Letter

Novel Phenyldiazenyl Fibrate Analogues as PPAR $\alpha/\gamma/\delta$ Pan-Agonists for the Amelioration of Metabolic Syndrome

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Supporting Information



ABSTRACT: The development of PPAR α/γ dual or PPAR $\alpha/\gamma/\delta$ pan-agonists could represent an efficacious approach for a simultaneous pharmacological intervention on carbohydrate and lipid metabolism. Two series of new phenyldiazenyl fibrate derivatives of GL479, a previously reported PPAR α/γ dual agonist, were synthesized and tested. Compound **12a** was identified as a PPAR pan-agonist with moderate and balanced activity on the three PPAR isoforms (α , γ , δ). Moreover, docking experiments showed that **12a** adopts a different binding mode in PPAR γ compared to PPAR α or PPAR δ , providing a structural basis for further structure-guided design of PPAR pan-agonists. The beneficial effects of **12a** were evaluated both *in vitro*, on the expression of PPAR target key metabolic genes, and *ex vivo* in two rat tissue inflammatory models. The obtained results allow considering this compound as an interesting lead for the development of a new class of PPAR pan-agonists endowed with an activation profile exploitable for therapy of metabolic syndrome.

KEYWORDS: PPARs, pan-agonist, docking, metabolic syndrome, type 2 diabetes mellitus, gene expression, PGE₂

P eroxisome proliferator-activated receptors (PPARs) are members of the nuclear receptor superfamily and play a decisive role in the regulation of lipid metabolism and glucose homeostasis. PPARs exist as three subtypes with different localizations and physiological functions, designated as PPAR α , PPAR γ , and PPAR δ .¹ These receptors are targets for the therapy of metabolic syndrome, a group of risk factors for cardiovascular disease and type 2 diabetes mellitus (T2DM) including insulin resistance, elevated fasting blood glucose, hypertension, obesity, and atherogenic dyslipidemia.¹ Furthermore, metabolic syndrome is accompanied by a lowgrade inflammatory state in white adipose tissue and liver, which may exacerbate insulin resistance and diabetes.² Fibrates

(e.g., fenofibrate and gemfibrozil), a class of lipid-lowering drugs, are PPAR α ligands.³ Thiazolidinediones (TZDs), such as rosiglitazone and pioglitazone, are insulin-sensitizing agents that activate PPAR γ .⁴ PPAR δ agonists have not yet reached the market, but some of them, for example, GW501516, GW0741, L165,041, GFT505, and MBX8025, are being investigated for a possible therapeutic usefulness in the treatment of metabolic

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disorders, inflammation, and angiogenesis (Figure S1).⁵ However, due to the adverse effects associated with these drug treatments,^{3,4} new research strategies have been undertaken with the aim to obtain new PPAR ligands with reduced side effects and improved beneficial effects. One of these strategies consists in the synthesis of PPAR α/γ dual or PPAR $\alpha/\gamma/\delta$ pan-agonists, which beneficially alter carbohydrate and lipid metabolism in a coordinated manner.⁶⁻⁸ In particular, pan-agonists would merge the agonist activities with regard to PPAR α , - γ , and - δ in a unique ligand but with a balanced activation profile, which could be useful for treatment of metabolic syndrome and T2DM. In the search for new PPAR ligands with these properties, we have recently discovered a series of derivatives of the antilipidemic drugs gemfibrozil or clofibric acid, in which the clofibric acid skeleton was combined with the lipophilic scaffolds derived from natural products, such as α -asarone, stilbene, chalcone, and their bioisosteres.^{9–12} In particular, the combination of the clofibric acid with a phenyldiazenyl function resulted in GL479 (R = H, X = O, Y = CH₂; Figure 1), a PPAR α/γ dual agonist.¹³



Figure 1. General features of synthetic PPAR agonists and structural development of the presented compounds.

Cocrystal structures of GL479 with PPAR α and - γ revealed that the ligand adopts different binding modes at these receptor subtypes, reflecting the distinct activation profile observed for each receptor.¹⁴

Herein we wish to report the synthesis and biological evaluation of a series of derivatives of GL479 (compound 11a, Table 1) in which (i) small substituents with different stereochemical properties were inserted at the para position of the distal aromatic ring of phenyldiazenyl moiety (compounds 11b-g) and (ii) the oxygen atom of the linker was moved to the para position of the phenoxy-propanoic acid (compounds 12a-f). Such design strategy was planned taking into account the crystallographic pose of GL479 into both PPAR α and - γ : the hydrophobic tail binds upward into arm II of PPAR α , where it explores a relatively large accessible volume. Instead, this group bends downward into arm III of PPARy, surrounded by hydrophobic residues such as I388, 1325, M329, and F266. Therefore, with the aim of probing further binding interactions in this region of the protein, we further explored the introduction of small substituents at the para position of the phenyldiazenyl group. Furthermore, C285 on H3, which assumes two possible conformations reflecting a certain degree of flexibility, and the oxygen in the linker of GL479 are at a favorable distance for H-bonding. This cysteine residue is also conserved in PPAR α . Thus, in the attempt to

promote possible interactions with the cysteine residues on H3, we varied the position of the oxygen atom on the linker.

Phenyldiazenyl derivatives 11a-g and 12a-f (Table 1) were easily obtained in good yields using simple synthetic procedures (see Supporting Information for the detailed procedures). The intermediates 2a-g and 3a-f were synthesized according to Scheme 1. Phenols 2a-g were synthesized by diazotization of *para*-substituted anilines with NaNO₂ in 6 N HCl at 0 °C and copulation of the obtained diazonium salts with phenol in 4 N NaOH at 0 °C. The alcohols 3a-f were synthesized by oxidation of *para*substituted anilines by oxone in dichloromethane/water at r.t. and subsequent reaction with 2-(4-aminophenyl)ethanol in CH₃COOH at r.t. (Scheme 1).

Esters 7 and 8 were obtained by S_N^2 reaction of phenols 4 or 5 with ethyl 2-bromo-2-methylpropanoate (6), in the presence of dry K_2CO_3 in DMF at reflux (7) or in the presence of KOH in EtOH at reflux (8) as reported in Scheme 2.

Esters 9a-g and 10a-f were synthesized by Mitsunobu reaction between 2a-g or 3a-f and the appropriate alcohol 7 or phenol 8. These esters were hydrolyzed in basic condition with 1 N NaOH in EtOH at room temperature to obtain the target acids 11a-g and 12a-f (Scheme 2).

All synthesized compounds were evaluated for their agonist activity toward the three subtypes of human PPARs (hPPARs) by a cell-based transactivation assay, according to a previously reported procedure.¹⁵ The results were compared with those obtained with clofibric acid (100 μ M), rosiglitazone (2 μ M), and L-165,041 (2 μ M), used as reference compounds for PPAR α , - γ , and - δ , respectively (Table 1). Maximum fold induction obtained with the reference agonists was set at 100%. The activity of all compounds was evaluated at 25 μ M, and only those showing efficacy percentage (E_{max}) higher than 20% were selected for determination of EC_{50} . Because of the different cell line used in this assay, GL479 turned out to be a partial agonist instead of a full agonist, as reported earlier;¹ moreover, this ligand displayed a significant activation of PPAR δ subtype that had not been previously investigated. Derivatives 11b-g displayed partial agonist activity toward PPAR α . The introduction of a chlorine or bromine resulted in about 3- and 5-fold increase of potency compared to lead compound GL479 (11b, EC₅₀ = $0.38 \pm 0.05 \ \mu\text{M}$; 11c, EC₅₀ = $0.81 \pm 0.06 \ \mu\text{M}$; GL479, EC₅₀ = $2.1 \pm 0.6 \ \mu\text{M}$) and a slight increase of efficacy, suggesting a favorable influence of the weak electron-withdrawing properties of these substituents on the activity. In contrast, the introduction of stronger electronwithdrawing substituents, such as trifluoromethyl and cyanide, afforded compounds with potency comparable to GL479 (11d, $EC_{50} = 2.3 \pm 1.3 \ \mu M$; 11e, $EC_{50} = 2.6 \pm 0.8 \ \mu M$), but a concomitant loss of efficacy from 56% to about 30%. The presence of the highly hydrophilic nitro group caused an almost complete loss of PPAR α activation (11f). Compound 11g, bearing the electron-donating group methoxy, showed a similar activity to the derivative 11d. Therefore, we argued that the electronic properties of the substituent in the para position of the distal aromatic ring of phenydiazenyl moiety do not seem to exert a crucial role on the activity. Surprisingly, compounds 11b-g showed poor PPAR γ activity as well as a weak activation of PPAR δ . In particular, the substitution with trifluoromethyl or methoxy increased the potency toward this receptor subtype, with 11d and 11g being the most potent.

Derivatives 12a-f displayed partial agonism on PPAR α . Compound 12a, a close analogue of GL479, showed about 8-

Table 1. In Vitro Transactivation Activity of Compounds with Various Lipophilic Tails and Linkers

N^NN^NCOOH

				hPPARa	hPPARγ	hPPAR δ
compd	R	Х	Y	$\mathrm{EC}_{50} \left(\mu\mathrm{M}\right)^{a} \left(E_{\mathrm{max}}\right)^{b}$	$EC_{50} (\mu M)^a (E_{max})^b$	$\mathrm{EC}_{50} \left(\mu\mathrm{M}\right)^{a} \left(E_{\mathrm{max}}\right)^{b}$
11a (GL ₄₇₉)	Н	0	CH ₂	$2.1 \pm 0.6 (56 \pm 9)$	n.d. (13 ± 1)	$6.4 \pm 1.2 \ (39 \pm 4)$
11b	Cl	0	CH ₂	$0.38 \pm 0.05 \ (60 \pm 6)$	n.d. (12 ± 1)	$3.1 \pm 0.4 \ (23 \pm 2)$
11c	Br	0	CH ₂	$0.81 \pm 0.06 \ (71 \pm 10)$	n.d. (12 ± 3)	$3.9 \pm 0.4 (29 \pm 2)$
11d	CF ₃	0	CH ₂	$2.3 \pm 1.3 (35 \pm 8)$	n.d. (8 ± 1)	$1.8 \pm 0.4 \ (24 \pm 3)$
11e	CN	0	CH ₂	$2.6 \pm 0.8 \ (25 \pm 7)$	n.d. (12 ± 1)	$6.5 \pm 1.2 \ (20 \pm 4)$
11f	NO ₂	0	CH ₂	n.d. (11 ± 7)	n.d. (14 ± 1)	$4.1 \pm 1.0 \ (21 \pm 3)$
11g	OCH ₃	0	CH ₂	$2.4 \pm 0.8 (32 \pm 2)$	n.d. (10 ± 1)	$1.4 \pm 0.2 \ (23 \pm 2)$
12a	Н	CH_2	0	$0.25 \pm 0.07 \ (69 \pm 12)$	$6.0 \pm 1.5 \ (24 \pm 3)$	$2.8 \pm 0.3 (35 \pm 2)$
12b	Cl	CH_2	0	$0.51 \pm 0.12 \ (52 \pm 9)$	n.d. (10 ± 1)	$1.2 \pm 0.3 \ (41 \pm 6)$
12c	Br	CH ₂	0	$1.9 \pm 0.2 \ (35 \pm 4)$	n.d. (8 ± 1)	$1.1 \pm 0.2 \ (29 \pm 3)$
12d	CF ₃	CH ₂	0	$2.2 \pm 1.2 (37 \pm 7)$	n.d. (9 ± 1)	n.d. (12 ± 4)
12e	CN	CH ₂	0	$1.4 \pm 0.1 \ (61 \pm 11)$	n.d. (12 ± 2)	$2.8 \pm 0.3 (50 \pm 3)$
12f	NO_2	CH ₂	0	n.d. (19 ± 8)	n.d. (10 ± 1)	$2.3 \pm 0.4 \ (28 \pm 3)$
clofibric acid				$50 \pm 6 (100 \pm 10)$	-	-
rosiglitazone				-	$0.04 \pm 0.02 \ (100 \pm 9)$	-
L165,041				-	-	$0.21 \pm 0.04 \ (100 \pm 4)$

 ${}^{a}\text{EC}_{50}$ values were determined by testing compounds in at least three separate experiments at five concentrations ranging from 1 to 50 or 100 μ M. The results are expressed as the mean \pm SEM. b Efficacy values were calculated as percentage of the maximum fold induction obtained with clofibric acid, rosiglitazone, and L165,041 as reference compounds for PPAR α , PPAR γ , and PPAR δ , respectively. n.d.: EC₅₀ not determined.





^aReagents and conditions: (a) NaNO₂, 6 N HCl, H₂O, 0-5 °C, 1 h; (b) phenol, 4 N NaOH, 0-5 °C, 1 h; (c) oxone, CH₂Cl₂, H₂O, r.t., 4 h; (d) 2-(4-aminophenyl)ethanol, CH₃COOH, r.t., 24–48 h.

fold potency increase and a concomitant slight increase of $E_{\rm max}$ (about 70%). Substitutions at the *para* position of the phenyldiazenyl moiety afforded compounds displaying lower potency and efficacy compared to **12a**. The chloro (**12b**) and cyano (**12e**) derivatives were the most active. Again, the presence of the hydrophilic and strong electron-withdrawing nitro group caused an almost complete loss of PPAR α activation (**12f**). As the activity is not strikingly affected from the electronic properties of the *para*-substituent of the distal aromatic ring of phenyldiazenyl moiety and given that the preparation of the derivative bearing the methoxy was somewhat troublesome, this last compound was not synthesized. Derivatives **12b**-f displayed very weak activation of PPAR γ ; only the unsubstituted derivative **12a** showed higher potency and efficacy than the lead compound GL479.





"Reagents and conditions: (a) dry K_2CO_3 , DMF, reflux, 4 h, (Y = CH₂CH₂OH) (4 \rightarrow 7); (b) KOH, EtOH, reflux, N₂, 72 h, (Y = OH) (5 \rightarrow 8); (c) PPh₃, DIAD, N₂, THF dry, r.t., 10–12 h; (d) 1 N NaOH, EtOH, r.t., 15–24 h.

Regarding PPAR δ , all derivatives 12a-f displayed a partial agonism profile toward this receptor subtype. Introduction of chlorine or bromine increased potency with respect to the unsubstituted derivative 12a, whereas the introduction of



Figure 2. Binding mode of compound 12a (magenta sticks) into the PPAR α (A) and $-\delta$ (B) LBDs, represented as green and dirty violet ribbon models, respectively. Only amino acids located within 4.5 Å of the bound ligand are displayed (white sticks) and labeled. H-bonds discussed in the text are depicted as dashed gray lines. H12 is shown in slate.



Figure 3. Expression of *PDK4* (A), *CPT1A* (B), and *GLUT1* (C) after treatment with PPAR agonist **12a** (100 μ M). The enhanced activity on the gene expression is compared with the commercial compounds L165,041 (2 μ M, A), GW7647 (2 μ M, B), or rosiglitazone (10 μ M, C) as reference positive controls.

trifluoromethyl group was detrimental. The cyano and nitro derivatives displayed a similar potency to 12a. Overall, this second series of derivatives (12a-f) displayed higher potency and a slightly higher efficacy than the corresponding derivatives of the first series (11a-f) toward PPAR δ . This suggests that the shift of the oxygen of the linker to the para position of the phenoxy group is more favorable to the interaction and resulting activation of PPAR δ . Interestingly, **12a** resulted more potent than GL479 but with basically the same significant and well-balanced activity on all three PPAR subtypes. In fact, 12a exhibits a weak/moderate efficacy, which is desirable given that moderately effective PPAR agonists demonstrate good tolerability and safety in a large population of patients.^{16–18} In addition, the partial agonist activity of 12a allows hypothesizing that this compound behaves like a selective PPAR modulator (SPPARM), which is a ligand that, compared with a full agonist, differentially induces specific receptor effects, ideally uncoupling the benefits of PPAR activation from the adverse side effects.¹⁹ Until a few years ago, SPPARM activity was pursued only for PPARy agonist, but nowadays, it is considered desirable also for the other two subtypes.²⁰ As far as we know, 12a is one of the few pan-agonists showing moderate activity toward all three PPAR subtypes. This is noteworthy if one considers that most pan-agonists introduced and then discontinued in clinical trials (e.g., indeglitazar, etc.)²¹ were full agonists at least toward one of the three subtypes.

To help interpretation of SAR data and to increase our understanding of the molecular basis for the observed partial agonism of **12a** toward PPAR α , $-\gamma$, and $-\delta$, we undertook docking simulations and cluster analysis (see details in the Supporting Information). Compound **12a** occupies arms I and II in both PPAR α and $-\delta$ (Figure 2), whereas it interacts

mainly with arms II and III in PPARy (Figure S7). As illustrated in Figure 2A,B, the clofibric acid moiety of 12a is located in the arm I, making conventional interactions with the polar side chains (S280, Y314, H440, and Y464 for PPAR α and T253, H287, H413, and Y437 for PPAR δ), the central phenoxy ring is placed in the center, and the hydrophobic tail part is buried in the arm II. In the PPAR α LBD complex, the gem-dimethyl substituents are directed into the lipophilic "benzophenone pocket", lined by F273, Q277, V444, and L456.²² However, in PPAR δ V444 is substituted with a methionine, which has a bulkier side chain and is less prone to accommodate the fibrate headgroup.²³ This might account for the potency increase of 12a toward PPAR α and for the decrease of PPAR δ potency. Noteworthy, the phenyldiazenyl moiety in PPAR α forms sulfur-aromatic interactions²⁴ with C275, C276, M355, and M330. This remarkable environment might rationalize the augmented potency of 12a in the PPAR α pocket, which is more lipophilic and less solvent-exposed than the corresponding pocket of PPAR δ .²⁵

In fact, with the exception of C276, sulfur-containing residues are not conserved in PPAR δ . The 10-fold potency decrease of **11a** compared to **12a** in PPAR α can be explained based on the valence-bond (or resonance) theory (see details in the Supporting Information).

Compound 12a binds to PPAR γ LBD in a different binding mode as compared to that of either PPAR α LBD or PPAR δ LBD because it does not interact directly with H12 (Supporting Information, Figure S7). Therefore, the low affinity and the attenuated transcriptional response of 12a and its derivatives toward PPAR γ might be ascribed to the peculiar binding mode, which is not able to replicate the spectrum of contacts of full agonists.²⁶

Next, we evaluated whether 12a could affect the expression of three PPAR target genes involved in the control of lipid and carbohydrate metabolism, namely, carnitine palmitoyltransferase 1A (CPT1A), pyruvate dehydrogenase kinase 4 (PDK4), and glucose transporter 1 (GLUT1).²⁷ CPT1A is a molecular component of the carnitine shuttle system that catalyzes the entry of fatty acids into the mitochondrial matrix and their subsequent enzymatic oxidation. Importantly, CPT1A gene expression is enhanced by PPAR α agonists.²⁷ PDK4 is a wellestablished PPAR δ target gene and contributes to the regulation of glucose and fatty acid metabolism and homeostasis.²⁷ GLUT1 is a constitutive glucose transporter whose expression is increased by PPAR γ ligands, resulting in enhanced glucose uptake and utilization into adipocytes.²⁸ Therefore, a screening of gene activation by real-time quantitative PCR (RTqPCR) analysis was performed for compound 12a in HepG2 human hepatocellular carcinoma cell line, using the PPAR δ agonist L165,041, the PPAR α agonist GW7647, and the PPAR γ agonist rosiglitazone as reference positive controls. As illustrated in Figure 3, treatment with 12a (100 μ M) showed enhanced activity on the expression of PDK4 and CPT1A similar to the agonists L165,041 (2 μ M) and GW7647 (2 µM, Figure 3A,B). Moreover, 12a significantly increased the mRNA expression level of GLUT1 as compared to rosiglitazone (10 μ M, P < 0.01; Figure 3C). The choice of performing the experiment at the fixed concentration of 100 μ M was based on the lower potency and/or efficacy of 12a in the transactivation assay compared to the reference compounds as reported in Table 1. However, these results indicate that 12a, even though less potent than positive controls, through its simultaneous and balanced activation of the three PPAR subtypes, may have a beneficial outcome on carbohydrate and lipid metabolism. In light of these results, for a deeper investigation of the biological properties of 12a, we decided to evaluate its anti-inflammatory activity. Numerous studies indicate that the role of PPARs in inflammation is particularly significant in the case of metabolic syndrome and atherosclerosis, which present an inflammatory component.² This allows hypothesizing that the beneficial effects of PPAR activation on insulin sensitivity are mediated, at least in part, by its anti-inflammatory activities. For this reason, we conducted ex vivo experiments for evaluating the ability of 12a to reduce the LPS-induced PGE2 production in liver and cortex of male adult Sprague-Dawley rats. LPS is often used as a stimulus for the induction of inflammationrelated disorders simulating metabolic diseases.²⁹

To assess the degree of inflammation, we measured the levels of prostaglandin PGE2, which can contribute to diabetes by inhibiting insulin secretion, leading to impaired glucose tolerance and insulin sensitivity.³⁰ Rats were sacrificed and both liver and cortex specimens were immediately collected and maintained in a humidified incubator in RPMI (Roswell Park Memorial Institute) buffer with added bacterial LPS (10 μ g/mL). During the incubation period, tissues were treated alternatively with 1 μ M WY14643, pioglitazone, and L165,041 as selective PPAR α , - γ , and - δ agonists, respectively, or with scalar concentrations of 12a (0.1-10 μ M). Compound 12a was able to significantly inhibit LPS-induced PGE2 production in both rat liver (Figure 4A) and cortex (Figure 4B), two tissues that play a central role in the regulation of glucose homeostasis and metabolism, in a nondose-dependent manner. The efficacy of 12a was comparable with that of WY14643 and pioglitazone, whereas the PPAR δ agonist L165,041 did not



Figure 4. Effects of **12a** (0.1–10 μ M) on LPS-induced production of PGE2 (pg/mg) in rat liver (A) and cortex (B), *ex vivo*. Data were reported as means ± SEM. Results were analyzed by analysis of variance (ANOVA) followed by Newman–Keuls posthoc test. ANOVA, *p* < 0.0001; posthoc ****p* < 0.001 vs LPS-treated group.

cause any decrease of the LPS-induced PGE2 production in both tissues.

In conclusion, this study led to the identification of **12a** endowed with moderate activity on the three PPAR subtypes (α, γ, δ) , which might represent a potential lead for the development of a new class of PPAR pan-agonists for treating metabolic syndrome and T2DM.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsmedchem-lett.8b00574.

Complete experimental procedures, characterization data for all compounds, biological procedures, docking and cluster analysis, as well as spectra (PDF)

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The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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ABBREVIATIONS

PPARs, peroxisome proliferator-activated receptors; T2DM, type 2 diabetes mellitus; TZDs, thiazolidinediones; SAR, structure-activity relationship; CPT1A, carnitine palmitoyltransferase 1A; PDK4, pyruvate dehydrogenase kinase 4; GLUT1, glucose transporter 1; LPS, lipopolysaccharide; LBD, ligand binding domain

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NOTE ADDED AFTER ASAP PUBLICATION

This paper was published ASAP on February 28, 2019 with errors in Table 1. The corrected paper was reposted on March 4, 2019.