelder-Mead algorithm was used to optimize model parameters and initial values set adjusting parameters visually. Statistical analysis of PK parameters was performed with “R: A language and environment for statistical computing” Version 3.2.3 (R Foundation for Statistical Computing, Vienna, Austria). Statistical analysis between dosing regimens was typically performed using a two-tailed students t-test.

3. Results

3.1. Estimated Dietary Intake of BaP Prior to and During the 72 Hour Pharmacokinetic Analysis Following Dosing with [^{14}C]-BaP

The BaP dietary ingestion for all 5 subjects was calculated as a range (low and high estimate). The average low estimate was 216 ± 89 ng/day (range 50–613) and the high estimate was 307 ± 113 ng/day (range 112–833) (Table 3). We found the greatest contributors to daily dietary intake of BaP in our subjects were black olives (11.3 ng/g), cooked quinoa (0.5 ng/g), coconut crème (3.4 ng/g) and pan-fried beef (0.4–0.6 ng/g).

3.2. Pharmacokinetics of [^{14}C]-BaP in the Presence or Absence of Smoked Salmon Containing a Complex PAH Mixture at 46 or 460 ng BaP_{eq}

The RPF approach to risk assessment of environmental complex PAH-containing mixtures relies on the assumption that individual PAHs behave independently, i.e., that one PAH does not impact the transport or ADME of another PAH. Typical biomarkers used to predict PAH-dependent carcinogenesis include CYP1 induction (CYP1A1 and CYP1B1, responsible for bioactivation of BaP to 7,8-dihydroxy-9,10-epoxy-7,8,9,10-tetrahydro-BaP, BaPDE) or BaPDE covalent binding to DNA (Kang et al., 2007; Kriek et al., 1993; Lee et al., 2002; Ross et al, 1995). There have been numerous studies done *in vitro* (Genies et al., 2016; Shimada and Guengerich, 2006; Shimada et al., 2008; Staal et al., 2006; 2007; Tarantini et al., 2011) and *in vivo* animal models (Nesnow et al., 1998; Schneider et al., 2002) that call into question the reliability of that assumption and RPF-dependent cancer risk assessment including our own work with human cancer cells (Mahadevan et al., 2005; 2007) and the mouse skin tumor model (Courter et al., 2007; Siddens et al., 2012; Tilton et al., 2015). In this study, we examined the pharmacokinetics of the prototypical PAH, BaP, which is the calibrator for cancer RPF determinations of PAHs (BaP has an RPF set at 1) in the presence of a food containing a complex PAH mixture of known composition and RPF. Two of the three volunteers (A & D) administered 125.02 g (460 ng BaP_{eq}) CTUIR smoked salmon prior to the capsule containing 46 ng [^{14}C]-BaP, exhibited markedly different pharmacokinetics compared to when dosed with [^{14}C]-BaP alone (Figure 1 and Tables 4 and 5). A more muted response was seen with volunteer E, primarily due to the high variability between the three cycles of smoked salmon (Figure 1, bottom panel). The fraction absorbed was decreased by 23–30% in volunteers A and D but actually slightly increased with volunteer E. The C_{max} was markedly reduced in all three volunteers and the T_{max} increased with volunteers A and D but remained the same with volunteer E. During that same time frame, consuming 22.5 g of salmon (46 BaP_{eq}) actually slightly increased (3–12%) the fraction of [^{14}C]-BaP_{eq} absorbed. Rates of oral absorption of [^{14}C]-BaP_{eq} were reduced with

increasing amounts of salmon vehicle (0.921 for neat, 0.590 for 22.5 g salmon vehicle, and 0.140 (volunteers A and D; 0.500 for A, D and E) for 125.02 g salmon vehicle). There was no statistical difference, using the non-compartment model, in $T_{1/2}$ of [^{14}C]-BaP_{eq} between volunteers that consumed smoked salmon with the [^{14}C]-BaP capsule and those who consumed the capsule neat ($p=0.82$).

With ingestion of smoked salmon the $K_{el\alpha}$ in the two-compartment model was markedly reduced (again, with the exception of volunteer E) whereas $T_{1/2\alpha}$ and $T_{1/2\beta}$ were increased (except Volunteer E) in a dose-response fashion. Together, these changes in pharmacokinetics could have been interpreted as not supporting the RPF approach to cancer risk assessment as it appears that, in the presence of a much greater amount and number of PAHs, the pharmacokinetics of orally administered [^{14}C]-BaP is markedly altered. This effect was dampened when the level of BaP_{eq} in the smoked salmon was reduced to 46 ng (Figure 2 and Tables 4 and 5). There was a slight, non-significant reduction in C_{max} . A smaller portion of smoked salmon with a 1:1 ratio of BaP_{eq} to [^{14}C]-BaP also resulted in a delay in absorption. The impact of metabolism on the pharmacokinetics of [^{14}C]-BaP with fasted individuals or those given the smoked salmon cannot be assessed from this study.

3.3. Pharmacokinetics of [^{14}C]-BaP in the Presence or Absence of Canned Salmon

To ensure that the alteration in [^{14}C]-BaP pharmacokinetics was due to PAHs in the smoked salmon and not any food matrix effect we dosed two individuals with 179.4 g of canned commercially obtained non-smoked salmon that had about 500-fold lower total PAHs (and no detectable carcinogenic PAHs). The impact on [^{14}C]-BaP pharmacokinetics was similar to the high smoked salmon intake (Figure 3 and Tables 4 and 5) indicating that the alteration in absorption and elimination from plasma was due to the food matrix itself and not the high levels of PAHs in the CTUIR smoked salmon.

3.4 Lack of Covalent Adduction of [^{14}C]-BaP_{eq} to DNA from PBMCs

Covalent adduction to DNA has been used as a biomarker of cancer risk for numerous carcinogens including BaP (Boysen and Hecht, 2003). We found no [^{14}C]-BaP_{eq} covalently bound to total DNA isolated from PBMCs out to 72 hours post-dosing (LOD 10 fg/mg DNA or 0.5 adducts/ 10^{11} nucleotides).

4. Discussion

BaP, a class 1 known human carcinogen (IARC, 2010), is found in almost all food and is especially high in charcoal-broiled meats or smoked meats and cheeses. The concentration of BaP in a particular food stuff varies markedly world-wide, e.g., eggs in the U.S. have been reported to contain an average of 0.03 ng/g (Kazerouni et al., 2001) whereas in Kuwait (post Iraq War) the number is over two orders of magnitude higher (7.49 ng/g) (Husain et al., 1997). White bread in the U.S. contains 0.10 ng BaP/g (Kazerouni et al., 2001) compared to Italy with 0.017 ng/g (Lodovici et al., 1995). Therefore, it is not surprising that reports of the estimated daily dietary intake of BaP range widely from 5–3,440 ng (Domingo and Nadal, 2015). In this study, based on our dietary questionnaire, the average intake for the five volunteers prior to study initiation was 221–306 ng/day. Using the recently revised cancer

risk slope factor of $1 \text{ (mg/kg-day)}^{-1}$ this would equate to a lifetime excess cancer risk of 3.1 to 4.4×10^{-6} .

The RPF approach to cancer risk assessment with complex environmental PAH mixtures assumes that there is a common mode of action and additivity applies across a wide range of structures (U.S. EPA, 2010). Thus, there should not be evidence of synergism or inhibition when comparing binary or complex mixtures. There are a number of studies using both *in vitro* markers of carcinogenic potency with human cell lines and *in vivo* tumor studies with animal models that raise serious doubts about this approach. The exquisite sensitivity of AMS allowed us, for the first time, to examine the impact (alteration in pharmacokinetics) of co-administration of a food containing a complex PAH mixture with BaP in humans at a defined dose. To that end, 46 ng of [^{14}C]-BaP was administered with or without smoked salmon containing either a 1:1 or a 10:1 ratio of BaP_{eq} as determined by the quantity and profile of PAH congeners by GC-MS in smoked salmon supplied by the CTUIR. In the study by Forsberg et al. (2012), 40 samples of Chinook salmon from the Columbia River, smoked for 22–33 hours with 2 types of wood in two different structures (tipi or shed), were analyzed for 62 PAHs. All PAHs detected were due to the smoking process as the unsmoked fillets were all below the limit of detection (2–10 ng/g for the PAHs analyzed).

Approximately 98% of the total PAHs in the smoked salmon were 2–4 ring PAHs. The larger carcinogenic PAHs (chrysene, benz[*a*]anthracene, benzo[*b*]fluoranthene, benzo[*a*]pyrene, benzo[*k*]fluoranthene and benzo[*b*]fluoranthene) were present at levels of 26–100 ng/g wet weight. As reported in this previous study, the major carcinogenic (assigned RPF) PAH in the CTUIR salmon used in this study contributing to the BaP_{eq} was fluoranthene (46 ng/g). BaP was below the limit of detection (2 ng/g) which Forsberg et al. (2012) found in 10% of the samples they analyzed.

A caveat in comparing pharmacokinetic data from rodents and humans in this current study is the orders of magnitude difference in dosage. Even so, in fasted individuals dosed with 46 ng [^{14}C]-BaP the T_{max} (1.25 hours) was similar to that (0.92–1.20 hours) found in non-fasted C57BL/6 mice (Fang and Zhang, 2010) after an oral dose (gavage in DMSO:corn oil, 1:9 v/v) of 15–30 mg/kg (approximately 7 orders of magnitude higher, on a mg/kg b.w. basis, than the human dose used here); the $T_{1/2}$ values were also similar (0.8 and 1.1 hours for mouse and human, respectively). Not unexpectedly, the C_{max} ratio was also about 7 orders of magnitude different (233 ng/mL versus 70 fg/mL). Pharmacokinetic studies in the rat indicate a much longer time for absorption with a T_{max} of 5–8 hours (Olesen et al., 2016; Ramesh et al., 2001). The Olesen et al. (2016) study dosed male Sprague-Dawley rats (fasted overnight and for 8 hours post-dosing) with 1.05 $\mu\text{g/kg b.w}$ [^3H]-BaP by oral gavage in ethanol whereas the Ramesh et al. (2001) study dosed male Fisher-344 rats (fasted for 16 hours prior to dosing) with 100 mg/kg b.w. BaP by gavage in peanut oil. Administration of smoked salmon containing 460 ng BaP_{eq} markedly reduced the C_{max} , increased T_{max} and essentially eliminated $K_{\text{el}\alpha}$ (two-compartment model) of [^{14}C]-BaP_{eq} in volunteers A and D (actually increased with volunteer E) all of which support rejection of the tenants of the RPF approach for PAH mixtures. A less marked delay in absorption and reduction in elimination was observed when the quantity of smoked salmon was reduced 10-fold (22.55 g of CTUIR smoked salmon, 46 ng BaP_{eq}). However, in order to establish that these alterations in [^{14}C]-BaP were due to the high levels of co-administered PAHs and not a food matrix effect, we

repeated the analysis with commercial canned salmon (containing at least 500-fold less total PAHs and no carcinogenic PAHs) at an amount (179.74 g) equivalent to the smoked salmon portion containing 460 ng BaP_{eq}. The impact was similar to alteration in pharmacokinetics seen with the 125.02 g co-administration of CTUIR smoked salmon. Thus, it appears that the food matrix is an important factor in oral BaP pharmacokinetics.

Rate and extent of oral absorption is dependent on chemical solubility, chemical concentration, and GI permeability (Mudie 2010). Here, we observed reduction of BaP bioavailability and rate of absorption with increasing dose vehicle. Salmon is a food source rich in lipid and there have been studies indicating enhanced GI absorption of PAHs with high-lipid containing foods or lipid vehicles (Laher et al., 1983) although not all studies have observed this effect (Laher et al., 1984). Since PAHs are highly lipophilic, salmon lipids could increase BaP solubility in the gut. However, the volume of salmon can dramatically reduce the BaP concentration, which we hypothesize is reducing the rate and extent of BaP absorption. Additionally, BaP could bind to insoluble salmon further reducing absorption.

A reduction in K_{elα} with increasing salmon vehicle was observed. This reduction of K_{elα} could be a result of reduced absorption or perhaps a change in overall BaP metabolism. A more definitive answer requires evaluation of BaP metabolites, which is not possible with this application of AMS as only total [¹⁴C] is measured. A number of metabolic pathways exist for BaP, some leading to detoxication and others to bioactivation (reviewed in ATSDR, 2013; U.S. EPA, 2017). A portion of BaP carcinogenicity/toxicity is also likely due to canonical AhR signaling as BaP and some metabolites are ligands (ATSDR, 2013; U.S. EPA, 2017). A great deal of research has documented the metabolism of BaP in hepatic and pulmonary tissues (reviewed in ATSDR, 2013). Extensive GI and hepatic metabolism could lead to a “first pass” effect wherein the [¹⁴C]-BaP_{eq} measured in plasma in this study would be expected to be comprised of predominantly BaP metabolites. The mouse and human GI have a complement of phase 1 and phase 2 BaP metabolizing enzymes although not as extensive as in liver (Buesen et al., 2002; Uno et al., 2008). This observation, plus estimates that greater than 95% of BaP (and other high MW carcinogenic PAHs) exposure in non-smoking humans is dietary (ATSDR, 2013), suggests that further work on carcinogenic PAH risk assessment needs to take into account the role of the GI in addition to the liver. In a previous study we micro-dosed (29 ng) human volunteers with [¹⁴C]-DBC and assessed the pharmacokinetics of [¹⁴C]-DBC_{eq} (Madeen et al., 2015). Almost all of the pharmacokinetic parameters were similar (caveat that only 1/5 volunteers were in both studies) to [¹⁴C]-BaP_{eq} except there was a non-significant trend with K_a (slower with [¹⁴C]-DBC_{eq}, p=0.057) and T_{max} (longer with [¹⁴C]-DBC, p=0.081). As the 6-membered ring PAH, DBC, has a log K_{ow} (7.2) higher than BaP (6.0), it would appear that absorption of an oral dose in the absence of food is not determined primarily by hydrophobicity.

Daniel W. Nebert and colleagues performed a series of elegant experiments on BaP toxicity and carcinogenesis in mice employing single, double and triple knockouts (*Cyp1a1*^{-/-}, *Cyp1a2*^{-/-}, *Cyp1b1*^{-/-}, *Cyp1a1*^{-/-}/*1a2*^{-/-}, *Cyp1a1*^{-/-}/*1b1*^{-/-}, *Cyp1a2*^{-/-}/*1b1*^{-/-} and *Cyp1a1*^{-/-}/*1a2*^{-/-}/*1b1*^{-/-}) of the mouse *Cyp1* genes (Dragin et al., 2008; Nebert et al., 2013; Shi et al., 2010a; Uno et al., 2001; 2004; 2006; 2008). Surprisingly, loss of expression of CYP1a1 enhanced toxicity and promoted carcinogenesis. Liver- and intestinal-specific

Cyp1a1 knockout mice further established the importance of GI Cyp1a1 metabolism of BaP in reducing uptake and distribution of BaP to target tissues (Shi et al., 2010b). A humanized *Cyp1a1* mouse has been employed to establish that the human gene is similar to the mouse in expression in the GI tract and in the metabolism of BaP (Dragin et al., 2007). The overall conclusion from these studies was that in the mouse GI metabolism of BaP is the predominant factor in toxicity and carcinogenesis. One caveat when comparing these studies to the present study is that the mouse studies were done with chronic high dose oral BaP exposure with presumed induction of CYP1a1 whereas it is unlikely that the single micro-dose of BaP employed here would significantly impact expression of BaP-metabolizing enzymes although the PAHs in the smoked salmon could induce GI (and perhaps liver) CYP1A1. Consistent with the tissue-specific mouse *Cyp1a1* knockout studies, intestinal-specific NADPH cytochrome P450 reductase null mice (which would eliminate all CYP-dependent metabolism) have shown intestinal BaP metabolism is important in pharmacokinetics of orally (but not intraperitoneally) administered BaP (Fang and Zhang, 2010). In the Fang and Zhang (2010) study, there was evidence that induction of Cyp1a1 metabolism of BaP could be induced as early as 2 hours following a 30 mg/kg oral dose. How might the salmon matrix impact intestinal CYP1A1-dependent BaP metabolism in humans?

In order to address these questions, we have initiated studies with “moving wire” technology developed at LLNL, in which UPLC can be interfaced with AMS allowing for quantification and identification (and thus pharmacokinetics) of [¹⁴C]-BaP metabolites (Thomas et al., 2011). It will be important to determine if alteration in [¹⁴C]-BaP metabolism occurs in the presence of smoked salmon containing high levels of multiple PAHs and how that impacts pharmacokinetics. Ramesh et al. (2001), performed a comprehensive examination of BaP and BaP metabolite levels in multiple tissues over time in the rat to determine pharmacokinetic values after oral dosing of rats with 100 mg/kg. Furthermore, they utilized HPLC to resolve and quantify tissue- and time-dependent levels of BaP and phase 1 and phase 2 metabolites. It will be of great interest to compare their results with respect to BaP metabolite profiles over time to plasma and urine of micro-dosed humans (a limitation of our study is the lack of availability of tissues).

5. Conclusion

Accelerator mass spectrometry was utilized to quantify [¹⁴C]-BaP_{eq} in plasma of humans over a 72 hour period following oral dosing with 46 ng in the presence or absence of quantities of smoked salmon containing a complex PAH mixture at BaP_{eq} of 1:1 or 1:10 ([¹⁴C]-BaP:BaP_{eq} smoked salmon). Co-administration of the large portion (125.02 g) of smoked salmon containing 460 ng BaP_{eq} altered the pharmacokinetics of [¹⁴C]-BaP_{eq}. The rate of absorption was decreased, the T_{max} increased and C_{max} and K_{1e} decreased. Reducing the amount of co-administered smoked salmon 10-fold reduced the extent of alteration in pharmacokinetic parameters. An initial interpretation of this study was that the presence of large amounts of complex PAHs altered processes involved in the ADME and/or transport of orally administered [¹⁴C]-BaP_{eq}. However, co-administration of commercially obtained canned salmon (with total PAH levels 500-fold lower than smoked salmon and with no carcinogenic PAHs), at amounts equivalent to the large portion of smoked salmon, had the

equivalent impact on [¹⁴C]-BaP_{eq} pharmacokinetics suggesting that this was a food matrix effect (perhaps the lipid-rich salmon somehow sequestered [¹⁴C]-BaP_{eq} or changed the route of absorption to primarily lymphatic).

To assess the utility of high dose animal data for pharmacokinetics, toxicokinetic and tumor studies for carcinogenic PAHs, it is critical that we have data from humans following administration of an environmentally relevant dose. A comprehensive comparison requires analysis of [¹⁴C]-BaP and its metabolites which is now possible with a UPLC-AMS “moving wire” approach and that will be the focus of future research. Finally, analysis of DNA isolated from PBMCs had no detectable (LOD = 0.5 adduct per 10¹¹ nucleotides) [¹⁴C]-BaP_{eq} covalent binding negating its usefulness as a biomarker of cancer risk.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Competing interests and funding statements:

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Abbreviations

ADME	absorption distribution metabolism and excretion
AFB₁	aflatoxin B ₁
AMS	accelerator mass spectrometry
ATDSR	Agency for Toxic Substances and Disease Registry
BaP	benzo[a]pyrene
BaP_{eq}	benzo[a]pyrene equivalents
BLOD	below the limit of detection
BLOQ	below the limit of quantitation
BMI	body mass index
CTUIR	Confederated Tribes of the Umatilla Indian Reservation
CYP	cytochrome P450

DBC	dibenzo[<i>def,p</i>]chrysene
EPA	Environmental Protection Agency
FDA	Food and Drug Administration
IARC	International Agency for Research on Cancer
IND	investigative new drug
IRB	institutional review board
JECFA	Joint FAO/WHO expert committee on food additives
LDAL	lowest daily allowable level
LLNL	Lawrence Livermore National Laboratory
LOD	limit of detection
OSU	Oregon State University
PAH	polycyclic aromatic hydrocarbon
PBMCs	peripheral blood mononuclear cells
PBPK	physiologically based pharmacokinetics
PK	pharmacokinetics
RPF	relative potency factor

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Highlights

Oral micro-dosing of humans with [¹⁴C]-BaP, at a level in the low range of lifetime dietary BaP exposure, was performed in the presence or absence of food (smoked salmon) containing a complex mixture of polycyclic aromatic hydrocarbons (PAHs) (1-or 10-fold the BaP_{eq}). This test of the relative potency factor approach (RPF) for oral cancer risk assessment for PAH mixtures appeared to call into question the validity of this method as smoked salmon impacted a number of pharmacokinetic parameters resulting in delayed absorption (higher T_{max}), reduced maximum blood levels (C_{max}) and reduction in the rate of elimination (longer T_{1/2} and slower K_e). However, administration of canned salmon containing no detectable PAHs, at identical amounts had similar impacts on pharmacokinetics of [¹⁴C]-BaP supporting a food matrix effect rather than interference by high levels of multiple PAHs. Physiologically-based pharmacokinetic models, toxicokinetics and risk assessments are currently performed with rodent models at doses orders of magnitude higher than actual human exposures. This unique dataset should be useful for further analysis of cancer risk in humans following exposure to environmentally relevant levels.

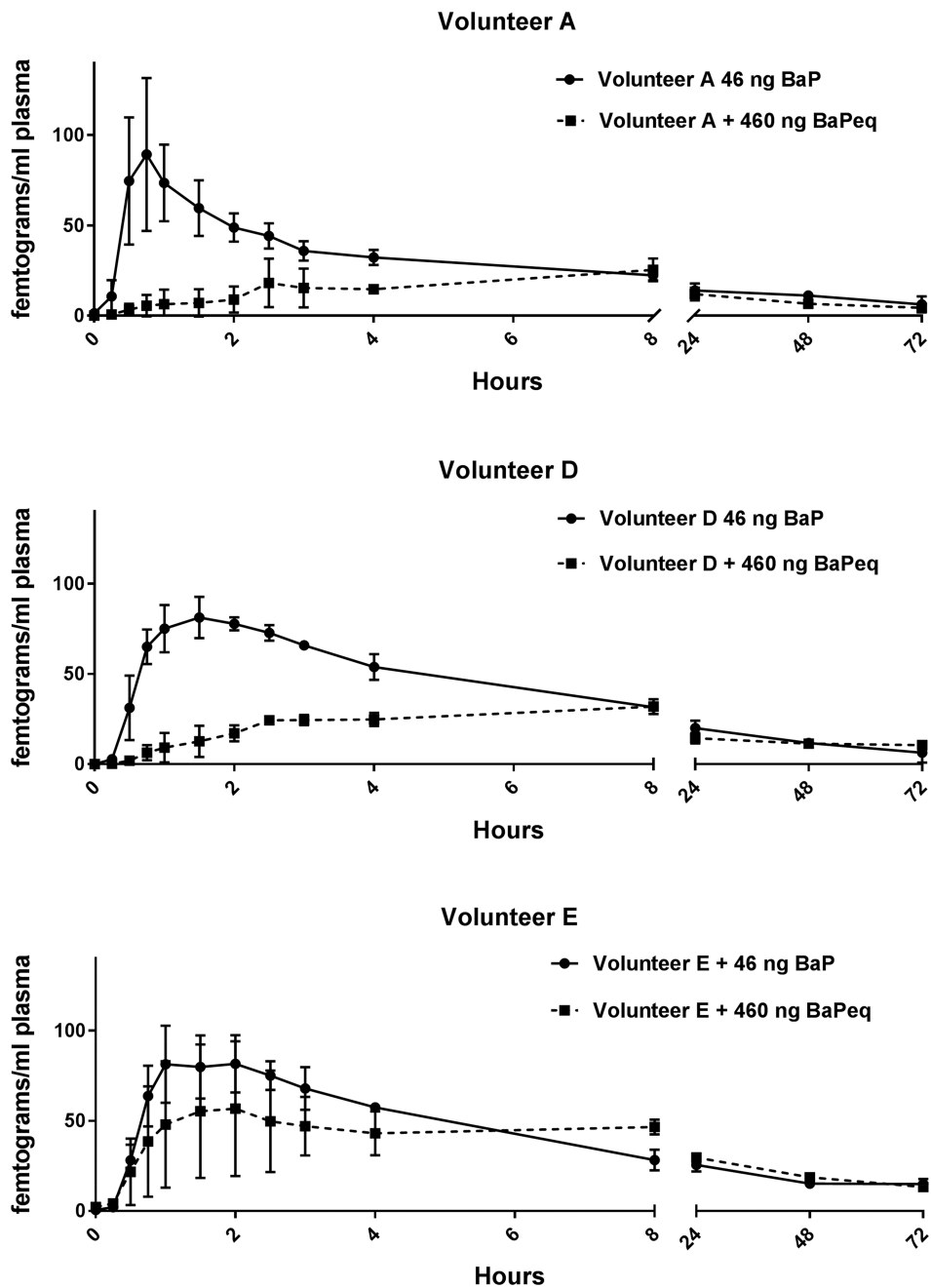


Figure 1. Pharmacokinetics in Plasma of [^{14}C]-BaP Administered Alone or with 10-X BaP_{eq} in Smoked Salmon

Plasma levels (fg/mL) of [^{14}C]-BaP_{eq} in volunteers A (Top Panel), D (Middle Panel) and E (Bottom Panel) over 72 hours after dosing with 46 ng (5 nCi) of [^{14}C]-BaP with (squares, broken line) or without (circles, solid line) co-administration of 125.02 g of CTUIR-smoked salmon containing a complex PAH mixture with 460 ng BaP_{eq} using published RPF values. The symbols represent the mean of three separate volunteer dosing trials and the bars the S.E. of the mean.

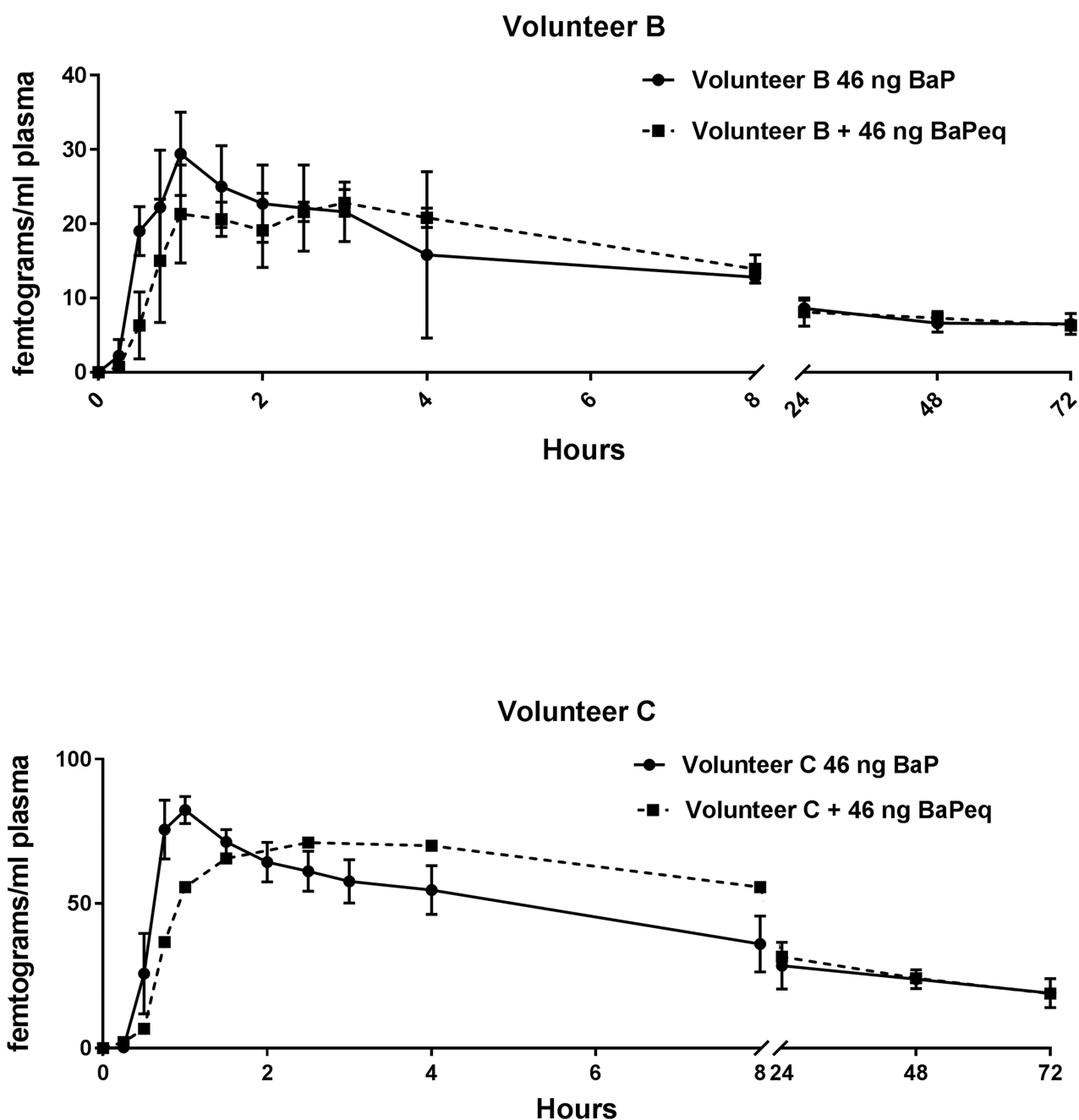


Figure 2. Pharmacokinetics in Plasma of [^{14}C]-BaP Administered Alone or with 1-X BaP_{eq} in Smoked Salmon

Top Panel- Plasma levels (fg/mL) of [^{14}C]-BaP_{eq} in volunteer B over 72 hours after dosing with 46 ng (5 nCi) of [^{14}C]-BaP with (squares, broken line) or without (circles, solid line) co-administration of 22.55 g of CTUIR-smoked salmon containing a complex PAH mixture with 46 ng BaP_{eq} using published RPF values. The symbols represent the mean of three separate volunteer dosing trials and the bars the S.E. of the mean. Bottom Panel- Plasma levels (fg/mL) of [^{14}C]-BaP_{eq} in volunteer C over 72 hours after dosing with 46 ng (5 nCi) of [^{14}C]-BaP with (squares) or without (circles) co-administration of 22.7 g of CTUIR-smoked salmon containing a complex PAH mixture with 46 ng BaP_{eq} using published RPF

values. Volunteer C only underwent 1 dosing trial (squares, broken line) with co-administration of 22.55 g smoked salmon. For clarity, for volunteer E with salmon, only 1 directional error bars were plotted.

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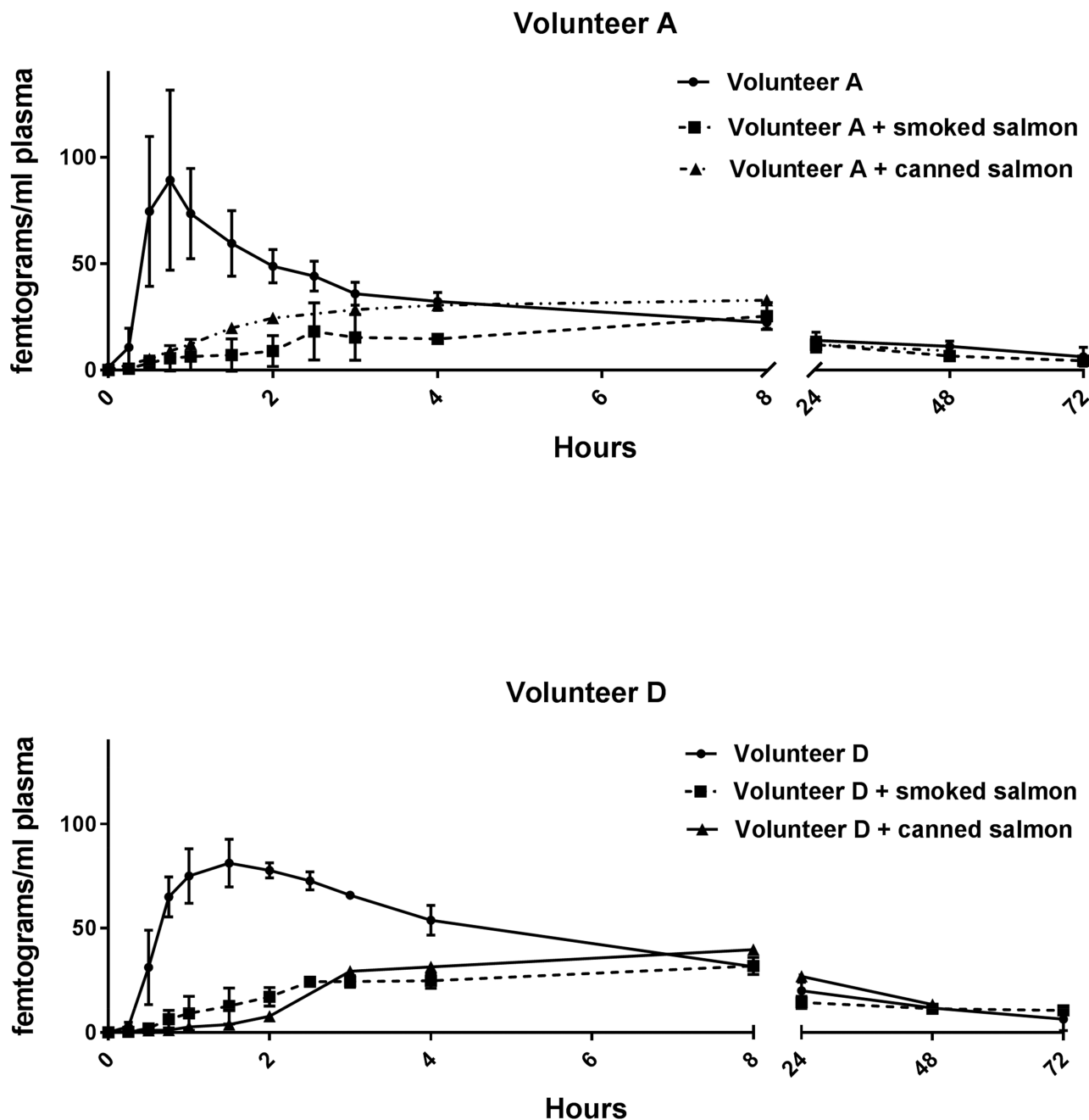


Figure 3. Pharmacokinetics in Plasma of [^{14}C]-BaP Administered Alone, with Smoked Salmon or Canned Salmon

Plasma levels (fg/mL) of [^{14}C]-BaP_{eq} in volunteers A (Top Panel) and D (Bottom Panel) over 72 hours after dosing with 46 ng (5 nCi) with [^{14}C]-BaP alone (circles, solid line) or immediately following consumption of 125 g of smoked salmon (460 BaP_{eq}, squares, dashed line) or 179.4 g of commercially obtained canned salmon (triangles, broken line). The symbols represent the mean of three separate volunteer dosing trials with [^{14}C]-BaP alone or with smoked salmon and the bars the S.E. of the mean. Co-administration of canned salmon was analyzed following a single dosing.

Table 1

Example of Food Diary and Calculation of BaP Intake

Food Item and Method of Preparation	Portion Size	Portion Size (g)	BaP ng/g	BaP ng/portion size	
				Low	High
Coffee, black	32 oz	960	0.011	10.56	
Breakfast Hash, pan fried	1 serving				
Pork Sausage	1/4 cup	32.5	0.02	0.65	
Egg	1 each	50	0.03	1.50	
Potato	1 medium	213	0.001–0.17	0.21	36.21
Grated Cheddar	1/4 cup	28.25	0.50	14.13	
Onion	2 Tbsp	20	0.46	9.20	

BaP Meal Intake Range 36.25 – 72.25 ng

CTUIR Smoked Salmon: PAH^a Composition from Two Separate Years and BaP_{eq} Calculated from Published RPFs

Table 2

Chemical Name	RPF	2014		2015	
		ng/g	BaP _{eq}	ng/g	BaP _{eq}
Naphthalene		75.4		170.00	
2-Methylnaphthalene		42.8		63.00	
1-Methylnaphthalene		35.7		46.00	
2-Ethylphenanthrene		11.71		BLOD	
2,6-Dimethylnaphthalene		45.6		90.00	
1,6-Dimethylnaphthalene		19.04		BLOQ	
1,4-Dimethylnaphthalene		BLOD		BLOD	
1,5-Dimethylnaphthalene		3.29		BLOD	
1,2-Dimethylnaphthalene		BLOD		BLOQ	
1,8-Dimethylnaphthalene		BLOD		BLOD	
2,6-Diethylnaphthalene		BLOD		BLOD	
Acenaphthylene		69.7		BLOQ	
Acenaphthene		11.35		BLOD	
Fluorene		43		74.00	
Dibenzothiophene		BLOD		BLOD	
Phenanthrene		102.3		150.00	
Anthracene		21.6		BLOQ	
2-Methylphenanthrene		13.32		20.00	
2-Methylanthracene		10.75		BLOQ	
1-Methylphenanthrene		6.49		BLOQ	
9-Methylanthracene		BLOD		BLOD	
3,6-Dimethylphenanthrene		BLOD		BLOD	
2,3-Dimethylphenanthrene		BLOD		BLOD	
Fluoranthene	0.08	24.6	1.97	46.00	3.68
9,10-Dimethylphenanthrene		BLOD		BLOD	

Chemical Name	RPF	2014		2015	
		ng/g	BaPeq	ng/g	BaPeq
Pyrene		29.3		36.00	
Retene		BLOD		BLOD	
Benzo(a)fluorene		BLOD		BLOD	
Benzo(b)fluorene		BLOD		BLOD	
Benzo(c)fluorene	20.00	BLOD		BLOD	
1-Methylpyrene		BLOD		BLOD	
Benzo(a)anthracene	0.20	BLOD		BLOD	
Cyclopenta(cd)pyrene	0.40	BLOD		BLOD	
Triphenylene		BLOD		BLOD	
Chrysene	0.10	BLOD		BLOD	
6-Methylchrysene		BLOD		BLOD	
5-Methylchrysene		BLOD		BLOD	
Benzo(b)fluoranthene	0.80	BLOD		BLOD	
7,12-Dimethylbenz(a)anthracene		BLOD		BLOD	
Benzo(k)fluoranthene	0.03	BLOD		BLOD	
Benzo(j)fluoranthene	0.30	BLOD		BLOD	
Benzo(i) and (e)aceanthrylene	60.00	BLOD		BLOD	
Benzo(e)pyrene		BLOD		BLOD	
Benzo(a)pyrene	1.00	BLOD		BLOD	
Indeno(1,2,3-c,d)pyrene	0.07	BLOD		BLOD	
Dibenzo(a,b)anthracene	10.00	BLOD		BLOD	
Benzo(a)chrysene		BLOD		BLOD	
Benzo(g,h,i)perylene	0.01	8.05	0.07	BLOD	
Anthanthrene	0.40	BLOD		BLOD	
Naphtho(1,2-b)fluoranthene		BLOD		BLOD	
Naphtho(2,3-j)fluoranthene		BLOD		BLOD	
Dibenzo(a,e)fluoranthene	0.90	BLOD		BLOD	
Dibenzo(a,l)pyrene	30.00	BLOD		BLOD	

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Chemical Name	RPF	2014		2015	
		ng/g	BaPeq	ng/g	BaPeq
Naphtho(2,3-k)fluoranthene		BLOD		BLOD	
Naphtho(2,3-e)pyrene	0.30	BLOD		BLOD	
Dibenzo(a,e)pyrene	0.40	BLOD		BLOD	
Coronene		BLOD		BLOD	
Dibenzo(e,l)pyrene		BLOD		BLOD	
Naphtho(2,3-a)pyrene		BLOD		BLOD	
Benzo(b)perylene		BLOD		BLOD	
Dibenzo(a,i)pyrene	0.60	BLOD		BLOD	
Dibenzo(a,b)pyrene	0.90	BLOD		BLOD	

^aPAHs with published RPFs are highlighted in gray

BLOD, below the limit of detection (2–10 ng/g)

BLOQ, below the limited of quantification (2–10 ng/g)

Table 3

Dietary Estimates of Daily BaP Exposure

Volunteer	Estimated Average Low and High Daily Consumption of Dietary BaP (ng/day)				
	Pre-Study ^a	Cycle 1 ^b	Cycle 4 ^c	Average \pm SD Low	Average \pm SD High
A	109–157	116–255	50–112	92 \pm 36	175 \pm 73
B	496–606	152–261	168–241	272 \pm 194	369 \pm 205
C	292–402	268–325	187–278	249 \pm 55	335 \pm 61
D	97–162	254–311	124–151	158 \pm 84	208 \pm 89
E	111–205	201–304	613–833	308 \pm 268	447 \pm 338
Average	221–306	198–291	232–319	216	307
\pm SD	174–195	65–31	168–195	89	113

^a Estimate from 3-day diary prior to first dosing with [¹⁴C]-BaP

^b Estimate from 3-day diary following first dosing with [¹⁴C]-BaP

^c Estimate from 3-day diary following first ingestion of smoked salmon with [¹⁴C]-BaP (smoked salmon not included in calculation)

Non-Compartmental Pharmacokinetics^a of [¹⁴C]-BaP Following Micro-Dosing with or without 46 or 460 ng [¹⁴C]-BaP_{eq} Smoked Salmon or Canned Salmon

Table 4

Subject	Fa (fraction absorbed)	T _{max} (hr)	C _{max} (fg/mL)	AUC _{0-72hr} (fgxhxmL ⁻¹)	AUC _{0-∞} (fgxhxmL ⁻¹)	T _{1/2} (hr)
Impact of Smoked Salmon Containing 460 ng BaP_{eq} on [¹⁴C]-BaP Pharmacokinetics						
A	-	0.75	79.2	1135	2118	50
A + 460 ng BaP _{eq}	0.700	8	25.3	794	992	30
D	-	1.5	78.1	1450	1832	31
D + 460 ng BaP _{eq}	0.773	8	31.9	1121	2779	103
E	-	2	81.6	1709	3052	60
E + 460 ng BaP _{eq}	1.13	2	56.7	1929	2738	40
Impact of Smoked Salmon Containing 46 ng BaP_{eq} on [¹⁴C]-BaP Pharmacokinetics						
B	-	1.0	29.4	649	1769	114
B + 46 ng BaP _{eq}	1.026	3	22.8	666	1880	125
C	-	1.0	82.4	2060	4306	79
C + 46 ng BaP _{eq}	1.121	2.5	71.1	2315	4074	60
Impact of Canned Salmon on [¹⁴C]-BaP Pharmacokinetics						
A + Canned Salmon	0.736	8	32.8	815	1108	23
D + Canned Salmon	0.837	8	39.7	1213	1699	27

Table 5
Two-Compartment Pharmacokinetics^a of [¹⁴C]-BaP Following Micro-Dosing with or without 46 or 460 ng [¹⁴C]-BaP_{eq} Smoked Salmon or Canned Salmon

Subject	K _a (hr ⁻¹)	K ₁₂ (hr ⁻¹)	K ₂₁ (hr ⁻¹)	K _{1e} (hr ⁻¹)	K _{o1a} (hr ⁻¹)	T _{1/2a} (hr)	V (L)	K _{elβ} (hr ⁻¹)	T _{1/2β} (hr)
Impact of Smoked Salmon Containing 460 ng BaP_{eq} on [¹⁴C]-BaP Pharmacokinetics									
A	1.394	0.773	0.164	0.083	1.006	0.689	304	0.014	51.1
A + 460 ng BaP _{eq}	0.126	0.086	0.011	0.012	0.108	6.416	822	0.001	570
D	0.672	0.422	0.128	0.089	0.620	1.118	280	0.018	38.0
D + 460 ng BaP _{eq}	0.154	0.134	0.021	0.001	0.156	4.445	603	0.0001	5725
E	0.591	0.466	0.077	0.050	0.586	1.183	235	0.007	106
E + 460 ng BaP _{eq}	1.221	0.006	0.174	0.025	0.248	2.798	727	0.017	40
Impact of Smoked Salmon Containing 46 ng BaP_{eq} on [¹⁴C]-BaP Pharmacokinetics									
B	1.280	0.307	0.102	0.019	0.423	1.639	1146	0.004	155.7
B + 46 ng BaP _{eq}	0.730	0.165	0.063	0.015	0.239	2.900	1333	0.004	172
C	0.666	0.486	0.145	0.034	0.658	1.054	294	0.007	93.1
C + 46 ng BaP _{eq}	0.450	0.161	0.091	0.034	0.274	2.529	354	0.01	62
Impact of Canned Salmon on [¹⁴C]-BaP Pharmacokinetics									
A + Canned Salmon	0.301	0.050	0.031	0.037	0.107	6.49	847	0.011	65
D + Canned Salmon	0.093	0.055	0.011	0.028	0.090	7.690	460	0.003	209