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Enantioselective, Ketoreductase-Based Entry into Pharmaceutical Building Blocks:

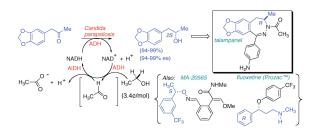
Ethanol as Tunable Nicotinamide Reductant

Sylvain Broussy, **Ross W. Cheloha**, and **David B. Berkowitz**^{*} Department of Chemistry, University of Nebraska, Lincoln, NE 68588-0304

Abstract

The use of NADH- and NADPH-dependent ketoreductases to access enantioenriched pharmaceutical building blocks is reported. Seven structurally diverse synthons are obtained, including those for atomoxetine (KRED 132), talampanel (RS1-ADH and CPADH), Dolastatin (KRED 132) and fluoxetine (KRED 108/132). Ethanol may be used as stoichiometric reductant, regenerating both nicotinamide cofactors, particularly under four-electron redox conditions. Its favorable thermodynamic and economic profile, coupled with its advantageous dual cosolvent role, suggests a new application for biomass-derived ethanol.

Abstract



As has been pointed out in a recent overview from the Merck Process Group,¹ advances in ketoreductase (KRED or alcohol dehydrogenase = ADH) technology have increased their potential for process chemistry. Asymmetric enzymatic reductions, *ex vivo*, are now more easily investigated in the research laboratory, and may be optimized there, under controlled conditions, offering a viable and complementary alternative to *in vivo* approaches, for example, in genetically engineered yeast² or *E. coli*.³ The *ex vivo* system circumvents issues of substrate, product and cosolvent toxicity, provided that enzyme activity and enantioselectivity are preserved.

We have a standing interest in the use of enzymes in asymmetric synthesis, for example, to access enantiomerically enriched podophyllum lignans⁴ or quaternary, α -vinyl amino acids.⁵ More recently, that focus has turned to ADH's, as catalytic reporting enzymes to facilitate

dbb@unlserve.unl.edu.

Supporting Information Available. Details of the synthetic and enzymatic chemistry, and spectroscopic and chiral HPLC characterization of products. This material is available free of charge via the internet at http://pubs.acs.org.

the evaluation of organometallic catalysts via ISES (In Situ Enzymatic Screening).^{5, 6} Parallel to these studies, we have undertaken to exploit ketoreductases in target-directed asymmetric synthesis. Indeed, the repertoire of enzymes in modern asymmetric synthesis continues to expand, including lipases,⁷ amidases,⁸ amine oxidases,⁹ alcohol¹⁰ and amine DH's,¹¹ epoxide hydrolases¹² and aldolases,¹³ among others.¹⁴

In this work, we have focused upon an array of ketones, the asymmetric reduction of which provides valuable pharmaceutical building blocks. In Table 1, each chiral secondary alcohol product is mapped (red shading) onto the pharmaceutical for which it is a synthon. The Aprepitant-leading ketone **1**, served as a model for our *ex vivo* conditions, giving high (*S*)-selectivity with CPADH and HLADH, consistent with reports from Merck¹⁵ and Rhodia,¹⁶ respectively. The second ketone screened serves as the substrate for a classic biocatalytic process (*Z. rouxi* whole cell route-Zmijewski group at Lilly¹⁷) for the production of Talampanel. Our screen identified two new DH's here, CPADH and RS-1 ADH, each of which also gives the correct antipode (*S*)-**4**, with high selectivity.

Ketones **5** and **7**, respectively, are precursors to building blocks for the promising chemotherapeutic candidate, Dolastatin 10, and Mitsubishi's broad spectrum fungicide MA-20565, respectively. In the former case, Genet has reported the use of stoichiometric DIP-Cl (92% ee),¹⁸ whereas Masui employs a diphenylprolinol-ligated borane reagent (92% ee).¹⁹ The highly enantioselective reductions seen here (KREDs 108 and 132) open up alternative "green" processes. Similarly, while both Ru(II)-diamine²⁰ and Rh-diamine-based²¹ asymmetric hydrogenations of **7** have been reported, reductions with CPADH, RS-1 ADH and KRED 132, uncovered in these studies, provide viable biocatalytic alternatives.

The final three entries (9, 11, 13) in Table 1 are precursors to either (*R*)-Strattera or (*R*)-Fluoxetine. While there are isolated reports of whole cell procedures for the asymmetric carbonyl reduction of 11, either with *Saccharomyces*²² or *Rhodotorula*²³ species, we find no previous literature descriptions of asymmetric biocatalytic reductions of either 9 or 13. In this regard, the success we have had with KRED 132, in both cases, is quite notable. The ee's are certainly competitive with those seen using Itsuno-Corey oxazaborolidine reduction (Senanayake),²⁴ in the former case, or Pd(II)-sparteine-mediated oxidative kinetic resolution (Stoltz),²⁵ in the latter.

With a half dozen promising new DH-based asymmetric reductions, in hand, we next set about to examine cofactor regeneration. The most commonly used nicotinamideregenerating reagents, with favorable thermodynamics, are collected in Figure 1, and compared with EtOH. Note that van der Donk and Zhao²⁶ have recently opened the door to phosphite-based reductions, with the most favorable redox potential of the group. While Wong and Whitesides²⁷ established the potential for using EtOH in biocatalytic reductions with water soluble substrates, use of this reductant for chemoenzymatic synthesis has lagged behind. However, EtOH is attactive here in (a) having a favorable redox potential, (b) being economically priced and readily available from the biomass fermentation stream, and (c) potentially serving a dual role as organic cosolvent. Regarding the first point, employing EtOH as a four electron reductant provides for more favorable thermodynamics, which result

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from the highly exergonic reduction of NAD(P) with acetaldehyde, provided that aldehyde DH (AlDH) activity is present.

This tunability of the EtOH reductant was examined in a model NMR experiment (Figure 2), with KRED 132 and ketone **9**. KRED 132 requires NADPH. We have found that LKADH can effectively be used to oxidize EtOH with NADP. In our hands, yeast AlDH also efficiently utilizes NADP. So, this LKADH/YAIDH couple was employed to access the full four electron reducing capacity of EtOH (panel A), and compared with the reaction under two electron redox conditions (no YAIDH, panel B, Le Chatelier effect alone). In fact, the reduction run under four electron reducing conditions proceeds much more rapidly. As expected, one sees the clear AcOH signature in the former case, attesting to the four electron redox cycle in play. Table 2 illustrates the use of these four electron conditions across three different substrates and four different DH's at the mmol scale.

In summary, the first viable ketoreductase-based entries into secondary alcohol building blocks for Dolastatin 10 (5), Prozac (9) and Strattera (13) are presented here, as are new biocatalytic entries into building blocks for Talampanel (3) and MA-20565 (7). The viability of using biomass-derived EtOH for cofactor regeneration is examined, and the advantage of using four electron redox cycles in such processes is demonstrated. Future studies will further probe the scope, limitations and optimal conditions for such "green" alternatives to transition metal- or boron hydride-based chiral carbonyl reductants for asymmetric process chemistry.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgment

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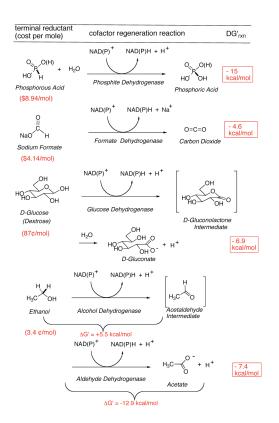


Figure 1.

Thermodynamics of nicotinamide cofactor regeneration – Tunability of the ethanol reductant.

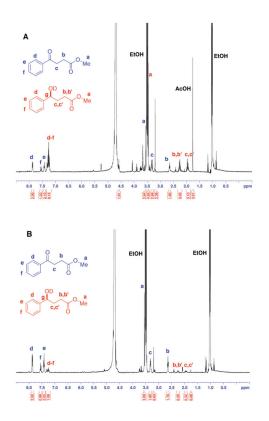


Figure 2.

Comparison of the KRED-132-mediated reduction of ketone **9** with NADPH (@ 2 mol %) regeneration using LKADH (50 mM KPO₄ in D²O, pD 7.5; 300 rpm, 30 °C, 3 h), both with (panel **A**) and without (panel **B**) YAIDH (see Supporting Information for details). Note the increased conversion and AcOH production under four electron reduction conditions.

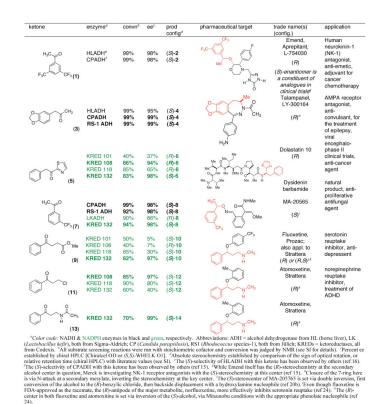




Table 2
Biocatalytic Reductions @ the mmol Scale – Ethanol as Four Electron Reductant

chiral product	ADH	regen	cofactor	yield	ee
F		system	(mol %)	5	
Сн ₃ (4)	CP- ADH	YADH/ YAIDH	NAD+ (0.4)	89%	94% (<i>S</i>)
OH CO ₂ Me (10)	KRED 132	LK ADH/ YAIDH	NADP+ (1)	86%	96% (<i>S</i>)
H ₃ C,OH (8)	RS-1 ADH	YADH/ YAIDH	NAD+ (1)	98%	99% (<i>S</i>)
H ₃ C OH (8)	LK ADH	(LK ADH)/ YAIDH	NADP+ (2)	64%	86% (<i>R</i>)

All reductions were performed on a 1 mmol scale at 30 °C, 300 rpm, pH 7.5 with the cofactor regeneration systems shown. See Supporting Information for details.

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