

## Putative Role of Bacterial Biosorbent in Metal Sequestration Revealed by SEM–EDX and FTIR

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Received: 2 January 2019 / Accepted: 10 January 2019 / Published online: 21 January 2019  
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**Abstract** Bacterial exopolysaccharides (EPS) play a critical role in sequestration of metals from contaminated environment. Considering these, this study was aimed at extracting EPS from metal tolerant *Pseudomonas aeruginosa* CPSB1 and *Azotobacter chroococcum* CAZ3 and to ascertain its role in metal removal. *P. aeruginosa* CPSB1 and *A. chroococcum* CAZ3 secreted 1306.7 and 1660  $\mu\text{g mL}^{-1}$  EPS, respectively in the presence of 200 and 100  $\mu\text{g mL}^{-1}$  Pb, respectively with glucose as C source. The binding of metal ions to bacterial EPS was validated by SEM and EDX. The functional group involved in metal chelation was revealed by FT-IR. The metal ions were adsorbed onto EPS and hence, EPS could play a crucial role in metal detoxification. Due to this novel trait, *P. aeruginosa* CPSB1 and *A. chroococcum* CAZ3 could be developed as bioinoculant to cleanup metal contaminated sites.

**Keywords** Exopolysaccharides · Heavy metals · Metal sequestration · *P. aeruginosa* · *A. chroococcum*

Heavy metals are significant environmental threat, which, when present beyond threshold levels in the environment, disrupt the ecological niches. However, certain metal tolerant bacterial strains have been found potentially magical in detoxifying polluted environment. For this, the metal tolerant bacterial strains have evolved multiple strategies to remediate metal contaminated soils. For example, metal

biosorption, extracellular precipitation, conversion of toxic metal ions into less toxic forms and flush out (efflux pumping) of metals to exterior environment are some of the approaches adopted by bacteria to thrive well even under metal stressed conditions [1]. Apart from these, bacterial cells also synthesize extracellular polymeric substances which allow them to survive even in the presence of stressor molecules by masking their toxic impact [2, 3]. Also, EPS plays a significant role in metal chelation, wherein, the ionic forms of metals bind to the complex polymeric structures of EPS. The secretion of EPS by bacterial cells is thus an interesting biological phenomenon to shield themselves from the harsh environment by forming a biofilm matrix on solid substrates. Moreover, the structure and composition of EPS is such that it amply allows easy sequestration of metal ions. Due to these functional properties, microbial polymers have largely been employed in metal removal from polluted environments [4]. Considering the importance of EPS in metal sequestration and detoxification, the present study was aimed at searching metal tolerant bacterial strains capable of synthesizing EPS under both metal stressed and conventional environments. Also, the uptake and localization of metals within EPS was determined by SEM and EDX while functional moieties of EPS were detected by FTIR analysis.

The production of EPS by metal tolerant bacterial strains under conventional and metal stressed conditions was determined by growing *P. aeruginosa* CPSB1 and *A. chroococcum* CAZ3 in 50 mL nutrient broth containing 5% glucose and treated with 0, 25, 50, 100 and 200  $\mu\text{g mL}^{-1}$  each of Cu, Cd, Cr, Ni and Pb. The bacterized flasks were incubated at  $28 \pm 2$  °C for 5 days on a rotary shaker incubator at 120  $\text{r min}^{-1}$ . The culture broth was spun at 8000 rpm for 30 min and three volume of chilled acetone

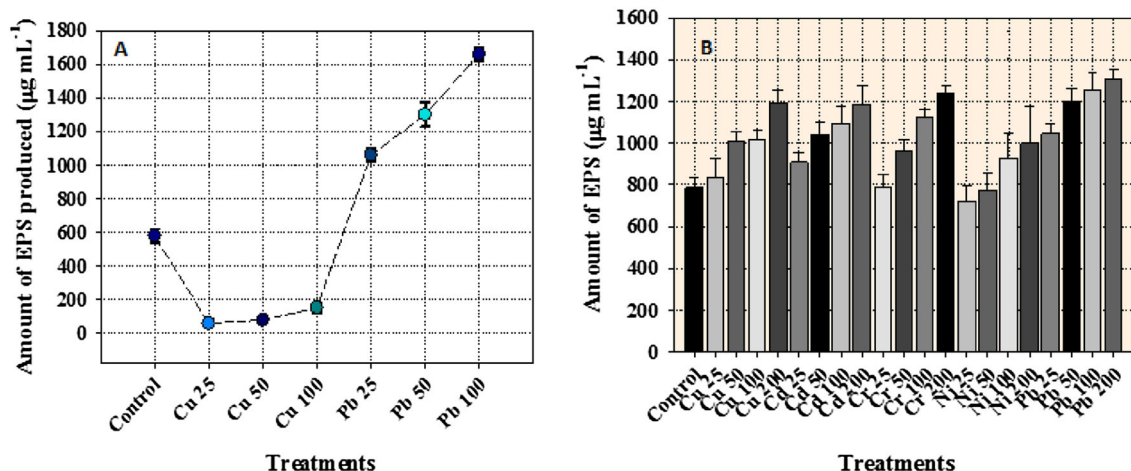
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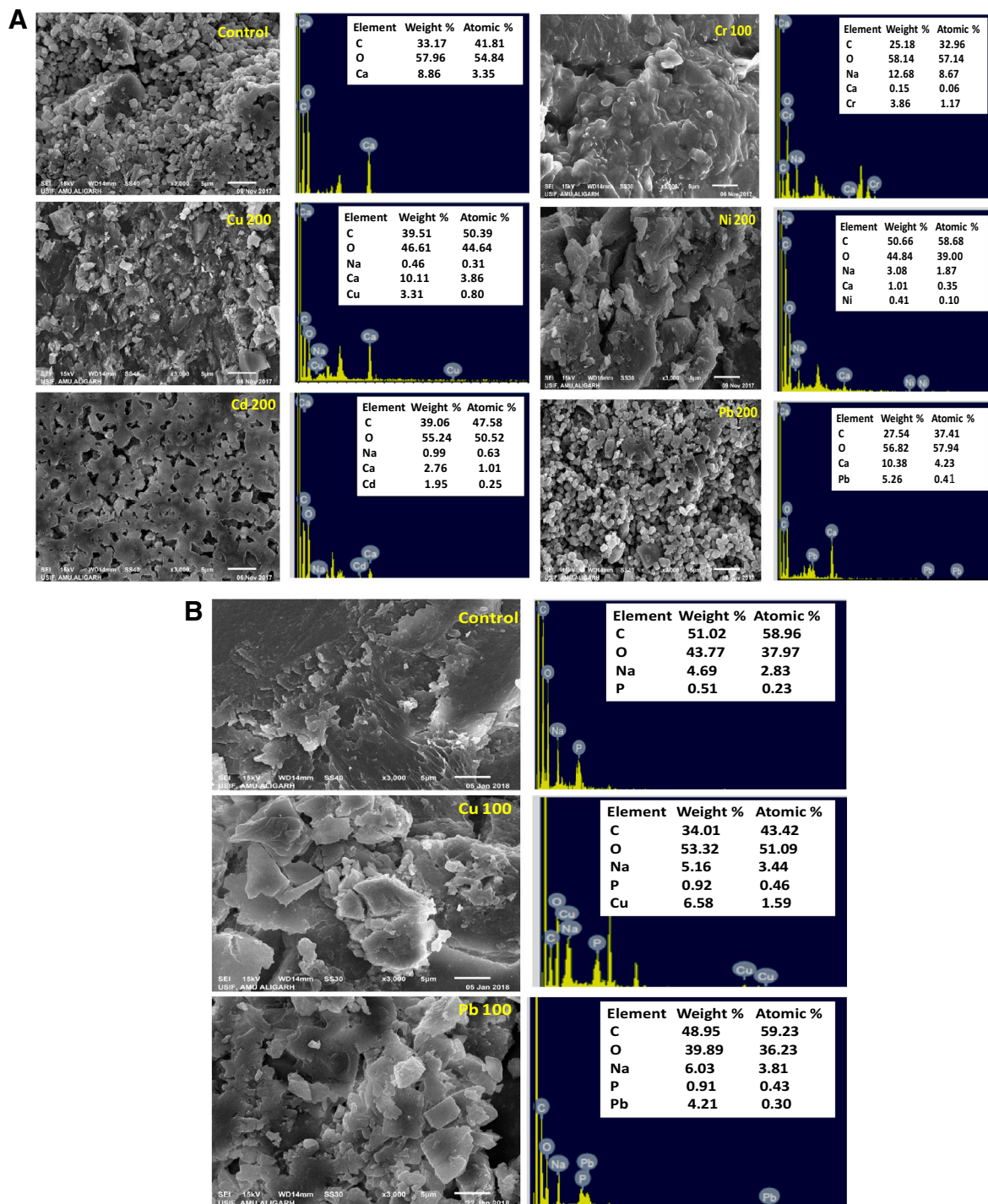
was added to one volume of supernatant and EPS was extracted. The extracted EPS was washed three times with distilled water and acetone and later transferred to a filter paper (No. 42) and weighed. The structural composition, metal distribution and sequestration of powdered EPS were later characterized by SEM, EDX and FTIR techniques.

Exopolysaccharides synthesized by microbial cells provide protection against stressful conditions including those against heavy metals and desiccation [5, 6] besides acting as a biosorbent [4]. Apart from these, the EPS synthesized by microbial communities also participates in biofilm formation (quorum sensing) and hence, acts as a barrier against antibacterial drugs. However, the QS inhibitors (QSIs) of diverse origins have been shown to act as potential antipathogens [7]. Considering these, the production of EPS by *P. aeruginosa* CPSB1 (GenBank Accession number KX821717) and *A. chroococcum* CAZ3 (GenBank Accession number MG252731) was assessed under metal stress. Interestingly, both Gram negative bacterial genera secreted considerable amounts of EPS even under metal stressed conditions though the level of production differed among bacterial strains and were greatly influenced by metal species. For example, *A. chroococcum* CAZ3 secreted  $153.3 \mu\text{g mL}^{-1}$  EPS when grown in the presence of  $100 \mu\text{g mL}^{-1}$  of Cu which was however, increased by 65% when it was grown in the presence of  $100 \mu\text{g mL}^{-1}$  of Pb when compared with control (Fig. 1a). Similarly, EPS production by *P. aeruginosa* CPSB1 was enhanced by 34, 34, 37, 21 and 40% when it was grown at  $200 \mu\text{g mL}^{-1}$  each of Cu, Cd, Cr, Ni and Pb, respectively relative to untreated control (Fig. 1b). While comparing the secretion of EPS by two strains, *A. chroococcum* CAZ3 in general demonstrated maximum production under metal stress. Among metals, Pb was found as the strongest inducer of EPS synthesis. The EPS micrographs of metal

treated bacterial cells under SEM revealed a definite polymeric structure whereas the uptake and localization of metal by *P. aeruginosa* CPSB1 (Fig. 2a) and *A. chroococcum* CAZ3 (Fig. 2b) was confirmed by EDX spectra. The fact that EPS could chelate heavy metal ions was strengthened by comparing the EDX spectra of EPS obtained from metal treated and control cells. The FT-IR data indicating functional groups was recorded in the range of  $1000\text{--}4000 \text{ cm}^{-1}$ . Multiple peaks corresponding to varied functional groups were obtained in the spectra of metal loaded and unloaded EPS. Some deviations and shifting in peaks of functional groups was observed in the presence of metals. The EPS extracted from *P. aeruginosa* CPSB1 for instance, displayed a broad peak in the range of wave number  $2819\text{--}1928 \text{ cm}^{-1}$  in the presence of Cu which could be designated to asymmetrical stretching of  $-\text{CH}_3$  group of fatty acids. Similarly in the presence of Cd and Cr, broad peaks in the range of  $2606\text{--}1874$  and  $2522\text{--}1866 \text{ cm}^{-1}$  respectively, were recorded which also corresponded to  $-\text{CH}_3$  stretching in fatty acids. Lead in contrast, resulted in a peak that ranged between  $2926$  and  $1775$  which represented asymmetrical stretching of  $> \text{CH}_2$  group. Several small peaks were also observed at different regions of EPS which were slightly shifted in position compared to control (Fig. 3). Similar variation in FTIR spectra representing modifications in the functional group moieties of metal loaded EPS extracted from *A. chroococcum* CAZ3 was observed (data not shown). The shifting in peaks relative to control could possibly be due to the disturbances in various functional groups attached to the surface of metal loaded EPS resulting due to the binding of metal ions with the bacterial EPS. The disappearance and irregularity in peaks of EPS extracted from other bacterial genera while growing in the presence of metal stress have also been reported by [8].



**Fig. 1** Quantification of EPS synthesized by *A. chroococcum* CAZ3 (a) and *P. aeruginosa* CPSB1 (b) under controlled and metal stressed environments

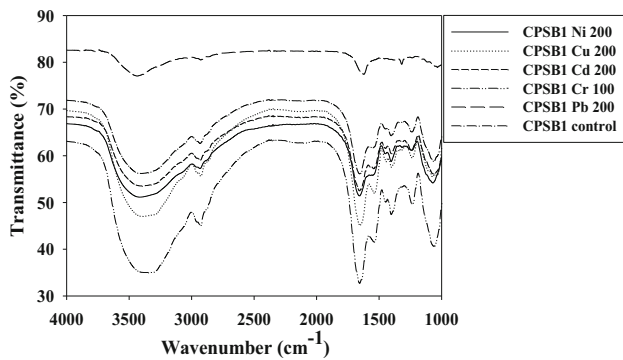


**Fig. 2 a** SEM micrographs and EDX spectra of EPS secreted by *P. aeruginosa* CPSB1 while growing under metal stressed and free conditions. The EDX shows elemental composition of EPS. **b** SEM

micrographs and EDX spectra of EPS secreted by *A. chroococcum* CAZ3 under metal stressed and free conditions. The EDX spectra represent the elemental composition of EPS

The secretion of EPS by metal tolerant Gram negative *P. aeruginosa* and *A. chroococcum* under metal stress and its further characterization by SEM, EDX and FTIR techniques explicitly revealed that EPS could serve as a potential and reliable biomolecule in safeguarding the

rhizosphere microflora from the destruction/damage caused by heavy metals while allowing them to maintain their metabolic activities even under stressful conditions. Summarily, the synthesis of EPS by bacterial cells is likely to serve as an important emerging eco-friendly approach for



**Fig. 3** FT-IR spectra of EPS extracted from metal treated and untreated cells of *P. aeruginosa* CPSB1 showing characteristic peaks corresponding to various functional groups involved in metal ion binding to EPS

heavy metal detoxification and hence, could practically be applied as a bioremediation strategy in metal contaminated environments.

**Acknowledgements** The funding for this work was provided by Department of Science and Technology, New Delhi, India as INSPIRE fellowship (DST/INSPIRE Fellowship/2014/IF140773). One of the authors AR is highly thankful to University Sophisticated Instruments Facility (USIF), Aligarh Muslim University, Aligarh and Macrogen Inc., Seoul, South Korea for the analyses.

#### Compliance with Ethical Standards

**Conflict of interest** The authors declare no competing interests.

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