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Structure and Dynamics of Membrane Proteins from Solid-State NMR

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Abstract

Solid-state NMR can elucidate membrane protein structures and dynamics in atomic detail to give mechanistic insights. By interrogating membrane proteins in phospholipid bilayers that closely resemble the biological membrane, solid-state NMR spectroscopists have revealed ion conduction mechanisms, substrate transport dynamics, oligomeric interfaces of seven-transmembrane-helix proteins, conformational plasticity that underlies virus-cell membrane fusions by complex protein machineries, and β -sheet folding and assembly by amyloidogenic proteins when bound to lipid membranes. These studies collectively show that membrane proteins exhibit extensive structural plasticity to carry out their functions. Solid-state NMR spectroscopy is ideally suited to elucidating this structural plasticity, local and global conformational dynamics, protein-lipid and protein-ligand interactions, and protonation states of polar residues. These mechanistically important structural studies are aided by new sensitivity enhancement techniques, resolution enhancement by ultrahigh magnetic fields, and the advent of 3D and 4D correlation techniques.

Membrane proteins carry out some of the most important biological functions such as transport of ions and metabolites and transduction of chemical signals and energy. Because of their lipid-associated nature, membrane proteins are difficult to crystallize and solubilize, thus posing significant challenges to structural biologists. So far most high-resolution membrane protein structures have come from X-ray crystallography, and cryo-electron microscopy has begun to contribute membrane protein structures in the 3–4 Å resolution range.

Compared to these techniques, solid-state NMR (SSNMR) spectroscopy provides unique, atomically detailed, dynamically sensitive, and biologically authentic information about membrane protein structure and dynamics. Two fundamental aspects of SSNMR spectroscopy make it especially well suited to membrane protein structural studies. First, the method probes chemical structure as well as angstrom-level three-dimensional structures by the local nature of nuclear spin interactions. The chemical shielding of nuclear spins by the surrounding electrons is sensitive to torsion angles, hydrogen bonding, and molecular packing. Thus, NMR chemical shifts are sensitive to even the most subtle structural changes in proteins. Dipolar couplings between nuclei such as ¹H, ¹³C, ¹⁵N, ³¹P, ²H and ¹⁹F can be measured with ± 0.1 Å precision for distances within 5 Å, and with ± 1 Å precision for

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distances between 5 and 15 Å. These nuclear spin interactions permit SSNMR studies of the active sites of membrane proteins in extraordinary detail. For example, protonation and deprotonation of titratable residues, motions of functionally important sidechains, and coordinated motions of entire domains for ligand transport, can be readily observed using a versatile array of SSNMR experiments. Second, SSNMR studies of membrane proteins is freed from the requirements of crystalline order and fast molecular tumbling, and can probe both rigid and partially dynamic proteins in lipid bilayers. Thus, membrane proteins can be studied in their native lipid environments. A variety of membranes can be used in SSNMR experiments: they can be lipid vesicles with compositions that mimic the biological membrane, whole cellular membranes, or bicelles with simple compositions. Since many membrane proteins exhibit structural plasticity depending on the lipid headgroup charge, cholesterol, membrane curvature, and other membrane properties, it is crucial to investigate membrane protein structures in native-like membranes by SSNMR.

SSNMR instrumentation and radiofrequency pulse sequences have recently seen dramatic advances in four areas: sensitivity enhancement by two orders of magnitude based on dynamic nuclear polarization from electrons (65); sensitivity enhancement by ¹H-detected experiments under ultrafast (~100 kHz) magic-angle spinning (MAS) (4); resolution enhancement by ultrahigh magnetic fields (above 23.5 Tesla, or ¹H Larmor frequency of 1.0 GHz); and resolution enhancement due to the advent of 3D and 4D correlation experiments (82; 85). These advances have allowed larger, multimeric, and more complex membrane proteins to be studied using SSNMR. Here we review SSNMR studies of several classes of membrane proteins that have yielded novel structural, dynamical and mechanistic information.

I. Ion Channels and Transporters

Ion channels conduct ions down their concentration gradients across the otherwise impermeable lipid membrane, using specific residues to control ion selectivity and channel opening and closing (90). Channels can be gated by a variety of stimuli such as membrane potential, pH, ligand, mechanical activity, and temperature (35). On the other hand, transporters move ligands against their electrochemical gradient by coupling conduction to an energetically favorable process, such as photon absorption, ATP hydrolysis, or movement of another ion down its concentration gradient (27). SSNMR has been used fruitfully to study pH-gated proton channels, pH and voltage-gated potassium channels, and multidrug-resistant transporters to reveal key mechanistic information about the ion conduction and substrate transport processes.

The Influenza M2 Proton Channel

The M2 protein of influenza A, B, and C viruses forms functionally homologous but sequence-distinct proton channels that are essential for the flu lifecycle. When a new virus is endocytosed into the host cell, the virus-envelope bound M2 activates in response to the low pH of the endosome and acidifies the virion, which in turn releases the viral ribonucleoprotein complex into the host cell. The wild-type M2 protein of influenza A virus, which is dominant in seasonal and pandemic flu outbreaks, is small (~100 residues) and

modular, and is blocked by the amantadine class of antiviral drugs (6). The N-terminal ectodomain (50) facilitates protein incorporation into the virion (32; 68). The single-pass TM domain contains the proton-selective residue, a single histidine, and the gating residue, a tryptophan, in a conserved HxxxW motif (Fig. 1a). The AM2 TM domain shows similar drug-sensitive and pH-dependent proton conductance as the full-length protein (59), thus making the TM domain a relevant model system for full-length M2. The cytoplasmic domain contains an amphipathic helix (AH) that mediates membrane scission during virus budding (9; 76; 95), and a cytoplasmic tail that participates in matrix protein 1 (M1) recognition and virus assembly (62).

The TM helix conformation is sensitive to the membrane environment—Crystal structures, solid-state NMR, and solution NMR structures (69; 81) of M2 in various membrane-mimetic environments have been determined (36). Crystal structures of the TM domain have progressively improved in resolution from 3.5 Å in octylglucoside (OG) to 1.0 Å in lipid cubic phases (1; 84; 86). SSNMR studies of the TM domain in lipid bilayers have shown that the TM conformation and dynamics are sensitive to the membrane composition, pH, and drug binding. High membrane fluidity and negatively charged lipids increase helicity by promoting the cationic state of His37 (9; 39; 55), while thicker bilayers decrease the helix tilt angle (7; 17; 78). Adding cholesterol increases the helicity (54), immobilizes the four-helix bundle (57), and stabilizes the tetramer (14; 45). Drug binding causes a kink to the middle of the TM helix (8; 42). For a combined TM-AH construct, a solution NMR structure solved in DHPC micelles shows the AH to be separated from the TM helix by an unstructured loop (81), while a SSNMR orientational structure solved in DOPC/DOPE bilayers shows the AH to lie parallel to the membrane surface, connected to the TM helix by a tight turn (83). In comparison, a MAS SSNMR structure solved in DPhPC membranes shows a dimer of dimers structure with two distinct orientations and depths for the TM helices (3). This symmetry breaking, not seen in other structures, may be caused by the DPhPC lipid, whose methyl-rich acyl chains can cause significant negative curvature to the membrane. Taken together, these studies show that the TM and AH domains have significant conformational plasticity in response to the pH and membrane environment (110), thus structural comparisons of different M2 constructs should be made in the same lipid membranes and at the same pH.

The ectodomain and cytoplasmic tail in M2 are dynamically disordered—2D

and 3D correlation experiments on ¹³C, ¹⁵N-labeled full-length AM2 showed that the ectodomain and cytoplasmic tail are unstructured and highly dynamic. Resonances beyond the α -helical chemical shifts of the TM and AH residues show random coil ¹³C chemical shifts, narrow linewidths, and low order parameters, indicating that the two extramembrane domains of M2 are intrinsically disordered (54). Site-specifically labeled ectodomain peptides and cytoplasmic tail peptides were ligated to the central TM-AH domain and confirmed this dynamic disorder (50; 51). Interestingly, despite this dynamic disorder, the extra-membrane domains modulate the structure and function of the TM domain. Chemical shift measurements of His37 and other key TM residues indicate that the ectodomain shifts the TM equilibrium towards the drug-bound conformation even in the absence of drug, while the cytoplasmic tail shifts the protonation equilibria of His37 towards higher pK_a's (50).

Thus, the ectodomain may preselect drug-binding competent conformation of the TM helix, while the cytoplasmic tail facilitates proton binding to His37, likely due to the acidic nature of the cytoplasmic tail.

Proton conduction mechanism and gating mechanism of M2—Solid-state NMR, MD simulations, and crystal structures that resolved water densities have provided rich information about how M2 is activated by low pH and how it conducts protons exclusively from the N-terminus to the C-terminus (72). ¹⁵N SSNMR spectra showed that the His37 imidazole ring shuttles protons into the virion (71) by microsecond-timescale protonation and deprotonation. This is manifested by pH-dependent histidine sidechain ¹⁵N and ¹³C chemical shifts and ¹⁵N chemical shift averaging at high temperature (23; 41; 103). The imidazole rings reorient at acidic pH to mediate proton transfer, as observed from motional averaging of the histidine sidechain dipolar couplings (40). Water-imidazole hydrogen bonding and proton exchange have been detected from water-imidazole ¹H-¹⁵N correlation peaks (37; 63; 100).

The His37 tetrad protonates successively like a tetraprotic acid, with a maximum of four proton-dissociation constants (pKa's). These pKa's have been measured using ¹⁵N chemical shift spectra as a function of pH (Fig. 1b) under various conditions to investigate how proton shuttling depends on the protein structure and membrane environment. Comparison of wild type and mutant channels probed the effects of the gating tryptophan residue on proton shuttling (58; 61; 101; 104). Comparison of TM and longer constructs revealed the influence of extramembrane domains (55). Comparison of influenza A and B M2 probed the effects of the amino acid sequence on proton transfer (102), while comparison of TM peptides bound to different lipid membranes examined the influence of the lipid environment on proton relay (41; 43). These studies all show that the third protonation event, occurring between pH 5 and 6, causes the largest increase of proton conductance and is therefore responsible for channel activation, but the exact values of the pKa's depend on the protein sequence and membrane composition. Cholesterol-containing membranes lower the pK_a's (41). Negatively charged lipids and the acidic cytoplasmic tail increase the pKa's (11; 55), likely by better attracting protons to the membrane. BM2 shows lower pK_a 's than AM2 (102), which may be due to faster proton relay to a second histidine that is C-terminal to the proton-selective histidine. When the gating Trp was mutated to Phe to allow bi-directional proton conductance, the His37 structural equilibrium was perturbed (Fig. 1c) (61). His37 shifted to the π tautomer and the neutral state at high pH, indicating faster deprotonation to the C-terminus, but became more cationic at low pH, indicating increased protonation from the C-terminus. Drug binding to this mutant channel left 50% of His37 cationic, in contrast to the wild-type, which became fully neutral upon drug binding. These results indicate that in wild-type M2, Trp41 blocks C-terminal acid activation of His37, causing asymmetric conductance of the channel (104).

The protonation equilibria of the proton-selective histidine can be described by a kinetic model that takes into account the asymmetric nature of proton shuttling (61):

$$His + H_{out}^{+} \stackrel{k_{on}}{\underset{k'_{off}}{\longrightarrow}} His H^{+} \stackrel{k_{off}}{\underset{k'_{on}}{\longrightarrow}} His + H_{in}^{+}.$$
(1)

Each equilibrium constant, measured from the ¹⁵N intensities of cationic and neutral histidines (Fig. 1b), is related to the rate constants as:

$$\frac{[His][H^+]}{[HisH^+]} \equiv K_a = \frac{k_{off} + k'_{off}}{k_{on} + k'_{on}}.$$
 (2)

This kinetic model explains all SSNMR-detected chemical equilibria of His37 in M2 channels. Since four protonation equilibria lead to five charge states of the histidine tetrad, the structure of each histidine residue is expected to depend on the tetrad charge state. Indeed, seven sets of His37 ¹³C and ¹⁵N chemical shifts have been resolved in cytoplasmic-tail-containing M2 (55) (Fig. 1d), and can be assigned to different charge states of the tetrad (Fig. 1e). Together, these SSNMR data indicate that the structural properties of M2 depend on the pH-sensitive His37 and the membrane composition.

Proton conduction requires histidine-water H-bonding, not imidazole-

imidazolium H-bonding-Strong imidazole-imidazolium hydrogen bonds have been proposed for the histidine tetrad by Cross and coworkers to explain the conduction mechanism. The rationale for this model was to account for the +2 charge of the His37 tetrad near neutral pH (43), as the first histidine pK_a measurement found high pK_a 's (~8.2) for the first two protonation events. It was argued that a low-barrier hydrogen bond between a neutral imidazole and a cationic imidazolium can stabilize the +2 charge in the middle of the membrane. However, the energetic rationale for this model has since been removed by new experimental evidence. First, more accurate measurements of His37 pK_a's in cholesterol-containing membranes that mimic the virus envelope composition (57) found lower pK_a 's (7.6 and 6.8) for the first two protonation events, indicating that the tetrad charge state is lower (about +1) at neutral pH (41; 55; 61; 102). Second, high-resolution crystal structures of the TM peptide showed a cluster of water molecules on both sides of the His37 tetrad, indicating proton delocalization among water molecules (1). So far, only a single ¹⁵N cross peak has been observed for the downfield ¹H chemical shift in 2D ¹H-¹⁵N correlation spectra, indicating that the hydrogen-bonded proton is not shared by two imidazole nitrogens, but by one nitrogen and one oxygen (63; 99). Moreover, at high temperature, the exchange-averaged ¹H chemical shift that correlates with the imidazole ¹⁵N signal lies at the water chemical shift of \sim 5 ppm, indicating that the hydrogen-bonding partner of imidazole is water instead of another histidine (37). Finally, torsion angle measurements (40) and crystal structures (1) showed that the H37 (χ_1, χ_2) angles are about 180°, which prevent the imidazole N-H group from pointing to another imidazole in a geometry to form N-H...N hydrogen bonds.

Potassium Channels

K⁺ ion movement through potassium channels across lipid membrane underlies many biological processes such as electrical signaling in the nervous system and cardiac action potentials. While crystal structures of a number of potassium channels are available (16; 56; 111), SSNMR is well suited to characterizing the conformational dynamics and membranedependent structural variations of K⁺ channels. The prokaryotic K⁺ channel KcsA has been the main model system for SSNMR studies, because it is smaller than mammalian voltagegated potassium (Kv) channels while retaining key ion selectivity and gating properties. A chimera between KcsA and Kv1.3, which grafts 11 residues in the turret region of Kv1.3 onto KcsA (Fig. 2a), has also served as a useful model for understanding toxin binding and lipid interaction of K⁺ channels (80).

Potassium channels can be inactivated by two mechanisms: a fast process involving the intracellular part (N-type inactivation for *Shaker* channels), and a slower process involving the extracellular K⁺ ion selectivity filter (C-type inactivation) (38). KcsA is activated by a high concentration of intracellular protons, which causes protonation of E118 and E120 (87), but also undergoes C-type inactivation, which causes loss of ion binding sites at the extracellular selectivity filter. These two gates together can adopt four states (15), as defined by electrophysiological data, under different combinations of pH and K⁺ concentrations. The two longest-living states are the resting state, which is adopted when the intracellular gate is closed and the extracellular gate is open, and the inactivated state, adopted when the intracellular gate is open can K⁺ ions be conducted. The region near the extracellular inactivation gate has been extensively studied by SSNMR. From the N-terminus to the C-terminus, these segments consist of TM helix 1, which corresponds to the S5 helix of Kv channels, a water-exposed turret, a short pore helix, the selectivity filter, and TM helix 2, which corresponds to the S6 helix of Kv channels (Fig. 2a).

Extensive chemical shift mapping of KcsA using 2D, 3D and 4D correlation SSNMR gave a wealth of information about the conformational equilibria among the four states and the structural coupling between the two gates. Bhate et al compared the acid-induced inactivated state of the gate with low [K⁺]-induced collapsed state of the protein (5) using an E71A mutant, which does not undergo C-type inactivation. Chemical shift data indicate that the mutant cannot access the collapsed state even when [K⁺] is low, suggesting that the acid-induced C-type inactivated state is structurally similar to the low [K⁺]-induced collapsed state. Between low and high K⁺ concentrations, the wild-type protein exhibits chemical shift changes of the E71 sidechain carboxyl, which is involved in a hydrogen-bonding network with water, D80 and W67. The sidechain ¹³CO chemical shift tensors of E71 and D80 (Fig. 2b) were measured from sideband intensities and showed that the E71 carboxyl is protonated even at pH 7.5, indicating an unusually high pK_{ab} while the D80 sidechain is deprotonated. Therefore, the protonation state of E71 is unchanged between high and low [K⁺], but the hydrogen-bonding network between the pore-helix and selectivity filter is altered, likely because of differences in K⁺ ion binding at the selectivity filter.

Chemical shift comparisons between the high $[K^+]$ /neutral pH state, the low $[K^+]$ /neutral pH state, and the high $[K^+]$ /low pH state gave information about structural similarities and

differences among these states (105). Extensive chemical shift changes were observed in the hinge, pH sensor and selectivity filter regions at low $[K^+]$ /neutral pH, when the channel occupies the collapsed state, compared to the high $[K^+]$ /neutral pH state. These structural changes were mirrored at high $[K^+]$ /low pH, indicating that there is allosteric coupling between the two modes of inactivation: the pH-dependent protonation change at E118 and E120 and the $[K^+]$ -dependent C-type inactivation.

In addition to pH and K⁺ concentration, the lipid composition of the membrane affects the KcsA conformational equilibria. Varying the lipid headgroup charge allowed all four gating states to be accessed (92). Cardiolipin and phosphatidylphosphate (PA) were found to increase the probability of the open-conductive state, while neutral lipids and high pH shifted the equilibrium towards one of the three non-conducting conformations. Residues that are diagnostic of these conformational states are found in the turret (e.g. Ala50) and the pore helix (e.g. Glu71). The lipid sensitivity of the protein structure was also examined by comparing the structures of KcsA and the KcsA-Kv1.3 chimera. The chimera contains a charge mutation in the pore helix, where a cationic R64 is changed to an anionic D64. The chimera shows cross peaks between D64 and R89, indicating a salt bridge, but no cross peak with lipids, while KcsA shows Ser/Thr cross peaks with asolectin lipids, indicating lipid binding (97). MD simulations suggest direct protein-lipid interactions that are affected by the electrostatic properties of the turret residues (91). The turret also shows significant structural plasticity. At high $[K^+]$ /low pH, long-range cross peaks such as D53-T85 are lost compared to the high $[K^+]$ /high pH state, indicating that the turret and the neighboring TM1 and TM2 helices undergo conformational changes to cause C-type inactivation.

Transporters

Transporters are chiefly responsible for multidrug resistance of cells, because small, hydrophobic, and positively charged drug molecules can be actively exported out of cells by these proteins in a non-specific manner (34). The guiding transporter hypothesis is that these proteins undergo conformational changes between at least two states, facing inward and outward, to move substrate molecules. EmrE is a small multidrug resistance transporter, and couples the import of two protons with the export of one substrate molecule from the cytoplasm to the periplasm (Fig. 3a). The protein contains four TM helices and forms an antiparallel, asymmetric dimer. TM helices 1–3 in each subunit are responsible for drug and proton antiport, while the fourth TM helix serves as the dimerization unit (Fig. 3b).

Solid-state and solution NMR experiments have described the topological structure, conformational dynamics, and mechanism of action of EmrE. Using oriented-membrane SSNMR experiments, Gayen et al showed that the four TM helices in the two subunits have different orientations relative to the bilayer normal. In magnetically oriented bicelles, each residue exhibits two sets of signals in 2D polarization inversion spin exchange at the magic angle (PISEMA) spectra, which correlate the ¹⁵N-¹H dipolar coupling with ¹⁵N chemical shifts (28) (Fig. 3c). This peak doubling was found for both ligand-free and ligand-bound proteins, indicating that the two subunits are asymmetric. Chemical cross-linking experiments showed that the two subunits are antiparallel. Binding of a substrate, tetraphenylphosphonium (TPP⁺), at a saturating molar ratio of one ligand per dimer affected

the orientation of TM3 in both subunits, indicating that ligand binding changes the protein conformation. It is noteworthy that oriented-membrane SSNMR is especially suited for investigating such asymmetric membrane proteins, since the structural asymmetry is directly encoded in the orientation-dependent NMR frequencies, but is not inherently detectable in isotropic chemical shift spectra that are measured under MAS.

After establishing the asymmetric and antiparallel topology of the EmrE dimer, it is important to characterize the protein conformational motion that underlies substrate transport. Here, solution NMR (64), oriented-membrane SSNMR, and MAS SSNMR (10) have all been used to probe EmrE motion. The chemical shifts and linewidths of EmrE bound to isotropic and oriented bicelles both showed that the apo protein undergoes conformational exchange at a rate of ~350 s⁻¹ at 37°C, while the TPP⁺-bound protein has 50-fold slower motion (10) (Fig. 3d). Fitting the linewidths as a function of temperature gave an activation energy of 28±5 kcal/mol, which is consistent with the expected free energy change when the hydrophobic area between the two subunits changes during the inwardoutward conformational conversion. The activation energy suggests that three out of four TM helices are involved in the conformational exchange, while TM4 is unchanged. The motional rates measured in isotropic bicelles by solution NMR agree well with rates measured from bilayer samples using ¹⁵N exchange NMR (10), indicating that EmrE motion is preserved between the two environments. The NMR-detected slow rate of 3.2–4.9 s⁻¹ (37–45°C) is overall consistent with the turnover rate of < 1 s⁻¹ for TPP⁺.

EmrE operates in a single-site alternating-access fashion, where a critical E14 mediates both proton import and substrate efflux (Fig. 3a). The ligand interaction and protonation behavior of E14 have been studied using SSNMR and mutagenesis. Two sets of E14 chemical shifts were observed that differ between the apo and ligand-bound states, indicating that this residue has two conformations in the dimer, and substrate binding creates additional states that are in slow exchange (52). The close proximity of E14 to the substrate TPP⁺ was demonstrated using dynamic nuclear polarization (DNP) experiments that enhance the ¹³C sensitivity (67). Cross peaks between ¹³C-labeled TPP⁺ aromatic carbons and the sidechain carboxyl of deprotonated E14 were observed in 2D ¹³C-¹³C correlation spectra. Recent studies also showed that the EmrE conformational motion is pH-dependent. The ¹⁵N NMR-detected exchange rate is faster (~220 s⁻¹) at low pH when E14 is protonated than at high pH (~40 s⁻¹) where E14 is unprotonated (29). Thus, the E14 protonation state affects the inward-outward conversion rate. These data also indicate that E14 has an elevated pK_a of 7.0 compared to the bulk pK_a, reminiscent of the pK_a increase of E71 in KcsA.

A much larger class of transporters, ATP-binding cassette (ABC) transporters, has also been studied by SSNMR. ABC transporters utilize the energy from ATP hydrolysis to transport substrate molecules. The transporter associated with antigen processing (TAP) is a 150 kDa heterodimeric protein that binds viral peptide antigens of >8 amino acids that are processed by the proteasome. TAP delivers the antigenic peptides to the endoplasmic reticulum as part of the adaptive immune response. Binding of an antigenic peptide to human TAP was studied using ¹³C, ¹⁵N-labeled peptide and unlabeled TAP (53). The ¹³C chemical shifts of the labeled peptide were detected using double-quantum (DQ) experiments, which suppressed the natural abundance TAP signals. The bound peptide was found to be extended,

spanning a length of 2.5 nm. Docking of this structure onto TAP suggested the binding site to be a negatively charged pocket at the dimer interface. These DQ ¹³C experiments required DNP to achieve high sensitivity, since the large molecular weight of the TAP limited the mass of the labeled antigenic peptide that can be used in the study.

II. Seven-Transmembrane-Helix (7TM) Proteins

While the small ion channels and transporters that have been characterized by SSNMR so far are marked by significant water-exposed protein surfaces and extensive structural plasticity, 7TM helix proteins are nearly diametrically opposite in both respects. Their water-exposed surface area is a small fraction of the whole protein and is mostly restricted to short loops between TM helices, and their structures are relatively insensitive to the membrane composition. So far, the 7TM proteins that have been studied by SSNMR fall into two classes: microbial rhodopsins and eukaryotic G-protein coupled receptors (GPCRs). Microbial rhodopsins serve as light-driven proton pumps, with bacteriorhodopsin (bR) and proteorhodopsin (PR) as prime examples, photosensors such as sensory rhodopsin, and light-gated non-specific cation channels such as channelrhodopsins (ChR) (22). These microbial rhodopsins all contain a retinal chromophore, whose photocycle is intimately related to the protein function. In comparison, eukaryotic GPCRs use ligand binding from outside the cell to activate an array of cellular responses (70), and are the target of ~40% of pharmaceutical drugs.

Because of the dominance of protein-protein interaction over protein-lipid and protein-water interactions, 7TM proteins typically give much narrower SSNMR spectral linewidths than small ion channels and transporters. This structural property, together with sparse isotopic labeling, enabled the determination of the atomic-resolution structure of Anabaena sensory rhodopsin (ASR), a cyanobacterial photoreceptor (94). Chemical-shift based torsion angle restraints and inter-residue cross-peaks as distance constraints were measured to determine the structure of each protein monomer (Fig. 4a). Since ASR forms trimers in lipid membranes, the intermolecular interface was determined by covalent attachment of a nitroxide spin label to S26C in helix A and measurement of paramagnetic relaxation enhancement (PRE) of neighboring spins. The PRE data showed that helices A and B of one subunit are packed against helices D and E of the adjacent subunit (93) (Fig. 4b). This intermolecular helix packing is the same as that of bR, but the trimer interface is rich with aromatic residues in ASR, in contrast to the methyl-rich interface in bR. The higher aromatic content of the ASR trimer interface was proposed to explain the higher stability of ASR in detergents. The chemical shifts of ASR bound to DMPC/DMPA liposomes were found to be very similar to the chemical shifts of protein bound to *E. coli* lipids (96), indicating that the ASR structure is mostly independent of the membrane composition, in contrast to the conformational plasticity of single-span TM proteins such as influenza M2.

The importance of understanding the oligomeric structure of 7TM proteins is further exemplified by the fact that microbial rhodopsins can adopt multiple oligomeric states. A case in point is green PR, which can form pentamers or hexamers depending on the protein to lipid ratio and membrane composition. Using DNP SSNMR and mixtures of ¹³C and ¹⁵N-labeled proteins, Glaubitz and coworkers measured intermolecular contacts (60). 2D

¹⁵N-¹³C transferred-echo double-resonance (TEDOR) spectra of pentameric PR showed cross peaks between Arg/Lys and Asp/Glu, indicating two salt bridges at the pentameric interface. Point mutations of these charged residues disrupted the pentameric assembly to favor monomeric or hexameric PR, confirming the importance of these salt bridges for the oligomeric assembly.

III. β-barrel Outer Membrane Proteins

While α -helical TM proteins are found in cytoplasmic membranes, β -barrel membrane proteins are found in the outer membranes of Gram-negative bacteria and in eukaryotic mitochondria membranes. These outer membrane proteins (OMPs) carry out many functions such as ion and metabolite transport, recognition of eukaryotic cells during infection, outer membrane biogenesis, and structural roles (46).

The voltage-dependent anion channel (VDAC) is a mitochondrial β -barrel protein that regulates metabolite exchange, calcium homeostasis, and mitochondria-mediated apoptosis. SSNMR has been used to investigate VDAC isoform 1 (VDAC1) in lipid bilayers to understand protein-lipid interactions and motions of specific protein domains. Static ²H spectra of d₅₄-DMPC indicate that VDAC1 caused an 8 °C increase and broadening of the lipid phase transition temperature (19). 2D ¹³C-¹³C correlation spectra of ¹³C, ¹⁵N-labeled VDAC1 in 2D crystals of DMPC showed little changes in cross-peak positions or intensities between 0 °C and 30 °C, indicating that the protein conformation and dynamics are unaffected by the membrane phase transition. In comparison, 2D crystals of DPhPC lipids affected some of the cross-peak positions and enhanced the spectral resolution, indicating that this lipid affects the protein structure, reminiscent of the large effect of DPhPC on the influenza M2 structure (3). Other experiments investigated the N-terminal structure of VDAC1, which was proposed to be important for channel gating. 2D ¹³C-¹³C proton assisted recoupling (PAR) spectra detected cross peaks between S193 and the N-terminal residue A14, supporting the position of the N-terminus inside the β -barrel (18). Phosphorylation or mutation of S193 to Glu has been proposed to close the channel, possibly due to structural changes in the N-terminus. However, mutation of \$193 to Glu or Ala showed only minor chemical shift perturbations for the N-terminal residues, and functional assays indicate that the mutant has similar conductance as the wild-type channel, thus refuting this model. Extensive backbone and sidechain resonance assignments for VDAC1 in 2D crystals of DMPC bilayers (20) found that the β -strand residues in the protein show very similar structure to LDAO-micelle bound VDAC1, while loops and solventexposed residues exhibit larger chemical shift differences.

IV. Viral Fusion Proteins

Enveloped viruses enter cells by fusing the virus envelope with the target cell membrane using viral fusion proteins. These proteins respond to environmental cues such as low pH and receptor binding to undergo a cascade of conformational changes to pull the two membranes into close proximity, cause membrane curvature and dehydration, and eventually merge the two membranes (98). The influenza virus hemagglutinin (HA), human immunodeficiency virus (HIV) gp41, and paramyxovirus (PIV) F, all share the common

features of a predominantly α-helical water-soluble ectodomains and trimeric organization of the pre-fusion and post-fusion states. They also contain two important membraneinteracting domains, an N-terminal fusion peptide (FP) that inserts into the target membrane during early stages of fusion, and a C-terminal transmembrane domain (TMD) that remains anchored in the virus envelope. In the post-fusion state, the ectodomain of the protein forms a trimer of hairpins, or six-helix bundle (6HB), which presumably brings the hydrophobic FP and TMD together in the same membrane.

Although crystal structures of the water-soluble ectodomain of a number of viral fusion proteins have been determined (33), the structures of the membrane-interacting FP and TMD have been much more elusive. SSNMR and solution NMR studies have been conducted extensively to characterize the structures and lipid interactions of these hydrophobic fusion protein domains. A common observation that has emerged from these studies is that the FPs and TMDs adopt different conformations depending on the membrane conditions and pH. Chemical shift measurements indicate that the HIV and influenza FPs are predominantly α -helical in negatively charged lipids and cholesterol-free membranes, but convert to the β -sheet conformation in cholesterol-containing membranes (26; 30; 31; 79). The parainfluenza virus 5 (PIV5) FP exhibits α -helical conformation in negatively charged lipids with and without cholesterol, but shifts to the β -sheet structure in negative-curvature lipids such as DOPC and DOPE (106; 107). This membrane-curvature induced conformational change was also observed for the PIV5 TMD, which adopts a sheet-helix-sheet structure in the most negative-curvature membrane (DOPE) but a predominantly helical structure in lamellar membranes, regardless of the membrane surface charge and cholesterol content (109).

Lipid mixing assays, mutagenesis, and measurement of membrane curvature and dehydration, shed light on the relevance of the α -helical and β -sheet conformations for membrane fusion. For the PIV5 FP and TMD, ³¹P NMR and small-angle X-ray scattering data show that the β -sheet conformation induces negative Gaussian curvature to phosphatidylethanolamine (PE) lipids (106; 109), which is important in hemifusion intermediates. 2D ³¹P-¹H correlation spectra correlating the lipid headgroup with water ¹H chemical shifts further indicate that the β -sheet rich conformation causes membrane dehydration compared to the helix-containing membrane. These data suggest that the β -sheet conformation is more important for fusion. For the HIV and influenza FP, both helical and sheet conformations were found to be fusogenic from lipid mixing assays, but fusion activity correlates with other structural features. ¹³C-¹⁵N and ¹³C-³¹P rotational-echo double-resonance (REDOR) data showed that β -strand FP of HIV gp41 associates in an antiparallel fashion (79) and is inserted into intermediate depths of the lipid membrane (26), with the deeper insertion correlating with stronger fusogenicity.

The influenza FP forms a helix-turn-helix motif in cholesterol-free membranes, with the angle between the two helices sensitive to pH. ¹³C-¹⁵N REDOR distance measurements between G16 and F9 and between A5 and M17 found that the peptide backbone adopts a mixture of closed and semi-closed states (31). Greater fusogenicity was observed at low pH where the semi-closed population is higher, suggesting that the larger ratio of hydrophobic to hydrophilic surface areas promotes fusion, possibly by stabilizing the membrane curvature needed for fusion (30).

Recent SSNMR experiments have increasingly focused on larger constructs of viral fusion proteins to address the question of whether the FP and TMD structural information obtained from peptide studies is preserved in the intact protein, and how this structure fits into the context of coordinated conformational changes of these multi-domain proteins. HIV gp41 has a relatively small ectodomain, and the FP has been successfully ligated to the two coiled coil segments (NHR and CHR) to probe the post-fusion state, while ligation of FP with NHR alone serves as a model of the prehairpin structure (75; 77). When the FP is tethered to both NHR and CHR, the resulting FP-hairpin was found to cause rapid fusion of anionic vesicles at low pH but not at high pH, nor to neutral vesicles, indicating that electrostatic interactions between the lipids and the ectodomain have significant effects on fusion. Chemical shift measurements indicate that the hairpin increased the helical propensity of the FP compared to the separate FP peptide, but the FP β -sheet remains antiparallel in the long construct, as shown by intermolecular ¹³CO-¹³CO dipolar coupling measurements (77). Thus, the ectodomain does not change the oligomeric assembly of the FP. Since the fulllength protein forms parallel coiled coils for the ectodomain, the antiparallel FP structure found from the SSNMR data was explained as interdigitation of at least two trimers in the membrane.

Compared to HIV, the PIV5 fusion protein has a much larger ectodomain, thus the postfusion state was investigated using a chimera that links the hydrophobic FP and TMD through a short and flexible linker (108). Interestingly, chemical shifts from 2D correlation spectra indicate that both domains exhibit stable α -helical conformations in all membranes examined, without a shift to the β -sheet conformation in negative-curvature lipids. This implies that each hydrophobic domain exerts an influence on the conformation of the other domain. However, no inter-domain cross peaks were observed in 2D correlation spectra up to a mixing time of 1.0 s, indicating that these FP and TMD helices are not tightly packed as the ectodomain 6HB. The chimera is well inserted into the membrane without perturbing the lamellar structure, which also differs from the property of the separate peptides. These results illustrate the dramatic difference in the structural properties and lipid interactions between the spatially separate peptides and the combined protein, indicating that the early fusion states of the protein have very distinct structures from the post-fusion structure. Future studies will be important for delineating the atomic details and membrane interactions of the prefusion and postfusion states of these fusion machineries.

V. Membrane-interacting amyloid-forming proteins: Aβ and α-synuclein

Although atomic-resolution structures have been determined for a number of amyloid fibrils using SSNMR (13; 89), recent research increasingly suggests that the neurotoxic agents in some of the most common neurodegenerative diseases may be prefibrillar and oligomeric states of these proteins rather than mature fibrils. For the Alzheimer's disease (AD) Aβ peptide, one hypothesis of the mechanism of toxicity is Aβ interaction with lipid membranes (49): the membrane surface was proposed to speed up fibril formation compared to aqueous solution, and this faster aggregation may in turn disrupt the membrane integrity. An atomicresolution structure of Aβ40 bound to DMPC/DMPG membranes was recently reported (66). The lipid-bound Aβ fibrils exhibit the same general β-loop-β architecture as the solution fibrils, but the C-terminal strand has a significant kink, which is absent in the

solution fibril. How this β -loop- β binds to and self-assembles in the lipid membrane is of significant interest. Binding and fibril formation depend on the peptide-lipid mixing protocol. Direct titration of monomeric A β to lipid vesicles gives slower fibrilization (days) and less membrane disruption, while peptide that is mixed with lipids in organic solvents first before switching to aqueous solution caused β -sheet formation within hours (73; 74). These differences in fibril-forming kinetics were manifested in thioflavin-T fluorescence, circular dichroism, ³¹P NMR spectra, and 2D ¹³C-¹³C correlation spectra. The lipid composition affects AB binding to the membrane and folding: negatively charged lipids such as POPG and thin bilayers such as DLPC facilitate fibril formation (48). The slower folding of titrated A β allowed the comparison of the folding rates of different domains of A β : A30 and L34, which reside in the C-terminal hydrophobic core, showed no chemical shift changes after 2 days, while D23, S26 in the loop region showed chemical shift changes over the entire incubation period (73). The depth of insertion and membrane topology of the AB peptide were measured using ¹³C-³¹P REDOR experiments, which showed that the L17-L34 segment has close contact with lipid headgroups (2; 73). Complementary to the SSNMR data of membrane-bound A β fibrils, solution NMR studies of A β in earlier stages of binding to zwitterionic lipids showed that the region between K16 and E22 has a conserved a-helical structure while the rest of the peptide is disordered (47). Many questions remain open about the precise structure of lipid-bound AB fibrils, the mechanism of fibril formation in the membrane, and whether membrane pores are induced by these fibrils.

The 140-residue protein α -synuclein (α S) is intrinsically disordered in solution and can misfold into parallel in-register β -sheets, which are the major constituent of Lewy bodies in Parkinson's disease (PD) (88). α S can also bind membranes, and this binding is implicated in the physiological function of a S in regulating synaptic vesicle clustering, recycling and neurotransmitter release (25). SSNMR and solution NMR have been used to characterize the propensities of different regions of α S to bind lipid membranes and adopt defined secondary structures. The protein consists of a long N-terminal domain, a central non-amyloid-B component (NAC) region, which is the core of the β -sheet fibril in the aggregated state (88), and a disordered C-terminus. 2D ¹³C-¹³C correlation spectra of POPC/POPA-bound aS followed the structural evolution of aS in the course of several days (12). aS exhibits ahelical chemical shifts for N-terminal and NAC-enriched Ala, Thr, and Val residues up to a day, after which the chemical shifts progressively change to β -sheet values and become stable after ~11 days (12). In contrast, fibrilization from solution did not show the α -helical intermediate. The depth of insertion of aS into the POPC/POPA bilayers was probed using ¹H-¹³C 2D correlation experiments with ¹H spin diffusion (44). Comparison of waterprotein and lipid-protein ¹H-¹³C cross peaks indicates that a S is well exposed to water but far from the lipid acyl chains at short incubation times, but the lipid-protein cross peaks become equal to water-protein cross peaks after 3 days, concomitant with the dominance of the β -sheet conformation. These results indicate that membrane-bound αS evolves from a surface-bound α -helical structure for the N-terminal and NAC domains to an inserted β sheet structure, with the end point of the structure similar to fibrils formed in solution.

Interesting structural differences were found when α S is bound to small unilamellar vesicles (SUV's) that mimic the composition (DOPE, DOPC, and DOPS) and curvature of synaptic vesicles. Chemical shifts measured from 2D 13 C- 13 C and 15 N- 13 C correlation spectra

indicate that the SUV-bound α S has a continuous and rigid α -helical structure for residues 6–25 (24; 25), and this domain is strongly associated with the membrane. The N-terminal α -helix is relatively independent of the lipid composition (21), which is consistent with the POPC/POPA study using large unilamellar vesicles. In comparison, residues 26–97 exhibit intermediate dynamics and cannot be detected in cross polarization (CP) and insensitive nuclei enhanced by polarization transfer (INEPT) spectra, in contrast to the POPC/POPA results. These intermediate dynamics are proposed to be relevant for α S's function in anchoring two synaptic vesicles. The depth of insertion of α S in SUV's was probed using PRE experiments (24). Gadolinium salts of PE-DTPA lipids caused stronger PRE broadening of hydrophilic residues than hydrophobic residues of the N-terminal α -helix, indicating a membrane surface location, while 16-doxyl-PC with a paramagnetic doxyl in the lipid tails did not cause line broadening of protein residues, indicating that α S does not insert deeply into these curved SUVs. This result again differs from the topology of α S in large membrane vesicles. Thus, both the conformation and membrane topology of α S are sensitive to the membrane composition and curvature.

Conclusion

Two common themes about membrane protein structures have emerged from SSNMR studies of membrane proteins in the last 10 years. First, the protonation state of polar residues is critical for the structure and function of many membrane proteins, as exemplified by the proton-selective histidine in the influenza M2 protein, glutamic acid residues in the pH sensor of KcsA, and the substrate-binding glutamate in EmrE. SSNMR is ideal for elucidating the protonation state and hydrogen bonding of these residues and will continue to make unique and insightful discoveries in this area. Second, structural plasticity is extensive for membrane proteins whose function requires substrate movement or membrane remodeling. a-helical ion channels and transporters exhibit both localized motion of functional residues and coordinated motions of protein backbones. The former is exemplified by the proton-selective histidine and gating tryptophan of influenza M2, while the latter is showcased by the EmrE transporter. This plasticity is manifested especially strikingly in viral fusion proteins, whose two hydrophobic domains can undergo dramatic helix-sheet transitions due to changes of the membrane lipid composition. Even for larger 7TM helix proteins, the oligomeric structure can vary, as seen in the case of proteorhodopsin. The factor that affects membrane protein structure most significantly is the lipid composition of the membrane. Thus, choosing the lipid composition that best mimics the natural membrane in which a protein carries out its function is crucial for obtaining biologically relevant structural information. As SSNMR techniques increasingly mature and become well adopted to determine high-resolution structures of membrane proteins, elucidating protein conformational dynamics and conformational plasticity, unraveling protein-lipid interactions, and illuminating how membrane proteins interact with ions, drug molecules, ligands, and other proteins, will become ever more important areas of discovery. For revealing the complex structural and dynamical basis of membrane protein function, SSNMR remains an unrivaled technique.

Appreviations:

DHPC	1,2-dihexanoyl-sn-glycero-3-phosphocholine
DOPC	1,2-dioleoyl-sn-glycero-3-phosphocholine
DOPE	1,2-dioleoyl-sn-glycero-3-phosphoethanolamine
DPhPC	1,2-diphytanoyl-sn-glycero-3-phosphocholine
DMPC	1,2-dimyristoyl-sn-glycero-3-phosphocholine
DMPA	1,2-dimyristoyl-sn-glycero-3-phosphate
DMPG	1,2-dimyristoyl-sn-glycero-3-phospho-(1'-rac-glycerol)
DLPC	1,2-dilinoleoyl-sn-glycero-3-phosphocholine
DOPS	1,2-dioleoyl-sn-glycero-3-phosphocholine
POPG	1-palmitoyl-2-oleoyl- <i>sn</i> -glycero-3-phospho-(1'- <i>rac</i> -glycerol)
POPA	1-palmitoyl-2-oleoyl-sn-glycero-3-phosphate
PE-DTPA	1,2-dimyristoyl- <i>sn</i> -glycero-3-phosphoethanolamine-N- diethylenetriaminepentaacetic acid
16-doxyl-PC	1-palmitoyl-2-stearoyl-[16-doxyl]- <i>sn</i> -glycero-3- phosphocholine
PDSD	proton-driven spin-diffusion
PISEMA	polarization inversion spin exchange at the magic angle
TEDOR	transferred-echo double-resonance
PAR	proton assisted recoupling
REDOR	rotational-echo double-resonance
INEPT	insensitive nuclei enhanced by polarization transfer

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Figure 1.

Structural studies of the influenza M2 proton channel by SSNMR. (a) The TM domain structure with bound drug (PDB: 2KQT), solved in DMPC bilayers. Important functional residues are indicated. (b) Determination of His37 pK_a's from pH-dependent His37 sidechain ¹⁵N chemical shifts. The W41F mutant data is shown as an example. Titration curves yield the His37 pK_a's, which differ for the W41F mutant and wild-type channels. (c) Histidine Ca and C β chemical shifts from 2D ¹³C-¹³C correlation spectra. A W41F mutation (black spectra) changes the His37 chemical shifts compared to the wild-type (red spectra), indicating that His37 can be protonated from the C-terminus in the mutant. (d) 2D ¹³C-¹³C correlation spectra of cytoplasmic-containing M2 as a function of pH. Seven states of His37 are observed. (e) Proposed His37 tetrads with different charge states, explaining the resolved seven sets of histidine chemical shifts.



Figure 2.

KcsA structure investigated using SSNMR. (a) Key domains and functional residues in KcsA that have been studied (PDB: 1K4C). (b) Measured sidechain carboxyl chemical shift anisotropies (CSA) (*black*) of E71 and D80, with the simulated CSA patterns overlaid in *red* (5). The simulated spectra of average protonated (*dark blue*) and deprotonated (*light blue*) carboxyl CSA tensors are shown on the right, based on a database of known crystal structures. Compared to this database, the δ_{11} component of E71 matches with protonated (*dark blue*) carboxyls, while E51 and D80 δ_{11} components match with deprotonated (*light blue*) carboxyls, indicating an elevated pK_a for E71. (c) The hydrogen-bonding network connecting the selectivity filter and the pore helix in the high K⁺ conductive state (PDB: 1K4C). Panels (b) and (c) reproduced with permission from Reference (5).



Figure 3.

(a) Single-site alternating-access model for substrate transport by EmrE. The transporter is only accessible to one side of the membrane at a time, and a conformational change is needed to switch between the two sides. (b) Model of the asymmetric antiparallel dimer structure of EmrE (PDB: 3B5D), in which E14 affects the conformational exchange rate. (c) Representative 2D PISEMA spectra of ¹⁵N-Val labeled EmrE in oriented bicelles. Most residues show peak doubling, indicating two distinct orientations relative to the bilayer normal (28). (d) Millisecond-timescale conformational change measured by static ¹⁵N exchange NMR on substrate-free (*red*) and substrate (TPP⁺)-bound EmrE (*grey*) (10). The exchange presumably reflects inter-conversion between the inward-facing and outward-facing states, and is ~50 times faster for the ligand-free protein than the TPP⁺-bound protein. Panel (c) reproduced with permission from Reference (28). Panel (d) reproduced with permission from Reference (10).



Figure 4.

Structure of microbial rhodopsin ASR determined by SSNMR. (a) Structure of the trimeric retinal-bound protein in DMPC/DMPA liposomes, with the retinal cofactors shown in green (PDB: 2M3G). (b) Side view of residues that line the trimer interface between helix B of one monomer and helices D and E of the adjacent monomer. The high aromatic content at the interface contributes to the high stability of the ASR trimer compared with the prototypical microbial rhodopsin, bR (94). (c) Representative 2D ¹³C-¹³C proton-driven spin-diffusion (PDSD) correlation spectrum of membrane-bound ASR with a mixing time of 500 ms, showing interhelical (*red*) and intrahelical (*blue*) cross peaks as structural restraints. Panel (c) reproduced with permission from Reference (94).