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Can light-based approaches overcome antimicrobial resistance?

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Abstract

The relentless rise of antibiotic resistance is considered one of the most serious problems facing mankind. This mini-review will cover three cutting-edge approaches that use light-based techniques to kill antibiotic-resistant microbial species, and treat localized infections. Firstly, we will discuss antimicrobial photodynamic inactivation using rationally designed photosensitizers combined with visible light, with the added possibility of strong potentiation by inorganic salts such as potassium iodide. Secondly the use of blue and violet light alone that activates endogenous photoactive porphyrins within the microbial cells. Thirdly the use of “safe UVC” at wavelengths between 200–230 nm that can kill microbial cells without damaging host mammalian cells. We have gained evidence that all these approaches can kill multi-drug resistant bacteria *in vitro*, do not themselves induce any resistance, and moreover can treat animal models of localized infections caused by resistant species that can be monitored by non-invasive bioluminescence imaging. Light-based antimicrobial approaches are becoming a growing translational part of anti-infective treatments in the current age of resistance.

1 Introduction to Microbial Drug Resistance

Alexander Fleming discovered the first antibiotic called “penicillin” in 1928 (1), and when mass production became possible, it was widely used by Allied Forces in World War II for treating wounds and post-surgical infections. After the war, penicillin began to be prescribed to the general population at large where it was widely hailed as a “wonder drug”, and commentators predicted a future free from the threat of infectious disease caused by bacteria. However when Fleming won the Nobel Prize in 1945 for his discovery, in his Nobel lecture he already warned of bacteria becoming “easily resistant to penicillin” (2). This resistance had been observed even before the antibiotic entered into broad clinical use. Since then a large number of antibiotics have been discovered, and dread infectious diseases caused by bacteria (pneumonia, tuberculosis, cholera, typhoid, diphtheria) do not wreak the large death toll they once did. Surgery, accidental trauma and childbirth have become much safer as a result of antibiotics.

However, over the last 60 years the development of antibiotic resistance has accelerated rather than slowed down (see Table 1). The reasons for this are many and various. Global

consumption of antibiotics in clinical medicine rose by nearly 40% between 2000 and 2010. but this figure masks patterns of declining use in some countries and rapid growth in others. There are large variations between different countries; the annual per-person consumption of antibiotics varies by more than a factor of 10 even in developed countries. Overuse and inappropriate use of antibiotics is facilitated in many places by their availability over-the-counter without prescription, and counterfeit and sub-standard antibiotics dominate the market in some regions. The speed and volume of international travel creates new opportunities for antibiotic-resistant pathogens to spread globally. Bacteria can easily share their genetic material with each other, creating new resistant strains at an unprecedented pace. The biggest problem of all is the fact that almost 80% of all antibiotics in the United States are not taken by people but rather used in livestock feedstuff, which was pointed out by Gloria Guglielmi (3). They are administered to cows, pigs, and chickens to make them grow more quickly, and to keep these animals healthy, in intrinsically unhealthy factory farming conditions. In 2013, more than 131,000 tons of antibiotics were used in food animals worldwide; by 2030, it is expected to be more than 200,000 tons.

The real value of antibiotics goes beyond simply preventing death and illness due to infection, in that antibiotics also permit the serious iatrogenic assault on the immune system that occurs in cancer treatment by chemotherapy or radiation therapy, or in organ transplantation, where they have helped to keep complication rates down (4). Therefore, antibiotics are an extremely valuable resource across the whole spectrum of modern medicine (5). However, multidrug-resistant (MDR) and pandrug-resistant (PDR) bacterial strains and the related infections they cause, have become emerging threats to public health throughout the world (6). These infections are associated with an approximately two-fold higher death rate, and with considerably prolonged hospital stays (7). Infections caused by antibiotic resistant microbes are often exceptionally hard to treat due to the limited range of therapeutic options (8). Therefore, there is an urgent need for an all-out search for alternative antimicrobial approaches to kill MDR strains, concentrating on methods that are unlikely to cause resistance to develop (9–11). Recently, Karen Bush *et al.* pointed out that novel non-antibiotic approaches to prevent and treat infectious disease should be considered high-priority international research and development goals (12).

A promising, innovative approach to achieve this goal is the use of light-based approaches to inactivate pathogenic and resistant microbes infecting living tissue, without causing any unacceptable damage to that tissue.

2 Light-based antimicrobial approaches

The part of the electromagnetic spectrum that is known as “light” ranges from UVC (200–280 nm), UVB (280 – 320 nm), UVA (320 – 400 nm), visible (400 – 750 nm), near infrared (NIR, 750 – 1200 nm) and mid/far IR (1200 – 10,000 nm). All these different wavelengths have been used in one form or another for killing various types of microbes. While bacteria have been by the most often studied microbial grouping, fungi, viruses and parasites have also been killed by light-based techniques.

Light based approaches must draw a balance between: (a) the quantum yield of microbial inactivation, in other words how many photons does it require to kill or inactivate a single microbe? (b) the penetration depth of the light into the tissue which increases with wavelength; (c) the propensity of the light to damage the host cells and tissues. UVC is highly active with a lethal dose of a few mJ of energy to kill or inactivate all known types of pathogens, but has poor penetration into tissue and a high possibility for to damage host cells. However so-called “safe UVC” at 220 nm does not damage host cells. Blue light has moderate activity needing 100s of J of energy, and also has relatively poor penetration, but does not much damage host tissue. Red light has no activity at all unless it is combined with a photosensitizing dye (PS), when the activity increases dramatically needing only a few J of energy. Penetration into tissue is good and damage to normal tissue is controllable depending on the PS structure. NIR light has moderate activity (100s of J) but has good penetration, and any damage to tissue depends on the heat produced which in turn depends on the power density.

One of the main advantages of light-based approaches is that most of them (but not all of them) are broad-spectrum in nature and can effectively destroy all kinds of microbes including bacteria (Gram-positive, Gram-negative, mycobacteria), fungi (yeasts and filamentous fungi), viruses (DNA and RNA) and parasites.

Another important advantage of light-based approaches is that the effectiveness of microbial destruction appears to be largely unaffected by the antibiotic resistance status of the particular microbe. Moreover, with the possible exception of UVC, it has proved impossible to artificially generate resistant microbes by the deliberate repeated administration of cycles of sub-lethal killing followed by regrowth. As many as 20 of these repeated cycles have been carried out with any significant resistance developing (13, 14).

3 Antimicrobial photodynamic inactivation (aPDI)

The discovery of photodynamic therapy (PDT) can be dated back to the year 1900, when workers in Munich, Germany observed that single-celled *Paramecia* (amoebae) were inactivated by the dye, acridine orange in the presence of light, but not in the dark (15). When it was found a few years later that the presence of oxygen was necessary for this effect to occur, the term “photodynamic” was coined (16). PDT was mainly developed as a cancer therapy in the 1960s and 1970s (17, 18). The photochemical mechanism of PDT was first reported in 1977 by Weishaupt et al (19). They identified singlet molecular oxygen as the most important cytotoxic agent in the destruction of cancer cells incubated with hematoporphyrin and exposed to red light. It was later realized that the key feature of compounds (dyes) such as the porphyrins used for PDT, was the existence of a long-lived triplet state that could be formed by an “intersystem crossing process” from the excited singlet state porphyrin, formed when the ground state porphyrin absorbed a photon of energy (20). Since ground state oxygen has a triplet electronic configuration, the excited porphyrin triplet state is allowed to undergo energy transfer with oxygen to produce the ground state singlet porphyrin and the excited state singlet oxygen ($^1\text{O}_2$). $^1\text{O}_2$ is a highly reactive molecule that can oxidize lipids, proteins and nucleic acids, thus causing the death of any kind of cells (bacteria, fungus, cancer, normal cells, etc). Because $^1\text{O}_2$ is highly reactive, it

can only damage tissue or cells in the precise location where it is produced (i.e. where both dye and light are present at the same time).

In addition to $^1\text{O}_2$, (which became known as the Type II photochemical mechanism), it was realized that other different reactive oxygen species (ROS) were also involved in the PDT effect depending on a number of factors. These ROS (including hydroxyl radicals, hydrogen peroxide, and superoxide anion) were produced by what became known as the Type I photochemical mechanism (21). It is thought that this process initially involves a 1-electron transfer from the PS triplet state to oxygen to form superoxide anion. Figure 1 shows a Jablonski diagram illustrating the photochemical mechanisms of PDT.

PDT was not much used as an antimicrobial approach for many years, because the PS that were commonly used for cancer therapy were not particularly effective at killing bacteria, especially Gram-negative strains (22). It was realized around 1990 that, although many Gram-negative bacteria are resistant to aPDI with commonly employed anionic or neutral PS (23), this resistance could be overcome by the use of PS with pronounced cationic charges, or the use of other means of permeabilizing Gram-negative cells (24, 25). These cationic charged PS can be taken up by Gram-negative cells by the process called “self-promoted uptake” (26). In this process, the cationic PS can displace the divalent cations (Ca^{2+} and Mg^{2+}) that hold together the lipopolysaccharide (LPS) structure forming part of the outer membrane permeability barrier in Gram-negative cells (27). The loss of the LPS allows much more PS to be taken up, and furthermore initial light-mediated damage also increases PS uptake allowing greater PDT damage to occur (28).

The vast majority of PS structures that have been tested for antimicrobial applications are molecules that possess intrinsic cationic charges. These molecules may be cationic porphyrins, chlorins, bacteriochlorins, phthalocyanines, phenothiazinium dyes, fullerenes, BODIPY-dyes, as well as some natural products (29).

There is one major difference between PDT, when it is used as an anticancer treatment, and when it is used as an anti-infection treatment (30). This concerns the mode of administration and the drug-light interval (time between applying the PS and shining the light). In order to treat cancer, it has been well established that the PS functions best when it is injected intravenously into the circulation and allowed to accumulate at the site of the tumor. Indeed, many PS were first clinically tested as fluorescent contrast agents that could “light up” undiscovered tumors in difficult anatomical locations (31, 32). In some cases drug-light intervals as long as 96 hours have been allowed to elapse between injection of the PS (m-tetrahydroxyphenylchlorin) and switching on the red laser light (33). In sharp contrast is the situation when PDT is used to treat localized infections. Here the PS is topically introduced into the infected area. This may be done by painting or dropping the PS solution onto the infected tissue surface, or by for instance a mouth rinse, or instilling into a hollow organ. A short amount of time (minutes) is allowed for the PS to attach and penetrate the microbial cells, which is a relatively rapid process, but not long enough for the PS to be taken up by host cells which is a relatively slow process. Maisch et al even used as short a time as 10 seconds for incubation of the PS with the microbial cells (34). Then when light is delivered

the ROS are preferentially generated inside the microbial cells, and no harm is done to host tissue.

The development of resistance to aPDI has not yet been observed despite attempts to deliberately encourage it. If aPDI was used in the treatment of infections, and the PS could only reach the target site at sub-lethal concentrations, any microorganisms surviving would have been exposed to sub-lethal doses of oxidative stress that would not result in total cell death. Since it is known that aPDI can damage DNA, it might also lead to increased mutational events, which could lead to selection for survival of more resistant or less susceptible strains (35). The most frequently employed method of testing for a microbe's ability to become resistant to a particular agent is by subjecting the said microbe to routine continual exposure to a particular agent.

Giuliani et al. analyzed whether *S. aureus*, and *P. aeruginosa* could become resistant to aPDI via repeated exposure to a tetracationic PS Zn(II) phthalocyanine derivative in combination with 30 J/cm² of 600–700 nm light. After 20 consecutive treatments at PS concentrations corresponding to the previously determined minimum inhibitory concentration (MIC), *S. aureus*, and *P. aeruginosa* were all incapable of developing resistance to aPDI. However, when the 20 exposures were repeated without light the MIC for *S. aureus* of the Zn (II) phthalocyanine derivative in the dark did increase. This shows that *S. aureus* may be able to develop some ability to protect itself against the dark toxic effect of a PS, perhaps by up-regulating efflux pumps or altering its membrane structure (14).

Cassidy *et al.* exposed *S. aureus* and *P. aeruginosa* to the PS meso-tetra (N-methyl-4-pyridyl) porphine tetra tosylate (TMP) and methylene blue (MB) for 72 h in an effort to “habituate” cells. It was shown that sub-lethal aPDI did not decrease the susceptibility to commonly employed antibiotics. Similarly, habituation with sub-lethal aPDI did not reduce susceptibility of *P. aeruginosa* isolates to aPDI protocols previously determined as lethal (eradication). A reduction in susceptibility to aPDI following habituation was apparent for two *S. aureus* isolates with MB and for 1 *S. aureus* isolate with TMP as the photosensitizer (36).

Pourhajibagher et al. studied the effects of aPDI with toluidine blue O (TBO) and light emitting diode irradiation, on virulence features and expression profiling of genes encoding potential virulence factors in a colistin-resistant extensively-drug resistant (XDR) clinical isolate of *A. baumannii* (CR-XDR-AB) and in *A. baumannii* ATCC 19606 by studying the cells surviving aPDI. Their results showed that the aPDI could lead to modulation of the virulence of *A. baumannii* strains in surviving cells in planktonic growth mode by suppressing the expression of the genes (*csuE*, *epsA*, and *abaI*) associated with biofilm formation, but not *blsA* (the gene corresponding to attenuation in biofilm formation) at the transcriptional level. Following aPDI, there was a significant up-regulation of *blsA* in CR-XDR-AB, but no up-regulation in the 19606 strain. Insignificant changes in the gene encoding bacterial heat shock protein (*dnaK*) and DNA repair gene (*recA*) in surviving cells demonstrated that sAPDI did not induce the typical cell stress fund with other antibacterial treatments, and there was no observable DNA damage when applied to *A. baumannii* strains in the planktonic growth mode. CR-XDR-AB cells surviving aPDI showed a reduction of

cell metabolic activity, increase in outer membrane permeability, and inhibition of efflux pump systems. aPDI reduced the minimum inhibitory concentrations of the most tested antimicrobials by 2-fold in CR-XDR-AB strain (37).

4 Potentiation of aPDI by addition of inorganic salts

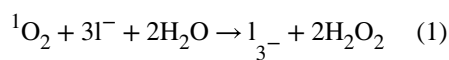
4.1 In vitro studies

The potentiation of aPDI by addition of inorganic salts was discovered in our laboratory. It started when we used sodium azide as a quencher of singlet oxygen (38). Azide has long been known as a physical quencher of singlet oxygen (39). However, when we tested whether azide would quench the bacterial killing mediated by the phenothiazinium dye, methylene blue (MB) and red light (660 nm), we were surprised to find that the killing was paradoxically potentiated (1–2 logs more killing), rather than the expected inhibition (40). Mechanistic investigations showed that there occurred a photoinduced electron transfer to form azide radicals, which could damage the bacterial cells. There was even some oxygen-independent photokilling in the presence of azide, but no killing at all in the absence of oxygen as expected. aPDI mediated by other phenothiazinium dye structures could also be potentiated by addition of azide (41). It is known that phenothiazinium dyes can carry out a mixture of Type 1 and Type 2 photochemistry (38). The fact that oxygen-independent photokilling could also be achieved in the presence of azide when functionalized fullerenes were used as the PS, supported this hypothesis (42). We then hypothesized that electron transfer reactions could occur from the photoexcited PS to other inorganic anions to produce reactive radicals. We tested this hypothesis using photoexcited titanium dioxide which is known as antimicrobial photocatalysis. Photocatalysis largely works by production of Type 1 ROS such as hydrogen peroxide and hydroxyl radicals. This occurs because TiO₂ can act as a large band-gap semiconductor, and when photoexcited by short-wavelength light (UVA, 360 nm), it can transfer electrons to oxygen reducing it to superoxide, while at the same time the positive holes can oxidize water to hydroxyl radicals. We showed that potassium iodide could produce dramatic potentiation of the microbial killing (up to six extra logs) of Gram positive and Gram-negative bacteria and also fungi (43). We needed to use relatively high concentrations of iodide (up to 100 mM) to achieve the maximum potentiation. We also showed that TiO₂ antimicrobial photocatalysis could be potentiated (although not to the same extent) by addition of potassium bromide (44). The mechanism involved the production of hypobromite (OBr⁻), a well-known antimicrobial species. The mechanism of iodide potentiation was a combination of the production of relatively stable species (free iodine and hypoiodide) and reactive iodine radicals (43).

We went on to show that aPDI mediated by MB could also be potentiated by KI (45). In this case the mechanism appeared to be a combination of the production of stable antimicrobial species (free iodine and hypoiodite) and short lived reactive species (iodine radicals). The concentration of KI is critical in determining the mechanism. If the concentration is relatively low (up to 10 mM) then iodine radicals are mainly responsible, but if the KI concentration is increased up to 100 mM (or even higher) then free iodine is mainly responsible.

Because of our results with TiO₂ photocatalysis, we assumed that the mechanism was via electron transfer (Type 1) but we subsequently revised our theory when we found that the antimicrobial effects of other different PS could also be dramatically potentiated by addition of KI (46). The first indication that the antimicrobial action of Type 2 PS could be dramatically potentiated by addition of KI came with Photofrin (47). Photofrin is a well-known PS used for many years to treat cancer with PDT. It is a derivative of hematoporphyrin which is water soluble, and has been found to localize in tumors when injected intravenously. Photofrin is considered to be a typical Type 2 PS which works mainly by the generation of singlet oxygen (48). Because Photofrin is an anionic compound it does not bind, penetrate or photokill Gram-negative bacteria, although it is quite effective at killing Gram-positive bacteria and fungi (49). When we added 100 mM KI to Photofrin and excited it with blue light (405 nm) we were able to eradicate several different species of Gram-negative bacteria (*E. coli*, *Proteus mirabilis*, *Klebsiella pneumoniae*, *Acinetobacter baumannii* and *Pseudomonas aeruginosa*) which were completely unharmed in the absence of KI (47).

To explain this impressive result, we compared the effects of three different formats. The first was termed “In” where all components of the system (bacteria, PS, salt) are present together during the light delivery. The second was “Spin” where the bacteria and PS were incubated together, centrifuged to remove unbound PS, resuspended in KI or other salt solution, and the light delivered. The idea here is to ensure that the PS is in contact with the bacteria, so that if iodine radicals are produced during illumination they will be close to their target. Lastly we have “After”, where the PS and KI (or other salt) were mixed together, illuminated, and either immediately or after different times the bacterial cells were added and mixed. Here the idea is to detect killing caused by stable (free iodine) or semi-stable (hypoiodite) species as compared to short-lived radicals. The results showed that Gram-negative bacteria were killed by free iodine or hypoiodite, while Gram-positive bacteria were more likely to be killed by iodine radicals. The overall chemical equation proposed to explain the oxidation of iodide by singlet oxygen is as follows:



It should be noted that hydrogen peroxide (another stable species) is also produced during this reaction, and there is evidence that hydrogen peroxide and iodine combine to exert synergistic effects against bacterial and yeast species (50).

We next went on to confirm our findings by testing a different (but still well-known) PS that has high activity against Gram-positive species, but no activity against Gram-negative species, namely Rose Bengal (RB). RB was also strongly potentiated by the addition of 100 mM KI producing eradication of several different Gram-negatives, under conditions where no killing was seen without KI (51). We then went on to test a hypothesis that the strength of the binding of the PS to the bacterial cell would govern the amount of potentiation that was obtained by addition of KI. If the PS bound strongly to the bacteria, we argued that potentiation should be found at lower KI concentrations because reactive iodine radicals

could carry out the killing. On the other hand, if the PS did not bind to the bacteria and remained in solution, then illumination would generate free iodine which would need a much higher KI concentration. We decided to test this by comparing two porphyrins with different charges, cationic TMPyP4 and anionic TPPS4 (52). Initially the results were as expected, with TMPyP4 (no salt) eradicating Gram-negatives and TPPS4 producing no killing. As expected both porphyrins eradicated bacteria when 100 mM KI was added. An unexpected result was seen when TPPS4 was tested in the “spin” format and eradicated *E. coli* when KI was added. This seemed to imply that an anionic porphyrin can bind to Gram-negative bacteria. The mystery was solved when we realized that although TPPS4 is a tetrasulfonic acid, it is not actually anionic, but rather behaves as if it is cationic, because of the presence of two ring pyrrole nitrogen atoms that can form hydrochloride salts. Hence although TPPS4 is not sufficiently cationic to kill Gram-negative bacteria in the absence of KI, it is sufficiently cationic to bind to Gram-negatives and kill the bacteria in the “after” format combined with the addition of KI.

There are other inorganic salts besides azide, iodide, and bromide that can potentiate aPDI. We showed that potassium thiocyanate KSCN could potentiate the killing of both Gram-positives and Gram-negatives by MB-aPDI (53). We originally hypothesized that the active species might be thiocyanate radicals (SCN^{\bullet}) formed by a Type 1 electron-transfer mechanism. However mechanistic investigations led us to revise that theory when we found that SCN- quenched singlet oxygen luminescence, dramatically increased oxygen consumption, and produced the spin-trapped ESR signal of an unexpected radical, sulfur trioxide radical anion ($\text{SO}_3^{\bullet-}$) (53).

We then tested another different pseudohalide salt which is isoelectronic with KSCN, namely potassium selenocyanate, KSeCN (54). KSeCN at concentrations up to 100 mM could potentiate aPDI giving up to 6 logs of extra killing of *E. coli* and MRSA. This was observed with aPDI mediated by several different PS structures: MB, RB and TPPS4. When a mixture of selenocyanate with these PS in solution was illuminated and then bacteria were added after the light in the “after format”, there was up to 6 logs of killing (Gram-negative > Gram-positive) but the antibacterial species decayed rapidly (by 20 minutes). Our hypothesis to explain this antibacterial activity was the formation of selenocyanogen (SeCN_2) by oxidation of SeCN- by singlet oxygen ($^1\text{O}_2$) as shown by quenching of the $^1\text{O}_2$ 1270 nm luminescence signal by SeCN- and increased photoconsumption of oxygen. The fact that lead tetraacetate mixed with SeCN- which is a literature preparation of (SeCN_2), also produced a short-lived antibacterial species, supported this hypothesis.

4.2 In vivo studies using potentiation of aPDI by inorganic salts.

Up to now we have demonstrated the in vivo applicability of inorganic salt potentiation of aPDT, using potassium iodide. The reason for choosing KI was as follows. KI is non toxic, and saturated KI solution is prescribed orally as a treatment for diverse fungal diseases and inflammatory conditions in dermatology (55–57). KI also has regulatory approval as a prophylactic medicine for release of radioactive iodine in cases of potential nuclear contamination (58).

For some years we have been utilizing a technology that employs stable bioluminescent bacteria or fungi, combined with sensitive low-light imaging (59) to test in vivo treatment of localized infections in small animal models by aPDT (60) (and some other different antimicrobial technologies (61). The advantages of this approach are: (1) it provides real-time information on bacterial viability and distribution within the tissue; (2) it is non-invasive and does not require sacrifice of animals or removal of biopsies; (3) it dramatically reduces animal numbers and each animal can be repeatedly imaged to follow the time course of infections. Figure 2 illustrates a typical experiment in which aPDT is carried out on an infected 3rd degree burn on the mouse back.

The combination of aPDI mediated by MB together with KI was studied in a mouse model of a murine burn infection using bioluminescent MRSA (45). The infected burn was treated with 50 μ M MB, with and without the addition of 10 mM KI, and excited with 660 nm light at up to 150 J/cm², while the control groups were dark controls with the same amount of MB plus KI and the light-alone control received 660 nm light to 150 J/cm². Figure 3 shows a set of five representative bioluminescence image time courses from burns (each time course from a single mouse in each of the 5 groups) infected with MRSA. The amount of light required to eradicate bacteria from the wound was much less when the MB plus KI solution was irradiated than when MB alone was irradiated.

Other similar studies used an abrasion wound infected with bioluminescent *A. baumannii* and treated with aPDT mediated by a functionalized fullerene excited by UVA or white light combined with KI (62), and an abrasion wound infected with bioluminescent *P. aeruginosa* and treated with aPDT mediated by RB excited by green light combined with KI (51). A mouse model of oral candidiasis caused by bioluminescent *C. albicans* was treated by MB-mediated aPDT combined with KI (63). A model of urinary tract infection in the female rat bladder caused by a bioluminescent strain of uropathogenic *E. coli* was treated by instillation of MB solution followed by a KI solution and then intravesicular delivery of red light using a fiber optic (64).

5 Blue light antimicrobial photoinactivation

5.1 In vitro studies

The advantages of using blue light alone to kill resistant microbes, is that the light is not as harmful to the host tissue or to the surroundings compared to UV light and moreover no added exogenous PS or dye is required (65, 66). Firstly, we must define what wavelengths can be called “blue light”. The most effective range is from ~390 nm to 420 nm (more accurately termed “violet” light), the next most effective range is from 450–480 nm and possible the least effective range is from 420–450 nm. Over the last five years a wide range of microbial cells, including Gram-positive bacteria, Gram-negative bacteria, mycobacteria, molds, yeasts and dermatophytes have been shown to be susceptible to blue light (66). Studies have been carried out in vitro using planktonic cells or biofilms, ex vivo, and in vivo using animal models (pre-clinical) and even in patients (clinical trials). The biological response to blue-light was firstly reported in 1881 by Charles Darwin when he described a blue light induced phototropic response in plants (67), and has since been identified in all three domains (eukaryotes, bacteria and archaea). Recent discoveries have been made

showing that within the genomes of many bacteria and fungi, are sequences encoding photoreceptor proteins. Other photoreceptors have been studied in *E.coli* (68) and *A. baumannii* (69) that possess a functional blue-light-sensing A (bslA) gene, coding for a protein called BLUF (blue light sensing using flavins). Blue light can regulate bacterial motility, suppress biofilm formation, and subsequently potentiate light inactivation of bacteria. On the other hand, the presence of blue light may also activate or increase bacterial virulence (69).

The lethality of blue light for bacteria has been reported both *in vitro* and *in vivo*. Blue light can mediate a broad-spectrum antimicrobial effect on both Gram-negative and Gram-positive bacteria. While the wavelength range of 390–420 nm has been reported to be the most effective antimicrobial spectral range, both 455 nm and 470 nm have also been found to have some antimicrobial effects on some bacterial species (e.g., *S. aureus*). The mechanism of the antimicrobial effect of blue light is that blue light excites endogenous intracellular metal-free porphyrins to behave as PS as described above for the case of aPDI. This photon absorption then leads to energy transfer from the porphyrin triplet state to oxygen producing $^1\text{O}_2$ in a similar manner to PDT (70). However, different bacteria demonstrate variable susceptibilities to blue light. Studies have reported that Gram-positive species, in general, were more susceptible to 405 nm light inactivation than Gram-negative species, which is generally consistent with the results obtained in a recent study (71, 72). It has also been theorized that the differences in inactivation kinetics may be due to organism-specific differences in porphyrin levels, different individual porphyrin sub-types (protoporphyrin, coproporphyrin or uroporphyrin), or different porphyrin subcellular localizations (9, 71). Moreover, it has been speculated that less oxygen-tolerant bacterial species may be particularly susceptible to the effects of ROS as some microaerophilic species have been found to possess fewer key antioxidant defenses than most aerobes (73, 74). For example, studies have shown blue light to be capable of inactivating the anaerobic oral pathogens *Prevotella*, *Porphyromonas*, and *Fusobacterium* as well as microaerophilic pathogens such as *P. acnes* and *H. pylori* (9, 75). However, inactivation results achieved with anaerobic/microaerophilic bacteria have not provided conclusive evidence as to whether oxygen-sensitive anaerobic bacteria are any more susceptible than aerobes as many of these anaerobic bacteria are also known to accumulate high levels of porphyrins.

Blue light is also lethal to many species of fungi, including filamentous fungi, although the effectiveness is dependent on the light wavelength, energy levels and microbial genus and species (76). Blue non-coherent light sources, such as halogen lamps, light-emitting diodes (LED), and plasma-arc curing (PAC) lights, are often used *in vivo* and *in vitro* studies. *In vitro* bactericidal efficacy of blue light for the bacteria *Escherichia coli*, MRSA and *P. aeruginosa*, even *Streptococcus mutans* in biofilms has been reported (77). Studies on the *in vitro* antimicrobial effects of blue light are summarized in Table 2.

5.2 Mechanisms of aBL

Broadly speaking the different wavelengths of aBL can be divided into two broad classes. The chromophores that have been proposed to be responsible for the aBL killing of bacteria are shown in Figure 4. Firstly, wavelengths in the range of 390–420 nm are absorbed by the

Soret band of metal-free porphyrins. The molar extinction coefficient of the porphyrin Soret band can be large, around 150,000–250,000 M⁻¹cm⁻¹. However, the specific types of porphyrins can lead to different photo-inactivation rates in different bacteria (i.e. *Staphylococcus*, *Streptococcus*, *Bacillus*, *Escherichia*, *Acinetobacter*, and *Aeromonas*) (78). Both *S. aureus* and *S. epidermidis* produced coproporphyrin as the predominant porphyrin species, whereas there was no predominant porphyrin produced in the Gram-negative *E. coli*, *Acinetobacter*, and *Aeromonas* strains. The amount of coproporphyrin produced by the *Staphylococcal* strains was two to three times higher than in the Gram-negative strains. Therefore, different bacteria produce different porphyrins and the peak absorption wavelengths of these porphyrins may differ. Therefore, different wavelengths may be required for their optimum photo-stimulation. The above-mentioned factors may be responsible for the variation in the bactericidal efficacy of blue light. There has been no comprehensive study of the amounts and identities of the porphyrins produced by bacteria in laboratory cultivation conditions as well as their natural physiological status in the environment, especially bacteria that are infecting a mammalian host (79). Do bacteria that can feed on blood and blood products such as hemin, accumulate more or less porphyrins compared to bacteria that grow on laboratory media? Secondly, wavelengths around 450 nm are absorbed by flavins such as (FMN) and flavoproteins. Although flavins have a double absorption band in the UVA and the blue, they have a minimum about 400 nm. While there are numerous studies looking at aPDI with exogenously added riboflavin and other flavins (80, 81), unequivocal proof that endogenous flavins are responsible for aBL killing is scarce (82). Thirdly, there was an interesting report that in the case of *S. aureus*, the yellow-orange pigment known as staphyloxanthin could act as a blue-light absorbing lethal chromophore (83).

5.3 Resistance to aBL

In order to assess the possible development of tolerance or resistance to antimicrobial blue light alone in *P. aeruginosa*, 10 repeated cycles of sub-lethal inactivation of bacteria in vitro, followed by bacterial re-growth, were carried out by Amin et al (13). This tolerance study showed no evidence of the development of resistance by *P. aeruginosa* to blue light after 10 consecutive cycles of sub-lethal inactivation. Their finding was in agreement with similar findings obtained previously when an *A. baumannii* strain was treated with blue light for 10 consecutive cycles of sub-lethal inactivation (84).

5.4 Blue light toxicity towards host cells and tissue

If the use of blue light for the treatment of infectious disease is to become accepted, it is important to understand the effects of blue light on host cells and tissues so that non-specific damage to the patient can be avoided (or at least not be a concern). However, only limited studies have been reported so far in this area.

In one in vitro study, Hockberguer *et al.* (85) showed that mouse (3T3 fibroblasts), monkey (kidney epithelial cells), and human (foreskin keratinocytes) irradiated with blue light (450nm-nm at 6.3 W/cm²) for 20 min produced hydrogen peroxide (H₂O₂), an important ROS which can cause cellular damage and could contribute to pathologies associated with exposure blue light. Another study was carried out by Wataha et al. (86) The investigators

treated 3T3 mouse fibroblasts with three dental blue (400nm-500 nm) light sources, and detected succinate dehydrogenase (SDH) activity of mitochondria after a range of fluences from 1.3 to 60 J/cm². Results showed that fluences ranging from 5 J/cm² to 15 J/cm² irreversibly suppressed SDH activity nearly 100% when compared to no-light controls up to 72 h post-exposure. A recent clinical study by Kleinpenning et al. (87) found no significant side effects of blue light on human skin. Eight healthy volunteers were irradiated with a blue light source (390–460 nm) with a peak emission of 420 nm on 5 consecutive days at 20 J/cm² daily. To analyze the irradiation effects, skin biopsies were analyzed for p53 expression, vacuolization, sunburn cells, elastosis, matrix metalloproteinase-1 (MMP-1), and melan-A expression. No inflammatory nor sunburn cells were found and no significant changes in p53 and MMP-1 expressions or sign of elastosis were observed after blue light exposure.

5.5 Animal studies of antimicrobial blue light for localized infections

As discussed above in the section about aPDT, we have developed several different small animal models of localized infections produced by stable bioluminescent bacteria and fungi that are suitable for testing light-mediated antimicrobial approaches. These models were ideal to test the in vivo ability of aBL to treat real life infections. Firstly, we tested community acquired MRSA infection of a skin abrasion on the mouse back. Exposure to 40 J/cm² of 415 nm LED on the same day of the infection, or to 108 J/cm² on the day after the infection both led to a reduction of 99% in the bioluminescence signal (88). Next, we studied 3rd degree burns on the mouse back infected with *P. aeruginosa* (89). Application of 55 J/cm² of 415 nm LED on the same day of infection led to a reduction of 99% in the bioluminescence signal, and importantly 100% of the mice survived, as opposed to only 18% of mice in the untreated control group (see Figure 5).

A similar result was achieved with *A. baumannii* infecting a 3rd degree mouse burn (90). *C. albicans* infection in a mouse burn could be reduced by 1.75 log₁₀ units by 432 J/cm² of 415 nm LED (91). The last study utilized an interesting animal model of infectious keratitis caused by inoculation of bioluminescent *P. aeruginosa* onto the surface of the mouse cornea (92) and is shown in Figure 6. The effectiveness of aBL (415 nm LED) was evaluated as a function of radiant exposure when aBL was delivered at 6 or 24 hours after bacterial inoculation. The aBL exposure calculated to reach the retina was within the safety limits laid down by the American National Standards Institute. Bacterial burden in the infected corneas was rapidly and significantly reduced (>2-log₁₀) after a single exposure of 180 J/cm² at 6 hours and 372 J/cm² at 24 hours. These animal studies are summarized in Table 3.

5.6 Clinical application of blue light for infectious diseases

Several studies have reported the benefits of blue light on acne vulgaris, an important dermatologic disorder in which the major pathogen is *P. acnes*. It has long been known that when acne lesions are examined under blue light or UVA light, distinct red fluorescence is observed (93). This fluorescence arises from porphyrins that have been synthesized by the *P. acnes* bacteria and have accumulated in the sebum secretions of the sebaceous glands. Fluorescence correlates better with the amount of sebum, rather than with the amount of *P. acnes* bacteria (94). Therefore, it is at present unclear if the beneficial effects of blue light on

acne rely mainly on PDT killing of the actual bacteria, or on some PDT effects on the sebaceous glands. Nevertheless, studies have shown that blue light does produce a clinically significant effect on acne (95).

Blue light also can inactivate *H. pylori*, which is the major pathogen causing atrophic gastritis and peptic ulcers and is increasingly resistant to antibiotics (96). Ganz et al. (97) treated 10 patients with blue light (405 nm, 40.5J/cm²) via an optical fiber inserted in the endoscope channel that illuminated a 1-cm diameter area of the gastric mucosa. Significant decreases (91%) in *H. pylori* colonies per gram of tissue in treated sites was observed by comparing biopsies from treated and adjacent untreated sites. Some patients showed reductions of bacterial burden approaching 99%. Another pilot trial was carried out by Lembo et al. (98) using a blue (408 nm) laser source with multi-segmented balloon that allowed whole stomach illumination to treat 18 patients with *H. pylori* infection. There was a generalized reduction in bacterial load in the stomach. The largest reduction in bacterial load was found in the antrum (>97%), followed by the body (>95%) and the fundus (>86%) with no apparent dose-response correlation. However, repeated breath tests showed *H. pylori* repopulated in days following illumination. These clinical studies showing that blue light can treat inflammatory and non-inflammatory acne lesions caused by *P. acnes*, and possibly digestive disorders caused by *H. pylori* are summarized in Table 4.

5.7 Other applications of aBL

Many of the proposed applications of aBL have been concerned with areas of antimicrobial killing or sterilization that are not directly connected to the goal of this review (prevention and treatment of antibiotic resistant infections) and might be considered off-topic. Therefore, we will not cover these in detail, but rather give one or two examples.

Medical applications include inactivation of bacterial contamination in ex vivo stored plasma (99), and sterilization of contact lenses (100). Agricultural produce and foodstuffs have been one of the most investigated areas so far. aBL has been studied for killing *E. coli* in milk and *Salmonella* in orange juice (101). There have also been studies of aBL for bacterial contamination on the surfaces of foodstuffs such as fresh-cut papaya (102), fresh-cut mango (103), and processed meat (hot dogs) (104). Another possible application has been to extend the life-time of fruits and vegetables post-harvest, which is usually restricted by the growth of molds or the development of rot (105).

6 Safe UVC

6.1 254 nm UVC

UVC light is electromagnetic irradiation between the wavelengths of 200 to 280 nm. It has been known for more than 100 years that UV light, particularly UVC is highly germicidal (106). UVC light with a wavelength 254-nm is readily produced from a low-pressure mercury vapor lamp, and is commonly used to inactivate and kill many microbial species, where it is popularly known as germicidal UV (107). The antimicrobial mechanism of UVC light relies on its ability to damage DNA caused by production of a variety of mutagenic and cytotoxic DNA lesions such as cyclobutane pyrimidine dimers (CPD) (108) and 6,4

photoproducts (109). CPD are known to interrupt the transcription, translation, and replication of DNA, leading to cell death (110). It has been reported that irradiation of operating rooms with germicidal UVC during orthopedic surgical procedures reduced the rate of surgical site infections (111).

Using mouse models, Dai et al. (112) investigated the potential of UVC light for the treatment of infections in highly contaminated superficial cutaneous wounds. Mouse models of partial-thickness skin abrasions infected with bioluminescent *P. aeruginosa* and *S. aureus* were developed. Approximately 10^7 bacterial cells were inoculated onto wounds measuring 1.2×1.2 cm on the dorsal surfaces of mice. UVC light was delivered at 30 min after bacterial inoculation. It was found that for both bacterial infections, UVC light at 2.59 J/cm^2 significantly reduced the bacterial burden in the infected mouse wounds by 10-fold in comparison to untreated wounds (112). Furthermore, UVC light increased the survival rate of mice infected with *P. aeruginosa* (58%) and increased the wound healing rate in mice infected with *S. aureus* (31%). In another study, Dai et al. (113) investigated the use of UVC irradiation (254 nm) for treatment of bioluminescent *C. albicans* infection in mouse third degree burns. A single exposure carried out on day 0 (30 min post-infection) gave an average 2.16-log_{10} (99%) loss of fungal luminescence when 2.92 J/cm^2 UVC had been delivered, while UVC at 24 h post-infection gave 1.94-log_{10} (96%) reduction of fungal luminescence after 6.48 J/cm^2 UVC was found to be superior to a topical anti-fungal drug, nystatin cream.

Nevertheless, it has also been shown that exposure of human cells to 254-nm UVC causes the formation of mutagenic and cytotoxic DNA lesions, which (if repeated for a sufficiently long time) can lead to the initiation and progression of skin cancer [12]. Although these DNA lesions can be rapidly and efficiently repaired by DNA repair enzymes usually by 24 hours and entirely by 48 hours, it would indeed be preferable if these DNA lesions could be avoided.

6.2 Safe short wavelength UVC

A new approach uses excimer lamps, which are quasimonochromatic light sources that can operate over a wide range of wavelengths in the UV and vacuum ultraviolet (VUV) spectral regions. The operation of excimer lamps is based on the formation of excited dimers (excimers) between two of the same atoms, or exciplexes formed between two or more dissimilar atoms (114). The best-known examples include rare gas excimers and rare-gas-halogen excimers. When the excited state semi-stable dimer splits apart returning to the ground state, a UV-photon is emitted. For example, Xe_2^* , Kr_2^* , Ar_2^* are excimer molecules, but XeCl^* , KrCl^* , XeBr^* , ArCl^* , Xe_2Cl^* are exciplex molecules. The spectrum of excimer lamp radiation is characterized by an intense narrow emission band (115). The full-width at half maximum of these emission bands depends on the kind of working molecule and excitation conditions and ranges from 2 to 15 nm.

The underlying rationale for “safe UVC” is that as the wavelength drops below that of the most commonly used UVC wavelength (namely the low-pressure mercury lamp at 254 nm) then the absorption coefficient of protein rises sharply, while the absorption coefficient of nucleic acids remains about the same. The consequence of this change in absorption, is that a 222-nm photon has the same chance of damaging a DNA molecule as a 254 nm photon,

but cannot penetrate as well through the protein that composes the cytosol of an eukaryotic cell. Since eukaryotic cells are considerable larger than prokaryotic bacterial cells, there is much more protein inside them. In eukaryotic cells the DNA is concentrated inside the nucleus which is separated from the plasma membrane. In prokaryotic cells the DNA is contained in the nucleoid, a region of cytoplasm where the chromosomal DNA is located. Bacteria do not have a membrane bound nucleus, but simply an area of the cytoplasm where the strands of DNA are found. Most bacteria have a single, circular chromosome that undergoes replication, although a few species do have two or more chromosomes. Smaller circular DNA strands, called plasmids, are also found in the cytoplasm.

It has been reported that the penetration ability of short-UVC light is reduced by half in only about 0.3 μm of tissue (116). Buananno and co-workers undertook a series of studies with 207-nm UVC light produced by a krypton-bromine excimer lamp (117, 118). They showed that 207 nm can inactivate bacteria efficiently (>99.9% killing), moreover this wavelength also showed less cytotoxic and mutagenic damage to human keratinocytes compared to 254 nm on the basis of equal energy density delivered.

Recently, a krypton-chlorine (Kr-Cl) excimer lamp was developed to emit 222-nm UVC light (114). We evaluated the bactericidal and DNA-damaging effects of 222-nm UVC light compared with 254 nm UVC light in MRSA-infected wounds in a mouse model (119). We demonstrated that 222-nm UVC light at 75 and 150 mJ/cm^2 showed efficient bactericidal activity on both normal mouse skin and in MRSA-infected wounds. Although the bactericidal activity of 222-nm UVC measured immediately after irradiation in infected wounds was slightly less than that of 254-nm UVC, the bacterial numbers in the wounds on days 5, 8 and 12 in 222-nm UVC wounds were comparable to or lower than those in 254-nm UVC wounds. Importantly, DNA damage to keratinocytes and other cells in the infected wound caused by 222-nm UV light was less severe than that caused by 254-nm UVC. Figure 7 illustrates the use of 222 nm safe UVC to kill bacteria in vitro and in vivo.

Because the skin surface is not completely smooth and is covered by intersecting grooves called “sulci cutis”, it might be possible that *S. aureus* cells were distributed in such a way within these sulci cutis grooves, or in hair follicles and were not irradiated with a sufficient dose of UVC to elicit the full bactericidal effect. In other words some bacteria could have been shielded by the skin surface irregularities.

It is known that wound exudate, a protein-rich fluid produced in response to tissue damage, is present in wounds. It contains various types of cells and many kinds of proteins including inflammatory mediators as well as proteolytic enzymes [29]. Thus it might be possible that proteins in the wound exudate absorb UVC light thus protecting the bacteria, and the remaining bacterial cells which were not eradicated, could proliferate and grow in the wound after UV irradiation.

In the wound healing process, epithelial cells from the wound edge begin to migrate over the surface of wound “burrowing a path” beneath the scab and the underlying granulation tissue (120). It has been previously reported that UV exposure might be beneficial for wound healing by inducing hyperplasia, granulation tissue formation, sloughing of necrotic tissue

and enhancing re-epithelialization (121). Our results demonstrated that irradiation with 254-nm UVC but not with 222-nm UVC impaired the burrowing or migration of keratinocytes across the wound suggesting that irradiation with 254-nm but not 222-nm UVC can induce DNA damage and inflammation in wounded dermis as well as epidermis. Although further investigation will be required for any application of 222-nm UVC light in clinical settings, the Xe-CI lamp is promising to reduce SSI and wound infections especially by multi-drug resistant bacteria. It remains to be tested whether “safe UVC” is sufficiently safe to make the adoption of skin and eye protection less critical than it is with 254 nm UVC.

7 Conclusions

This review has highlighted three different applications of light that can all be used in the fight against drug-resistant pathogens. Each of them has their own particular strengths and weaknesses. aPDT is the most complicated as it requires the use of a separate exogenous PS, and moreover if KI potentiation is used, the addition of a sufficient concentration of KI solution. It can however deliver complete eradication at relatively low fluences, and red light can penetrate well into tissue. It is at present uncertain if the PS and the KI can be combined into some type of nano-drug delivery system such as liposomes or micelles. This may be difficult as a high concentration of KI (100s of mM) is required to obtain the best potentiation effect. The use for treating a bladder infection was attractive, as the bladder is ideally suited to contain an aqueous solution at the same time as light is delivered (64). However, the urothelial lining of the bladder may be more sensitive to the damaging effects of PDT compared to skin and soft tissue.

Blue light alone has the advantages of being simple to deliver, and modern blue LEDs can provide substantial power densities from portable devices in a reliable manner. However, the penetration depth of blue light into tissue is very limited (< 1mm) and other approaches will be necessary to get sufficient blue light to infections that are anything else than purely superficial. Interstitial optical fibers, light delivery microneedles, and tissue optical clearing are some possibilities (66). UVC penetrates even less than blue light but its antimicrobial killing efficiency is several orders of magnitude higher. Even though UVC devices are clinically approved for treatment of wounds, they have not been much used in wound care because of fears about the possible toxic and carcinogenic effects of UVC. Perhaps the introduction of “safe UVC” at 207 nm or 222 nm will overcome this reluctance.

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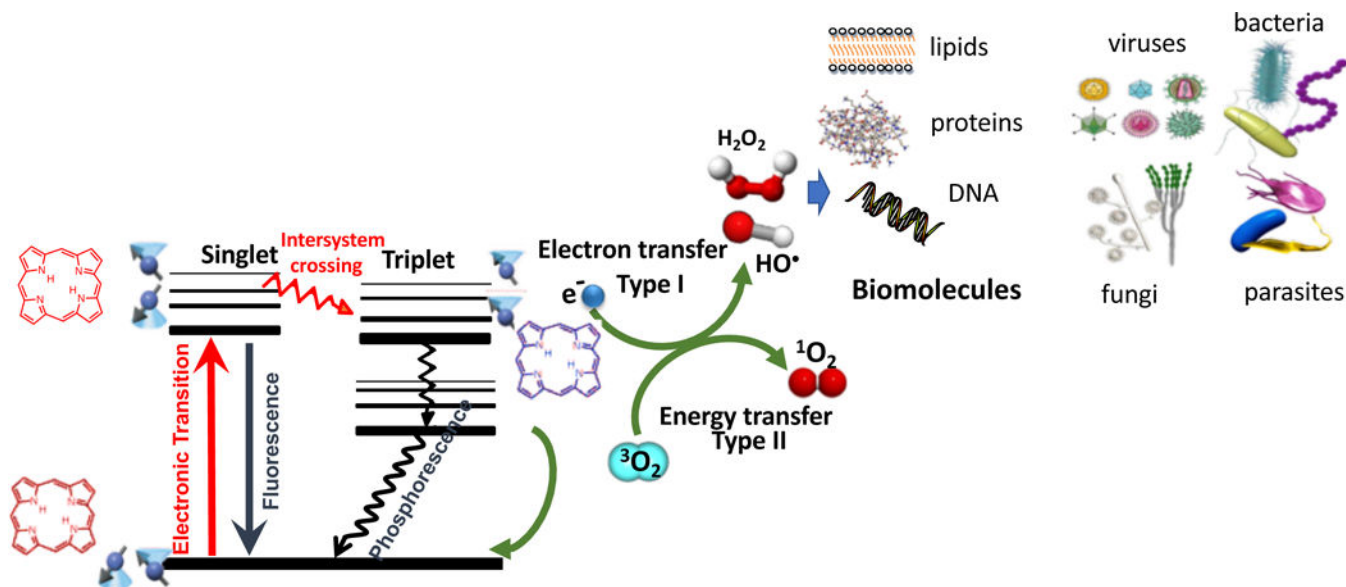


Figure 1. Jablonski diagram

A ground-state PS absorbs a photon, transitions to the short-lived (nsec) excited singlet state that can undergo intersystem crossing to the long-lived (μsec) excited triplet state. The triplet PS can undergo energy transfer with ground state triplet oxygen ($^3\text{O}_2$) to form reactive singlet oxygen ($^1\text{O}_2$, Type 2) or else can undergo an electron transfer reaction to form HO \cdot , superoxide and H $_2\text{O}_2$ (Type 1). These ROS ($^1\text{O}_2$ and HO \cdot) can damage lipids, proteins and nucleic acids leading to destruction of all types of microbial cells, viruses and parasites.

**Figure 2. PDT experiment**

Illustration of an aPDT experiment in a mouse model of an infected burn produced by two heated brass blocks. Generally the bacteria are applied to the surface of the burn by a pipette, followed after a short time by application of the PS in a similar fashion, and after another relatively short time by light delivery.

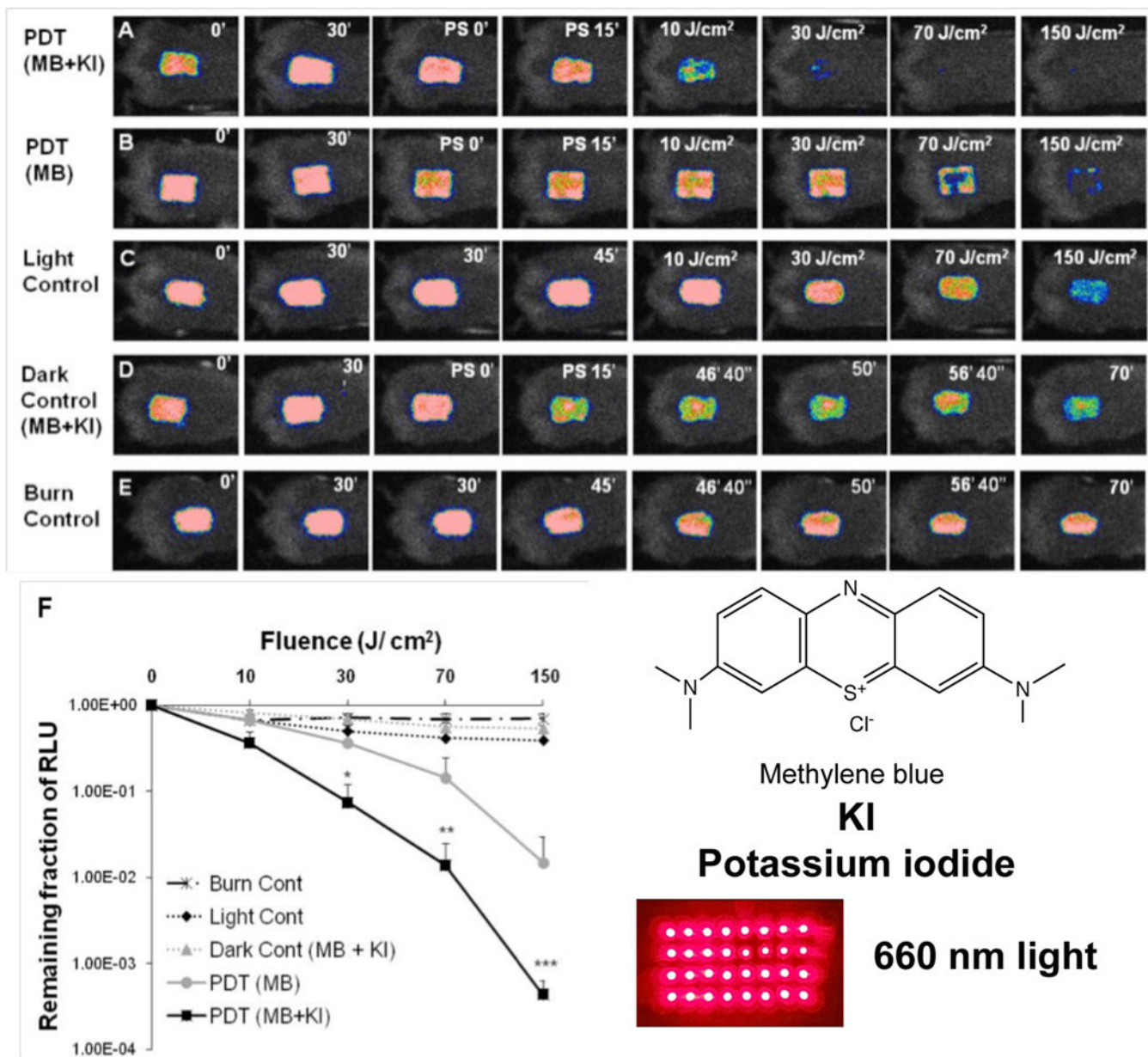


Figure 3. PDT mediated by MB plus KI in vivo

(A-E) Successive bacterial bioluminescence images of representative mouse burns infected with $10(8)$ CFU of luminescent MRSA (USA 300) treated with: (A) PDT using mixture of MB ($50 \mu M$) + KI ($10 mM$) or (B) PDT using MB ($50 \mu M$) at 30 min after bacterial inoculation + 15 min from PS application. PDT was carried out with a combination of $50 \mu L$ of a mixture containing MB + KI or MB alone and $150 J/cm^2$ red light ($660 \pm 15 nm$, $100 mW/cm^2$). (C) Light alone; (D) applied with mixture of MB + KI, but without red light illumination (dark control); (E) burn control without any treatment. (F). Dose-response plot of mean bacterial bioluminescence of mouse burns infected with MRSA (USA 300) after treatment with: light alone, mixture of MB ($50 \mu M$) + KI ($10 mM$) (dark control), PDT using

MB (50 μ M) alone or mixture of MB (50 μ M) + KI (10 μ M). From (45) no permission necessary.

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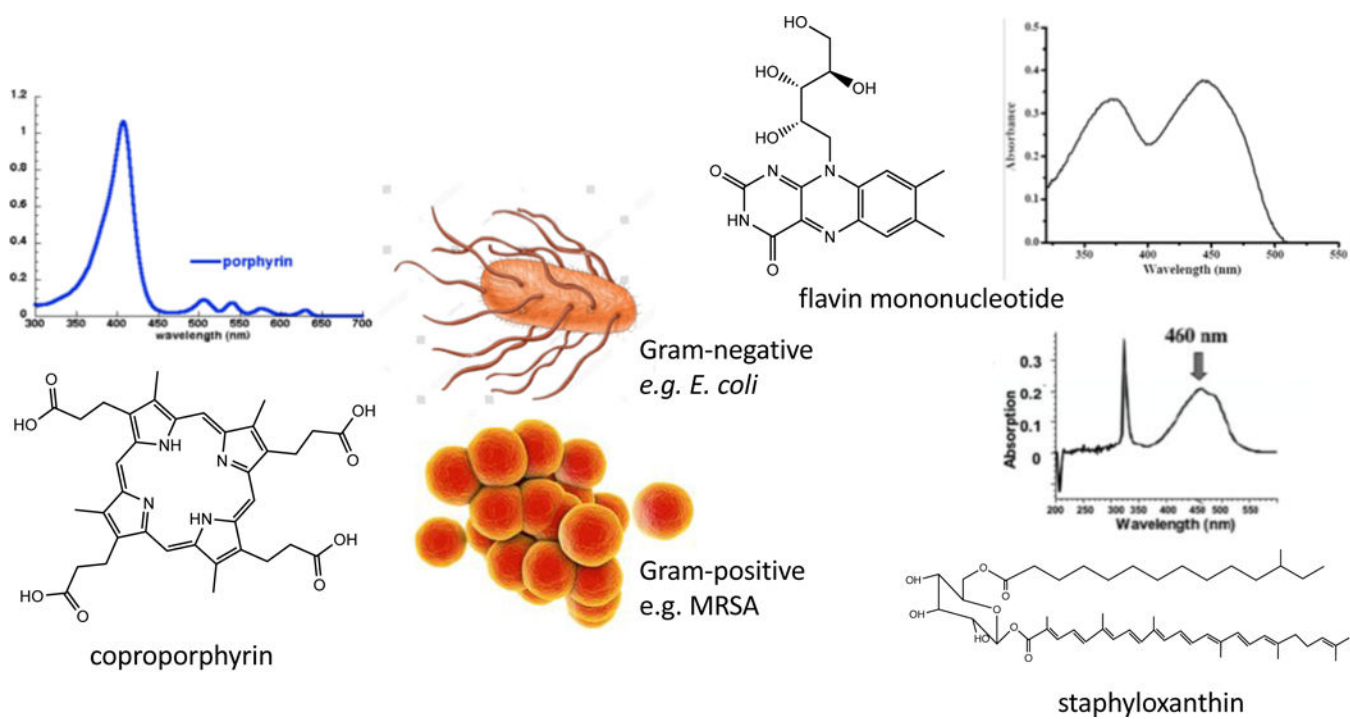


Figure 4. Chromophores for blue light inactivation of bacteria
 Chemical structures and absorption spectra of coproporphyrin, flavin mononucleotide, and staphyloxanthin.

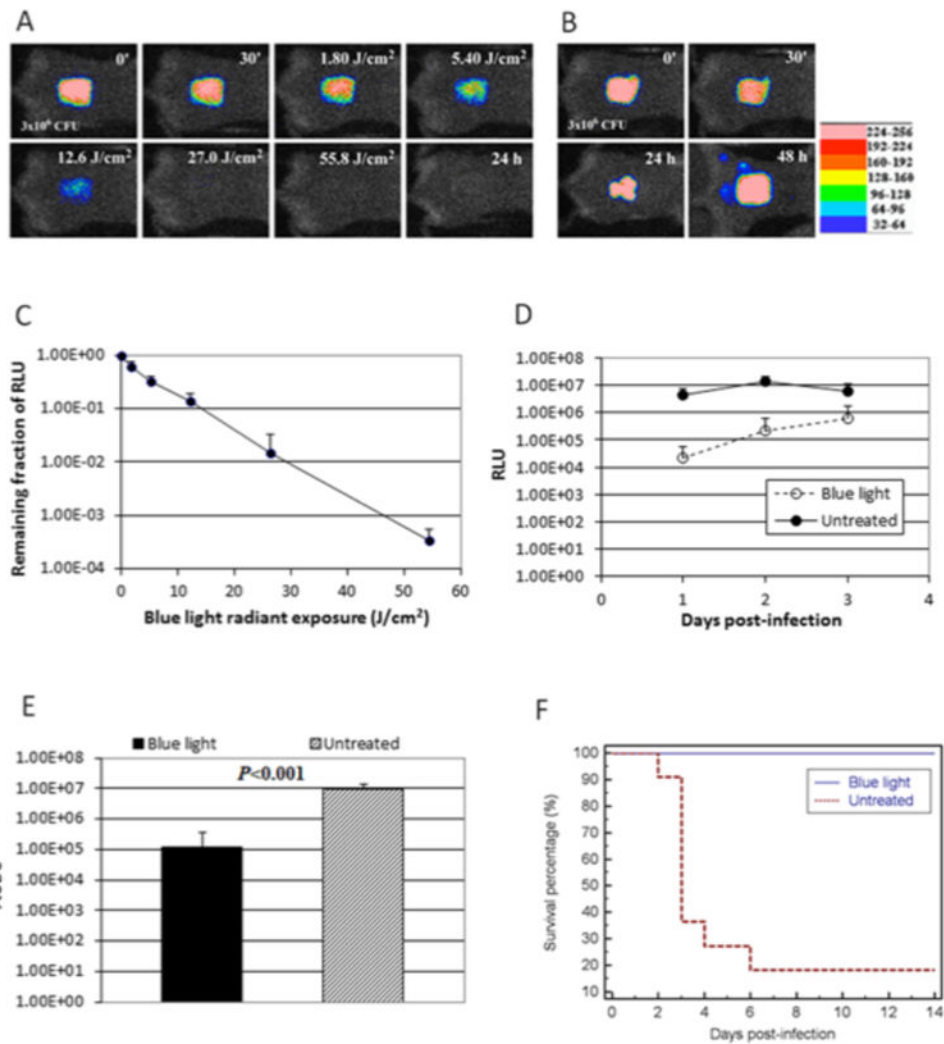


Figure 5. Wound infection treated with aBL

A, B. Successive bacterial luminescence images of representative mouse burns infected with 3×10^6 CFU of luminescent *P. aeruginosa* with and without blue light (415 nm) exposure, respectively. Blue light was delivered at 30 min after bacterial inoculation. (C) Dose response of mean bacterial luminescence of mouse burns infected with 3×10^6 CFU of *P. aeruginosa* and exposed to blue light (415 nm) at 30 min after bacterial inoculation. (D) Time courses of mean bacterial luminescence of the infected skin abrasions with and without blue light exposure, respectively (E) Mean areas under the bioluminescence versus time curves (from day 1 to day 2 in the two-dimensional coordinate system in panel D), representing the overall bacterial burden of mouse wounds. Bars, standard deviations. (F) Kaplan-Meier survival curves of blue light-treated ($n = 11$) and untreated ($n = 11$) mouse burns ($P = 0.0001$). From (89) no permission necessary.

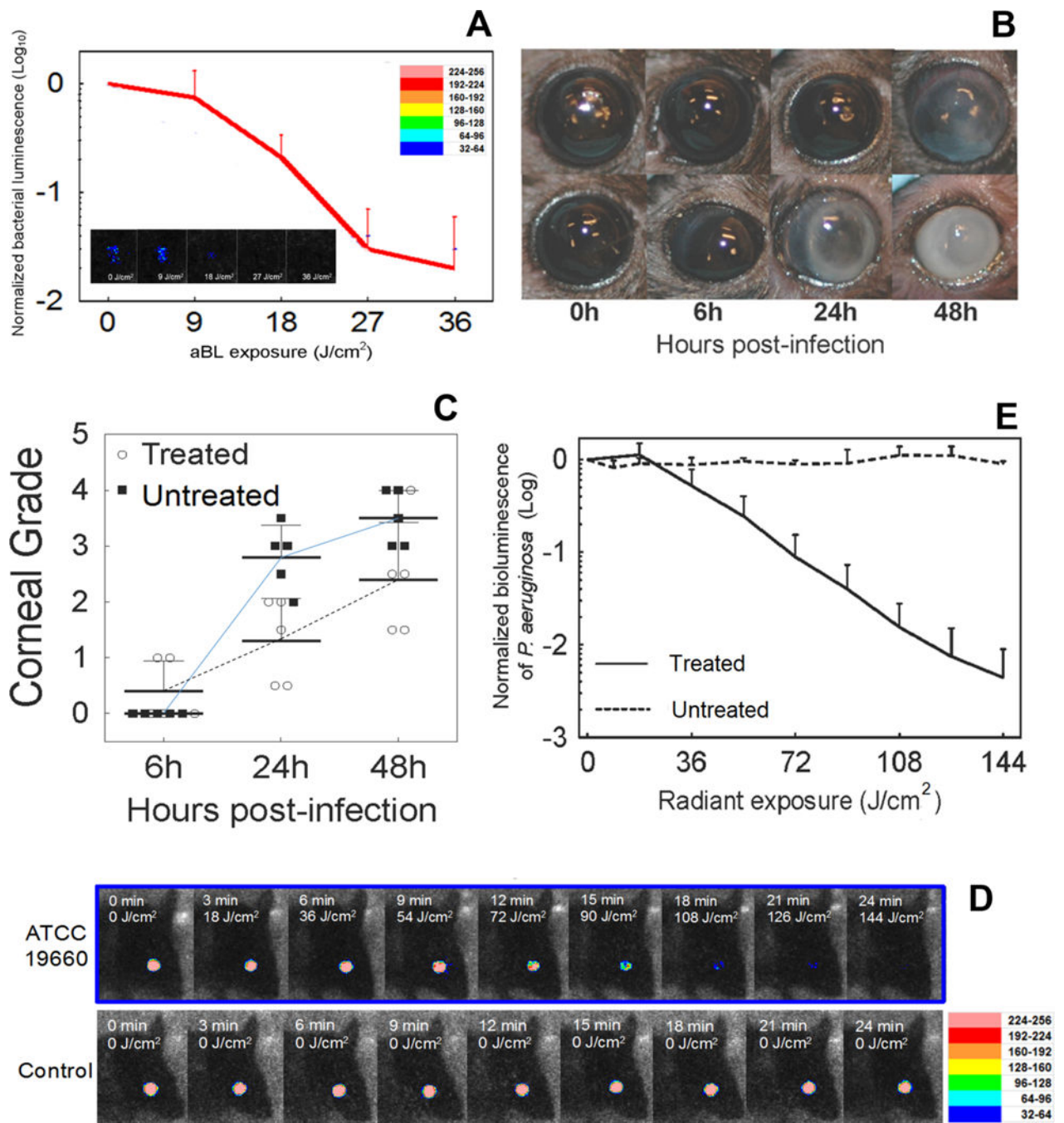


Figure 6. aBL for infectious keratitis

(A) Influence of radiant exposure of aBL on the bacterial luminescence intensity of mouse eyes infected with 5×10^5 CFU *P. aeruginosa* and treated with $36 \text{ J}/\text{cm}^2$ aBL at 6 hours post inoculation. The inset shows the bacterial luminescence images from a representative mouse eye infected with 5×10^5 CFU *P. aeruginosa* and treated with $36 \text{ J}/\text{cm}^2$ aBL at 6 hours post inoculation. Bacterial luminescence images were taken when 0, 9, 18, 27, and $36 \text{ J}/\text{cm}^2$ aBL had been delivered. The pseudo-color scale bar indicates the bioluminescence intensity value of a single pixel in the bioluminescence images. (B) Photographs of two representative

mouse eyes infected with 5×10^5 CFU *P. aeruginosa* treated with aBL (36 J/cm^2) at 6 hours post infection (top row) or not treated (bottom row), indicating less opacity and damage in the aBL-treated mouse eye than in the untreated one. (C) Corneal pathology scores of the infected mouse eyes treated with 36 J/cm^2 aBL at 6 hours post infection or not treated showing the higher mean corneal pathology scores of the untreated mouse eyes than the aBL-treated mouse eyes Analysis of the interaction between aBL treatment and time post infection shows that the effect of aBL treatment increased with the time post infection. Data were analyzed using the general linear model. (D) The bacterial luminescence images of representative mouse eyes infected with 5×10^5 CFU *P. aeruginosa*, treated with 144 J/cm^2 aBL 24 hours later or left untreated. (E) Influence of radiant exposure of aBL on the bacterial luminescence intensity of the mouse eyes infected with 5×10^5 CFU *P. aeruginosa*, treated with 144 J/cm^2 aBL at 24 hours post inoculation or left untreated. Bacterial luminescence images were taken when 0, 18, 36, 54, 72, 90, 108, 126, and 144 J/cm^2 aBL had been delivered. From (92) no permission necessary.

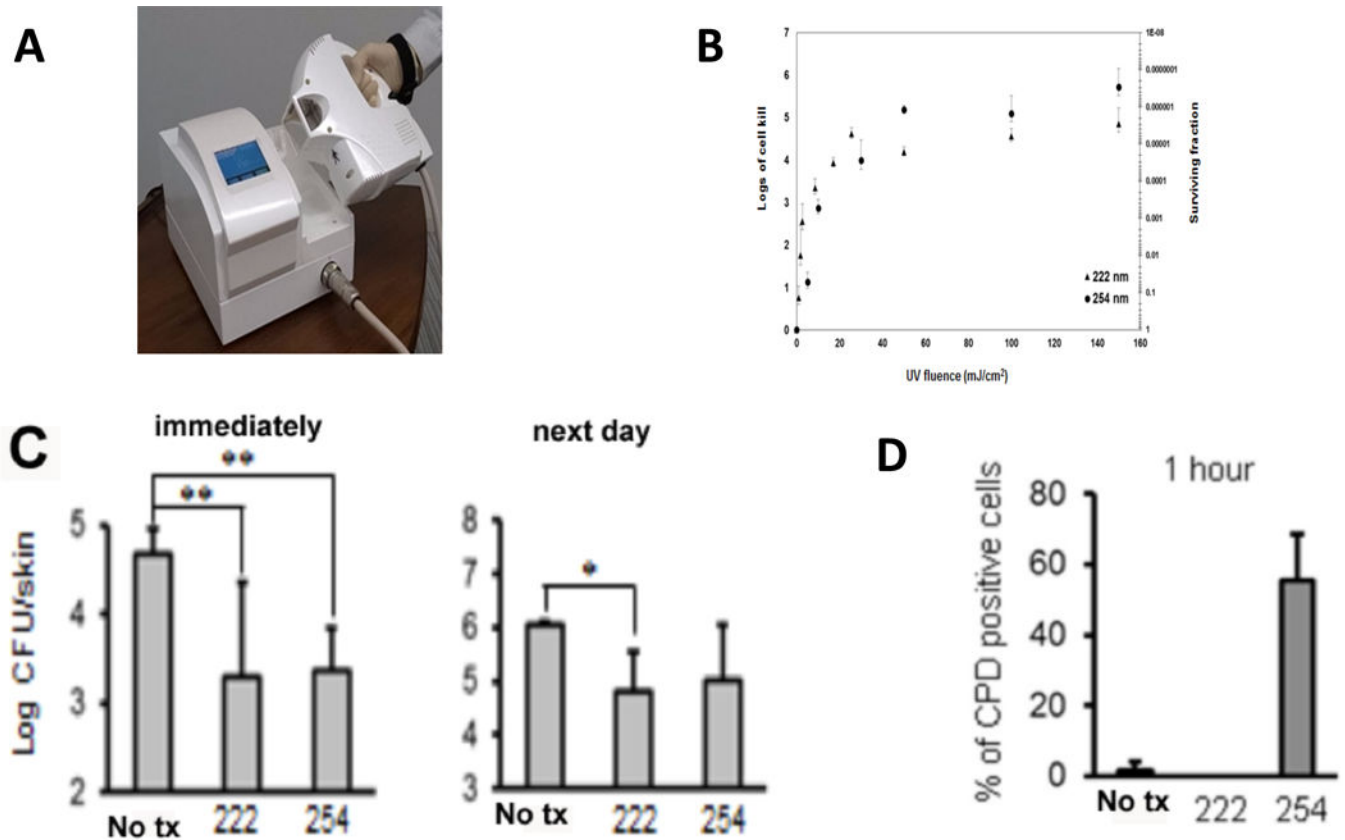


Figure 7. Comparison of 222 nm and 254 nm UVC

(A) Photograph of 222 nm Kr/Cl excimer lamp. (B) Comparison of 222 nm and 254 nm UVC for in vitro inactivation of *S. aureus* on an agar plate. (C) *S. aureus* CFU extracted from a mouse wound either immediately or 24 hours after treatment with 750 mJ/cm² of either 222 nm or 254 nm UVC. (D) % of CPD-positive cells quantified by counting 10 random high-power (×400) fields of tissue sections removed 1 hour post exposure to 150 mJ/cm² of either 222 nm or 254 nm UVC. From (119) no permission necessary.

Table 1. Timeline of the introduction of new antibiotics and the first reports of resistance emerging.

Antibiotic introduced	Antibiotic resistance identified
1943 penicillin	1940 penicillin-resistant <i>Staphylococcus</i>
1950 tetracycline	1959 tetracycline-resistant <i>Shigella</i>
1953 erythromycin	1978 erythromycin-resistant <i>Pneumococcus</i>
1960 methicillin	1962 methicillin-resistant <i>Staphylococcus</i>
1967 gentamycin	1979 gentamycin-resistant <i>Enterococcus</i>
1972 vancomycin	1988 vancomycin-resistant <i>Enterococcus</i>
1985 ceftazidime	1987 ceftazidime-resistant <i>Enterobacteriaceae</i>
1985 imipenem	1998 imipenem-resistant <i>Enterobacteriaceae</i>
1996 lefloxacin	1996 lefloxacin-resistant <i>Pneumococcus</i>
2000 linezolid	2001 linezolid-resistant <i>Staphylococcus</i>
2003 daptomycin	2005 daptomycin-resistant <i>Staphylococcus</i>
2010 ceftaroline	2011 ceftaroline-resistant <i>Staphylococcus</i>

Table 2.

Studies of the in vitro antimicrobial effect of blue light

Light Source	Radiant exposure	Bacterial species/strains	Inactivation efficacy	Ref
405-nm diode laser	20 J/cm ²	<i>H. pylori</i>	>99.9%	Hamblin <i>et al.</i> (70)
405 nm light-emitting device	15 J/cm ² at lamp aperture	<i>P. gingivalis</i>	>75%	Fukui <i>et al.</i> (122)
380–520 nm broadband light	4.2–42 J/cm ²	<i>P. gingivalis</i> , <i>P. intermedia</i> , <i>P. nigrescens</i> , <i>P. elainogenica</i> , <i>S. constellatus</i>	<i>P. intermedia</i> and <i>P. nigrescens</i> : >5 log ₁₀ at 4.2 J/cm ² ; <i>P. melaninogenica</i> : >5 log ₁₀ at 21 J/cm ² ; <i>P. gingivalis</i> : 1.83 log ₁₀ at 42 J/cm ²	Soukos <i>et al.</i> (123)
400–500 nm lamps	260 and 1300 mW/cm ² for up to 3 min	<i>P. gingivalis</i> , <i>F. nucleatum</i> , <i>S. mutans</i> , <i>E. faecalis</i>	The minimal inhibitory dose for <i>P. gingivalis</i> and <i>F. nucleatum</i> was 16–62 J/cm ² ; for <i>S. mutans</i> and <i>S. faecalis</i> was 159–212 J/cm ²	Feuerstein <i>et al.</i> (124)
405-nm superluminescent diode light	50.4–55.2 J/cm ²	MRSA USA 300; MRSA IS-853	92.1% for USA 300; 93.5% for IS-853	Enwemeka <i>et al.</i> (125)
470-nm superluminescent diode light	55 J/cm ²	MRSA USA 300; MRSA IS-853	90.4% for both strains	Enwemeka <i>et al.</i> (126)
405 and 470 nm light	15 J/cm ²	<i>S. aureus</i> , <i>P. aeruginosa</i>	<i>S. aureus</i> : 90% at 405 nm, 62% at 470 nm; <i>P. aeruginosa</i> : 95.1% at 405 nm, 96.5% at 470 nm	Guffey and Wilborn (127)
407–420 nm		five <i>P. acnes</i> strains	decreased by 15.7% immediately and 24.4% at 60 min after the irradiation.	Kawada <i>et al.</i> (128)
407–420 nm	75 J/cm ²	<i>P. acnes</i>	less than 2-log ₁₀ units (99%) illuminated once; decreased by 4-log ₁₀ units (99.99%) after two illuminations and by 5-log ₁₀ units (99.999%) after three illuminations	Ashkenazi <i>et al.</i> (129)
390–410 nm LED	54–108 J/cm ²	<i>A. baumannii</i> , <i>Enterobacter cloacae</i> , <i>Stenotrophomonas maltophilia</i> , <i>P. aeruginosa</i> , <i>E. coli</i> , <i>S. aureus</i> , <i>E. faecium</i> , <i>Klebsiella pneumoniae</i> , <i>Elizabethkingia meningoseptica</i> .	Most (71%) showed a decrease of 5 log ₁₀ units	Halstead <i>et al.</i> (130)
405–425nm LED	110 J/cm ²	<i>A. baumannii</i>	7.64-log ₁₀ CFU	Dai <i>et al.</i> (89)
405–425nm LED	70 J/cm ²	<i>P. aeruginosa</i> , <i>A. baumannii</i> , methicillin-resistant <i>S. aureus</i> (MRSA), <i>Candida albicans</i>	> 4-log ₁₀ CFU	Zhang <i>et al.</i> (90, 91)
393–413 nm LED	133 J/cm ²	<i>S. aureus</i> , <i>S. pneumoniae</i> , <i>E. coli</i> , <i>P. aeruginosa</i>	4.75-log ₁₀	Bameck <i>et al.</i> (131)
	118 – 2214 J/cm ²	<i>S. aureus</i> , <i>S. epidermidis</i> , <i>E. faecalis</i> ,	Complete inactivation (> 4-log ₁₀ CFU) in suspension was achieved in all of the isolates tested.	Gupta <i>et al.</i> (132)

Light Source	Radiant exposure	Bacterial species/strains	Inactivation efficacy	Ref
395–414 nm LED	68 J/cm ²	<i>S. pneumoniae</i> , <i>Corynebacterium striatum</i> , <i>E. coli</i> , <i>K. pneumoniae</i> , <i>P. aeruginosa</i> , <i>Serratia marcescens</i> <i>C. albicans</i> , β -lactam-resistant <i>E. coli</i>	> 6-log ₁₀ CFU reduction	Rhodes et al (133)

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Table 3.

Reports of the in vivo use of aBL in animal models of localized infection

Light Source	Animal model	Bacterial species/strains	Inactivation efficacy	Ref
405–425 nm LED	Infected skin abrasion on mouse back	Bioluminescent MRSA USA300LAC::lux	> 2 log ₁₀ reduction of bioluminescence after 40 J/cm ² (day 0) or 108 J/cm ² (day 1)	Dai et al (88)
405–425 nm LED	Infected 3 rd degree burn on mouse back	Bioluminescent <i>P. aeruginosa</i>	Bioluminescence was reduced 100-fold by 55 J/cm ² . Mouse survival was 100% vs 18.2% in untreated mice	Dai et al (89)
405–425 nm LED	Infected 3 rd degree burn on mouse back	Bioluminescent <i>A. baumannii</i>	55.8 J/cm ² reduced bacterial levels in burn	Zhang et al (90)
405–425 nm LED	Infected 3 rd degree burn on mouse back	Bioluminescent <i>C. albicansi</i>	A single exposure of 432 J/cm ² aBL reduced the fungal burden in infected mouse burns by 1.75-log ₁₀	Zhang et al (91)
405–425 nm LED	Infectious keratitis in mice	Bioluminescent <i>P. aeruginosa</i>	Bacterial luminescence in the infected corneas was reduced (>2-log ₁₀) after a single exposure of aBL.	Zhu et al (92)

Table 4

Clinical studies with antimicrobial blue light

Light source	Clinical trial type	Disease	Treatment regimen	No. of patients/ acne location	Total follow-up time	outcome	Side-effects	Ref
407–420 nm metal halide lamp	open clinical trial	Mild to moderate acne	twice a week up to 5 weeks	30	8 weeks	Acne lesions were reduced by 64%.	Two patients experienced dryness.	(128)
Handheld 412 nm LED	prospective, single-center, open-label study	Mild-to-moderate facial acne.	2J/cm ² /day full-face treatment	32	8 weeks	100 percent of subjects improved,	Three minimal adverse events	(134)
home use blue-LED	IRB approved randomized self-control study	Inflammatory facial acne	Four Treatments twice daily	30	until resolution	significant reduction in lesion size and erythema and improvement	N/A	(135)
hand-held, blue- LED	open clinical trial	mild-to-moderate facial acne	twice daily dose of ~29 J/cm ² ,	33	8 weeks	90% improvements	N/A	(136)
407 to 420 nm LED	prospective, randomized, open comparative study	inflammatory facial acne grades II or III	8 sessions of light therapy for 15 minutes each, twice a week,	60	4 weeks	Equivalent improvement to benzoyl peroxide	23.3% patients with mild desquamation and/or dryness	(137)
415 nm LED	open-label study		48 J/cm ² , twice a week for 4–8 weeks	45	8 weeks	mean improvement score was 3.14 at 4 weeks and 2.90 at 8 weeks	N/A	(138)
409–419 nm LED	open-label study	mild to moderate acne	eight 10–20 treatments over 4 weeks	30	12 weeks	Significant decrease in lesion count continued to week 12.	N/A	(139)
405 nm laser	controlled, prospective, blinded, trial	<i>H. pylori</i> stomach infection	40 J/cm ² was delivered to a 1-cm diameter spot in the gastric antrum via optical fiber passed through an endoscope	10	Comparison of biopsies removed at treatment	mean reduction in CFU/g tissue was 91%	none	(97)
408 nm laser	controlled, prospective pilot trial	<i>H. pylori</i> stomach infection	Whole stomach illumination via endoscopic balloon and diffusing fibers 10–30 kJ	18	Comparison of biopsies removed at treatment and urea breath test	Reduction in CFU/g tissue ranged from 93% to 86% depending on location. Urea breath test dropped after Tx but recurred after 5 weeks	Transient, minor	(98)