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# Shift towards larger diatoms in a natural phytoplankton assemblage under combined high-CO<sub>2</sub> and warming conditions

#### SCARLETT SETT<sup>1,\*</sup>, KAI G. SCHULZ<sup>2</sup>, LENNART T. BACH<sup>1</sup> AND ULF RIEBESELL<sup>1</sup>

<sup>1</sup>GEOMAR HELMHOLTZ CENTRE FOR OCEAN RESEARCH KIEL, DÜSTERNBROOKER WEG 20, 24105 KIEL, GERMANY AND <sup>2</sup>CENTRE FOR COASTAL BIOGEOCHEMISTRY, SCHOOL OF ENVIRONMENT, SCIENCE AND ENGINEERING, SOUTHERN CROSS UNIVERSITY, P.O. BOX 157, LISMORE, NSW 2480, AUSTRALIA

\*CORRESPONDING AUTHOR: ssett@ifam.uni-kiel.de

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An indoor mesocosm experiment was carried out to investigate the combined effects of ocean acidification and warming on the species composition and biogeochemical element cycling during a winter/spring bloom with a natural phytoplankton assemblage from the Kiel fjord, Germany. The experimental setup consisted of a "Control" (ambient temperature of ~4.8 °C and ~535  $\pm$  25 µatm  $\rho$ CO<sub>2</sub>), a "High-CO<sub>2</sub>" treatment (ambient temperature and initially 1020  $\pm$  45 µatm  $\rho$ CO<sub>2</sub>) and a "Greenhouse" treatment (~8.5 °C and initially 990  $\pm$  60 µatm  $\rho$ CO<sub>2</sub>). Nutrient replete conditions prevailed at the beginning of the experiment and light was provided at *in situ* levels upon reaching  $\rho$ CO<sub>2</sub> target levels. A diatom-dominated bloom developed in all treatments with *Skeletonema costatum* as the dominant species but with an increased abundance and biomass contribution of larger diatom species in the Greenhouse treatment. Conditions in the Greenhouse treatment accelerated bloom development with faster utilization of inorganic nutrients and an earlier peak in phytoplankton biomass compared to the Control and High CO<sub>2</sub> but no difference in maximum concentration of particulate organic matter (POM) between treatments. Loss of POM in the Greenhouse treatment, however, was twice as high as in the Control and High CO<sub>2</sub> treatment at the end of the experiment, most likely due to an increased proportion of larger diatom species in that treatment. We hypothesize that the combination of warming and acidification can induce shifts in diatom species composition with potential feedbacks on biogeochemical element cycling.

KEYWORDS: ocean acidification; warming; diatoms; spring bloom; mesocosms

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## INTRODUCTION

Climate change, driven primarily by anthropogenic utilization of fossil fuels over the past 250 years, is changing sea surface temperatures (i.e. ocean warming) and seawater carbonate chemistry (i.e. ocean acidification). The IPCC 2014 projects an additional increase in sea surface temperature of 1–6 °C and a decrease in surface pH of ~0.3 units by the end of this century (Field *et al.*, 2014). Both of these environmental drivers are expected to affect marine primary producers (Riebesell and Tortell, 2011; Boyd and Brown, 2015) with potential consequences for marine biogeochemical element cycling in the future ocean (Riebesell *et al.*, 2009; Gehlen *et al.*, 2011; Rees, 2012).

Indirect effects of warming are expected through enhanced stratification of the water column and consequent alteration of nutrient supply and light availability, potentially affecting growth at the base of the marine food web and modulating energy transfer to higher trophic levels (Chust et al., 2014). Below optimum temperatures for maximum metabolic rates (Boyd et al., 2013), one of the direct effects of warming is the enhancement of metabolic activities (Eppley, 1972) which is considered to be more pronounced in heterotrophs than autotrophs (Pomeroy and Wiebe, 2001). Numerous laboratory and mesocosm-based experiments have been carried out to examine the effects of warming on natural plankton communities and some of the most common and recurrent patterns observed were increased primary production (PP) (Lewandowska et al., 2012), shifts in species composition (Sommer et al., 2007; Lassen et al., 2010), shifts in the partitioning of organic matter into dissolved and particulate pools (Wohlers et al., 2009) and decreased biomass build-up (Keller et al., 1999; Sommer and Lengfellner, 2008a; O'Connor et al., 2009; Lassen et al., 2010; Wohlers-Zöllner et al., 2012; Biermann et al., 2014) with one exception (Taucher et al., 2012). Taucher et al. (2012) suggested that the enhanced biomass build-up with warming during their experiments, contradictory to previous results, was most likely due to differences in the phytoplankton community composition.

In addition to warming, ocean acidification is also expected to affect marine ecosystems in the upcoming decades (Riebesell and Tortell, 2011). While studies looking at the effects of ocean acidification on natural phytoplankton communities have shown that there can be a fertilizing effect of increasing CO<sub>2</sub> on PP (Hein and Sand-Jensen, 1997; Schippers *et al.*, 2004; Tortell *et al.*, 2008; Egge *et al.*, 2009; Engel *et al.*, 2013) others pointed towards a negative effect of rising proton (H<sup>+</sup>) concentration particularly on calcifying phytoplankton (Kottmeier *et al.*, 2014; Bach, 2015). At the community level, shifts in species composition were observed in some studies (Tortell *et al.*, 2002, 2008) with higher CO<sub>2</sub> concentrations benefiting diatoms (Tortell *et al.*, 2008; Feng *et al.*, 2009; Yoshimura *et al.*, 2013). Most importantly, shifts in phytoplankton composition could affect the partitioning of organic carbon into the dissolved and particulate pools (Engel *et al.*, 2005; Riebesell *et al.*, 2007; Bellerby *et al.*, 2008; Kim *et al.*, 2011) which could have consequences for the export of organic carbon pump in the future. Ocean acidification and warming are two processes expected to occur simultaneously and thus their effects on natural phytoplankton communities should ideally be studied in combination (Hare *et al.*, 2007; Feng *et al.*, 2009; Kim *et al.*, 2011; Paul *et al.*, 2015; Riebesell and Gattuso, 2015).

By means of an indoor mesocosm experiment, we were able to follow the temporal development of a spring bloom containing a natural phytoplankton assemblage of a coastal eutrophic area in the Kiel Bight under control, ocean acidification (i.e. "High  $CO_2$ ") and combined acidification and warming (i.e. "Greenhouse") conditions. The increases in temperature and  $pCO_2$  applied in our experiment are within the range expected by the end of this century as projected by the IPCC 2014 (Field *et al.*, 2014).

#### MATERIALS AND METHODS

#### **Experimental setup**

The indoor mesocosm experiment was carried out between 12 January and 24 February 2012 at the GEOMAR Helmholtz Centre for Ocean Research Kiel. Six mesocosms (see end of section for details) with a volume of 1400 L each (1.5 m diameter, 1 m depth) were set up in temperature controlled rooms and filled simultaneously with seawater from Kiel Bight containing a representative late winter/early spring plankton community of bacteria, phytoplankton and microzooplankton. Although mesozooplankton was present in the water column at this time of year, most of it was probably trapped on the fine-sand filter collecting large detritus particles before filling the mesocosms. Dissolved inorganic nutrients at the beginning of the experiment were naturally available and therefore not manipulated. Each mesocosm had a computer-controlled light system equipped with full spectrum light bulbs (10 solar tropic T5 Ultra, 4000 K, 2 solar nature T5 Ultra, 9000 K, JBL) covering the full natural spectrum of photosynthetically active radiation (PAR: 400-700 nm). The light programme allowed for simulation of a daily triangular light curve with maximum light intensity (170–200  $\mu$ mol photons m<sup>-2</sup> s<sup>-1</sup> at the surface) around noon (Brock, 1981) and a daily light dose increasing over time. Light was provided at conditions representing the mean of a mixed water column of ~10 m depth. Each mesocosm was equipped with a slowly rotating propeller which ensured gentle mixing throughout the experiment, allowing small particles to remain in suspension and bigger particles to sink through the water column (Sommer et al., 2007). Mesocosms were set up in duplicates with the following treatments: "Control" (initial in situ temperature of  $4.8 \pm 0.3$  °C and initial in situ  $pCO_2$  of 535  $\pm$  25 µatm), "High CO<sub>2</sub>" treatment (initial temperature of 5.4  $\pm$  0.3 °C and initially  $1020 \pm 45 \,\mu \text{atm} \, p \text{CO}_2$ ) and "Greenhouse" treatment  $(8.5 \pm 0.4 \,^{\circ}\text{C}$  and initially  $990 \pm 60 \,\mu\text{atm} \,p\text{CO}_2)$ . Elevated temperature and CO<sub>2</sub> scenarios were chosen according to IPCC 2014 projections for the end of this century (Field et al., 2014).

Temperature and salinity were measured every sampling day with a combined conductivity and temperature probe (WTW Germany). Samples for biogeochemical parameters and phytoplankton analysis were taken daily at 9 a.m. during the bloom period (Days 12–20) and every other day otherwise. Samples were taken from an intermediate depth (0.5 m below surface) either with a membrane pump directly into measuring vials or with a silicone tube by gravity into sample canisters. The canisters were gently mixed before any subsampling took place to ensure representative homogenous samples.

Low replication in the experimental setup (i.e. duplicates instead of triplicates per treatment) prevented us from conducting statistical analysis on our data. The low replication stemmed from unforeseen technical issues (i.e. propellers and light system failing) that, although immediately addressed during the experiment, could not be solved. The unbalanced design of the experiment (i.e. lack of a warming alone treatment) was due to logistical constraints of the available infrastructure. In view of this, we have applied a rather conservative approach in our data interpretation, remaining mostly descriptive and pointing out treatment effects only in cases where differences among treatments are rather obvious (i.e. no overlapping standard deviation). Nonetheless, we still believe our results provide valuable information about changes in community composition due to changing environmental conditions.

## Temperature and CO<sub>2</sub> manipulations

Target temperature for the Greenhouse treatment (i.e. +4 °C above *in situ* conditions) was attained within 12 hrs after filling the mesocosms and remained constant throughout the experiment. For the CO<sub>2</sub> manipulation,

25 L of seawater (same water used for filling the mesocosms) were aerated with pure  $CO_2$  (99.99%) for  $\sim$ 24 hrs before the manipulation. The CO<sub>2</sub>-saturated seawater was used to manipulate total dissolved inorganic carbon (DIC) in the High  $CO_2$  and Greenhouse treatments while leaving total alkalinity (TA) unaffected. The CO<sub>2</sub>-saturated seawater was added to the mesocosms at an intermediate depth with a silicone tube and the manipulation performed gradually over 2 days to avoid rapid changes and acute stress for the plankton community. During the CO<sub>2</sub> manipulation, in all mesocosms, light conditions during the day were kept at 10%  $(17-20 \,\mu\text{mol photons m}^{-2} \,\text{s}^{-1})$  of the maximum surface irradiance to acclimate to the new carbonate chemistry conditions. Once target CO<sub>2</sub> levels were attained (Experiment day 3), the regular daily light regime was started. Transparent floating lids (polyethylene, 0.3 mm thick) were placed on the water surface for the duration of the experiment to minimize gas exchange with the atmosphere. A gas tracer (N<sub>2</sub>O) was used to estimate gas exchange fluxes (Czerny et al., 2013). Assuming that mesocosms in the same treatment behaved similarly in terms of gas exchange fluxes, N<sub>2</sub>O was only added to one mesocosm per treatment and any corrections (see below) were applied also to the replicate mesocosm.

#### Carbonate chemistry

DIC samples were sterile filtered (0.2 µm) with a membrane pump at a flow rate of ~50 mL min<sup>-1</sup> into 50 mL Schott Duran bottles with overflow and measured immediately using an infrared detection method for CO<sub>2</sub> (AIRICA-automated infrared inorganic carbon analyzer) with ~2 µmol kg<sup>-1</sup> precision. N<sub>2</sub>O was used as a gas tracer to estimate air/water gas exchange of CO<sub>2</sub> and the estimated loss/gain in CO<sub>2</sub> was used to calculate the biologically driven changes (i.e. photosynthesis and respiration) on DIC concentrations (i.e. DIC<sub>bio</sub>). DIC concentrations presented in this study are therefore representative for biological consumption, unless stated otherwise.

Samples for TA were filtered through GF/F filters and measured within 2 days in a Metrohm Basic Titrino 794 titration device according to Dickson *et al.* (2003). Measurements for TA and DIC were corrected with certified reference material (Dickson *et al.*, 2003). Calculation for carbonate chemistry speciation was carried out with the programme CO2SYS (Lewis and Wallace, 1998) from measured DIC, TA, temperature, salinity and nutrients (i.e.  $PO_4$  and Si(OH)<sub>4</sub>), using the dissociation constants by Mehrbach *et al.* (1973) as refitted by Dickson and Millero (1987).

#### **Biological and chemical analyses**

Dissolved inorganic nutrients nitrate  $(NO_3)$ , phosphate  $(PO_4^{3-})$  and silicic acid  $(Si(OH)_4)$  as well as dissolved organic carbon (DOC), nitrogen (DON) and phosphorus (DOP) were filtered through pre-combusted (450 °C, 6 hrs) GF/F filters with a membrane pump at a rate of  $\sim 100 \,\mathrm{mL\,min^{-1}}$  directly into vials. Except for DOC samples, which were fixed with 85% phosphoric acid (0.5% v/v final concentration) and refrigerated, all samples were frozen at -20 °C until analysis. Nutrients were measured on a segmented flow autoanalyzer (SEAL OuAAtro) according to Hansen and Koroleff (1999). DOC samples were analysed using the high-temperature catalytic oxidation method on a Shimadzu TOC-V analyzer according to Wurl and Min Sin (2009). The method was calibrated against deep-sea reference material provided by the University of Miami. DON/DOP samples were pressurecooked with Oxisolv (MERCK) and once oxidized, both organic nitrogen and phosphorus were measured as nitrate and phosphate according to Hansen and Koroleff (1999). DON/DOP concentrations were obtained by subtracting the respective inorganic nutrient concentrations.

Samples for particulate organic carbon (POC), nitrogen (PON) and phosphorus (POP) were filtered at low vacuum (200 mbar) onto pre-combusted (450 °C, 6 hrs) GF/F filters and stored at -20 °C until analysis. Filters for POC/ PON were fumed with hydrochloric acid (37% HCl) for 2 hrs to remove all inorganic carbon and subsequently dried overnight at 60 °C. Samples were analysed with a EuroEA analyzer according to Sharp (1974). POP filters were pressure-cooked for 30 min with Oxisolv (Merck) and POP oxidized to orthophosphate, which was measured photometrically according to Hansen and Koroleff (1999). Samples for particulate biogenic silica (BSi) were filtered at low vacuum (200 mbar) onto cellulose acetate filters and stored at -20 °C until analysis. Samples were cooked in a NaOH solution for 2 hrs and 15 min at ~85 °C to extract all biogenic silica into solution and then analysed as silicate according to Hansen and Koroleff (1999).

Initial POM values were calculated by averaging concentrations of all treatments on the first 2 days of the experiment ( $\pm$ standard deviation). Build-up of POM was calculated by subtracting the average of the first 2 days (once target  $pCO_2$  levels were established) from measurements at each experimental day. In most cases, maximum POM build-up was observed after nutrient depletion and maximum levels lasted for more than 2 experimental days.

#### Phytoplankton composition analysis

Phytoplankton species composition was analysed by pigment analysis [high-pressure liquid chromatography (HPLC)], size categorization by flow cytometry and Utermöhl microscopy. HPLC samples were filtered onto pre-combusted (450°C, 6 hrs) GF/F filters and stored at -80 °C until extraction. Photosynthetic pigments were separated via HPLC with 100% acetone (HPLC grade) according to the method of Van Heukelem and Thomas (2001). Phytoplankton composition was calculated with the programme CHEMTAX (Mackey *et al.*, 1996) which estimates taxon-specific contributions to total Chlorophyll *a* (Chl *a*) according to the presence and amounts of marker pigments.

Samples for flow cytometry were fixed with 25% glutaraldehyde (0.1% v/v final concentration) and immediately frozen at -80 °C until analysis. Samples were thawed for 10 min and mixed gently before measuring with an Accuri C6 flow cytometer (BD Biosciences). Autotrophic organisms were detected by their red fluorescence (fluorescence channel FL3). Using forward scatter as an indicator, three size classes were determined. Group I represented the smallest phytoplankton, approximately in the pico size range ( $<2 \mu m$ ). Groups II and III represented phytoplankton approximately in the nano  $(2-20 \,\mu\text{m})$  and micro size class (>20 µm), respectively. The contribution of each group to total Chl a fluorescence was calculated by multiplying the cell abundance within a gate and its corresponding average red fluorescence (fluorescence channel FL3).

Phytoplankton larger than  $2 \,\mu\text{m}$  were counted using the inverted microscopy method according to Utermöhl (1958) on six different occasions during the experiment (Collet, 2013), 4 sampling days from the beginning of the experiment until maximum Chl *a* and 2 sampling days during the decline of the bloom. In order to provide a comparison to the counts obtained with the flow cytometer, diatoms were grouped according to size. Group II from the microscopy counts included diatoms <20  $\mu$ m in size such as *Skeletonema costatum* and small *Chaetoceros* sp (~5  $\mu$ m), while Group III included diatoms >20  $\mu$ m such as *Thalassionema nitzschioides, Thalassiosira* sp, *Nitzschia longissima, Navicula transitans, Fragilaria* sp, *Guinardia* sp and *Chaetoceros* sp (>20  $\mu$ m).

## RESULTS

# CO<sub>2</sub> manipulation and initial experimental conditions

Averaged target  $pCO_2$  levels (first 2 sampling days after  $CO_2$  manipulation, ±standard deviation) in the High  $CO_2$  and Greenhouse treatments (1020 ± 45 and 990 ± 60 µatm  $pCO_2$ , respectively) were established within the first 3 days of the experiment and remained relatively

constant for ~10 days. Thereafter,  $pCO_2$  levels decreased to an averaged minimum of 40 ± 10 µatm in all treatments (Fig. 1A). Starting pH<sub>total</sub> values (total scale) were 7.93 ± 0.02 in the Control, 7.67 ± 0.02 in the High CO<sub>2</sub> and 7.69 ± 0.02 in the Greenhouse treatment. pH<sub>total</sub> increased up to maximum values of 8.98 ± 0.04 in the Control, 8.94 ± 0.02 in the High CO<sub>2</sub> and 8.87 ± 0.02 in the Greenhouse treatment (Fig. 1B). TA slightly increased after nutrient depletion on Day 15 from an initial average of 2072.2 ± 3.3 µmol kg<sup>-1</sup> to final concentrations of 2083.8 ± 2.4 µmol kg<sup>-1</sup> (Fig. 1C). A summary



**Fig. 1.** Development of (**A**) CO<sub>2</sub> partial pressure (pCO<sub>2</sub>), (**B**) pH<sub>totalscale</sub> and (**C**) TA during the experiment. pCO<sub>2</sub> and pH were calculated from measured TA and DIC (see Methods for details). Vertical lines indicate time of nitrate depletion; red line corresponds to Greenhouse treatment, green line to High CO<sub>2</sub> treatment and the black line to the Control.

of initial nutrient and  $pCO_2$  conditions as well as temperature and salinity over the experimental period is presented in Table I.

#### Characterization of bloom development

Phytoplankton growth was accompanied by a rapid decrease in dissolved inorganic nutrients (Figs 2A–C and 3). Faster nutrient uptake in combination with an earlier onset of the bloom and earlier timing of maximum biomass by ~2–5 days was observed under the Greenhouse conditions, whereas overall consumption of dissolved inorganic nutrients was the same between treatments (Fig. 2A–C). In all treatments, PO<sub>4</sub> was depleted 1 day earlier than any of the inorganic nitrogen species. All inorganic nutrients were exhausted by Day 15 in the Greenhouse treatment and 3–4 days later in the Control and High CO<sub>2</sub>. Chl *a* increased on average from ~0.09 ± 0.01 µg L<sup>-1</sup> to peak average concentrations of 37 ± 2.5 µg L<sup>-1</sup> in all treatments (Fig. 3). The timing of peak Chl *a* concentrations was similar among the treatments.

# Development, build-up and decline of particulate and dissolved organic matter

A concomitant increase in particulate organic matter (POM) was observed along with decreasing inorganic nutrients and increasing Chl *a*. The temporal development of PON, POP and BSi directly followed the drawdown of the corresponding inorganic nutrients (Fig. 2D–F). The onset and timing of peak concentrations was slightly earlier,  $\sim$ 3 days, under the Greenhouse conditions but with no differences in absolute maximum concentrations compared to the Control and High CO<sub>2</sub>.

PON increased from an initial average of 1.6  $\pm$  0.22 µmol N L<sup>-1</sup> to maximum average concentrations

Table I: Summary of initial experimental conditions. Mean water temperature for each treatment throughout the experiment, initial partial pressure of  $CO_2$  after  $CO_2$  manipulation (p $CO_2$ ), initial dissolved inorganic nutrient concentrations and salinity.

	Control	High CO <sub>2</sub>	Greenhouse					
Mean water temperature (°C) Initial pCO <sub>2</sub> (µatm)	$4.8 \pm 0.3$ 536 ± 25	5.4 ± 0.3 1017 ± 46	8.5 ± 0.4 990 ± 57					
Initial nutrient concentrations (µmol L <sup>-1</sup> )								
NO <sub>3</sub> <sup>-</sup>	11 ± 0.2							
PO <sub>4</sub> <sup>-3</sup>	$0.78 \pm 0.05$							
Si(OH) <sub>4</sub>	30.0 ± 1.5							
NH4 <sup>+</sup>	$2.4 \pm 0.2$							
Salinity	20.0							



Fig. 2. Development of major dissolved inorganic nutrients and POM concentrations during the experiment. (A) nitrate, (B) phosphate, (C) silicate, (D) particulate organic nitrogen, (E) particulate organic phosphorus and (F) biogenic silica. Vertical lines indicate time of nitrate depletion; red line corresponds to Greenhouse treatment, green line to High  $CO_2$  treatment and black to Control.

of 14.1  $\pm$  0.65 µmol N L<sup>-1</sup> in all treatments. PON concentrations at the end of the experiment were on average 5.9  $\pm$  1.7 µmol L<sup>-1</sup> in the Control, 5.8  $\pm$  0.83 µmol L<sup>-1</sup> in the High CO<sub>2</sub> and 3.5  $\pm$  0.77 µmol L<sup>-1</sup> in the Greenhouse treatment.

POP increased on average from 0.05  $\pm$  0.01 µmol P L<sup>-1</sup> to maximum concentrations of 0.59  $\pm$  0.07 µmol P L<sup>-1</sup> in the Control, 0.58  $\pm$  0.04 µmol P L<sup>-1</sup> in the High CO<sub>2</sub> and 0.54  $\pm$  0.04 µmol P L<sup>-1</sup> in the Greenhouse treatment. Concentrations of POP at the end of the experiment were almost twice as high in

the Control (0.37 ± 0.07 µmol P L<sup>-1</sup>) and High CO<sub>2</sub> (0.32 ± 0.07 µmol P L<sup>-1</sup>) compared to the Greenhouse treatment (0.15 ± 0.05 µmol P L<sup>-1</sup>). After PO<sub>4</sub> exhaustion, DOP slightly increased to maximum concentrations of 0.16 ± 0.07 µmol P L<sup>-1</sup> in all treatments and remained stable throughout the experiment with no distinct difference between treatments (data not shown).

BSi, indicative for diatom biomass, followed the development of Chl *a* from the beginning of the experiment up to the peak of the bloom. BSi increased on average from  $1.0 \pm 0.1 \ \mu$ mol L<sup>-1</sup> to maximum concentrations of



**Fig. 3.** Temporal development of total Chl *a* during the experiment. Vertical lines indicate time of nitrate depletion; red line corresponds to Greenhouse treatment, green line to High  $CO_2$  treatment and black to Control.

28.5 ± 1.2 µmol L<sup>-1</sup> in the Control, 27.9 ± 1.0 µmol L<sup>-1</sup> in the High CO<sub>2</sub> and 30.0 ± 0.6 µmol L<sup>-1</sup> in the Greenhouse treatment. BSi concentrations remained relatively high after the peak of the bloom while the concentrations of pigments associated to diatoms decreased similarly to Chl *a* (Figs 3 and 6A). BSi comprised not only living diatoms but also silicate frustules which remained suspended in the water column for some time before sinking out. BSi concentrations at the end of the experiment were more than twice as high in the Control and High CO<sub>2</sub> (14.8 ± 3.6 and 11.9 ± 3.9 µmol L<sup>-1</sup>, respectively) compared to the Greenhouse treatment (4.9 ± 1.9 µmol L<sup>-1</sup>).

#### Carbon cycling

During the first part of the experiment (Days 1–20) calculated CO<sub>2</sub> out-gassing did not exceed 12 µmol L<sup>-1</sup> d<sup>-1</sup> and during the second half of the experiment (Days 20–44) calculated CO<sub>2</sub> in-gassing did not exceed 20 µmol L<sup>-1</sup> d<sup>-1</sup>. Higher out-gassing occurred in the High CO<sub>2</sub> and Greenhouse treatments, where CO<sub>2</sub> gradients were larger. However, the lid in one of the Control mesocosms was not properly covering the water surface and a small opening was created near the propeller, increasing gas exchange in this mesocosm towards the end of the experiment. DIC concentrations in this Control steadily increased after Day 20 of the experiment, most likely from in-gassing, since no difference in POC or DOC concentrations was observed between this and the replicate mesocosm (Fig. 4).

Photosynthetic uptake of DIC was accompanied by an increase in POC and DOC concentrations (Fig. 4A–C). Slightly faster DIC uptake and an earlier peak in POC



**Fig. 4.** Carbon dynamics during the experiment for (**A**) DIC (corrected for gas exchange), (**B**) particulate organic carbon and (**C**) DOC. Colour coding according to Fig. 1. Vertical lines indicate time of nitrate depletion; red line corresponds to Greenhouse treatment, green line to High  $CO_2$  treatment and black to Control. Please note that concentrations of one of the Controls deviate from its duplicate after the bloom peak due to a defective lid (see section *Carbon cycling* in the Results section for details).

concentrations were observed in the Greenhouse treatment. Photosynthetic carbon fixation during the experiment decreased DIC concentrations from 2050.3  $\pm$  9 µmol L<sup>-1</sup> to 1657.6  $\pm$  72 µmol L<sup>-1</sup> in the Control and from 2113.1  $\pm$  2 µmol L<sup>-1</sup> to 1627  $\pm$  4 µmol L<sup>-1</sup> in the High CO<sub>2</sub> and 2095.5  $\pm$  1 µmol L<sup>-1</sup> to 1633.4  $\pm$  10 µmol L<sup>-1</sup> in the Greenhouse treatments (Fig. 4A). A slightly higher DIC drawdown of 485.7  $\pm$  5.8 µmol L<sup>-1</sup> was calculated in the High CO<sub>2</sub> compared to 389.9  $\pm$  99 µmol L<sup>-1</sup> and 462.0  $\pm$  10 µmol L<sup>-1</sup> in the Control

and Greenhouse treatments, respectively. The large standard deviation in the Control stems from differences in DIC concentrations between the duplicate mesocosms due to issues with the covering lid mentioned in the previous paragraph.

POC increased from an initial average of  $13.2 \pm 1.1 \,\mu\text{mol} \text{ C L}^{-1}$  to maximum concentrations of  $289.7 \pm 13.6 \,\mu\text{mol} \text{ C L}^{-1}$  in the Control,  $305.2 \pm 15.8 \,\mu\text{mol} \text{ C L}^{-1}$  in the High CO<sub>2</sub> and  $301.6 \pm 26.1 \,\mu\text{mol} \text{ C L}^{-1}$  in the Greenhouse treatment (Fig. 4B). Maximum build-up of POC occurred ~5 days after inorganic nutrient depletion and remained relatively constant for 10–15 days, then decreased rapidly towards the end. DOC steadily increased from an average of  $220.6 \pm 7.3 \,\mu\text{mol} \text{ C L}^{-1}$  to maximum concentrations of  $271.3 \pm 16 \,\mu\text{mol} \text{ C L}^{-1}$  in the Greenhouse and slightly higher in the High CO<sub>2</sub> treatments with  $302.3 \pm 7 \,\mu\text{mol} \text{ C L}^{-1}$  at the end of the experiment (Fig. 4C).

## Elemental stoichiometry of POM

Inorganic nutrients were assimilated into the respective POM pools according to Redfield proportions up to Day 15 of the experiment. Thereafter, due to nutrient limitation, POC:PON and POC:POP increased strongly, while PON:POP remained relatively stable (Fig. 5). Both POC: POP and PON:POP started at values almost twice as high as the respective Redfield ratio (106:1 and 16:1), suggesting a low contribution of living phytoplankton biomass to the POM pool at the beginning of the experiment (Fig. 5B and C). Carbon fixation continued after nutrient depletion, leading to maximum POC:PON of up to 30–40 and POC:POP of up to 600–800. Except for the faster bloom onset in the Greenhouse treatment, no differences in maximum ratios were observed between the treatments (Fig. 5A–C).

# Phytoplankton composition derived from HPLC analysis

Chl *a* was almost exclusively produced by diatoms with minor contributions from cryptophytes, prasinophytes, dinophytes and chlorophytes (Fig. 6A–E). Diatoms contributed up to ~35 µg L<sup>-1</sup> to the total average of  $37 \pm 3 µg L^{-1}$  Chl *a* (Fig. 6A). Cryptophytes were the second group after diatoms contributing up to ~1.2 µg L<sup>-1</sup> to Chl *a* while the minor contributions from the other phytoplankton groups did not exceed ~0.5 µg L<sup>-1</sup> (Fig. 6B–E). Contributions of prasinophytes were higher in the Greenhouse treatment compared to the Control and High CO<sub>2</sub> treatments while the opposite was observed for cryptophytes showing twice the concentrations in the Control and High CO<sub>2</sub>



**Fig. 5.** Elemental stoichiometry of POM during the experiment. Colour coding according to Fig. 1. Vertical lines indicate time of nitrate depletion; red line corresponds to Greenhouse treatment, green line to High  $CO_2$  treatment and black to Control. Grey horizontal lines represent Redfield proportions of (**A**) 6.6, (**B**) 106:1 and (**C**) 16:1.

compared to the Greenhouse treatment (Fig. 6D and E). Chl a contributions of chlorophytes and dinophytes showed a similar response, increasing after the peak of the bloom with higher concentrations in the Greenhouse treatment (Fig. 6B and C), though at relatively low-absolute concentrations.

# Phytoplankton composition based on flow cytometry and microscopy

Phytoplankton abundances as measured by flow cytometry increased to peak concentrations similarly to the development of Chl *a* (Fig. 7A–C). Abundances in Group I (<2  $\mu$ m) decreased after reaching peak



**Fig. 6.** Chl *a* equivalent concentrations of: (**A**) Diatoms, (**B**) Chlorophytes, (**C**) Dinophytes, (**D**) Cryptophytes and (**E**) Prasinophytes analysed by CHEMTAX. Vertical lines indicate time of nitrate depletion; red line corresponds to Greenhouse treatment, green line to High CO<sub>2</sub> treatment and black to Control.

concentrations between Days 15 and 18 while abundances in Group II ( $\sim 2-20 \,\mu\text{m}$ ) and Group III ( $> 20 \,\mu\text{m}$ ) remained relatively stable for  $\sim 10-15$  days. The contribution of Group I to total Chl fluorescence (i.e. cell abundances multiplied by average red fluorescence) at the beginning of the experiment was up to 30% but steadily decreased towards the peak of the bloom, accounting for <2% thereafter (Fig. 7D). Group II showed peak abundances 1 day after nutrient depletion. Abundances of Group II were  $\sim 25\%$  lower in the Greenhouse treatment compared to those in the Control and High CO<sub>2</sub> treatments (Fig. 7B). The contribution of Group II to Chl fluorescence during the bloom peak was >60% in the Control and High CO<sub>2</sub> and  $\sim 45-50\%$  in the Greenhouse treatment (Fig. 7E). Conversely, Group III showed higher abundances in the Greenhouse treatment during peak Chl a (and thereafter) compared to the Control and High CO<sub>2</sub> treatments (Fig. 7C). The contribution to Chl fluorescence during peak concentrations by Group III was  $\sim 50\%$  in the Greenhouse treatment and <40% in the Control and High CO<sub>2</sub> (Fig. 7F). Since more than 95% of the Chl could be attributed to diatoms from the peak of the bloom towards the end of the experiment, Groups II and III as identified by flow cytometry are most likely primarily composed of diatoms.

Although absolute values were not the same between microscopic and flow cytometry counts, a tendency towards larger diatom species in the Greenhouse treatment was observed with both methods. The shift towards larger diatoms in the Greenhouse treatments was most likely driven by the increased abundances of *Thalassiosira* sp, *Thalassionema nitzschioides* and *Nitzschia longissima* (Table II).



**Fig. 7.** Cell counts according to group sizes from flow cytometry (left panel): (**A**) Group I ( $<2 \mu$ m), (**B**) Group II ( $2-20 \mu$ m) and (**C**) Group III ( $>20 \mu$ m) and percent contribution to total Chlorophyll fluorescence (right panel): (**D**) Group I, (**E**) Group II and (**F**) Group III. Colour coding according to Fig. 1. Vertical lines indicate time of nitrate depletion; red line corresponds to Greenhouse treatment, green line to High CO<sub>2</sub> treatment and black to Control.

Microzooplankton abundances in our experiments (comprising flagellates, dinoflagellates and ciliates) increased at the beginning of the experiment up to the peak of the bloom and decreased rapidly thereafter, similarly to the results reported by Lewandowska and Sommer (2010), showing no difference between treatments.

## DISCUSSION

Ocean acidification and warming have the potential to significantly affect marine biogeochemical element cycling

by modifying the balance between primary production and heterotrophic processes and/or changing plankton community composition (Riebesell and Tortell, 2011). Our study examined the combined effects of these two drivers on the species composition and biogeochemical dynamics during a spring phytoplankton bloom with a natural plankton community from a temperate coastal ecosystem in the Baltic Sea. The majority of experiments in this region focused on the individual effects of either warming (Wohlers-Zöllner *et al.*, 2012; Sommer *et al.*, 2012b) or CO<sub>2</sub> (Schulz and Riebesell, 2013; Engel *et al.*, 2014). So far only the study by Paul *et al.* (2015)

Experiment day	1	11	16	17	19	31	44
Group II (mean abun	idance ± SD)						
Chaetoceros sp ~	5μm						
Control	-	$6 \pm 5$	39 ± 18	37 ± 23	37 ± 18	8±5	5 ± 1
High CO <sub>2</sub>	-	4 ± 3	28 ± 3	16 ± 1	28 ± 11	11 ± 9	-
Greenhouse	-	$5 \pm 4$	25 ± 18	15 ± 5	14 ± 1	2 ± 1	-
Skeletonema cost	tatum						
Control	2 ± 1	67 ± 30	1032 ± 1013	5686 ± 1727	2245 ± 1282	3989 ± 1474	682 ± 453
High CO <sub>2</sub>	1 ± 0.5	106 ± 22	1487 ± 60	2430 ± 309	3019 ± 2659	1929 ± 357	25 ± 1
Greenhouse	1 ± 0.5	349 ± 109	$2393 \pm 670$	3456 ± 2410	2833 ± 511	516 ± 124	-
Group III (mean abur	ndance ± SD)						
Chaetoceros sp >	20 µm						
Control	_	-	62 ± 9	94 ± 27	73 ± 23	106 ± 29	$66 \pm 14$
High CO <sub>2</sub>	-	10 ± 7	57 ± 15	72 ± 11	107 ± 16	72 ± 51	34 ± 19
Greenhouse	-	19 ± 1	115 ± 15	95 ± 23	74 ± 21	35 ± 11	-
<i>Thalassiosira</i> sp							
Control	-	28 ± 3	$305 \pm 156$	730 ± 50	657 ± 153	$695 \pm 56$	383 ± 89
High CO <sub>2</sub>	-	$42 \pm 4$	$440 \pm 41$	960 ± 23	784 ± 149	655 ± 357	$398 \pm 69$
Greenhouse	1 ± 1	119 ± 38	1054 ± 80	1022 ± 257	1061 ± 283	1116 ± 357	174 ± 101
Thalassionema nit	tzschioides						
Control	-	3 ± 1	12 ± 9	12 ± 1	23 ± 7	25 ± 5	-
High CO <sub>2</sub>	-	-	18 ± 8	18 ± 2	13 ± 1	5 ± 2	-
Greenhouse		7 ± 4	$45 \pm 20$	37 ± 5	$41 \pm 22$	33 ± 17	-
Nitzschia longissir	ma						
Control	-	6 ± 1	15 ± 4	28 ± 5	18 ± 1	24 ± 4	34 ± 18
High CO <sub>2</sub>	-	4 ± 1	$23 \pm 2$	31 ± 3	$23 \pm 6$	37 ± 10	27 ± 9
Greenhouse	-	$44 \pm 4$	271 ± 98	289 ± 23	393	335 ± 27	148 ± 141
Navicula transitan	S						
Control	-	3 ± 1	15 ± 3	26 ± 17	49 ± 3	73 ± 11	$22 \pm 2$
High CO <sub>2</sub>	-	7 ± 4	13 ± 11	39 ± 20	$55 \pm 28$	85 ± 21	6 ± 3
Greenhouse	-	11 ± 2	26 ± 8	32 ± 3	38 ± 16	$24 \pm 6$	-
<i>Fragilaria</i> sp							
Control	-	4 ± 3	$38 \pm 27$	71 ± 20	$57 \pm 5$	57 ± 6	18 ± 9
High CO <sub>2</sub>	-	-	15 ± 11	43 ± 36	47 ± 17	17 ± 15	1 ± 1
Greenhouse	-	13 ± 7	$42 \pm 23$	33 ± 3	$38 \pm 28$	$23 \pm 6$	-
<i>Guinardia</i> sp							
Control	-	2 ± 1	94 ± 18	179 ± 63	203 ± 116	277 ± 71	318 ± 133
High CO <sub>2</sub>	-	$6 \pm 4$	145 ± 69	421 ± 186	69 ± 49	868 ± 90	$324 \pm 74$
Greenhouse	-	12 ± 1	$140 \pm 35$	178 ± 6	115 ± 81	$105 \pm 51$	4 ± 1

Table II: Summary of microscopic counts of diatom species for Groups II and III. Mean abundances represent the average of the duplicates on the specific experiment day ±standard deviation.

examined the combined influence of both drivers on a natural autumn Baltic Sea plankton assemblage.

# Phytoplankton community composition and shifts in size

A phytoplankton bloom, dominated by diatoms (98% of total Chl *a*), developed in our experiments under all treatment conditions (Fig. 6). The chain-forming diatom *S. costatum* typically dominates spring blooms in the Kiel Bight (Wasmund *et al.*, 2008) and was also the most abundant species under all treatment conditions during our experiments (60–70% of diatom abundance). There is significant evidence from the field showing that over the past 50 years, diatom abundance positively correlated with warming and increasing wind conditions in the North Sea (Hinder *et al.*, 2012). Phytoplankton biomass almost doubled over the past century with diatoms

contributing to up to 80% of the carbon biomass during spring blooms in the Kiel Bight, the same region of the present study (Wasmund et al., 2008). Both of these studies also reported increasing abundances of Skeletonema spp, the dominating group on our experiments. However, in our experiments we observed a shift towards larger diatom species (mainly Thalassiosira sp and Nitzschia longissima, Table II) in the Greenhouse treatment. Due to experimental constraints we did not include a "warming only" treatment in our design. Thus, we could not disentangle the influence of warming and acidification individually on this phytoplankton community. Nonetheless, in the following section we will discuss whether the individual or the combined effect of both stressors could have driven the shift in phytoplankton size in the Greenhouse treatment of our experiments.

The shift towards larger diatom species was somewhat contrary to previous studies which reported that warming alone generally decreased abundances of large diatom species, reduced cell size and benefited smaller phytoplankton (Sommer and Lengfellner, 2008a; Daufresne et al., 2009; Lewandowska and Sommer, 2010; Morán et al. 2010). However, the increase of small-sized phytoplankton in the latter experiments was mostly associated with higher grazing rates of zooplankton on larger phytoplankton species allowing smaller ones to proliferate (Keller et al., 1999; O'Connor et al., 2009; Sommer and Lewandowska, 2010). Furthermore, the addition of adult copepods in those experiments might have generated a selective grazing pressure on larger phytoplankton since both copepods and ciliates in this region have shown to preferentially feed on larger phytoplankton during their adult lifetime (Sommer et al., 2002; Sommer and Lengfellner, 2008b; Aberle et al., 2007).

While physiological studies with single species give valuable information on tolerance ranges and help us understand underlying mechanisms for the response of species to different stressors (Torstensson et al., 2013; Li et al., 2018), these results cannot be extrapolated to how a species would respond when being part of a more complex (mixed) plankton community. During CO<sub>2</sub> enrichment experiments with natural phytoplankton communities, both positive and negative effects of increasing CO2 were observed on a number of functional groups and species. For instance, small picoeucarvotes profited from higher CO<sub>2</sub> levels (Yoshimura et al., 2013; Crawfurd et al., 2017; Schulz et al., 2013, 2017). The specific mode of carbon concentrating mechanism (CCM) employed for inorganic carbon acquisition was speculated to be among the reasons for this response, relying more on passive diffusion in very small cells. However, not only relatively small but also relatively large phytoplankton cells have been found to directly profit from higher CO<sub>2</sub> concentrations. Incubation experiments with natural Southern Ocean phytoplankton communities showed that large chain-forming Chaetoceros sp profited at high-CO<sub>2</sub> conditions at the expense of smaller diatoms such as Pseudo-nitzschia subcurvata and Cylindrotheca closterium (Tortell et al., 2008; Feng et al., 2010). Similarly, Wu et al. (2011) showed that although increasing CO<sub>2</sub> consistently enhanced growth rates of five different diatom species ranging in size, the largest enhancement (i.e. 33% compared to 5%) was observed for the largest species. This was hypothesized to be related to an alleviation of carbon limitation in larger cells which have a thicker diffusive boundary layer compared to smaller cells.

Incubation experiments with natural Bering Sea phytoplankton communities revealed either a negative or no effect of elevated  $CO_2$ , increased temperature or the combination of both on diatom cell numbers in comparison to a control (Hare *et al.*, 2007). A positive effect of elevated CO<sub>2</sub> alone, but not in combination with increased temperature, was observed in a North Atlantic community (Feng et al., 2009). In the latter experiment, however, it is important to note, that nutrient levels applied with silicate to nitrate ratios of about 0.15 are rather disadvantageous for diatom growth, also indicated by lower diatom biomass towards the end of the experiment in most incubations compared to the beginning. Also, a shift towards larger species within the same functional group (i.e. diatoms), as observed in our experiment, is not easily detected by looking at overall abundances. For example, Feng et al. (2009) reported an increase in overall diatom abundance only in the High  $CO_2$  treatment but the shift towards larger species (i.e. Pseudo-nitzschia to Cylindrotheca ratio) was more pronounced in the combined treatment ("Greenhouse"), although the absolute diatom abundances were three times lower than in the High CO<sub>2</sub>. Additionally, incubation experiments with a natural plankton community from the subtropical North Atlantic showed that if the community was initially dominated by diatoms, elevated CO<sub>2</sub> tended to favour the larger diatom species (Eggers et al. 2014).

# Temporal development of the bloom and POM dynamics

The temporal development of POM during the nutrient replete phase of the experiment (Days 1-15) followed a typical natural diatom-dominated spring bloom exhibiting rapid drawdown of inorganic nutrients and accompanied by exponential autotrophic growth (as represented by Chl a) and biomass build-up of POC, PON, POP, and BSi. Once inorganic nutrients were exhausted between Days 15 and 18 (depending on treatment), peak concentrations in the respective particulate pools were reached (Fig. 2). During the first part of the experiment (i.e. exponential growth phase), higher temperature in the Greenhouse treatment accelerated nutrient uptake and caused maximum POM concentrations being reached  $\sim$ 2–5 days earlier compared to the Control and the High  $CO_2$  treatments. Such a response (i.e. earlier onset of the bloom and faster biomass build-up) was consistently observed in mesocosm experiments investigating warming (Sommer et al., 2012a) and is most likely due to enhanced metabolic activities at higher temperature. Even so, a week or less (2-3 days) acceleration of such phytoplankton bloom events (i.e. less time from bloom initiation to bloom peak) in nature would be negligible and does not seem to be of present concern based on long-term monitoring records on spring bloom regions (Wiltshire et al., 2008).

Peak concentrations of PON, POP and BSi were observed 1 day after the respective inorganic nutrients were depleted with no differences in maximum POM concentrations between the treatments (Fig. 2 D-F). Mesocosm experiments investigating the impact of warming on plankton communities from the Kiel Bight consistently yielded a decrease in phytoplankton biomass with increasing temperature during spring blooms (Sommer and Lengfellner 2008; Sommer et al., 2012b) and autumn blooms (Paul et al., 2015) with only one study showing the opposite response during a summer bloom (Taucher et al., 2012). Taucher et al. (2012) suggested that the contrasting results were most likely due to differences in the phytoplankton assemblage. While most of the spring experiments were dominated by the typical "spring bloom" species S. costatum, the summer experiment by Taucher et al. (2012) was dominated by the diatom Dactyliosolen fragilissimus.

DIC uptake and POC build-up continued for another 5 days after nutrient depletion and thereafter POC concentrations remained constant for ~15 days while DIC concentrations remained low (Fig. 4B-C). Chl a, however, declined sharply directly afterwards (Fig. 3). These results are contrasting to previous warming experiments where POM concentrations were more closely coupled to Chl a and decreased immediately after peak concentrations (Wohlers et al., 2009; Biermann et al., 2014). In the latter experiments, however, mesozooplankton (i.e. copepods) was added after initial filling of the mesocosms to investigate the influence of warming on grazing activities. Therefore, phytoplankton growth was topdown controlled in these experiments and biomass build-up restricted (Sommer and Lewandowska, 2010; Lewandowska et al., 2014). The lower grazing pressure in our experiment favoured fast autotrophic biomass build-up in all treatments followed by a phase of relatively stable POM concentrations for 10-15 days after the bloom peak (Fig. 2D, E and F). Relatively stable POM concentrations after the bloom peak were also observed in similar experiments with nutrient induced diatom blooms under low-grazing pressures (Engel et al., 2002; Wohlers-Zöllner et al., 2011). As in our experiment, post-bloom conditions most likely lead to the retention of diatom aggregates in the water column (Smayda and Boleyn, 1966; Sarthou et al., 2005).

Towards the end of the experiment and after the phase of stable POM concentrations, the loss of POM in the Greenhouse treatment was about twice as fast as in the Control and High  $CO_2$  treatments. As this loss of POM did not correspond to an increase in any of the dissolved matter (both organic and inorganic) pools, faster degradation at elevated temperature was unlikely the cause. Thus, two possible mechanisms remain that could

explain these results, either the initiation of aggregation processes and therefore the earlier sinking/loss of POM from the water column (Piontek *et al.*, 2009), was driven by higher temperatures or by the accompanied phytoplankton community shift towards larger diatoms.

Diatoms are known to be important organic matter exporters as they often contribute a high fraction of the biomass produced during spring blooms (Buesseler, 1998; Sarthou et al., 2005 and references therein). Boyd and Newton (1995) followed the development of the spring bloom in 2 consecutive years in the North East Atlantic and showed that although primary production was comparable in both years the export of organic carbon was more efficient, from the surface to the deep ocean, when the phytoplankton assemblage was dominated by large diatom species. Thus, the higher abundance of larger diatoms in our experiments could have been a crucial player in determining faster loss processes in the Greenhouse treatment. However, while cellspecific sinking rates between small and large cells might have been relevant in our 1 m deep tanks, extrapolation of our results to sinking rates in nature are to be made cautiously.

#### CONCLUSION

It was recently suggested that the interactive effects of environmental drivers, rather than the sum of the individual effects, best explained plankton responses in aquatic ecosystems (Christensen et al., 2006). In our experiment with a natural phytoplankton assemblage from a coastal eutrophic ecosystem, we observed that in the combined ocean acidification and warming treatment a shift in size towards larger diatoms was favoured. This shift was not observed in the High CO<sub>2</sub> treatment and while we did not have a warming treatment alone, based on evidence from similar experiments investigating warming and  $CO_2$  separately, we hypothesize that the combination of both factors was important for the shift in size. The potential for warming and CO<sub>2</sub> induced changes in species composition and size distribution during bloom events in the future ocean is likely to influence biogeochemical element cycling with consequences for CO<sub>2</sub> partitioning between ocean and atmosphere.

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