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# SAR Studies of *N*-[2-(1*H*-Tetrazol-5-yl)phenyl]benzamide Derivatives as Potent G Protein-Coupled Receptor-35 Agonists

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**(5)** Supporting Information

**ABSTRACT:** G protein-coupled receptor-35 (GPR35) has emerged as a potential target in the treatment of pain and inflammatory and metabolic diseases. We have discovered a series of potent GPR35 agonists based on a coumarin scaffold and found that the introduction of a 1*H*-tetrazol-5-yl group significantly increased their potency. We designed and synthesized a new series of N-[2-(1*H*-tetrazol-5-yl)phenyl]benzamide derivatives through a two-step synthetic approach, and characterized their agonistic activities against GPR35 using a dynamic mass redistribution (DMR) assay. *N*-(5-bromo-2-(1*H*-tetrazol-5-yl)phenyl)-4-methoxybenzamide (**56**) and *N*-(5-bromo-2-(1*H*-tetrazol-5-yl)phenyl)-2-fluoro-4-methoxyben-



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zamide (63) displayed the highest agonistic potency agonist GPR35 with an EC<sub>50</sub> of 0.059  $\mu$ M and 0.041  $\mu$ M, respectively. The physicochemical properties of selected compounds were calculated to evaluate their druglikeness, suggesting that compounds 56 and 63 have good druglike properties. Together, *N*-[2-(1H-tetrazol-5-yl)phenyl]benzamide derivatives are potentially great candidates for developing potent GPR35 agonists.

KEYWORDS: GPR35, dynamic mass redistribution, N-[2-(1H-tetrazol-5-yl)phenyl]benzamide derivatives, Lipinski's rule

**G** protein-coupled receptors (GPCRs) as the most successful druggable receptor family have an important position in drug development.<sup>1</sup> However, the physiological functions of many GPCRs, in particular orphan receptors,<sup>2</sup> are far from clear. Discovering potent probe molecules for these orphan receptors has great significance in understanding their physiological functions and for drug development. The orphan G protein-coupled receptor 35 (GPR35) was first discovered in 1998, and it later has been implicated in a variety of diseases,<sup>3</sup> such as cardiovascular diseases,<sup>4</sup> gastric cancer,<sup>3,5</sup> artery disease,<sup>6</sup> and type 2 diabetes.<sup>7</sup>

In the past decade, several endogenous ligands have been discovered to activate GPR35; these ligands include kynurenic acid,<sup>5</sup> lysophosphatidic acid,<sup>8</sup> multiple tyrosine metabolites,<sup>9</sup> and the mucosal chemokine CXCL17<sup>10</sup> but with variable potency. Several classes of synthetic agonists including the phosphodiesterase 5 inhibitor zaprinast (1),<sup>11</sup> the antiasthma drugs doxantrazole (2), and pemirolast (3)<sup>12</sup> have also been reported to display GPR35 agonistic activity. Furthermore, zaprinast is the most widely used reference agonist to elucidate the biological functions of GPR35. Many more GPR35 agonists were discovered through high-throughput screening,<sup>13,14</sup> and

only a few structure–activity relationship (SAR) studies have been reported in the literature so far.  $^{15,16}$ 

Our previous studies have shown that tetrazole 4 possessed significant agonism at GPR35.<sup>17</sup> Several other studies have shown that halogen atom substitution also increased the agonistic activity in certain positions.<sup>14–16</sup> The introduction of lipophilic residues and hydrogen bond accepting groups at specific positions in the molecule could probably enhance the activity of compounds.<sup>15,16</sup> Inspired by these findings, a variety of N-[2-(1H-tetrazol-5-yl)phenyl]amido derivatives with different halogen atoms at the 4- or 5-positions were synthesized. The compounds were prepared starting from **6a–6c** and **8**.

The 2-(1*H*-tetrazol-5-yl)aniline derivatives 7a and 7b were obtained from 6a and 6b via the [3 + 2] cycloaddition of nitriles and sodium azide catalyzed by aluminum chloride at 90 °C with excellent yields (84% and 91%, respectively) (Scheme 1).

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Scheme 1. Synthesis of Compounds 7a and 7b<sup>a</sup>



"Reagents and conditions: (a) NaN<sub>3</sub>, AlCl<sub>3</sub>, THF, 90 °C, 5 h, yield 81–86%.

The amines 6a-6c and 8 were reacted with various acid chlorides in pyridine as the base and solvent at room temperature, yielding a series of *N*-(2-cyanophenyl)amido derivatives (19-41) (Scheme 2, Scheme 3). The *N*-[2-(1*H*-





"Reagents and conditions: (a)  $R_3COCl,$  pyridine, overnight, yield 68–87%. (b) NaN\_3, AlCl\_3, THF, 90  $^\circ C$ , 5 h, yield 69–91%





<sup>*a*</sup>Reagents and conditions: (a) 4-Methoxybenzoyl chloride, pyridine, overnight. (b) NaN<sub>3</sub>, AlCl<sub>3</sub>, THF, 90 °C, 5 h.

tetrazol-5-yl)phenyl]benzamide derivatives (42-64) were synthesized from N-(2-cyanophenyl)benzamide derivatives

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(19-41) via a [3 + 2] cycloaddition, also with good yields (69-91%) (Scheme 2, Scheme 3).

All *N*-[2-(1*H*-tetrazol-5-yl)phenyl]amido compounds obtained were tested using a dynamic mass redistribution (DMR) assay<sup>18</sup> using HT-29, a cell line endogenously expressing GPR35.<sup>19</sup> The reference agonist zaprinast was also used to perform a DMR desensitization assay.<sup>19</sup> Results showed that all *N*-[2-(1*H*-tetrazol-5-yl)phenyl]amido compounds obtained not only gave rise to a dose-dependent DMR in HT-29 but also desensitized the cells to subsequent stimulation with 1  $\mu$ M zaprinast. Notably, the potency to trigger DMR for all compounds was found to be almost equal to that to desensitize the zaprinast response (Table 1, Table 2), suggesting that these compounds are GPR35 agonists.

The DMR antagonist assay further showed that the known GPR35 antagonist ML-145 (**5**, Figure 1) dose-dependently and completely blocked the DMR arising from all *N*-[2-(1*H*-tetrazol-5-yl)phenyl]amido derivatives, each at its respective  $EC_{80}$  to  $EC_{100}$  concentration.<sup>20</sup> This suggests that the DMR assays of these *N*-[2-(1*H*-tetrazol-5-yl)phenyl]amido derivatives were specific to the activation of GPR35.

Inspired by our previous findings, a tetrazolyl group was first introduced into the simplest compounds **6a** and **6b**, yielding compounds **7a** and **7b**. Compared to the inactive compounds **6a** and **6b**, **7a** and **7b** both showed moderate potency with an  $EC_{50}$  of 29.99  $\mu$ M and 44.06  $\mu$ M, respectively. This result suggests that the introduction of the tetrazolyl to the molecule significantly improves the agonistic activity.

Next, an amide linker was introduced into compound 7b, given that lipophilic residues and hydrogen bond-accepting groups may also play an important role in activating GPR35.<sup>15</sup> The propanamide (14, EC<sub>50</sub> 0.62  $\mu$ M) showed more than 50-fold increase in potency compared to 7b. The introduction of the bulkier alkyl substituents further improved potency. For instance, isopropyl (15, EC<sub>50</sub> 0.38  $\mu$ M) and cyclohexyl (16, EC<sub>50</sub> 0.44  $\mu$ M) substituted compounds displayed more than 100-fold increased potency compared to 7b. Compounds with an aromatic group in this position also had relatively higher potency, such as furyl (17, EC<sub>50</sub> 1.06  $\mu$ M), thienyl (18, EC<sub>50</sub> 0.52  $\mu$ M) and phenyl (42, EC<sub>50</sub> 0.87  $\mu$ M).

Since the benzoic acid derivatives are easy to obtain, we chose the phenyl-substituted derivative 42 (N-(5-bromo-2-(1H-chose + 2 N + chose +

# Table 1. Potency of 2-Substituted 2-(1H-Tetrazol-5-yl)aniline Derivatives in DMR Assays

			Br HNN NH <sub>2</sub> N-N 7a	Br H NH <sub>2</sub> N-N 7b	$ \begin{array}{c}                                     $	
compd	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	$EC_{50}^{a}$ ( $\mu$ M)	desensitization $IC_{50}^{\ b}$ ( $\mu$ M)	antagonist $\mathrm{IC_{50}}^{c}(\mu\mathrm{M})$
7a				$29.99 \pm 2.80$	$35.56 \pm 2.54$	$0.67 \pm 0.19$
7b				$44.06 \pm 4.06$	$75.34 \pm 9.31$	$0.67 \pm 0.25$
14	Br	Н	ethyl	$0.62 \pm 0.06$	$0.70 \pm 0.07$	$0.47 \pm 0.11$
15	Br	Н	isopropyl	$0.38 \pm 0.02$	$0.28 \pm 0.02$	$0.58 \pm 0.06$
16	Br	Н	cyclohexyl	$0.44 \pm 0.03$	$0.20 \pm 0.01$	$0.43 \pm 0.08$
17	Br	Н	furyl	$1.06 \pm 0.13$	$0.44 \pm 0.03$	$0.29 \pm 0.08$
18	Br	Н	thienyl	$0.51 \pm 0.02$	$0.26 \pm 0.03$	$0.50 \pm 0.05$

 ${}^{a}\text{EC}_{50}$  to trigger DMR.  ${}^{b}\text{IC}_{50}$  to desensitize upon cells repeated stimulation with 1  $\mu$ M zaprinast.  ${}^{c}\text{IC}_{50}$  of known GPR35 antagonist 5 to block the agonism. The data respresent mean  $\pm$  sd from two independent measurements, each with four replicates (n = 8).

Table 2. Potency of N-(2-(1H-Tetrazol-5-yl)phenyl)benzamide Derivatives in the DMR Assay



<sup>*a*</sup>EC<sub>50</sub> to trigger DMR. <sup>*b*</sup>IC<sub>50</sub> to desensitize cells upon cells repeated stimulation with 1  $\mu$ M zaprinast. <sup>*c*</sup>IC<sub>50</sub> of known GPR35 antagonist 5 to block the agonist-induced DMR. The data represents mean  $\pm$  sd from two independent measurements, each with four replicates (n = 8).



Figure 1. Selected GPR35 agonists with potencies at GPR35 (1-4) and antagonist (5).

tetrazol-5-yl)phenyl)benzamide) for further modification. *N*-(5-Bromo-2-(1*H*-tetrazol-5-yl)phenyl)benzamide derivatives with different substituents in the *o-*, *m-*, and *p*-position were investigated (mono- or disubstitutions).

For the monosubstitutions, the same substituent in different positions of the benzene ring could significantly change the activity. The following rank order of potency among compounds with the same substituent in different positions (o-, m-, and p-position) was observed: m-methyl (44,  $EC_{50}$  0.86  $\mu$ M)  $\approx$  *p*-methyl (45, EC<sub>50</sub> 1.33  $\mu$ M) > *o*-methyl (43, EC<sub>50</sub> 2.05  $\mu$ M) (Figure S1a); p-chloro (48, EC<sub>50</sub> 0.52  $\mu$ M) > ochloro (46, EC<sub>50</sub> 0.87  $\mu$ M) > *m*-chloro (47, EC<sub>50</sub> 2.18  $\mu$ M) (Figure S1b); p-fluoro (52, EC<sub>50</sub> 0.45  $\mu$ M)  $\approx$  o-fluoro (50,  $EC_{50} 0.35 \ \mu M) > m$ -fluoro (51,  $EC_{50} 0.61 \ \mu M)$  (Figure S1c); *p*methoxyl (56, EC<sub>50</sub> 0.059  $\mu$ M) > o-methoxyl (54, EC<sub>50</sub> 0.86  $\mu$ M) > *m*-methoxyl (55, EC<sub>50</sub> 1.94  $\mu$ M) (Figure S 1d); *p*trifluoromethyl (54, EC<sub>50</sub> 3.65  $\mu$ M) > *m*-trifluoromethyl (55,  $EC_{50}$  4.60  $\mu$ M). These results suggest that substituents in the para-position led to a considerable increase in potency except for the methyl-derivatives with better tolerance in the metaposition. For N-(5-bromo-2-(1H-tetrazol-5-yl)phenyl)benzamide derivatives with varying substituents in the paraposition, the rank order of potency was as follows: methoxyl  $(56, EC_{50} 0.059 \ \mu M) >$ fluoro  $(52, EC_{50} 0.45 \ \mu M) >$ chloro (48, EC<sub>50</sub> 0.52  $\mu$ M) > methyl (45, EC<sub>50</sub> 1.33  $\mu$ M) > trifluoromethyl (54, EC<sub>50</sub> 3.65  $\mu$ M), suggesting that methoxyl in the para-position was favorable for the agonistic activity.

For the disubstitution of the benzamide ring, the rank order of potency of *o*,*p*-dichloro (49), *o*,*p*-difluoro (53), and *m*,*p*dimethoxy (57) was as follows: *o*,*p*-difluoro (53, EC<sub>50</sub> 0.13  $\mu$ M) > *o*,*p*-dichloro (49, EC<sub>50</sub> 0.31  $\mu$ M) > *m*,*p*-dimethoxy (57, EC<sub>50</sub>

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0.83  $\mu$ M), suggesting that the *o,p*-positions led to a visible increase in activity compared to the *o*- or *p*-monosubstituted compounds. However, the dimethoxy in *m,p*-positions (57) led to a large reduction in activity compared to compound 56, suggesting that bulky *m,p*-substituted compounds are not well tolerated but *o,p*-disubstitutions are.

Compounds 50 and 56 showed the highest agonistic potency among the derivatives. Therefore, we devised the synthesis of compound 63, which was substituted with a fluoro in the *o*position and a methoxy in the *p*-position. As expected, compound 63 displayed further improved potency (EC<sub>50</sub> 0.041  $\mu$ M, Figure 2).



**Figure 2.** (a) Real time kinetic responses of **63** at different doses in HT-29 cells. (b) DMR amplitudes of compound **63** as a function of doses compared with the dose-dependent desensitization of zaprinast DMR by **63**, and the dose-dependent inhibition of the DMR of 100 nM compound **63** by compound **5**. The data represents mean  $\pm$  sd from two independent measurement, each with four replicates (n = 8).

There is evidence that the halogen atom bromine is very important for the retention of compound activity agonist GPR35.<sup>14–17,21</sup> In order to study the effect of the change of the bromine atom on the activity of the compound, we chose compound **56** for further study. Results showed that transforming the substitution position of the bromine atom (**58**, EC<sub>50</sub> 0.30  $\mu$ M), replacing the bromine with a fluorine (**59**, EC<sub>50</sub> 0.20  $\mu$ M), or removing the bromine (**60**, EC<sub>50</sub> 0.65  $\mu$ M) all decreased potency.

Compound 37, a synthetic precursor of **60** was also tested and found to be devoid of activity, demonstrating the necessity of the tetrazole. The activity was completely lost when the tetrazolyl was replaced by cyano (Figure S1e). In addition, changing the position of the tetrazoyl also significantly reduced the activity of the compound (**64**, EC<sub>50</sub> 17.58  $\mu$ M).

We further examined the activity of compounds **53** and **56** through the ERK phosphorylation. Compound **53** and compound **56** increased ERK phosphorylation (Figure. 3a). As control, the known GPR35 agonist zaprinast also triggered ERK phosphorylation (Figure 3b). Moreover, the GPR35 antagonist ML-145 attenuated ERK phosphorylation induced by these compounds (Figure. 3b). These results suggested that compounds **53** and **56** induced the phosphorylation of ERK via the activation of GPR35.

In order to evaluate the druglikeness of a compound, several physiochemical properties such as the partition coefficient (clogP), the ligand efficiency (LE) and the ligand-lipophilicity efficiency (LLE) were introduced.<sup>22–24</sup> These parameters were calculated for the compounds with relatively high potency, including compounds 14–16, 18, 42, 44, 46, 48–54, 56–60, and 63 (Table 3). Lipinski's Rule of Five considered that the clogP value of a druglike compound should be lower than 5.<sup>25,26</sup> The clogP values of all these selected compounds were found to be within this range. In fact, most of these compounds

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**Figure 3.** (a) Western blot of ERK1/2 and p-ERK1/2 after treatment with compounds **53** and **56** for 15 min at concentrations of 100 nM, 1  $\mu$ M respectively. (b) Western blot of ERK1/2 and p-ERK1/2 after treatment with the known GPR35 agonist Zaprinast (Zap) at 1  $\mu$ M and Western blot of ERK1/2 and p-ERK1/2 after treatment with compounds for 10 min at the same concentrations as in (a) in the presence of ML-145 (25  $\mu$ M) for 5 min.

Table 3. Physicochemical Properties of Selected Compounds

compd	pEC50	clogPa	LE	LLE
14	6.21	1.27	0.37	4.94
15	6.42	1.58	0.36	4.84
16	6.36	2.78	0.30	3.58
18	6.29	1.98	0.31	4.31
42	6.06	2.23	0.29	3.83
44	6.06	2.73	0.28	3.33
46	6.06	2.19	0.28	3.87
48	6.28	3.02	0.29	3.26
49	6.50	2.93	0.28	3.57
50	6.46	2.04	0.29	4.42
51	6.21	2.45	0.28	3.76
52	6.35	2.45	0.29	3.90
53	6.88	2.21	0.30	4.67
54	6.07	2.29	0.26	3.78
56	7.23	2.36	0.31	4.87
57	6.08	2.01	0.24	4.07
58	6.52	2.36	0.28	4.16
59	6.70	1.64	0.29	5.06
60	6.19	1.31	0.28	4.88
63	7.38	2.15	0.32	5.23
Calculated b	v the Chembi	odraw Ultra 11	.0.	

exhibited a clogP value between 2 and 3, which is favorable for orally administered drugs.

The LE combines physiochemical with pharmacological properties, and it represents the binding force between each atom and receptors and calculates it as follows:  $LE = pEC_{50}/N$  (N = non-hydrogen atoms). The LE values of most selected compounds were found to be about 0.3, except for compounds 54 and 57.

LLE (LEE =  $pEC_{50}$  – clogP), another useful parameter, combines the potency and lipophilicity. Among these compounds, 14, 15, 56, 59, 60, and 63 showed a LLE value of 5. A suitable drug candidate should have a LE > 0.3 and LLE > 5. In summary, the new compounds 56 and 63 with potent potency showed a very good clogP and suitable LE and LLE values.

In conclusion, a series of N-[2-(1H-tetrazol-5-yl)phenyl]amido derivatives were synthesized as potent GPR35 agonists through a two-step synthesis method. SAR analysis showed that a bromine substituent in the 5-position and a *p*-methoxybenzamide in the 2-position would significantly improve the

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activity (56, EC<sub>50</sub> 0.059  $\mu$ M). The *o*,*p*-distribution such as *o*,*p*-dichloro (49) and *o*,*p*-difluoro (53) could also improve the potency when compared with the *o*- or *p*-monosubstituted analogs. Combining these findings, compound 63 was synthesized and found to display the highest potency (EC<sub>50</sub> 0.041  $\mu$ M) and good physicochemical properties. Together, this study provides a new series of potent GPR35 agonists, such as 56 and 63, which may become useful leading ligands to further elucidate the physiological roles of GPR35.

# ASSOCIATED CONTENT

# **S** Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsmedchem-lett.7b00510.

Experimental details and characterization data for the reported compounds; NMR spectra; and biological assays (PDF)

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#### Notes

The authors declare no competing financial interest.

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# ABBREVIATIONS

GPR35, G protein-coupled receptor 35; GPCR, G proteincoupled receptor; SARs, structure activity relationships; DMR, dynamic mass redistribution; THF, tetrahydrofuran; LE, ligand efficiency; LLE, ligand-lipophilicity efficiency

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