

Structures and disulfide cross-linking of *de novo* designed therapeutic mini-proteins

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Recent advances in computational protein design now enable the massively parallel *de novo* design and experimental characterization of small hyperstable binding proteins with potential therapeutic activity. By providing experimental feedback on tens of thousands of designed proteins, the design-build-test-learn pipeline provides a unique opportunity to systematically improve our understanding of protein folding and binding. Here, we review the structures of mini-protein binders in complex with Influenza hemagglutinin and Bot toxin, and illustrate in the case of disulfide bond placement how analysis of the large datasets of computational models and experimental data can be used to identify determinants of folding and binding.

Introduction

Two iterations of massively parallel design-build-test-learn cycles resulted in hundreds of high-affinity designed mini-protein binders [1] against influenza H1 hemagglutinin (HA) and the receptor-binding domain of botulinum neurotoxin B (BoNT H_CB). Mini-proteins Bot.671.2 (PDB: 5VID) and HB1.6928.2.3 (PDB: 5VLI) obtained in these experiments were co-crystallized with their respective targets. For Bot.671.2 (see Fig. 1A), the asymmetric unit of the crystal contains five copies of the target H_CB, four in complex with the mini-protein binder. All the copies of Bot.671.2 match the design closely (RMSD-monomer_{C α} < 1.0 Å). For HB1.6928.2.3 (see Fig. 1B), the asymmetric unit contains one copy of the molecule in complex with HA which closely recapitulates the computational design (RMSD-monomer_{C α} < 1.1 Å). Deep mutational scanning [2] revealed that the mini-protein binders can tolerate a large number of mutations without negatively affecting binding; the exceptions are amino acids at

the binding interface and polar amino acids in the vicinity of the binding surface (likely due to electrostatic effects [3]). In agreement with control experiments in which lower design success was obtained with simple gly/ser loops than with Rosetta designed loops, amino acids in many of the loop positions cannot be indiscriminately replaced by glycines or serines.

We studied the effect of systematically introducing all geometrically allowable disulfide bridges [4] into each of the BoNT designs (3193 combinations of disulfides for 141 initial nondisulfided designs). We found that 23% of all the pairs tested improved binding (or stability), 51% did not affect binding, and 26% were detrimental. We found a strong correlation between the number of interaction hotspot residues spanned by the disulfide and binding (see Fig. 2B). For example, ~64% of the designs that have average binding energy per-residue < -3.25 kcal·mol⁻¹ and whose disulfides enclose at least four hotspots improved protein

binding or stability. Starting from Bot.728.3 (see Fig. 2A), we used this information to synthesize an improved disulfide protein variant named Bot.2110.4 (disulfide C5-C28) and co-crystallized it with H_cB (PDB: 5VMR, see Fig. 2C). The experimental X-ray structure closely resembles the design model (RMSD-

monomer_{C α} < 1.4 Å), has a correctly formed disulfide bond, and thermal denaturation experiments show that it is hyperstable (T_m > 95 °C). Thus, even for small mini-proteins [4,5], there are many potential ways to introduce covalent staples to further enhance stability and function.

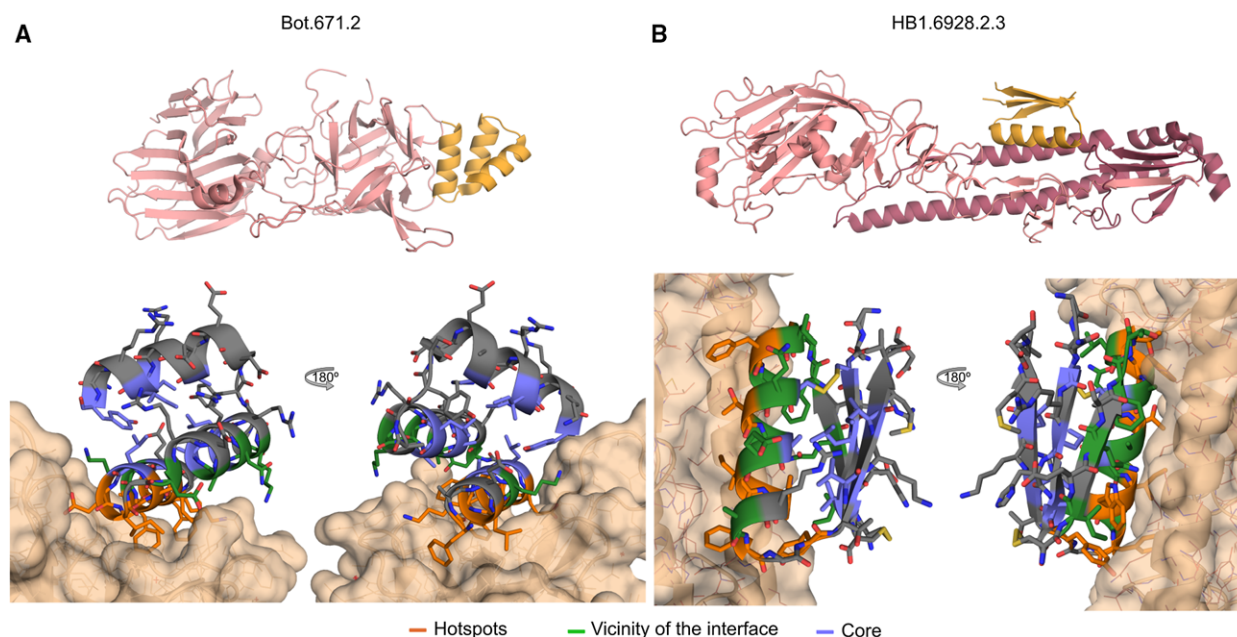


Fig. 1. Structure of *de novo* designed binders. (A) The co-crystal structure of the binder Bot.671.2 in complex with BoNT H_cB. Top: The binders (orange color) in the context of the target (pink color). Bottom: Close-up view of binding site. One (chains G and D) of the four copies of the complex present in the asymmetric unit of the crystal is shown. (B) Same as 'A', for the co-crystal structure of mini-protein binder HB1.6928.2.3 in complex with HA.

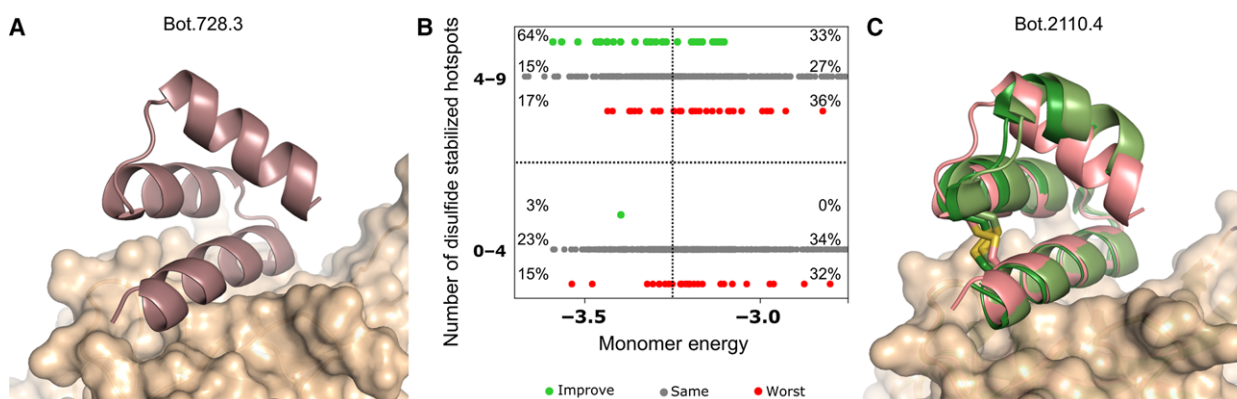


Fig. 2. Disulfide stabilization of *de novo* designed mini-protein binders. (A) The designed structure (cartoon representation) for the binder Bot.728.3 (the nondisulfide parent of Bot.2110.4, see 'C'). (B) Effect on function of introduced disulfides on BoNT. X-axis, the Rosetta monomer energy (kcal·mol⁻¹ per-residue). Y-axis, the number of hotspot residues between the two cysteines forming the disulfide, green, disulfide designs with improved binding (or stability), red, disulfide designs with decreased binding, grey, designs with no change in binding. Disulfide designs with lower energy and more hotspots enclosed are more likely to be improved. (C) Co-crystal structure of binder Bot.2110.4 (cartoon representation) bound to H_cB (surface representation). Two copies of Bot.2110.4 in the asymmetric unit are shown in different shades of green, the computational design is in pink. The disulfide (sticks representation) is very similar in the design model and crystal structure.

Methods

Crystallographic structures are available in the RCSB protein databank (www.rcsb.org) [6]. The dataset of computational (Rosetta) and experimental metrics is publicly available in the zenodo repository (www.zenodo.org, <https://doi.org/10.5281/zenodo.838815>). Data analysis was performed in Python (<http://www.python.org>) using the iPython shell [7] and the NumPy module [8]. For the systematic scan of disulfides in BoNT binders (see Fig. 2B), a given design was considered to improve (or vice-versa) if the change in binding was at least two categories (where the possible binding categories are as follows: not binding, 100, 10, 1, and 1 nM + protease resistant).

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