ORIGINAL ARTICLE



# Inhibitory effect of chemical and natural anti-browning agents on polyphenol oxidase from ginger (Zingiber officinale Roscoe)

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Abstract Food browning is undesirable as it causes deterioration in food quality and appearance. This phenomenon was related to polyphenol oxidase (PPO), which catalyzes conversion of phenolic compounds into  $o$ -quinones. The present work evaluated the use of chemical and natural anti-browning agents to prevent the browning of ginger PPO. Sodium metabisulfite (5 mM) is a better chemical inhibitor compared to L-cysteine and sodium chloride as 55.00% of ginger PPO inhibition was achieved. The percentage of inhibition increased as the concentration of antibrowning agents increases. The addition of heated onion, chili pepper and pineapple extracts exhibited a stronger inhibitory effect on ginger PPO than unheated extracts. Heated chili pepper extract was the best natural inhibitor found in this study and it inhibited the ginger PPO (47.97%) mixed-competitively. Natural anti-browning agents have potential to be used to control the browning of ginger as well as other vegetables and fruits.

Keywords Polyphenol oxidase - Ginger - Anti-browning agents - Inhibition

# Introduction

Ginger, scientifically named as Zingiber officinale Roscoe, belongs to the Zingiberaceae family which is a large group consisting of many well-known perennial herbs (Singh

 $\boxtimes$  Chen Wai Wong wongcw@ucsiuniversity.edu.my et al. [2012](#page-6-0)). The plant is grown mainly for its rhizome which was widely used for cosmetic, flavouring, medicinal and ornamental purpose. It was found that ginger exhibits antioxidant, antimicrobial, antiemetic anti-tumour and antiinflammatory properties. It also helps in enhancing gastrointestinal motility, treating digestion problem, reducing cholesterol and reducing nausea (Ramachandran [2008](#page-6-0)). Therefore, ginger was widely used as a culinary spice and herbal remedy due to its high nutritional and medicinal value.

In food industry, food quality assurance during food processing and storage was critical. Food browning is currently a major concern in food processing and storage sector. It is defined as the phenomenon where brownish black pigment formed on the surface of food under chemical reactions that occurred within (Corzo-Martinez et al. [2012\)](#page-5-0). Normally, browning reaction occurs in fruits, vegetables, seafoods and cereals. Browning effect can deteriorate the marketability, food nutritional value, appearance and consumers' acceptance on the food, causing reduction in shelf life of food and affect economic situation of a country.

Polyphenol oxidase, PPO (EC 1.14.18.1) is a coppercontaining oxidoreductase which is universally found in plants (Othman [2014](#page-6-0)). It is also known as tyrosinase, catechol oxidase, and monooxygenase. It catalyses the hydroxylation of monophenol to diphenol and subsequently oxidation of diphenol into quinone compound, which will then polymerize and form dark pigment melanin (Araji et al. [2014](#page-5-0)). Polyphenol oxidase is a plastid-localized enzyme which normally being separated from its substrate, phenolic compounds by cell compartments (Escobar et al. [2008](#page-5-0)). Mechanical injuries and tissue damage such as slicing, peeling and cutting on plant cells cause the

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interaction between PPO and phenolic compounds, thus browning reaction occur.

Control of enzymatic browning was essential as to maintain food quality and to ingratiate consumers' needs. Food industry are currently trying to eliminate the usage of chemical in food due to the awareness of the side effects that may brought by the chemicals. The PPO inhibitors occurring in natural resources have been studied in several plants such as potato (Lee et al. [2002](#page-5-0)), pear (Kim et al. [2005\)](#page-5-0) and banana (Lee [2007](#page-5-0)). However, research on the natural anti-browning agents on ginger is yet to be established. There is limited scientific information on how the natural inhibitors may influence the PPO activity for ginger. Thus, the inhibitory effects of natural inhibitors (honey, onion, chili pepper and pineapple extracts) on ginger PPO were examined in this study.

# Materials and methods

# Plant material and natural inhibitors

Ginger (Zingiber officinale Roscoe) originated from Bentong, Pahang, Malaysia was used in this study. It was purchased from a local morning market (Pandan Perdana, Kuala Lumpur). Red onion (Allium cepa) and chili pepper (Capsicum annuum), were also bought from the local morning market (Pandan Perdana, Kuala Lumpur). Red onion and chili pepper were originated from China and Malaysia, respectively. Honey (Capilano Manuka honey Australia) and pineapple (Ananas comosus) from Malaysia were bought from Giant supermarket (Taman Connaught, Cheras, Kuala Lumpur). All chemicals used in this study were of analytical grade and were used as obtained.

# PPO extraction

The extraction method was adapted from Lin et al. ([2016\)](#page-5-0) with slight modification. 100 g of ginger was first washed, peeled and cut into small pieces. The ginger was then homogenized with 2 g of Polyvinylpolypyrrolidone (Sigma Aldrich, St. Louis, MO, USA) in 200 mL of pre-chilled  $(4 °C)$  0.1 M phosphate buffer at pH 6.8 under 22,000 rpm for 1 min using a LB-8011ES industrial blender (Waring Laboratory, Torrington, CA, USA). The homogenate was centrifuged at 7000 rpm for 20 min at  $4 °C$  (Centrifuge Universal 320R, Hettich Zentrifugen, Tuttlingen, Germany). The supernatant obtained was filtered under vacuum with Buchner filtering system (WP6211560 Vacuumpressure pump, Millipore Sigma, Burlington, MA, USA). The filtrate was used as the crude PPO extract.

#### PPO assay

PPO activities were assayed with 4-methylcatechol and pyrocatechol at 410 nm as the substrates by using spectrophotometric method with slight modifications (Joseph et al. [2012](#page-5-0)). The reaction mixture included 0.1 mL enzyme solution, 1.0 mL of 0.1 M substrate solution in 1.9 mL 0.1 M phosphate buffer at pH 6.8. Then, the mixture was transferred into a cuvette and absorbance readings were measured at 15 s intervals for 5 min using a spectrophotometer (Secomam UviLine 9400, Champigny-sur-Marne, France) at room temperature. The blank consisted of 1 mL substrate solution in 2 mL of phosphate buffer. The initial velocity was calculated from the slope of the absorbance value against time curve. One unit (U) of PPO activity was defined as amount of enzyme that caused a 0.001 absorbance change per min (Kim and Kim [2013\)](#page-5-0).

#### Effect of anti-browning agents

Chemical inhibitors used were L-cysteine, sodium chloride and sodium metabisulfite. PPO activities were assayed with inhibitor at three different concentrations (1, 3 and 5 mM) spectrophotometrically for 4-methylcatechol and pyrocatechol, respectively. Reaction mixture included 0.1 mL crude PPO, 1.9 mL substrate in 0.1 M buffer at pH 6.8, and 1 mL inhibitor. The percentage inhibition (%) values for each inhibitors were calculated using the PPO activity without inhibitor as the initial PPO activity.  $I_{50}$  values, which is the concentration of inhibitor needed to show 50% inhibitory effect on enzyme, was obtained from the percentage inhibition. Michaelis–Menten constant  $(K<sub>m</sub>)$ , maximum velocity ( $V_{\text{max}}$ ), dissociation constant ( $K_i$ ) and type of inhibition for each inhibitor were also determined.

## Effect of natural inhibitors

Honey, red onion extract, red chili pepper extract and pineapple juice were used as natural inhibitors. Honey was tested with three different concentrations, which were 1, 3 and 5% (v/v) (Gacche et al. [2009](#page-5-0)). Onion, chili pepper and pineapple extracts were used freshly and after heating with a concentration of 10 mg/mL.

Extraction of onion, chili pepper and pineapple were carried out according to the method of Yapi et al. ([2015\)](#page-6-0) with slight modifications. 90 g of each natural inhibitor was first homogenized with 90 mL pre-chilled  $(4 \degree C)$  distilled water at maximum speed (22,000 rpm) for 1 min. The homogenate was filtered through cloth. The filtrate was centrifuged at 7000 rpm for 20 min at  $4^{\circ}$ C. The supernatant obtained was filtered through Buchner funnel with filter paper and the filtrate was used as the fresh inhibitor. Heated inhibitors were prepared by incubating the fresh extracts of inhibitors at 95  $\degree$ C for 15 min. The inhibitors were subjected to centrifugation at 10,000 rpm for 60 s if there is precipitation. 50  $\mu$ L of heated inhibitors were added into 4.95 mL distilled water to prepare 5 mL of 10 mg/mL inhibitors.

PPO activity was assayed and expressed as percentage inhibition (%). Same as the study of inhibitory effect of anti-browning agents, Michaelis–Menten constant  $(K_m)$ , maximum velocity ( $V_{\text{max}}$ ) and dissociation constant ( $K_i$ ) were also determined.

# Statistical analysis

Statistical analyses of all assays were performed with Microsoft Office Excel. All experiment was performed with triplicate measurements. The data collected were presented as mean  $\pm$  standard deviation (SD) and also as percentage relative activity.

# Results and discussion

# PPO extraction

Enzymes are pH dependent as changes in pH may cause impairment of enzyme-catalyzed biochemical reactions (Buchholz et al. [2012\)](#page-5-0). Buffer, which can maintain pH of a solution at or near to certain value, was used in the extraction method to maintain a suitable pH for PPO. As referred to Chikezie et al. [\(2013](#page-5-0)), most PPO extracted from food source were found to have maximum activity near neutral pH value, so phosphate buffer of pH 6.8 was used as extraction buffer in this study. Enzymes are also temperature dependent. Hence, pre-chilled Waring blender and phosphate buffer at  $4^{\circ}$ C were used in the extraction process to maintain temperature as heat may be produced during homogenization process. All studies of PPO were also carried out at  $4^{\circ}$ C to avoid denaturation of enzyme.

Polyvinylpolypyrrolidone (PVPP) was added during homogenization process into the buffer solution. PVPP is a protective agent which it can bind to phenolic compounds, avoid the browning reaction between PPO and its substrates, phenolic compounds (Rojas-Grau et al. [2008](#page-6-0)).

### PPO activity assay

Ginger PPO activities determined in this study were 13,450 EU/min/mL for 4-methylcatechol and 9040 EU/ min/mL for pyrocatechol, respectively. This indicates that ginger PPO exhibits a higher catalytic reaction towards 4-methylcatechol.

#### Effect of anti-browning agents

Table [1](#page-3-0) shows the inhibitory effect of various antibrowning agents on ginger PPO with 4-methylcatechol and pyrocatechol as substrates. It can be seen that the inhibition percentage increased as concentration of inhibitors increased from 1.0 to 5.0 mM. Sodium metabisulfite shows the greatest effect in inhibiting ginger PPO as it exhibited 46.36% inhibition at 1.0 mM. Furthermore, sodium metabisulfite has the smallest  $I_{50}$  and  $K_i$  value as compared to sodium chloride and L-cysteine, which signifies its strongest binding affinity to enzymes (Sharma [2012](#page-6-0)). Lcysteine and sodium chloride might not be good inhibitors for ginger PPO as low inhibition percentage was obtained (6.63–28.18%). These further prove that sodium metabisulfite has greater efficiency in inhibiting ginger PPO.

From Table [1](#page-3-0), Ginger PPO with sodium chloride and Lcysteine exhibited the lowest  $K_m$  value by using 4-methylcatechol and pyrocatechol, respectively. V<sub>max</sub> values were decreased from the initial  $V_{\text{max}}$  value of control for all anti-browning agents used which indicates the number of substrates used by PPO to produce products per unit time were reduced (Egelhofer [2016](#page-5-0)).

L-cysteine is a thiol containing compound which it can react with copper in active site of PPO, inactivate catalytic ability of the enzyme (Saeidian [2016](#page-6-0)). L-cysteine also able to inhibit browning effect by interacting with quinone compound, make it unable to undergo polymerization and produce melanin pigment (Gacche et al. [2006](#page-5-0)). In this study, mixed-type inhibition and un-competitive-type inhibition of L-cysteine on ginger PPO using 4-methylcatechol and pyrocatechol, respectively were observed (Table [1\)](#page-3-0). Inhibition of polyphenol oxidase activity by Lcysteine has been reported in browning control of acanthus PPO (Saeidian [2016\)](#page-6-0) and wild mushroom PPO (Dedeoglu and Guler [2009\)](#page-5-0).

Sodium chloride was sodium salt, which act as agent of firmness to inhibit browning effects. It is reported by Ayala-Zavala and Gonzalez-Aguilar ([2010\)](#page-5-0) that chloride was a weak inhibitor. Sodium chloride was found to be uncompetitive inhibitors for ginger PPO (Table [1\)](#page-3-0). This result was similar to those reported by Yong and Hye ([2007\)](#page-6-0) on potato PPO, which stated that NaCl may interfere with enzyme–substrate complex and cause inactivation of PPO.

Sodium metabisulfite  $(Na_2S_2O_5)$  is a type of sulfites which is reported to be highly effective in control of browning effect. Sulfites are reducing agent which it can react with *o*-quinone and form diphenols which can help to prevent the polymerization of  $o$ -quinone into coloured pigment (Kilic-Akyilmaz and Gulsunoglu [2015\)](#page-5-0). In this study, sodium metabisulfite non-competitively inhibited ginger PPO using both 4-methylcatechol and pyrocatechol

Substrates	Inhibitors	$\left[\Pi\right]$ (mM)	$I_{50}$ (mM)	Inhibition $(\%)$	$V_{max}$ (EU/min/ $mL$ )	$K_{m}$ (mM)	$K_i$ (mM)	Type of inhibition
4-Methylcatechol	Control				12,903.23	7.37		
	L-cysteine	1	8.82	12.27	10,752.69	7.53	5.00/4.42	Mixed
		3		20.90	10,204.08	7.85	11.34/8.65	
		5		28.18	9803.92	7.89	15.82/ 12.23	
	Sodium chloride	1	8.74	10.91	11,764.71	6.84	10.33	Un-competitive
		3		20.91	10,000.00	6.12	10.33	
		5		28.18	9259.26	5.56	12.70	
	Sodium metabisulfite	1	3.62	46.36	6896.55	7.16	1.14	Non-competitive
		3		50.00	6666.67	7.08	3.21	
		5		55.00	5970.15	7.16	4.31	
Pyrocatechol	Control				11,299.44	4.24		
	L-cysteine	1	10.44	6.63	10,416.67	3.75	11.80	Un-competitive
		3		16.84	9569.38	3.32	16.59	
		5		23.47	8771.93	3.07	17.35	
	Sodium chloride	$\mathbf{1}$	10.68	11.22	10,000.00	4.00	7.70	Un-competitive
		3		18.37	9478.67	3.63	15.62	
		5		23.47	8403.36	3.36	14.51	
	Sodium metabisulfite	1	10.28	16.84	9345.79	4.21	4.78	Non-competitive
		3		21.94	8771.93	4.21	10.42	
		5		24.49	8196.72	4.22	13.21	

<span id="page-3-0"></span>Table 1 Effect of chemical inhibitors on Zingiber officinale R. PPO

as substrates (Table 1). Inhibitory effect of sodium metabisulfite has been studied in the browning control of grape PPO (Sanchez-Ferrer and Garcia-Carmona [2010](#page-6-0)) and apple PPO (Cortez-Vega et al. [2008](#page-5-0)).

# Effect of honey on PPO activity of ginger

Inhibition parameters of honey on ginger PPO were reported in Table 2. 5% (w/v) honey exhibited a lower percentage of inhibition for 4-methylcatechol (1.4%) than that of pyrocatechol (42.57%). This indicates that honey exhibits a greater inhibition towards ginger PPO using pyrocatechol compared to 4-methylcatechol. There was an increase in the inhibition percentage with the increase of honey concentrations for both substrates used.

The  $V_{max}$  value was unchanged and  $K_m$  value was increased for PPO with honey using 4-methylcatechol. It can be concluded that honey was a competitive inhibitor (Table 2) and able to diminish the rate of catalysis by reducing the proportion of enzymes molecules bound to a substrate (Wong and Lee [2014\)](#page-6-0). However, honey exhibited non-competitively inhibition towards ginger PPO when

Table 2 Effect of honey which used as natural inhibitor on Zingiber officinale R. PPO activity

Substrates	<b>Inhibitors</b>	Concentration $(\%)(w/v)$	Inhibition $(\%)$	$V_{\text{max}}$ (EU/min/mL)	$K_m$ (mM)	Type of inhibition
4-Methylcatechol	Control	-		8547.01	4.53	-
	Honey		1.40	8264.46	5.69	Competitive
		3	2.10	8000.00	6.2	
		5	2.10	8000.00	7.2	
Pyrocatechol	Control	-	-	8333.33	4.39	-
	Honey		28.38	5847.95	4.37	Non-competitive
		3	32.43	5633.80	4.39	
		5	42.57	4938.27	4.42	

<span id="page-4-0"></span>pyrocatechol was used as the substrate, whereby the  $K<sub>m</sub>$ value was unchanged (Table [2](#page-3-0)). Honey did not compete with substrate for enzyme but bound with enzyme at a site other than active site (Engelking [2015\)](#page-5-0).

Mechanism of honey in inhibiting PPO varies depends on the variety of honey used, source of PPO and substrates used (Lozano [2006](#page-5-0)). Haard and Simpson [\(2000\)](#page-5-0) have suggested that honey acts as chelating agent to inhibit PPO. A small peptide with molecular weight of 600 found in honey may chelate copper of PPO active site thus inactivate the enzyme. Moreover, high amount of components such as phenolic acids, flavonoids and vitamins found in honey has implied its antioxidant properties (Lee [2013\)](#page-5-0).

# Effect of onion, chili pepper and pineapple extracts on PPO activity of ginger

Table 3 shows the inhibitory effect of onion, chili and pineapple extracts on PPO activity of ginger. All heated extracts exhibited stronger inhibition percentage than nonheated extracts for both 4-methylcatechol and pyrocatechol. It can be concluded that heated extracts could have reduced the capability of the binding of enzyme to substrates greater than the fresh extracts. This could probably due to the occurrence of non-enzymatic browning, Maillard reaction in the extracts. Maillard reaction products (MRPs) were found to exert great antioxidant properties which can control browning reaction in food (Phisut and Jiraporn [2013\)](#page-6-0). Shaimaa et al. ([2016\)](#page-6-0) has also reported that heating process has increased flavonoids and phenolics content in chili pepper extract as well as increased in its antioxidant capability, which would cause an increase in inhibition percentage as well. Similar results have been found by

Wong and Lee [\(2014](#page-6-0)) on cassava leaves PPO and Kim et al. [\(2007](#page-5-0)) on peach juice PPO whereby heated onion extracts inhibited PPO activity markedly.

According to Table 3, chili pepper extract shows the greatest inhibition percentage (22.90–47.97%) on ginger PPO for both 4-methylcatechol and pyrocatechol as compared to onion and pineapple extracts. This was in agreement with those reported by Mercimek et al. [\(2015](#page-6-0)) in which 44% inhibition on potato PPO was obtained with chili pepper extract. Chili pepper extract was an un-competitive and mixed inhibitor for ginger by using 4-methylcatechol and pyrocatechol, respectively (Table 3). A large amount of phenolic and flavonoid compounds as well as high ascorbic acid content were found to be the causes of antioxidant properties in chili pepper, which inhibit the formation of undesired brown pigment (Shaimaa et al. [2016\)](#page-6-0).

Ginger PPO inhibited by chili pepper extract exhibited the lowest  $K_m$  and  $V_{max}$  values for both non-heated and heated extracts by using 4-methylcatechol. This suggesting a stronger affinity of PPO towards 4-methylcatechol, which prevent the dissociation of enzyme–substrate complex. In contrast, chili pepper extract inhibited ginger PPO exhibited the highest  $K_m$  and lowest  $V_{max}$  value by using pyrocatechol for both non-heated and heated extracts. The reduction in binding affinity of ginger PPO and pyrocatechol was occurred in the presence of chili pepper extract.

From Table 3, it was found that both heated and nonheated onion extract exhibited similar type of inhibition, which was mixed-type inhibition. However, onion was found to competitively inhibited yam PPO (Yapi et al. [2015](#page-6-0)) and non-competitively inhibited cassava leaves PPO (Wong and Lee [2014](#page-6-0)). Onion possesses sulfhydryl or thiol

<b>Substrates</b>	<b>Inhibitors</b>	Heated/non-heated	Inhibition $(\%)$	$V_{max}$ (EU/min/mL)	$K_{m}$ (mM)	Type of inhibition
4-Methylcatechol	Control			12,658.23	6.35	
	Onion	Non-heated	14.02	11,494.25	6.80	Mixed
		Heated	15.89	11,111.11	6.90	
	Chili	Non-heated	22.90	9615.38	4.44	Un-competitive
		Heated	28.50	9009.01	4.12	
	Pineapple	Non-heated	17.76	10,000	6.35	Non-competitive
		Heated	21.50	9259.26	6.29	
Pyrocatechol	Control	$\overline{\phantom{m}}$	-	8928.27	4.39	$\qquad \qquad -$
	Onion	Non-heated	28.38	6250	4.59	Mixed
		Heated	33.11	5882.35	4.62	
	Chili	Non-heated	40.54	5128.21	5.95	Mixed
		Heated	47.97	4597.70	6.10	
	Pineapple	Non-heated	25.00	6389.78	4.31	Non-competitive
		Heated	27.03	5882.35	4.37	

Table 3 Effect of 10 mg/mL onion, chili and pineapple extract which act as natural inhibitor to Zingiber officinale R. PPO activity

<span id="page-5-0"></span>(–SH) group which able to bind copper at PPO active site, interfere with catalytic activity of PPO. Flavonoid compound and thiol groups found in onion have identified its antioxidant ability which can also help to control browning reaction (Akhtar 2015).

Non-competitive inhibition of pineapple juice on ginger PPO was obtained for both substrates used as shown in Table [3](#page-4-0). Unchanged  $K_m$  values and reduced  $V_{max}$  values indicates the non-competitive inhibitors did not reacted with the enzyme at the active site. Pineapple contains ascorbic acid, reviewing the antioxidant properties of pineapple which can control PPO activity (Antoniolli et al. 2012). Chaisakdanugull et al. (2007) has suggested that main organic acids like malic and citric acids also played an important role in inhibiting banana puree PPO. Cysteine and other sulfhydryl compounds present in pineapple extract also contribute to PPO inhibition. The effectiveness of pineapple juice in controlling browning reaction has been tested on rose apple fruit PPO by Supapvanich et al. [\(2012](#page-6-0)).

## Conclusion

This study concluded that heated chili pepper extract was the best natural inhibitor to inhibit ginger PPO. Sodium metabisulfite showed a stronger inhibition on ginger PPO in this study. However, the use of natural inhibitors such as honey, onion, chili pepper and pineapple extract can be used to replace the sulphite-containing agents which could be harmful to human health.

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