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Structure activity study of carbonic anhydrase IX: Selective inhibition with ureido-substituted benzenesulfonamides

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Abstract

Ureido-substituted benzenesulfonamides (USBs) show great promise as selective and potent inhibitors for human carbonic anhydrase hCA IX and XII, with one such compound (SLC-0111/U-F) currently in clinical trials (clinicaltrials.gov, NCT02215850). In this study, the crystal structures of both hCA II (off-target) and an hCA IX-mimic (target) in complex with selected USBs (U-CH₃, U-F, and U-NO₂), at resolutions of 1.9 Å or better, are presented, and demonstrate differences in the binding modes within the two isoforms. The presence of residue Phe 131 in hCA II causes steric hindrance (U-CH₃, 1765 nM; U-F, 960 nM; U-NO₂, 15 nM) whereas in hCA IX (U-CH₃, 7 nM; U-F, 45 nM; U-NO₂, 1 nM) and hCA XII (U-CH₃, 6 nM; U-F, 4 nM; U-NO₂, 6 nM), 131 is a Val and Ala, respectively, allows for more favorable binding. Our results provide insight into the mechanism of USB selective inhibition and useful information for structural design and drug development, including synthesis of hybrid USB compounds with improved physiochemical properties.

Keywords

Benzenesulfonamides; Carbonic anhydrase IX; hCA IX-Mimic; Structure-activity relationships; X-ray crystallography; Cancer therapy

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Accession codes

Coordinates and structure factors for all structures have been deposited to the PDB with the following accession codes: 5JN1, 5JN3, 5JMZ, 5JN7 and 3N4D.

Author contributions

M.Y.M. wrote the paper, performed and analyzed the experiments. B.P.M., N.L., and L.S. performed and analyzed the experiments. F.C., and C.T.S. designed and synthesized the compounds. S.C.F., and R.M. designed the study and reviewed the paper. All authors discussed the results and approved the final version of the manuscript.

Appendix A. Supplementary data

Supplementary data related to this article can be found at <http://dx.doi.org/10.1016/j.ejmech.2017.03.026>.

1. Introduction

A commonly observed feature of solid tumors is regions comprised of proliferating and highly metabolic cells. These cells typically display a hypoxic phenotype, which leads to increased acidity in the tumor microenvironment and a metabolic shift that favors a progression to metastasis [1]. This hypoxic phenotype results when tumor cells outgrow their blood supply leading to low oxygen levels, and in response, undergo oncogenic alternations in their metabolism to over express proteins essential for survival [1,2]. Human Carbonic Anhydrase (hCA IX; EC 4.2.2.1) is one such protein, which is mainly induced by hypoxia and functions as a pH regulatory factor [3]. The expression of hCA IX has been shown to be crucial for tumor cell survival and proliferation due to its ability to normalize the intracellular pH (pHi), while maintaining low extracellular pH (pHe) [4]. This extracellular acidification of the tumor microenvironment promotes local invasion and metastasis, while decreasing the effectiveness of adjuvant therapies. These features contribute to poor clinical outcomes in cancer patients [5]. Although hCA XII expression is not modulated by hypoxia, it is also shown to be upregulated in many aggressive tumors and, similar to hCA IX, it functions to regulate both intra and extracellular pH.

Like other α -CAs, hCA IX and hCA XII belong to a family of zinc metalloenzymes that catalyze the reversible hydration of carbon dioxide to bicarbonate and a proton via a ping-pong mechanism, a key reaction for pH regulatory mechanisms [6]. In contrast to other hCA isoforms, hCA IX expression in normal cells is limited to the stomach and gastrointestinal tract. However, it is upregulated in many solid and aggressive tumors including the brain, breast, bladder, cervix, colon, colorectal, head and neck, pancreas, kidney, lung, ovaries, stomach and oral cavity [7]. Its overexpression in cancer is linked to hypoxia, extracellular acidification, chemotherapeutic resistance, poor prognosis and clinical outcome [4,8]. Conversely, hCA XII is expressed in many normal tissues including pancreas, colon and breast. Because the expression of hCA IX is increased in hypoxic tumors and plays a role in tumor progression, acidification, and metastases, it already serves as a target for imaging and predicting therapeutic outcome [4,9]. hCA XII may also serve as a therapeutic target because of its elevated regulation in cancer. Further, the extracellular location and hCA IX's limited expression in normal tissues, makes both isozymes favorable for drug targeting and delivery in aggressive cancer treatment [7,10]. Indeed, inhibition of hCA IX and hCA XII activity in tumor cells has been shown to reduce tumor growth and proliferation, thus establishing hCA IX and potentially hCA XII as anti-cancer drug targets [9,11].

Several classes of carbonic anhydrase inhibitors have been developed and are used clinically for treatment of diseases [9,12]. However, most of these inhibitors are not hCA IX and hCA XII specific which will reduce their efficacy or produce off-target side effects when used as anti-cancer agents due to inhibition of other carbonic anhydrases (particularly that of the ubiquitously expressed cytosolic hCA II) [9,13]. Difficulties in designing and developing hCA IX/hCA XII-specific inhibitors stem from the fact that the 15 human CA isoforms are all structurally homologous with many conserved residues residing within the active site [14]. To this extent, a structure-based drug design approach has shown to be useful for the development of isoform selective hCA IX and hCA XII inhibitors. Specifically, using compounds that contain a zinc-binding group (ZBG) coupled to a variable chemical moiety,

known as the compound “tail” can facilitate specific interactions with specific residues within the active site. This is the predominant method for developing selective hCA IX inhibitors – a method known as the “tail approach” [14].

Recently, ureido-substituted benzenesulfonamides (USBs) have been shown to selectively inhibit hCA IX and XII [11]. This class of inhibitors also showed both anti-proliferative and anti-metastatic activity in mouse models of breast cancer [11]. One of these USBs, name SLC-0111, is in Phase 1 clinical trials (see, [clinicaltrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT02215850), NCT02215850) for the treatment of solid tumors over expressing hCA IX and XII (Table 1) [11,15]. Despite the success of these compounds, the structure-activity relationships (SARs) of these inhibitors have not been extensively studied, a necessity towards improving USBs for “next generation” hCA IX and XII inhibitors.

Here, we provide insights into the SARs of USBs in hCA IX compared to hCA II and identify important interactions between these compounds and hCA IX and XII, which provides the mechanism(s) for their selective inhibition. Further, we utilize this information to suggest ways to improve this class of compounds to increase both selectivity and potency towards hCA IX and XII. These include addition of an NO₂ group to improve hCA IX specificity and/or addition of F/CH₃ groups to improve specificity, solubility and physiochemical properties. We utilize three USB derivatives with varying chemical properties denoted, U-CH₃, U-NO₂, and U-F (SLC-0111) (Fig. 1) and determined their X-ray crystallography structures in complex with hCA II and the hCA IX-mimic (hCA II with active site substitutions Ala65Ser, Asn67Gln, Glu69Thr, Ile91Leu, Phe131Val, Lys170Glu, and Leu204Ala to closely resemble the active site of hCA IX; hCA II numbering) [16]. Information from these studies provide a foundation for SARs between clinically used USBs and hCA IX and will be of value in the design of next generation inhibitors for the treatment of aggressive cancers.

2. Results

2.1. Selective hCA IX and XII inhibition

A series of USBs have been designed and synthesized previously by Pacchiano et al. [11,17] The rationale for their design was influenced by the observations of hCA isoform specific inhibition and the dependence on the substitution pattern of the urea moiety. The general structure includes a primary sulfonamide ZBG conjugated to benzene, which incorporates a 4-substituted ureido moiety and an R group(s) present in the tail region (Fig. 1, Supplemental Fig. S1) [11]. The 3 inhibitors investigated here were U-CH₃ (4-{{[3,5-methylphenyl]carbamoylamino} benzenesulfonamide) where R is two methyl groups (Fig. 1B, Supplemental Figs. S1A and S1B), U-F (4-{{[4-fluorophenyl]carbamoyl]amino} benzenesulfonamide) where R is 4-fluorophenyl (Fig. 1C, Supplemental Figs. S2C and S2D), and U-NO₂ (4-{{[3-nitrophenyl] carbamoyl] amino} benzenesulfonamide) where R is 3-nitrophenyl (Fig. 1D, Supplemental Fig. S1E) [11c]. All three USBs showed selective inhibition of hCA IX and XII over hCA II and displayed variable selectivity ratios with dependency on the substituted R group (Table 2). U-NO₂ is the most potent hCA IX inhibitor with a K_i of 1 nM and also a good inhibitor of hCA XII with a K_i of 6 nM, but also the least selective against hCA II with II/IX and II/XII

ratios of 15 and 2, respectively. U-CH₃ is also a potent inhibitor of hCA IX and XII, with K_i's of 7 and 6 nM, respectively and displayed the least specificity for hCA II with a K_i value of 1765 nM and showed similar selectivity ratios of 250 and 300 for hCA IX and hCA XII, suggesting that the increased hydrophobicity (with the CH₃ substituents) provides specificity for tumor-associated isozymes over the ubiquitously expressed hCA II. U-F showed ~2-fold higher specificity compared to U-CH₃ for hCA XII (K_i = 4 nM) and good affinity for hCA IX (K_i = 45 nM) compared to hCA II (K_i = 960 nM), the selectivity ratios are 240 and 20 fold for hCA IX and hCA XII, respectively. Based on the ability of U-F and U-NO₂ to selectively bind to hCA IX (and XII) over hCA II, these inhibitors were further studied *in vivo* using mouse models of metastatic breast cancer that overexpress hCA IX [11]. The inhibitors were shown to suppress tumor growth in a concentration-dependent manner and effectively attenuate formation of metastasis *in vivo*. The authors for this study indicated that administration of the compounds did not result in significant weight loss, suggesting that the inhibitors were not generally toxic [11]. These observations may be attributable to the observed increased isoform selectivity by these inhibitors as shown by CA inhibition studies performed using applied photophysics stopped-flow technology (Tables 1 and 2) [11,17].

2.2. X-ray crystallographic results

Structures of the three ureidosulfonamide compounds (U-CH₃, U-F and U-NO₂) in complex with both hCA II and the hCA IX-mimic were determined using X-ray crystallography to resolutions of 1.9 Å or better (statistics summarized in Table 3). Unambiguous omit (Fo-Fc) electron density maps were apparent for all compounds within the active site of both hCA II and the hCA IX-mimic, and showed a 90% occupancy in the final refined structures (Supplemental Fig. S1). As commonly observed with sulfonamide-based hCA inhibitors, the compounds were shown to block hCA catalysis through binding directly to the catalytic zinc of both hCAs displacing the zinc-bound OH⁻/H₂O via the primary sulfonamide group [18]. For all compounds the benzene (C1–5) sulfonamide core are within Van der Waals distances from the conserved side chains of Leu 198, Val 121, Val 143 and Thr 200 (Figs. 2 and 3). Additional electron density consistent with a bound glycerol molecule, from the cryoprotectant solution used during data collection, was observed adjacent to the aromatic ring of the benzene sulfonamide in the active site of hCA II but not hCA IX-mimic (data not shown) [19].

2.2.1. Compound U-CH₃—A structural comparison of the binding of U-CH₃ within the active sites of hCA II and IX (mimic) demonstrated distinct differences which most likely contribute to the ~250-fold selectivity of binding for hCA IX over hCA II (Table 2, Figs. 2A and 3A). Contributing to this conformational change in U-CH₃ is residue Phe 131 in hCA II, which reduces the accessible opening of the active site and acts as a binding anchor to the tail (Fig. 2A). In hCA IX, this residue is Val 131 which opens the active site cavity (Fig. 3A). Having fewer constraints on the conformational torsion angles in hCA IX allows the tail of U-CH₃ to rotate (Table 4) and release a trapped solvent molecule in a hydrophobic pocket (near Leu 198 when bound in hCA II). This ~90° rotation also permits the oxygen of the ureido group to H-bond with the nitrogen of Gln 92 (2.8 Å), and Leu 91 and Val 131 to form specific hydrophobic interactions with the CH₃ substituted R groups (Fig. 3A). These

observations, taken together, explain the increase of interactions for U-CH₃ bound in hCA IX compared to hCA II, measured using PDBePISA [20], and therefore the ~250-fold selectivity of binding for hCA IX over hCA II (Table 2 and Supplemental Table 1).

2.2.2. Compound U-F—The change in conformational rotamers for U-F binding between hCA IX over hCA II were larger than for U-CH₃, with the fluorophenyl moiety rotated more than 100° towards Pro 202 in hCA II (because of the jutting Phe 131) (Table 4). Compound U-F therefore becomes sandwiched between the two groups in hCA II, whereas in hCA IX the tail sits in the hydrophobic pocket created by Val 131 and Gly 132 (Fig. 3A and B). The binding of U-F to hCA II does not result in the displacement of all the active site waters some of which are shown to participate in hydrogen bonding. Lack of solvent displacement most likely contributes to the lower K_i for U-F binding to hCA II compared to U-CH₃ and also reduces its selectivity for hCA IX (Table 2, Figs. 2B and 3B).

2.2.3. Compound U-NO₂—The crystal structures clearly show why compound U-NO₂ is a good inhibitor for both hCA II and IX, as the compound shows little conformational change in either active site (Table 4). Residues Val 121, Val 143, and Leu 198, interact with the core benzene (Figs. 2C and 3C). Additional hydrophobic interactions are also observed between Phe 131 and Leu 91 and Val 131 and the nitrophenyl ring of U-NO₂ in hCA II and IX, respectively (Figs. 2C and 3C). In addition, the presence of Phe 131 in hCA II does not distort the tail conformation, which allows U-NO₂ to directly H-bond with Gln in hCA II, by a bridging water molecule in hCA IX. It should be noted that the binding of U-NO₂ within the active site of hCA II has been previously reported [11c,17]. But, the present comparison with hCA IX now demonstrates why the K_i's are similar and why there is little to no isoform selectivity between hCA II and IX (Table 2).

3. Discussion

3.1. Structure activity relationships of the hCA active site

The ureido linker present between the sulfamoylphenyl fragment and the R moiety of the ureidosulfonamide inhibitors allow for rotatable R tails (Table 4). This permits the compounds to adopt a range of conformations and participate in diverse interactions within the enzyme active site, with no significant conformational changes observed for the active site side chains, other than the removal of the bulky hydrophobic residue Phe 131 (Figs. 4 and 5). For instance, U-CH₃, which has a K_i for hCA II of 1765 nM and a significantly lower K_i for hCA IX of 7 nM, adopts a significantly different conformation upon binding the hCAs, which is most likely due to Phe 131 in hCA II (Fig. 4A and Table 4). Superposition of the structures of hCA IX-mimic and hCA II in complex with U-CH₃ clearly shows Phe 131 in hCA II would cause steric hindrance if it bound in the same conformer in hCA IX, hence the conformational changes in both compound U-CH₃ and U-F and the interactions that can be formed (Fig. 4). The structure of hCA II bound to U-CH₃ (average B factor of 27.7 Å²) showed weak interactions and stabilization of the tail moiety by Phe 131 and Gly 132, instead of Leu 91 and Val 131 as seen in the crystal structure of hCA IX-mimic bound U-F (average B factor of 13.4 Å²) (Figs. 2A and 3A). The difference in average B factors between bound compound UF in the active site of hCA II and hCA IX-mimic is therefore a

good indicator of the conformer stability of the compound specific to each isoform. To this extent it is assumed that different favorable/unfavorable contacts between the inhibitor tail moiety and the enzyme active site lead to different inhibition profiles for this class of sulfonamides. This phenomenon was also observed in the superimposed structures of U-F bound hCA II and hCA IX-mimic in which a $\sim 90^\circ$ shift of the fluorophenyl tail moiety in the active site of hCA II occurred (Fig. 4B). The conformational change also caused differences in hCA II residues interacting and possibly further stabilizing the tail moiety of U-F. The observed interacting hCA II residues are Phe 131 and Pro 202 and for hCA IX-mimic, which has a lower K_i value (hCA IX-mimic = 45 nM and hCA II = 960 nM), are Val 131 and Gly 132 (Figs. 2B and 3B). In the superimposed structures of U-NO₂ bound hCA II and hCAIX-mimic, little to no conformational change of the inhibitor was detected (Fig. 4C). This observation correlates to the reduced isoform selectivity of this compound when compared to U-CH₃ and U-F. In summary, our results suggest a close correlation between the previously determined kinetic and our X-ray crystallographic data with the selectivity profiles of benzenesulfonamides incorporating ureido tails. The flexibility of the ureido linker allows for variable conformations of the tail to be adapted providing acceptable isoform specificity and selectivity for these compounds, which are desirable for a drug candidate. It is also clear from this and previous studies that designing isoform selective hCA IX inhibitors requires the exploitation of residues in the CA selective pocket (15 Å away from the active site zinc), which confer the desired hCA IX and hCA XII specificity (Table 5) [14,21]. It is predicted that the reason for the preferred inhibitor interactions with the hydrophobic and not the hydrophilic pocket is the absence of bulky residues in the hCA IX and hCA XII active sites such as Phe 131 that reduce steric hindrance of the compound (Table 5) [14,21]. In hCA IX, residue 131 is a Val and hCA XII has an Ala at that site (Table 5). Although both residues are relatively small with short hydrophobic side chains, Ala is also considered nucleophilic. These small residues allow the compounds to freely enter the active site in a conformation that favors interactions with residues in the hydrophobic pocket to induce selectivity. The residue differences between hCA IX and hCA XII, but especially at site 131, may also confer specificity for one membrane bound isoform over the other. These observations, when taken together, provide necessary parameters for designing potent hCA IX and hCA XII specific and selective inhibitors for the treatment of cancer.

4. Conclusions

Its limited expression in normal tissue, its extracellular catalytic domain, its upregulation in many aggressive tumors, and a therapeutically favorable response when inhibited, rationalize targeting hCA IX for cancer treatment [7,22]. Specifically, inhibition of the catalytic activity of hCA IX has been shown to decrease tumor growth, prevent metastases and increase overall survival rates [4,8]. Thus, targeting hCA IX activity in tumors that overexpress this enzyme provides a useful approach for treatment of several aggressive cancers. Here, we present novel structural information on a series of ureido-substituted benzene sulfonamide based (USB) inhibitors previously determined as selective inhibitors of hCA IX and hCA XII over off-target hCA II and I [11]. Our findings identify key interacting residues and pinpoint the advantage of using compounds with flexible linker regions. We have shown in this study how ureido sulfonamide based inhibitors may provide an avenue to overcome CA

isoform specificity, as they display both nanomolar affinity and preferential binding for the tumor associated membrane bound isoforms. For instance, U-CH₃, which is the most selective of the three inhibitors, exhibits greater than a 200-fold increase in selectivity for hCA IX and hCA XII over hCA II. Compound U-NO₂ is the most potent inhibitor of hCA IX, but it is also the least isozyme selective. U-F displays increased potency for hCA XII. The crystal structures of U-CH₃, U-F and U-NO₂ bound to the active site of hCA IX in comparison to hCA II show that residue 131 may play a critical role in determining compound selectivity. As previously mentioned, the difference in residues at position 131 in the active site of hCA IX (Val 131) and hCA II (Phe 131) cause conformational changes in the inhibitors. These changes dictate the types of interactions that can be formed to stabilize the inhibitors within the active site of hCAs. Furthermore, the introduction of strong electron-withdrawing groups (3-NO₂ and 4-F) or highly electron-donating (3,5-dimethyl) causes a polarization within the entire molecule and extended conjugation system. This also means better solubility of these compounds in aqueous solution.

In designing the next generation of hCA IX and hCA XII inhibitors that incorporate substituted benzene sulfonamide moieties, it is important that the chemical properties of these inhibitors also be considered. For instance, compound U-NO₂ is the most potent hCA IX inhibitor but it is also the least soluble among the three inhibitors. To improve its solubility/selective properties, a hybrid compound could be synthesized, in which either one or multiple F and/or CH₃ group(s) are incorporated into the terminal benzene ring, with improved Van der Waal contacts. This may increase its hydrophilicity and perhaps isoform selectivity. Another approach might be the addition of an NO₂ at position C1/C2/C3 of the substituted benzene moiety in U-CH₃ or U-F compounds. These substitutions or others may aid in the development of more potent hCA IX and hCA XII specific and selective inhibitors with improved physiochemical properties. In summary, these studies provide new insight into USB selective CA inhibition and for the first time “map” key residue interactions of the hCA IX active site in complex with USBs [21]. This information will provide important parameters needed to design and develop antiproliferative and antimetastatic drugs that target hCA IX.

5. Experimental section

5.1. Protein expression, purification and CA IX-mimic design [23]

HCA II and hCA IX-mimic cDNA were transformed into *Escherichia coli* BL21 (DE3) competent cells in Luria broth medium containing ~0.1 mg/ml ampicillin at 37 °C to an optical density of ~0.6 at 600 nm. Protein expression was induced using 0.1 mg/ml isopropyl β-D-1-thiogalactoside (IPTG) and 1 mM zinc sulphate. The cells were incubated for an additional 3 h and harvested by centrifugation. Purification was performed using affinity chromatography, proteins were eluted (400 mM sodium azide, 50 mM Tris, pH 7.8), buffer exchanged (into 50 mM Tris-HCl, pH 7.8) and then concentrated via centrifugal ultra-filtration using a 10 kDa MWCO filter. Protein purity was checked by SDS-PAGE (stained with coomassie blue) and concentration determined by UV/Vis spectroscopy, and measured to be 29 and 20 mg/ml for hCA II and IX-mimic, respectively. The concept of the engineered hCA IX-mimic was conceived by Genis et al., and developed further by Pinard et

al., and constructed by site-directed mutagenesis of the active site of hCA II [23]. The substituted residues in the hCA IX-mimic are analogous to wild type hCA IX, generating an extremely stable construct that readily crystallizes [10,16,23]. Active site mutations in the hCA IX-mimic include; Ala65Ser, Asn67Gln, Glu69Thr, Ile91Leu, Phe131Val, Lys170Glu, and Leu204Ala.

5.2. X-ray crystallography

Crystals of purified hCA II and hCA IX-mimic were obtained using the hanging drop vapor diffusion method, with 1.6 M Na-Citrate, 50 mM Tris, and pH 7.8 as the precipitant solution and incubated at room temperature. Crystals formed after 5 days [10,16,23]. 10 mM stock concentrations of each inhibitor (provided by Dr. Claudiu T. Supuran) were prepared in a 100% dimethyl sulfoxide (DMSO) solution. Prior to being added to the crystal drop, stock solutions were diluted in a mixture 1:1 deionized water and precipitant solution to the desired compound concentrations of ~10 μ M. Crystals were then equilibrated with the precipitant-inhibitor solution 48 h prior to data collection. Diffraction data was collected either “in-house” using an RU-H3R rotating Cu anode ($\lambda = 1.5418 \text{ \AA}$) operating at 50 kV and 22 mA utilizing a 0.3 mm R-Axis VI⁺⁺ image plate detector (Rigaku, USA) or on the F1 beamline ($\lambda = 0.9177 \text{ \AA}$) at the Cornell High Energy Synchrotron Source (CHESS). Data sets were integrated, merged and scaled using the HKL2000 software package, all to a $P2_1$ space group [24]. Initial phases were calculated using molecular replacement using PDB entry 3KS3 [18] (hCA II) with waters and ligands removed, as a search model. Initial phase determination, model refinements, and generation of ligand restraint files were performed using PHENIX suite of programs [25]. Models for ligand-protein complexes were generated using Coot and LigPlus [26]. Coot was also used to determine bond lengths and angles used for analysis as well as PDB files for ligands [26,27]. Figures were generated using PyMol [26]. Diffraction data and final model statistics are summarized in Table 2.

5.3. Ligand binding studies

The buried and accessible surfaces areas of all compounds in their respective crystal structures were calculated using the online *Proteins, Interfaces, Surfaces and Assemblies* algorithm provided by the Protein Data Bank in Europe (*PDBePISA*) [20]. The ΔG values indicate the solvation free energy gain upon formation of the interface but not the effect of satisfied hydrogen bonds and salt bridges across the interface. ASA and BSA indicate the accessible surface and buried surface areas of the proteins upon inhibitor binding.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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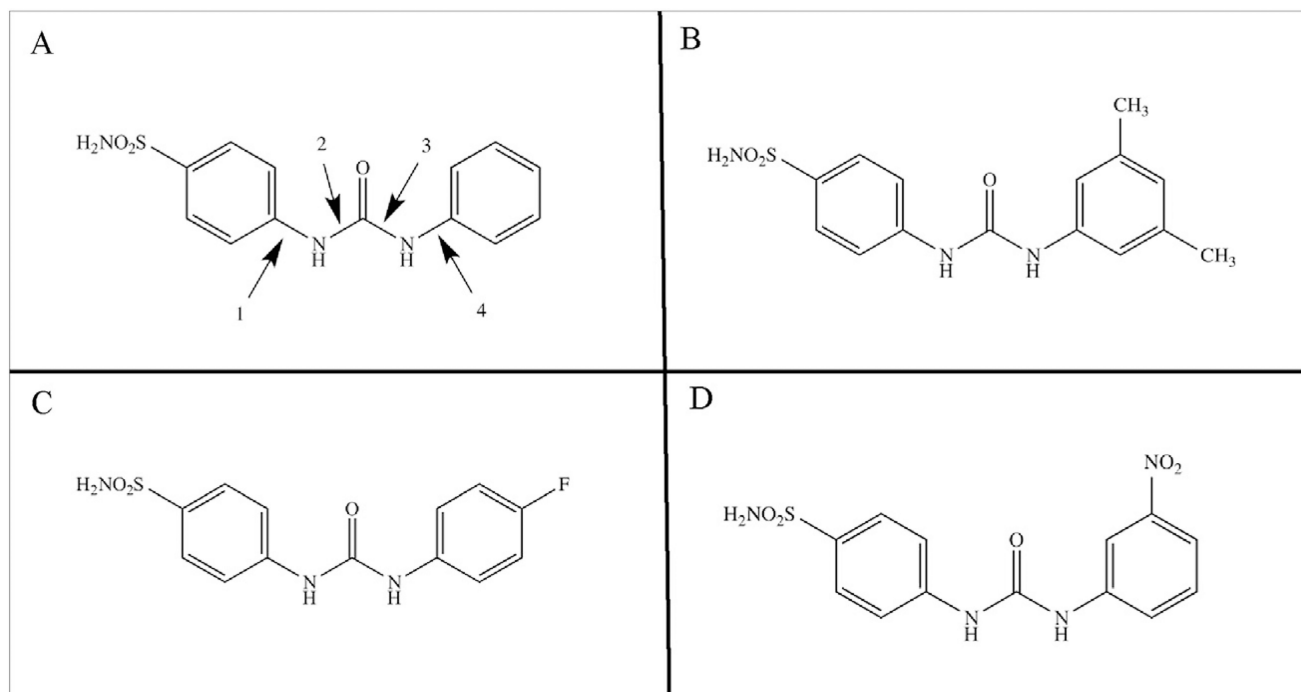


Fig. 1. Structures of ureido-substituted benzenesulfonamide inhibitors. The general construct shows the primary sulfamate ZBG is conjugated to benzene incorporating 4-substituted ureido moieties A) and an R moiety present in the ureido tail of the compound (B–D). Numbers 1, 2, 3 and 4 refer to torsion angles in Table 3.

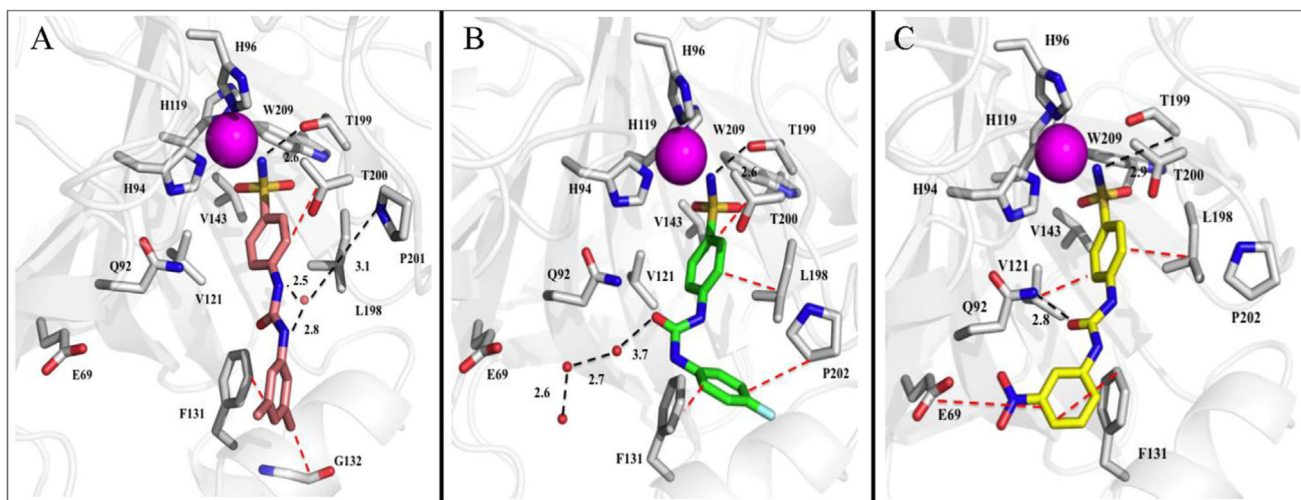


Fig. 2. Ureido-substituted benzenesulfonamides bound in the active site of hCA II (grey). A) Compound U-CH₃ (pink) B) Compound U-F (green) and C) Compound U-NO₂ (yellow) with active site residues and relative bond distances (Å) labeled. Specific interactions of compounds with the active site of hCA II are indicated as black dotted lines and hydrophobic interactions shown as red dotted lines. The zinc (magenta) and solvent (red) are represented as spheres. The figure was made using PyMol.26. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

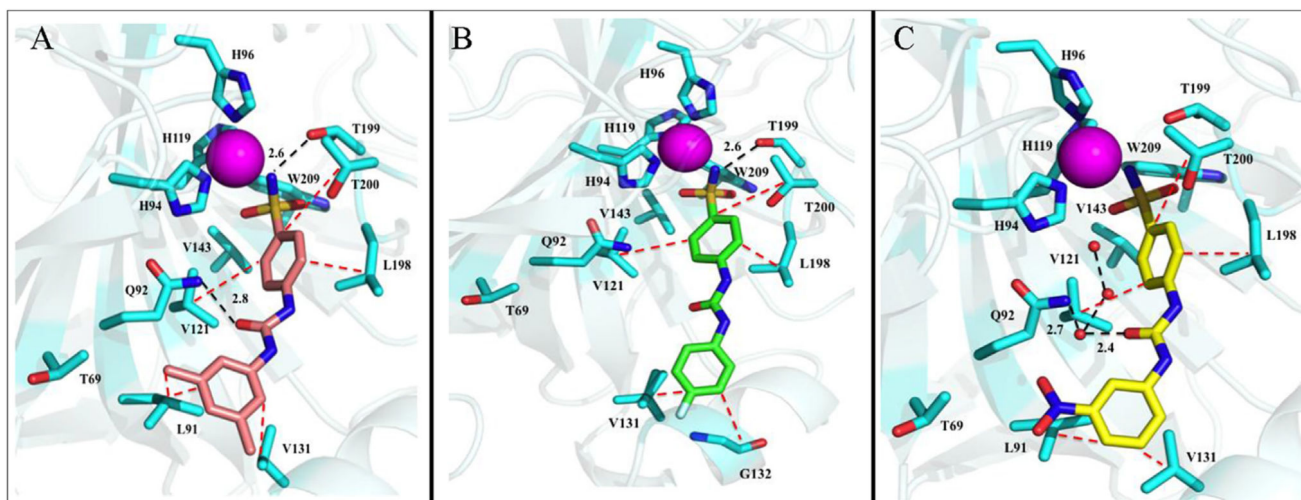


Fig. 3. Ureido-substituted benzenesulfonamides bound in the active site of hCA IX-mimic (cyan). A) Compound U-CH₃ (pink) B) Compound U-F (green) and C) Compound U-NO₂ (yellow) with active site residues and relative bond distances (Å) labeled. Specific interactions of compounds with the active site of hCA II are indicated as black dotted lines and hydrophobic interactions shown as red dotted lines. The zinc (magenta) and solvent (red) are represented as spheres. The figure was made using PyMol.26. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

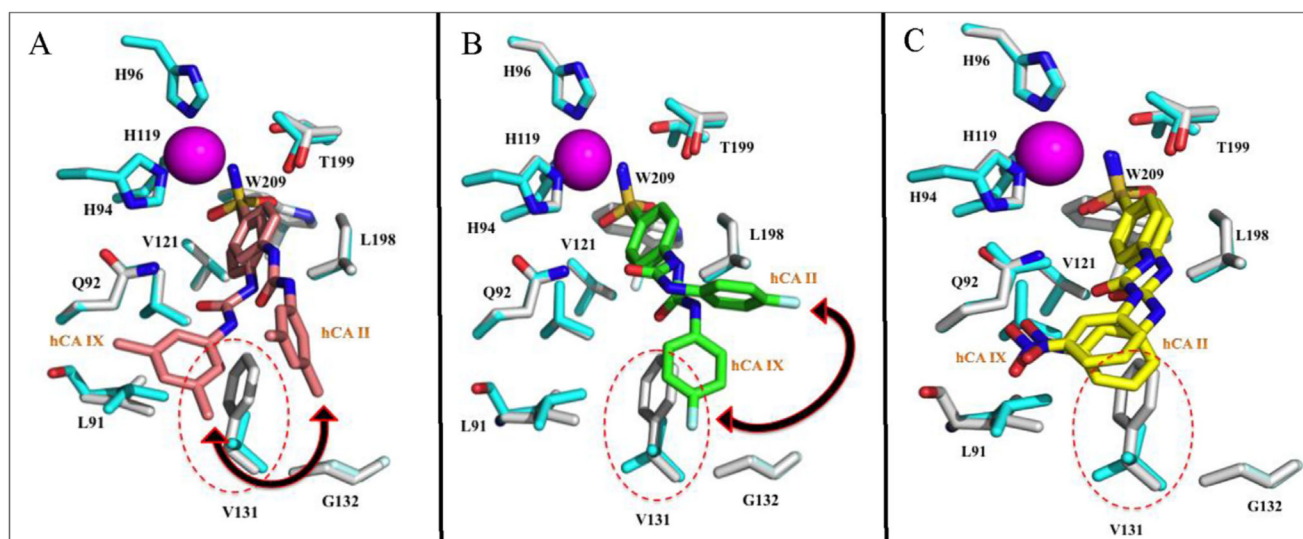


Fig. 4. Overlay of compounds A) U-CH₃ (pink), B) U-F (green) and C) U-NO₂ bound to the active site of hCA II (grey) onto inhibitor bound hCA IX-mimic (cyan) with specific residues labeled. Active site zinc shown as magenta sphere, red dotted circles indicate residues within the active site of hCA IX-mimic that differ from hCA II. Black double-headed arrow indicates the direction of conformational change of the compounds bound to hCA IX-mimic in comparison to hCA II. The figure was made using PyMol.26. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

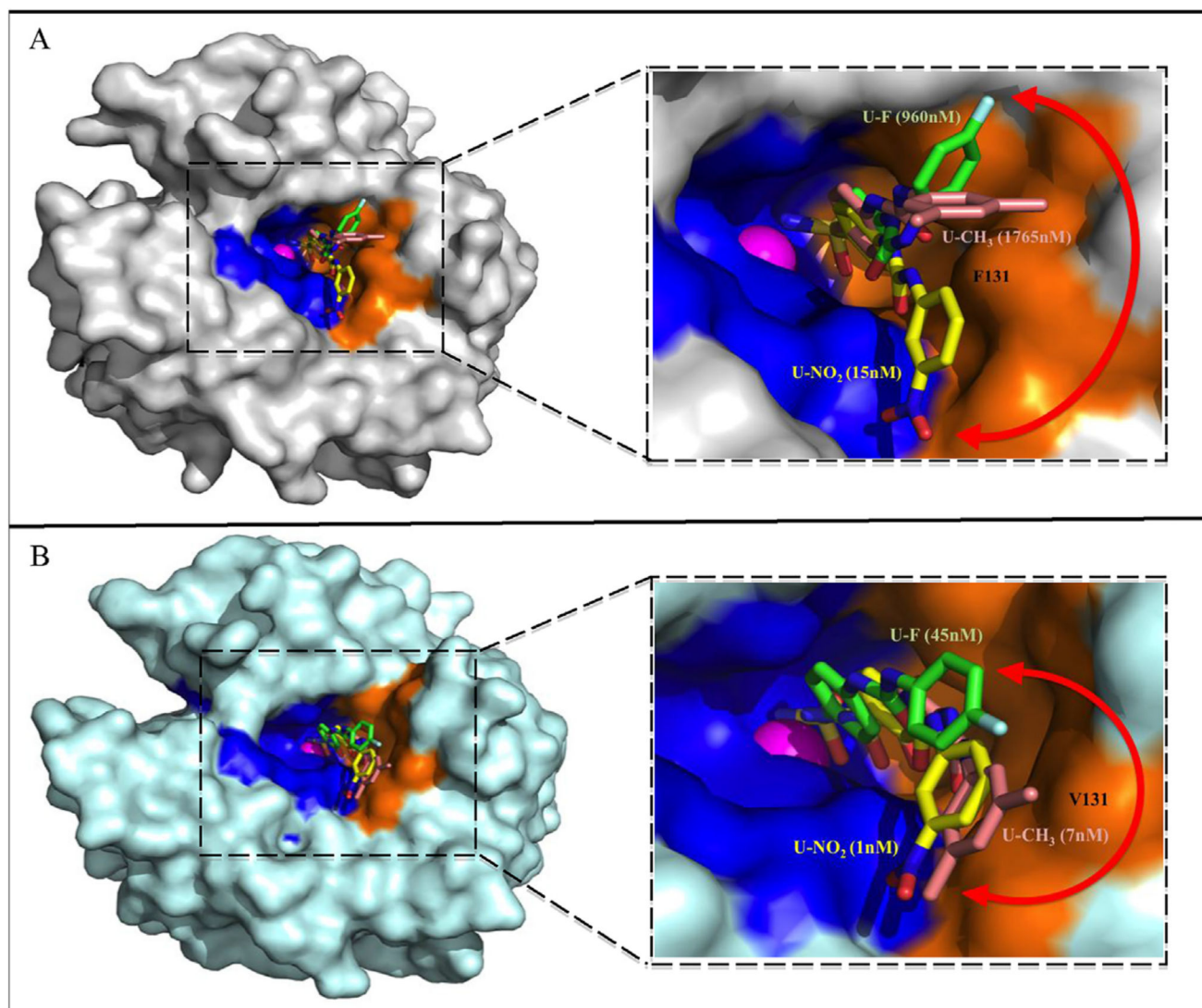


Fig. 5. Surface representation of ureido-substituted benzenesulfonamide compounds bound to the active site of A) hCA II (grey) and B) hCA IX-mimic (cyan). Compound U-CH₃ (pink), c U-F (green), and U-NO₂ (yellow). Zinc (magenta sphere), hydrophilic (blue) and hydrophobic (orange) residues are as shown. Red double-headed arrows indicate isoform specificity relative to residue 131 (labeled in black). The K_i values of each compound bound to hCA II and hCA IX are also labeled. The figure was made using PyMol.26. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

Table 1Inhibitions for hCA isoforms with ureido-substituted benzenesulfonamide compound **SLC-0111/U-F**.

Isoform	K_i (nM)^a
hCA I	5080
hCA II	960
hCA III	7918
hCA IV	286
hCA VA	2545
hCA VB	908
hCA VI	6490
hCA VII	8550
hCA IX	45
hCA XII	4
hCA XIII	8755
hCA XIV	257

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Inhibition and isozyme selectivity ratio for hCA II, IX, and XII with ureido-substituted benzenesulfonamide compounds **U-CH₃**, **U-F**, and **U-NO₂**.

Table 2

Compound	K_i (nM) ^a	Selectivity Ratio ^b						
		hCA II	hCA IX	hCA XII	II/IX	II/XII	IX/XII	XII/IX
U-CH₃	1765	7	6	6	250	300	1	1
U-F	960	45	4	4	20	240	10	<1
U-NO₂	15	1	6	6	15	2	<1	5

^a K_i values obtained from Pacchiano et al., [11]: **U-CH₃** (3,5-Me₂C₆H₃), **U-F** (4-FC₆H₄), **U-NO₂** (3-O₂NC₆H₄).

^b Selectivity as determined by the ratio of K_i s for hCA isozyme relative to hCA IX and hCA XII.

Table 3

X-ray crystallography statistics of USBs bound hCA IX-mimic and hCA II structures.

Sample PDB ID	hCA IX-mimic: U-CH ₃ 5JN1	hCA IX-mimic: U-5JN3	hCA IX-mimic: U-F 5JNF	hCA IX-mimic: U-NO ₂ 5JMZ	hCA II: U-CH ₃ 5JN7	hCA II: U-F 3N4D
Space Group	$P2_1, a = 42 \pm 0.8, b = 42 \pm 0.8, c = 72 \pm 0.5; \beta = 104 \pm 0.3$					
Resolution (Å)	29.1–1.51 (1.53–1.51)	19.9–1.60 (1.63–1.60)	19.9–1.90 (2.00–1.90)	29.1–1.52 (1.55–1.52)	20.0–1.55 (1.58–1.55)	
Total Reflections	65546	53075	61159	55980	63099	
R_{sym}^a (%)	5.8 (34.7)	6.0 (50.0)	5.8 (19.8)	6.9 (48.9)	7.8 (27.2)	
I/σ	31.0 (3.7)	29.5 (3.1)	19.4 (6.3)	28.6 (3.2)	25.5 (5.4)	
Completeness (%)	100.0 (100.0)	100.0 (100.0)	96.6 (96.7)	96.4 (97.3)	98.2 (98.7)	
R_{cyst}^b (%)	16.1 (20.4)	16.9 (22.8)	16.7 (18.2)	16.5 (24.5)	16.1 (16.9)	
R_{free}^c (%)	18.4 (20.0)	19.6 (20.2)	19.9 (23.8)	18.9 (33.0)	18.8 (20.0)	
# of Protein Atoms	2048	2123	2046	2053	2093	
# of Water Molecules	185	187	170	208	205	
# of Ligand Atoms	22	21	23	22	21	
Ramachandran stats (%)	Favored, allowed, outliers					
Avg. B factors (Å ²): Main-chain, Side-chain, Solvent, Ligand ^d	14.2, 19.0, 26.3, 13.4	15.3, 19.6, 23.5, 23.7	18.6, 23.3, 28.7, 37.9	14.8, 19.3, 25.6, 27.7	13.8, 17.9, 23.3, 23.1	

^a $R_{\text{sym}} = (\sum |I - \langle I \rangle| / \sum \langle I \rangle) \times 100$.^b $R_{\text{cyst}} = (\sum |F_o - F_c| / \sum |F_o|) \times 100$.^c R_{free} is calculated in the same way as R_{cyst} except it is for data omitted from refinement (5% of reflections for all data sets).^dValues in parenthesis correspond to the highest resolution shell.

Table 4

Torsion angles ($^{\circ}$) of the USB compounds **U-CH₃**, **U-F**, and **U-NO₂** bound to the active sites of hCA II and hCA IX-mimic, respectively. Definition for torsion angles given in Fig. 1.

Angle	hCA II: U-CH ₃	hCA IX-mimic: U-CH ₃	hCA II: U-F	hCA IX-mimic: U-F	hCA II: U-NO ₂	hCA IX-mimic: U-NO ₂
1	16	-38	-45	5	-34	-36
2	178	-177	179	176	-179	180
3	178	176	-1	177	-176	179
4	54	-33	151	-56	-19	-33

Table 5

Residues within the active site of hCA II, hCA IX and hCA XII that differ and make up the selective pocket (*hCA II numbering*).

Residues	hCA II	hCA IX	hCA XII
67	Asn	Gln	Lys
91	Ile	Leu	Thr
131	Phe	Val	Ala
135	Val	Leu	Ser

Information adapted from Pinard et al. [14].

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