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Stereoselective Synthesis of Bicyclo[6.1.0]nonene Precursors of the Bioorthogonal Reagents s-TCO and BCN

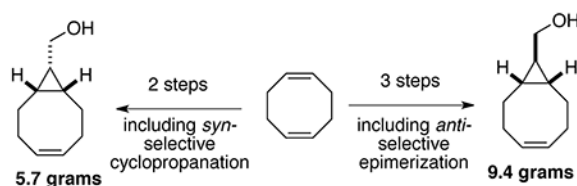
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Abstract

The cyclooctyne BCN and the *trans*-cyclooctene s-TCO are broadly used reagents for bioorthogonal chemistry. A bottleneck for the synthesis of these reagents had been a poorly selective cyclopropanation reaction with ethyl diazoacetate and catalytic $\text{Rh}_2(\text{OAc})_4$. Here, we describe that low catalyst loadings (0.27 mol%) of $\text{Rh}_2(\text{S-BHTL})_4$ provides the BCN precursor with 79:21 *syn:anti* selectivity. The synthesis of the s-TCO precursor was best achieved through a sequence of $\text{Rh}_2(\text{OAc})_4$ (0.33 mol%) catalyzed cyclopropanation, followed by ester hydrolysis under epimerizing conditions. Both sequences could be carried out on multigram scale.

Graphical Abstract



trans-Cyclooctenes and cyclooctynes have emerged as important and broadly useful coupling reagents for bioorthogonal chemistry—unnatural reactions that proceed smoothly in biological context without interfering with native functionality. Ideally, bioorthogonal coupling partners should be stable and nontoxic. Additionally, it is desirable for bioorthogonal labeling to proceed rapidly at the low concentrations that are most relevant to biological study.

The conformationally strained *trans*-cyclooctene 's-TCO' (**1**, Scheme 1) reacts with tetrazines with rates as fast as $k_2 3.3 \times 10^6 \text{ M}^{-1}\text{s}^{-1}$ in water at 25 °C—the fastest bioorthogonal reactions reported to date. s-TCO has been used as a probe compound for labeling in live cells, radiochemistry, and for the creation of patterned hydrogels and biomimetic fibers through interfacial bioorthogonal chemistry. The cyclooctyne BCN (**2**, Scheme 1) reacts rapidly in bioorthogonal reactions with azides ($k_2 2.0\text{--}2.9 \text{ M}^{-1}\text{s}^{-1}$), and also with tetrazines, and has found broad application in bioorthogonal chemistry.

Supporting Information Available:

General Experimental Methods and copies of ^1H NMR and ^{13}C NMR spectra are provided, as is the CIF file for $\text{Rh}_2(\text{S-BHTL})_4$. This material is available free of charge via the Internet at <http://pubs.acs.org>.

s-TCO and BCN are both prepared by short synthetic sequences starting with 1,5-cyclooctadiene as outlined in Scheme 1. A limitation of these syntheses is the poor diastereoselectivity of cyclopropanation, which puts a bottleneck on the scalability of the syntheses. Thus, the $\text{Rh}_2(\text{OAc})_4$ catalyzed reaction of ethyl diazoacetate with 1,5-cyclooctadiene (COD) gives cyclopropanes *syn*-**3** and *anti*-**3** in high yield but only 55:45 *anti*:*syn* selectivity. The *anti*-diastereomer **3** can be reduced to alcohol **4** and applied to the preparation of s-TCO (**1**), whereas the *syn*-diastereomer **3** can be reduced to alcohol **5** and used for preparing BCN (**2**). The *syn*-diastereomer of s-TCO has also been prepared from **5** and applied in ^{18}F PET imaging applications.

Reported herein are diastereoselective syntheses of s-TCO and BCN precursors **4** and **5**, respectively. Key to the synthesis of the BCN precursor was the identification of a *syn*-selective Rh(II)-tetracarboxylate catalyst for the synthesis of *syn*-**3**. For the s-TCO precursor, a 2-pot sequence of saponification/epimerization and LAH reduction gave **4** in high yield and excellent diastereoselectivity. The syntheses of **4** and **5** can be easily carried out on large scale, greatly removing the bottlenecks to the preparation of the s-TCO and BCN.

This study was initiated by screening a family of dirhodium tetracarboxylate catalysts for the cyclopropanation of COD. Screening reactions were carried out by adding a solution of ethyl diazoacetate in COD to dram vials that had been charged with dirhodium catalysts in COD. The results of screening efforts are summarized in Scheme 2. GC-assay yields for the screening reactions varied from 60–97% for all of the catalysts shown in Scheme 2 except for $\text{Rh}_2(\text{S-DOSP})_4$, which gave product in only 25% yield, and $\text{Rh}_2(\text{CAPY})_4$ (not displayed) which gave <1% yield. The primary consideration for these screening efforts was the optimization of diastereoselectivity, not yield, as vial-based screening reactions often proceed in relatively low yield due to the formation of maleate/fumarate products that can be minimized by using controlled syringe-pump addition of the diazo compound in larger scale reactions.

Of the catalysts surveyed (Scheme 2), Hashimoto-type catalysts derived from *tert*-leucine were most effective in promoting *syn*-selective cyclopropanation of COD. As shown in Scheme 2, the most *syn*-selective catalyst is the recently described and very sterically encumbered complex $\text{Rh}_2(\text{S-BHTL})_4$, which gives **3** with 79:21 *syn*:*anti* selectivity. It is plausible that the very bulky BHTL ligands favor *syn*-selectivity by enforcing side-on approach of the alkene to the Rh-carbene. Less selective but commercially available catalysts are $\text{Rh}_2(\text{S-PTTL})_4$ and $\text{Rh}_2(\text{S-NTTL})_4$, which give **3** with 68:32 *syn*:*anti* selectivity. Of the various catalysts that were surveyed, only $\text{Rh}_2(\text{TPA})_4$ gave appreciable levels of selectivity for the *anti*-diastereomer, giving **3** with 31:69 *syn*:*anti* selectivity. Rh_2TPA_4 is known to lead to unique selectivity in cyclopropanation reactions. The high *anti*-selectivity observed in the present study may be due to the ability of the triphenylacetate ligands to enforce the end-on approach of the alkene to the carbene by engaging in attractive substrate-catalyst interactions. With $\text{Rh}_2(\text{S-BHTL})_4$, various solvents were surveyed (CH_2Cl_2 , THF, ethyl acetate, toluene, hexanes) but none gave **3** with appreciably higher *syn*-selectivity. Additionally, running the cyclopropanation catalyzed by $\text{Rh}_2(\text{S-BHTL})_4$ at 0 °C

in neat COD did not improve the selectivity for the *syn*-diastereomer. For large scale synthesis, it was determined that performing the reaction in neat COD was optimal.

The optimized synthesis of *syn*-**3** is shown in Scheme 3. Through syringe pump addition of ethyl diazoacetate in neat COD using 0.27 mol% of $\text{Rh}_2(\text{S-BHTL})_4$, it was possible to obtain **3** with 79:21 *syn:anti* selectivity. After chromatography, *syn*-**3** was obtained in 65% yield. Lithium aluminum hydride reduction of *syn*-**3** gave BCN precursor **5** in 86% yield on a scale that gave 5.7 grams of product (Scheme 3).

For the synthesis of diastereomerically pure **4**, we initially explored the use of *anti*-selective catalyst $\text{Rh}_2(\text{TPA})_4$. Ultimately, it was more efficient to completely avoid separation of diastereomers by saponifying the mixture of *syn*- and *anti*-**3** under epimerizing conditions to produce the *anti*- bicyclo[6.1.0]non-4-ene-9-carboxylic acid **6**. $\text{Rh}_2(\text{OAc})_4$ (0.33 mol%) catalyzed cyclopropanation of ethyl diazoacetate in neat COD gave **3** as a 47:53 *syn/anti* mixture on 80 mmol scale. Previously, Dehmlow and Plückerbaum had shown that *endo-t*-butyl bicyclo[6.1.0]non-4-ene-9-carboxylate could be epimerized during ester hydrolysis. This is possible because *anti*- bicyclo[6.1.0]non-4-ene-9-carboxylic acid **6** acts as a thermodynamic sink, allowing almost total conversion of the diastereomeric ester mixture to the anti isomer. Similarly, we observed that upon treatment with $\text{KO}t\text{Bu}$ in wet ether the *syn/anti* mixture of **3** converged to the diastereomerically pure acid **6** (Scheme 4). Reduction of **6** with LiAlH_4 gave *s*-TCO precursor **4** in 91% yield and on 9.4 gram scale.

In summary, a scalable and more efficient method has been described for the large scale synthesis of the bicyclo[6.1.0]nonene precursors to the bioorthogonal reagents *s*-TCO and BCN. Synthesis of the BCN precursor was enabled by the discovery that $\text{Rh}_2(\text{S-BHTL})_4$ provides useful levels of *syn*-selectivity for the cyclopropanation of 1,5-cyclooctadiene with ethyl diazoacetate. Key to the development of the *s*-TCO precursor was the employment of ester hydrolysis conditions that also epimerize *syn*- and *anti*-**3** to a single *anti*-diastereomer of carboxylic acid **6**.

Experimental Section

General Considerations:

All reactions were conducted under N_2 in flame-dried glassware. Tetrahydrofuran was distilled under nitrogen from a mixture of sodium and benzophenone. Unstabilized OmniSolv diethyl ether was purchased from MilliporeSigma and dried by passing through a column of activated molecular sieves. Ether stabilized with ethanol should *not* be used. Stabilized 1,5-cyclooctadiene was purchased from TCI. All other reagents were purchased from Aldrich or Alfa Aesar and used as received. Compounds were chromatographed via silica gel (ICN SiliTech 32–62D, 60Å). High resolution mass spectra were obtained using a Thermo Q-Exactive Orbitrap high resolution mass spectrometer using a heated electrospray (HESI) source.

***syn*-Selective Synthesis of Ethyl (1R, 8S, 9S, 4Z)-Bicyclo [6.1.0]non-4-ene-9-yl-10-ate (3-*syn*) and Ethyl (1R, 8S, 9R, 4Z)-Bicyclo [6.1.0]non-4-ene-9-yl-10-ate (3-*anti*)**—A dry 500 mL round-bottomed flask was charged with $\text{Rh}_2(\text{S-BHTL})_4$ (295 mg,

0.225 mmol) and a 2 cm magnetic stirbar under N₂. 1,5-Cyclooctadiene (88.2 g, 815 mmol) was added via syringe and the resulting mixture allowed to stir until the catalyst was completely dissolved. The GC standard, dodecane, (14.1 g, 82.7 mmol) was injected. The round-bottomed flask was covered in aluminum foil. A syringe containing ethyl diazoacetate [10.9 g, 10.0 mL, 82.7 mmol (molarity corrected for 13% CH₂Cl₂ content)] was covered in aluminum foil, and the neat diazo compound was added to the solution at a rate of 2.7 mL/hr. The resulting mixture was allowed to stir at rt for an additional 14 h after addition was complete, at which point GC analysis indicated that the ethyl diazoacetate had been completely consumed. Analysis of the crude mixture by gas chromatography indicated a 79:21 *syn:anti* ratio of diastereomers. The crude product was directly loaded onto a chromatography column containing silica gel (39.5 cm tall, with a circumference of 26 cm) using hexane to assist the transfer of material. In a single column, flash chromatography was carried out with pure hexanes until COD had eluted, followed by a gradient to 2% diethyl ether in hexanes to give 10.40 g (53.5 mmol, 65%) of *syn*-**3** and 2.69 g (13.8 mmol, 17%) of *anti*-**3** as clear oils, along with 174 mg of a *syn/anti* mixture. ¹H NMR and ¹³C NMR agreed with the spectra reported in the literature.

(1R, 8S, 9S, 4Z)-bicyclo[6.1.0]non-4-ene-9-ylmethanol (5)—A dry round-bottom flask was charged with 95% lithium aluminum hydride powder (6.10 g, 153 mmol) under nitrogen. THF (200 mL) was transferred to the flask via cannula and the suspension was allowed to chill in an ice bath. *syn*-**3** (8.43 g, 43.4 mmol) was dissolved in THF (40 mL) and slowly added to the solution via syringe. The mixture was allowed to stir for 1 h, at which point the ice bath was removed, and stirring continued for 1 h while the mixture warmed to rt. The flask was again chilled by an ice bath, and a large excess of sodium sulfate decahydrate was added in small portions as a solid in order to quench excess LiAlH₄. Once quenching was complete, as indicated by a thick white suspension, the crude product was vacuum filtered through a bed of celite. The crude was then concentrated onto silica gel and chromatographed (25% diethyl ether in hexanes) to yield 5.70 g (37.4 mmol, 86%) of the title compound as a clear oil. ¹H NMR (CDCl₃, 600 MHz, δ): 5.64–5.63 (m, 2H), 3.71 (dd, *J* = 7.7, 5.3 Hz, 2H), 2.36 (ddt, *J* = 15.5, 7.8, 3.7 Hz, 2H), 2.13–2.07 (m, 2H), 2.02–1.96 (m, 2H), 1.61–1.54 (m, 2H), 1.16–1.09 (m, 2H), 1.03–0.99 (m, 2H). ¹³C NMR (CDCl₃, 150 MHz, δ): 129.9 (dn), 60.4 (up), 27.8 (up), 24.0 (up), 20.9 (dn), 19.1 (dn). IR (CH₂Cl₂, cm⁻¹): 3334, 3006, 2930, 1653, 1022. HRMS-(ESI/Ion Trap) *m/z*. [M-OH]⁺ calcd for C₁₀H₁₅, 135.1170; found, 135.1168.

Unselective Synthesis of Ethyl (1R, 8S, 9S, 4Z)-Bicyclo [6.1.0]non-4-ene-9-yl-10-ate (3-*syn*) and Ethyl (1R, 8S, 9R, 4Z)-Bicyclo [6.1.0]non-4-ene-9-yl-10-ate (3-*anti*)—A 500 mL flame-dried round-bottom flask was charged with Rh₂(OAc)₄ (122 mg, 0.276 mmol) and a magnetic stirbar under N₂. 1,5-Cyclooctadiene (88.2 g, 815 mmol) was added via syringe and the resulting mixture was allowed to stir until the catalyst was completely dissolved. A GC standard, dodecane, (14.1 g, 82.7 mmol) was injected. The round-bottomed flask was covered in aluminum foil. The syringe containing ethyl diazoacetate in 13% CH₂Cl₂ [10.9 g, 10.0 mL, 82.7 mmol (molarity corrected for CH₂Cl₂ content)] was covered in aluminum foil and added to the solution at a rate of 1.5 mL/h, and the resulting mixture was allowed to stir at rt for an additional 1 h after addition was

complete, at which point GC analysis indicated that the ethyl diazoacetate had been completely consumed. Analysis of the crude mixture by gas chromatography indicated a 47:53 *syn:anti* ratio of diastereomers. The crude product was directly loaded onto a column of silica gel (28 cm tall with a circumference of 14.5 cm) using hexane to assist the transfer of material. Without attempting to separate diastereomers, flash chromatography with hexanes followed by a very gradual gradient to 20% diethyl ether in hexanes gave 15.50 g (79.8 mmol, 96%) of isomers of **3**, a clear oil.

(1R, 8S, 9R, 4Z)-Bicyclo[6.1.0]non-4-ene-9-carboxylic acid (6)—A flame-dried round-bottom flask was charged with potassium *tert*-butoxide (22.89 g, 204.0 mmol) and a magnetic stirbar under N₂. Dry diethyl ether (100 mL) was added to the flask via syringe, followed by H₂O (1.5 mL, 83.2 mmol), resulting in a thick white suspension. The *syn/anti* mixture of **3** (13.21 g, 68.0 mmol) was added via syringe, followed by addition of additional diethyl ether (50 mL) by syringe. The mixture was allowed to stir overnight at rt with moderate stirring (18 hours total). The next day, the thick tan suspension was concentrated to dryness by rotary evaporation and re-dissolved in 10% NaOH (15 mL). Diethyl ether (~100 mL) was added and the mixture was transferred to a separatory funnel. The organic layer, which initially contained some suspended solids, was separated and extracted with additional 10% NaOH (3 × 15 mL). After three extractions, all of the solids had completely dissolved. The aqueous layers were combined and chilled in an ice bath, followed by acidification by dropwise addition of 12 M HCl until ~pH 2. The mixture was chilled as a white precipitate formed. The white precipitate was filtered and rinsed with small portions of cold H₂O (2 × 5 mL) and was then recrystallized in methanol and water. The white crystals were filtered and dried, yielding 10.06 g (60.52 mmol, 89%) of the title compound. mp 70–72 °C (lit. 83–85 °C). ¹H NMR (CDCl₃, 600 MHz, δ): 5.64–5.62 (m, 2H), 2.33–2.29 (m, 2H), 2.22–2.18 (m, 2H), 2.10–2.06 (m, 2H), 1.65–1.63 (m, 2H), 1.52–1.48 (m, 2H), 1.20–1.18 (m, 1H). ¹³C NMR (CDCl₃, 150 MHz, δ): 181.2 (up), 130.2 (dn), 29.1 (dn), 28.5 (up), 28.0 (dn), 26.9 (up). IR (CH₂Cl₂, cm⁻¹): 3012, 2934, 1699, 1653. HRMS-(ESI/Ion Trap) *m/z*: [M]⁺ calcd for C₁₀H₁₅O₂, 167.1066; found, 167.1067.

(1R, 8S, 9R, 4Z)-Bicyclo[6.1.0]non-4-ene-9-ylmethanol (4)—A flame-dried round-bottom flask was charged with 95% lithium aluminum hydride powder (9.54 g, 239 mmol) under nitrogen. THF (400 mL) was transferred via cannula and the suspension was allowed to chill in an ice bath. Compound **6** (11.24 g, 67.62 mmol) was dissolved in THF (50 mL) and slowly to the solution via syringe. The mixture was allowed to stir for 1 h, at which point the ice bath was removed, and stirring continued for 2 h while the mixture warmed to rt. While chilling the solution in an ice bath, a large excess of sodium sulfate decahydrate was added in small portions as a solid in order to quench excess LiAlH₄. Once quenching was complete, as indicated by a thick white paste, the crude product was vacuum filtered through a bed of celite. The crude was then concentrated onto silica gel and chromatographed (25% diethyl ether in hexanes) to yield 9.40 g (61.7 mmol, 91%) of the title compound as a clear oil. ¹H NMR and ¹³C NMR agreed with that reported in the literature.

Synthesis of (S)-BHTL

To a dry round-bottom flask was added (*L*)-*tert*-leucine (1.13g, 8.6 mmol) and *cis*-5-norbornene-endo-2,3-dicarboxylic anhydride (1.41g, 8.6 mmol) under N₂. Triethylamine (0.15 mL) and toluene were added via syringe and the mixture fitted with a Dean-Stark trap and heated to reflux overnight. The toluene was then removed *in vacuo* and the residue was partitioned between 5% hydrochloric acid and ether. The aqueous layer was further extracted with ether (x3). The organic layers were combined, washed with brine, dried, and concentrated. The crude solid was recrystallized in methylene chloride and hexanes to provide 1.64 g (5.90 mmol, 69% yield) of the title compound as a white solid. mp 183 °C. ¹H NMR (CDCl₃, 600 MHz, δ): 12.54 (s, 1H), 6.09–6.05 (app d, 2H), 4.09 (s, 1H), 3.47–3.28 (m, 4H), 1.59–1.55 (m, 2H), 0.96 (s, 9H). ¹³C NMR (CDCl₃, 150 MHz, δ): 177.1 (up), 168.3 (up), 135.0 (dn), 134.4 (dn), 59.1 (dn), 52.0 (up), 44.7 (dn), 34.8 (up), 27.6 (dn). IR (CH₂Cl₂, cm⁻¹): 2991, 2966, 1771, 1741, 1653. HRMS-(ESI/Ion Trap) *m/z*: [M]⁺ *m/z*, calcd for C₁₅H₂₀O₄N, 278.1387; found, 278.1387.

Synthesis of Rh₂(S-BHTL)₄

A round bottom flask was charged with rhodium acetate dimer (0.35 g, 0.79 mmol), (*S*)-BHTL (1.10 g, 3.97 mmol) and chlorobenzene. The flask was fitted with a dropping funnel that was stoppered with glass wool and filled with sodium carbonate. The dropping funnel was fitted with a reflux condenser and the mixture was heated to reflux overnight. The dropping funnel and reflux condenser was removed and chlorobenzene was removed by distillation. The flask was cooled and the residue was dissolved in methylene chloride. The solution was washed with aqueous sodium bicarbonate (x3) and then brine. The organic layer was dried, concentrated, and recrystallized in methylene chloride and hexanes, yielding 0.95 g of the title compound as a light green solid (0.72 mmol, 92%). mp > 250 °C. ¹H NMR (CDCl₃, 600 MHz, 360K, δ): 6.07 (app s, 2H), 4.24 (s, 1H), 3.27–3.15 (m, 4H), 2.98 (s, 1H), 1.57–1.51 (m, 2H), 0.86 (s, 9H). ¹³C NMR (CDCl₃, 150 MHz, δ): 187.1 (up), 176.9 (up), 176.2 (up), 170.4 (up), 135.4 (dn), 134.6 (dn), 61.6 (dn), 59.8 (up), 52.2 (up), 44.9 (dn), 34.9 (up), 28.0 (dn), 20.8 (dn), 14.1(dn). IR (CH₂Cl₂, cm⁻¹): 2963.2, 1772.3, 1700.9, 1652.9 HRMS-(ESI/Ion Trap) *m/z*: [M]⁺ calcd for C₆₀H₇₃O₁₆N₄Rh₂, 1311.3131; found, 1311.3126.

Catalyst screening Experiments

A 1-dram vial with septum cap and a stir bar was purged with nitrogen using a needle inlet and outlet. A solution of rhodium catalyst (0.3 mg) dissolved in 1,5-cyclooctadiene (264 mg, 0.3 ml, 2.43 mmol) was added to the vial, and the mixture was allowed to stir. Separately, a stock solution of ethyl diazoacetate [550 mg, 0.51 mL, 4.2 mmol (molarity corrected for CH₂Cl₂ content)], dodecane (715 mg, 0.95 mL, 4.2 mmol), and 1,5-cyclooctadiene (3.52 g, 32.5 mmol) was prepared. To the reaction vial, 0.4 mL of the stock solution was added in one injection. The resulting mixture was allowed to stir overnight, followed by assay by GC.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

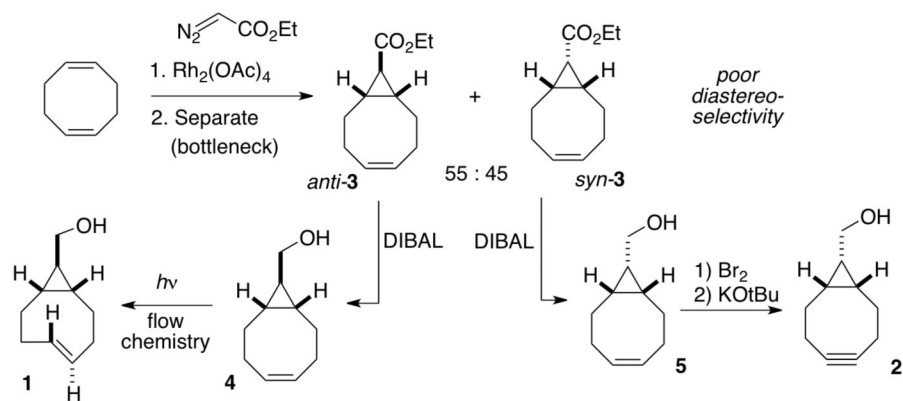
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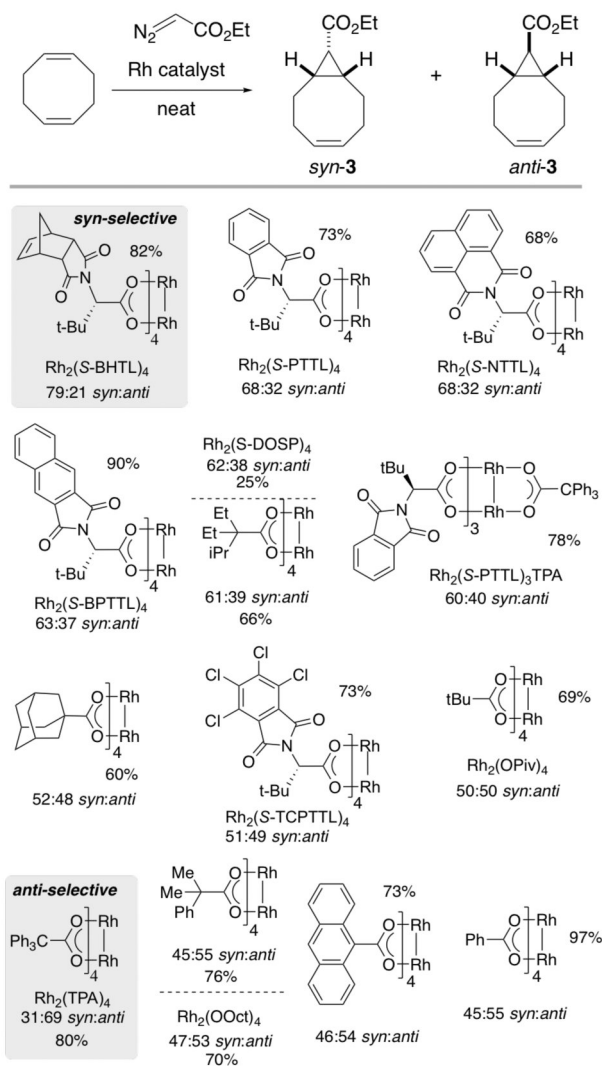
REFERENCES

1. (a)Sletten EM; Bertozzi CR *Angew. Chem. Int. Ed.*, 2009, 48, 6974.(b)Patterson DM; Nazarova LA; Prescher JA *ACS Chem. Biol.*, 2014, 9, 592. [PubMed: 24437719] (c)Lang K; Chin JW, *ACS Chem. Biol.*, 2014, 9, 16. [PubMed: 24432752] (d)McKay CS; Finn MG *Chem. Biol.*, 2014, 21, 1075. [PubMed: 25237856] (e)Ramil CP; Lin Q *Chem. Commun.*, 2013, 49, 11007.(f)Rossin R; Robillard MS *Curr. Opin. Chem. Biol.*, 2014, 21, 161. [PubMed: 25159021] (g)Meyer J-P; Adumbeau P; Lewis JS; Zeglis BM *Bioconj. Chem.*, 2016, 27, 2791.(h)Vrabel M; Carell T *Cycloadditions in Bioorthogonal Chemistry*, Springer International Publishing, Switzerland, 2016.
2. (a)Taylor MT; Blackman ML; Dmitrenko O; Fox JM *J. Am. Chem. Soc.* 2011, 133, 9646–9649. [PubMed: 21599005] (b)Darko A; Wallace S; Dmitrenko O; Machovina MM; Mehl RA; Chin JW; Fox JM *Chem. Sci.* 2014, 5, 3770. [PubMed: 26113970]
3. (a)Blizzard RJ; Backus DR; Brown W; Bazewicz CG; Li Y; Mehl RA *J. Am. Chem. Soc.* 2015, 137, 10044. [PubMed: 26237426] (b)Seitchik JL; Peeler JC; Taylor MT; Blackman ML; Rhoads TW; Cooley TB; Refakis C; Fox JM; Mehl RA *J. Am. Chem. Soc.* 2012, 134, 2898. [PubMed: 22283158] (c)Murrey HE; Judkins JC; Am Ende CW; Ballard TE; Fang Y; Riccardi K; Di L; Guilmette ER; Schwartz JW; Fox JM; Johnson DS *J. Am. Chem. Soc.* 2015, 137, 11461. [PubMed: 26270632] (d)Siegl SJ; Dzijak R; Vázquez A; Pohl R; Vrabel M *Chem. Sci.* 2017, 8, 3593. [PubMed: 30155204]
4. (a)Denk C; Svatunek D; Filip T; Wanek T; Lumpi D; Fröhlich J; Kuntner C; Mikula H *Angew. Chem. Int. Ed.* 2014, 53, 9655.(b)Wang M; Svatunek D; Rohlfing K; Liu Y; Wang H; Giglio B; Yuan H; Wu Z; Li Z; Fox JM *Theranostics* 2016, 6, 887. [PubMed: 27162558]
5. (a)Zhang H; Dicker KT; Xu X; Jia X; Fox JM *ACS Macro Lett.* 2014, 3, 727. [PubMed: 25177528] (b)Liu S; Zhang H; Remy RA; Deng F; Mackay ME; Fox JM; Jia X *Adv. Mater.* 2015, 27, 2783. [PubMed: 25824805] (c)Zhang H; Trout WS; Liu S; Andrade GA; Hudson DA; Scinto SL; Dicker KT; Li Y; Lazouski N; Rosenthal J; Thorpe C; Jia X; Fox JM *J. Am. Chem. Soc.* 2016, 138, 5978. [PubMed: 27078610]
6. (a)Dommerholt J; Schmidt S; Rinske T; Hendriks LJA; Rutjes FPJT; van Hest JCM; Lefebvre DJ; Friedl P; van Delft FL *Angew. Chem. Int. Ed.* 2010, 49, 9422.(b)Dommerholt J; van Rooijen O; Borrmann A; Guerra CF; Bickelhaupt FM; van Delft FL *Nat. Commun.* 2014, 5, 5378. [PubMed: 25382411] (c)Lang K; Davis L; Wallace S; Mahesh M; Cox DJ; Blackman ML; Fox JM; Chin JW *J. Am. Chem. Soc.* 2012, 134, 10317. [PubMed: 22694658] (d)Plass T; Milles S; Koehler C; Schultz C; Lemke EA *Angew. Chem. Int. Ed.* 2011, 50, 3879.(e)Chen W; Wang D; Dai C; Hamelberg D; Wang B *Chem. Commun.*, 2012, 48, 1737.(f)Jawalekar AM Reubsaet E; Rutjes FPJT; van Delft FL *Chem. Commun.*, 2011, 47, 3198.(g)Kim CH; Axup JY; Dubrovskaya A; Kazane SA; Hutchins BA; Wold ED; Smider VV; Schultz PG *J. Am. Chem. Soc.* 2012, 134, 9918. [PubMed: 22642368] (h)Witte C; Martos V; Rose HM; Reinke S; Klippel S; Schroder L; Hackenberger CP R. *Angew. Chem. Int. Ed.* 2015, 54, 2807.(i)Shelbourne M; Brown T, Jr.; El-Sagheer AH; Brown T *Chem. Commun.*, 2012, 48, 11184.(j)Sherratt AR; Chigrinova M; Mackenzie DA; Rastogi NK; Ouattara MTM; Pezacki AT; Pezacki JP *Bioconj. Chem.* 2016, 27, 1222. [PubMed: 27017898]
7. Adly FG; Gardiner MG; Ghanem A *Chem. Eur. J.* 2016, 22, 3447–3461. [PubMed: 26833989]

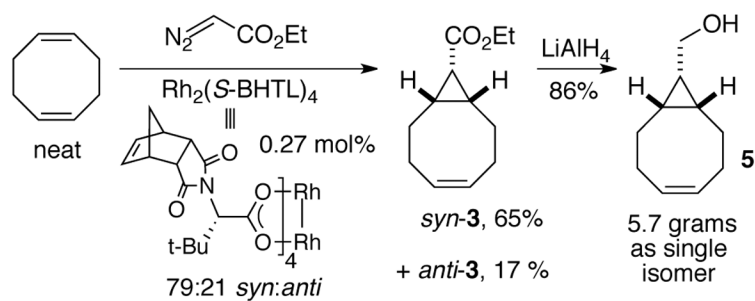
8. Consistent with its sterically congested structure, the ^1H NMR spectrum of $\text{Rh}_2(\text{S-BHTL})_4$ coalesces at 360K. X-ray crystallography confirmed the structure reported in reference 7. See Supporting Information.
9. Bonge HT; Hansen TJ *Org. Chem* 2010, 75, 2309.
10. Panne P; DeAngelis A; Fox JM *Org Lett* 2008, 10, 2987. [PubMed: 18547051]
11. Dehmlow EV; Plückebaum OJ *Prakt. Chem* 1996, 338, 303.
12. Dehmlow EV; Plückebaum OJ *Chem. Res.*, Miniprint 2001, 451.

**Scheme 1.**

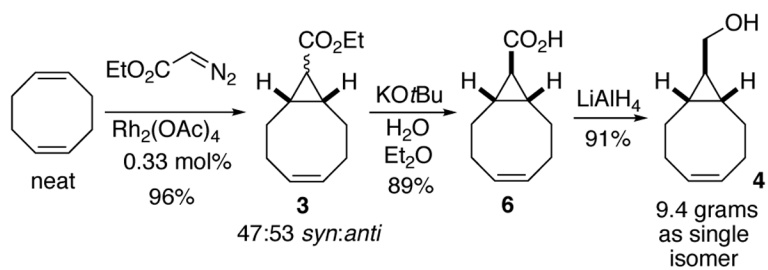
Scalability of published syntheses of s-TCO (1) and BCN (2) are limited by poor diastereoselectivity of the initial cyclopropanation reaction involving 1,5-COD and ethyl diazoacetate

**Scheme 2.**Catalyst Screening for Diastereoselective Cyclopropanation of 1,5-Cyclooctadiene^a

^a Unless otherwise noted, all yields are GC-assay yields measured against a dodecane standard. No effort was made to optimize the yields across the series of catalysts.



Scheme 3.
Large scale synthesis of *syn-3* and BCN precursor **5**



Scheme 4.
Large scale synthesis of s-TCO precursor **4**.