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Microfluidic Diatomite Analytical Devices for Illicit Drug Sensing with ppb-Level Sensitivity

Xianming Kong^{a,b}, Xinyuan Chong^b, Kenny Squire^b, and Alan X. Wang^{b,*}

^aCollege of Chemistry, Chemical Engineering and Environment Engineering, Liaoning Shihua University, Fushun, Liaoning 113001, P. R. China

^bSchool of Electrical Engineering and Computer Science, Oregon State University, Corvallis, OR, 97331, USA

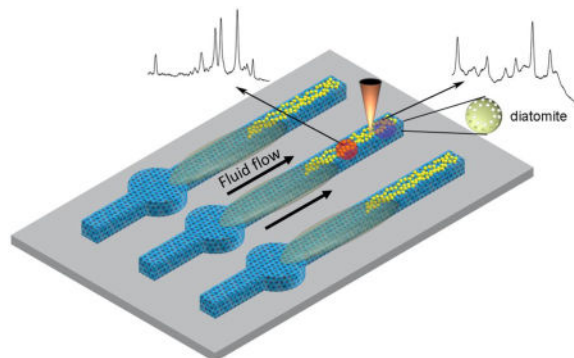
Abstract

The escalating research interests in porous media microfluidics, such as microfluidic paper-based analytical devices, have fostered a new spectrum of biomedical devices for point-of-care (POC) diagnosis and biosensing. In this paper, we report microfluidic diatomite analytical devices (μ DADs), which consist of highly porous photonic crystal biosilica channels, as an innovative lab-on-a-chip platform to detect illicit drugs. The μ DADs in this work are fabricated by spin-coating and tape-stripping diatomaceous earth on regular glass slides with cross section of $400 \times 30 \mu\text{m}^2$. As the most unique feature, our μ DADs can simultaneously perform on-chip chromatography to separate small molecules from complex biofluidic samples and acquire the surface-enhanced Raman scattering spectra of the target chemicals with high specificity. Owing to the ultra-small dimension of the diatomite microfluidic channels and the photonic crystal effect from the fossilized diatom frustules, we demonstrate unprecedented sensitivity down to part-per-billion (ppb) level when detecting pyrene (1ppb) from mixed sample with Raman dye and cocaine (10 ppb) from human plasma. This pioneering work proves the exclusive advantage of μ DADs as emerging microfluidic devices for chemical and biomedical sensing, especially for POC drug screening.

Graphical abstract

*Corresponding author: wang@eecs.oregonstate.edu.

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1. Introduction

In recent years, the escalation of research interests in porous media microfluidics[1, 2], especially microfluidic paper-based analytical devices (μ PADs)[3–5], have fostered a new spectrum of biomedical devices for point-of-care diagnosis and biosensing. μ PADs can be fabricated by simple, low-cost processes using conventional photo- or soft lithographic techniques, utilizing either photoresists[6] or wax printing[7]. Advantages of using μ PADs for microfluidic channels include: 1) ubiquitous and extremely cheap cellulosic materials; 2) capillary flow which enables fluid transport without using any external pump; and 3) compatible with many chemical and biomedical applications. Many different chemical and biological assays have been performed using μ PADs, including for the detection of glucose[8], protein (albumin) [9], cholesterol[10], and heavy metals[11]. They have also been used as platforms for ELISA[12]. Especially, I. M. White's group used inkjet-printed paper-based surface-enhanced Raman scattering (SERS) substrates for chromatographic separation and detection of target analytes from complex samples[13], which opened a new route for on-chip chemical sensing.

Other than μ PADs, porous silica materials and devices also have attracted considerable attention for biosensing due to the use of their large surface area and pore volume to achieve high sensitivity[14, 15]. The high porosity, which allows for the immobilization of target molecules not only on the external surface of the substrate but also inside of the pores, enables the loading of large amounts of sensing molecules, giving instant responses and high sensitivity. The optical transparency, on the other hand, permits optical detection through the bulk of the material. In addition, the surface groups and biocompatibility also makes porous silica one of the most potential materials for biosensing. Moon et al. have fabricated polymer and colloidal silica porous composite for nucleic acid biosensing[16]. Yang et al. have synthesized porous SiO_2 material and used it as enzyme immobilization carriers to fabricate glucose biosensors[17]. However, the pores in sol-gel derived silica lack a high degree of order, which results in random paths and consequently non-uniform diffusion of the analytes. A fraction of the sensing molecules might even be unreachable, leading to low response and poor spatial resolution [18].

Diatoms are unicellular, photosynthetic, bio-mineralized marine organisms that possess a biosilica shell, which is called the frustule. The two-dimensional (2-D) periodic pores on

diatom surface enable it unique optical, physical, and chemical properties [19, 20]. In recent decades, a variety of biosensors with ultra-high sensitivity using diatom biosilica have been reported [21]. Zhen et al. developed photoluminescence-based diatom biosensors that have been successfully applied for 2, 4, 6-trinitrotoluene (TNT) sensing [22]. De Stefano et al. have fabricated highlyselective biosensor for immuno-complex detection by modifying diatom frustules (*Coscinodiscus concinnus*) with antibodies [23]. From the optics perspective, the photonic crystal feature of diatoms could provide additional SERS enhancement when hybridized with plasmonic nanostructures [24, 25]. Our group has developed an in-situ growth method for depositing silver nanoparticles (Ag NPs) on diatom for ultrasensitive, label-free TNT sensing [26, 27]. Other than natural photonic crystal structures from living diatoms, diatomite consists of fossilized remains of ancient diatoms as geological deposits with billions of tons of reserve on earth. Therefore, diatomite is a type of naturally abundant photonic crystal biosilica, which has been widely used in industry as water filters, adsorbents, and medicine[28–30]. Diatomite has similar properties to diatoms such as highly porous structure, excellent adsorption capacity, and photonic crystal effects [31, 32].

In this study, we report microfluidic diatomite analytical devices (μ DADs), which consist of nano-porous photonic crystal biosilica channels for label-free biosensing of illicit drugs from complex biological samples using on-chip chromatography in conjunction with SERS sensing method. Previously, bio-inspired photonic crystals have been integrated into microfluidic systems as lab-on-a-chip system [33] and SERS has been employed for drug sensing [34]. In this research, Cocaine ($C_{17}H_{21}NO_4$) is chosen as the target analyte in our study, which is an alkaloid derived from coca leaves. Cocaine is one of the most widely used illicit drugs all over the world according to the latest World Drug Report from the United Nations Office on Drugs and Crime (UNODC). Cocaine is a potent stimulant of the central nervous system that leads to a state of increased alertness and euphoria. Its effect is similar to that of amphetamines but with shorter duration. In this study, we report using μ DADs for on-chip chromatography-SERS to separate and detect cocaine from real biofluidic samples. The μ DADs achieve nearly 1,000 times better limit of detection (LOD) than normal chromatography plates to 1~10 ppb level, which is comparable or even higher than that of many laboratory analysis techniques[35], which will be discussed in Section 3.6.

2. Materials and methods

2.1. Materials and reagents

Tetrachloroauric acid ($HAuCl_4$) was purchased from Alfa Aesar (USA). Trisodium citrate ($Na_3C_6H_5O_7$), anhydrous ethanol, hexane and acetate were purchased from Macron (USA). Celite209 (diatomite), carboxymethyl cellulose, pyrene, 4-mercaptobenzoic acid (MBA), plasma and cocaine were obtained from Sigma-Aldrich(USA). The chemical reagents used were of analytical grade. Water used in all experiments was deionized and further purified by a Millipore Synergy UV Unit (Millipore-Sigma USA) to a resistivity of $\sim 18.2 M\Omega \cdot cm$.

2.2. Preparation and Characterization of Gold Nanoparticles (Au NPs)

The glassware used through the NP synthesis process was cleaned with aqua regia (HNO₃/HCl, 1:3, v/v) followed by rinsing thoroughly with Milli-Q water. Au NPs with an average diameter of 60 nm were prepared using sodium citrate as the reducing and stabilizing agent according to the literature with little modification[36]. Briefly, a total of 100 mL of 1 mM chloroauric acid aqueous solution was heated to boiling under vigorous stirring. After adding 4.1 mL of 1% trisodium citrate, the pale yellow solution turned fuchsia within several minutes. The colloids were kept under reflux for another 15 min to ensure complete reduction of Au³⁺ ions followed by cooling to room temperature. For practical point-of-care (POC) sensing, the Au NPs will be concentrated by centrifuge and stored in refrigerators with expected life time of more than 1 month.

2.3 Fabrication of μ DADs

The diatomaceous earth substrates were fabricated by spin coating diatomite on glass slides. The diatomite was dried at 150 °C for 6 h in an oven before spin-coating the glass slides. After cooling to room temperature, 11.55 g of diatomite was first dispersed in 20 mL of 0.4% aqueous solution of carboxymethyl cellulose and then deposited onto the glass slide by spin-coating at 1300 rpm for 20 seconds. The porous photonic crystal biosilica channels were fabricated by a simple tape-stripping method as shown in Figure S1: the glass slides were first covered by an adhesive tape; then 400 μ m wide channel array was cut by a razor blade through the tape; after spin-coating with diatomite, the tape was removed gently, leaving 400 \times 30 μ m² cross section diatomite channel array on the glass substrate. Last, the μ DADs were dried in shade and activated at 110 °C for 3 h to improve the adhesion of diatomite to the glass slide.

2.4 μ DADs for on-chip chromatography-SERS biosensing

The on-chip chromatography-SERS sensing method was designed for ultra-sensitive detection of analytes from mixtures or complex biofluid as shown in Scheme 1. First, 0.2 μ L liquid sample was spotted onto the reservoir (circular region) of the μ DAD. After drying in air, the bottom tips of the μ DADs were immersed in the solvent which migrates along the porous channels towards the other end of the μ DADs due to capillary forces. After that, the μ DADs were taken out from the solvent and dried in air. The separated analyte spots along the porous channels were marked under ultraviolet illumination at 380 nm wavelength and visualized by iodine colorimetry. Then 2 μ L of concentrated Au NPs in solution were dropped onto the corresponding spots directly. An alternative method to avoid dispensing the colloid solution is to pre-deposit Au NPs using inkjet printer at the designated spots. However, this process requires precise calibration of the analyte migration rate and will be investigated in our future research. A Horiba Jobin Yvon(USA) Lab Ram HR800 Raman microscope equipped with a CCD detector (uEye cmos, Germany) was used to acquire the Raman spectra, and a 50 \times objective lens (Olympus Mplan, Japan) was used to focus the laser onto the SERS substrates. A 785 nm laser was chosen as the excitation wavelength and the laser spot size was 2 μ m in diameter. The confocal pinhole was set to a diameter of 200 μ m. The acquired data was processed with Horiba LabSpec 5 software. Fluorescence spectra were acquired using the previous method [37]. Briefly, we focus light to a diatom surface

with the 50× objective lens using the Horiba Jobin Yvon Lab Ram HR800 Raman system with 325 nm UV line.

2.5 Other Instruments

UV-vis absorption spectra were recorded on NanoDrop 2000 UV-Vis spectrophotometer (Thermo Scientific USA) using quartz cells of 1 cm optical path. Scanning electron microscopy (SEM) images were acquired on FEI Quanta 600 FEG SEM (Thermo Scientific, USA) with 15–30 kV accelerating voltage. The microscopy images were obtained using Olympus (Japan) IX73 microscope with 20× objective lens.

3. Results and discussion

3.1 Synthesis of Au colloid

SEM and UV-vis absorption spectroscopy were employed to characterize the quality of the prepared Au NPs. The SEM image (Figure S2) indicates that the Au NPs have a spherical shape with uniform size distribution and their diameters are estimated to be 50–60 nm. The UV-vis spectra of Au colloids were shown in Figure S3. The wavelength and intensity of the maximum absorption of the plasmonic NPs depends on the size, shape, concentration and surrounding dielectric environment around the nanoparticles. The localized surface plasmon resonance (LSPR) peak of the prepared Au colloids is located at 549 nm. These values correspond to relatively uniform, mono-dispersed Au colloids with diameters of approximately 50–60 nm.

3.2 Micro- & Nano-structures of the μ DADs

The morphology of μ DADs was characterized by SEM as shown Figure 1(a). The width of the porous diatomite channels was nearly 400 μ m. The reservoir (circular region) with diameter of 1 mm was used for sample dispensing. The porous diatomite channels mainly consist of disk-shaped diatomite biosilica. The morphology of the diatomite biosilica was shown in Figure 1(b) and the 2D periodic pores with sub-micron diameters on diatomite enables guided-mode resonances (GMRs) of photonic crystals[38], which has similar effect to diatom biosilica as we have reported previously [26]. In order to verify the photonic crystal effect of diatomite, the near field optical microscopic image of a single diatomite is shown in Figure S4. The light pattern comes from the high order diffraction of the photonic crystals, which agrees with the results from Stefano's group. [39]. Therefore, the nanostructures of diatomite provide photonic crystal effects, although it may not be perfect. The highly porous structure and uniform pore size of diatomite divide the stationary phase into smaller entities, thus decreasing the length of each diffusional segment paths [40], which enables more homogenous fluid flows into the pores of diatomite. Therefore, the eluent flows more smoothly and uniformly along μ DADs due to capillary forces without any external pump. In order to verify the fluid flow within the porous photonic crystal biosilica channels, 100 ppm pyrene solution was used as the fluidic sample. After fluid flowing, the porous photonic crystal biosilica channels were illuminated by UV light as shown in Figure 1(c). The fluorescence color from pyrene was observed in contrast to the glass substrate, which indicated that the 3D μ DADs porous structure enabled pump-free fluid flow successfully.

3.3 On-Chip Chromatography using the μ DADs

Polychromatic hydrocarbons (PAHs) are a class of aromatic compounds consisting of two or more aromatic or heterocyclic rings. The detection of various PAHs has significant engineering potential as PAHs are harmful to the environment and public health. Unfortunately, the low binding affinity between PAHs and the surface of metallic substrates prevents efficient SERS detection of PAHs from mixtures as the spectra from co-existing components interfere with the SERS spectra from the PAHs[41]. We first investigated the potential of using SERS to detect MBA, Pyrene and their mixture. Figure 2(a) shows the SERS spectra of MBA, pyrene and their mixture. MBA is a commonly used Raman probe molecule because of its strong binding affinity with metallic surfaces and intense Raman signals. The peaks located at 1074 and 1587 cm^{-1} are associated with the C-C ring-breathing modes of MBA[42]. For mixture (Pyrene : MBA=1:1) solution, the metallic surface coverage was dominated by MBA because covalent bonds can be formed easily between the Au NPs and the mercapto group of MBA. Thus only a very weak Raman peak from pyrene was observed from the SERS spectra of the mixture. It is difficult to distinguish the Raman peak of pyrene from the mixture by normal SERS without separating MBA.

When the fluid flows along the μ DADs via the capillary force, the diatomite functions as the stationary phase for chromatography because the abundance of hydroxyl groups on diatomite surfaces make it highly polar. After the mixture sample has been dropped to the reservoirs of μ DADs, the organic eluent flows along the channels. More polar compound molecules will have stronger interaction with the diatomite and will migrate at a slower speed. We first investigated the separation effect of μ DADs with pyrene and MBA mixture. Hexane and ethyl acetate ($v/v = 6:1$) were used as the eluent for the separation of pyrene from the mixture. After complete fluid flow, a UV lamp and iodine colorimetry was used to visualize different analyte spots corresponding to pyrene and MBA. Pyrene migrated faster and was located further from the original dropping point at the reservoir because the lower molecular polarity induces weaker affinity with polar diatomite surface. The SERS spectra at corresponding spots were collected on the surface of μ DADs as shown in Figure 2(b). The characteristic peaks of pyrene at 590 cm^{-1} and 1230 cm^{-1} are clearly observed, which means that the μ DADs can successfully be used as the stationary phase for on-chip chromatography.

3.4 Confinement of the analyte by μ DADs

In general, the intensity of SERS signals $I_{SERS(vs)}$ can be estimated as [43,44]:

$$I_{SERS}(V_S) \propto N_M \times |A(V_L)|^2 \times |A(V_S)|^2 \times \delta_{ads}^R \quad \text{Equation (1)}$$

where N_M is the number of molecules involved in the SERS measurement, δ_{ads}^R is the Raman cross section of the molecule that is being detected, and $A(V_L)$ and $A(V_S)$ are the electrical field enhancement factors at the excitation laser and Stokes frequency for the Raman signal enhancement. These parameters usually are intrinsic factors which are nearly constant for the same SERS substrate and the target molecules other than N_M [44]. In most on-chip

chromatography SERS devices, the plasmonic nanoparticles are dispensed onto the analyte spots after chromatography separation. The SERS spectra collected from each spot will only come from the target molecules at the surface of the chromatography chip. This means that the overall SERS intensity will be dependent on the amount of target molecules in close proximity to the plasmonic NPs at the surface of the chromatography plate. We have reported previously that thinner diatomite layers will effectively concentrate the analyte at the surface of the chromatography plate[45]. Compared with thin film plate, the μ DADs we fabricated can confine the liquid flow within a $400 \times 30 \mu\text{m}^2$ cross section channel, which significantly enhances the target molecule concentration at the surface of μ DADs.

The confinement of the analyte molecules by μ DADs was investigated by fluorescence microscopy and spectra. First, 0.2 μL and 1 μL of pyrene solution with 200 ppm, 20 ppm and 2 ppm concentrations was dispensed onto the μ DADs and normal diatomite chromatography plate respectively. After eluent migrating, the substrate was illuminated by a UV laser, as shown in Figure 3(a). From the optical images, we can observe fluorescence spots on the two chips. For 20 ppm concentration of pyrene, the fluorescence spot from μ DADs chip is brighter than that from normal chromatography plate. With the concentration of pyrene down to 2 ppm, fluorescence spot on μ DADs was still obvious while there was no observable fluorescence spot from the normal chromatography plate.

Such confinement effect to target molecules was also confirmed by fluorescence spectra as shown in Figure 3. The samples used to acquire the fluorescence spectra were the same as those for the fluorescence images. In Figure 3(b), the intensity of fluorescence spectra of pyrene decreases with reduced pyrene concentration. When the concentration of pyrene is 2 ppm, only weak fluorescence spectra of pyrene were observed. As shown in Figure 3(c), the fluorescence spectra of pyrene from μ DADs at the spot of 2 ppm pyrene still showed intense fluorescence signals. In principle, the number of pyrene molecules on the normal chromatography chip (1 μL) should be higher than spotted onto μ DADs (0.2 μL), but the intensity of the fluorescence spectra was in the opposite manner. A smaller amount of pyrene in the μ DADs shows higher fluorescence intensity than that from 20 ppm pyrene on normal chromatography chip, therefore it proves that narrow micro-channels have stronger confinement effect of target molecules.

3.5 Ultrasensitive on-chip sensing of pyrene from mixture by μ DADs

The μ DADs were employed for sensing pyrene from mixture. We compared the SERS spectra obtained from the μ DADs as shown in Figure 4(a) and (b) with normal diatomite chromatography chip (Figure S5). In Figure 4, all the characteristic bands of MBA and pyrene exhibited monotonous decrease in intensity as the mixture concentration decreases. The detection limit from pyrene/MBA mixture is down to 1 ppb on the μ DADs, and only 2 ppm on the normal diatomite chromatography plate (Figure S5). The experimental results demonstrate more than 1,000 times improvement of the detection sensitivity using the μ DADs compared to normal diatomite chromatography plates. We attribute this dramatic improvement to the strong microchannel confinement of the fluid flow, which prevents the lateral diffusion of the target molecules within the μ DADs.

3.6 On-chip sensing of cocaine from biofluid

In the case of sensitive detection of cocaine, the current detection platforms are gas chromatography (GC) [46], high performance liquid chromatography (HPLC) and chromatography in tandem with mass spectrometry (MS) [47] [48]. Although the chromatography and mass spectrometry are accurate and reliable, they are expensive, time-consuming and require skilled personnel. Although accurate lab analysis techniques are available, instant, cost-effective and ease-of-use methods for on-site testing of cocaine from biofluid such as saliva, plasma and urine are yet to be developed for forensics and medical diagnosis. Here, μ DADs were employed for on-chip detection of cocaine from human plasma. Cyclohexane and ethanol (v/v = 6:1) were used as the eluent for the separation of cocaine from plasma. In our experiment, cocaine was intentionally added into human plasma to obtain different concentrations (10 ppb to 100 ppm). The macromolecules such as albumin and enzymes in plasma cannot diffuse on the μ DADs due to the high molecular weight. Good separation and detection of cocaine is achieved using μ DADs. The SERS spectra were shown in Figure 5. The Raman peak at 1008 cm^{-1} was assigned to the aromatic ring breathing of cocaine. As shown in Figure 5(a), the characteristic band of the cocaine exhibited a monotonous decrease in intensity following the decrease of the cocaine concentration in plasma. The detection limit for cocaine in plasma, which is defined as the signal-to-noise (SNR) ratio of 3 as marked by the blue line in Figure 5(b), was 10 ppb using the μ DADs. As a comparison, Brunetto et al. have developed column-switching LC method for cocaine detection, in which the LOD could achieved 80 ppb[49]. The 10 ppb LOD from μ DADs is even better than that from the aforementioned LC laboratory analysis method. According to the report by National Highway Traffic Safety Administration[50], smoking 50 mg of cocaine would result in peak cocaine concentration in plasma at 230 ppb after 45 minutes, and the half-life-time for cocaine is approximately one hour. Therefore, our μ DADs is sensitive enough to monitor cocaine from blood serum five hours after cocaine abusing.

4. CONCLUSIONS

In this pilot study, we have developed a new type of microfluidic devices, μ DADs, for ultrasensitive, label-free, ease-of-use and rapid sensing of illicit drugs from complex biofluidic samples. The μ DADs are fabricated via a simple method by spin-coating and tape-stripping diatomite on glass. The μ DADs can simultaneously separate small molecules from the complex background and acquire the SERS spectra of the target chemicals with high specificity after the deposition of plasmonic nanoparticles. Furthermore, the μ DADs exhibit extremely high confinement of the analyte due to the ultra-small dimension of the diatomite microfluidic channels, which effectively increase the concentration of target molecules at the sensor surface. The experimental results achieved ultra-high detection sensitivity down to 1 ppb, which represents an improvement factor of more than 1,000 times when compared to the normal chromatography plate device. To demonstrate the significant engineering potentials for forensic sensing, we have achieved ultra-sensitive detection of cocaine in human plasma with LOD of 10 ppb, which is even better than many laboratory analytical methods such as HP-LC and GC-MS. Such facile μ DADs using hybrid plasmonic-diatomite biosilica, as a new type of cost-effective and ultra-sensitive microfluidic devices with

multiplex sensing capabilities, will play a pivotal role in chemical and biological sensing, especially for POC drug screening.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

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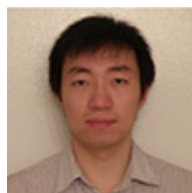
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Biographies



Xianming Kong is an Associate Professor at the College of Chemistry, Chemical Engineering and Environment Engineering at Liaoning Shihua University, Fushun, P. R. China. He was a Postdoctoral Scholar at the School of Electrical Engineering and Computer Science at Oregon State University from 2015 to 2017. He received his PhD degree in Physical Chemistry from Nanjing University, Nanjing, China, in 2012. From 2012–2015, he worked as a postdoctoral researcher in the School of Chemical Technology, Aalto University, Finland. His current research work focuses on the development of surface-enhanced Raman scattering (SERS) sensors for chemical and biological sensing.



Xinyuan Chong received his Ph.D. degree from the School of Electrical Engineering and Computer Science at Oregon State University in 2017. He received his B.S. and M.S. in Physics from Tsinghua University in 2008 and 2011 respectively. His current research interests include plasmonic-enhanced optical sensor for chemical detection, gas sensing, and biomedical applications.



Kenny Squire is a PhD candidate in the School of Electrical Engineering and Computer Science at Oregon State University. He received his B.S. degree in Electrical Engineering from Brigham Young University in 2015. His research focuses on optical biosensors utilizing surface plasmon resonance and surface-enhanced Raman spectroscopy.



Alan X. Wang is an Associate Professor of the School of Electrical Engineering and Computer Science at Oregon State University since 2011. He received his Ph.D. degree in Electrical and Computer Engineering from the University of Texas at Austin in 2006. From 2007–2011, he was with Omega Optics, Inc., where he served as the Chief Research

Scientist for 9 SBIR/STTR projects. His research interests include nanophotonic devices for optical interconnects, and optical sensors for chemical and biological detection. His current research activities are sponsored by the National Science Foundation, the National Institutes of Health, Oregon Nanoscience and Microtechnologies Institute, the National Energy Technology Laboratory, and industrial sponsors such as Hewlett-Packard. He has more than 70 journal publications and 70 conference presentations, and also holds three U.S. patents. He is a senior member of IEEE Photonics, SPIE and OSA.

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Research Highlight

- A microfluidic analytical device based on photonic crystal biosilica micro-channel array
- Ultra-high sensitivity for illicit drug sensing down to 1~10 ppb in human plasma
- Cost-effective platform for point-of-care applications

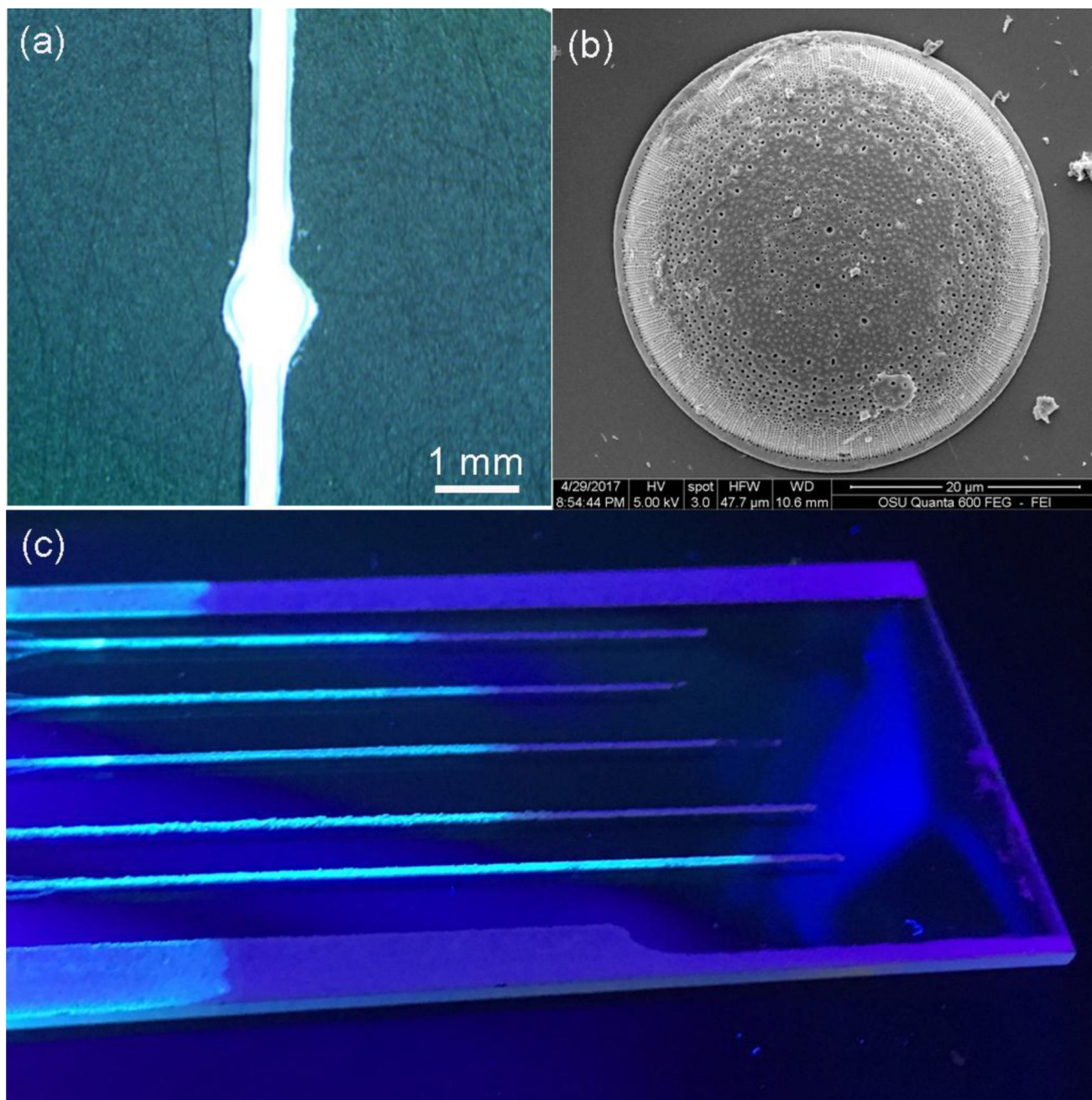


Figure 1. Optical image of the μ DADs (a) and SEM image of the honeycomb-like diatomite (b), which forms the micro-channels of μ DADs, and optical image of μ DADs after 100 ppm pyrene migration illuminated by UV light (c)

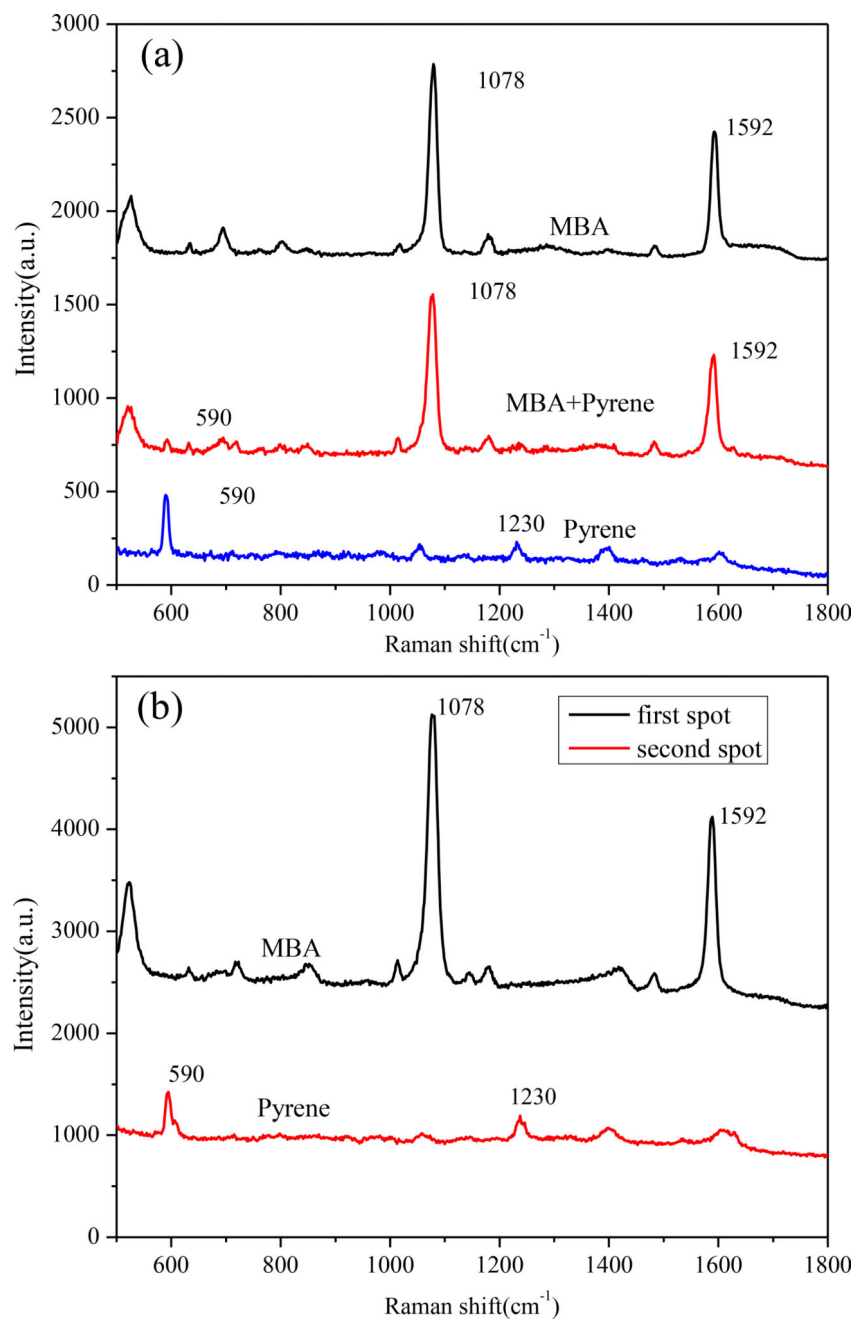


Figure 2. (a) SERS spectra of pure substance of MBA, pyrene and the mixture; and (b) SERS spectra of different spots on μDADs after chromatography separation

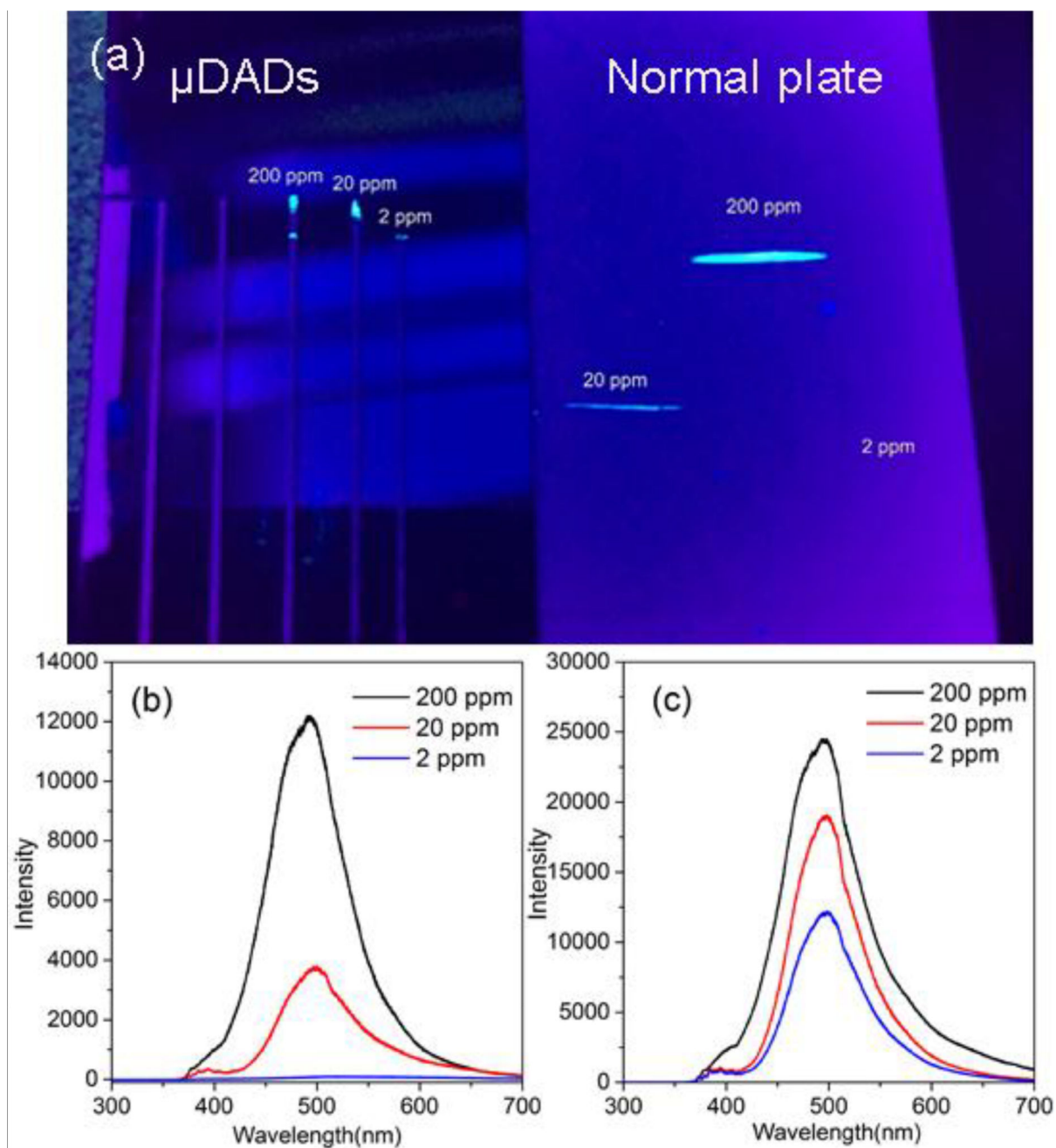


Figure 3.

(a) Photographic images of different concentrations of pyrene separated by μ DADs and normal diatomite plates. The spots after separation are visualized by UV light; fluorescence spectra of different concentrations of pyrene separated by normal diatomite chromatography plates (b) and μ DADs (c).

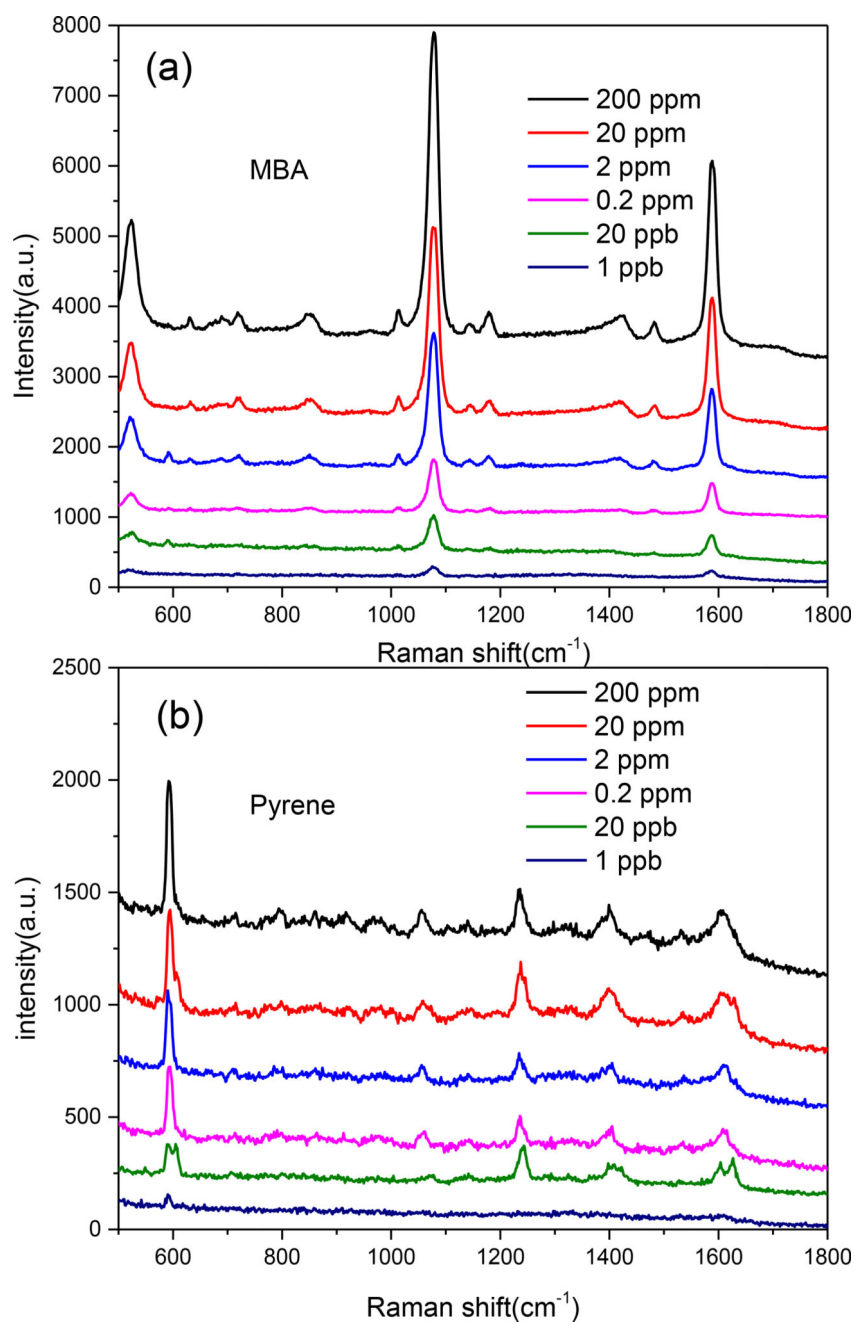


Figure 4. SERS spectra of MBA on the first spot (a) and pyrene on the second spot (b) from mixture (pyrene : MBA = 1:1) at different concentrations separated by μ DADs

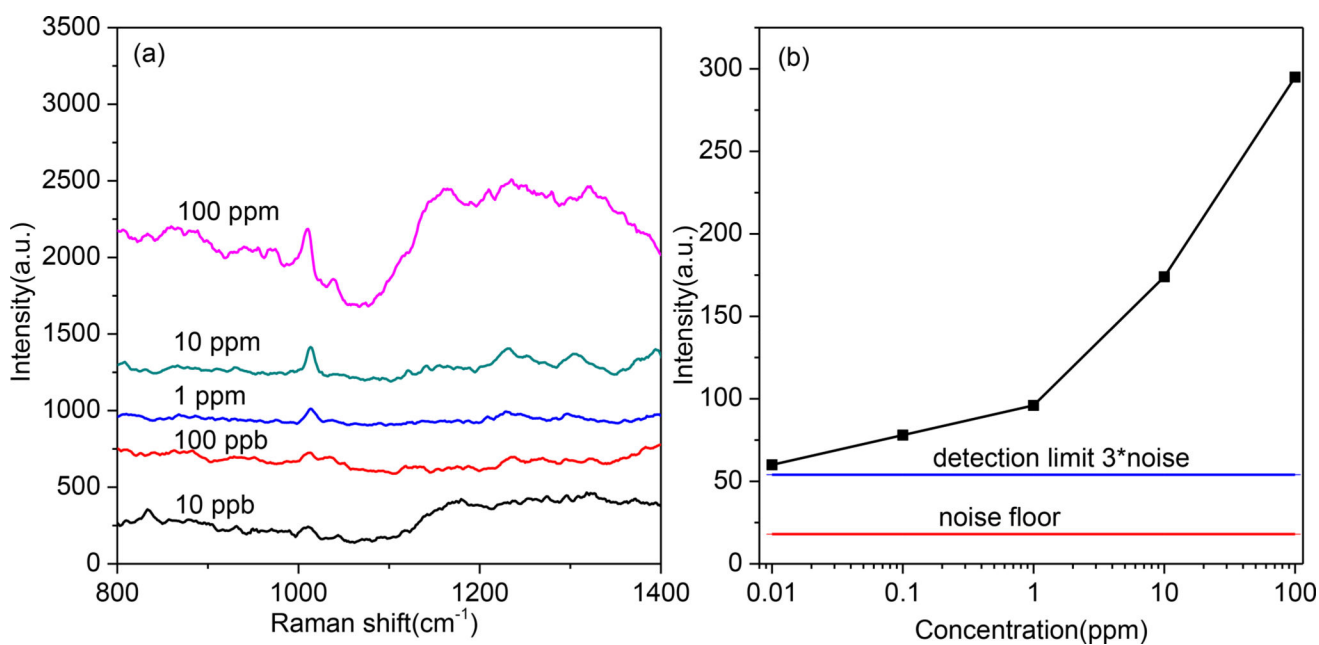
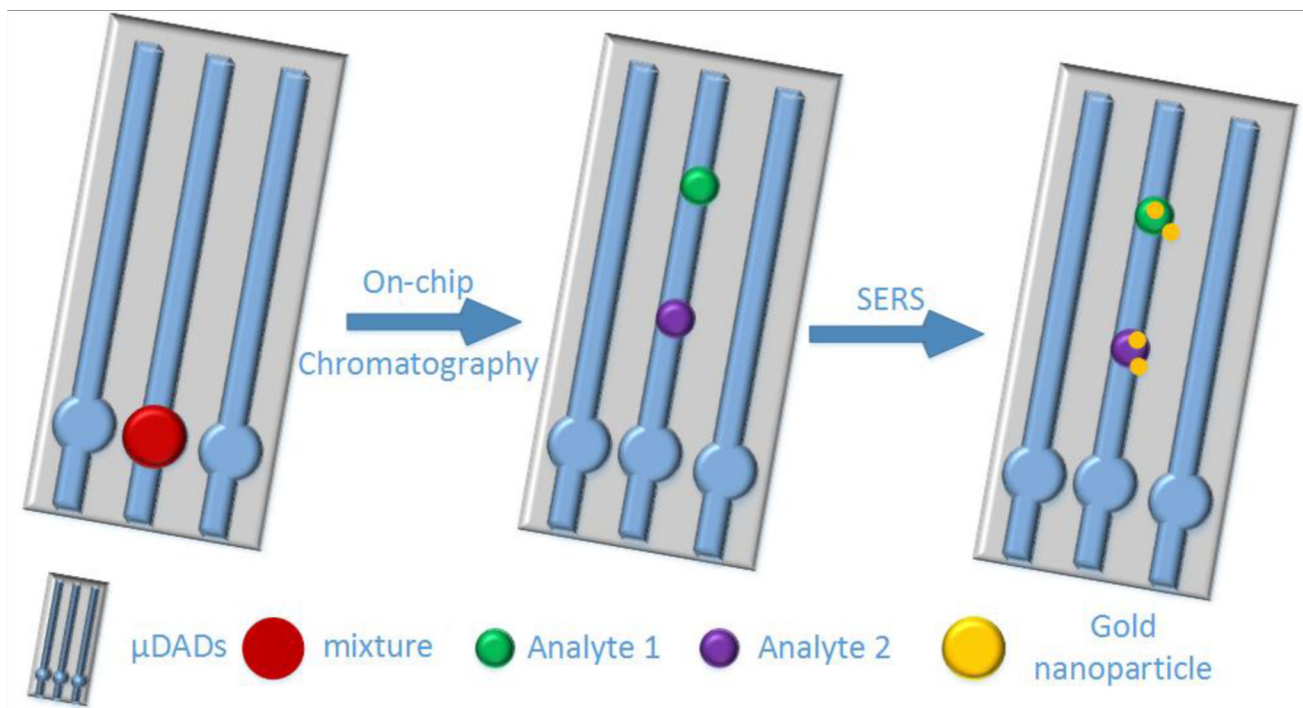


Figure 5. SERS spectra of human plasma with different concentrations of cocaine separated by μ DADs (a) and the SERS intensity as a function of logarithm scale Cocaine concentration in Plasma (b). The detection limit was indicated as the line in (b).



Scheme 1.
Schematic illustration of on-chip chromatography-SERS biosensing using the proposed μDADs