



Sulfide Protects *Staphylococcus aureus* from Aminoglycoside Antibiotics but Cannot Be Regarded as a General Defense Mechanism against Antibiotics

Julia Weikum,^{a*} Niklas Ritzmann,^{a*} Nils Jelden,^{a,b} Anna Klöckner,^{a,b*} Sebastian Herkersdorf,^a Michael Josten,^{a,b,c} Hans-Georg Sahl,^{b,c} Fabian Grein^{a,b}

^aInstitute for Pharmaceutical Microbiology, University Hospital Bonn, University of Bonn, Bonn, Germany

^bGerman Center for Infection Research, Bonn-Cologne Site, Bonn, Germany

^cInstitute of Medical Microbiology, Immunology, and Parasitology, University Hospital Bonn, University of Bonn, Bonn, Germany

ABSTRACT Sulfide production has been proposed to be a universal defense mechanism against antibiotics in bacteria (K. Shatalin, E. Shatalina, A. Mironov, and E. Nudler, *Science* 334:986–990, 2011, doi:10.1126/science.1209855). To gain insight into the mechanism underlying sulfide protection, we systematically and comparatively addressed the interference of sulfide with antibiotic activity against *Staphylococcus aureus*, as a model organism. The impact of sulfide and sulfide precursors on the antibiotic susceptibility of *S. aureus* to the most important classes of antibiotics was analyzed using modified disk diffusion assays, killing kinetic assays, and drug uptake studies. In addition, sulfide production and the impact of exogenously added sulfide on the physiology of *S. aureus* were analyzed. Sulfide protection was found to be limited to aminoglycoside antibiotics, which are known to be taken up by bacterial cells in an energy-dependent process. The protective mechanism was found to rely on an inhibitory effect of sulfide on the bacterial respiratory chain, leading to reduced drug uptake. *S. aureus* was found to be incapable of producing substantial amounts of sulfide. We propose that bacterial sulfide production should not be regarded as a general defense mechanism against antibiotics, since (i) it is limited to aminoglycosides and (ii) production levels vary considerably among species and, as for *S. aureus*, may be too low for protection.

KEYWORDS *Staphylococcus aureus*, aminoglycosides, resistance mechanisms

The rapid emergence of antibiotic-resistant bacteria has become a global threat. Therefore, studies on antibiotics and bacterial resistance modes have become a central topic for research, in hopes of finding alternative therapies (1–3). The formation of reactive oxygen species (ROS) is considered to occur as a general downstream effect in bacteria challenged with bactericidal antibiotics (4). In 2011, a novel antibiotic resistance mechanism mediated by hydrogen sulfide (H₂S) was proposed for several pathogenic bacteria, including *Staphylococcus aureus* (5). It was proposed that sulfide reduces the cellular formation of ROS by interfering with the Fenton reaction and by stimulating the ROS-scavenging enzymes superoxide dismutase and catalase (5).

Sulfide is a weak acid and the equilibrium between its protonated and deprotonated states depends greatly on the pH. Here, we use the term sulfide to describe the sum of the different protonated forms present in solution (H₂S, HS⁻, and S₂⁻). Sulfide is a major part of the microbial sulfur cycle, as several bacteria can use sulfide as their sole electron source (6). It is produced during dissimilatory sulfate reduction, where it results from the stepwise reduction of sulfate via sulfite and an unusual protein trisulfide (7), and can also be formed as the product of cysteine degradation either via cystathionine-

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Address correspondence to Fabian Grein, grein@uni-bonn.de.

* Present address: Julia Weikum, Centre for Molecular Medicine Norway, Oslo, Norway; Niklas Ritzmann, Institute for Molecular Microbiology and Biotechnology, University of Münster, Münster, Germany; Anna Klöckner, MRC Centre for Molecular Bacteriology and Infection, Imperial College London, London, United Kingdom.

J.W. and N.R. contributed equally to this work.

β -synthase (CBS) and cystathionine- γ -lyase (CSE) or via 3-mercaptopyruvate sulfurtransferase (3MST) (5). These enzymes have also been shown to produce sulfide in mammals. Here, sulfide has been identified as a third gasotransmitter, in addition to nitric oxide (NO) and carbon monoxide (CO), and various physiological and pathophysiological functions have been attributed to sulfide (8–10). It is well known for its toxicity, which has been associated with its ability to interact with heme proteins, most notably cytochromes in the respiratory chain (11–14).

Aminoglycosides target the 30S subunit of bacterial ribosomes, leading to a cascade of pleiotropic effects that account for the bactericidal activity (15). They are among the antibiotics that are commonly used to treat infections caused by Gram-positive cocci such as *S. aureus*. Before they can interact with their intracellular targets, aminoglycosides need to cross the cytoplasmic membrane. This process is energy dependent, requiring a threshold electrochemical potential across the cytoplasmic membrane. In line with that, compounds that inhibit the respiratory chain, such as cyanide, or uncouple the electrochemical gradient, such as carbonyl cyanide *m*-chlorophenyl hydrazone (CCCP), have been shown to protect bacteria from aminoglycoside toxicity (reviewed by Taber et al. [16]). Furthermore, anaerobes are usually not susceptible to aminoglycosides. Given the known inhibitory effect of sulfide on the respiratory chain and the dependence of aminoglycoside uptake on respiration, we hypothesized that sulfide impairs the *S. aureus* respiratory chain and consequently reduces uptake of the drug.

RESULTS

Sulfide production has been proposed to be a universal defense mechanism against antibiotics in various bacteria, including *S. aureus* (5), such that interference with sulfide formation may become a useful strategy to restore the antibiotic susceptibility of pathogenic bacteria. In search of novel antistaphylococcal targets, we aimed to systematically address the impact of sulfide on antibiotic susceptibility of different *S. aureus* strains. To this end, we established a modified disk diffusion assay that allows continuous incubation with gaseous sulfide (Fig. 1), thereby diminishing effects that occur due to the oxidation of sulfide in liquid medium. The system is based on the continuous production of gaseous sulfide by *Escherichia coli*, which can have a direct impact on a regular disk diffusion assay in which *S. aureus* is plated on Mueller-Hinton (MH) agar plates. In *E. coli*, sulfide is produced mainly from cysteine by 3MST, encoded by *sseA* (5). Therefore, the $\Delta sseA$ strain served as a control to exclude the impact of other volatile compounds produced by *E. coli*. To verify the system, sulfide quantification was performed by replacing the inner MH agar plate with a petri dish filled with a solution of zinc acetate (2% in H₂O) and using the methylene blue assay method the next day. An intense blue color, indicative of sulfide, was observed when the *E. coli* wild-type (WT) strain was used, while no color change was observed with the control *E. coli* $\Delta sseA$ strain (data not shown).

Surprisingly, we observed sulfide-mediated protection of all *S. aureus* strains tested only against the class of aminoglycoside antibiotics. In contrast, the strains tended to be more sensitive to the other classes of antibiotics in the presence of sulfide (Fig. 1). We obtained highly similar values for *S. aureus* RN4220, the methicillin-sensitive *S. aureus* (MSSA) strain used by Shatalin et al. (5), and the closely related strain HG003, which we then chose as a model organism for the following studies. Killing kinetic assays were performed to verify the observations from the agar diffusion assay. Therefore, a culture in the exponential growth phase was treated with the aminoglycoside gentamicin and sulfide; the latter was added either at the beginning of the experiment, to a final concentration of 1 or 4 mM, or by periodic addition of 1 mM sulfide every 30 min. A protective effect was detectable in the presence of all sulfide concentrations, in comparison to the untreated sample (Fig. 2A). However, only periodic addition of sulfide led to constant protection. This corresponds to the sulfide concentrations measured simultaneously with the killing kinetics (Fig. 2B). Under oxic conditions, sulfide is known to quickly oxidize to a mixture of various compounds, such

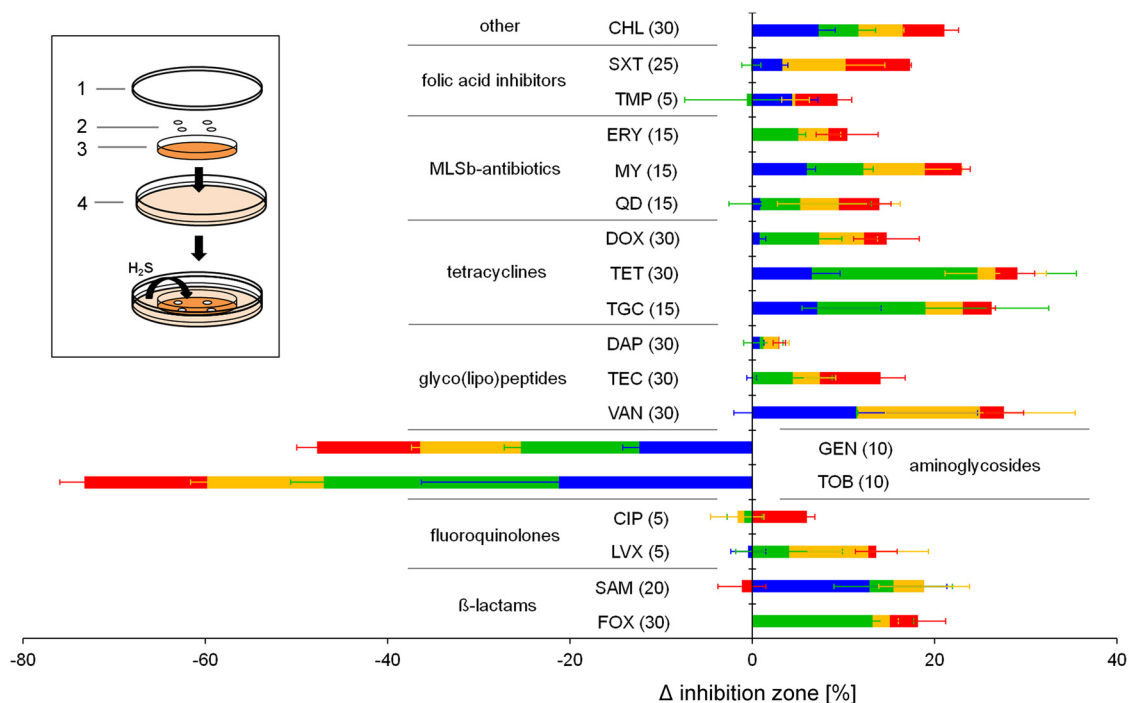


FIG 1 Impact of externally added sulfide on the susceptibility of *S. aureus* to several antibiotics. The susceptibility of *S. aureus* HG003 (red), RN4220 (orange), Newman (green), and USA300 (blue) to various classes of antibiotics was assessed. Cells were plated on MH agar plates and incubated with or without the external addition of sulfide. FOX, cefoxitin; SAM, ampicillin-sulbactam; LVX, levofloxacin; CIP, ciprofloxacin; TOB, tobramycin; GEN, gentamicin; VAN, vancomycin; TEC, teicoplanin; DAP, daptomycin; TGC, tigecycline; TET, tetracycline; DOX, doxycycline; QD, quinupristin-dalfopristin; MY, lincomycin; ERY, erythromycin; TMP, trimethoprim; SXT, trimethoprim-sulfamethoxazole; CHL, chloramphenicol. Numbers in parentheses indicate antibiotic contents (in micrograms). Shown are the mean differences (\pm standard deviations) in the inhibition zones between sulfide-treated and nontreated plates. Experiments were performed in triplicate. The inset shows a schematic illustration of the assay. 1, lid of the petri dish; 2, antimicrobial susceptibility disks; 3, *S. aureus* plated on MH agar; 4, 145-mm-diameter petri dish containing 35 ml of LB inoculated with the *E. coli* Δ ss_{eA} strain or W3110 supplemented with cysteine. Sulfide produced by *E. coli* affects the *S. aureus* disk diffusion assay.

as polysulfide, thiosulfate, and sulfate (17, 18). In line with that, the sulfide concentrations at time zero were significantly lower than the theoretically calculated values and, during the experiment, sulfide levels declined quickly within the first minutes after addition in all samples. In the case of periodic addition, however, the sulfide concentration never dropped below a certain level, and constant protection was observed over the entire time course. This finding indicates that a certain minimal sulfide concentra-

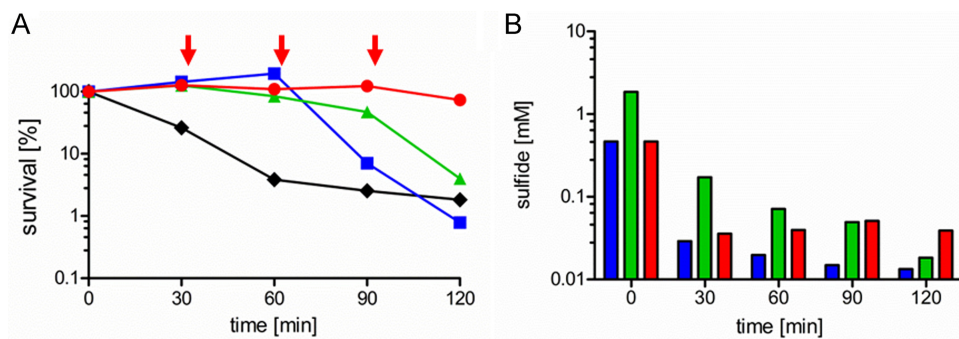


FIG 2 Correlation of protection of *S. aureus* HG003 from gentamicin toxicity with sulfide stability. (A) Survival of *S. aureus* grown to an OD₆₀₀ of 0.5 in LB after treatment with 25 mg/liter gentamicin and different concentrations of sulfide (1 mM [blue line], 4 mM [green line], or periodic addition of 1 mM sulfide at 30-min intervals, as indicated by arrows [red line]) or without addition of sulfide (black line). (B) Sulfide concentrations measured during the killing experiment shown in panel A. Colors correspond to the different sulfide concentrations added, as described for the experiment in panel A. The data show a representative experiment of three independent experiments with comparable results.

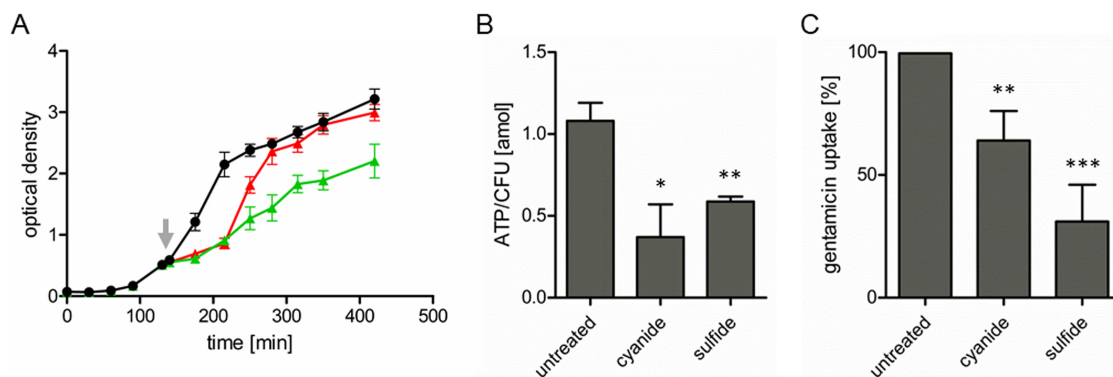


FIG 3 Effects of sulfide and cyanide on *S. aureus* growth, ATP concentrations, and gentamicin uptake. (A) Cell growth was monitored by measuring the OD₆₀₀. In the exponential growth phase (OD₆₀₀ of ~0.5), the culture was split (indicated by the arrow) and 1 mM sulfide (red line) or 0.1 mM cyanide (green line) was added or the cells were left untreated (black line). The data are means \pm standard deviations of three independent experiments. (B) Effects of sulfide and cyanide on ATP levels in *S. aureus*. Cells were grown to an OD₆₀₀ of 0.5 and then incubated with 0.1 mM cyanide or 1 mM sulfide for 10 min or left untreated. ATP levels are expressed in attomoles of ATP per CFU, which was determined simultaneously. (C) Impact of sulfide and cyanide on gentamicin uptake by *S. aureus*. Uptake of bodipy FL-labeled gentamicin was determined under conditions as in panel B. Values determined for the untreated control were set as 100%, and the treated samples are shown in relation to the control. Data are means \pm standard deviations of three independent experiments. Statistical analysis was analyzed by *t* test: *, *P* < 0.05; **, *P* < 0.005; ***, *P* < 0.0005.

tion is necessary for permanent protection against aminoglycoside toxicity. By comparing the measured sulfide concentrations with the degree of protection, we estimated this minimal concentration to be 30 to 50 μ M. Furthermore, it was verified using cyanide that the inhibition of the respiratory chain protects *S. aureus* against gentamicin toxicity like sulfide (data not shown), as reported previously (19–21).

The correlation of the protective effect with sulfide stability shows that indeed sulfide is the protective agent and not the emerging oxidation products. This is in line with results from killing assays performed in lysogeny broth (LB) that had been preincubated with sulfide for 2 h. No sulfide was detected in this medium, showing that the sulfide was fully oxidized, and no protection against gentamicin was observed (data not shown). The observations described above indicated a protective mechanism in addition to those proposed by Shatalin et al. (5), which we aimed to reveal next.

Sulfide is well known to be cytotoxic due to its interactions with heme groups (22, 23), leading to inhibition of cytochromes, which are essential parts of the respiratory chain. It is noteworthy that the uptake of aminoglycosides is dependent on the membrane potential of the bacterial cell (16, 24, 25), which in turn is directly dependent on respiration. Therefore, we hypothesized that the additional protective mechanism of sulfide is based on inhibition of the respiratory chain, ultimately leading to reduced drug uptake. In line with the cytotoxic character of sulfide, reduced growth of *S. aureus* was observed in the presence of sulfide and in the presence of cyanide, which is another known inhibitor of the respiratory chain. Similar growth delays were observed upon addition of sulfide or cyanide (Fig. 3A). After approximately 2 h, however, the growth rate in the presence of sulfide was restored to levels similar to those in the untreated sample, presumably due to the aforementioned oxidation of sulfide.

We next measured the impact of sulfide on the aerobic respiration of *S. aureus*, and we found the oxygen consumption to be 181 ± 19 nmol/min in the presence of sulfide, compared to 217 ± 7 nmol/min in the untreated sample. Cellular ATP concentrations, determined as a measure of the energy status of the cells, were found to be significantly reduced upon addition of cyanide or sulfide (Fig. 3B). To confirm that the reduced ATP concentrations were not due to decreased bacterial cell numbers, CFU in the presence of sulfide or cyanide were determined, and similar values ($6.1 \times 10^7 \pm 0.4 \times 10^7$ and $5.8 \times 10^7 \pm 0.8 \times 10^7$ CFU/ml, respectively) were obtained, in comparison to the untreated sample ($6.5 \times 10^7 \pm 0.4 \times 10^7$ CFU/ml). Therefore, sulfide impairs

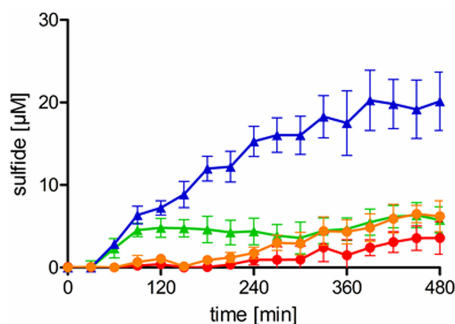


FIG 4 Sulfide production by *S. aureus* and *E. coli*. Sulfide was quantified in the supernatant of *S. aureus* cultures grown in LB supplemented with cysteine (orange line) or cystine (red line) and in *E. coli* cultures supplemented with cysteine (blue line) or cystine (green line). The data are means \pm standard deviations of two independent experiments.

aerobic respiration, leading to reduced energy levels in *S. aureus*, without strongly influencing cell viability.

To confirm that the protection mechanism is based on reduced drug uptake, aminoglycoside uptake by *S. aureus* was directly studied by fluorescence microscopy using bodipy FL-labeled gentamicin. Therefore, we first used cyanide to inhibit the *S. aureus* respiratory chain and found significantly reduced drug uptake (Fig. 3C). This showed that, for the labeled version of gentamicin also, a functional respiratory chain is needed for efficient drug uptake. When cells were pretreated with sulfide, we observed drastically reduced uptake of labeled gentamicin, in comparison to the untreated sample (Fig. 3C). In conclusion, our results show that there is specific protection of *S. aureus* against aminoglycoside antibiotics that is mediated by inhibiting respiration and consequently reducing the energy status of the cell, ultimately affecting drug uptake.

Shatalin and colleagues, in all of their experiments with *S. aureus*, used DL-propargylglycine (PAG) and aminooxyacetate (AOAA), which are known inhibitors of the sulfide-producing enzymes CBS and CSE, respectively (5, 26). Therefore, we studied the impact of these inhibitors on the vitality of *S. aureus*. While no toxic effect of PAG was found at concentrations up to 128 mg/liter, the MIC of AOAA was found to be 16 mg/liter. Checkerboard dilution assays carried out with AOAA and gentamicin or ampicillin revealed no synergistic effects of the inhibitor and the antibiotic, with fractional inhibitory concentration (FIC) index values of 2.15 ± 0.09 and 1.125 ± 0.13 , respectively.

While sulfide was exogenously added in the previous experiments, we finally addressed the question of whether *S. aureus* is capable of producing enough sulfide to protect itself against antibiotics. In line with previous observations (5), sulfide production was observed, using lead acetate paper strips, when *S. aureus* was grown with proper supplementation with cysteine or cystine (data not shown). However, this method displays several drawbacks, i.e., it measures only gaseous sulfide, it is only semiquantitative, and, most importantly, it sums up sulfide production over several hours. Therefore, we measured sulfide production of *S. aureus* HG003 using the methylene blue method and using *E. coli* as the control for sulfide production (Fig. 4). Of note, the sulfide concentration in the *S. aureus* culture never exceeded $6 \mu\text{M}$, regardless of whether cysteine or cystine was added. In contrast, up to $20 \mu\text{M}$ sulfide was detected in the *E. coli* culture grown in LB supplemented with cysteine. Addition of sublethal concentrations of gentamicin did not induce sulfide production (data not shown), as would be expected if sulfide production were an active resistance mechanism against aminoglycosides. These results clearly show that, under the conditions used, *S. aureus* is not capable of producing sulfide at a concentration that leads to protection against aminoglycosides. In line with that finding, the addition of neither cysteine nor cystine to the growth medium protected *S. aureus* from gentamicin-mediated killing (data not shown).

DISCUSSION

Sulfide has gained considerable scientific attention since it has been identified as a third gasotransmitter in mammalian physiology (27–29). Like the other two gasotransmitters, CO and NO, sulfide was shown to affect diverse physiological functions (28). Besides sharing a role in signal transduction, the three gasses are all well known for their toxicity, which in the case of sulfide is mainly based on its tendency to bind to cytochromes and other metalloenzymes and to reduce protein disulfide bridges (13, 30). Sulfide was proposed to be a universal defense against antibiotics in bacteria (5). While different bacteria and representative antibiotics were used in that study, here we used the Gram-positive pathogen *S. aureus*, to systematically and comparatively address the sulfide-mediated protection against the most important classes of antibiotics. We found sulfide-mediated protection against aminoglycosides and observed a clear correlation of sulfide stability and protection, which showed that sulfide is the protective agent and not its oxidation products. This is of importance because it was recently shown that the role of sulfide oxidation has been largely underestimated and some characteristics associated with sulfide should rather be linked to polysulfides (31, 32). The fact that we found protection only against aminoglycosides suggested a mechanism different from the ones postulated previously (5). Those authors proposed universal protection against antibiotics mediated by inhibition of the Fenton reaction via decreases in the intracellular levels of cysteine, iron, and H₂O₂ and the activation of ROS-detoxifying enzymes (5). However, it is well known that a threshold membrane energization is needed for the uptake of gentamicin (24, 25), which was shown here to be impaired by sulfide. This mechanism reflects the situation in *S. aureus* small-colony variants (SCVs), which are resistant to aminoglycosides. SCVs are subpopulations of *S. aureus* that grow slowly because of, for example, a defect in menadione or heme biosynthesis. Therefore, essential components of the respiratory chain are missing, leading to reduced membrane potential and ultimately reduced drug uptake (33, 34).

The different observations made in this study and by Shatalin and colleagues (5) with regard to *S. aureus* may be explained by the fact that all of their findings for *S. aureus* were based on the use of PAG and AOAA. These compounds are known inhibitors of the sulfide-producing enzymes CBS and CSE. Despite being widely used to inhibit CBS, AOAA is well known to be a nonspecific inhibitor of aminotransferases and other pyridoxal 5'-phosphate-dependent enzymes, as it blocks enzyme activity by covalently binding to the cofactor (35, 36); therefore, it is not specific for CBS. Most importantly, we observed that *S. aureus* does not produce sulfide at substantial levels, which was already shown earlier (37). Therefore, the physiological effects of PAG and AOAA in *S. aureus* cells cannot be attributed to a defect in sulfide production. In addition, checkerboard dilution assays carried out with inhibitors and antibiotics showed that there were no synergistic effects of these compounds. Therefore, it can be assumed that the increased antibiotic susceptibility of *S. aureus* in the presence of the inhibitors (5) is due to the addition of the individual toxic effects.

Despite the fact that *S. aureus* is not capable of producing substantial amounts of sulfide endogenously, the sulfide-mediated protection against aminoglycosides, which are often used as adjunctive therapy to treat staphylococcal infections, may be of clinical importance. This is because we have estimated that a minimal concentration of 30 to 50 μM is needed to fully protect *S. aureus*. Most importantly, the sulfide concentration in the blood of healthy humans was found to be in the range of 30 to 60 μM (38), although levels as high as 100 μM have been reported (39). It is thought that sulfide plays an important role in inflammation, and it was shown in a mouse model that the administration of bacterial lipopolysaccharides increased plasma sulfide concentrations significantly (40). In line with that finding, increased sulfide levels (150 μM) have been measured in the blood of patients suffering from septic shock (40). Moreover, a recent study reported high concentrations of sulfide in the sputum of

children suffering from cystic fibrosis (CF) (41). Those authors reported sulfide production rates of $0.3 \mu\text{M}/\text{min}$ and final concentrations of up to $300 \mu\text{M}$ *in vitro*. Importantly, *S. aureus* is the primary respiratory pathogen in young CF patients, and aminoglycosides are routinely used to treat bacterial infections of CF lungs. Of note, aminoglycosides are usually not used as monotherapy to treat infections caused by *S. aureus*. Instead, they are often used in combination with antibiotics targeting cell wall biosynthesis (e.g., β -lactams or vancomycin). Although the precise mechanism of synergy is not fully understood, it has been shown that, for enterococci, a damaged cell wall leads to increased uptake of streptomycin (42). In such a scenario, sulfide-mediated protection is likely abrogated.

Sulfide can freely diffuse across membranes only in its fully protonated form (43). At the physiological pH of blood (pH of ~ 7.36), $\sim 30\%$ of sulfide is present in its fully protonated form, as H_2S (44). Because the intracellular pH of *S. aureus* is ~ 7.8 , it is expected that more sulfide would be deprotonated, leading to accumulation of sulfide within the cell. It must be noted, however, that all *S. aureus* strains encode an enzyme system thought to be responsible for sulfide detoxification (45).

In addition to the effect on aminoglycoside uptake, sulfide might affect bacterial susceptibility to other cationic molecules, such as lantibiotics and in particular host defense peptides, which have also been shown to require a certain threshold membrane potential (46–48). Moreover, the mechanism presented here is most likely not limited to *S. aureus* but also may protect other bacteria from aminoglycoside antibiotics. An identical mechanism has been shown to be the reason for NO-mediated protection of bacteria from aminoglycosides (20). It has been shown that nitrosylation of cytochromes in the respiratory chain of *Salmonella* by NO is responsible for reduced drug uptake and an NO donor protects *S. aureus* from aminoglycoside killing (20). Besides sulfide and NO, the third mammalian gasotransmitter, CO, is known to specifically inhibit cytochromes (13). Therefore, it can be speculated that CO can protect bacteria from aminoglycosides as well.

Together, our results show that the sulfide-mediated protection mechanisms postulated earlier (5) need to be extended by the mechanism described in this study. The fact that we observed protection only against aminoglycoside antibiotics clearly shows that, at least in *S. aureus*, the resistance mechanisms proposed by Shatalin and colleagues (5) can be neglected. Together with the fact that *S. aureus* is not capable of producing substantial amounts of sulfide endogenously, we propose that sulfide production should not be regarded as a universal protection mechanism against antibiotics.

MATERIALS AND METHODS

Bacterial strains. *Staphylococcus aureus* HG003 (49), RN4220 (50), Newman (51), USA300 (52), and SG511 (Robert Koch Institute, Berlin, Germany) and the *Escherichia coli* strains W3110 (53) and ΔsseA (54) were used in this study. Bacteria were cultivated in LB at 37°C unless otherwise stated. Liquid cultures were shaken at 160 rpm.

Chemicals. ATP, PAG, AOAA, sodium sulfide nonahydrate, and sodium hydrosulfide hydrate were purchased from Sigma-Aldrich. Sulfide solutions were freshly prepared before each experiment. Bodipy FL succinimidyl ester and antimicrobial susceptibility disks were purchased from Thermo Fisher. For daptomycin, sterile cellulose disks (Carl Roth) were loaded with $30 \mu\text{g}$ of daptomycin (Cubist Pharmaceuticals). Gentamicin and all other chemicals were purchased from Carl Roth.

Susceptibility testing. MICs were determined by standard broth microdilution, according to the Clinical and Laboratory Standards Institute guidelines (55). To check the relationships between AOAA and antibiotics, checkerboard microdilution assays were performed. FIC indices were calculated as described previously (56), and interactions were defined as synergistic for FIC index values of ≤ 0.5 and nonsynergistic for values between 0.5 and 4.

For modified disk diffusion assays, *S. aureus* strains were grown in MH broth to an optical density at 600 nm (OD_{600}) of 0.5. The cell suspension was diluted 10-fold, and $100 \mu\text{l}$ of the suspension was spread on MH agar plates before antimicrobial susceptibility disks were applied. The plates were placed in polypropylene petri dishes (diameter, 145 mm) containing 35 ml of a cell suspension of either *E. coli* W3110 (providing continuous sulfide production) or the ΔsseA strain (control). For these suspensions, *E. coli* strains were grown in LB to an OD_{600} of 0.5 and diluted 100-fold in fresh LB. In the case of *E. coli* W3110, LB was supplemented with 10 mM L-cysteine as a substrate for sulfide production, which was omitted for the control ΔsseA strain because we found that small amounts of sulfide are formed

nonenzymatically in sterile LB supplemented with cysteine. Plates were incubated for 15 h at 37°C, and the inhibition zones were measured.

For killing kinetic assays, *S. aureus* was grown until an OD₆₀₀ of 0.5 was reached. Cells were treated with 1 mM sodium hydrosulfide hydrate, 0.1 mM sodium cyanide, 1 mM cystine, or 2 mM cysteine, and gentamicin sulfate was added to a final concentration of 25 mg/liter. At different time points, samples were taken, serially diluted in 0.9% NaCl solution, and streaked on LB agar plates. Colony counts were determined after overnight incubation at 37°C. Counts at time zero were set as 100%.

Sulfide quantification. Sulfide levels were quantified with a modified methylene blue assay (57). Samples were taken simultaneously with the killing kinetic assays and centrifuged. To 250 μ l of 2% zinc acetate solution, 395 μ l of the supernatant and 100 μ l of dimethyl-*p*-phenylenediamine chloride (0.2% solution in 20% H₂SO₄) were added. Five microliters of ammonium iron(III) sulfate (10% solution in 2% H₂SO₄) was added, and the solution was incubated for 20 min at room temperature. To determine the sulfide concentration, absorption at 670 nm was measured. Detection of sulfide via lead acetate paper strips was performed as described previously (5).

Labeling of gentamicin with bodipy FL. Labeling of gentamicin with bodipy FL was performed as described previously (58). The sample was mixed with chloroform in a 1:1 ratio and incubated for 1 h. The upper layer (hydrophilic phase), in which gentamicin and fluorescently labeled gentamicin were present, was used for future experiments. Successful labeling of gentamicin was verified by matrix-assisted laser desorption ionization–time of flight (MALDI-TOF) mass spectrometry. The preparations containing fluorescently labeled gentamicin and control preparations (lacking bodipy FL succinimidyl ester) displayed identical MIC values.

Uptake of bodipy FL-labeled gentamicin. *S. aureus* cells were grown to an OD₆₀₀ of 0.5. Bacterial cultures (500 μ l) were incubated for 10 min with either 1 mM freshly prepared sodium hydrosulfide hydrate or 0.1 mM sodium cyanide. The cells were then incubated for 15 min with 20 μ l fluorescently labeled gentamicin. Cells were washed twice with fresh LB, and microscopy was performed as described previously (59). For each experiment, at least 2.5×10^4 cells were analyzed. Photographs were analyzed using ImageJ software, calculating the ratio between the total area of cells (phase-contrast channel) and the integrated density of fluorescent cells (bodipy FL channel).

Determination of cellular ATP concentrations. Cellular ATP concentrations were measured with the BacTiter-Glo microbial cell viability assay (Promega), according to the manufacturer's protocol. *S. aureus* HG003 was grown to an OD₆₀₀ of 0.5, and 1 ml of culture was incubated for 10 min with either 1 mM sodium hydrosulfide hydrate or 0.1 mM sodium cyanide. One hundred microliters of cell suspension was mixed with 100 μ l of BacTiter-Glo reagent, the mixture was incubated for 5 min, and luminescence was measured using a Tecan Spark microplate reader.

Oxygen consumption. Oxygen consumption of *S. aureus* was determined by measuring the oxygen partial pressure using an oxygen electrode connected to an oxygen measurement controller (digital model 20; Rank Brothers). Five milliliters of bacterial culture (OD₆₀₀ of 0.5) was incubated for 1 min at 37°C in the presence or absence of 1 mM sodium hydrosulfide hydrate. The culture was then transferred to the incubation chamber, which was set at 37°C and rinsed with air. The chamber was closed, and oxygen consumption was measured, with constant stirring. Values were plotted, and the slope of the linear part of the graph was used for the calculation of oxygen consumption. Values were normalized to the OD₆₀₀ of the cell suspension.

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We declare that no conflicts of interest exist.

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