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## Cross-talk of Cannabinoid and Endocannabinoid Metabolism is Mediated via Human Cardiac CYP2J2

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## Abstract

Phytocannabinoids have well-known cardiovascular implications. For instance, 9tetrahydrocannabinol (9-THC), the principal component of cannabis, induces tachycardia in humans. In order to understand the impact of phytocannabinoids on human cardiovascular health, there is a need to study the metabolism of phytocannabinoids by cardiac cytochromes p450 (CYPs). CYP2J2, the primary CYP of cardiomyocytes, is responsible for the metabolism of the endocannabinoid, anandamide (AEA), into cardioprotective epoxides (EET-EAs). Herein, we have investigated the kinetics of the direct metabolism of six phytocannabinoids (9-THC, 8tetrahydrocannabinol, cannabidiol, cannabigerol, and cannabichromene) by CYP2J2. CYP2J2 mainly produces 11/1"-OH metabolites of these phytocannabinoids. These phytocannabinoids are metabolized with greater catalytic efficiency compared to the metabolism of AEA by CYP2J2. We have also determined that the phytocannabinoids are potent inhibitors of CYP2J2-mediated AEA metabolism, with 9-THC being the strongest inhibitor. Most of the inhibition of CYP2J2 by the phytocannabinoids follow a noncompetitive inhibition model, and therefore dramatically reduce the formation of EET-EAs by CYP2J2. Taken together, these data demonstrate that phytocannabinoids are directly metabolized by CYP2J2 and inhibit human cardiac CYP2J2, leading to a reduction in the formation of cardioprotective EET-EAs.

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WRA designed and performed the experiments, made figures, and analyzed the data and wrote the manuscript. ATW performed experiments, made figures, analyzed data, and wrote the manuscript. AD designed the experiments and wrote the manuscript. All authors analyzed the results and approved the final version of the manuscript.

The authors claim no conflict of interest.

#### Keywords

Endocannabinoids; cannabinoids; tetrahydrocannabinol; cannabidiol; cytochrome P450 2J2; anandamide

## 1. INTRODUCTION

*Cannabis sativa* has been used for centuries throughout human history for both its psychoactive effects and medicinal properties. Increasingly, legalization of cannabis for medical and recreational use is gaining worldwide support, in conjunction with trends of increased cannabinoid potency. Therefore, studying the effects of cannabinoids derived from cannabis on human health is of medical and scientific interest.

Cannabinoids are broadly classified into three categories depending on their source: (1) endocannabinoids (eCB) that are endogenously produced derivatives of polyunsaturated fatty acids (PUFAs) in animals; (2) phytocannabinoids (pCBs) that are derived from plants; and (3) synthetic cannabinoids. Psychoactive pCBs include 9-tetrahydrocannabinol (9-THC), the primary psychoactive component of the plant, 8-tetrahydrocannabinol (8-THC), and cannabinol (CBN). Some of the most abundant non-psychoactive pCBs in cannabis include cannabichromene (CBC), cannabidiol (CBD), and cannabigerol (CBG) (Figure 1).

Phytocannabinoids have well-known cardiovascular implications that have been difficult to interpret due to variations regarding their effects in different species. For instance, the cardiovascular effects of THC in animals versus humans are contradictory [1, 2]. 9-THC induces tachycardia in humans, and only reproduces similar results in conscious monkeys; and prolonged exposure resulted in a reduction in elevated heart rate, as is seen in humans with developed tolerance [3]. In other animal models, 9-THC induces bradycardia [4–6]. Interpreting animal model data is further complicated using anesthesia. Experiments using anaesthetized [5] versus non-anaesthetized [4] rats did and did not exhibit tolerance to bradycardia symptoms, respectively, despite increased 9-THC administration. This lack of consensus in cross-species studies, changing variables in experimental design, and the psychoactivity of pCBs have obfuscated focus on discerning the exact cardiovascular implications of cannabis. Therefore, in order to understand the impact of pCBs on human cardiovascular health, there is a need to study the metabolism of pCBs by human cardiac enzymes. Of interest are the cytochromes P450 (CYPs), the primary enzymes that are involved in drug metabolism in the human body.

CYPs are known for their ability to metabolize diverse xenobiotics, synthesize steroids, and be involved in fatty acid metabolism [7]. CYPs generally require electrons donated by cytochrome P450 reductase (CPR) in order to oxidize their substrates. Previously, it was demonstrated that pCBs inhibit the metabolism of drugs by microsomal CYPs (1A1 [8, 9], 1A2 [8], 1B1 [10], 2A6 [11], 2B6 [10–12], 2C8 [12], 2C9 [13–15], 2C11 [16], 2C19 [17], 2D6 [18], 3A4 [12, 19], 3A5 [12, 19], and 3A11 [20]). Currently, there is absence of any mechanistic study on the metabolism of pCBs by CYP2J2, the most abundant CYP expressed in the cardiomyocytes of the heart [8, 9].

CYP2J2 is involved in the metabolism of both  $\omega$ -3 and  $\omega$ -6 eCBs leading to the formation of eCB epoxides that are vasodilatory, anti-platelet aggregatory, anti-inflammatory, and overall cardioprotective [21]. Anandamide (AEA) was the first eCB discovered. It is derived from the  $\omega$ -6 fatty acid, arachidonic acid (AA) (Figure 1A) [22]. AEA was shown to be metabolized by several CYPs, including CYP2J2, forming different regioisomers of epoxyeicosatrienoyl ethanolamides (EET-EAs) (Figure 1A) [23, 24]. CYP2J2 has also been shown to metabolize several drugs, and many of which are known to be cardiotoxic [25–28].

Despite structural differences between eCBs and pCBs, both of these classes of cannabinoids interact with the endocannabinoid system (ECS) in the body. The ECS system consists of an ensemble of eCBs and eCB-like mediators, their corresponding receptors, and metabolic enzymes involved in ligand formation and degradation [29]. The ECS is involved in homeostatic functions dynamically regulating the functionality of the immune, reproductive, gastrointestinal, and central nervous systems, in addition to that of the brain, liver, and heart [30]. Physiological modulation by the ECS is largely dependent upon the nature and location of the diseased state [31].

The activity of the ECS is primarily elicited via molecular recognition of both pCBs and eCBs by cannabinoid receptors 1 and 2 (CBR1 and CBR2). ECS regulatory roles in the cardiovascular system are complex as the two receptors elicit varied responses [32, 33]. Activation of CBR1 translates to cardiovascular complications by decreasing blood flow and inducing vasoconstriction [34–38]; while CBR2 activity suppresses inflammatory responses of endothelial cells and monocytes [36, 39–41] and is also suggested to provide a protective role against cardiovascular diseases through induced vasodilation [40]. Thus, conflicting natures of CBR1 and CBR2 activation make the ECS a contributor to the generation, as well as amelioration, of cardiovascular disease.

Previously, we showed that CYP2J2 converts AEA into EET-EAs and similar epoxides from  $\omega$ -3 fatty acid ethanolamides that are derived from DHA and EPA [42]. During inflammation, the levels of these epoxides are comparable to AEA [42]. Inasmuch as CYP2J2 generates cardioprotective epoxides from eCBs and also metabolizes several drugs, CYP2J2 is a prime enzyme for studying the potential effects of pCBs on eCB metabolism. It has been hypothesized that AEA metabolism by unknown CYPs is inhibited by pCBs [43]. Herein, we determine the direct metabolism of selected pCBs by CYP2J2 and evaluate their effects on AEA metabolism. We have determined that the 6 pCBs tested (9-THC, 8-THC, CBC, CBD, CBG, and CBN) are all metabolized by CYP2J2 to form new hydroxyl products of the pCBs, with various di-oxygenated products as well. The catalytic efficiencies of the pCB turnovers are similar to or greater than that of AEA. Using CYP2J2-nanodiscs, we measured the kinetics of AEA metabolism by CYP2J2. Further, we have determined that the pCBs are potent inhibitors of AEA metabolism. We determined that CBG is a competitive inhibitor of AEA, whereas 9-THC, 8-THC, CBD, CBN, and CBC are noncompetitive inhibitors. Of these, 9-THC is the strongest inhibitor of AEA metabolism and reduces CYP2J2-mediated AEA metabolism to 20% of the uninhibited activity. Our study demonstrates that the pCBs, especially 9-THC, inhibits CYP2J2 and prevent the metabolism of endogenous substrates such as AEA.

## 2. EXPERIMENTAL PROCEDURES

Materials, CYP2J2 and CPR expression, nanodiscs and Soret titrations can be found in the Supplementary Materials.

#### 2.1 Metabolism of pCBs by CYP2J2

Initial metabolism identification of pCB products were determined using a lipidreconstituted system as previously described [44]. Briefly, CYP2J2 was reconstituted in 20% POPS using 0.6  $\mu$ M CYP and 0.6  $\mu$ M CPR. 60  $\mu$ M of each pCB was pre-incubated with CYP2J2 and  $\mu$ M CPR for 10 min at 37°C in 0.5 mL of 0.1 M potassium phosphate buffer (pH 7.4). Reactions were initiated upon the addition of 1 mM NADPH and allowed to react for 60 min. Reactions were terminated with 0.1 mL glacial acetic acid and extracted thrice with 0.5 mL ethyl acetate. The samples were centrifuged at 3K RPM and 4°C for 5 min to separate the layers, and the organic layers were combined and dried down under a stream of N<sub>2</sub> gas. Metabolites were resuspended in 0.1 mL acetonitrile and were identified using LC-MS/MS analysis as described below.

#### 2.2 Kinetics of pCB metabolism

Kinetics of pCB metabolism were determined using a lipid-reconstituted system as previously described [44]. Metabolites were quantified using an HPLC method (see below). 0.2  $\mu$ M of CYP2J2 and 0.6  $\mu$ M of CPR were incubated with varying concentrations of pCBs individually (5–60  $\mu$ M) for 10 min at 37°C in 0.5 mL of 0.1 M potassium phosphate buffer (pH 7.4). Reactions were initiated with 0.5 mM NADPH and allowed to react for 30 min. Reactions were terminated and extracted using ethyl acetate as stated above. Metabolism of the pCBs was determined to be linear for up to 45 min for each pCB.

#### 2.3 Kinetics of AEA metabolism

AEA metabolism by CYP2J2 was determined in Nanodiscs using 0.2  $\mu$ M of CYP2J2-NDs and 0.6  $\mu$ M CPR. AEA was pre-incubated with CYP2J2-ND/CPR for 10 min at 37°C in 0.5 mL of 0.1 M potassium phosphate buffer (pH 7.4). For inhibition experiments, AEA was pre-incubated with the CYP2J2-ND/CPR system along with 30  $\mu$ M of the tested pCB. Reactions were initiated with 0.5 mM NADPH and allowed to react for 30 min. EET-EAs were extracted using ethyl acetate as stated above, but were resuspended in 180-proof ethanol for LC-MS/MS quantification.

#### 2.4 HPLC quantification of pCB metabolism

Metabolism products were analyzed via high-performance liquid chromatography (HPLC) consisting of an Alliance 2695 analytical separation module (Waters, Milford, MA) and a Waters 996 photodiode diode array detector (Waters). Metabolites were separated in reverse phase using a Phenomenex Prodigy® 5µm ODS-2,  $150 \times 4.60$  mm column (Phenomenex, PN 00F-3300-E0, Torrance, CA). Mobile Phase A consisted of H<sub>2</sub>O + 0.1% formic acid and Mobile Phase B consisted of acetonitrile + 0.1% formic acid. Metabolites were separated using a linear gradient of 25% Mobile Phase B to 100% Mobile Phase B over 20 min, and then held at 100% Mobile Phase B for an additional 10 min. 9-THC, 8-THC, and CBD hydroxyl metabolites were monitored at  $\lambda = 281$  nm; CBG hydroxyl products were

monitored at  $\lambda = 278$  nm; and CBN and CBC at  $\lambda = 283$  nm. Standard curves for quantification were made using a 9-THC-11-OH standard for 9-THC-OH, 8-THC-OH, and CBD-OH; CBG standard for CBG-OH; CBN standard for CBN-OH; and CBC standard for CBC-OH. There was no significant variation in the absorbance intensities among 9-THC, 8-THC, and CBD. Elution times are reported in the Section 3.1.1 (Figure 2).

#### 2.5 High-resolution LC-MS and LC-MS/MS analysis of metabolites

Masses of products were determined using the Q-Exactive MS system (Thermo. Bremen, Germany) in the Metabolomics Laboratory of Roy J. Carver Biotechnology Center, University of Illinois at Urbana-Champaign as previously described [44]. Metabolites were separated using the above HPLC method and the Phenomenex Prodigy® 5 $\mu$ m ODS-2, 150 × 4.60 mm column. Negative and positive electrospray ionization was employed. Masses were considered significant if ± 5 ppm of the calculated exact mass.

#### 2.6 LC-MS/MS quantitation of EET-EAs

Samples were analyzed with the 5500 QTRAP LC/MS/MS system (Sciex, Framingham, MA) in Metabolomics Lab of Roy J. Carver Biotechnology Center, University of Illinois at Urbana-Champaign. Software Analyst 1.6.2 was used for data acquisition and analysis. The 1200 series HPLC system (Agilent Technologies, Santa Clara, CA) includes a degasser, an autosampler, and a binary pump. The LC separation was performed on an Agilent Eclipse XDB-C18 ( $4.6 \times 150$ mm,  $5\mu$ m) with mobile phase A (0.1% formic acid in water) and mobile phase B (0.1% formic acid in acetontrile). The flow rate was 0.4 mL/min. The linear gradient was as follows: 0-2min, 90%A; 8min, 55%A; 13–25min, 40%A; 30min, 30%A; 35min, 25%A; 36–43min, 90%A. The autosampler was set at 10°C. The injection volume was 10 µL. Mass spectra were acquired under positive electrospray ionization (ESI) with the ion spray voltage of +5500 V. The source temperature was 450 °C. The curtain gas, ion source gas 1, and ion source gas 2 were 32, 65, and 50, respectively. Multiple reaction monitoring (MRM) was used for quantitation: m/z 364.2  $\rightarrow$  m/z 62.0. Internal standard 14,15-EET-EA-d8 was monitored at m/z 372.1  $\rightarrow$  m/z 63.1.

#### 2.8 Competitive and noncompetitive inhibition equations used in studies

All inhibition studies are best described by a competitive inhibition model (Equation 1) or a noncompetitive inhibition model (Equation 2)

$$B = \frac{B_{max}[S]}{K\left(1 + \frac{[I]}{K_i}\right) + [S]} \quad (\text{Equation 1})$$

$$B = \frac{B_{max}[S]}{(K + [S])\left(1 + \frac{[I]}{K_i}\right)} \quad (\text{Equation 2})$$

where  $B_{max}$  and K are the Michaelis-Menten kinetic parameters of the substrate [S] without inhibitor, [I] is the concentration of inhibitor, and  $K_i$  is the inhibition constant. For kinetic experiments, B and K represent the velocity and  $K_m$ , respectively; for Soret binding experiments, B and K represent A and  $K_S$ , respectively.

### 3. RESULTS

#### 3.1.1 CYP2J2 metabolizes pCBs primarily to 1'/1"-OH metabolites

We first tested the direct endpoint metabolism of the pCBs by CYP2J2-CPR. We determined that the pCBs— 9-THC, 8-THC, CBD, CBG, CBN, and CBC—are all substrates of CYP2J2 (Figures 2–4). LC-MS/MS analysis indicates that the following classes of oxidation products were produced for each pCB: mono-oxygenation, carboxylation, and di-oxygenation (masses of the products are within 5 ppm of the predicted masses). Chromatograms for the mono-oxygenated products are shown in Figure 2. MS/MS spectra of the major mono-oxygenated products are shown in Figures 3 and 4. Excel files containing all the raw chromatogram and spectra data for the standards and metabolism experiments can be found in the Supplementary Material. A detailed analysis of the MS/MS data is provided in the Supplementary Material.

A resorcinolic ion (193.12 m/z) is produced upon the fragmentation of the 9-THC and 9-THC-11-OH standards (Figure 3A–B) [45]. The major 9-THC-OH product of the CYP2J2 metabolism shows a mass corresponding to mono-oxygenation of the resorcinolic ion followed by the elimination of a water molecule (209.12 m/z  $\rightarrow$  191.11 m/z) (Figure 3A– B). The 9-THC standard additionally fragments between Carbons 1' and 2' to produce an ion of 259.17 m/z. The CYP2J2 product shows a mass indicating mono-oxygenation (275.16 m/z) of this ion, which is not present in the 9-THC-11-OH standard, followed by the elimination of a water molecule (257.16 m/z). This identifies the CYP2J2 product as 9-THC-1'-OH. Furthermore, the 257.16 m/z peak is the largest peak in the CYP2J2 product spectrum, but not the 9-THC-11-OH standard. Fragmentation between Carbons 1' and 2' is likely facilitated by the elimination of the 1'-OH as a water molecule, thereby making it the most abundant ion.

The 8-THC, CBD, CBG, and CBC (Figures 3C–F) standards show similar fragmentation patterns as the 9-THC standard (Figure 3B). Likewise, the major mono-oxygenated products for these compounds show similar fragmentation patterns as the CYP2J2 product for 9-THC, as mentioned above (Figures 3C–F). Therefore, the metabolites for these compounds can be identified as 8-THC-1'-OH, CBD-1"-OH, CBG-1"-OH, and CBC-1"-OH. However, there are additional ions in the CBD sample that are reminiscent of ions in the 9-THC-11-OH standard, notably the resorcinolic 193.12 m/z ion (Figures 3A and 3D). Therefore, it is likely that CYP2J2 additionally produces CBD-7-OH, the analogous product compared to 9-THC-11-OH. The CBD-7-OH and CBD-1"-OH co-elute and could not be separated further.

CBN fragments differently when compared to the other pCBs due to the extended conjugation of the dibenzene moiety (Figure 4A). As a result, no ion corresponding to the separation of the resorcinol and terpenoid halves is observed in either the CBN standard or

the product (Figure 4). There are two important ions of the product that are monooxygenated fragments of the parent compound (237.13 m/z  $\rightarrow$  253.12 m/z and 195.12 m/z  $\rightarrow$  211.11 m/z) (Figure 4B). These two fragments share the dibenzene moiety and Carbons 11 and 1' (Figure 4). There are no other fragments of either the standard or the product that could distinguish between Carbons 11 and 1'. Therefore, the product is CBN-11-OH, CBN-1'-OH, or a combination of the two. The pentyl tail of CBN fragments to produce an ion of 71.09 m/z in the CBN standard, which is 69.07 m/z in the product (Figure 4B). This could be the result of a hydroxyl cleavage at the 1' position and subsequent olefination of this fragment, which supports CBN-1'-OH as the identity of the product.

#### 3.1.2 Formation of other pCB metabolites by CYP2J2

All pCBs were metabolized by CYP2J2 to produce ions suggestive of pCB carboxylation by CYP2J2. 9-THC-11-COOH was identified and compared to a standard, which is verified by a fragment resembling the carboxylated species without the pentyl side chain (Figure S1, 9-THC, Ion 9). Other carboxylated metabolites for each of the pCBs follow similar fragmentation, where an ion lacking an aliphatic carbon chain can be detected. Some pCBs (CBC, CBG, CBN) possess fragmentation suggestive of multiple carboxylation sites, due to the presence of representative ions being restricted to either positive or negative ion modes.

Each pCB was also metabolized by CYP2J2 into di-hydroxylated products in addition to the 1'/1" hydroxylation. Although MS analysis suggests their presence, the identities of these di-hydroxylated metabolites could not be determined from their MS/MS spectra due to lack of standards.

Overall, the relative abundances of these ions are much lower compared to the major hydroxylated products. Findings related to the formation of other metabolites are summarized in Figure 5 but are presented in greater detail in the Supplementary Material.

#### 3.2 Kinetics of pCB metabolism by CYP2J2

We next determined the kinetics of metabolism for the pCBs. To do this, we developed an ultraviolet/visible wavelength (UV-Vis) HPLC quantification method based on the absorbance of the resorcinol moiety ( $\lambda = 278-283$  nm) as described in the Materials and Methods section. The initial mono-oxygenated products were analyzed to determine the steady-state kinetics (Figure 6, Table 1). These correspond to 9-THC-1'-OH; 8-THC-1'-OH; CBN-1'-OH or CBN-11-OH; CBD-1"-OH and CBD-7-OH; CBG-1"-OH; and CBC-1"-OH. The other metabolites were below the detection limit.

The pCBs were effectively metabolized with differing kinetic characteristics. The  $V_{max}$  values ranged from 80.9 to 710 pmol  $\cdot$  min<sup>-1</sup>  $\cdot$  nmol<sub>CYP2J2</sub><sup>-1</sup> and the  $K_m$  values ranged from 8.27 to 84.2  $\mu$ M. 9-THC displayed the greatest catalytic efficiency (18.9 min<sup>-1</sup>  $\cdot$  nM<sup>-1</sup>). This value is twice that of the other dibenzopyran cannabinoids, 8-THC and CBN, which had catalytic efficiencies of 9.50 and 9.13 min<sup>-1</sup>  $\cdot$  nM<sup>-1</sup>, respectively. CBC, CBD, and CBG varied in their efficiencies from 4.41 to 14.9 min<sup>-1</sup>  $\cdot$  nM<sup>-1</sup>. Compared to the kinetics we had previously determined for AA, the canonical endogenous substrate of CYP2J2, these pCBs are metabolized more efficiently [46].

#### 3.3 Binding of pCBs to CYP2J2-Nanodiscs (CYP2J2-NDs)

We next determined the binding of the pCBs and AEA to CYP2J2 using UV-Vis Soret titration binding. To produce accurate binding spectra, we used CYP2J2-NDs, which solubilize CYP2J2 in a membrane mimic (Figure 7A). Substrates binding to CYPs typically displace a water molecule from the 6<sup>th</sup> coordination position of the iron-heme, producing a high-spin iron. This results in a characteristic shift in the Soret absorbance from ~417 nm to ~390 nm. The pCBs were titrated into CYP2J2-NDs and the spin-state equilibrium constants  $(K_S)$  were determined, which is used as an approximation of the  $K_D$  for CYPs. CBD, CBG, and CBN produced a measurable Soret shift upon binding (Figure 7B-E). At saturating concentrations, CBD, CBG, and CBN produced 32%, 48%, and 38% high-spin shift, respectively. Among these, CBD and CBG showed the tightest binding ( $K_S$  of 11.5 ± 3.0  $\mu$ M and  $11.1 \pm 1.2 \,\mu\text{M}$  for CBD and CBG, respectively). The binding of 9-THC, 8-THC, CBC, and AEA did not produce a significant change in the high-spin content. Therefore, to measure their relative binding affinity, we measured their inhibition of ebastine binding to CYP2J2, a technique we had previously used to determine the binding constant of AA [46]. All the pCBs displayed binding affinities that are greater than AEA, with most between 10 and 20 µM, and 9-THC being the weakest (Figure 7E).

#### 3.4 Inhibition of AEA metabolism by pCBs

Having established that the pCBs are substrates of CYP2J2, we next determined the inhibition of the CYP2J2-mediated AEA metabolism by the pCBs using a CYP2J2-ND/CPR system (Figure 8A). AEA is metabolized by CYP2J2 with a  $V_{max}$  of  $135 \pm 14.0$  $\text{pmol}_{\text{EET-EAs}} \cdot \text{min}^{-1} \cdot \text{nmol}_{\text{CYP2J2}}^{-1}$  and a  $K_m$  of 30.9 ± 7.3 µM (Figure 8A, Table 1), which is a greater catalytic efficiency compared to AA,[46] but less than that of most of the pCBs. The predominant metabolite is 14,15-EET-EA (51.7% of the total) with 5,6-EET-EA being the least (0.4%) (Figure 8A). The metabolism of AEA was repeated in the presence of 30 µM of each pCB in separate experiments (Figure 8B–G, Table 1). CBG inhibited AEA metabolism competitively with a  $K_i$  of 10.8 ± 1.4 µM. The rest of the pCBs noncompetitively inhibited AEA metabolism. The apparent  $V_{max}$  of the AEA metabolism was reduced to 20–45% of the uninhibited enzyme in the presence of 9-THC, 8-THC, CBN, CBD, and CBC (Figure 8B–G). Of these, 9-THC inhibited most strongly with a  $K_i$ of 9.86  $\pm$  0.45  $\mu$ M. The next strongest was CBD with a  $K_i$  that is almost double that of 9-THC (16.9  $\pm$  1.1  $\mu$ M for CBD). CBN, CBC, and 8-THC inhibited with  $K_i$  values around 20  $\mu$ M. For all these pCBs, the  $K_i$  values are in close approximation to their determined binding affinities, except for 9-THC, which is comparatively stronger (Table 1).

## 4. DISCUSSION

Phytocannabinoids are known for their psychotropic effects. However, their role in overall human physiology remains to be elucidated. For instance, pCBs have strong effect on the human cardiovascular physiology, including inflammation; additionally, the principal component of cannabis, 9-THC, induces tachycardia in humans. Previous studies have produced variable results regarding their effects on cardiovascular health, which is partly due to the dearth of information regarding most of the pCBs, their metabolites, and their modulation of the endocannabinoid system.

Herein, we have performed direct *in vitro* assays to determine the effects of pCBs on a human cardioprotective enzyme, CYP2J2. Given that CYP2J2 is known to metabolize drugs and convert eCBs, such as AEA, to cardioprotective epoxides makes this enzyme a prime target for investigating the impact of pCBs on the metabolism of AEA. We have determined that the 6 most common pCBs, 9-THC, 8-THC, CBN, CBD, CBG, and CBC, are mostly converted to 1'/1"-OH metabolites by CYP2J2. We further determine that these pCBs demonstrate greater catalytic efficiencies compared to typical endogenous CYP2J2 substrates. Finally, we determined that these pCBs potently inhibit AEA metabolism. While CBG inhibits competitively, the other 5 pCBs inhibit via a noncompetitive model, thereby shutting down AEA metabolism through CYP2J2. Of these pCBs, 9-THC is the strongest inhibitor of AEA metabolism.

8/ 9-THC-11-OH are the best studied metabolites of the THCs as these metabolites are psychotropic and are the most abundant metabolite formed by CYPs in liver microsomes. [47-49] Other CYP-mediated oxidations of THCs the liver involve C-ring oxidations and acyl hydroxylation, namely 3'-OH (CYP1A1) and 4'-OH (CYP2C11) [50-52]. Other organs, such as the brain and lung, produce primarily hydroxylated metabolites on the acyl chain, notably -4'-OH and -5'-OH [53, 54]. However, despite being found as a metabolite of pCBs, the 1'/1" hydroxylation was found to be a minor metabolite of other CYPs that have been tested, such as CYP1A1/2 [48]. Likewise, CBD is hydroxylated on both the C ring and pentyl side chain, and CBD-1"-OH has been primarily shown to be produced by CYP1A1/2, albeit it is still not the major metabolite of these CYP enzymes [8]. The primarily 1'/1" hydroxylation is therefore unique to CYP2J2 and may be used as a probe regarding its activity. In smaller quantities, we also propose that CYP2J2 generates 8/ 9-THC-11-COOH, CBN-11/5"-COOH, CBD-7-COOH, CBG-8'/9'/10'-COOH, and CBC-9'/5"-COOH. Throughout the literature, THC and CBD are the best studied cannabinoids with respect to CYP metabolism. Herein, we measure the CYP2J2-mediated metabolism of CBC, CBG, and CBN that are not extensively studied but nonetheless very important constituents of cannabis.

All the pCBs inhibited CYP2J2-mediated AEA metabolism. Interestingly, two modes of AEA inhibition by the pCBs were observed. CBG inhibited AEA metabolism competitively. Compared to the other pCBs, CBG has a structure that is the most linear and most resembles the structure of AEA. We have previously determined that the competitive nature of PUFAs arises from their flexibility, which allows them to occupy most of the active site volume [46]. Likewise, CBG and AEA may be competitive for similar reasons.

The other pCBs inhibit AEA metabolism in a noncompetitive manner. Classically, noncompetitive inhibition arises from the binding of the inhibitor to an allosteric site, preventing the metabolism of a substrate when both are bound. CYP2C9's metabolism of diclofenac was shown to be noncompetitively inhibited by 6-hydroxyflavone in an atypical manner. 6-hydroxyflavone was shown to bind into a site that prevents full access of diclofenac to the heme [55]. Since these pCBs are also substrates of CYP2J2, they must bind near the heme. Therefore, in addition to the classical mode of noncompetitive inhibition, the pCBs may be binding the active site along with AEA and preventing AEA from being metabolized. We had previously demonstrated that 7-deoxydoxorubicin aglycone

concurrently binds the active site with AA and alters AA metabolism by changing the binding conformation of AA [44]. This is mediated in part to the polycyclic and planar structure of 7-deoxydoxorubicin aglycone. The rest of the pCBs other than CBG likewise have polycyclic structures, giving them less flexibility.

Another intriguing characteristic of the AEA inhibition by these pCBs is the differences in the inhibitory strength concerning the dibenzopyran varieties, 9-THC, 8-THC, and CBN. Among these, 9-THC has a  $K_i$  value for the inhibition of AEA that is twofold tighter than 8-THC or CBN. Coincidentally, the catalytic efficiency of the 9-THC metabolism was determined to be twice that of the metabolism of 8-THC or CBN. These data demonstrate that the positioning and frequency of double bonds in the C-ring of the dibenzopyran pCBs governs the metabolism of these pCBs and the inhibition of AEA metabolism, with the 9 position being crucial.

All the pCBs tested herein potently inhibit AEA metabolism by CYP2J2. These data have intriguing physiological implications, particularly in relation to trafficking and molecular recognition of cannabinoid-like ligands within the ECS. The pCBs have been shown to stall the catabolism of AEA by inhibiting fatty-acid-binding proteins that aid in the delivery of AEA to fatty acid amide hydrolase (FAAH) [56], and FAAH has been shown to be weakly inhibited by CBD [57]. As a result, the extended lifetime of AEA [57] results in greater targeting of CBR1, which is further amplified by the activation of CBR1 by select pCBs themselves. Through noncompetitive inhibition, our data shows that in addition to pCB signaling through CBR1 receptors, they also prevent CYP2J2-mediated AEA metabolism. As EET-EAs are CBR2-preferring agonists [30, 42, 58], pCBs prevent the formation of CBR2-preferring ligands in addition to increasing actions at CBR1. Interestingly, a study performed on Rhesus monkeys shows that 8-THC-1'-OH is the least effective at inducing behavioral changes associated with cannabis use [59]. As CBR1 is responsible for the psychotropic effects of cannabinoids, this presumes that 8-THC-1'-OH is the weakest agonist of CBR1. Such crosstalk is not only mediated by CYP metabolism, but also by CBR signaling. It has been shown previously that selective blockade of CBR1 receptor and activation of CBR2 is ideal for the treatment and management of cardiovascular disease [60]. Therefore, the inhibited formation of CBR2-preferring EET-EAs and stronger CBR1 activation by select pCBs may explain the increase risk of myocardial infarction after the first hour of cannabis use. CYP2J2 then clears the pCBs by converting them to weaker CBR1 agonists and resumes normal EET-EA metabolism after the first hour and the cardiotoxic risk is lowered. Taken together, this study indicates that pCBs will elicit complex cardiovascular physiology that will be partially mediated through inhibition of human cardiac CYP2J2.

### Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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## ABBREVIATIONS

8-THC	8-tetrahydracannabinol
9-THC	9-tetrahydracannabinol
AEA	anandamide
AA	arachidonic acid
CBC	cannabichromene
CBD	cannabidiol
CBG	cannabigerol
CBR	cannabinoid receptor
CBN	cannabinol
-СООН	carboxy
СҮР	cytochrome P450
CPR	cytochrome P450 reductase
eCB	endocannabinoid
ECS	endocannabinoid system
EET	epoxyeicosatrienoic acid
EET-EA	epoxyeicosatrienoyl ethanolamide
-OH	hydroxy
LC-MS/MS	liquid chromatography-tandem mass spectrometry
ND	Nanodisc
рСВ	phytocannabinoid
PUFA	polyunsaturated fatty acid

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0 -

CH<sub>3</sub>

CH3

Endocannabinoids (eCBs) А HO 8,9-EET-EA 5,6-EET-EA HC CYP2J2 CH CH HO Anandamide AEA 11,12-EET-EA 14,15-EET-EA CH в Phytocannabinoids (pCBs) **Dibenzopyran Numbering** 7<sub>CH</sub> 11 А H<sub>3</sub>C  $CH_3$ 8// ัด H₃C 13 4 10 1' 3' 5' 5 9 Δ9-Tetrahydrocannabinol **∆9-THC** CBD CH<sub>3</sub> З 10' OH H<sub>3</sub>C н НΟ H₃Ċ  $H_3($ CH. 8 Й₃С 5 Δ8-Tetrahydrocannabinol Cannabigerol **∆8-THC** CBG

## CH<sub>3</sub> OH H<sub>3</sub>C CH H<sub>2</sub>( Cannabinol

CBN



8 HO CH 1" 3" 5" Cannabichromene CBC

#### **Figure 1. Chemical structures**

(A) Endocannabinoids (eCBs): anandamide (AEA) and epoxyeicosatrienoyl ethanolamides (EET-EEAs). (B) Phytocannabinoids (pCBs). The numbering for each pCB is given, and 8tetrahydrocannbionol (8-THC) and cannabinol (CBN) follow analogous numbering as 9tetrahydrocannabinol ( 9-THC). These psychoactive pCBs follow the dibenzopyran numbering system. The non-psychoactive pCBs in cannabis include cannabidiol (CBD), cannabigerol (CBG), and cannabichromene (CBC). They follow a monoterpenoid numbering scheme. The resorcinol ring for each pCB is designated as the A-ring.



pCBs

## Mono-oxygenated-pCBs

Substrate	Mono-oxygenated	Product	Predicted m/z
	Product	Formulae	[M + H]⁺
CBC	CBC-OH	C <sub>21</sub> H <sub>30</sub> O <sub>3</sub>	331.22677
∆8-THC	Δ8-THC-OH	$C_{21}H_{30}O_3$	331.22677
Δ9-THC	Δ9-THC-OH	C <sub>21</sub> H <sub>30</sub> O <sub>3</sub>	331.22677
CBN	CBN-OH	C <sub>21</sub> H <sub>26</sub> O <sub>3</sub>	327.19547
CBD	CBD-OH	C <sub>21</sub> H <sub>30</sub> O <sub>3</sub>	331.22677
CBG	CBG-OH	C <sub>21</sub> H <sub>32</sub> O <sub>3</sub>	333.24242







Figure 3. MS/MS spectra of the major mono-oxygenated products of 9-THC, 8-THC, CBD, CBG and CBC

(A) Schematic of 9-THC fragmentation pathways for the 9-THC and 9-THC-11-OH standards and the CYP2J2-mediated 9-THC-OH product. Analogous ions are identified for the other pCBs. (B) Spectra of the 9-THC standard, 9-THC-11-OH standard, and the major CYP2J2 product of 9-THC. (C–F) Spectra of the (C) 8-THC-OH product, (D) CBD-OH product, (E) CBG-OH product, and (F) CBC-OH product. Corresponding standard spectra for (C–F) are shown as grey dashes and grey font. Elution times of products are given. Masses that indicate 1'/1"-OH as the product are labelled in red and underlined. Masses were considered significant if within 5 ppm of the predicted masses.

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## Figure 4. MS/MS spectra of the CBN standard (std) and the major mono-oxygenated CYP2J2 product (CBN-OH) $\,$

(A) Spectra of the CBC standard and scheme of major fragmentations. (B) The major CYP2J2 product of CBC and major fragmentations. Elution times of products are given. Blue, dashed brackets indicate sites of fragmentation. Masses that indicate 1'-OH are labelled in red and underlined. Masses were considered significant if within 5 ppm of the predicted masses.



#### Figure 5. Major products of CYP2J2-pCB metabolism

Question marks (?) are positioned to suggest multiple species derived from the fragmentation pathways for each metabolite (reference Supplementary Materials). Metabolites are as follows (left to right, row by row): 9-THC-1'-OH, 9-THC-11-COOH, 9-THC-DiOH; 8-THC-1'-OH, 8-THC-11-COOH, 8-THC-DiOH; CBN-1'-OH, CBN-5"/11-COOH, CBN-DiOH; CBD-1'-OH, CBD-7-COOH, CBD-DiOH; CBG-1'-OH, CBG-8'/9'/10'-COOH, CBG-DiOH; CBC-1'-OH, CBC-5"/9'-COOH, CBC-DiOH. Structures of -DiOH products are representative as the position of the second OH could not be determined.



Figure 6. Kinetics of pCB metabolism

Mono-oxygenated products identified by LC-MS/MS analysis for the metabolism of (A) 9-THC, (B) 8-THC, (C) CBN, (D), CBD, (E) CBG, and (F) CBC were quantified using UV-HPLC as stated in the Materials and Methods section. Data fit to the Michaelis-Menten equation. Error represents the SEM of 3 experiments.



#### Figure 7. Soret binding titration

(A) CYP2J2 in Nanodiscs. (B) CBD, (C) CBG, and (D) CBN produced a measurable Soret shift with 32%, 48%, and 38% high-spin shift, respectively. (E) Binding parameters as determined directly from the Soret titration or through inhibition of ebastine (EBS) binding. Data displayed one-site binding and data shown are the mean ± SEM of 3 titrations.

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(A) Kinetics of AEA metabolism. A schematic of CYP2J2-Nanodiscs and CPR metabolizing AEA is given. The total amount of EET-EAs are plotted. The percent of each regioisomer to the total amount of EET-EAs is given to the right. (B–G) Inhibition of AEA metabolism by the indicated pCB. AEA data from Panel A is shown as grey for comparison and the model of inhibition is given. 30  $\mu$ M of each pCB was used to determine inhibition. Error represents the SEM of 3 experiments.

Table 1

Kinetic parameters of experiments.

		Metaboli	m		AEA Metab	olism Inhibition
	Metabolite(s)	$V_{max}a$	$K_m$ ( $\mu$ M)	$k_{cat} \cdot K_m^{-1b}$	$K_i$ ( $\mu$ M)	Type
AEA	EET-EAs	$135 \pm 14$	$30.9 \pm 7.3$	4.47	1	1
9-THC	HO-'1-	$211 \pm 28$	$11.9 \pm 4.8$	18.7	$9.86 \pm 0.45$	Noncompetitive
8-THC	HO-'1-	$80.9\pm14.3$	$8.52\pm5.30$	9.50	$20.2 \pm 1.9$	Noncompetitive
CBN	-1'-OH or -11-OH	$710\pm248$	$77.8 \pm 42.2$	9.13	$19.4\pm2.0$	Noncompetitive
CBD	-7-OH + -1"-OH	$372\pm100$	$57.2\pm26.5$	6.50	$16.9 \pm 1.1$	Noncompetitive
CBG	-1"-OH	$123 \pm 14$	$8.27 \pm 3.24$	14.9	$10.8\pm1.4$	Competitive
CBC	HO-"I-	$371 \pm 134$	$84.2\pm46.2$	4.41	$21.1 \pm 1.4$	Noncompetitive
a (pmol · mir	1-1 · nmolCYP2J2 <sup>-1</sup>					

 $b_{(nM^{-1} \cdot min^{-1})}$