

## Chemical Probes for Redox Signaling and Oxidative Stress

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### Abstract

**Significance:** Cellular reactive oxygen species (ROS) mediate redox signaling cascades that are critical to numerous physiological and pathological processes. Analytical methods to monitor cellular ROS levels and proteomic platforms to identify oxidative post-translational modifications (PTMs) of proteins are critical to understanding the triggers and consequences of redox signaling.

**Recent Advances:** The prevalence and significance of redox signaling has recently been illuminated through the use of chemical probes that allow for sensitive detection of cellular ROS levels and proteomic dissection of oxidative PTMs directly in living cells.

**Critical Issues:** In this review, we provide a comprehensive overview of chemical probes that are available for monitoring ROS and oxidative PTMs, and we highlight the advantages and limitations of these methods.

**Future Directions:** Despite significant advances in chemical probes, the low levels of cellular ROS and low stoichiometry of oxidative PTMs present challenges for accurately measuring the extent and dynamics of ROS generation and redox signaling. Further improvements in sensitivity and ability to spatially and temporally control readouts are essential to fully illuminate cellular redox signaling. *Antioxid. Redox Signal.* 30, 1369–1386.

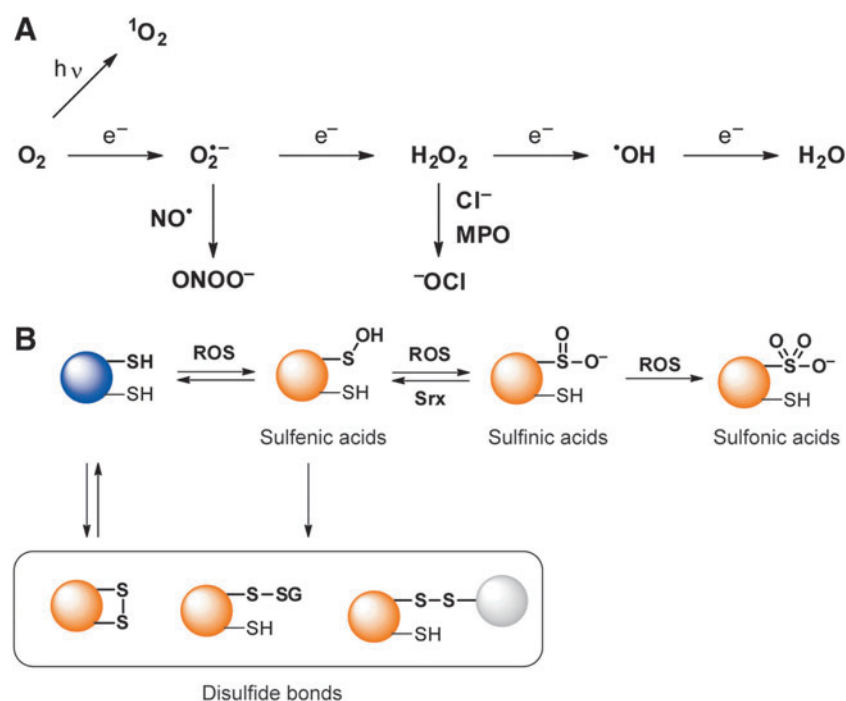
**Keywords:** oxidative stress, cysteine oxidation, redox proteomics, redox sensors

### Introduction

REACTIVE OXYGEN SPECIES (ROS) are produced endogenously during a variety of cellular processes, and primary among these is aerobic respiration. In eukaryotic systems, the majority of cellular ROS is believed to be generated during the transfer of electrons through components of the electron transport chain localized in the inner mitochondrial membrane. This electron transport process results in the reduction of molecular oxygen ( $O_2$ ) to water, and it drives adenosine triphosphate (ATP) synthesis to fuel key physiological processes. However, partial reduction of  $O_2$  can result in the production of numerous ROS, including superoxide ( $O_2^{\bullet-}$ ) and hydrogen peroxide ( $H_2O_2$ ) (Fig. 1A). Further, controlled ROS production also occurs at reduced nicotinamide adenine dinucleotide phosphate (NADPH) oxidase (NOX) enzymes, which oxidize NADPH to nicotinamide adenine dinucleotide phosphate ( $NADP^+$ ), and reduce  $O_2$  to produce  $O_2^{\bullet-}$  as a primary product. In various physiological processes, including cellular proliferation and migration, cytosolic regulatory subunits are assembled on membrane catalytic NOX proteins to activate ROS production (60, 118).

Importantly, localization of NOX subunits provides spatial regulation within cells, such as the assembly of regulatory and catalytic subunits at the leading edge of migrating cells (125).

To counteract the damaging effects of excessive ROS generation, organisms possess a system of small-molecule antioxidants, including ascorbic acid and glutathione, as well as redox enzymes, including glutaredoxins (Grxs), thioredoxins (Trxs), and peroxiredoxins (Prxs) (126). In unicellular organisms, ROS sensors trigger activation of antioxidant systems when ROS concentrations exceed physiological levels. For example, the  $H_2O_2$  sensors OxyR (*e.g.*, prokaryotes) and Orp1 (*e.g.*, yeast) directly sense peroxide levels and activate the production of antioxidant enzymes such as catalase (18). In multicellular organisms, antioxidant enzymes play important roles in spatiotemporal regulation of redox signaling. Evidence suggests that the subcellular localization of antioxidant enzymes restricts ROS to a particular cellular compartment (126). In addition, antioxidant enzymes such as Prxs are inactivated by oxidation or phosphorylation, and they form temporal “hot spots” for redox signaling through feedback inhibition (102, 133).



**FIG. 1. ROS and oxidative PTMs of cysteine.** (A) ROS are derived from molecular oxygen through excitation and reduction. (B) Oxidative PTMs of cysteine highlighted in this review. PTMs, post-translational modifications; ROS, reactive oxygen species. Color images are available online.

These highly controlled systems of cellular ROS generation and attenuation result in strict spatiotemporal regulation of cellular ROS levels. It was initially believed that the high reactivity of ROS resulted mainly in irreversible damage to biomolecules; however, it is now realized that ROS can play critical roles in cell signaling and regulation. This process, known as redox signaling, is a key component of numerous physiological and pathological processes, including growth-factor signaling and inflammation. One of the major effectors of ROS in redox signaling are thiol groups of cysteine residues in proteins, which can be transformed to form distinct oxidative post-translational modifications (PTMs) such as S-sulfenylation, S-sulfinylation, and S-sulfonylation, as well as disulfide-bond formation (Fig. 1B). S-sulfenylation (sulfenic acid [RSOH]) is a reversible PTM under cellular redox potentials, at which sulfenic acids can readily be reduced to the corresponding thiol by cellular antioxidants. Sulfenic acids can also be converted to intra- or intermolecular disulfide bonds with thiol groups of redox-active small molecules and proteins. Further oxidation of sulfenic acids results in S-sulfinylation and S-sulfonylation. Sulfinic acids (RSO<sub>2</sub>H) are only reversed by sulfiredoxin (Srx) (14), and sulfonic acids (RSO<sub>3</sub>H) are considered irreversible in cells. Among the oxidative PTMs of cysteine (Cys), the reversibility of S-sulfenylation renders it suitable for temporal signal transductions akin to phosphorylation. Well-studied examples of S-sulfenylation include protein tyrosine phosphatase 1B (PTP1B) and epidermal growth factor receptor (EGFR). PTP1B is inhibited by H<sub>2</sub>O<sub>2</sub> through S-sulfenylation of the catalytic Cys215 on growth-factor stimulation (63), and EGFR is activated through S-sulfenylation of Cys797 (93). Both these S-sulfenylation events serve to further magnify phosphorylation events that are key to growth-factor signaling.

To elucidate the ubiquity and functional consequences of redox signaling resulting from cellular ROS generation, it is critical to be able to visualize ROS production and analyze

subsequent oxidative PTMs of cysteine. ROS detection has been enabled by small-molecule and protein-based probes, which are coupled to spectroscopic readouts that provide sensitive ROS detection in living cells (*in situ*). Oxidative PTMs resulting from increased ROS levels can be interrogated by applying chemical probes coupled with proteomic technologies such as gel electrophoresis and mass spectrometry (MS). In this review, we summarize the available array of chemical probes for analyzing both ROS generation and downstream oxidative cysteine PTMs. Detailed mechanisms, advantages, and limitations of each method will be discussed where relevant.

### Chemical Probes for ROS

ROS and the related reactive nitrogen species (RNS) typically display high reactivity with biomolecules (Fig. 1A). ROS include O<sub>2</sub><sup>•-</sup>, H<sub>2</sub>O<sub>2</sub>, hydroxyl radical (•OH), hypochlorite (•OCl), and singlet oxygen (<sup>1</sup>O<sub>2</sub>). RNS include peroxynitrite (ONOO<sup>-</sup>), which is generated *via* the reaction of nitric oxide (•NO) with O<sub>2</sub><sup>•-</sup>. Due to the distinct reactivity characteristics of each ROS and RNS, these species can display unique effects on physiological processes. For example, O<sub>2</sub><sup>•-</sup> and H<sub>2</sub>O<sub>2</sub> are relatively mild reactants and are typically implicated in redox signaling that is mediated through oxidative PTMs of cysteine. In contrast, •OCl is a stronger oxidant that is produced by myeloperoxidase (MPO) in phagosomes, and it attacks invading microbes that are engulfed by neutrophils and macrophages. Due to the unique reactivity characteristics of each ROS, it is imperative to develop methods that are specific for each species. To this end, a number of chemical probes that are specific to distinct ROS have been developed and utilize electron paramagnetic resonance (EPR), chemiluminescence, fluorescence spectroscopy, and microscopy platforms to monitor ROS levels. In this review, we focus on chemical probes for O<sub>2</sub><sup>•-</sup> and

$\text{H}_2\text{O}_2$ , which are considered the most relevant to redox signaling (18, 102).

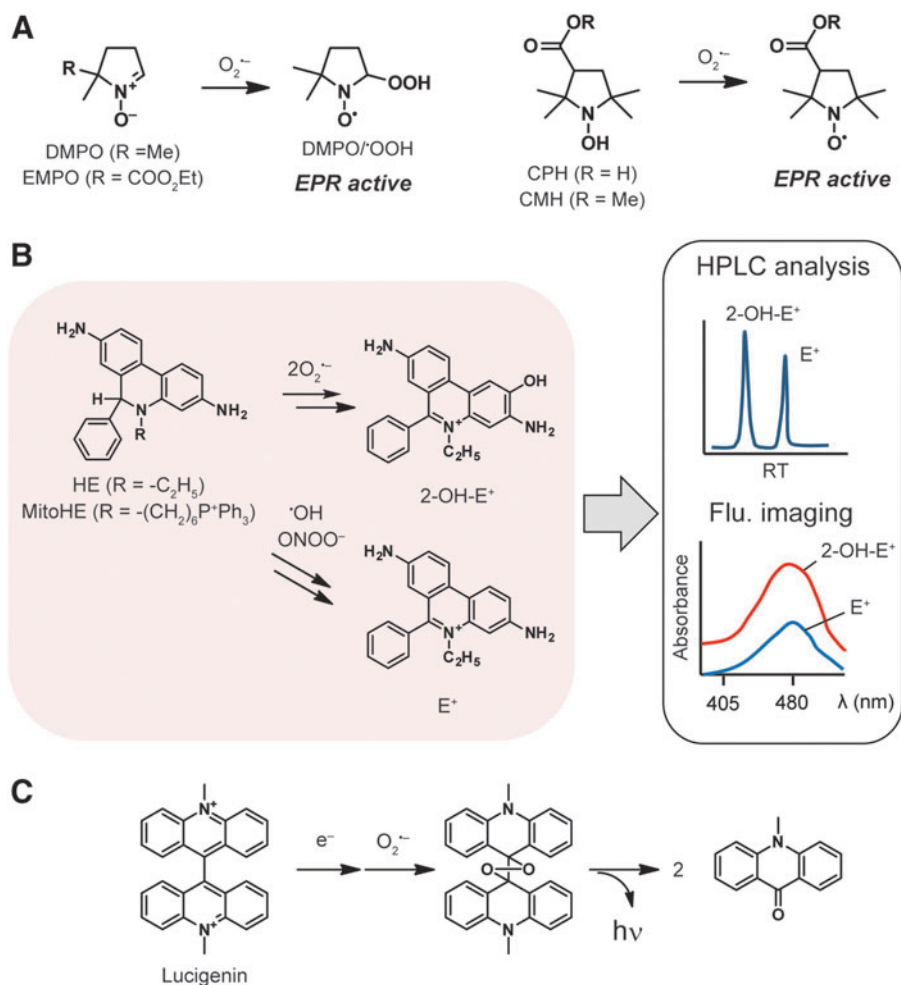
### Chemical Probes for $\text{O}_2^{\bullet-}$

#### EPR probes for $\text{O}_2^{\bullet-}$

EPR detects the absorbance of microwave energy by radicals with unpaired electrons in an applied magnetic field. Because ROS are very short lived and unable to be directly detected in biological samples, chemical probes that form stable adducts with ROS, termed spin traps, are used for biological ROS detection by EPR. A nitron-based probe, 5,5-dimethyl-1-pyrroline-N-oxide (DMPO), is the most widely used EPR probe for  $\text{O}_2^{\bullet-}$ . DMPO reacts with unstable free radicals, such as  $\text{O}_2^{\bullet-}$  and  $\bullet\text{OH}$ , to form relatively stable radical adducts, DMPO/ $\bullet\text{OOH}$  and DMPO/ $\bullet\text{OH}$ , respectively, which can be characterized by EPR (Fig. 2A). The DMPO probe has been applied to detect free radicals in a wide variety of cells, including endothelial cells (105, 144) and myocytes (128). One of the limitations of DMPO is the short half-life of DMPO/ $\bullet\text{OOH}$  ( $t_{1/2} \sim 45$  s), which lowers the detection sensitivity in EPR measurements. To overcome this limitation, a derivative of DMPO, 5-(diethoxyphosphoryl)-5-methyl-1-pyrroline-N-oxide (DEPMPO), has been developed, which generates a more stable adduct with  $\text{O}_2^{\bullet-}$  ( $t_{1/2} \sim 15$  min) (29). Another limitation of DMPO is the potential

decomposition of DMPO/ $\bullet\text{OOH}$  to DMPO/ $\bullet\text{OH}$  through reductive degradation by biological antioxidants. The susceptibility to reductive degradation hampers sensitive detection *in situ*, resulting in false negatives when  $\text{O}_2^{\bullet-}$  levels are low (9). The possibility of degradation also requires control experiments with superoxide dismutase (SOD) mimics to distinguish  $\text{O}_2^{\bullet-}$  from  $\bullet\text{OH}$  (119). Further, nitron-based probes are limited by the slow reaction rates displayed with  $\text{O}_2^{\bullet-}$  (e.g.,  $k=74 \text{ M}^{-1}\cdot\text{s}^{-1}$  for a nitron-based derivative, EMPO) (22).

Cyclic hydroxylamine-based probes, including 1-hydroxy-3-carboxy-2,2,5-tetramethyl-pyrrolidine hydrochloride (CPH), provide an alternative method for  $\text{O}_2^{\bullet-}$  detection with high reliability and sensitivity (59) (Fig. 2A). CPH reacts with  $\text{O}_2^{\bullet-}$  with a fast reaction rate ( $k=3.2\times 10^3 \text{ M}^{-1}\cdot\text{s}^{-1}$ ) to give a stable radical product ( $t_{1/2}=330$  min in smooth muscle cells) (23). CMH (1-hydroxy-3-methoxycarbonyl-2,2,5,5-tetramethylpyrrolidine), a plasma membrane-permeable derivative of CPH, has been applied to  $\text{O}_2^{\bullet-}$  detection in cultured cells and tissue samples (24, 59). Cyclic hydroxylamine-based probes have overcome drawbacks of nitron-based probes, such as slow reaction with  $\text{O}_2^{\bullet-}$  and reductive degradation in living organisms. However, the reaction rates of hydroxylamine-based probes are still slower than that of spontaneous dismutation of  $\text{O}_2^{\bullet-}$  under physiological conditions ( $k=10^5\text{--}10^6 \text{ M}^{-1}\cdot\text{s}^{-1}$ ) (113). In addition, these probes can be oxidized by



**FIG. 2. Chemical probes for superoxide.** (A) EPR-based spin traps form stable radicals with superoxide. (B) HE and MitoHE react with superoxide to form fluorescent products. (C) Lucigenin releases a photon on reaction with superoxide. EPR, electron paramagnetic resonance; HE, hydroethidine. Color images are available online.

multiple ROS and auto-oxidation as well, which necessitates control experiments with specific ROS-scavengers, such as SOD and catalase.

Triarylmethyl (TAM) radical-based probes have also been reported; the TAM radicals are EPR active and lose the EPR signal through the reaction with  $O_2^{\bullet-}$  at fast reaction rates ( $k = 3.1 \times 10^5 M^{-1} \cdot s^{-1}$  for a TAM derivative, OXO63) (103). The TAM radical-based probe, CT-02H, showed selectivity to  $O_2^{\bullet-}$  and  $ROO^{\bullet}$  among ROS and was applied to the detection of cellular  $O_2^{\bullet-}$  release on menadione stimulation (70). A major limitation of TAM radical-based probes is the signal-decreasing (turn-off) response to  $O_2^{\bullet-}$ , which lowers sensitivity of detection due to high background signals and hampers the wide application of TAM probes in living cells.

#### Fluorescent and chemiluminescent probes for $O_2^{\bullet-}$

Fluorescence imaging is a facile and sensitive methodology to detect ROS generation by coupling ROS-reactive chemical probes with fluorescence spectroscopy. Hydroethidine (HE) is a widely used fluorescence probe for  $O_2^{\bullet-}$  both *in vitro* and *in situ* (Fig. 2B). HE is oxidized by  $O_2^{\bullet-}$ , to form the highly fluorescent 2-hydroxyethidium (2-OH- $E^+$ ) with a high reaction rate ( $k = 2.6 \times 10^6 M^{-1} \cdot s^{-1}$ ) (138, 139). However, other ROS such as  $ONOO^-$ ,  $OH^{\bullet}$ , and  $H_2O_2$  oxidize HE to give a fluorescent product, ethidium ( $E^+$ ). High-performance liquid chromatography can be used to separate 2-OH- $E^+$  from  $E^+$  and provide a quantitative and specific measure of  $O_2^{\bullet-}$  in the system (28).  $E^+$  and 2-OH- $E^+$  can also be distinguished by fluorescence spectroscopy by excitation at 405 nm instead of 480 nm, which significantly reduces the fluorescence signal from  $E^+$  (85). However, control experiments utilizing SOD are still required to confirm specific detection of  $O_2^{\bullet-}$ . A mitochondria-targeted HE derivative, Mito-HE (also known as MitoSOX) contains a positively charged triphenylphosphonium (TPP) group, which allows Mito-HE to accumulate in mitochondria (142). Mito-HE has been applied to the detection of mitochondrial  $O_2^{\bullet-}$  production in a wide range of cells, including aortic endothelial cells (86) and cultured cerebellar granule neurons (48).

Hydrocyanines are reported as probes for several ROS, including  $O_2^{\bullet-}$  (57). The reduced cyanines are readily prepared by reduction of cyanine dyes with sodium borohydride, resulting in a drastic change in cyanine absorption and fluorescence. One derivative, hydro-carbocyanine 3 (Cy3), displays a fluorescence increase on treatment with  $O_2^{\bullet-}$ ,  $OH^{\bullet}$ , and t-butyl peroxy radical (TBO $^{\bullet}$ ). One advantage of hydrocyanines over HE is the increased stability in aqueous solutions; the half-lives of hydro-Cy3 and hydro-carbocyanine 7 (Cy7) in neutral aqueous solutions are  $\sim 3$  days, whereas that of HE is 30 min. Deuterated hydrocyanines have been reported to be more stable against auto-oxidation, which reduces background signals and allows for more sensitive detection of ROS (56). Hydro-Cy3 has been applied to live-cell imaging of rat aortic smooth muscle cells stimulated by angiotensin II. Another advantage of hydrocyanines is near-infrared absorption and fluorescence, which enables applications of hydrocyanines for *in vivo* imaging of ROS production in mice (35, 57).

Sulfonate-based protecting groups are utilized as a design strategy for  $O_2^{\bullet-}$  probes. Screening a library of benzene-sulfonyle fluorescein (BES) derivatives identified BES-So and

BES- $H_2O_2$  as fluorescent probes for  $O_2^{\bullet-}$  and  $H_2O_2$ , respectively (74, 75). In addition, HKSOX-1, a fluorescein protected by triflate groups, is reported to be responsive to  $O_2^{\bullet-}$  and applied to *in vivo*  $O_2^{\bullet-}$  imaging in zebrafish (44).

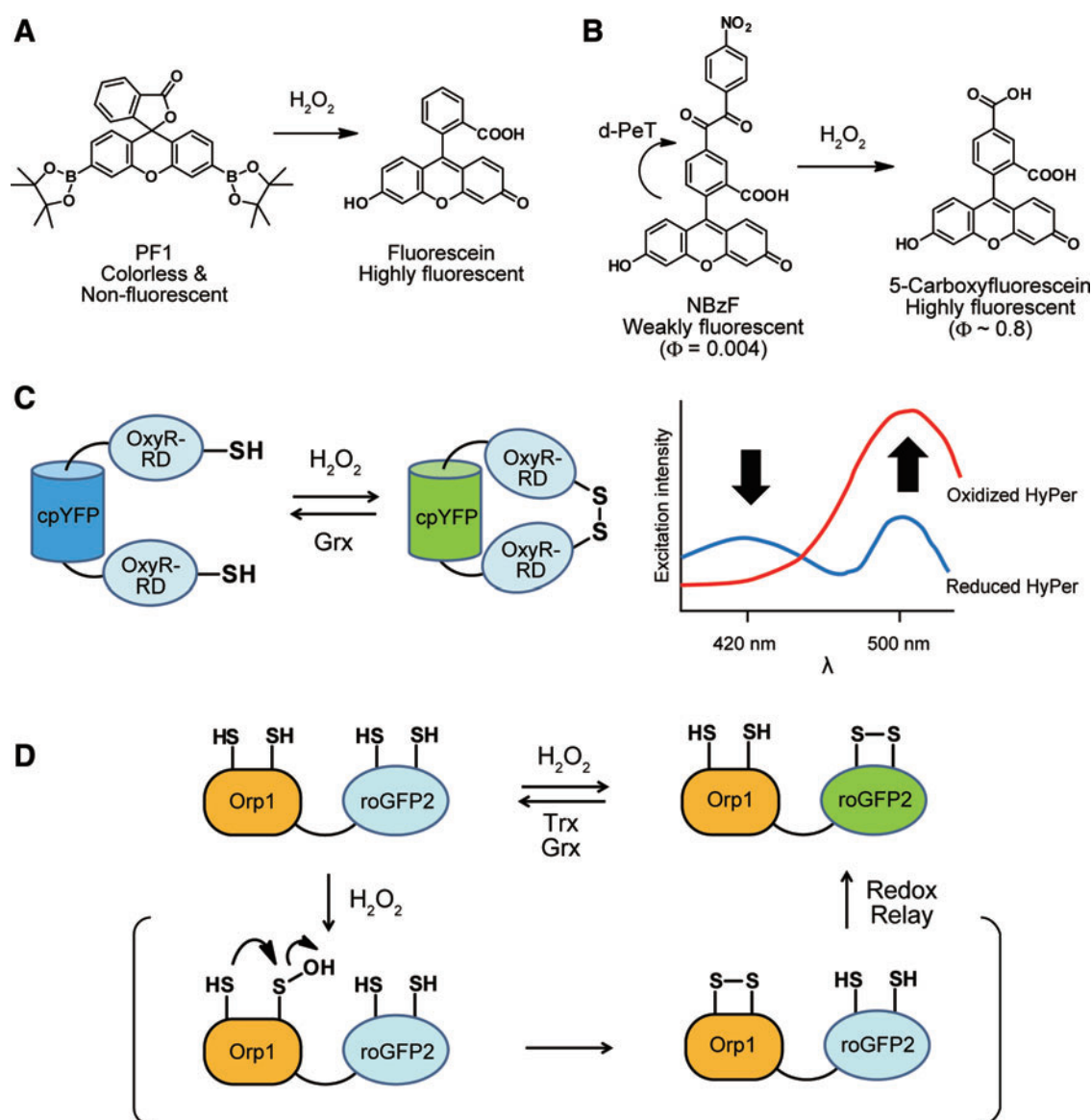
Chemiluminescent probes are widely used for  $O_2^{\bullet-}$  detection. The most common chemiluminescent probe for  $O_2^{\bullet-}$  is lucigenin (Fig. 2C). In the initial step, lucigenin is reduced by  $O_2^{\bullet-}$  to a cation radical, which then reacts with a second  $O_2^{\bullet-}$  molecule to form a dioxetane structure. A photon is emitted when the high-energy dioxetane structure is cleaved. Although lucigenin-based chemiluminescence is facile to monitor, lucigenin itself can produce  $O_2^{\bullet-}$  through a process known as redox cycling. This redox cycling phenomenon leads to the overestimation of  $O_2^{\bullet-}$  levels in the sample (120), and it is important to minimize artifactual production of  $O_2^{\bullet-}$  from lucigenin by choosing appropriate experimental conditions (67). Another class of chemiluminescent probes for  $O_2^{\bullet-}$  are derived from coelenterazine, which is a luciferin that is responsible for the bioluminescence of marine organisms. Coelenterazine has been reported as a chemiluminescent probe for  $O_2^{\bullet-}$  with enhanced light emission relative to lucigenin (52). One advantage of coelenterazine is that it does not produce artificial  $O_2^{\bullet-}$  through redox cycling (120). Derivatization of coelenterazine has resulted in the improved probes, 2-methyl-6-phenylimidazo[1,2-a]pyrazin-3(7H)-one (CLA) (116) and 6-(4-methoxyphenyl)-2-methyl-imidazo[1,2-a]pyrazin-3(7H)-one (MCLA) (89); the optimized derivative MCLA has a significantly enhanced emission (>100-fold) over coelenterazine. CLA and MCLA have been applied to  $O_2^{\bullet-}$  detection in a variety of biological samples such as living cells (53, 54) and tissues (58, 130). It is reported that MCLA can emit light by the reaction with ROS other than  $O_2^{\bullet-}$  (49, 90); thus, control experiments with SOD and other specific quenchers of ROS are required.

#### Chemical Probes for Hydrogen Peroxide

##### Small-molecule-based probes for $H_2O_2$

The most widely used chemical probes for  $H_2O_2$  are reduced dyes such as dihydrodichlorofluorescein (DCFH $_2$ ) and dihydrorhodamine123. Oxidation of these reduced dyes by ROS generates a fluorescence increase. However, these reduced dyes have several critical drawbacks when considered as fluorescence probes for  $H_2O_2$ . First, DCFH $_2$  has poor reactivity and specificity to  $H_2O_2$ ; DCFH $_2$  is even more responsive to highly ROS, including  $OH^{\bullet}$ ,  $ONOO^-$ , and  $^-OCl$ , than  $H_2O_2$  (112). Thus, DCFH $_2$  should be regarded as an indicator of gross ROS amount rather than an  $H_2O_2$ -specific probe. In addition, DCFH $_2$  can be oxidized by light irradiation, which requires intensive protection from light throughout the experimental workflow. In addition, the irradiation during fluorescence readings can also trigger increased DCFH $_2$  fluorescence, rendering the interpretation of imaging results complicated and unreliable (5, 112).

In attempts to overcome these drawbacks, chemical reactions that are specific for  $H_2O_2$  have been sought as the underlying basis for fluorescence-based  $H_2O_2$  probes. One example is the deprotection of boronate esters through oxidation by  $H_2O_2$ . Peroxyfluor-1 (PF1) and peroxyresorufin-1 (PR1) contain boronate ester groups, which can be converted to phenols by  $H_2O_2$  to produce the highly fluorescent molecules, fluorescein ( $\lambda_{abs} = 488$  nm,  $\lambda_{em} = 520$  nm) and resorufin



**FIG. 3. Fluorescent probes for  $\text{H}_2\text{O}_2$ .** (A) Boronate-protected fluorophores are deprotected by  $\text{H}_2\text{O}_2$ . (B) The benzil group is cleaved by  $\text{H}_2\text{O}_2$  to form highly fluorescent 5-carboxyfluorescein. (C) HyPer enables ratiometric fluorescent measurements of  $\text{H}_2\text{O}_2$  *in situ*. (D) Orp1 forms a disulfide bond on reaction with  $\text{H}_2\text{O}_2$ , which is relayed to the fused roGFP2, leading to a fluorescence change.  $\text{H}_2\text{O}_2$ , hydrogen peroxide; roGFP2, redox-sensitive green fluorescent protein 2. Color images are available online.

( $\lambda_{\text{abs}} = 571 \text{ nm}$ ,  $\lambda_{\text{em}} = 585 \text{ nm}$ ), respectively (82) (Fig. 3A). PF1 and PR1 show a large fluorescence increment (>1000-fold) with reaction rate constants  $k = 0.54 \text{ M}^{-1} \cdot \text{s}^{-1}$ , and  $1.0 \text{ M}^{-1} \cdot \text{s}^{-1}$ , respectively. The response of PF1 is reported to be selective for  $\text{H}_2\text{O}_2$ , with relatively small fluorescence increments observed for  $\cdot\text{NO}$  treatment *in vitro*. Peroxy Green 1 (PG1), a derivative based on TokyoGreen (124), has also been applied for fluorescence imaging of endogenous  $\text{H}_2\text{O}_2$  produced by A431 cells and cultured rat hippocampal neurons under epidermal growth factor (EGF) stimulation (83). Similarly, a wide variety of fluorophores have been protected with boronates; for example, rhodol-based derivatives, Peroxy Yellow 1 (PY1) and Peroxy Orange 1 (PO1), were developed for multi-color imaging, generating products that emit yellow ( $\lambda_{\text{abs}} = 519 \text{ nm}$ ,  $\lambda_{\text{em}} = 548 \text{ nm}$ ) and orange fluo-

rescence ( $\lambda_{\text{abs}} = 540 \text{ nm}$ ,  $\lambda_{\text{em}} = 565 \text{ nm}$ ), respectively (21). PY1 and PO1 have been applied to live-cell imaging of endogenous  $\text{H}_2\text{O}_2$  in multiple types of cells, including RAW264.7 macrophages stimulated by phorbol 12-myristate 13-acetate (PMA), and EGF-stimulated A431 cells. A major limitation of boronate-based probes is the slow reaction rate with  $\text{H}_2\text{O}_2$  ( $k = 0.1\text{--}1 \text{ M}^{-1} \cdot \text{s}^{-1}$ ). This slow reaction rate results in low sensitivity under physiological conditions where  $\text{H}_2\text{O}_2$  is rapidly removed by antioxidant enzymes such as Prxs (132). Another limitation is that boronate ester-based probes have a fluorescence response to  $\text{ONOO}^-$  as well. Multiple reports show the application of boronate-based probes to the detection of  $\text{ONOO}^-$  (114, 137, 143); thus, control experiments with  $\text{ONOO}^-$  scavengers are required to differentiate  $\text{H}_2\text{O}_2$  and  $\text{ONOO}^-$  when using boronate-based probes.



Benzil chemistry can also be used as a design strategy for H<sub>2</sub>O<sub>2</sub> probes. Benzil ( $\alpha$ -dibenzoyl) has been reported to generate benzoic acid by H<sub>2</sub>O<sub>2</sub> in alkaline conditions through an acyclic (Baeyer-Villiger) mechanism (109). To exploit this reaction for H<sub>2</sub>O<sub>2</sub> sensing, benzil groups were appended to fluorescein to generate 5-benzoylcarbonylfluorescein derivatives (3) (Fig. 3B). The benzil group acts as an H<sub>2</sub>O<sub>2</sub>-reactive group while also quenching fluorescence through a donor-excited photoinduced electron transfer (d-PeT) mechanism (123, 124). In the d-PeT process, an electron from the excited fluorophore transfers to the lowest unoccupied molecular orbital of the benzil group, which results in non-radiative relaxation of the excitation energy (123). A derivative with a nitro group on the benzil moiety, 5-(4-nitrobenzoyl)carbonylfluorescein (NBzF), displays a 150-fold fluorescence increase ( $\lambda_{\text{abs}} = 495 \text{ nm}$ ,  $\lambda_{\text{em}} = 519 \text{ nm}$ ) in the presence of H<sub>2</sub>O<sub>2</sub> with a reaction rate constant  $k = 3.2 \text{ M}^{-1} \cdot \text{s}^{-1}$ . In addition to H<sub>2</sub>O<sub>2</sub>, NBzF shows a relatively small fluorescence increment in the presence of *t*-butyl hydroperoxide (*t*-BuOOH) and ONOO<sup>-</sup>, but it does not respond to •NO *in vitro*. NBzF was applied for live-cell imaging of endogenous H<sub>2</sub>O<sub>2</sub> in PMA-stimulated RAW264.7 cells and EGF-stimulated A431 cells. However, NBzF has limitations that are similar to boronate-based probes, including the relatively slow reaction rate with H<sub>2</sub>O<sub>2</sub> and the necessity of control experiments with ROS scavengers.

The enzyme-mediated reaction of a small molecule with H<sub>2</sub>O<sub>2</sub> can provide a robust and sensitive strategy for H<sub>2</sub>O<sub>2</sub> detection. For example, Amplex Red is oxidized by H<sub>2</sub>O<sub>2</sub> in the presence of horseradish peroxidase (HRP) to produce the fluorescent product, resorufin (140). This Amplex Red assay is a reliable method to detect extracellular H<sub>2</sub>O<sub>2</sub> with a limit of detection of  $\sim 5 \text{ pmol}$ . However, HRP is inactivated in the highly reducing cytosolic environment, likely due to the reduction of functionally essential disulfide bonds on HRP (43), rendering Amplex Red not applicable to cytosolic H<sub>2</sub>O<sub>2</sub> detection.

Due to the distinct physiological roles of H<sub>2</sub>O<sub>2</sub> in different subcellular locations, it is important to visualize H<sub>2</sub>O<sub>2</sub> production in a particular organelle. To this end, organelle-targeted probes for H<sub>2</sub>O<sub>2</sub> have been developed. One strategy to target small-molecule-based probes to a particular organelle is to utilize organelle-directing groups. Mitochondria peroxy yellow 1 (MitoPY1) is a boronate-based probe conjugated with a TPP group, which is well known to localize in mitochondria due to its positive charge (20). MitoPY1 successfully detected exogenously added H<sub>2</sub>O<sub>2</sub> in the mitochondria. Another strategy for directing probes to distinct organelles is through the use of protein tags such as the SNAP-tag. The SNAP-tag is an engineered human DNA repair protein, human O<sup>6</sup>-alkylguanine-DNA alkyltransferase (hAGT), which covalently binds to small molecules with an O<sup>6</sup>-benzylguanine group (51). Boronate-based probes are conjugated with SNAP-tag substrates (SNAP PG1 and PG2) (115) and directed to SNAP-tagged proteins with distinct subcellular localizations. This approach has been used to target boronate probes to the cytosol, nucleus, the extracellular face of the plasma membrane, as well as the endoplasmic reticulum and mitochondrial inner membrane, for exogenous H<sub>2</sub>O<sub>2</sub> detection in each specific subcellular location. Similarly, organelle-specific imaging of endogenous H<sub>2</sub>O<sub>2</sub> has been achieved. An organelle-targeted benzil-based probe, NBzF-BG, which has an O<sup>6</sup>-benzylguanine group

conjugated with NBzF, has been applied to detect endogenous H<sub>2</sub>O<sub>2</sub> produced during phagocytosis (2). NBzF-BG has been presented on the extracellular side of the plasma membrane of RAW264.7 macrophages, localizing NBzF-BG on the inside of the phagosomal membrane on phagocytosis of opsonized microbeads. NBzF-BG localized inside phagosomes has enabled the visualization of H<sub>2</sub>O<sub>2</sub> production in phagosomes.

#### Genetically encoded fluorescence probes for H<sub>2</sub>O<sub>2</sub>

Genetically encoded fluorescent protein-based H<sub>2</sub>O<sub>2</sub> probes, such as HyPer and redox-sensitive green fluorescent protein 2 (roGFP2)-Orp1, are powerful imaging tools with several advantages such as high sensitivity, ratiometric measurements, and ability to localize to a particular organelle. HyPer consists of a circularly permuted yellow fluorescent protein (cpYFP) inserted into the regulatory domain of OxyR, a prokaryotic H<sub>2</sub>O<sub>2</sub> sensor (11) (Fig. 3C). H<sub>2</sub>O<sub>2</sub> oxidizes Cys199 of the OxyR regulatory domain to form a sulfenic acid, followed by disulfide-bond formation with Cys208. The disulfide formation changes the conformation and excitation properties of HyPer, which enables ratiometric measurements with two excitation wavelengths (488/405 nm). The fluorescence response of HyPer is relatively sensitive to H<sub>2</sub>O<sub>2</sub> concentrations, and HyPer is not responsive to other ROS including O<sub>2</sub><sup>•-</sup>, •NO, and ONOO<sup>-</sup>, and oxidized glutathione (GSSG). Several variants of HyPer have been developed with larger signal enhancement (HyPer-2 and HyPer-3) (13, 77) and different excitation and emission wavelengths (HyPer-Red) (27). The fast reaction rates ( $k = 1.2\text{--}5.0 \times 10^5 \text{ M}^{-1} \cdot \text{s}^{-1}$ ) with H<sub>2</sub>O<sub>2</sub> and the ratiometric response confer HyPer probes that are suitable for various biological applications. HyPer has been applied for live-cell imaging in bacterial and mammalian cells, and it has also been targeted to organelles such as the cytoplasm, mitochondria, nucleus, endoplasmic reticulum, peroxisomes, and plasma membrane (12). The HyPer probes have also been applied to imaging of whole animals, including *Caenorhabditis elegans* (55) and zebrafish (88). A limitation of HyPer probes is that the fluorescence signal of HyPer can be affected by pH; thus, control experiments with pH sensors such as SypHer (79) are required to interpret HyPer-based imaging results.

Another strategy for genetically encoded H<sub>2</sub>O<sub>2</sub> probes is based on roGFPs (25, 41) coupled with a yeast peroxidase Orp1 (38). Orp1 is oxidized to form a sulfenic acid by H<sub>2</sub>O<sub>2</sub> at the catalytic Cys36, which rapidly forms an intramolecular disulfide bridge with the resolving Cys82. When fused to roGFP2, the disulfide bridge of Orp1 is efficiently relayed to roGFP2, which results in a ratiometric fluorescence signal change (390/480 nm) (Fig. 3D). The response of roGFP2-Orp1 is specific for H<sub>2</sub>O<sub>2</sub> among oxidants such as GSSG, cystine, and dehydroascorbic acid (38). One advantage of roGFP2-Orp1 over HyPer is insensitivity to pH changes within the physiological range. Applications of roGFP2-Orp1 include the visualization of physiological H<sub>2</sub>O<sub>2</sub> production in *Drosophila* larvae and tissues (6, 8, 122). Although roGFP2-Orp1 and HyPer have been successfully applied to H<sub>2</sub>O<sub>2</sub> imaging *in vivo*, reaction rates of their sensing moiety are still  $>100$ -fold slower than that of typical 2-Cys Prxs ( $k = \sim 10^7 \text{ M}^{-1} \cdot \text{s}^{-1}$ ), which are abundant in the cytosol. Recently, improved roGFP2-based H<sub>2</sub>O<sub>2</sub> probes, roGFP2-Tsa2 $\Delta$ C<sub>R</sub> and

roGFP2-Tsa2 $\Delta$ C<sub>p</sub> $\Delta$ C<sub>R</sub>, have been reported (84). These probes utilize a typical 2-Cys Prx Tsa2, with mutations at the peroxidatic cysteine (C<sub>p</sub>) and/or the resolving cysteine (C<sub>R</sub>), as an H<sub>2</sub>O<sub>2</sub>-sensitive moiety. These probes are ~50% oxidized under unstressed conditions, which allows for dynamic monitoring of both increases and decreases in basal H<sub>2</sub>O<sub>2</sub>. By using the roGFP2-Tsa2 $\Delta$ C<sub>R</sub> probe, it has been shown that cellular growth rate correlates with basal mitochondrial H<sub>2</sub>O<sub>2</sub> levels in yeast cells.

### Chemical Probes for Oxidative PTMs of Cysteine

Cysteine residues have unique chemical reactivity, including high nucleophilicity and redox sensitivity, and play important roles in many protein functions, such as nucleophilic and redox catalysis, structural-disulfide formation, and binding to metal ions (91). Oxidative PTMs of cysteine play central roles in redox signaling by regulating the activity of key proteins within signaling pathways. One well-characterized example of cysteine-mediated redox signaling is in EGF signaling. In A431 epidermoid carcinoma cells, PTP1B is inhibited (63) and EGFR is activated (93) through S-sulfenylation, resulting in augmentation of the phosphorylation signals resulting from EGF binding to EGFR. Therefore, methods for identifying and quantifying oxidative cysteine PTMs are critical to understanding redox signaling. Numerous chemical probes with a variety of reactive handles have been developed and applied to globally monitor oxidative cysteine PTMs.

Chemical probes for oxidative cysteine PTMs can be categorized into two groups: (i) cysteine-reactive probes to visualize general cysteine PTMs and (ii) probes that are specific for a particular oxidative PTM of cysteine. Probes in the first category are based on monitoring the intrinsic cysteine reactivity with electrophilic chemical probes that form stable covalent adducts with reduced cysteines. Oxidation of a particular cysteine residue is reflected by a loss in nucleophilicity and subsequent decrease in probe labeling. These methods allow for identification of any cysteine PTM that reduces nucleophilicity, and therefore can be applied to all oxidative cysteine modifications. However, secondary methods need to be employed to identify the exact modification that is causing the measured loss in cysteine reactivity. In the second category, PTM-specific reactions are harnessed to design chemical probes that are specific for each oxidative cysteine PTM, including S-sulfenylation, S-sulfinylation, and S-glutathionylation. Contrary to the general cysteine-reactive probes, PTM-specific probes can focus on a particular PTM, lower the complexity of proteome samples, and enhance sensitivity and reliability of analysis. However, probes to directly monitor inter- and intra-molecular disulfides, and sulfonic acids, are currently lacking, rendering these PTMs intractable to existing PTM-specific methods. To date, both general and specific methods rely on two-dimensional electrophoresis and MS-based analytical methods for identification and quantification of labeled proteins.

### Chemical-Proteomic Methods for General Oxidative PTMs of Cysteine

#### *Two-dimensional gel electrophoresis*

Two-dimensional gel electrophoresis (2DE)-based methods have been commonly used in proteomic research. 2DE

methods are also frequently applied to the analysis of oxidative cysteine PTMs for visualization with radioactive or fluorescent reporters (17). In typical procedures for autoradiography, non-oxidized cysteine residues are capped by iodoacetamide (IAM) or *N*-ethylmaleimide (NEM). Then, oxidized cysteines are reduced by reducing agents such as dithiothreitol (DTT) or tris(2-carboxyethyl)phosphine (TCEP). Subsequently, the newly reduced cysteine residues are labeled with radioactive <sup>14</sup>C-IAM or <sup>14</sup>C-NEM for visualization by autoradiography (61, 65). Due to non-specific reduction by DTT or TCEP (34), these methods visualize multiple forms of oxidative cysteine PTMs, including disulfide bonds, sulfenic acids, and nitrosothiols. To achieve more facile visualization of 2D gels, radioactive probes have been replaced by fluorescent cysteine-reactive probes such as 5-iodoacetamidofluorescein (10), monobromobimane (136), and boron dipyrromethene (BODIPY) FL C1-IA (42). The signal intensity of radiation or fluorescence from proteins can be normalized by the protein abundance estimated from Coomassie Brilliant Blue (CBB) staining to provide a qualitative estimate of the stoichiometry of oxidation (42, 65).

2DE-based methods can be applied to directly compare the degree of cysteine oxidation between two samples. In a method termed redox differential in-gel electrophoresis (redox-DIGE), a pair of cysteine-reactive probes conjugated with orthogonal fluorophores such as Cy3-maleimide and carbocyanine 5 (Cy5)-maleimide are used for labeling two samples, respectively (16, 30, 46). After blocking reduced cysteines, oxidized cysteines in two distinct cell lysates are labeled with two different fluorophores. The two labeled samples are combined and analyzed on the same 2D gel, enabling direct visualization of differences in oxidative cysteine PTMs across two samples. Redox-DIGE has been applied to identify mitochondrial proteins that are sensitive to oxidative modifications by endogenous ROS (46). However, this redox-DIGE analysis lacks information about protein abundance; thus, fluorescence signals cannot be normalized with the protein amounts. To overcome this limitation, another differential labeling strategy was developed, whereby both oxidized and reduced cysteine residues are differentially labeled with two distinct fluorophores (62). A ratio between the two fluorescence signals gives a quantitative measure of the stoichiometry of oxidation by normalizing for protein abundance.

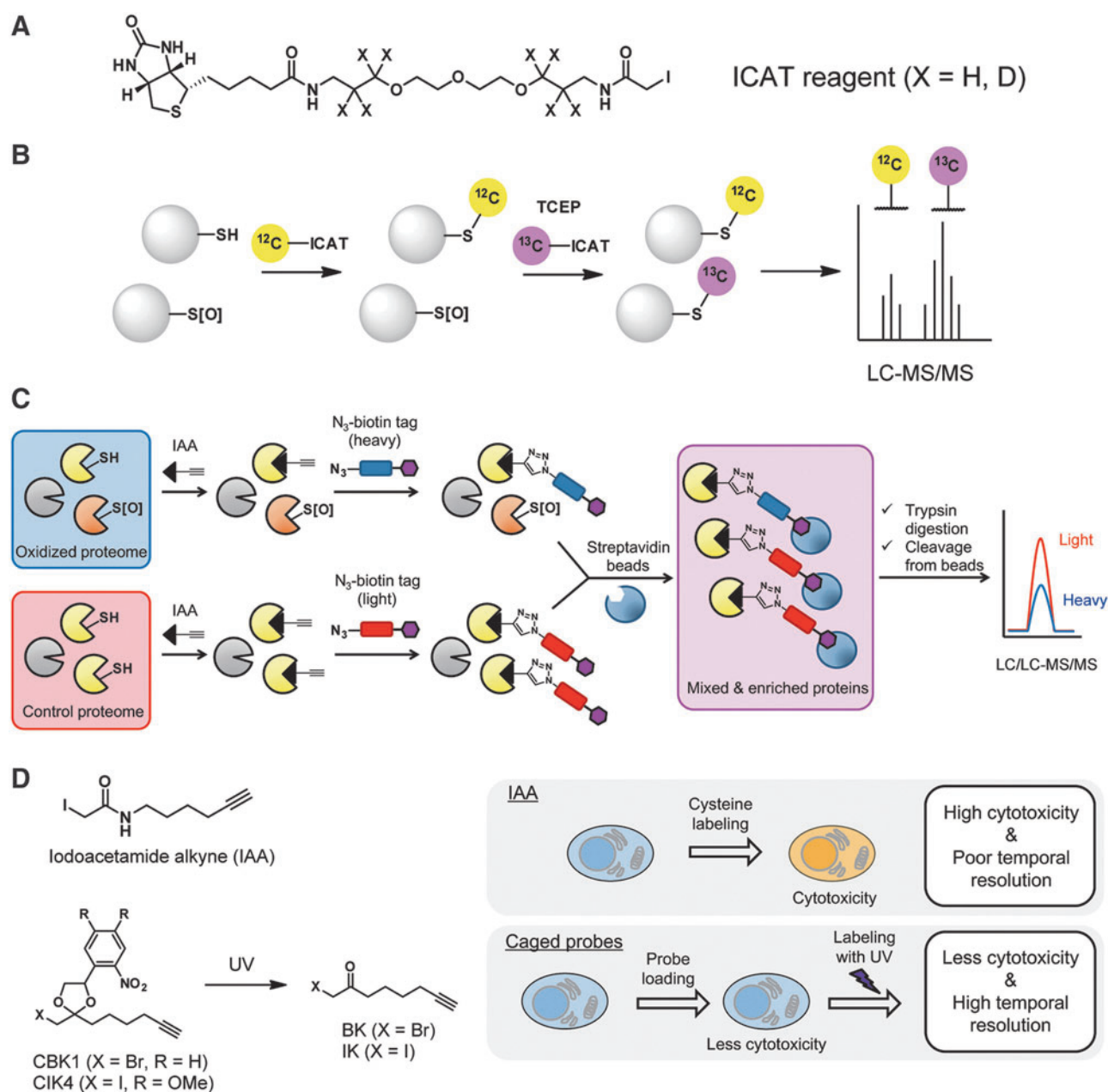
The 2DE-based methods described earlier commonly have major drawbacks. The limited resolution of 2DE results in insufficient separation of proteins with similar physicochemical properties. Further, limited detection sensitivity of gel-imaging methods prohibits the detection of low abundant proteins. In addition, even if protein spots are found to be oxidized by 2DE, the identification of each protein has to be performed individually by MS. Time-consuming 2DE procedures and subsequent individual identification of each protein by MS make 2DE methods low-throughput. Further, given that a protein has multiple cysteines, a signal of the protein on 2DE can represent an average of multiple cysteine residues, and identifying the exact cysteine(s) that are oxidized can be challenging.

#### *Isotope-coded affinity tags*

To overcome the multiple drawbacks of 2DE, a method based on isotope-coded affinity tags (ICAT) has been

developed. The ICAT method utilizes biotinylated IAM derivatives that are labeled with isotopically light and heavy linkers (39). Peptides tagged with the ICAT probes were enriched by streptavidin beads and analyzed by quantitative MS (Fig. 4A). Isotopically labeled ICAT probes enable relative quantification of oxidized cysteines between the light- and heavy-labeled samples. The ICAT strategy was applied to identify cysteines that are susceptible to oxidation on  $H_2O_2$  treatment (111). A modified ICAT strategy termed OxICAT (oxidative isotope-coded affinity tag) has been developed for

differential labeling of oxidized and reduced cysteine in the same sample (64). In this OxICAT strategy, reduced cysteines are labeled with a light ICAT reagent, and oxidized cysteine residues are reduced with TCEP and labeled with a heavy ICAT reagent (Fig. 4B). This differential labeling enables quantification of the ratio between reduced and oxidized forms of each cysteine, and this ratio can then be compared across multiple samples. To better monitor protein abundance changes in tandem with oxidation variations, isotopic dimethyl labeling has been coupled with ICAT,



**FIG. 4. Chemical-proteomic platforms for general cysteine PTMs.** (A) Structures of ICAT reagents. (B) OxICAT is a modified ICAT technology for oxidative cysteine PTMs. (C) A quantitative proteomic platform, isoTOP-ABPP, enables comprehensive identification and quantification of cysteine PTMs. (D) Caged electrophilic probes are developed to achieve *in situ* labeling of cysteine and analysis of cysteine PTMs within a physiological context. CBK1, caged  $\alpha$ -bromomethylketone alkyne; IAA, iodoacetamide alkyne; ICAT, isotope-coded affinity tags; isoTOP-ABPP, isotopic tandem orthogonal proteolysis activity-based protein profiling; LC-MS/MS, liquid chromatography-tandem mass spectrometry; OxICAT, oxidative isotope-coded affinity tag; TCEP, tris(2-carboxyethyl)phosphine; UV, ultraviolet. Color images are available online.



enabling protein abundance measurements before cysteine enrichment (32, 33). Another strategy for protein quantification is a method termed GELSILOX in which a stable  $O_2$  isotope is incorporated during the trypsin digestion step (78). These ICAT, OxICAT, and GELSILOX methods overcome major limitations of 2DE by achieving both the identification and quantification of multiple oxidized proteins in a single experiment, and, in addition, provide higher sensitivity than gel-based methods for the detection of low abundance proteins. However, detection sensitivity can be decreased by low ionization efficiency of certain peptides, multiple cysteines on the same peptide can complicate site identifications, and a cysteine cannot be identified if the residue is on a very short or long tryptic peptide.

#### *Isotopic tandem orthogonal proteolysis activity-based protein profiling*

ICAT probes have a bulky biotin reporter that may inhibit labeling of sterically hindered cysteines. An alternative MS-based method, termed isotopic tandem orthogonal proteolysis activity-based protein profiling (isoTOP-ABPP), relies on a simpler iodoacetamide alkyne (IAA) probe (131). In isoTOP-ABPP, proteome samples are labeled with IAA, before incorporation of a biotin-azide tag *via* copper(I)-catalyzed alkyne-azide cycloaddition (CuAAC) (Fig. 4C). The biotin moiety facilitates subsequent enrichment of labeled proteins by streptavidin beads, after which proteins then undergo an on-bead tryptic digestion. The biotin-azide tag has cleavable linkers that are susceptible to cleavage by treatment with tobacco etch virus protease or sodium dithionite (98, 131), which releases the enriched peptides for MS analysis. By using isotopically light- or heavy-labeled biotin-azide tags, relative quantification across two samples is achieved. An isoTOP-ABPP ratio, *R*, is calculated from signal intensities of light- and heavy-tagged peptides for each identified cysteine. The *R* value provides a quantitative measure of changes in cysteine reactivity, and the isoTOP-ABPP method can identify and quantify cysteine reactivity changes of ~1000 to 1500 cysteine residues. The isoTOP-ABPP method and its derivatives have been applied for identification of bacterial proteins that are susceptible to oxidation by  $H_2O_2$  (19). This method was also applied to identify *S*-nitrosation sites on numerous proteins on treatment with transnitrosation donors such as *S*-nitrosoglutathione (GSNO), identifying *S*-nitrosation sites on proteins including 3-hydroxyacyl-CoA dehydrogenase type 2 (HADH2) and cathepsin D (CTSD) (141).

Although isoTOP-ABPP can quantify and identify more than 1000 cysteines in a single tandem liquid chromatography-tandem mass spectrometry (LC/LC-MS/MS) run, coverage of the whole proteome by this method can still be limited due to the high complexity and heterogeneity of the proteome samples. Organelle isolation strategies can be coupled with isoTOP-ABPP to improve coverage of reactive cysteines in a particular organelle such as mitochondria (7). Another limitation of isoTOP-ABPP has been the usage of cytotoxic IAA probe, which is not suitable for live-cell applications. Thus, applications of isoTOP-ABPP have been primarily performed in lysate samples. However, because oxidative PTMs of cysteine are highly labile to environmental redox changes caused by cell lysis, it is ideal to perform the initial labeling step directly in living cells. Photolabile caged electrophilic probes

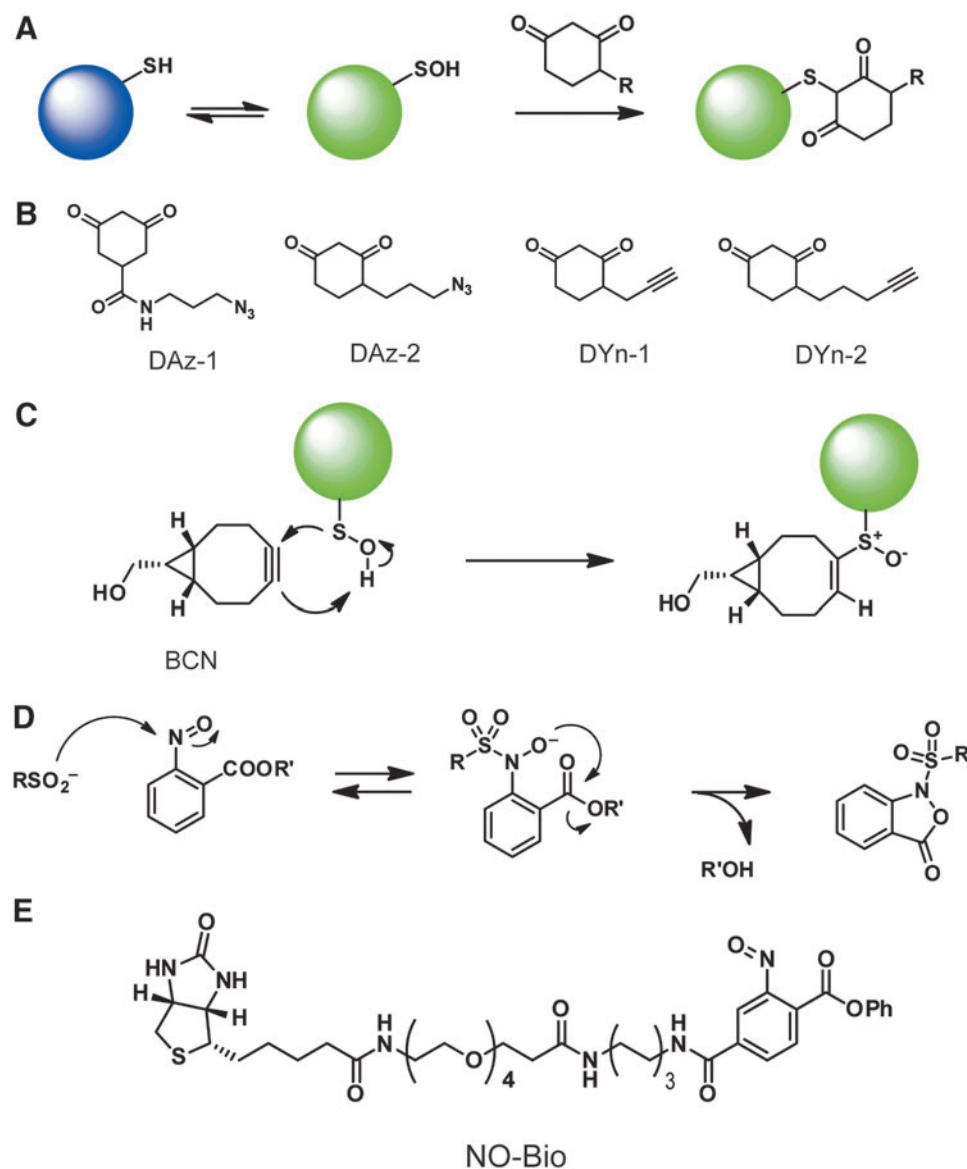
have been developed to achieve cysteine labeling *in situ* (1, 4). A caged electrophile named caged  $\alpha$ -bromomethylketone alkyne, or CBK1, is not cysteine reactive and, therefore, can be loaded into live cells with negligible cytotoxicity (Fig. 4D). CBK1 is activated to be cysteine reactive by ultraviolet (UV) light irradiation *in situ* to trigger cysteine labeling, and the labeled cysteine residues are identified by LC/LC-MS/MS. An additional advantage of this labeling strategy with caged probes is the high temporal resolution of cysteine labeling, which is required for accurate analysis of transient and reversible cysteine oxidations in redox signaling. The practical usability of CBK1 has been demonstrated by analyzing redox signaling in A431 epidermoid carcinoma cells stimulated by EGF. CBK1 coupled with isoTOP-ABPP identified hundreds of cysteine residues with *R* values, indicating the degree of oxidative modification for each cysteine under EGF stimulation. More recently, an optimized caged probe, caged  $\alpha$ -iodoketone alkyne 4 (CIK4), has been reported with improved cysteine reactivity and uncaging efficiency, which enables more efficient detection of cysteine residues by LC/LC-MS/MS (1). Though caged electrophilic probes enable cysteine labeling *in situ* with short exposure time to cytotoxic agents and high temporal resolution, the limitation is the usage of UV light, which may cause cytotoxic effects on live cells.

#### **Chemical Proteomic Methods for a Specific Oxidative PTM of Cysteine**

##### *Chemical probes for sulfenic acids*

The 2DE, ICAT, and isoTOP-ABPP methods discussed earlier all rely on changes to the intrinsic reactivity of a cysteine induced by oxidation, and, therefore, do not differentiate cysteine PTMs such as sulfenic acids, sulfinic acids, and disulfides. Complementary strategies focus on a specific cysteine PTM by utilizing chemical probes that selectively react with a single PTM. The reversibility of sulfenic acids renders this PTM the ability to mediate cellular signal transduction events that are akin to phosphorylation and acetylation. One strategy for the specific labeling of sulfenylation is based on the selective reduction of sulfenic acids by arsenite, which is analogous to the differential tagging strategies described earlier (108). In this method, sulfenic acids are reduced by arsenite under denaturing conditions, then labeled with a biotin-maleimide probe. Another strategy is the direct labeling with chemical probes that are specific for sulfenic acids, including 5, 5-dimethyl-1, 3-cyclohexanedione, also known as dimedone, as a reactive moiety (Fig. 5A). The enolate form of dimedone facilitates nucleophilic attack onto sulfenic acids to form a covalent adduct. Dimedone or other 1,3-diketone compounds are conjugated with fluorophores or biotin analogues for in-gel visualization and enrichment with avidin for MS-based proteomic experiments (37, 87, 95, 96).

The presence of bulky reporter groups such as fluorophores and biotin analogues are not ideal because they result in low labeling efficiency by blocking access to buried sulfenic acids, as well as limiting the plasma membrane permeability of the probes. Sulfenic acids are highly labile to redox changes, and, therefore, labeling *in situ* is important for accurate analysis. To this end, dimedone derivatives with minimal modifications have been developed; azide-tagged probes (azide-tagged dimedone [DAz]-1 and DAz-2) (66, 100) and alkyne-tagged probes (alkyne-tagged dimedone [DYn]-1 and



**FIG. 5. Chemical probes for sulfenic acids and sulfenic acids.** (A) The reaction between sulfenic acids and dimedone. (B) Structures of dimedone-based probes. (C) A strained alkyne-based probe, BCN, can be conjugated to sulfenic acids. (D) The reaction between sulfenic acids and 2-nitroso benzoic acid derivatives. (E) NO-Bio is a nitroso-based chemical probe for sulfenic acids. BCN, bicyclo[6.1.0]nonyne; DAz, azide-tagged dimedone; DYn, alkyne-tagged dimedone; NO-Bio, biotinylated 2-nitroso terephthalic acid. Color images are available online.

DYn-2) (93) have chemical handles for incorporation of reporters *via* a Staudinger reaction or CuAAC in subsequent steps (Fig. 5B). These probes are membrane permeable and can be applied to the labeling of sulfenic acids directly in living cells.

A strategy for differential labeling of sulfenic acids has been developed by utilizing isotopically coded dimedone- and iododimedone-bearing probes, which are reactive with sulfenic acids and reduced cysteines, respectively (110). Based on this strategy, the degree of sulfenylation at a particular cysteine residue can be accurately quantified. In a similar manner, dimedone and IAM conjugated with lantamide-1,4,7,10-tetraazacyclododecane-1,4,7,10-tetraacetic acid (DOTA) have been applied to quantify sulfenylation using inductively coupled plasma-MS (26).

Dimedone-based probes have been applied to various biological systems to uncover sulfenylation sites. For example, a biotinylated dimedone probe has been utilized to reveal that Cys124 of protein kinase B  $\beta$  (Akt2) is sulfenylated under physiologically relevant platelet-derived growth factor (PDGF) stimulation (129). The sulfenylation of Cys124 inhibits the ac-

tivity of Akt2, showing a novel regulatory mechanism for kinase regulation in addition to phosphorylation. As another example, a biotinylated dimedone probe has been applied to T cells to identify increased sulfenylation on protein tyrosine phosphatases such as Src homology domain 2-containing protein tyrosine phosphatase (SHP)-1 and SHP-2, and actin on T cell activation (81). This study demonstrated that reversible sulfenylation is an important regulatory factor for proliferation and function of cluster of differentiation 8 ( $\text{CD}8^+$ ) T cells. An alkyne-functionalized dimedone probe, DYn-2, revealed that Cys797 of EGFR is sulfenylated under EGF stimulation and sulfenylation augments the kinase activity of EGFR (93). Lastly, isotopically light- and heavy-labeled DYn-2 probes were coupled to advanced LC-MS/MS-based methods to globally identify and quantify the exact sites of sulfenylation within proteomes (134).

Although dimedone-based probes have been successfully applied to proteomic experiments, one limitation of dimedone is the demonstrated slow reaction rate with sulfenic acids, which requires the use of the chemical probes at high concentrations to achieve efficient labeling. To overcome this limitation, a library of  $\sim 100$  cyclic C-nucleophiles have been evaluated for

reactivity with sulfenic acids to find several novel compounds with more than 200-fold enhanced reactivity than that of dimedone (36). Strained bicycle[6.1.0]nonyne (BCN) derivatives were also reported as a new reactive group with sulfenic acids (Fig. 5C). The BCN derivatives can react with sulfenic acids with a faster reaction rate than dimedone by over two orders of magnitude (97). It is noteworthy that BCN is reported to undergo thiol-yne addition and can be non-selectively consumed by thiols (121). These new chemistries can offer an alternative strategy for monitoring sulfenylation in addition to the widely used dimedone-based probes.

#### Chemical probes for sulfenic acids

S-sulfinylation (RSO<sub>2</sub>H) was regarded an artifactual hyper-oxidation of cysteine that occurs during sample handling, but increasing evidence indicates that S-sulfinylation is not only physiologically relevant but also important in redox signaling (47, 71). It was revealed that ~5% of cysteine residues are oxidized to sulfenic acids in a soluble fraction of rat liver (40). Among S-sulfinylated proteins are Prxs, which are a family of H<sub>2</sub>O<sub>2</sub>-scavenging enzymes. Due to high abundance in the cytosol and very fast reaction rates with H<sub>2</sub>O<sub>2</sub> ( $k = \sim 10^7 \text{ M}^{-1} \cdot \text{s}^{-1}$ ), Prxs remove H<sub>2</sub>O<sub>2</sub> very rapidly to keep the concentration of cytosolic H<sub>2</sub>O<sub>2</sub> relatively low (132). Therefore, it has been proposed that Prxs activity may be locally regulated under redox signaling through PTMs such as S-sulfinylation (135) and phosphorylation (133). Although sulfenic acids are not reduced by most biological antioxidants, the S-sulfinylation on Prxs can be reduced through an ATP-dependent reduction by Srx (14).

To capture sulfenic acids in a specific manner, the reported reaction between 2-nitroso benzoic acid derivatives and sulfenic acids has been adapted (72) (Fig. 5D). The reactive nitroso warhead covalently modifies sulfenic acids to form stable nitroso-benzene derivatives. This ligation reaction is shown to be selective to sulfenic acids among various nucleophilic and/or redox-active small molecules. An observed side reaction involves the oxidation of cysteine to cystine concomitant with the reduction of nitroso compounds. A nitroso compound, 2-nitroso terephthalic acid (NO-Ph), was shown to successfully label a sulfenic acid on a recombinant Gpx3 mutant; however, NO-Ph also formed a stable sulfonamide adduct with a reduced cysteine on Gpx3. Thus, the blocking of reduced cysteine with NEM before the labeling of sulfenic acids is required for efficient MS analysis. The reaction termed sulfenic-acid-nitroso ligation (SNL) has been utilized to design a sulfenic acid-specific probe, biotinylated 2-nitroso terephthalic acid (NO-Bio), in which a 2-nitroso benzoic acid derivative is conjugated with a biotin analogue (73) (Fig. 5E). The NO-Bio probe has been applied to the labeling of sulfenic acids on recombinant Parkinson disease protein 7 (DJ-1) oxidized by H<sub>2</sub>O<sub>2</sub> and in cell lysates induced by 2,2'-dipyridyl disulfide (DPS), and to profile oxidation among normal and matched tumor cells. The SNL reaction can also be utilized to monitor S-nitrosation (76). Biotinylated probes with sulfenic acid (Biotin-SO<sub>2</sub>H) and GSNO (Biotin-GSNO) have been applied to quantitative MS using stable isotope labeling with amino acids in cell culture (SILAC), revealing relative occupancy ratios of S-sulfinylation and S-nitrosation in 293T cells. The SNL chemistry has not yet been applied to a wide variety of biological samples;

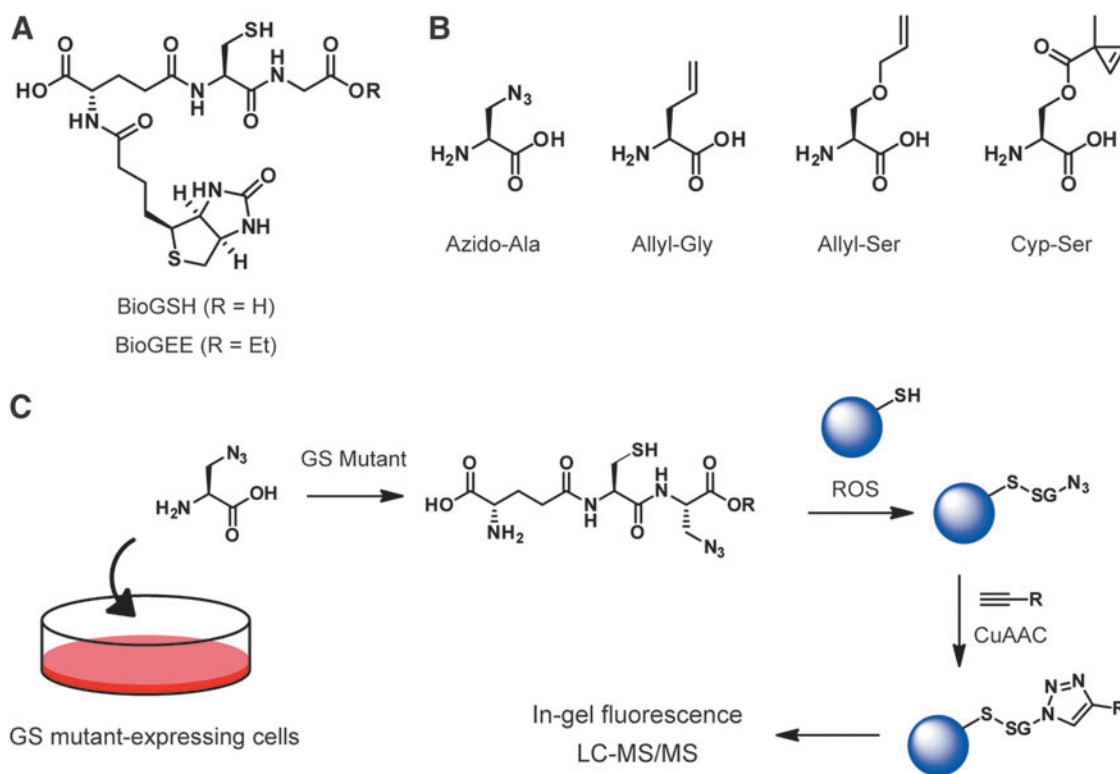
however, it opens the door to proteome-wide analyses focusing on S-sulfinylation and S-nitrosation.

#### Chemical probes for S-glutathionylation

S-glutathionylation is known to regulate important protein functions that are involved in various physiological processes such as gene expression, signaling, energy metabolism, and cell survival (92). For example, glutathionylation of nuclear factor- $\kappa$ B (NF- $\kappa$ B) inhibits DNA-binding activity and downstream transcriptional activation (94). Another key protein in this pathway, inhibitor of  $\kappa$ B kinase (IKK), is also glutathionylated, rendering the NF- $\kappa$ B pathway highly susceptible to redox changes in the cellular environment (101). The extent of modification with glutathione depends on the ratio of reduced (GSH) and oxidized forms (GSSG) of glutathione.

Various chemical-proteomic strategies have been developed to analyze S-glutathionylation within a biological context. In early methods based on autoradiography, GSH tagged with <sup>3</sup>H or <sup>35</sup>S was utilized for labeling *in vitro*, and <sup>35</sup>S-cysteine was applied for metabolic incorporation *in situ* (31). GSH is synthesized from glutamate, cysteine, and glycine through two subsequent ATP-dependent reactions catalyzed by  $\gamma$ -glutamylcysteine synthetase and glutathione synthetase (GS) (80). The method for metabolic incorporation of <sup>35</sup>S-cysteine requires treatment with cycloheximide to inhibit incorporation of <sup>35</sup>S-cysteine into proteins, ensuring that <sup>35</sup>S-cysteine is only incorporated into GSH. Proteome samples modified with <sup>35</sup>S-labeled GSH are analyzed by sodium dodecyl sulfate polyacrylamide gel electrophoresis and imaged by autoradiography. This method was applied to monitor S-glutathionylation in human monocytes (104), revealing that glyceraldehyde-3-phosphate dehydrogenase (GAPDH) is S-glutathionylated during a respiratory burst (99). However, this method has major drawbacks, including the necessity to inhibit protein synthesis, which greatly affects cellular metabolism, and the lack of a handle to enrich modified proteins for global MS identification.

To overcome these drawbacks of radioactive GSHs, biotinylated GSH (BioGSH) analogues have been developed. GSH and GSSG can be conjugated with a biotin group at the primary amine by treatment with sulfosuccinimidyl-6-(biotinamido)hexanoate (sulfo-NHS-biotin) (15). A membrane-permeable precursor of GSH, glutathione ethyl ester (GEE), can also be modified by sulfo-NHS-biotin to form biotinylated glutathione ethyl ester (BioGEE) (117) (Fig. 6A). The biotin groups can be utilized for enrichment of glutathionylated proteins by streptavidin beads, followed by reductive cleavage of the enriched proteins from the beads and analysis by gel electrophoresis and MS. Non-reducing Western blotting enables sensitive detection of glutathionylated proteins in combination with streptavidin-HRP or anti-biotin antibodies. These BioGSH derivatives have been applied to the identification of glutathionylation *in vitro* (45, 127) and in cells. BioGEE was applied to live HeLa cells under oxidative stress by exogenous H<sub>2</sub>O<sub>2</sub> or tumor necrosis factor- $\alpha$  (TNF- $\alpha$ ) treatment to identify annexin A2 and Trx peroxidase 2 as glutathionylated proteins (117). Limitations of this method include the modification with a bulky biotin group that may perturb a subset of proteins tagged with these reagents, and the exogenous addition of redox-active GSH derivatives that affect the native GSH/GSSG ratio.



**FIG. 6. Chemical probes for S-glutathionylation.** (A) Structures of biotinylated glutathione derivatives, BioGSH and BioGEE. (B) Glycine surrogates with bioorthogonal handles for metabolic incorporation. (C) The glycine surrogates can be incorporated into GSH catalyzed by mutated GSs in living cells, which enables proteomic analyses of native glutathionylation. BioGEE, biotinylated glutathione ethyl ester; BioGSH, biotinylated GSH; CuAAC, copper(I)-catalyzed alkyne-azide cycloaddition; GS, glutathione synthetase; GSH, reduced glutathione. Color images are available online.

An elegant method to overcome the limitations of BioGSH derivatives has been developed by utilizing a mutated GS that catalyzes coupling of an azide-tagged alanine derivative (azido-Ala) to  $\gamma$ -glutamylcysteine ( $\gamma$ Glu-Cys) (106) (Fig. 6B, C). The GSH derivative containing an azide handle (azide-GSH) for CuAAC allows for identification of glutathionylated proteins by gel and MS analysis. The strategy was utilized for analysis of glutathionylation in living cells under exogenous  $H_2O_2$  stimulation and glucose depletion (107). Alkene-bearing Gly surrogates (allyl-containing glycine [Allyl-Gly], allyl-containing serine [Allyl-Ser], and cyclopropane-containing serine [Cyp-Ser]) can also be enzymatically incorporated into GSH by mutated GSs, which enable bioorthogonal tagging by tetrazine-based reporters (50) (Fig. 6B).

Selective reduction of glutathione-containing disulfides mediated by Grx has been applied to proteomic analysis of glutathionylation (68). In the Grx-based method, on blocking non-oxidized cysteine residues with NEM, glutathione-containing disulfide bonds are reduced by Grx. Freshly formed cysteines are labeled with a biotin-tagged NEM probe, resulting in subsequent enrichment of labeled proteins. The Grx-based method has been applied to analyze glutathionylation under oxidative stress induced by diamide (69).

## Summary

Redox signaling is involved in various physiological processes such as antioxidant induction, cellular proliferation, and migration. Due to the highly reactive and unstable nature

of ROS, it has been challenging to measure physiological ROS generation in a reliable and quantitative manner. Taking advantage of the distinct reactivities of different ROS, chemical probes have been developed and applied to detect and quantify biological ROS production. To date, various detection methods for  $O_2^{\bullet -}$  are available and widely used with EPR, fluorescence, and chemiluminescence instrumentation. Fluorescence probes for  $H_2O_2$  have been designed by utilizing  $H_2O_2$ -specific chemical reactions and engineered  $H_2O_2$ -sensitive proteins. Despite the low reactivity of  $H_2O_2$ , fluorescent protein-based  $H_2O_2$  probes are highly sensitive to  $H_2O_2$ . Currently available ROS probes have several advantages and limitations in terms of sensitivity, selectivity, and usability; thus, it is important to choose the appropriate methods for a given application. Control experiments with ROS scavengers and a combination of multiple ROS-detection methods can increase the reliability of the measurements. Besides the detection of ROS generation, proteome-wide analysis of oxidative cysteine PTMs is also critical to uncover the mechanisms and downstream consequences of redox signaling. Increased evidence shows that cysteine PTMs such as sulfenylation, sulfinylation, and glutathionylation can transiently regulate protein activity with critical downstream consequences. The dynamic nature of oxidative PTMs of cysteine has made it a challenge to perform comprehensive and quantitative analysis of these PTMs within native biological environments. Recent advances in chemical probes for *in situ* labeling of cysteine and its PTMs will provide facile and reliable approaches to capture native PTMs within

a physiological context. In parallel, advances in MS instrumentation and quantitative proteomic strategies will enhance the efficiency of identification and quantification of cysteine PTMs. Analytical platforms for ROS and cysteine PTMs highlighted in this review are complementary to each other, and their combined use will offer a robust and comprehensive strategy to dissect redox signaling. The detection of ROS and oxidative cysteine PTMs will benefit from analytical advances that can improve the sensitivity of fluorescence and MS-based analyses, and further, methods to provide spatial and temporal control over measurements will be instrumental in furthering our understanding of redox signaling.

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#### Abbreviations Used

2DE	= two-dimensional gel electrophoresis
2-OH-E <sup>+</sup>	= 2-hydroxyethidium
Akt2	= protein kinase B $\beta$
ATP	= adenosine triphosphate
BCN	= bicycle[6.1.0]nonyne
BES	= benzenesulfonyl fluorescein
BioGEE	= biotinylated glutathione ethyl ester
BioGSH	= biotinylated GSH
CBK1	= caged $\alpha$ -bromomethylketone alkyne
CLA	= 2-Methyl-6-phenylimidazo[1,2-a]pyrazin-3(7H)-one
C <sub>p</sub>	= peroxidatic cysteine
CPH	= 1-hydroxy-3-carboxy-2, 2, 5-tetramethylpyrrolidine hydrochloride
C <sub>R</sub>	= resolving cysteine
CuAAC	= copper(I)-catalyzed alkyne-azide cycloaddition
Cy3	= carbocyanine 3
Cys	= cysteine
DAz	= azide-tagged dimedone
DCFH <sub>2</sub>	= dihydrodichlorofluorescein
DMPO	= 5,5-dimethyl-1-pyrroline-N-oxide
d-PeT	= donor-excited photoinduced electron transfer
DTT	= dithiothreitol
DYn	= alkyne-tagged dimedone
E <sup>+</sup>	= ethidium
EGF	= epidermal growth factor
EGFR	= epidermal growth factor receptor
EPR	= electron paramagnetic resonance
Grx	= glutaredoxin

**Abbreviations Used (Cont.)**

GS = glutathione synthetase  
 GSH = reduced glutathione  
 GSNO = S-nitrosoglutathione  
 GSSG = oxidized glutathione  
 H<sub>2</sub>O<sub>2</sub> = hydrogen peroxide  
 HE = hydroethidine  
 HRP = horseradish peroxidase  
 IAA = iodoacetamide alkyne  
 IAM = iodoacetamide  
 ICAT = isotope-coded affinity tag  
 isoTOP-ABPP = isotopic tandem orthogonal  
   proteolysis activity-based  
   protein profiling  
 LC/LC-MS/MS = tandem liquid chromatography-tandem  
   mass spectrometry  
 MCLA = 6-(4-methoxyphenyl)-2-methyl-  
   imidazo[1,2-a]pyrazin-3(7H)-one  
 MitoPY1 = mitochondria peroxy yellow 1  
 MS = mass spectrometry  
 NADPH = reduced nicotinamide adenine  
   dinucleotide phosphate  
 NBzF = 5-(4-nitrobenzoyl)carbonylfluorescein  
 NBzF-BG = O<sup>6</sup>-benzylguanine-conjugated NBzF  
 NEM = N-ethylmaleimide  
 NF-κB = nuclear factor-κB  
 •NO = nitric oxide  
 NO-Bio = biotinylated 2-nitroso terephthalic acid  
 NO-Ph = 2-nitroso terephthalic acid  
 NOX = NADPH oxidase  
 O<sub>2</sub> = oxygen

O<sub>2</sub><sup>•-</sup> = superoxide  
<sup>-</sup>OCI = hypochlorite  
 •OH = hydroxyl radical  
 ONOO<sup>-</sup> = peroxy nitrite  
 OxICAT = oxidative isotope-coded affinity tag  
 PF1 = Peroxyfluor-1  
 PG1 = Peroxy Green 1  
 PMA = phorbol 12-myristate 13-acetate  
 PO1 = Peroxy Orange 1  
 PR1 = Peroxyresorufin-1  
 Prx = peroxiredoxin  
 PTM = post-translational modification  
 PTP1B = protein tyrosine phosphatase 1B  
 PY1 = Peroxy Yellow 1  
 redox-DIGE = redox differential in-gel electrophoresis  
 RNS = reactive nitrogen species  
 roGFP = redox-sensitive green fluorescent  
   protein  
 ROS = reactive oxygen species  
 SHP = Src homology domain 2-containing  
   protein tyrosine phosphatase  
 SNL = sulfinic-acid-nitroso ligation  
 SOD = superoxide dismutase  
 Srx = sulfiredoxin  
 Sulfo-NHS-biotin = sulfosuccinimidyl-6-(biotinamido)-  
   hexanoate  
 TAM = triarylmethyl  
 TCEP = tris(2-carboxyethyl)phosphine  
 TPP = triphenylphosphonium  
 Trx = thioredoxin  
 UV = ultraviolet