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Unraveling the Molecular Mechanisms of Thermo-responsive Properties of Silk-Elastin-Like Proteins by Integrating Multiscale Modeling and Experiment

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Abstract

Adaptive hydrogels tailor-made from silk-elastin-like proteins (SELPs) possess excellent biocompatibility and biodegradability with properties that are tunable and responsive to multiple simultaneous external stimuli. To unravel the molecular mechanisms of their physical response to external stimuli in tandem with experiments, here we predict and measure the variation in structural properties as a function of temperature through coarse-grained (CG) modeling of individual and crosslinked SE_{8Y} and S₄E_{8Y} molecules, which have ratios of 1:8 and 4:8 of silk to elastin blocks respectively. Extensive structural reshuffling in single SE_{8Y} molecules led to the increased compactness of the structure, whereas S₄E_{8Y} molecules did not experience any significant changes as they already adopted very compact structures at low temperatures. Crosslinking of SE_{8Y} molecules at high concentrations impeded their structural transition at high temperatures that drastically reduced the degree of deswelling through extensive suppression of the structural shuffling and the trapping of the molecules in high potential energy states due to inter-molecular constraints. This integrative experimental and computational understanding of the thermal response in single molecules of SELPs and their crosslinked networks should lead to further improvements in the properties of SELP hydrogels through predictive designs and their wider applications in biomaterials and tissue engineering.

Introduction

Hydrogels are colloidal suspensions of one or more polymers dispersed in fluids such as water. Owing to their high compliance and hydrophilicity, hydrogels have been widely

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Conflicts of interest

There are no conflicts to declare.

applied in biomedical applications such as drug delivery, tissue engineering, and implants.¹ However, significant advances in biomaterial applications demand far greater versatility than traditional static hydrogels manufactured from synthetic polymers. For instance, microneedles derived from silk are capable of imbibing bodily fluids for timed-release of drugs.² These advances have driven the development of nature-inspired hydrogels whose properties are tunable and responsive to multiple simultaneous external stimuli. Stimuli-responsive hydrogels are also increasingly prepared from bio-inspired composites which provide excellent biocompatibility and biodegradability.³

Protein-based biopolymers manufactured through genetic engineering are one such category of stimuli-responsive hydrogels that are designed through bio-inspiration. Hydrogels derived from silk-elastin-like proteins (SELPs) are notable examples. SELPs are derived from tandemly repeating units of elastin-like (GXGVP) and silk-like (GAGAGS) domains where the elastin-like domains are elastic with dynamic features⁴ and the silk-like domains convey mechanical stiffness through the formation of β -sheet secondary structures. An example of the dynamic changes observable in SELPs is the reversible transition in structure, from a soluble protein to a contracted aggregate, around a transition temperature, T_t .⁵ The stimuli-response of SELPs can be tuned further by changing the “X” residue located in the elastin-like domain to allow for sensitivity towards changes in pH, ionic strength, and phosphorylation to just name a few options.⁶ Therefore, the naming convention of SELPs helps to broadly clarify the multitude of possible combinations. For instance, if the S:E ratio is 1:8 and tyrosine (Y) is the “X” amino acid in the elastin block, the resulting molecule is termed as SE_{8Y}.

Numerous experimental studies have detailed SELPs’ macromolecular structures derived from processes such as self-assembly through aggregation,^{7–9} electrospinning,¹⁰ and wet-spinning.¹¹ In contrast, the precise mechanisms behind SELPs’ physical transitions on the molecular level in response to external stimuli have only begun to be unravelled recently through computational methods.^{5, 12} The primary means were through fully atomistic molecular dynamics (FAMD) simulations of SELP molecules with high molecular weight, coupled with advanced large-scale replica exchange methods.¹³ Prior to these numerical studies, only the structural transitions of elastin-like peptides (ELPs) due to temperature have been studied, largely through much smaller scale sampling or low molecular weights using MD.^{14–17} However, even with advanced sampling methods such as Replica Exchange MD (REMD),¹³ analysing the properties of ELPs and SELPs with high molecular weight still requires significant computational resources and the data produced are frequently noisy.^{5, 12, 18} Additionally, although many properties of the SELP hydrogel are tunable, it was shown that the initial concentration of the SELP molecules strongly influences the swelling ratio of the corresponding hydrogel.⁵ For example, increasing the initial concentration of SE_{8Y} from 2% to 10% results in a significant decrease in the degree of swelling by 60%. There is no clear understanding of why this phenomenon occurs, particularly the sub-nanoscale mechanisms for the aggregation of SELP molecules, as the concentration increases.

Therefore, this study addressed two significant problems. First, we showed that coarse-grained MD (CGMD) of individual SELP molecules, SE_{8Y} with 644 residues and S₄E_{8Y}

with 576 residues, enabled the rapid characterization of their thermal response due to potential energy landscapes that were significantly less rugged. There was a clear transition temperature beyond which the SE_{8Y} molecule collapsed into a tightly-packed globular structure from an extended state, while the absence of a transition in S₄E_{8Y} was captured as well. This CG model was validated against both experimental and FAMD studies with excellent agreement. Second, through the combination of both experimental studies and the efficient conformational sampling of this CG model, we provided molecular evidence that the clustering and crosslinking of SELP molecules impeded their structural transition at high temperatures, leading to drastic reductions in the degree of deswelling. The molecules were trapped in states with high potential energy due to this aggregation. This understanding of the thermal response in single molecules of SELPs and their crosslinked networks should lead to further improvements in the properties of SELP hydrogels through predictive design and their wider applications in biomaterials and tissue engineering.

Results and Discussions

Multiscale Characterization of SELPs' Structural Transition

To characterize the thermal response of single molecular SELPs, SE_{8Y} and S₄E_{8Y} SELPs were synthesized with seamless cloning strategies and purified through inverse temperature transition cycling as described previously^{7, 19} and outlined in the experimental section. Transitions in these SELPs were determined through differential scanning calorimetry (DSC). The DSC data (Fig. 1a) showed a clear structural transition temperature, T_b at 294 K for SE_{8Y}, characterized by the endotherm during heating and the exotherm during cooling. In contrast, there were no such transitions for S₄E_{8Y} (Fig. 1b) as neither an endotherm nor exotherm were present. The structural transition in SE_{8Y} was also evident by determining the hydrodynamic radius, R_H , with dynamics light scattering (DLS) at 277 K and 330 K (Fig. 1c). On the one hand, there was a clear reduction in R_H of ~70% beyond the structural transition temperature of the single SE_{8Y} molecule, indicating that it underwent a structural collapse. On the other hand, the R_H of S₄E_{8Y} did not show a clear reduction (Fig. 1d), which further confirmed the absence of any structural transition at high temperatures. Peaks that were larger than 10 nm only indicated that some SELP proteins had aggregated.

In order to understand the molecular mechanisms underlying these differences in structural response to changes in temperature, the structural transition characteristics of the SELPs were characterized through FAMD and CGMD across the temperature range of 277 K to 330 K, as outlined in the experimental section and the Supporting Information.

The structural collapse beyond the transition temperature observed experimentally was also identified in both FAMD and CGMD, where the SE_{8Y} molecule became significantly more compact at high temperatures (Fig. 2a and b) compared to the considerably more extended structure at 277 K. This compactness was characterized by two fundamental metrics of protein geometry: the radius of gyration (R_g) and the solvent-accessible surface area (SASA) of the SELPs. From the FAMD results (Fig. 2b), the SE_{8Y} molecule showed a rather gradual decrease of less 10% in the molecule's R_g across the simulated temperature range. In contrast, the R_g and the SASA (Fig. 2c and d) determined by CGMD simulations predicted a clearly distinguishable structural transition at a transition temperature, T_b of 296 K. This T_b

is in remarkable agreement with our experimental value of 294 K from DSC. Between the temperatures of 277 K and 330 K, the difference in R_g was more than 20%, while the SASA was reduced by almost 10%.

Moreover, this transition was suppressed for S_4E_{8Y} (Fig. 2c and d) with no distinct trends in either the R_g or the SASA across the entire temperature range. This result correlates with the absence of a structural transition indicated by our DSC and DLS experimental data. Even at the lowest temperature of 277 K, S_4E_{8Y} already adopted a very compact structure as the R_g was more than 30% smaller than that of SE_{8Y} while the SASA was 20% lower. The fact that the thermo-responsive structural transition was clearly captured in SE_{8Y} with a distinct transition temperature, while the absence of any transitions was also captured for S_4E_{8Y} , clearly demonstrates that the PLUM potential can adequately capture the dynamic properties of SELPs, in excellent agreement with our experimental studies. Therefore, the PLUM model is validated for further studies here on SELP hydrogel crosslinking, as well as for future work on various thermo-responsive properties in other SELP constructs.

The molecular mechanisms of the structural collapse can be observed in the contact maps of the SELP molecules (Fig. 3 and Fig. S3) that illustrate the distance matrix between all possible pairs of α -carbon atoms in all residues. For SE_{8Y} , the contact maps showed a gradual increase in the number of contacts made between the α -carbon atoms in each amino acid and α -carbon atoms at further neighbours (Fig. 3a and b). This was characterized by the increase in the yellow populated region away from the diagonal of the matrix. This implied that any particular residue i within the SELP molecule established increasing numbers of contacts with residues that were more than ± 3 neighbours away (i.e. more than 3 residues away from i) as the temperature increased. This structural reshuffling led to the increasing compactness of the structure, thereby reducing the R_g and the SASA in SE_{8Y} . In contrast, the compact structure of S_4E_{8Y} was a result of the large number of contact with far neighbours even at the lowest temperature of 277 K (Fig. 3c). This is the reason for the low initial R_g and SASA of S_4E_{8Y} . Moreover, structural reshuffling at the higher temperature of 330 K only resulted in a decrease in number of contacts with far neighbours while increasing near neighbour contacts as indicated by a localized redistribution of the yellow regions (Fig. 3d), thereby accounting for the lack of a clear structural transition from a looser structure to a more compact conformation observed in S_4E_{8Y} . Therefore, these results clearly demonstrate that having higher ratios of silk to elastin blocks greatly reduces the thermal responsiveness that is imbued by the presence of greater amounts of elastin domains with more dynamic structures.

Concentration Dependence of Crosslinked SELPs' Structural Transition

Having characterized the structural dynamics at the level of a single molecule, the SE_{8Y} molecules were subsequently enzymatically crosslinked into hydrogels at 2% concentration with horseradish peroxidase (HRP). As HRP is a tyrosine-specific crosslinker, the phenolic sidechains of tyrosine residues form radicals in the presence of HRP and crosslinks are established when these radicals react with one another.

Heating the hydrogels caused a shrinkage clearly observable when comparing the hydrogel size at 277 K and at 330 K (Fig. 4a, top row). In excellent correlation with the absence of a

structural transition temperature in a single S_4E_{8Y} molecule, the S_4E_{8Y} hydrogel showed no response to changes in the temperature (Fig. 4a, bottom row). More crucially, the deswelling process was heavily influenced by the concentration of the SELP molecules. As shown in Fig. 4b, the SE_{8Y} hydrogel synthesized at a significantly higher concentration of 20% showed a diminished deswelling response in comparison with the hydrogel synthesized from a concentration that was ten times lower. With a concentration of 2%, the deswelling ratio was more than 95%, whereas at a concentration of 20%, this ratio was significantly reduced to only 60%.

To determine the molecular mechanisms behind this suppressed transition, CG crosslinked models consisting of SE_{8Y} molecules were generated and this process is described in detail in the experimental methods. An example of the model is also illustrated in Fig. S2. By performing the simulations in the same temperature range of 277 K to 330 K, suppressed structural transitions were also observed in the crosslinked CG model, in good agreement with the trend observed in our experiments. A single SE_{8Y} molecule showed a reduction in its radius of gyration from 28.4 Å at the temperature of 277 K to 22.0 Å at 330 K, which was a decrease of 23% (Fig. 4b and Fig. S4a). However, the average radius of gyration for ten crosslinked molecules only showed a minimal reduction: from 30.8 Å at 277 K to 28.6 Å at 330 K, which was a reduction of merely 7%. Differences in the thermal response between our experimental and simulation data arise from the fact that the CG models are unable to capture the aggregation of multiple SELP clusters into even larger clusters and the crosslink density cannot be precisely determined and modelled. These aggregations will significantly increase the amount of deswelling as observed in our experimental SELP hydrogel. However, the trends of reduced deswelling with when SELP concentration increased ten-fold are in good agreement.

To understand the discrepancy in the thermal response at varying concentrations, the radius of gyration of each crosslinked molecule was analysed at the two temperatures of 277 K and 330 K (Fig. S4b and c). Almost every molecule experienced inhibition in their thermal response such that the shrinkage was significantly reduced after the T_b , suggesting that the diminished response was a collective phenomenon rather than a response of molecules individually. Above the structural transition temperature corresponding to that of a single SE_{8Y} molecule (i.e. 296 K), the inability of individual molecules to collapse was due to inter-molecular constraints preventing significant conformational changes in the protein backbones.

This was readily apparent from the contact maps at 277 K and 330K (Fig. 5a and b) where there was extensive suppression of the shuffling to increase the number of contacts with far neighbours above the transition temperature, as observed from the lack of increase in yellow regions away from the main diagonal axis. This was in stark contrast to the distinct increase observed in a single, non-crosslinked SE_{8Y} molecule (Fig. 2c and d) as discussed above.

To characterize the extent that inter-molecular constraints influenced structural perturbations, the potential energy of each SE_{8Y} molecule was determined (Fig. 5c) and averaged over all molecules in the crosslinked system. It can be observed that the collective inter-molecular constraints in crosslinked hydrogels caused the molecules to be trapped in a

state of higher average potential energy. Below the transition temperature, the energetic behaviours of both systems were largely similar, showing slight increases compared to the baseline at 277 K. Above the transition temperature, the two curves diverged dramatically, where the increase at the highest temperature of 330 K was 30% and 50% from the baseline for single and crosslinked systems, respectively.

Conclusions

Tunable hydrogels that are responsive to multiple simultaneous external stimuli can be customized from SELPs, with the additional benefits of biocompatibility and biodegradability. Integrative experimental and computational approaches can help to optimize the design of these SELP hydrogels, especially with a bottom-up characterization of their molecular properties. This process can be significantly aided by advanced algorithms for modelling SELPs' nanoscale characteristics. To demonstrate this, we harnessed potential energy landscapes that are less rugged to speedily characterize and uncover the fundamental molecular mechanisms underlying SELPs' sequence- and concentration-dependent physical response to external stimuli. CGMD simulations of individual and crosslinked SE_{8Y} and S₄E_{8Y} molecules were performed together with experimental synthesis and characterizations. Experimental DSC and DLS data showed a distinct structural transition temperature, T_b , at 294 K for single SE_{8Y} molecules, beyond which the hydrodynamics radius, R_H decreased by 70%. Such a transition was not found for S₄E_{8Y} molecules. Similarly, CG modelling of single SE_{8Y} molecules found that the radius of gyration, R_g , and the SASA decreased beyond a T_t of 296 K by 20% and 10% respectively, collapsing into a tightly-packed globular structure from an extended state. The T_t and the corresponding structural changes were in close agreement with our experimental and FAMD data. Good agreement between CGMD and experimental data was also obtained in the observed absence of structural transitions in single S₄E_{8Y} molecules even at high temperatures of 330 K. Contact maps for both SELP sequences showed that structural reshuffling was very extensive in SE_{8Y} molecules, leading to the increased compactness of the structure. In contrast, S₄E_{8Y} molecules did not experience significant changes due to their compact structures even at low temperatures, where the R_g was more than 30% smaller than that of SE_{8Y} while the SASA was 20% lower at the temperature of 277 K. These results extensively validated the ability of the PLUM potential to accurately capture the single molecular temperature transitions of SELPs. Combined experimental studies and coarse-grained modelling of crosslinked SE_{8Y} hydrogels demonstrated that the clustering and crosslinking of SE_{8Y} molecules at high concentrations impeded their structural transition at high temperatures, leading to drastic reductions in the degree of deswelling. Experimentally, the deswelling ratio was more than 95% beyond the T_t at a concentration of 2%, reducing to only 60% at a concentration of 20%. The average R_g for the CG hydrogel model also showed minimal deswelling of 7%. Contact maps showed extensive suppression of the shuffling and did not allow for increased number of contacts with far neighbours above the T_t . By analysing the potential energies, this suppressed transition was attributed to the trapping of the molecules in high potential energy states due to inter-molecular constraints as a result of dense packing. Trends in average potential energy per molecule of both systems were largely similar below T_b , showing slight increases compared to the baseline at

277 K. Above the transition temperature, the potential energy increased by 30% and 50% from the baseline for single and crosslinked systems, respectively, at the highest temperature of 330 K. Through this detailed characterization of the effects of SELP sequences and concentration on the thermo-responsiveness of SELP hydrogels, we demonstrated the strong predictive capabilities that integrative experimental and computational approaches can provide for the design, optimization, and customization of SELP hydrogels for advanced applications in biomaterials and tissue engineering. Furthermore, this multiscale approach can potentially be applied to enhance the development of other protein-based materials.

Materials and methods

Synthesis of SELP polymers

SELP genes and expression plasmids were constructed using our previously established procedures.⁷ The purity of the proteins was monitored via SDS-PAGE, and the molecular weights of the proteins were determined by MALDI-TOF (Bruker Corporation, Billerica, MA).

Preparation of enzymatically cross-linked SELP hydrogels

To obtain crosslinked SELP hydrogels according to previously reported procedures,²⁰ the lyophilized SELP powder was dissolved in deionized water at 4 °C for 4 h to form a SELP stock solution. Horseradish peroxidase (HRP) type VI lyophilized powder (Sigma-Aldrich, St. Louis, MO) was mixed with deionized water to form a 40 mg·mL⁻¹ HRP stock solution with a concentration of 10,000 U·mL⁻¹. In accordance with our previous studies⁵, to fabricate a 10% SELP hydrogel, 6 μL of HRP stock solution was added to 100 μL 10% SELP stock solution, and then the crosslinking reaction of SELP was initiated by adding 0.2 μL of 30 wt% H₂O₂ solution to the SELP and HRP mixture with a final H₂O₂ concentration of 18 × 10⁻³ M. The reaction mixture was mixed by gentle pipetting prior to gelation. The enzymatically cross-linked SELP hydrogels were formed by incubation at 4 °C overnight. To fabricate 2% and 20% SELP hydrogels, the same SELP:HRP:H₂O₂ ratio and crosslinking reaction protocol was used for SELP gelation. An inversion test was used to qualitatively characterize the gelation time for each SELP hydrogel. The formed SELP hydrogels were dialyzed in DI water for 24h to remove the residual, unreacted HRP.

Differential scanning calorimetry

DSC measurements were performed with Nano DSC II Model 6100 (Calorimetry Sciences Corp., Lindon, UT) to capture the inverse transition temperature of SELP hydrogels. The 2% SELP hydrogels were equilibrated at the initial temperature for 10 min, and then heated/cooled in the sample chamber at a rate of 2 °C min⁻¹. The same volume of solvent was placed in the reference chamber during each scan. The baseline scans were taken with the solvent under the same condition and subtracted from the sample scans.

Dynamic light scattering

Dynamic Light Scattering (DLS) was carried out on a DynaPro Titan instrument (Wyatt Technology, Santa Barbara, CA) equipped with a temperature controller. Cuvettes with 1 mm path length were used. All samples were filtered through 0.2 μm Millex[®] syringe filters

(EMD Millipore, Darmstadt, Germany) before measurement. SELP solutions (0.2 mg/mL) were stabilized at each temperature for 10 minutes prior to measurement. To obtain the hydrodynamic radii, the intensity autocorrelation functions were analyzed using the Dynamics software (Wyatt Technology, Santa Barbara, CA).

Swelling and deswelling properties

The swelling ratio of the SELP hydrogels at 4 °C was determined by the ratio of the weight of the hydrogel equilibrated in deionized water to the weight of corresponding as-prepared gel. For deswelling studies, the swollen samples equilibrated in deionized water at 4°C were transferred to 4°C buffer solution, and the deswelling ratio was determined by the ratio of the weight of the hydrogel in buffer to the weight of the hydrogel equilibrated in deionized water. To investigate the stimuli responsive properties, the hydrogels were equilibrated at 4, 10, 15, 22, 37, 50, and 60°C in deionized water and buffers. The equilibrium deswelling ratio upon environmental stimulus is defined as the ratio of the weight of the hydrogel under the specific stimuli to the weight of the hydrogel equilibrated in deionized water at 4°C.

Fully atomistic molecular simulation setup

Both FAMD and CGMD simulations were performed in order to characterize phenomena on different time scales, as laid out in Fig. S5. The FAMD simulations followed our established procedures.^{5, 12, 18} Using CHARMM version 35b1,²¹ extended straight chain conformations of SE₈Y and S₄E₈Y were constructed from elastin and silk blocks. The SE₈Y chain was a 14-mer alternating silk-elastin chain that had the sequence of [(GVGVVP)₄(GYGVVP)(GVGVVP)₃(GAGAGS)]₁₄, while the S₄E₈Y chain had the sequence of [(GVGVVP)₄(GYGVVP)(GVGVVP)₃(GAGAGS)₄]₉. The same SELP sequences were studied in both simulations and experiments. Energy minimization through the steepest descent algorithm was performed to prevent steric clashes.

Subsequently, REMD simulations were carried out in the canonical (NVT) ensemble with the CHARMM19 all-atom energy function.²¹ The implicit solvent was implemented as the EEF1 force field with a Gaussian effective solvent energy function.²² REMD improves conformational sampling by integrating Monte Carlo exchanges into a classical molecular dynamics simulation scheme. This scheme helps the sampled protein to escape local free energy minima in the replicas with high temperatures, thus allowing wider sampling of the conformational space. Full details of the forcefield and the REMD methodology are laid out in the SI. Twenty-four temperature replicas were created and their temperatures were exponentially distributed between 280 and 480 K.²³ Replica exchanges were attempted every 0.5 ps with 2 fs time step to achieve a 15% exchange acceptance rate with the Metropolis criterion. K-means clustering was performed on the ensemble of structures from the last 1,000 exchanges at the lowest temperature replica. Clustering was based on mutual similarity by root-mean-square deviation (RMSD of 3 Å) and performed in the MMTSB tool set.²⁴

The lowest-energy representative structure in the most populated cluster was selected for further refinement with REMD in explicit solvent to correct for local structural approximations. These simulations were implemented in GROMACS version 5.01²⁵ with

the CHARMM27 force field.²⁶ The representative protein was solvated in a rectangular water box with fully periodic boundary conditions. The entire model system contained approximately 200,000 atoms. After energy minimization through the steepest descent algorithm, the solvent was equilibrated while the protein was fixed. This was done in two equilibration stages of 100 ps each with a time step of 1 fs. The NVT ensemble was implemented in the first stage to stabilize the system's temperature. The NPT ensemble was implemented in the second stage to stabilize the system's pressure. Subsequently, the restraints on the protein were removed and equilibrated for an additional 100 ps in the NPT ensemble. The Berendsen thermostat²⁷ and the Parrinello-Rahman barostat²⁸ were used for temperature and pressure coupling, respectively. Covalent bonds with hydrogen atoms were constrained with the LINCS algorithm.²⁹ The short range electrostatic interactions and Lennard-Jones interactions were evaluated with a cutoff of 10 Å. Particle-mesh Ewald summation³⁰ was used to calculate long range electrostatic interactions with a grid spacing of 1.6 Å and a fourth order interpolation.

Finally, REMD was performed with 120 replicas in explicit solvent, starting from representative structures derived from implicit solvent REMD, at temperatures exponentially distributed from 280 K to 400 K.²³ A 2 fs time step was used, each replica was simulated for 20 ns, and exchanges were attempted after 2 ps equilibration runs. Exchange acceptance ratios, based on the Metropolis criterion, were between 20–30%, which signifies adequate sampling. K-means clustering was performed on the ensemble of structures in the final 2 ns to group structures into clusters according to a RMSD of 12 Å for the low-temperature replica at 280 K and for the high-temperature replica at 330 K. Representative structures with the lowest potential energy were chosen from most populated clusters. Analysis of representative structures was carried out using the MMTSB script package.²⁴

Coarse-grain molecular simulation setup

Following the extensive FAMD simulations, the representative structure of a single SELP molecule at 280 K was coarse-grained using the scheme from the PLUM potential.³¹ In this CG model, heavy atoms in each amino acid of the protein backbone are fully represented while their corresponding sidechains are coalesced into a single CG bead. This leads to an intermediate-resolution CG model of four beads per amino acid with implicit water solvent interactions. The full details of the PLUM potential are laid out in the SI. There are multiple benefits of using this approach. First, the rugged potential energy landscape encountered in AAMD is smoothed due to significantly reduced interactions in CG simulations. Conformational sampling is greatly enhanced, where the estimated speedup factor was estimated to be three orders of magnitude.³² For instance, the time to fold the protein (AAQAA)₃ was experimentally determined to be on the order of 100 – 500 ns at room temperature, and yet CG simulations with the PLUM potential were able to achieve the same protein folding from an extended conformation within 500 ps of simulation time.³² Second, computational resources are drastically diminished due to greatly enhanced sampling together with significant reductions in the number of interactions that have to be computed.

Using the PLUM potential implemented in the open-source CGMD software package Extensible Simulation Package for Research on Soft matter (ESPREsSo),^{33, 34} the transition

temperature of this CG model was validated against experimental and FAMD data by simulating a single SELP molecule in the temperature range of 277 to 330 K with a simulation time of 5 ns each. Using a conservative estimate based on the reported speedup factor,³² 5 ns of simulation time corresponds to more than 1 μ s of real sampling time of an already extensively equilibrated SELP structure from FAMD. A time step of 1 fs was used and non-bonded interactions were slowly uncapped over 1 ps to minimize structural perturbations upon starting the simulation.

To simulate dityrosine-crosslinked clusters of SE_{8Y} molecules, tyrosine residues that were exposed on the surface of a single SE_{8Y} molecule were determined first in PyMol.³⁵ A surface exposed residue was defined as being within 0.35 nm from the surface of the molecule. Ten SE_{8Y} molecules were then placed one after another in an iterative fashion using Packmol³⁶ and PyMol³⁵ scripts, such that randomly selected surface tyrosine sidechain beads were able to be packed at less than 10 Å to each other without creating overlaps in any of the other beads (see Fig. S2). These sidechain beads were then crosslinked by the addition of a harmonic spring potential to mimic bonding. The transition temperature of these crosslinked clusters was determined again in the same manner as for a single SE_{8Y} molecule.

Molecular dynamics data analysis tools

For all the FAMD and CGMD simulations above, radius of gyration, solvent accessible surface area, and contact maps were analyzed using the open-source GRONingen MACHine for Chemical Simulations (GROMACS) analysis tools,²⁵ PyMol,³⁵ in-house Bash and Python scripts, and script libraries such as MDTraj.³⁷ All visualization of molecular models was performed using combinations of Visual Molecular Dynamics (VMD),³⁸ PyMol,³⁵ and in-house TCL, Python, Bash, and Matlab scripts.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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References

1. Caló E, Khutoryanskiy VV. *European Polymer Journal*. 2015; 65:252–267.
2. Tsioris K, Raja WK, Pritchard EM, Panilaitis B, Kaplan DL, Omenetto FG. *Advanced Functional Materials*. 2012; 22:330–335.
3. Koetting MC, Peters JT, Steichen SD, Peppas NA. *Materials Science and Engineering: R: Reports*. 2015; 93:1–49. [PubMed: 27134415]
4. Megeed Z, Cappello J, Ghandehari H. *Advanced Drug Delivery Reviews*. 2002; 54:1075–1091. [PubMed: 12384308]

5. Huang W, Tarakanova A, Dinjaski N, Wang Q, Xia X, Chen Y, Wong JY, Buehler MJ, Kaplan DL. *Advanced Functional Materials*. 2016; 26:4113–4123. [PubMed: 28670244]
6. Wang Q, Xia X, Huang W, Lin Y, Xu Q, Kaplan DL. *Advanced Functional Materials*. 2014; 24:4303–4310. [PubMed: 25505375]
7. Xia X-X, Xu Q, Hu X, Qin G, Kaplan DL. *Biomacromolecules*. 2011; 12:3844–3850. [PubMed: 21955178]
8. Lin Y, Xia X, Wang M, Wang Q, An B, Tao H, Xu Q, Omenetto F, Kaplan DL. *Langmuir*. 2014; 30:4406–4414. [PubMed: 24712906]
9. Hwang W, Kim B-H, Dandu R, Cappello J, Ghandehari H, Seog J. *Langmuir*. 2009; 25:12682–12686. [PubMed: 19803470]
10. Ner Y, Stuart JA, Whited G, Sotzing GA. *Polymer*. 2009; 50:5828–5836.
11. Qiu W, Teng W, Cappello J, Wu X. *Biomacromolecules*. 2009; 10:602–608. [PubMed: 19186950]
12. Tarakanova A, Huang W, Qin Z, Kaplan DL, Buehler MJ. *ACS Biomaterials Science & Engineering*. 2017; 3:2889–2899.
13. Sugita Y, Okamoto Y. *Chem Phys Lett*. 1999; 314:141–151.
14. Yeo GC, Tarakanova A, Baldock C, Wise SG, Buehler MJ, Weiss AS. *Science Advances*. 2016; 2:e1501145. [PubMed: 26998516]
15. Li NK, Quiroz FG, Hall CK, Chilkoti A, Yingling YG. *Biomacromolecules*. 2014; 15:3522–3530. [PubMed: 25142785]
16. Li B, Alonso DOV, Daggett V. *J Mol Biol*. 2001; 305:581–592. [PubMed: 11152614]
17. Huang J, Sun C, Mitchell O, Ng N, Wang ZN, Boutis GS. *The Journal of Chemical Physics*. 2012; 136:085101. [PubMed: 22380064]
18. Tarakanova A, Huang W, Weiss AS, Kaplan DL, Buehler MJ. *Biomaterials*. 2017; 127:49–60. [PubMed: 28279921]
19. Huang WW, Tarakanova A, Dinjaski N, Wang Q, Xia XX, Chen Y, Wong JY, Buehler MJ, Kaplan DL. *Adv Funct Mater*. 2016; 26:4113–4123. [PubMed: 28670244]
20. Partlow BP, Hanna CW, Rnjak-Kovacina J, Moreau JE, Applegate MB, Burke KA, Marelli B, Mitropoulos AN, Omenetto FG, Kaplan DL. *Advanced Functional Materials*. 2014; 24:4615–4624. [PubMed: 25395921]
21. Brooks BR, Bruccoleri RE, Olafson BD, States DJ, Swaminathan S, Karplus M. *J Comput Chem*. 1983; 4:187–217.
22. Lazaridis T, Karplus M. *Proteins: Structure, Function, and Bioinformatics*. 1999; 35:133–152.
23. Patriksson A, van der Spoel D. *Physical Chemistry Chemical Physics*. 2008; 10:2073–2077. [PubMed: 18688361]
24. Feig M, Karanicolas J, Brooks CL Iii. *Journal of Molecular Graphics and Modelling*. 2004; 22:377–395. [PubMed: 15099834]
25. Abraham MJ, Murtola T, Schulz R, Páll S, Smith JC, Hess B, Lindahl E. *SoftwareX*. 2015; 1–2:19–25.
26. MacKerell AD, Bashford D, Bellott M, Dunbrack RL, Evanseck JD, Field MJ, Fischer S, Gao J, Guo H, Ha S, Joseph-McCarthy D, Kuchnir L, Kuczera K, Lau FTK, Mattos C, Michnick S, Ngo T, Nguyen DT, Prodhom B, Reiher WE, Roux B, Schlenkrich M, Smith JC, Stote R, Straub J, Watanabe M, Wiórkiewicz-Kuczera J, Yin D, Karplus M. *The Journal of Physical Chemistry B*. 1998; 102:3586–3616. [PubMed: 24889800]
27. Berendsen HJC, Postma JPM, Vangunsteren WF, Dinola A, Haak JR. *J Chem Phys*. 1984; 81:3684–3690.
28. Parrinello M, Rahman A. *J Appl Phys*. 1981; 52:7182–7190.
29. Hess B. *Journal of Chemical Theory and Computation*. 2008; 4:116–122. [PubMed: 26619985]
30. Essmann U, Perera L, Berkowitz ML, Darden T, Lee H, Pedersen LG. *The Journal of Chemical Physics*. 1995; 103:8577–8593.
31. Bereau T, Deserno M. *The Journal of Chemical Physics*. 2009; 130:235106. [PubMed: 19548767]
32. Bereau T. *Unconstrained Structure Formation in Coarse-Grained Protein Simulations*, Thesis. 2011
33. Limbach HJ, Arnold A, Mann BA, Holm C. *Comput Phys Commun*. 2006; 174:704–727.

34. Arnold A, Lenz O, Kesselheim S, Weeber R, Fahrenberger F, Roehm D, Košovan P, Holm C. Meshfree Methods for Partial Differential Equations VI. Griebel M, Schweitzer MA, editors Vol. 89. Springer Berlin Heidelberg: 2013. 1–23. ch. 1
35. DeLano WL. The PyMOL molecular graphics system. 2002. <http://pymol.org>
36. Martínez L, Andrade R, Birgin EG, Martínez JM. J Comput Chem. 2009; 30:2157–2164. [PubMed: 19229944]
37. McGibbon Robert T, Beauchamp Kyle A, Harrigan Matthew P, Klein C, Swails Jason M, Hernández Carlos X, Schwantes Christian R, Wang L-P, Lane Thomas J, Pande Vijay S. Biophys J. 2015; 109:1528–1532. [PubMed: 26488642]
38. Humphrey W, Dalke A, Schulten K. Journal of Molecular Graphics. 1996; 14:33–38. [PubMed: 8744570]

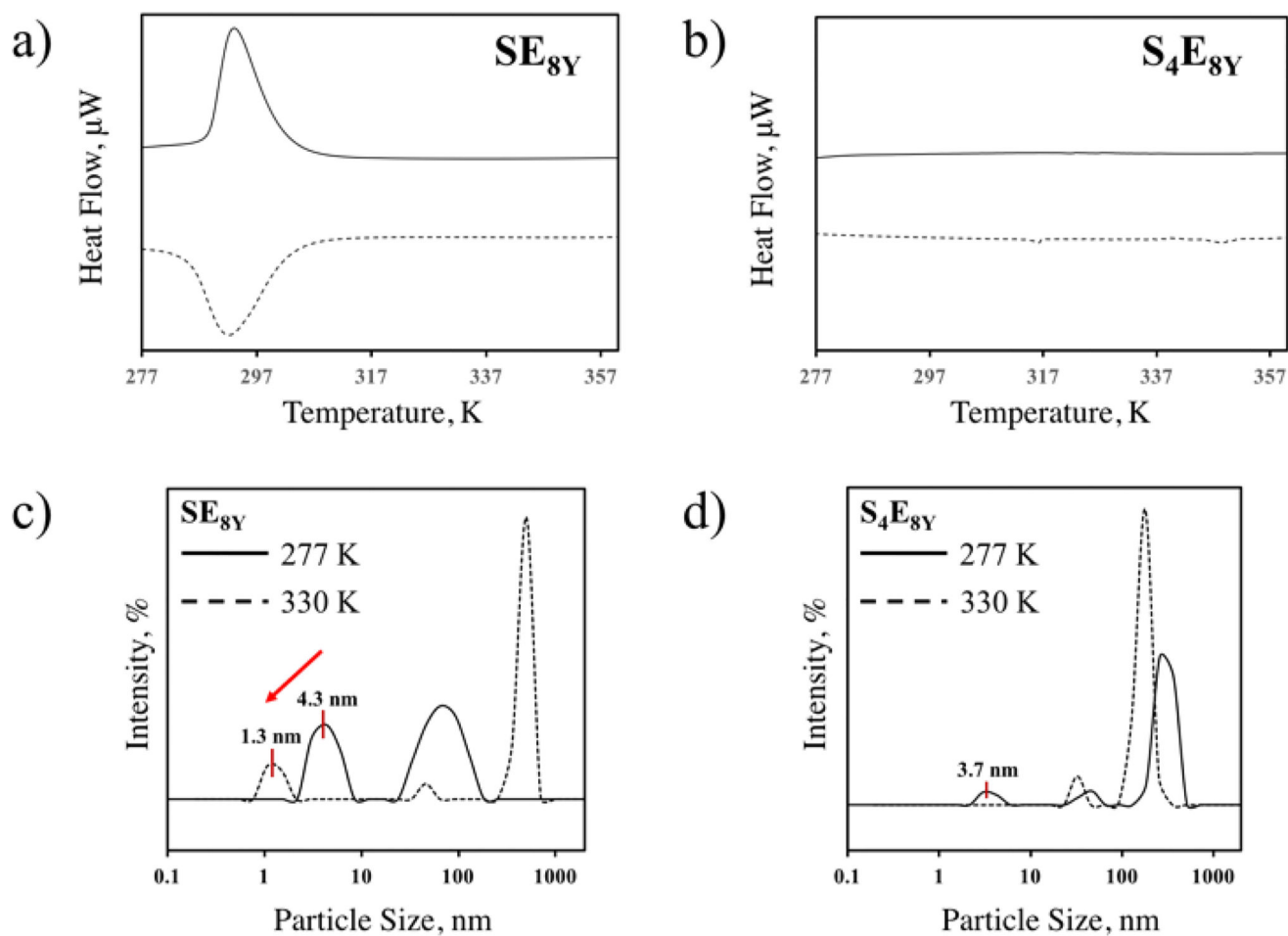


Fig. 1. Comparison of the DSC heating curves for (a) SE_{8Y} and (b) S₄E_{8Y}, showing a clear structural transition at 294 K for SE_{8Y}, whereas there was no distinct transition for S₄E_{8Y}. The hydrodynamic radius from DLS data for (c) SE_{8Y} shows a reduction in size from 4.3 nm to 1.3 nm, whereas (d) S₄E_{8Y} shows no transition. In (c) and (d), peaks that are larger than 10 nm are a result of protein aggregation.

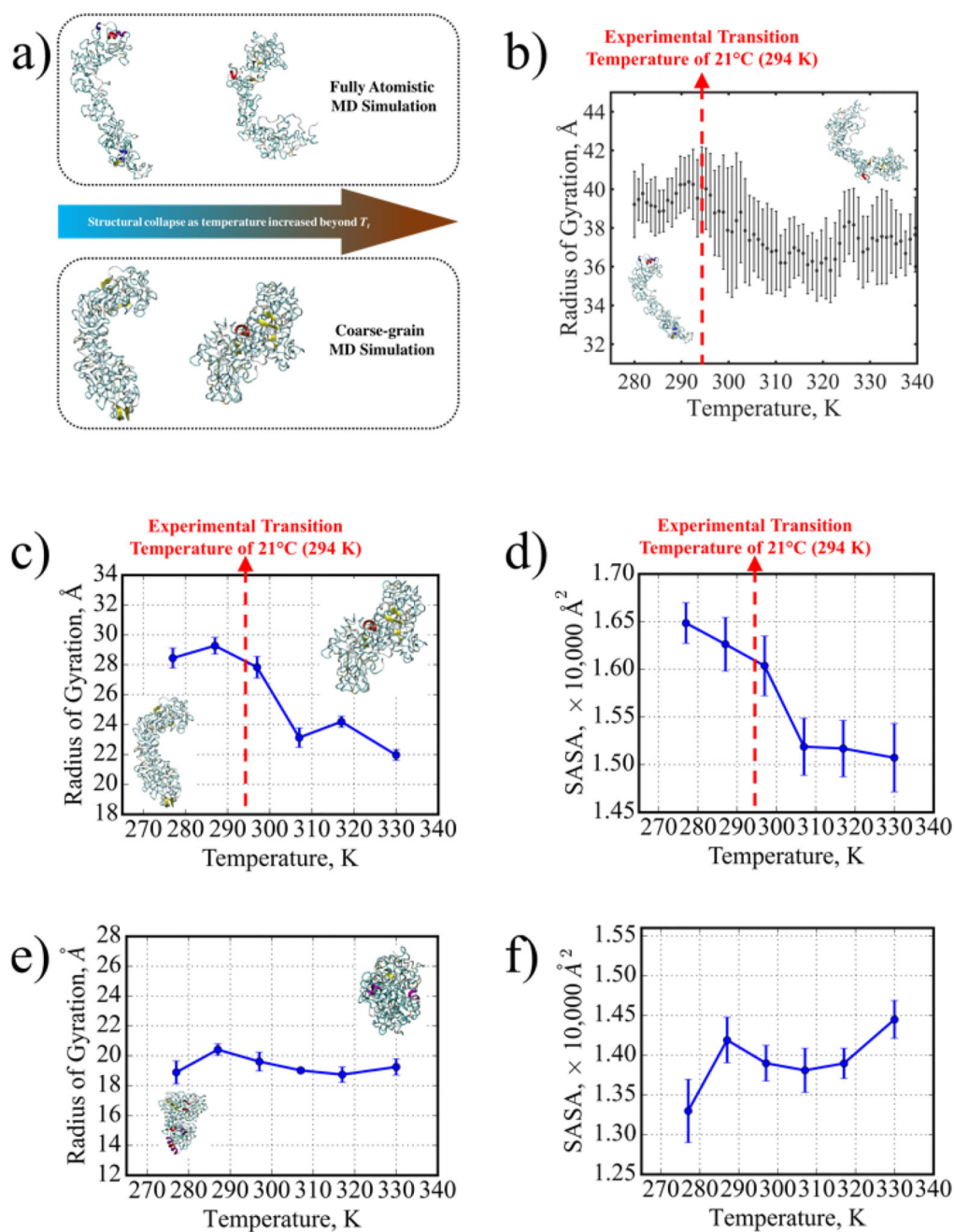


Fig. 2.

(a) A distinct structural transition temperature for SE₈Y was captured by both FAMD and CGMD in terms of (b) the radius of gyration from FAMD and corresponding transitions in (c) the radius of gyration and (d) the SASA from CGMD. This is contrasted with the absence of distinct transitions in S₄E₈Y in terms of the (e) radius of gyration and (f) SASA as well. Insets in both (c) and (e) show the respective SELP molecule's conformations at 277 K (left) and 330 K (right).

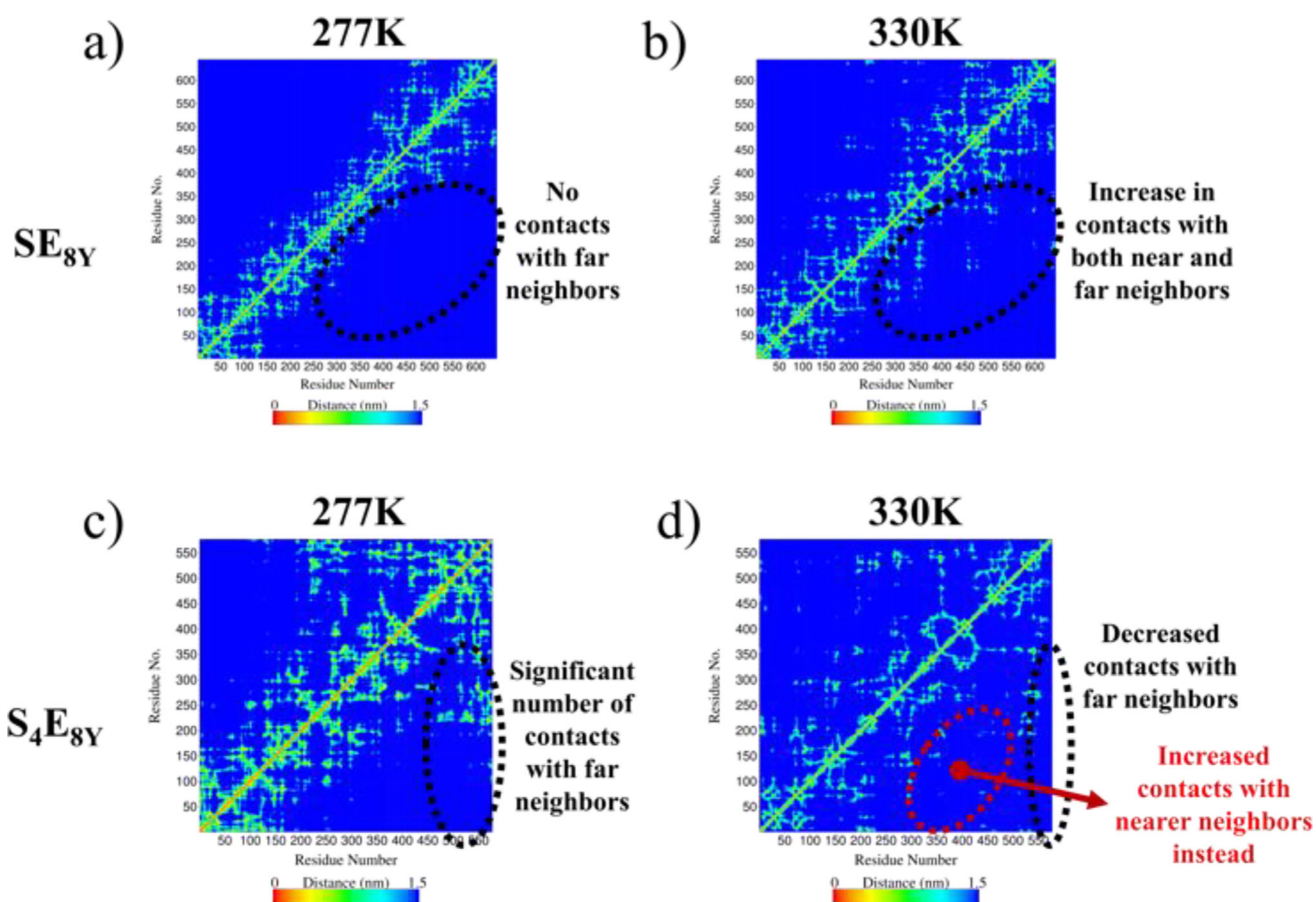


Fig. 3. Contact maps showing the matrices of contact distances between the α -carbon atoms of each residue in (a), (b) SE_{8Y} and (c), (d) S_4E_{8Y} at the temperatures of 277 K and 330 K, respectively.

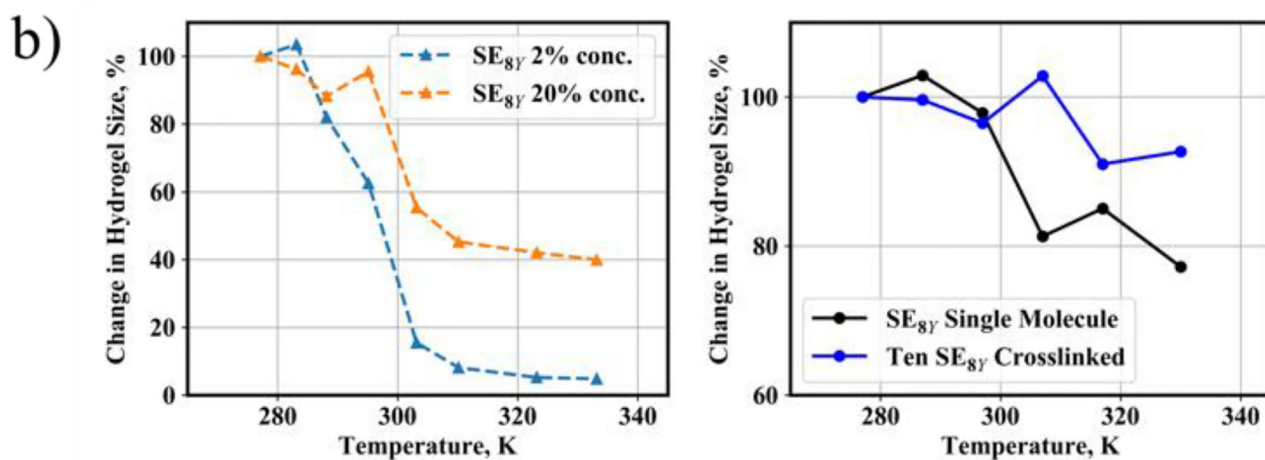
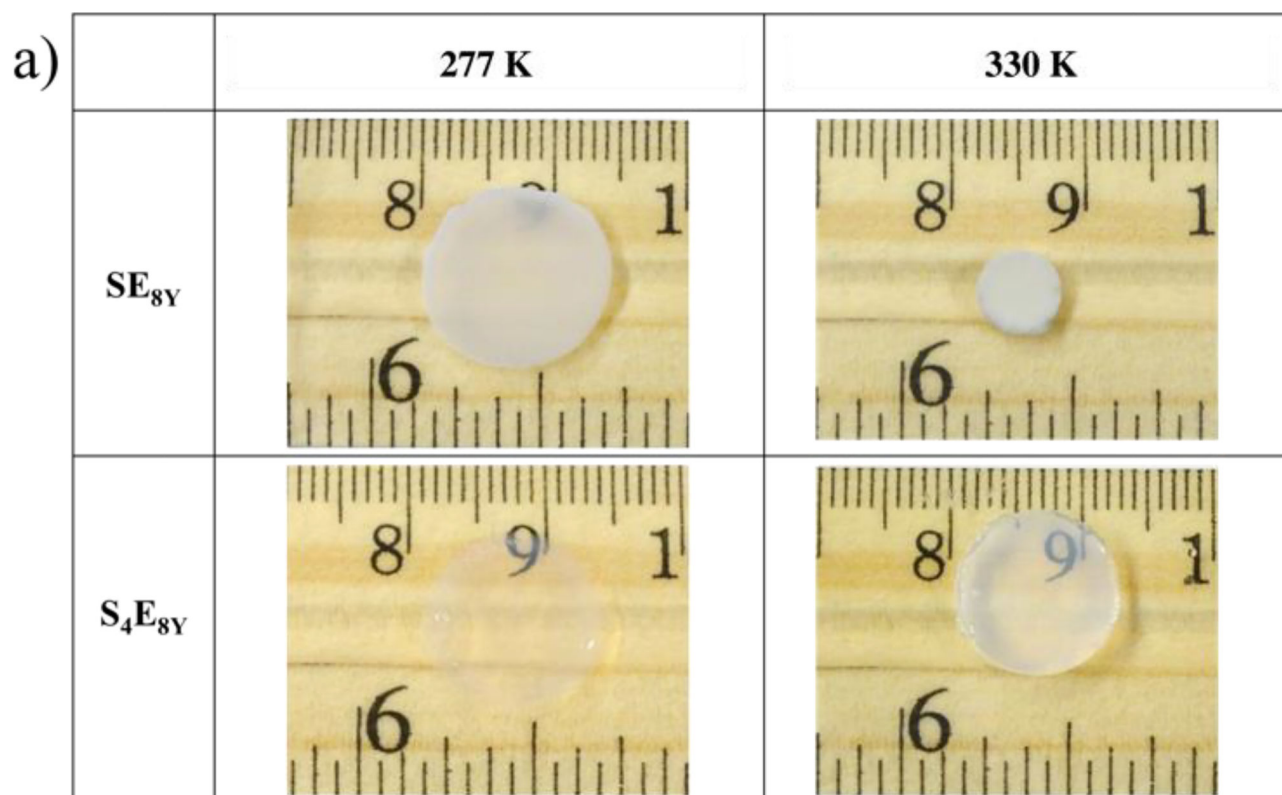


Figure 4.

(a) Deswelling of SE_{8Y} and S_4E_{8Y} hydrogels at 277 K and 330 K. (b) Variations in the size of experimental (left) and simulated (right) hydrogels that have different concentrations of SE_{8Y} as the temperature increased, where the crosslinked CG model showed significantly reduced deswelling above the transition temperature ($P < 0.0001$), in good correlation with the trends from our experiments.

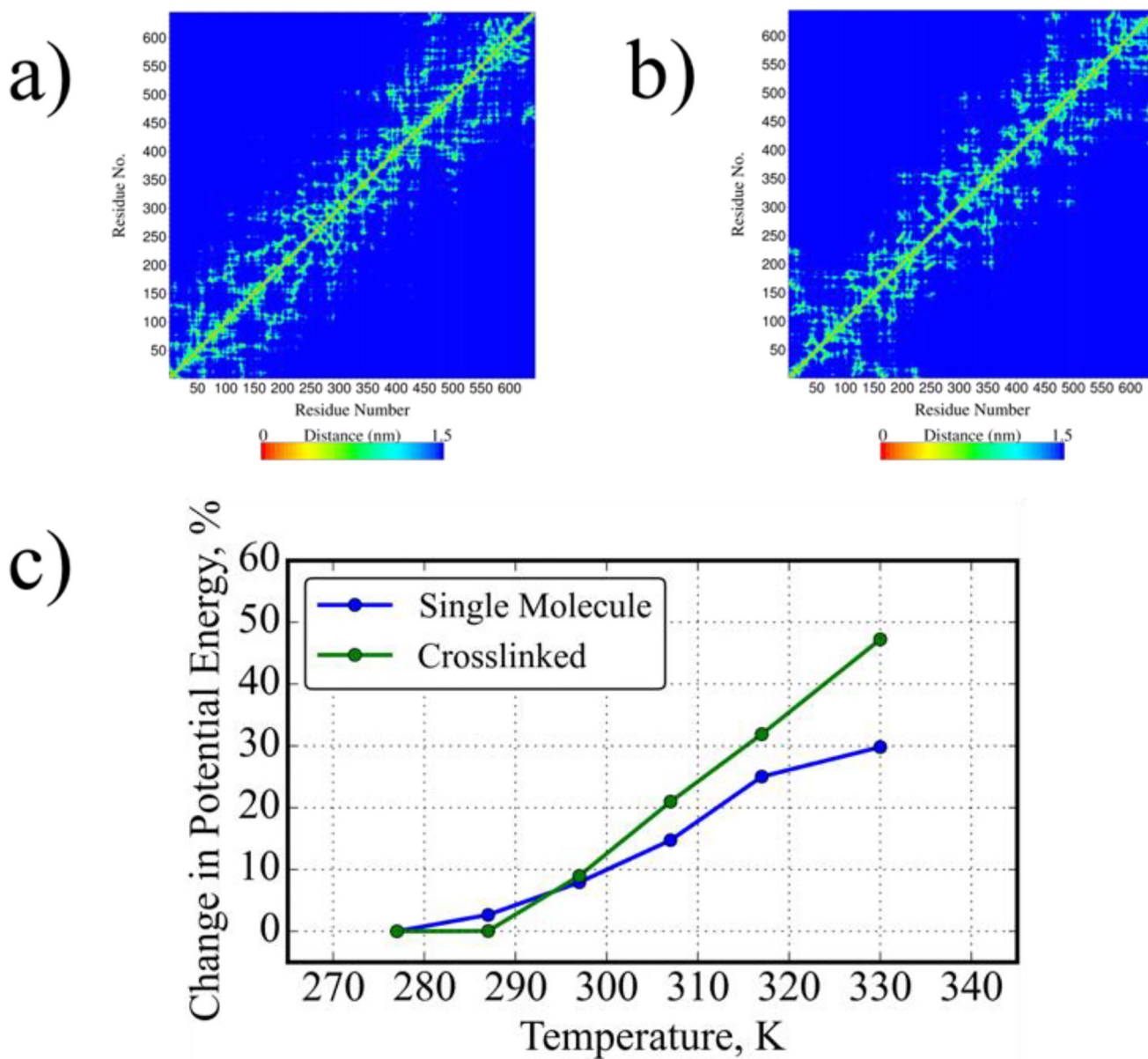


Figure 5. (a), (b) Contact maps showing the matrices of contact distances between the α -carbon atoms of each residue in one of the crosslinked SE_{8Y} molecules at the temperatures of 277 K and 330 K, respectively. Minimal variations in contact maps of the same chain (a) at 277K and (b) at 330K were observed, indicating that the chains had very minor variations in compactness beyond the transition temperature. (c) Change in potential energy of the SE_{8Y} molecules with increasing temperature, where molecular clustering caused each molecule to be trapped in states of much higher potential energy above the transition temperature, hence they were unable to deswell.