

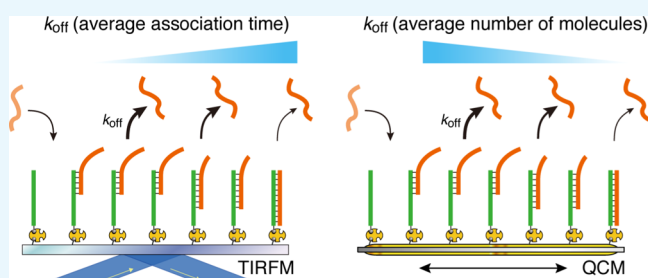
Probing Multiple Binding Modes of DNA Hybridization: A Comparison between Single-Molecule Observations and Ensemble Measurements

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Supporting Information

ABSTRACT: Interactions between biomolecules are generally analyzed by ensemble measurements, assuming that the interactions occur in a single binding manner. However, such interactions may occur via multiple binding modes. We investigated the kinetics of DNA hybridization as a multiple dynamic model of biomolecular interactions. Two kinetic analyses were performed with a single-molecule observation using total internal reflection fluorescence microscopy (TIRFM) and with ensemble measurements using a quartz-crystal microbalance (QCM) biosensor. We observed the DNA hybridization of 8 and 12 bp DNAs with random sequences and dA₁₂–dT₁₂ and calculated the kinetic parameters, including the dissociation rate constant (k_{off}). Hybridization of 8 bp DNA proceeded mainly via a single binding mode. However, hybridization of 12 bp DNA indicated at least two different binding modes and dA₁₂–dT₁₂ hybridization showed multiple binding modes. For the multiple binding interactions, the kinetic parameters obtained from TIRFM and QCM were different because kinetic parameters obtained from QCM indicate average number of molecules, whereas those from TIRFM indicate average association time. The present study revealed the details of multiple interactions, which can be utilized for better understanding of not only DNA hybridization but also biomolecular interaction mechanisms.



INTRODUCTION

Biomolecular interactions influence biological functions and are more complex than chemical reactions owing to molecular conformational changes and fluctuations among biological macromolecules.^{1,2} Therefore, investigating kinetic biomolecular interactions is required to understand biological systems in detail. Biosensors, such as a surface plasmon resonance (SPR) biosensor, have been used for quantitative analysis of reaction kinetics as well as binding association.³ A quartz crystal microbalance (QCM) biosensor can also be used for quantitative kinetic analysis. A host molecule-immobilized QCM enables the detection of interactions with guest molecules as a function of time because the resonance frequency decreases linearly with the increase in mass on the sensor surface at the nanogram level.^{4,5} Kinetic analysis yields the association and dissociation rate constants and dissociation constants (k_{on} , k_{off} , and K_{d} , respectively). We have previously used a 27 MHz QCM device to study the kinetics of DNA–DNA,⁶ DNA–protein,⁷ DNA–intercalator,⁸ and other biomolecules.^{9–11} However, biosensors including SPR and QCM detect the total amount of substance on the sensing area, resulting in an ensemble measurement of the binding/dissociation behaviors. Hence, it is difficult to assess the behavior of each biomolecule during interactions via multiple binding modes, resulting from conformational fluctuations, using an ensemble measurement.

Experimental methods designed for single-molecule detection can be utilized to identify the binding mode of each molecule. Atomic force microscopy and optical tweezers are emerging as powerful tools to detect a single-molecule event.¹² Furthermore, single-molecule detection targeting DNA molecules, whose physical properties are well defined among biomolecules, has been reported as follows. Protein nanopores such as α -hemolysin^{13,14} and porin A¹⁵ have gained attention as a single-molecule analytical tool for DNA detection and sequencing. Previously, an array of femtoliter-sized reaction vessels was utilized to detect the single-molecule DNA hybridization by enzymatic signal amplification.¹⁶ In addition, localized SPR¹⁷ and micromechanical technique¹⁸ were used for in situ sensing of single binding events of nanoparticle-labeled DNA strands. Carbon nanotube field-effect transistor provided information about the single-molecule binding and dissociation of DNA hybridization in terms of activation energy.¹⁹

Total internal reflection fluorescence microscopy (TIRFM) was developed to observe a single-molecule behavior using a fluorescence labeling technique.²⁰ In TIRFM, the guest molecules labeled with a fluorescent dye can be detected in the range of an evanescent field on a glass plate immobilized

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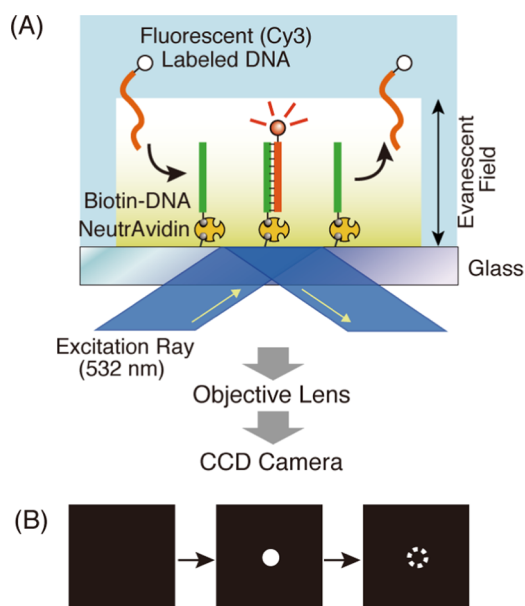


Figure 2. Schematic illustration of the procedure for single-molecule detection of DNA hybridization. (A) Evanescent field generated on the glass surface by the excitation ray (532 nm) contributes to a single-molecule detection during hybridization between Cy3-labeled DNA and biotinylated DNA immobilized on the glass plate. Fluorescence is transmitted through the objective lens, which is detected by the CCD camera. (B) A hybridization event is indicated by a bright spot on the CCD camera images.

molecules was used to construct the DNA-displayed glass plate (area: 1.13 cm²). In case of maximum immobilization, the distance between the probe DNAs was calculated to be 554 nm (occupied area: 2.7×10^5 nm²/probe-DNA molecule). The immobilization amount is low enough for one probe DNA to be located per ~ 10 pixels of the CCD camera to obtain one fluorescent-labeled DNA per pixel.

Single-Molecule Observation of DNA Hybridization.

We observed the DNA hybridization of Cy3-labeled DNAs with probe DNA on the glass plate of the reaction cell by TIRFM (Figure 1C). When the reaction cell with the probe DNA (8 mer) was filled with 70 μ L of 1 nM Cy3-labeled target-DNAs (8 mer) solution, we found many countable bright spots in the CCD-camera image (Figure S1A). However, in the case of no probe-DNA immobilization, very few bright spots were observed from the Cy3-labeled target DNAs displaying nonspecific binding with the plate in the same condition (Figure S1B). These results indicate that (1) the Cy3-labeled DNAs bound to probe DNAs, (2) the Cy3 fluorescent dye emitted fluorescence by evanescent light, and (3) its emission was detected with the CCD camera (Figure 2).

Next, we obtained the CCD-camera images continuously at 30 s intervals, resulting in the movies showing many blinking spots. The intensity of a blinking spot in one area of the images was plotted against time in the case of a single-molecule hybridization of 8 mer–8 mer, 12 mer–12 mer, and dA₁₂–dT₁₂ (Figure 3). They showed short or long pulses. The similar intensity of the short or long pulses in each graph reflected fluorescent emission from one molecule of Cy3-labeled target DNA. Thus, a high-intensity step in contrast to the baseline would indicate DNA hybridization, and the duration of the step shows the association time for binding between the target DNA and the probe DNA. We found that the association time of the

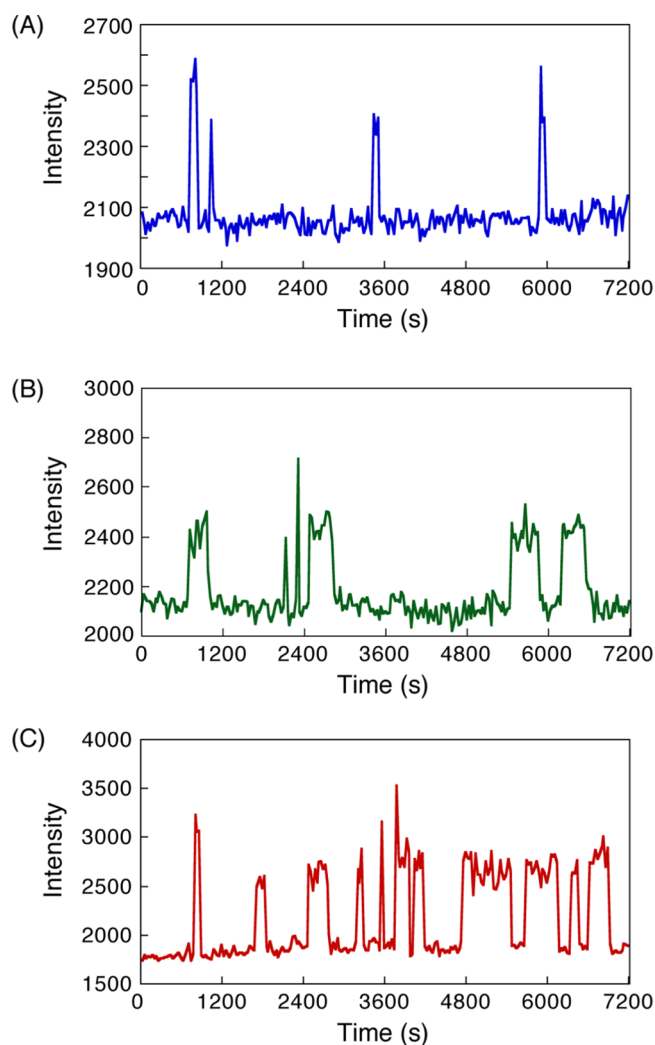


Figure 3. Time courses of fluorescent intensity of bright spots corresponding to the binding and dissociation of single-molecule hybridization in (A) 8 mer–8 mer, (B) 12 mer–12 mer, and (C) dA₁₂–dT₁₂ DNAs. Time-lapse images were obtained at an interval of 30 s. Experimental conditions: 200 mM NaCl Tris-EDTA (TE) buffer (10 mM Tris-HCl, pH 8.0, 1 mM ethylenediaminetetraacetic acid (EDTA), 200 mM NaCl) [Cy3-DNA] = 1 nM at 20 °C.

hybridization in 8 mer–8 mer was shorter than that in 12 mer–12 mer (Figure 3A,B). In addition, the association time in dA₁₂–dT₁₂ seemed to be a mixture of short and long pulses (Figure 3C). These results suggest that the association time depends on the DNA length and that hybridization in dA₁₂–dT₁₂ could occur between DNA strands with various lengths.

Histogram Analysis of Time-Lapse Images of TIRFM.

We analyzed the movies obtained from the time-lapse images of each single-molecule hybridization to obtain the association time. According to eq 6, we plotted $\ln(\text{counts}/n)$ against association time (t_a) for 8 mer–8 mer, 12 mer–12 mer, and dA₁₂–dT₁₂ (Figure 4A–C). For comparison, each vertical axis was normalized with the maximum value of (counts/ n) of 1. We drew a straight line on the plots for 8 mer–8 mer hybridization according to the theoretical formula (eq 6) (Figure 4A). The slope corresponding to an apparent dissociation rate constant (k_{off}) was 0.013 s⁻¹. In contrast, we could not draw single straight line in the plots for 12 mer–12 mer hybridization because a bimodal distribution was found

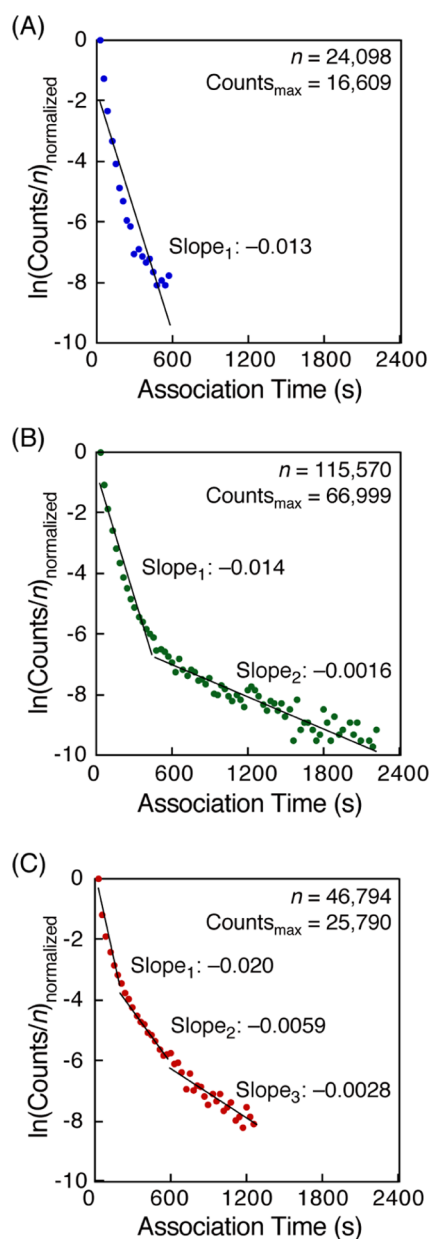


Figure 4. Histogram analyses of association time for single-molecule hybridization of (A) 8 mer–8 mer, (B) 12 mer–12 mer, and (C) dA₁₂–dT₁₂ DNAs. Each vertical axis was normalized with the maximum value of (counts/*n*) of 1. Slope values were obtained according to eq 6 in Experimental Section. Time-lapse images were obtained at an interval of 30 s. Experimental conditions: 200 mM NaCl TE buffer (10 mM Tris–HCl, pH 8.0, 1 mM EDTA, 200 mM NaCl) [Cy3–DNA] = 1 nM at 20 °C.

in the graph (Figure 4B). Therefore, we fitted the plots by two straight lines. The values of the two slopes were 0.014 s⁻¹, which was similar to that of the 8 mer–8 mer hybridization, and a smaller value of 0.0016 s⁻¹; this suggests the existence of different binding modes (fast and slow dissociation modes) in the 12 mer–12 mer hybridization. Moreover, in the case of dA₁₂–dT₁₂, the plots showed a hyperbolic curve, suggesting the existence of multiple binding modes (Figure 4C). To analyze the curve simply, we divided the plots into three areas expediently and fitted with three straight lines with slopes 0.020, 0.0059, and 0.0028 s⁻¹. The same tendency was found when hybridization was observed using different time-lapse

intervals in case of 8 mer–8 mer, 12 mer–12 mer, and dA₁₂–dT₁₂ DNA hybridization (Figures S2–S4, respectively).

Kinetic Analysis for Dissociation Rate Constants of DNA Hybridization. Irradiation with a strong laser for single-molecule detection in TIRFM results in the quenching of the fluorescent probe at a certain frequency.²⁷ Therefore, the disappearance of bright spots in the TIRFM images could be attributed to both dissociation of fluorescence-labeled molecules from probe molecules on the plate and the photo-bleaching of a fluorescent probe. To obtain the actual dissociation rate constant (*k*_{off}) from the time-lapse TIRFM movies, we performed the histogram analysis with various time-lapse intervals (τ_{tl}) under constant exposure time (τ_{ex}) for each hybridization (Figures 4, and S2–S4) and plotted the *k*_{off}· τ_{tl} values against the τ_{tl} values, according to eq 7 (Figure 5). Each

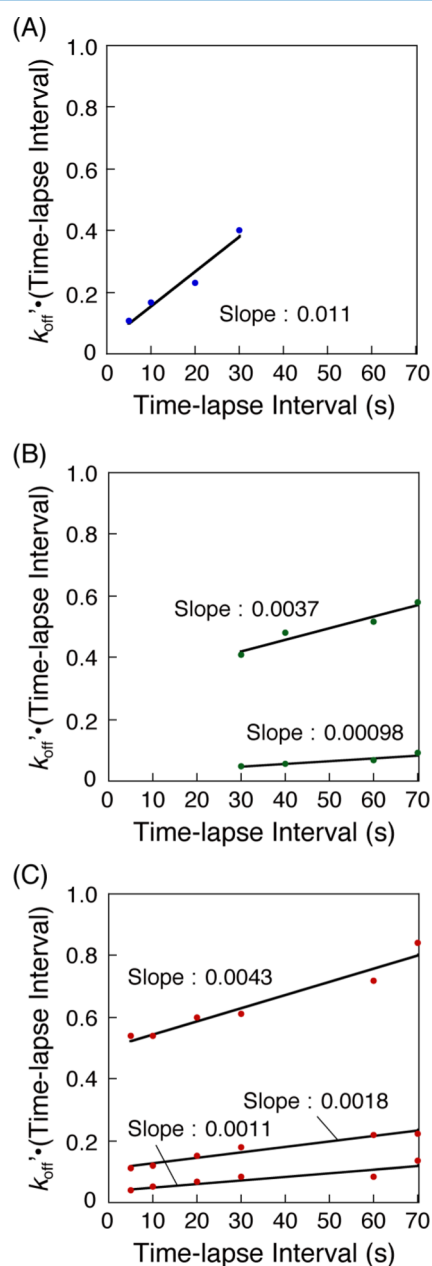


Figure 5. Time-lapse interval plots to obtain actual *k*_{off} values for (A) 8 mer–8 mer, (B) 12 mer–12 mer, and (C) dA₁₂–dT₁₂ DNA hybridization according to eq 7 in Experimental Section.

Table 1. Dissociation Rate Constants for DNA Hybridization, as Obtained by TIRFM^a

DNA	$k_{\text{off}}^{\text{fast}} (\times 10^{-3} \text{ s}^{-1})$	$k_{\text{off}}^{\text{middle}} (\times 10^{-3} \text{ s}^{-1})$	$k_{\text{off}}^{\text{slow}} (\times 10^{-3} \text{ s}^{-1})$	$k_{\text{off}}^{\text{avg}} (\times 10^{-3} \text{ s}^{-1})$
8 mer–8 mer	11.1 ± 1.7 (100%)			11.1 ± 1.7
12 mer–12 mer	3.7 ± 0.7 (90%)		1.0 ± 0.3 (10%)	1.8 ± 0.2
dA ₁₂ –dT ₁₂	4.3 ± 0.5 (52%)	1.8 ± 0.2 (41%)	1.1 ± 0.2 (7%)	2.7 ± 0.7
d(ATG) ₄ –d(CAT) ₄	8.4 ± 1.5 (86%)		0.8 ± 0.1 (14%)	1.2 ± 0.1

^aEach kinetic parameter was obtained from eqs 6–7 and 10–11 in Experimental Section and is reported with experimental errors. The percentage represents relative abundance. Experimental conditions: 10 mM Tris–HCl, pH 8.0, 1 mM EDTA, 200 mM NaCl, 20 °C.

apparent k_{off}' for the single binding mode in the 8 mer–8 mer hybridization, two binding modes in the 12 mer–12 mer hybridization, and three binding modes in the dA₁₂–dT₁₂ hybridization was obtained from each histogram and was plotted with each time-lapse interval (Figure 5).

The actual k_{off} values of each case were calculated from the linear plots as a slope of a fitted line (k_{off} for 8 mer–8 mer, $k_{\text{off}}^{\text{fast}}$ and $k_{\text{off}}^{\text{slow}}$ for 12 mer–12 mer, $k_{\text{off}}^{\text{fast}}$, $k_{\text{off}}^{\text{middle}}$, and $k_{\text{off}}^{\text{slow}}$ for dA₁₂–dT₁₂). We also calculated the average of k_{off} values ($k_{\text{off}}^{\text{avg}}$) obtained by linear fitting of the entire plots without dividing them into two or three (Table 1). Based on the results of DNA hybridization using 8 and 12 bp DNAs, the k_{off} values decreased with increasing length. This is reasonable because higher number of base pairs contributes to the stability of the hybridization, resulting in a smaller dissociation rate constant. In the case of 12 mer–12 mer hybridization, the two k_{off} values ($k_{\text{off}}^{\text{fast}}$ and $k_{\text{off}}^{\text{slow}}$) should correspond to two binding modes with different base-pair length: partial hybridization ($k_{\text{off}}^{\text{fast}}$) and perfect hybridization ($k_{\text{off}}^{\text{slow}}$). The $k_{\text{off}}^{\text{fast}}$ and $k_{\text{off}}^{\text{slow}}$ values for dA₁₂–dT₁₂ were similar to those for 12 mer–12 mer. The hybridized dA₁₂–dT₁₂ DNAs with the $k_{\text{off}}^{\text{middle}}$ value could exist in 41% of the relative abundance in the total dA₁₂–dT₁₂ hybridization. Thus, these results suggest that the interaction mode in the hybridization of dA₁₂–dT₁₂ may not only be partial and perfect hybridization ($k_{\text{off}}^{\text{fast}}$ and $k_{\text{off}}^{\text{slow}}$, respectively) but may also include other types of bindings such as a zipperlike-mismatch hybridization ($k_{\text{off}}^{\text{middle}}$) (Figure 6).

QCM Measurements of DNA Hybridization for Comparison with Those from TIRFM. A QCM biosensor allows an ensemble measurement of the binding and dissociation behaviors by detecting the total amount of substance on the sensing area. We performed a kinetic measurement for the DNA hybridization using a QCM technique with the condition same as that used in the TIRFM experiments. The time-course curves of the QCM frequency changes were based on the length of DNA and its sequence during hybridization (Figure S5A,C,E). We calculated the kinetic parameters k_{on} , k_{off} and K_{d} from these results according to eqs 10–12 (Figure S5B,D, and F, Table 2). The k_{off} values of the 8 and 12 bp DNA hybridization from QCM are similar to those from TIRFM. However, the k_{off} value for dA₁₂–dT₁₂ from QCM is larger than that from TIRFM. We found that the k_{on} and k_{off} values of dA₁₂–dT₁₂ hybridization were close to those of the 8 bp but not to 12 bp DNA hybridization. This suggests that the binding of dA₁₂–dT₁₂ occurs mainly via zipperlike hybridization, detected by the QCM device as a large mass volume. In contrast, the histogram of association time in the TIRFM experiments should be fitted with a straight line on the plots showing a longer association time (Figure 4). Thus, the kinetic parameters obtained from an ensemble measurement, such as that using QCM, provide the average number of molecules and those from single-molecule observation, such as those via TIRFM, provide the average

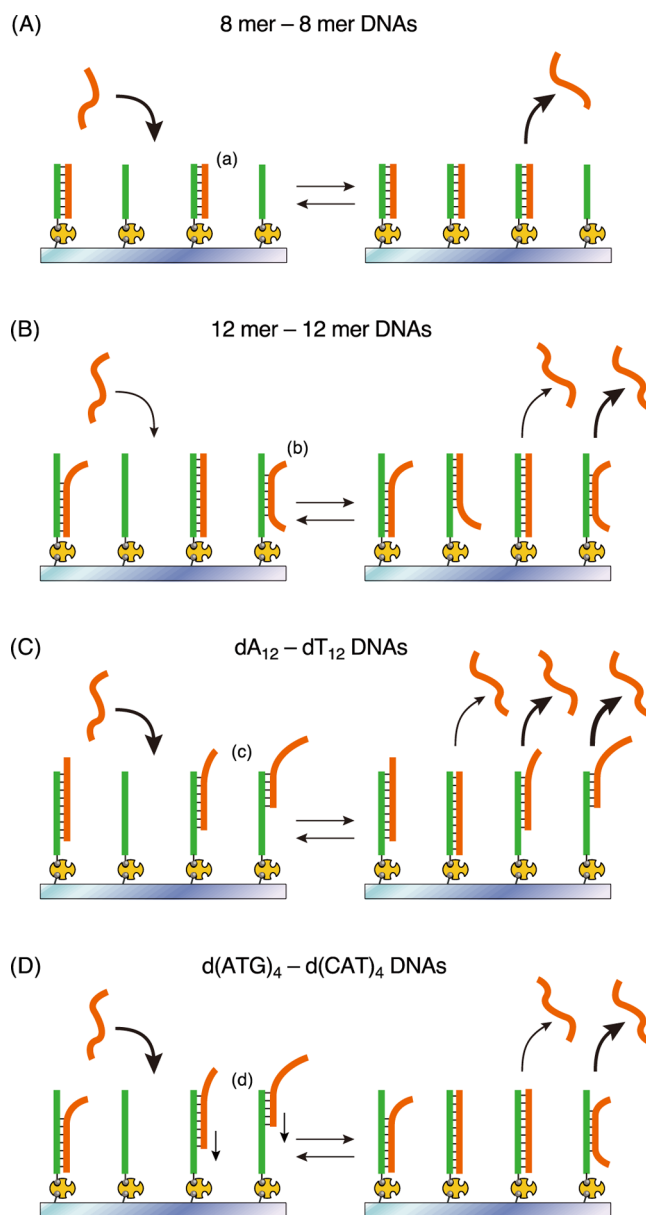


Figure 6. Schematic illustrations of multiple binding behaviors of DNA hybridization of (A) 8 mer–8 mer, (B) 12 mer–12 mer, (C) dA₁₂–dT₁₂, and (D) d(ATG)₄–d(CAT)₄ DNAs. (a) Perfect hybridization, (b) partial hybridization, (c) zipperlike-mismatch hybridization, and (d) dangling and then sliding block-like hybridization.

association time. Therefore, the difference between the kinetic parameters obtained by QCM and TIRFM could become more remarkable in case of interactions involving various binding modes. We are interested in investigating whether the multiple binding process occurring during cellular signal transduction

Table 2. Kinetic Parameters of DNA Hybridization, as Obtained by the QCM Method^a

DNA	k_{on} ($\times 10^3 \text{ M}^{-1} \text{ s}^{-1}$)	k_{off} ($\times 10^{-3} \text{ s}^{-1}$)	K_{d} (nM)
8 mer–8 mer	230 \pm 10	29 \pm 9	126 \pm 38
12 mer–12 mer	80 \pm 10	0.7 \pm 0.4	9 \pm 5
dA ₁₂ –dT ₁₂	410 \pm 20	52 \pm 15	127 \pm 37
d(ATG) ₄ –d(CAT) ₄	310 \pm 7	2.9 \pm 2.2	9 \pm 7

^aEach kinetic parameter was obtained from eqs 10–12 in Experimental Section and is reported with experimental errors. Experimental conditions: 10 mM Tris–HCl, pH 8.0, 1 mM EDTA, 200 mM NaCl, 20 °C.

depends on the average number of molecules or on the average association time.

Finally, we discussed the possibility of DNA sliding via a zipperlike-mismatch hybridization followed by perfect hybridization, which showed stability. The k_{on} value of the 12 mer DNA was smaller than that of the 8 mer DNA because of the low probability of collision required for perfect hybridization. In contrast, in dA₁₂–dT₁₂ hybridization, most of the target DNAs acted as 8 bp DNAs corresponding to a zipperlike-mismatch hybridization in the ensemble measurement of binding and dissociation behaviors (Table 2), and a few of the target DNAs acted as 12 bp corresponding to perfect hybridization with $k_{\text{off}}^{\text{slow}}$ (Table 1). We observed the DNA hybridization using d(ATG)₄ and d(CAT)₄. We expected the block-like hybridization to shift by 3 bases (Figure S5G,H). The k_{on} value in d(ATG)₄–d(CAT)₄ was close to that of 8 mer–8 mer (Table 2). Moreover, the k_{off} value was similar to that of 12 mer–12 mer. Collectively, these results suggest that target-d(CAT)₄ binds with 6 or 9 bp region via an interaction corresponding to the block-like mismatch hybridization (easy binding) and then proceeds to full-match hybridization with the same K_{d} value as that of 12 bp DNA (stable binding). In the TIRFM observations (Figures S6 and S7), bimodal linear-fitting for the plots in the histogram was obtained for d(ATG)₄–d(CAT)₄ DNAs, indicating two dissociation modes (partial and perfect hybridization) kinetically (Table 1). Thus, the dA₁₂–dT₁₂ hybridization indicates various binding/dissociation modes due to the zipperlike hybridization with a slight sliding. However, the d(ATG)₄–d(CAT)₄ hybridization showed initial dangling behavior followed by sliding and stable (partial and perfect) hybridization (Figure 6).

CONCLUSIONS

We observed hybridization of DNA with various lengths and sequences by two different methods: single-molecule observation using TIRFM and ensemble measurement using QCM. The resultant kinetic parameters obtained through TIRFM were compared with those obtained from QCM measurement. Because the hybridization of 8 mer–8 mer DNA consisted of a single binding mode, the hybridization of 12 mer–12 mer DNA showed both fast and slow dissociation rate constants. In addition, the hybridization of dA₁₂–dT₁₂ DNA showed multiple binding modes. In the case of multiple binding interactions, kinetic parameters obtained from QCM and TIRFM were different because the kinetic parameters obtained from QCM and TIRFM indicate the average number of molecules and average association time, respectively. Single-molecule detection in addition to ensemble measurements of biomolecular interactions presents more detailed information of

the binding modes, which are essential to understand biomolecules more precisely. We are interested in assessing whether the biological function depends on the average number of molecules or the average association time. It would be necessary to clarify the correlation between single-molecule kinetics and biological functions in situ.

EXPERIMENTAL SECTION

Materials. NeutrAvidin and EZ-Link NHS-PEG4-biotin were purchased from Thermo Fisher Scientific (Waltham, MA). Biotinylated DNA and fluorescent-tagged oligonucleotides were purchased from Eurofin Genomics (Tokyo, Japan). Manicure was purchased from Shiseido (Tokyo, Japan). 3-Amino-propyltriethoxysilane (APTES) was purchased from Sigma Aldrich (St. Louis, MO). Boric acid was purchased from Nacalai Tesque (Kyoto, Japan). 4-(2-Hydroxyethyl)-1-piperazineethanesulfonic acid (HEPES) and ethylenediaminetetraacetic acid (EDTA) were purchased from Dojindo (Kumamoto, Japan). A cover glass (18 mm \times 18 mm, no. 1S, 0.175 \pm 0.015 mm thickness) and a slide glass (76 mm \times 26 mm, S-0314 Neo no. 2, 1.0–1.2 mm thickness) were purchased from Matsunami Glass Industries (Osaka, Japan). 1-Ethyl-3-[3-dimethylaminopropyl]carbodiimide (EDC) was purchased from Tokyo Chemical Industry (Tokyo, Japan). *N*-Hydroxysuccinimide (NHS) was purchased from Wako Pure Chemical Industries (Osaka, Japan). All of the other reagents were purchased from Wako Pure Chemical Industries, Ltd. (Osaka, Japan), unless otherwise specified, and used without further purification. Type 1 ultrapure water (Milli-Q) was used in all of the experiments.

Cover Glass Modified with an Amino Group. To clean up a cover glass, it was sonicated in methanol for 20 min and rinsed in Milli-Q water. Then, the cover glass was immersed in 100 mM KOH, followed by sonication for 20 min. The stain-free cover glass was stored in Milli-Q water until further use. Next, the cover glass was immersed in 106 mL of mixture comprising 100 mL methanol, 5 mL acetic acid, and 1 mL APTES to modify the surface with an amino group (Figure 1B,C).

Reaction Cell Assembled with Cover Glass and Slide Glass. A slide glass with a hole (diameter: 12 mm) in the center was cleaned up by sonication in detergent solutions, acetone, and 100 mM KOH aqueous solutions for 20 min each, and the slide glass was stored in Milli-Q water until further use. In a clean booth, the slide glass was attached with the amino-group-modified cover glass by using manicure as the adhesive at the bottom around the hole to form a reaction cell (Figure 1B). The reaction cell was air-dried in a clean booth.

Immobilization of Biotinylated Probe DNA. The reaction cell equipped with the amino-group-modified cover glass at the bottom was filled with 70 μL NHS-PEG4-biotin solution (1 mg mL⁻¹) buffered with 10 mM borate NaOH (pH 8.5) for 2 h to biotinylate the amino groups on the glass surface (Figure 1C). After rinsing out with the HEPES buffer (10 mM HEPES–NaOH, pH 7.5, 100 mM NaCl), the reaction cell was filled with 70 μL HEPES buffer, followed by the addition of 1 μL of 10 mg mL⁻¹ NeutrAvidin solution and incubation at room temperature (20–25 °C) for 1 h to immobilize NeutrAvidin on the glass surface via avidin–biotin interaction. Then, the reaction cell was rinsed with 200 mM NaCl TE buffer (10 mM Tris–HCl, pH 8.0, 1 mM EDTA, 200 mM NaCl) and the reaction cell was filled with 70 μL of 200 mM NaCl TE buffer containing 10 pM 5'-biotinylated probe DNA

to immobilize the biotinylated probe DNA on the surface. After incubation for 1 h, the reaction cell was rinsed with 200 mM NaCl TE buffer and filled with 70 μ L of the same buffer. This reaction cell was used for DNA hybridization experiments.

Single-Molecule Observation of DNA Hybridization by TIRFM. A combination of an inverted fluorescence microscope IX71 (Olympus Co., Tokyo, Japan), an objective lens UApo N 100XOTIRF (Olympus Co.), a Samba 532-nm laser unit (Cobolt Inc., Solna, Sweden), an Electron Multiplier-CCD camera Imagem (HAMAMATSU Photonics K.K., Shizuoka, Japan) was used as a TIRFM system. A biotinylated probe-DNA-immobilized cover glass of the reaction cell was placed in contact with the immersion oil on the objective lens of TIRFM. The cover glass was irradiated by a laser light at 532 nm from the lower side through the objective lens, resulting in the generation of an evanescent field on the upper surface of the glass plate (Figure 2A). When the Cy3-labeled target DNAs were added to the buffer solution in the reaction cell, CCD-camera images were recorded by laser irradiation with 60.53 ms exposure time. Cy3-labeled DNAs hybridized with probe DNAs immobilized on the glass plate appeared as bright spots in the images (Figure 2B). Time-lapse images were recorded at 5, 10, 20, 30, 40, 60, and 70 s. Software MetaMorph was used for analyzing the resultant images.

TIRFM Result-Based Kinetic Analysis of DNA Hybridization. The association time (t_a) of Cy3-labeled DNAs with probe DNAs on the substrate was evaluated as continuous bright spots detected on the time-lapse images. t_a depends on the binding affinity, and the distribution of the t_a can be analyzed statistically.

In general, the dissociation reaction of double-stranded DNA (dsDNA) is expressed as eq 1.



The decrease in dsDNAs per unit time is proportional to the abundance of dsDNAs with the dissociation rate constant (k_{off}) as a proportionality constant, as shown in eq 2.

$$-\frac{d}{dt}[\text{dsDNA}] = k_{\text{off}}[\text{dsDNA}] \quad (2)$$

By dividing both sides of eq 2 by the total amount of the DNA immobilized on the substrate and using the probability P_{dsDNA} ($=[\text{dsDNA}]/[\text{total amount of the DNA immobilized on the substrate}]$), eq 3 can be derived as follows.

$$-\frac{d}{dt}P_{\text{dsDNA}} = k_{\text{off}}P_{\text{dsDNA}} \quad (3)$$

The probability P can adapt not only to space in total molecules but also to time in a single molecule from ergodicity. Thus, the probability $P_{\text{dsDNA}}(t_a)$, which implies frequency of a single dsDNA molecule hybridized with association time t_a , is described by eqs 4 and 5 obtained by integrating eq 3

$$P_{\text{dsDNA}}(t_a) = A \cdot \exp(-k_{\text{off}} \cdot t_a) \quad (4)$$

$$\ln(P_{\text{dsDNA}}) = \ln A - k_{\text{off}} \cdot t_a \quad (5)$$

where A is an integration coefficient.

In single-molecule TIRFM experiments, $P_{\text{dsDNA}}(t_a)$ can be calculated by dividing [counts] by n , where [counts] is the number of bright spots with association time t_a and n is the total number of bright spots counted. Thus, eq 5 can be replaced with eq 6

$$\ln\left[\frac{\text{counts}}{n}\right] = \ln A - k_{\text{off}}' \cdot t_a \quad (6)$$

Therefore, we can obtain the k_{off}' values from the slope of linear plots obtained by plotting $\ln(\text{counts}/n)$ against t_a from eq 6.

In single-molecule TIRFM observation, quenching of bright spots occurs not only because of dissociation of fluorescence-labeled molecules from an evanescent field but also because of the fluorescence bleaching with laser light.²⁷ Because the effect of bleaching depends on laser exposure time (τ_{ex}) and a time-lapse interval (τ_{tl}) related to the irradiation frequency, the actual dissociation rate constant (k_{off}) is related to the apparent k_{off}' obtained from eq 6 by following eq 7²⁵

$$k_{\text{off}}' \cdot \tau_{\text{tl}} = k_{\text{b}} \cdot \tau_{\text{ex}} + k_{\text{off}} \cdot \tau_{\text{tl}} \quad (7)$$

where k_{b} is the photobleaching rate constant. We calculated the actual k_{off} values using eq 7.

27 MHz QCM Measurements of DNA Hybridization for Comparison of Kinetic Parameters. A 27 MHz QCM experiment was performed to obtain the conventional kinetic parameters of DNA hybridization for comparison with the kinetic parameters obtained from the single-molecule observation in TIRFM. An AFFINIX Q4 system (Initium Co. Ltd., Tokyo, Japan) was used in this study. A sensor cell designed for the AFFINIX Q4 system has a volume of 500 μ L and is equipped with a quartz plate (60 μ m thick and 8.7 mm in diameter) with a gold electrode of 5.7 mm² area attached at the bottom of the cell. The QCM measurements were performed as mentioned previously.^{3,8} Briefly, 3,3'-dithiodipropionic acid was immobilized on a cleaned gold electrode, and carboxylic acids were activated as *N*-hydroxysuccinimide esters on the surface using EDC and NHS. NeutrAvidin was reacted with the activated esters on the QCM plate in the HEPES buffer. The biotinylated DNA was immobilized on the NeutrAvidin-immobilized QCM in the 200 mM NaCl TE buffer and frequency changes (ΔF_{water}) were monitored over time. When the ΔF_{water} value corresponding to an amount of biotinylated DNA reached 100 Hz (a predetermined value), 10 μ M free biotin was added to regulate the immobilization amount. After rinsing the biotinylated DNA-immobilized QCM cell with 200 mM NaCl TE buffer, frequency changes in QCM in response to the addition of complementary DNA strands were measured over time.

Binding kinetics were analyzed using the time courses of frequency decreases (mass increases), as described previously.^{3,8} Briefly, the binding mode of the hybridization is given by eq 8.



The amount of dsDNAs formed at time t after the injection is described by eqs 9–12.

$$[\text{dsDNA}]_t = [\text{dsDNA}]_{\infty} (1 - e^{-1/\tau t}) \quad (9)$$

$$\Delta m_t = \Delta m_{\infty} (1 - e^{-1/\tau t}) \quad (10)$$

$$\frac{1}{\tau} = k_{\text{on}}[\text{target-DNA}]_0 + k_{\text{off}} \quad (11)$$

$$K_{\text{d}} = \frac{k_{\text{off}}}{k_{\text{on}}} \quad (12)$$

The binding relaxation time (τ) can be calculated from the time course of Δm in each concentration by a curve-fitting method based on eq 10. The k_{on} and k_{off} can be obtained from the slope and intercept of the linear correlation of the reciprocal plot of binding relaxation time ($1/\tau$) against each concentration, using eq 11 (Figure S5). K_{d} values were calculated from the obtained k_{on} and k_{off} values, using eq 12.

■ ASSOCIATED CONTENT

🔍 Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsomega.8b00135.

CCD-camera images for DNA hybridization of Cy3-labeled DNAs. Histogram analyses of association time for 8 mer–8 mer, 12 mer–12 mer, dA_{12} – dT_{12} , and $d(ATG)_4$ – $d(CAT)_4$ DNAs; typical time courses of frequency of a QCM plate and kinetic analysis of DNA hybridization; TIRFM observation for the $d(ATG)_4$ – $d(CAT)_4$ hybridization (PDF)

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Notes

The authors declare no competing financial interest.

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