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Resin cements formulated with thio-urethanes can strengthen porcelain and increase bond strength to ceramics

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Abstract

Objectives—The use of thio-urethane oligomers has been shown to significantly improve the mechanical properties of resin cements (RCs). The aim of this study was to use thio-urethane-modified RC to potentially reinforce the porcelain-RC structure and to improve the bond strength to zirconia and lithium disilicate.

Methods—Six oligomers were synthesized by combining thiols - pentaerythritol tetra-3-mercaptopropionate (PETMP, P) or trimethylol-tris-3-mercaptopropionate (TMP, T) –with di-functional isocyanates - 1,6-Hexanediol-diisocyanate (HDDI) (aliphatic, AL) or 1,3-bis(1-isocyanato-1-methylethyl)benzene (BDI) (aromatic, AR) or Dicyclohexylmethane 4,4'-Diisocyanate (HMDI) (cyclic, CC). Thio-urethanes (20 wt%) were added to a BisGMA/UDMA/TEGDMA organic matrix. Filler was introduced at 60 wt%. The microshear bond strength (μ SBS), Weibull modulus (m), and failure pattern of RCs bonded to zirconia (ZR) and lithium disilicate (LD) ceramics was evaluated. Biaxial flexural test and fractographic analysis of porcelain discs bonded to RCs were also performed. The biaxial flexural strength (σ_{bf}) and m were calculated in the tensile surfaces of porcelain and RC structures ($Z=0$ and $Z=-t/2$, respectively).

Results—The μ SBS was improved with RCs formulated with oligomers P_AL or T_AL bonded to LD and P_AL, P_AR or T_CC bonded to zirconia in comparison to controls. Mixed failures predominated in all groups. σ_{bf} had superior values at $Z=0$ with RCs formulated with oligomers P_AL, P_AR, T_AL, or T_CC in comparison to control; σ_{bf} increased with all RCs composed by thio-urethanes at $Z=-t/2$. Fractographic analysis revealed all fracture origins at $Z=0$.

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Conflict of interest

The authors declare no conflict of interest.

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Conclusion—The use of specific thio-urethane oligomers as components of RCs increased both the biaxial flexural strength of the porcelain-RC structure and the μ SBS to LD and ZR.

Clinical Significance—The current investigation suggests that it is possible to reinforce the porcelain-RC pair and obtain higher bond strength to LD and ZR with RCs formulated with selected types of thio-urethane oligomers.

Keywords

ceramic interface strengthening; mechanical reliability; bond to ceramics; resin cements; thio-urethane oligomers

1. Introduction

Resin cements (RCs) have gained clinical popularity in Dentistry because of their ability to bond both to the tooth structure and to the majority of restorative materials when associated with chemical components such as silanes and phosphate-based primers [1]. The use of RCs is indicated for several clinical applications that include adhesion of complete crowns, fixed partial prosthesis, partial coverage, and intracanal posts. The main advantages of RCs over other traditionally used cements, such as zinc-phosphates and glass-ionomers, are the reduced risk of microleakage, and the decreased sorption, solubility, and staining, which results in the maintenance of the interface integrity [2].

Glass-ceramics and polycrystalline ceramics are available for use in clinical practice. Loss of retention might represent a technical complication, compromising the lifetime of the restoration [3]. Glass-ceramics have been widely used for years due to the possibility of producing total or partial coverage, maintaining the maximum integrity of the remaining coronal structure. Considering that glass-ceramic restorations have been extensively used in partial coverage with preparations that do not have sufficient mechanical retention, high bond strength values are important for restoration successful clinical outcome. However, the current bond protocols are not very efficient for polycrystalline ceramics, since they are acid resistant and therefore cannot be treated by conventional etching [4]. Moreover, silane treatments are also not adequate for materials such as highly crystalline zirconia, which do not form covalent bonds with the alkoxy silane group of the coupling agent [4]. This problem has led to extensive research in the recent years focusing on the improvement of bonding strategies for this class of ceramic material, which has been used mostly for full coverage. More recently, partial polycrystalline structures have also been recommended [5], increasing the need for resin cements that provide higher bond strength and retention stability.

Besides providing bonding stability for ceramic restorations, RCs have been recognized to significantly strengthen bonded restoration [6,7]. Studies have demonstrated that the increase in the elastic modulus of RCs has a favorable impact on the mechanical strength of a dental feldspathic porcelain [6,7]. Moreover, cement characteristics such as the reduced volumetric shrinkage and, as a consequence, low polymerization stress might have a positive impact on the ceramic strength [8]. These cement characteristics are of great interest, especially for restorations made of dental porcelains or glass-ceramics. These materials

provide excellent esthetic characteristics and have been used for years in different applications. However, they are intrinsically fragile structures, i.e., have very low fracture toughness in comparison to other dental ceramics.

Recent publications showed that RCs formulated with thio-urethane oligomers as part of the organic matrix presented increased mechanical properties when compared to materials composed only by traditional monomers. Mechanical properties such as flexural strength, flexural modulus, and fracture toughness are the ones mostly improved by the addition of thio-urethanes [9,10]. These improvements are achieved because of a combination of factors, including the low T_g of the additive [10], which provides opportunities for stress relaxation within the glassy methacrylate network [11], but also possibly due to active strand behavior on the thio-urethane oligomer [12]. In addition to that, the fact that the additive is pre-polymerized leads to a reduction in volumetric shrinkage and polymerization stress, observed in conjunction with an increase in the degree of conversion and delayed gelation/vitrification, which are in turn a function of the chain-transfer-capable thiol pending functionalities [9].

The objective of the present study was to determine the bond strength of resin cements to ceramic materials as function of the cement formulation (six oligomers) and the ceramic substrate (zirconia or lithium disilicate). In addition, the Weibull parameter of bilayered discs (0.15 mm of cement and 0.8 mm of ceramic) was determined for porcelain/RC specimens using the six experimental cement compositions. The discs were tested with the cement layer positioned on the tensile side of the biaxial flexural strength test. The hypothesis was that the properties evaluated would be significantly improved by the formulation of RCs containing thio-urethanes.

2. Materials and Methods

2.1 Resin cements composition

The experimental resin cements formulated for the study were composed of Bis-phenol A diglycidyl dimethacrylate (Bis-GMA; Esstech, Essington, PA, USA), urethane dimethacrylate (UDMA; Esstech) and tri-ethylene glycol dimethacrylate (TEGDMA; Esstech) in a 50:30:20 mass ratio. Photoinitiators were added to the monomers as follows: 0.6 wt% of a tertiary amine (EDMAB - ethyl 4-dimethylaminobenzoate; Avocado, Heysham, England), 0.2 wt% of di-camphoroquinone (Polysciences Inc., Warrington, PA, USA), and 0.5 wt% inhibitor (BHT-2,6-di-tert-butyl-4-methylphenol; SigmaAldrich, St. Louis, MO, USA).

Six different oligomers were synthesized in solution in the presence of catalytic amounts of trimethylamine. Multi-functional thiols were combined with di-functional isocyanates in 4 x volume of dichloromethane (very diluted solution) as presented in Table 1. In addition, the isocyanate:thiol ratio was kept at 1:2 (by mol) to avoid gelation of the oligomer during reaction, according to the Flory-Stockmeyer theory [13], leaving pendant thiols. Oligomers were purified by precipitation in hexanes and rotaevaporation, and then characterized by ¹H-NMR and mid-IR spectroscopy [14]. The disappearance of the isocyanate peak at 2270 cm⁻¹ and the appearance of resonance signals at 3.70 ppm were used as evidence for completion

of isocyanate reaction and thio-urethane bond formation, respectively [15]. The thiol group (SH) concentration for each oligomer was determined using a titration method with Ellman's reagent well established in the literature [16]. Thio-urethane oligomers were added to the methacrylate organic phase in proportion of 20wt%, as defined in our previous investigation [9]. The final oligomer product presented as a viscous liquid at room temperature, and was completely miscible with methacrylate monomers. A resin cement formulated without oligomer served as control.

Filler was introduced to all groups at 60 wt% (Barium glass 0.7 μm , density 3.0 g/ml, refractive index 1.553 - V117 4107, Esstech), with the aid of a mechanical mixer (DAC 150 Speed mixer, Flacktek, Landrum, SC, USA) for 5 min at 2400 rpm. All procedures were carried out under yellow lights.

2.2 Microshear bond strength (μSBS) to ceramics

Forty-two specimens (14 mm long, 12 mm wide and 3 mm thick) of yttrium-stabilized tetragonal polycrystalline zirconia (Y-TZP) (inCoris TZI mono L, Sirona Dental Inc, USA) or lithium disilicate ceramic (LD) (IPS e.max CAD, HT, shade A1, Ivoclar Vivadent, Liechtenstein) were produced ($n=6$ per group). Materials were processed as per the manufacturer's instructions. After sintering, the Y-TZP bonding surface was sandblasted with 30 μm silica-coated aluminum oxide particles (CoJetTM Sand S30, 3M ESPE, USA) for 15 s (at a distance of 5 mm and 2.5 bar of pressure) and ultrasonically cleaned for 10 min. Samples were silanated (Silano, Angelus, Brazil) and coated with a layer of 10-MDP (Metal/Zirconia Primer, Ivoclar Vivadent). LD specimens were etched with 10% hydrofluoric acid (Condac porcelain, FGM Dental Products, Brazil) for 20 s, cleaned with water and oil-free compressed air for 30 s and silanated (Silano, Angelus).

RC application on the ceramic treated surfaces was performed by filling plastic cylinders (0.8 mm wide and 1 mm long, Tygon[®] Tubing, Bio-Rad Laboratories, USA) with RC – 04 RC cylinders for each ceramic specimen, totalizing 24 bonded cylinders in each group. The RC was light activated for 60 s at 800 mW/cm² (Bluephase, Ivoclar Vivadent). The plastic cylinders were removed after RC polymerization. Samples were water-stored for one week at a temperature of 37[°]C. Specimens were tested in a universal testing machine at a crosshead speed of 0.5 mm/min using a metal wire. Data were obtained in MPa by dividing the force in Newton (N) by the bonding surface area (A) ($A = \pi * r^2$, where r is the RC cylinder radius = 0.4 mm). The mean fracture strength (MPa) in each specimen was calculated.

The fractured surfaces were evaluated by stereomicroscope (Olympus, Japan) and the failure patterns classified as adhesive, cohesive in cement, cohesive in ceramic, or mixed.

2.3 Porcelain-RC biaxial flexural strength

Two hundred and ten disc-shaped porcelain specimens were produced. The porcelain powder (Super Porcelain EX-3, Noritake, Japan) was mixed with the modeling liquid to produce a thick slurry, which was condensed into a metallic mold (12-mm diameter, 1-mm thickness). The mold was overfilled and placed on a vibrating table for 90 seconds, with excess liquid removed with absorbent paper. The surface was leveled with a razor blade to

produce discs of uniform thickness. Each disc was removed carefully from the mold, placed on a refractory substrate, and fired in a ceramic furnace (Vacumat 40, Vita Zahnfabrik) according to the manufacturer's instruction. The discs were cooled to room temperature and visually inspected. Discs with defects or visible cracks observed under a stereo microscope were discarded and replaced. The discs were manually wet ground on both sides with 320-grit SiC abrasive papers (Norton S.A., Brazil) to produce flat surfaces 0.8 ± 0.1 mm in thickness. Wet polishing was performed with 600- and 1200-grit SiC abrasive papers for 60 seconds on each surface. Final dimensions were checked with a digital caliper (Absolute Digimatic, Mitutoyo, Japan), and the porcelain discs were randomly divided into seven groups ($n=30$ per group) according to the RC tested.

The bonding surface of each porcelain disc was etched with 10% hydrofluoric acid (Condac porcelain, FGM) for 60 s, washed for 30 s with water and oil-free compressed air for 30 s and silanated (Silano, Angelus). The resin cement coating was applied with the aid of a digital caliper were a cement layer of $150 \mu\text{m}$ was standardized [17]. The RC was light activated with 800 mW/cm^2 (Bluephase, Ivoclar Vivadent) by 60 s through the porcelain disc. Samples were stored for 24 h in dark containers until final degree of conversion was obtained. The biaxial flexural strength (σ_{bf}) of the porcelain-RC specimens was determined on the mechanical testing machine using a ball-on-ring setup. The discs were centrally placed on a 10-mm diameter knife-edged support and loaded with a spherical indenter (4-mm diameter) at a crosshead speed of 1 mm/min.

The σ_{bf} was calculated according to the analytical solutions described and tested before [7, 18–23]. First, the E of the porcelain (E_1^*) and RC (E_2^*) were calculated as a function of the Poisson's ratio of the porcelain and RC, according to Equation 1:

$$E_1^* = \frac{E_1}{1 - \nu_1^2} \quad E_2^* = \frac{E_2}{1 - \nu_2^2}, \quad (\text{Eq. 1})$$

where E_1 is the E of porcelain [24], and E_2 the E of the RC defined previously according to ISO 4049 [10]. ν_1 and ν_2 the Poisson's ratios of the porcelain (0.25) [24] and resin-based luting agents (0.27) [25]. The neutral plane (m) of the coated porcelain specimens was calculated as a function of the porcelain and RC thicknesses (t_1 and t_2) and the calculated E_1^* and E_2^* , using Equation 2:

$$m = \frac{E_1^*(t_1)^2 - E_2^*(t_2)^2}{2(E_1^*t_1 + E_2^*t_2)}. \quad (\text{Eq. 2})$$

The σ_{bf} was calculated at z-axial positions at the center of the discs, where the porcelain surface at the bonded interface is located (position $z=0$) and the RC surface that contacts the ring of the ball-on-ring setup is located (position $z=-t_2$), according to Equations 3, 4, and 5:

$$\sigma_{bf} = \frac{-3P(1+\nu)(z-tn)}{2\pi(t_1+t_2)^3} \left[1 + 2\ln\left(\frac{a}{b}\right) + \frac{1-\nu}{1+\nu} \left(1 - \frac{b^2}{2a^2} \right) \frac{a^2}{R^2} \right] \quad (\text{Eq. 3})$$

$$\times \left[\frac{E_1^*(E_1^*t_1 + E_2^*t_2)(t_1 + t_2)^3}{(E_1^*t_1^2)^2 + (E_2^*t_2^2)^2 + 2E_1^*E_2^*t_1t_2(2t_1^2 + 2t_2^2 + 3t_1t_2)} \right] \quad (z = 0),$$

$$\sigma_{bf} = \frac{-3P(1+\nu)(z-tn)}{2\pi(t_1+t_2)^3} \left[1 + 2\ln\left(\frac{a}{b}\right) + \frac{1-\nu}{1+\nu} \left(1 - \frac{b^2}{2a^2} \right) \frac{a^2}{R^2} \right] \quad (\text{Eq. 4})$$

$$\times \left[\frac{E_2^*(E_1^*t_1 + E_2^*t_2)(t_1 + t_2)^3}{(E_1^*t_1^2)^2 + (E_2^*t_2^2)^2 + 2E_1^*E_2^*t_1t_2(2t_1^2 + 2t_2^2 + 3t_1t_2)} \right] \quad (z = -t_2),$$

$$\nu = \frac{\nu_1 t_1 + \nu_2 t_2}{t_1 + t_2}. \quad (\text{Eq. 5})$$

For the fractographic analysis, the fractured specimens retrieved from the biaxial flexural test were coated with gold and observed at varied magnifications under optical microscopy and SEM (JSM-6610; JEOL Ltd., Tokyo, Japan) to identify fracture features and to determine the critical flaw origin. An evaluation of the number of fracture fragments for each sample was also performed.

2.4 Statistical analysis

Data from microshear bond strength tests were submitted to one-way ANOVA and Tukey's test ($\alpha=0.05$). Data passed the normality (LD: $p=0,837$; ZR: $p=0,069$) and homoscedasticity tests (LD: $p=0,743$; ZR: $p=0,903$). Confidence intervals (95% CI) were calculated for σ_{bf} of the ceramic discs at axial positions $z=0$ and $z=-t_2$. Groups were considered significantly different when the 95% confidence interval bounds did not overlap. A Weibull analysis was performed for μSBS and σ_{bf} -data using the software Weibull++ (Reliasoft, Tucson, AZ, USA); m was calculated using maximum likelihood estimation and a two-parameter Weibull model, and 95% upper and lower confidence bounds were calculated using the likelihood ratio [26].

3. Results

3.1 Microshear bond strength (μSBS) to ceramics

When bonded to LD (Figure 1A), RCs formulated with oligomers P_AL or T_AL showed significant higher μSBS values in comparison to that obtained for the control group. Among the groups containing oligomer-based RCs, only P_CC and P_AL showed significantly different μSBS , whereas the other oligomers results in similar bond strength values.

Considering the zirconia substrate, it is possible to note in Figure 1B that only RCs composed by the oligomers P_AL, P_AR or T_CC obtained higher bond strength in comparison to the control. No statistical difference was observed among the bond strengths obtained for the different types of thio-urethanes.

The m showed no differences in μ SBS reliability for RCs composed by oligomers in comparison to control, either for LD or ZR. Within LD, the RCs formulated with P_CC and T_AR were superior to P_AR.

The failure patterns after μ SBS testing in LD surfaces (Figure 2A) were similar for the different experimental groups, with predominance of mixed failures for all groups, ranging between 75 and 85% of all failures. The remaining failures were classified as adhesive. For zirconia substrate (Figure 2B), there was predominance of mixed failures in all groups. However, in the samples bonded to the RC composed by the oligomer P_AL, which showed the highest μ SBS, cohesive failures in RC predominated over the adhesive ones.

3.2 Porcelain-RC biaxial flexural strength

The results of σ_{bf} and m in the porcelain-RC specimens ($z=0$ and $z=-t_2$) are shown in Table 3. Porcelain coated by RCs composed by the oligomers P_AL, P_AR, T_AL, and T_CC showed significantly higher σ_{bf} at $z=0$ as compared to control group, with the RC formulated with oligomer P_AL yielding the highest porcelain-RC strength. The mechanical reliability (m) of the structure at $z=0$ was not affected by the type of RC used. At axial position $z=-t_2$, significant differences were observed for σ_{bf} with RC composed by oligomers leading to superior values in comparison to the control. The structural reliability at $z=-t_2$ was independent of the RC tested, with no significant differences in m values among groups.

Samples of porcelain-RC of control group presented a greater number of fractures in three or four-or-more fragments (13 samples) than those containing RCs formulated with oligomers, in which two fragments predominated (24 to 28 samples), as presented in Table 4.

Representative images of the SEM fractographic analysis are shown in Figure 3. Failures originated on the ceramic surface at the ceramic-RC interface ($Z=0$), which was subjected to tensile stress, and propagated throughout the material. Pores at the porcelain-RC interface seemed sometimes associated with failure origins.

4. Discussion

The results of this study demonstrate that thio-urethane-based RCs are capable of reinforcing ceramic/RC bilayers and improving the bond strength between the resin cement to both zirconia and lithium disilicate substrates. Therefore, the hypothesis of this study was accepted. Improving the bond strength of resin cements to zirconia has been a major challenge in dental materials research over the years. The inert nature of this type of ceramic substrate limits the chemical interaction with primers, adhesives and resin cements [4]. Moreover, limited topographic roughness is observed after mechanical and/or chemical-mechanical surface treatments due to the high hardness and reduced (almost absent) vitreous

phase in zirconia surfaces [27]. This study showed that is possible to increase the bond strength of RC to zirconia by producing resin cements that contain TU oligomers, especially oligomers P_AL, P_AR, or T_CC. One possible mechanism to explain this is the improved toughness of the TU-modified cement, which can provide a means for energy absorption during mechanical loading [28]. These materials have been widely used for applications where high resistance to impact is desirable, exactly because of the increased toughness provided by the flexible thiocarbamate bonds [29].

The same improvement was observed in bond strength values when oligomer-based RCs were bonded to a lithium disilicate (LD) glass-ceramic. LD has been largely utilized in dentistry because of good aesthetics and good mechanical strength associated with the ability of chemically adhere to the resin cement via silane application. The most commonly used restorations produced with LD are bridges, full or partial crowns, and veneers. LD has a significant amount of vitreous phase in its composition and, therefore can be acid etched and silane-treated. The oligomer versions P_AL or T_AL resulted in significant improvement in bond strength of the cement to the LD. This may be due to several reasons. The pendant thiol functionalities from the oligomer backbone are available to react with the vinyl bonds on the methacrylates, via chain-transfer reactions [29]. This mechanism is well-known to delay the point in conversion where gelation and vitrification takes place, with two main advantages: first, the final conversion is extended, which is desirable in the dental application since higher conversions have been correlated with increased properties [9,10,14] and decreased cytotoxicity [30]. Second, the resulting polymer network becomes more homogeneous, as demonstrated by narrower tan delta peaks by dynamic mechanical analysis [29]. Third, the already mentioned improvement in toughness and fracture toughness [14] may also contribute to the stability of the bonded interface. And finally, the decrease in polymerization stress, especially in a confined situation such as the cementation line, must have also played a role. This improvement in the mechanical properties of the cement can be speculated to have played the major role in increasing the bonding to ceramics in the present study.

The failure pattern analysis of the μ SBS specimens showed a majority of fractures involving both the RC/ceramic interface and the RC bulk structure (mixed fractures). It might be assumed that when comparing both the RC bulk and ceramic/cement interface they have similar strength, and therefore there is equal probability of fracture starting in both locations. Moreover, two main forces act into the RC and interface during μ SBS tests explaining the failure pattern observed in the study. First, tensile forces predominate over shear ones in the interface during the test and irregularities in the interface could lead to the failure initiation. The load applied might lead to relevant amount of stress concentration into the RC, making it favorable for the fracture to happen within the bulk of the structure [31]. In addition, the RC containing oligomer P_AL when bonded to zirconia showed a higher amount of cohesive fractures because of the higher bond strength.

Results of biaxial flexure strength tests showed a significant improvement in the σ_{bf} of the ceramic/RC bilayers produced with thio-urethane-based cements. In comparison with the control group, RCs composed by oligomers P_AL, P_AR, T_AL, and T_CC led to significant higher σ_{bf} at the ceramic surface ($Z=0$). In addition, all RCs composed by

oligomers increased the σ_{b1} in the RC surface ($Z=-t_2$). Micro topographical roughness features are formed in the ceramic surface after conditioning with hydrofluoric acid and removal of the vitreous phase. The resin cement fills these spaces, forming an interpenetrated resin/ceramic layer that pose an obstacle to crack propagation [19,22]. Therefore, cements with higher mechanical properties, such as those composed by TUs, are likely to produce a stronger interpenetrating layer that results in increased biaxial flexure of the bilayered structure [7].

RCs composed by oligomers showed significant lower volumetric shrinkage and polymerization stress in previous studies [9,10,14]. The higher volumetric shrinkage in the control group might have been responsible for creating interfacial defects with poor RC infiltration into the ceramic surface where crack propagation might initiate [8]. Additionally, the polymerization stress might be responsible for residual tensile stress build up at the ceramic-RC interface and around inorganic filler particles of the cement [32]. In this way, it is possible that such a higher interfacial stress in the control group may have facilitated crack propagation at the interface and around filler particles near the bonded region, therefore reducing the biaxial flexure strength of the bilayer. [33,34].

Another property that can account for this strengthening effect on the ceramic-RC bilayer is the Young's modulus (E) of the RC [6,7]. RCs with higher E might withstand higher stress concentration in the RC layer and, therefore, reduce the magnitude of stress reaching the ceramic structure [7]. Therefore, the load necessary to fracture the specimen is increased, as observed in the present study. One example of the role of E on the ceramic-RC strengthening is the number of retrieved pieces of each sample in the experiment. Samples of porcelain-RC of control group had a greater number of specimens fractured in three or four-or-more fragments (13 samples) than those containing RCs formulated with oligomers, in which two fragments predominated (24 to 28 samples). The reduction in the number of fracture fragments generated by samples of experimental groups during biaxial flexural testing suggests reduced energy storage in the ceramic bulk, instead translated into elastic strain in the RC with higher E prior to failure [35,36]. In other words, the closer the elastic modulus of porcelain and RC, the lower the amount of stress concentration in the porcelain-RC interface.

The representation of fractographic analysis in Fig 3 shows that the fracture origin is more frequently located on the ceramic surface. This was expected as the elastic deformation of porcelain is much lower than that of the RC during the biaxial flexure test. Moreover, the presence of bubbles or porosity at ceramic-RC interface served favored stress concentration in this region, leading to crack propagation through the ceramic and RC structures.

5. Conclusions

When compared to a RC composed by an organic matrix containing only traditional monomers (BisGMA/UDMA/TEGDMA), the formulation of RCs with selected thio-urethane oligomers provided increased μ SBS to zirconia and to lithium disilicate. Moreover, the use of RCs composed by oligomers increased the flexural strength of porcelain-RC bilayers.

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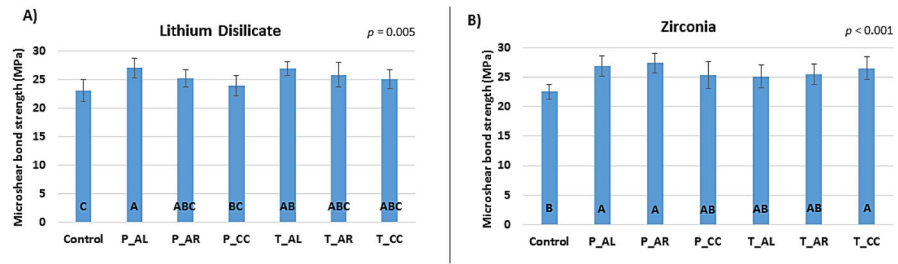


Figure 1. Microshear bond strength in (A) lithium disilicate and (B) zirconia.

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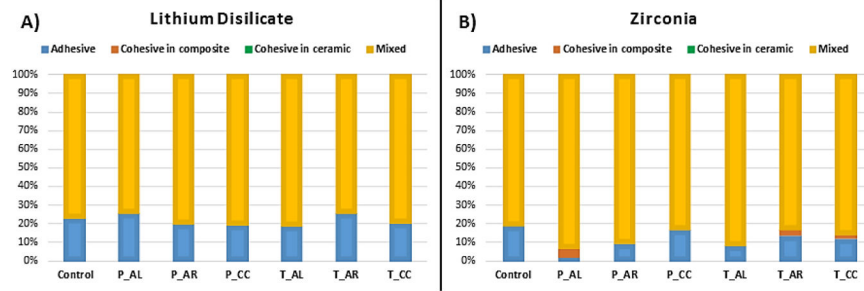


Figure 2. Failure patterns for microshear bond strength test in (A) lithium disilicate and (B) zirconia.

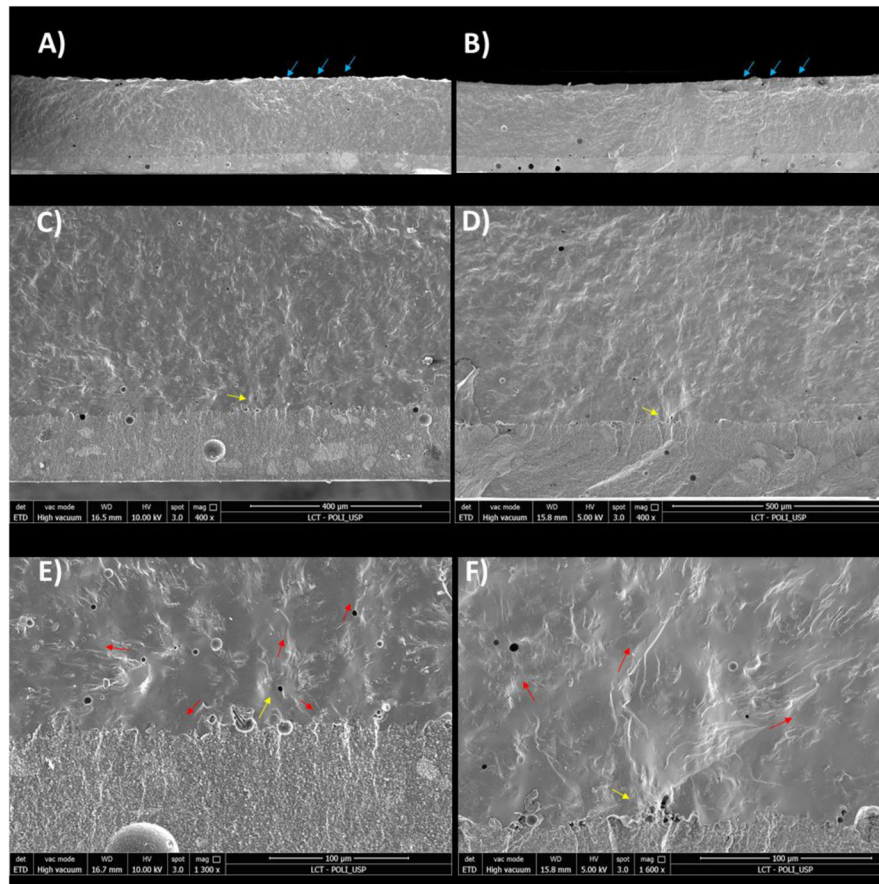


Figure 3.

SEM images of fracture surfaces. Figures 'A', 'C' and 'E' represent the control group; figures 'B', 'D' and 'F' represent a sample of porcelain bonded to a RC formulated with oligomer P_AL. Blue arrows (in Figures 'A' and 'B') indicate the compression curl; yellow arrows (Figures 'C'-'F') indicate the fracture origin; red arrows (in Figures 'E' and 'F') indicate wake hackles.

Table 1

Combination of isocyanates with thiols for the composition of the six thio-urethane oligomers used in the study.

Thiol (abbreviation)	Isocyanate (abbreviation)	Thio-urethane formed
Pentaerythritol tetra-3-mercaptopropionate (PETMP)	1,6-Hexanediol-diisocyanate (aliphatic, AL)	P_AL
Pentaerythritol tetra-3-mercaptopropionate (PETMP)	1,3-bis(1-isocyanato-1-methylethyl)benzene (aromatic, AR)	P_AR
Pentaerythritol tetra-3-mercaptopropionate (PETMP)	Dicyclohexylmethane 4,4'-Diisocyanate (cyclic, CC)	P_CC
Trimethylol-tris-3-mercaptopropionate (TMP)	1,6-Hexanediol-diisocyanate (aliphatic, AL)	T_AL
Trimethylol-tris-3-mercaptopropionate (TMP)	1,3-bis(1-isocyanato-1-methylethyl)benzene (aromatic, AR)	T_AR
Trimethylol-tris-3-mercaptopropionate (TMP)	Dicyclohexylmethane 4,4'-Diisocyanate (cyclic, CC)	T_CC

Table 2Weibull modulus (m) for microshear bond strength.

Group	m	
	Lithium Disilicate	Zirconia
Control	2.9 (2.0 – 3.8) AB	2.4 (1.7 – 3.1) A
P_AL	3.5 (2.4 – 4.6) AB	2.8 (2.0 – 3.5) A
P_AR	2.1 (1.4 – 2.7) B	3.4 (2.5 – 4.3) A
P_CC	4.9 (3.4 – 6.3) A	3.7 (2.8 – 4.6) A
T_AL	3.3 (2.3 – 4.3) AB	2.8 (2.1 – 3.5) A
T_AR	4.8 (3.3 – 6.1) A	3.1 (2.3 – 3.9) A
T_CC	3.9 (2.6 – 5.1) AB	3.4 (2.4 – 4.0) A

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Table 3Means (95% confidence intervals) for biaxial flexural strength (σ_{bf}) and Weibull modulus (m), n=30

Group	Axial position Z=0		Axial position Z=-t2	
	σ_{bf} (MPa)	m	σ_{bf} (MPa)	m
Control	119.7 (112.2 – 127.2)C	7.7 (5.8 – 10.2) A	9.5 (8.9–10.1)C	7.6 (5.7 – 10.2) A
P_AL	146 (137.7–154.4)A	7.7 (6.0 – 9.9) A	15.1 (14.2–15.9)AB	7.7 (6.0 – 9.9) A
P_AR	140.1 (132.8–147.4)AB	8.3 (6.3 – 10.9) A	14.4 (13.6–15.1)AB	8.3 (6.3 – 11.0) A
P_CC	127.5 (120.9–134)BC	8.5 (6.3 – 10.9) A	14.5 (13.8–15.3)AB	8.3 (6.3 – 10.9)A
T_AL	135.7 (130.1–141.3)AB	10.4 (8.0 – 13.4) A	13.7 (13.1–14.3)B	10.4 (8.0 – 13.4) A
T_AR	130.9 (124.9–136.9)BC	8.9 (6.7 – 11.9) A	15.5 (14.7–16.2)A	8.9 (6.7 – 11.9) A
T_CC	136 (129.4–142.6)AB	9.4 (7.1 – 12.4) A	15.6 (14.7–16.4)A	7.9 (6.0 – 10.5) A

Table 4

Number of disc fragments after tested in biaxial flexure.

Group	Number of disc fragments		
	02	03	04-or-more
Control	17	09	04
P_AL	26	02	02
P_AR	24	05	01
P_CC	25	05	
T_AL	26	04	
T_AR	28	02	
T_CC	25	05	

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