# $\alpha$ , $\gamma$ -Diketocarboxylic Acids and Their Esters Act as Carbonic Anhydrase IX and XII Selective Inhibitors

Alessio Nocentini,<sup>†,‡</sup> Alessia Lucidi.<sup>§</sup> Francesca Perut.<sup>∥</sup> Annamaria Massa.<sup>∥</sup> Daniela Tomaselli.<sup>§</sup> Paola Gratteri,<sup>‡</sup> Nicola Baldini, <sup>||,⊥</sup> Dante Rotili,\*<sup>,§</sup> Antonello Mai,\*<sup>,§</sup> and Claudiu T. Supuran<sup>\*,†</sup>

† Department of NE[UR](#page-3-0)OFARBA, Section of Pharmaceutical an[d N](#page-3-0)utraceutical Science[s, Univ](#page-3-0)ersity of Florence, Polo Scientifi[co](#page-3-0), [V](#page-3-0)ia U. Schiff 6, 50019 Sesto Fiorentino, Firenze, Italy

‡ Department of NEUROFARBA, Section of Pharmaceutical and Nutraceutical Sciences, Laboratory of Molecular Modeling Cheminformatics & QSAR, University of Florence, Polo Scientifico, Via U. Schiff 6, 50019 Sesto Fiorentino, Firenze, Italy

 $^{\$}$ Dipartimento di Chimica e Tecnologie del Farmaco "Sapienza" Università di Roma, 00185 Rome, Italy

∥ Laboratorio di Fisiopatologia Ortopedica e Medicina Rigenerativa, IRCCS Istituto Ortopedico Rizzoli, via di Barbiano 1/10, 40136 Bologna, Italy

 $^{\perp}$ Department of Biomedical and Neuromotor Sciences, University of Bologna, 40123 Bologna, Italy

**S** Supporting Information



ABSTRACT: Among human carbonic anhydrase (CA) inhibitors, the  $\alpha$ , $\gamma$ -diketocarboxylic acids and esters are still poorly investigated. Here, we report the first compounds of this class (1−6) acting as potent inhibitors at low nanomolar level against the cancer-related human CA IX and XII, and 2−3 magnitude orders selective toward the cytosolic isoforms hCA I and II. At enzymatic level, the α,γ-diketoacids 1−3 were more effective inhibitors compared to the corresponding ethyl esters 4−6. The phenyl- and α-naphthyl-containing compounds (1, 3, 4, and 6) behaved as dual hCA IX/XII inhibitors, while the β-naphthyl analogues (2 and 5) exhibited hCA IX-selective inhibition. In MG63 and HOS osteosarcoma (OS) cell lines, the ethyl esters 5 and 6 displayed dose-dependent reduction of viability and proliferation after 72 h treatment, with 6 being more potent than 5 likely for its dual hCA IX/XII inhibition.

KEYWORDS: Metalloenzyme, carbonic anhydrase, carboxylic acid, selective inhibition, antitumor

arbonic anhydrase (CA) inhibitors (CAIs) possess a range of pharmacologic applications for the treatment and prevention of diseases connected with enhanced secretion of electrolytes, pH regulation, metabolic modulation, etc.<sup>1−4</sup> In fact, many CAIs are used as diuretics, antiglaucoma agents, an[ti](#page-3-0)epileptics, antiobesity, and antitumor drugs. $1^{-4}$  In a[dd](#page-3-0)ition to primary sulfonamides and their isosters (sulfamates and sulfamides), several other chemotypes with [CA](#page-3-0) inhibitory properties and diverse mechanisms of CA inhibition were reported in the last years.<sup>5,6</sup> Indeed, sulfonamides and their congeners, as well as dithio-/monothiocarbamates, xanthates, hydroxamates, and some c[arb](#page-3-0)oxylates coordinate the zinc ion from the enzyme active site potently inhibiting the CA catalytic activity. $5,6$  Compounds that inhibit CAs by anchoring to the zinc-coordinated water molecule are the phenols, polyamines, sulfoco[um](#page-3-0)arins, some carboxylic acid esters, and the thioxocoumarins, $7,8$  whereas the coumarins show a distinct

CA inhibition mechanism, acting as prodrug inhibitors, and after hydrolysis to the corresponding 2-hydroxycinnamic acids, occlude the entrance to the active site.<sup>9,10</sup> The fourth inhibition mechanism, out of the active site binding, was also reported for a carboxylic acid derivative.<sup>11</sup> [Th](#page-3-0)us, from the above data it is possible to observe that the carboxylates and their derivatives may interact with these [enz](#page-3-0)ymes by at least four different inhibition mechanisms, making them of great interest for the design of inhibitors with diverse profiles for the many CA isoforms present in vertebrates (15  $\alpha$ -class isoforms) or other organisms (with seven distinct CA genetic families

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Received: January 21, 2019 Accepted: February 19, 2019 Published: February 19, 2019 <span id="page-1-0"></span>present in organisms all over the tree of life). $5,6$  Indeed, several drug design studies of carboxylic acids as CAIs were reported in the last years based on aromatic/heterocy[clic](#page-3-0) scaffolds, with several such derivatives also being characterized by X-ray crystallographic techniques.11<sup>−</sup><sup>14</sup> In some cases, highly isoform-selective inhibitors were detected,<sup>12−14</sup> as well as compounds with enhanced [in](#page-3-0)[hib](#page-4-0)itory activity against pathogenic versus human enzymes.<sup>14</sup>

With the only exception of the carboxylic acid derivatives reported by Sechi et al., whic[h sh](#page-4-0)owed inhibitory activity in the low micromolar/submicromolar range against a panel of human and fungal CAs,<sup>14</sup> the  $\alpha$ ,*γ*-diketo acid scaffold has been poorly investigated for the interaction with CAs so far. Therefore, we designed [a f](#page-4-0)ew 6-aromatically substituted-2,4 dioxo-5-hexenoic acids and their ethyl esters 1−6 (Figure 1) as



Figure 1. New  $\alpha$ , $\gamma$ -diketocarboxylic acids and ethyl esters 1–6 as CAIs (R = H, ethyl; R<sub>1</sub> = phenyl,  $\alpha$ - or  $\beta$ -naphthyl).

potential CAIs, studied their inhibition mechanism by using kinetic and computational techniques, and evaluated their effects in two human osteosarcoma (OS) cell lines, MG63 and HOS, highly expressing the cancer-related CA isoforms IX and XII.

The general synthetic pathway followed for the preparation of derivatives 1−6 is depicted in the Scheme 1. The (E)-4-(4-



<sup>a</sup>Reagents and conditions: (a) 2.8 M SnCl<sub>2</sub> in 37% HCl, EtOH, 0  $^{\circ}$ C, 24 h; (b) appropriate aroyl chloride, Et<sub>3</sub>N, dry DCM, 0 °C, 2–24 h; (c) diethyl oxalate, EtONa, dry THF, N<sub>2</sub> atmosphere, rt, 3-5 h; (d) 0.7 M LiOH, THF/MeOH, rt, 6−8 h.

nitrophenyl)but-3-en-2-one 7, prepared as reported in the literature, $15$  was dissolved in ethanol and treated with a solution of stannous chloride dihydrate in concentrated hydrochl[ori](#page-4-0)c acid at 0 °C to obtain the reduced product (E)-4-(4-aminophenyl)but-3-en-2-one 8, which was then subjected to acylation reaction with the appropriate commercial aroyl chloride in the presence of triethylamine in dry DCM at 0 °C in order to afford the intermediates 9−11. The compounds 4−6, obtained by treating 9−11 with diethyl oxalate in the presence of sodium ethoxide in dry THF at room temperature under nitrogen atmosphere, were finally hydrolyzed to the corresponding  $\alpha$ , $\gamma$ -diketocarboxylic acids 1–3 with 0.7 M lithium hydroxide in a mixture of THF and methanol.

Full experimental details regarding the synthesis and physicochemical characterization of final (1−6) and intermediate (8−11) compounds are outlined in Supporting Information.

The CA inhibition profile of compounds 1−6 w[as evaluated](http://pubs.acs.org/doi/suppl/10.1021/acsmedchemlett.9b00023/suppl_file/ml9b00023_si_001.pdf) [by applying](http://pubs.acs.org/doi/suppl/10.1021/acsmedchemlett.9b00023/suppl_file/ml9b00023_si_001.pdf) a stopped flow  $CO_2$  hydrase assay<sup>16</sup> against four physiologically significant CA isoforms (I, II, IX, and XII), in comparison with acetazolamide (AAZ) as refer[en](#page-4-0)ce drug.

The following SAR can be gathered from the inhibition data shown in Table 1: (i) The cytosolic hCA I and II are the least

Table 1. Inhibitory Potencies of Derivatives 1−6 versus hCA I, II, IX, and XII

> OH  $\Omega$



<sup>a</sup>Mean from three different assays, by a stopped flow technique (errors were in the range of ±5−10% of the reported values).

inhibited isoforms by α,γ-diketocarboxylic acids 1−3 and their ethyl esters 4−6. Most derivatives act as low micromolar inhibitors, with inhibition constant  $(K_I)$  values spanning between 2.08 and 9.4  $\mu$ M. Compounds 1, 3, and 6 do not inhibit hCA I below 10  $\mu$ M, and 4 exhibits the same behavior against hCA II.

Carboxylic acids and mostly esters are known to be relatively less effective CA inhibitors than sulfonamides, with most derivatives of these classes of compounds acting in the micromolar range.<sup>11−14</sup> The present results against the cytosolic isoforms confirm these evidences. No significant difference stood ou[t b](#page-3-0)[etw](#page-4-0)een acids 1−3 and esters 4−6 in the inhibition of hCA I. Conversely, hCA II was more than 2-fold better inhibited by compounds 1−3 than their ethyl esters. Switching the phenyl group to  $\alpha$ - or  $\beta$ -naphthyl moiety does not affect the inhibitory properties against hCA I and II within the two subsets of molecules. (ii) The tumor-associated hCA IX and XII are potently inhibited by both subsets of derivatives. Compounds 1−6 inhibited hCA IX in the low nanomolar range, exhibiting  $K_I$  values spanning between 11.3 and 90.2 nM. The nature of the outer aromatic ring does not affect the hCA IX inhibitory properties of acid derivatives 1−3, being their  $K<sub>I</sub>$  values not significantly diverse (11.3−12.8 nM). Solely the  $\beta$ -naphthyl ethyl ester 5 exhibits analog  $K_I$  (12.6) nM) to carboxylic acids 1−3, whereas the switching of the aromatic moiety to the phenyl (4) and  $\alpha$ -naphthyl (6) ones worsens the inhibitory activity by three- (37.9 nM) or 7-fold (90.2 nM), respectively. (iii) With the exception of the  $\beta$ naphthyl derivatives 2 and 5 ( $K_I$  values = 767.6 and 249.1 nM, respectively), the hCA XII is potently inhibited by the other screened derivatives that show comparable inhibitory properties in the low nanomolar range (7.5−34.6 nM) independently from the nature of the zinc binding group (ZBG). Anyway, it should be noted that the phenyl ester derivative 4 is 4-fold less effective than its carboxylic acid cognate 1. (iv) A clear selectivity of action against hCA IX and XII over the cytosolic hCA I and II stands out from the data in Table 1 for most of the reported derivatives, with  $\beta$ -naphthyl derivatives 2 and 5 that can be considered hCA IX-selective i[nhibitor](#page-1-0)s.

The *in silico* assessment of the binding mode of compounds 2 and 5 into the hCA II and IX highlights that 2-hydroxy-4 oxohexa-2,5-dienoic acids, and their ethyl esters are able to occupy the catalytic region of the binding pockets by coordinating the zinc ion and interacting with residues nearby. Docking studies followed by a refinement in a VSGB solvent model show that the coordination occurs between the metal ion and the deprotonated carboxylic group of 2 or the oxogroup of the ethyl ester of 5 (Figure 2). The molecular



Figure 2. Docking of 2 (A) and 5 (B) into hCA IX. Docking of 2 into  $hCA$  II  $(C)$ .

architecture of the two active sites thoroughly affects the binding modes. The hCA II/IX Phe131/Val130 mutation modulates the H-bonds network that the 2-hydroxy-4-oxohexa-2,5-dien portions can form within the pockets. The presence of several H-bonds effectively reinforces the carboxylate coordination to the metal ion of 2 and 5 in hCA IX (Figure 2A,B). The carboxy or carbethoxy moieties of 2 and 5, respectively, accept two H-bonds from the backbone NH of Thr200 and Thr201. The side chain OH group of Thr201 is involved in an interesting pattern of H-bonds with the 2-hydroxy-4-oxohexa-2,5-diene portions of the ligands under investigation, acting both as donor and acceptor group. In particular, the OH group acts as both H-bond donor toward the 2-hydroxy and to the 4 oxo moieties of 2, while it participates to a three center Hbond involving the analogue deprotonated groups in 5. The ethyl moiety of 5 accommodates into the pocket lined by Val121, Val142, and Trp210. The naphthamidophenyl fragment of both molecules orients toward the hydrophobic half of hCA IX active site, with van der Waals interactions taking place with Val130, Asp131, and Arg129.

The above-mentioned hCA II/IX Phe131/Val130 mutation makes the hCA II binding site less roomy if compared to that one of hCA IX preventing, de facto, the positioning of the ligand as described for hCA IX. Nonetheless, the carboxylates maintain the zinc-coordination and a H-bond with the backbone NH Thr200. The 2-hydroxy-4-oxohexa-2,5-diene portions lack the proper H-bond distances with Thr201 because of the rotation undergone by the ligands to accommodate the naphthamidophenyl core toward His64, Ans62, and Asn67 (Figure 1C). These evidence support the observed CA IX/II selective inhibition profiles. In fact, the binding mode of 2 [within the](#page-1-0) hCA II active site prevents the coordination bond stabilization, which enhances the hCA IX inhibition efficacy of the compounds more than two-orders of magnitude if compared to that toward hCA II.

Chemotypes endowed with a selectivity ratio spanning between 2 and 3 orders of magnitude for hCA IX and XII over both I and II have a great potential as starting points for the design of novel CAIs as antitumor agents devoid of undesired side effect related to promiscuous activity. Since in a previous paper we have demonstrated that there is a strong rationale for the use of CA IX inhibitors in human OS models, $17$  we tested the  $\beta$ -naphthyl derivatives 2 and 5 together with the  $\alpha$ naphthyl ethyl ester 6 (0−100  $μM$ , 72 h) in two [di](#page-4-0)fferent OS cell lines (MG63 and HOS) that highly express CA IX and/or XII (see Figure S1 in Supporting Information). Likely for the reduced cell permeability due to its acidic nature, 2 did not show any inhibitory eff[ect on OS cell growth](http://pubs.acs.org/doi/suppl/10.1021/acsmedchemlett.9b00023/suppl_file/ml9b00023_si_001.pdf), while the two ethyl esters 5 and 6 affected MG63 and HOS cell viability in a dose-dependent way (Figure 3). In particular, 5 decreased by



Figure 3. Cell viability assay. The treatment of OS cells with the two ethyl esters 5 and 6 for 72 h reduced cell viability in a dose-dependent way, as revealed by the indirect acid phosphatase viability assay. Treatment with 2 did not produce any cytotoxicity on OS cells.

50% the viability of both the tested cell lines at 50  $\mu$ M, while 6 at 25  $\mu$ M arrested the viability of MG63 and HOS cells at 16.4 and 19.7%, respectively, thus being the most potent of the two compounds.

In order to confirm this viability indirect test, the effectiveness of 5 and 6 was then evaluated by erythrosine B dye vital staining in both MG63 and HOS OS cell lines. Both compounds displayed strong antiproliferative effects in the low micromolar range after 72 h treatment (Figure 4), with 6  $[IC_{50}]$ 



Figure 4. Cell proliferation assay. The percentage of cell growth with respect to untreated cells is significantly lower ( $p < 0.05$ ) after the treatment with compound 5 or 6 for all the tested doses.

values = 14.7  $\mu$ M (MG63) and 15.6  $\mu$ M (HOS)] being more potent than 5  $[IC_{50}$  values = 45.8  $\mu$ M (MG63) and 19.2  $\mu$ M (HOS)]. When tested in the same cell lines to determine their putative effect on apoptosis induction through detection of caspase 3 and 7 activity, neither 6 (at 25  $\mu$ M) or 5 (at 50  $\mu$ M) induced any pro-apoptotic activity (see Figure S2 in Supporting Information). Thus, the decreased number of

<span id="page-3-0"></span>cells after compounds 5 and 6 treatment could be a consequence of a cell cycle arrest.

The evidence that the dual CA IX/XII inhibitor  $\alpha$ -naphthyl ester 6 resulted more effective than the CA IX-selective  $\beta$ naphthyl regioisomer 5 in inhibiting the proliferation of both the OS cell lines suggests the importance of the simultaneous inhibition of both the hCA IX and XII isoforms, at least in this context.

In summary, we have reported herein a new series of  $\alpha$ . diketocarboxylic acids and ethyl esters (1−6) potent at nanomolar level and selective toward the human CA isoforms IX and XII, highly expressed in tumor cells, and scarcely potent against the cytosolic CAs I and II. This unique inhibitory profile displayed by 1−6 is different from previously reported α,γ-diketoacids and esters (see compounds 1–3, 6a,b in ref 14), which showed unselective, low micromolar inhibition against hCA I, II, IX, and XII, with only one compound (6a in [ref](#page-4-0) 14) potent against hCA IX and XII at submicromolar  $level.<sup>1</sup>$ 

I[n g](#page-4-0)eneral, among the new  $\alpha$ , $\gamma$ -diketoacids the carboxylic acids 1−3 displayed similar or higher potency than the corr[esp](#page-4-0)onding ethyl esters 4−6 against the hCA IX and XII, and the phenyl- and  $\alpha$ -naphthyl-substituted compounds of the two series (1, 3, 4, and 6) behaved as dual hCA IX/XII inhibitors, while the  $\beta$ -naphthyl analogues (2 and 5) exhibited hCA IX-selective inhibition. Molecular modeling studies performed on 2 and 5 docked into hCA II and hCA IX highlighted their mechanism of action, due to coordination between the enzyme zinc ion and the carboxylate group of 2 or the carbonyl group of 5. In comparison with the binding mode within the hCA II active site, the better accommodation (more interactions and H-bonds) of the ligands into the hCA IX binding site accounted for their higher selectivity toward hCA IX. When tested in MG63 and HOS OS cells to determine their effects on cell viability, 2 was totally inactive likely due to its acidic nature (low cell permeability), while 5 and 6 displayed dose-dependent reduction of viability after 72 h treatment. As antiproliferative agents, the  $\alpha$ -naphthyl ester 6 was more potent than the  $\beta$ -naphthyl analogue 5, probably because 6 was a dual hCA IX/XII inhibitor, while 5 exerted selective hCA IX inhibition.

## ■ ASSOCIATED CONTENT

## **6** Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsmedchemlett.9b00023.

[Experimental proced](http://pubs.acs.org)ures for th[e synthesis and phys](http://pubs.acs.org/doi/abs/10.1021/acsmedchemlett.9b00023)[icoche](http://pubs.acs.org/doi/abs/10.1021/acsmedchemlett.9b00023)mical characterization of all new compounds; material and methods for CA inhibition assays, in silico studies, cell based assays (PDF)

## ■ AUTHOR INFORMATION

#### Corresponding Authors

\*(D.R.) Phone: +39 06 49913237. E-mail: dante.rotili@ uniroma1.it.

\*(A.M.) Phone: +39 06 49913391. E-mail: an[tonello.mai@](mailto:dante.rotili@uniroma1.it) [uniroma1.it.](mailto:dante.rotili@uniroma1.it)

 $*(C.T.S.)$  Phone: +39 055 4573729. E-mail: cl[audiu.supuran@](mailto:antonello.mai@uniroma1.it) [uni](mailto:antonello.mai@uniroma1.it)fi.it.

## ORCID<sup>®</sup>

[Paola](mailto:claudiu.supuran@unifi.it) Gratteri: 0000-0002-9137-2509

Antonello Mai: 0000-0001-9176-2382 Claudiu T. Supuran: 0000-0003-4262-0323

#### Author Contrib[utions](http://orcid.org/0000-0001-9176-2382)

A.N., D.R., A.Mai, F.P[., N.B., and C.T.S.](http://orcid.org/0000-0003-4262-0323) assisted manuscript preparation, data interpretation, and project overview. A.L., D.T., and D.R., compound synthesis. A.N., P.G., and C.T.S., in vitro profiling for hCAs inhibition and molecular modeling. F.P., A.M., and N.B., biological assays on OS cell lines. D.R., A.Mai, and C.T.S. conceived the study and followed the whole project. All authors have approved the final version of the manuscript.

#### Notes

The authors declare no competing financial interest.

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## ■ ABBREVIATIONS

AAZ, acetazolamide; CA, carbonic anhydrase; CAI, carbonic anhydrase inhibitors; OS, osteosarcoma

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