

Fabrication and characterization of a titanium dioxide (TiO₂) nanoparticles reinforced bio-nanocomposite containing *Miswak* (*Salvadora persica* L.) extract – the antimicrobial, thermo-physical and barrier properties

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Objective: The microbial, physico-chemical and optical corruptions threaten a variety of foods and drugs and consequently the human biological safety and its accessible resources. The human being's tendency towards bio-based materials and natural plant-extracts led to an increase in the usage of antimicrobial biocomposites based on medicinal herbs. *Miswak* (*Salvadora persica* L.) extract (SPE) has been proved effective for its antimicrobial and other biological activities. Therefore, in this study, titanium dioxide (TiO₂) nanoparticles (TONP) and SPE were applied to fabricate antimicrobial carboxymethyl cellulose (Na-CMC) based bio-nanocomposites which would simultaneously promote some thermo-physical and barrier properties.

Methods: CMC-neat film (C1), CMC/TONP-2% (C2) and CMC/TONP-2% with 150, 300 and 450 mg/mL SPE (SPE150, SPE30 and SPE450, respectively) were fabricated. The physical and mechanical properties; elemental mapping analysis (MAP), X-ray diffraction (XRD), scanning electron microscopy (SEM), thermal gravimetric analysis (TGA-DTG); fourier transform infrared (FTIR), energy-dispersive X-ray (EDX) and UV-vis spectroscopies were done to further validate the results.

Results: Addition of TONP (2%) improved the blocking of UV light at 280 nm while SPE-containing nanocomposites completely blocked it. FTIR, XRD and SEM confirmed the formation of homogeneous films and high miscibility of applied materials. TONP led to an increase in Young's modulus (YM) and stress at break (SB) while SPE decreased them and enhanced the elongation to break (EB) (flexibility) of the active nanocomposites. Compared to CMC-film, the thermo-gravimetric analysis (TGA-DTG) showed a higher thermal stability for CMC/TONP and CMC/TONP/SPE nanocomposites. The EDX spectroscopy and elemental mapping analysis (MAP) proved the existence and well-distributedness of Na, K, Cl, S, Ti, F and N elements in SPE-activated nanocomposites. The pure SPE and SPE-activated nanocomposites showed a favorable antimicrobial activity against both gram-positive (*Staphylococcus aureus*) and negative (*Escherichia coli*) bacteria.

Conclusion: The CMC-TiO₂-SPE nanocomposites were homogeneously produced. Combination of TiO₂ nanoparticles and dose-dependent SPE led to an improvement of thermal stability, and high potential in antimicrobial and UV-barrier properties. These results can generally highlight the role of the fabricated antimicrobial bio-nanocomposites as a based for different applications especially in food/drug packaging or coating.

Keywords: antimicrobial biomaterials, biopolymeric nanocomposite, carboxymethyl cellulose, *Miswak* (*Salvadora persica* L.), titanium dioxide, food packaging

Introduction

The quality of perishable stuffs like foods and drugs is always affected by ambient parameters such as light, oxygen, water vapor, temperature and micro-organisms. Packaging as a protective barrier can overcome their probable degenerative issues. The hazards of traditional petroleum-based materials for human and bio-eco-systems led to increasing focus on biodegradable materials.¹ Some of the natural and edible hydrocolloid biopolymers are recently used for film production including polysaccharides (such as cellulose and its derivatives, chitosan, alginate, guar gum, starch and alginate) and proteins (such as gelatin and collagen).² Among them, carbohydrates are better candidates because of their better film-forming, barrier and physical properties.³⁻⁵

As cellulose is the most accessible, biodegradable and nonhazardous natural polymer, its derivatives are considered good cases for application in the area of materials and packaging. Carboxymethyl cellulose (CMC), an ether-cellulose derivative,⁶ is well-studied as an accessible carbohydrate with good film-forming, mechanical and transparency properties.^{5,7,8} The nano-structured cellulose (CNF),^{3,9} chitin,¹⁰ copper oxide (CuO),¹¹ titanium dioxide (TiO₂),¹² silicon dioxide (SiO₂),¹³ silver (Ag)¹⁴ and some other nanoparticles¹⁵ are used for improving antimicrobial or thermo-physical properties of polymers.

The nontoxicity of TiO₂ nanoparticles (TONP) for human is presented by Food and Drug Organization (FDA). Furthermore, photo-catalysis and antimicrobial as well as UV-barrier properties of TONP made it the center of attention in different studies.¹⁶ As several recent studies showed, TONP in low content (1–5%) could improve the thermo-physical properties of biopolymeric films.^{17,18} The TONP together with montmorillonite (MMT) has enhanced the thermal, mechanical and UV-barrier properties of the starch nanocomposites in which the best content of TONP is obtained at 2% wt of polymer which also decreased the water vapor permeability (WVP) of the composites.¹⁸ Recently, different content of TONP (5, 10 and 15 wt%) has been applied into soluble soybean polysaccharide in which lower content of TONP (5%) led to improving the melting point, glass transition, tensile strength and WVP of the film.¹⁷

The human request is packaging systems which not only physically protect the food and drug stuffs but also enhance their shelf-life. These issues have led to creation of new packaging systems called active packaging. Since

the micro-organisms are one of the main hazards for human health and food/drug safety, the antimicrobial packaging has been considered as one of the most common types of active systems. These can deliver antimicrobial agents into the package content. On the other hand, medicinal herbs extracts and essential oils contain a wide range of bioactive compounds. They have become desirable cases for antimicrobial packaging and composites researches. Blending of the plant-based extracts or essential oils into the polymers in order to produce antimicrobial films is well studied with satisfactory results.

Miswak (*Salvadora persica* L.) (*S. persica*) with other common names known as “*Arak*”, “*Siwak*”, “*Siwaki*”, “*Sewak*”, “*Meswak*”, “*Miswaak*”, “*Misswak*”, “*Mswaki*”, “*Miswaki*”, a small shrub tree from the family of Salvadoraceae, is native to the Middle East and can be found in Africa, Saudi Arabia, India and south of Iran.¹⁹

Miswak root has been in use as a toothbrush (or chewing stick) for many years. World Health Organization (WHO) has also validated the chewing sticks as safe tools for oral hygiene.^{20,21}

Non-toxicity of *S. persica* extracts (SPEs) and their anti-inflammatory, antimicrobial and antioxidant activities have been previously reported.²¹⁻²⁵ It has several bioactive organic chemicals such as salvadorine, salvadorea, cyanates, saponins, alkaloids, amides, tannins, glycosides (eg, salvadoraside and salvadoside), terpenes, flavonoids and fatty acids (eg, oleic, linoleic and stearic acids) and a broad range of non-organic chemicals including potassium chloride (KCl), sodium chloride (NaCl), calcium (Ca), potassium (K), sodium (Na), phosphorus (P), fluoride (F), silica (Si), sulfur (S), magnesium (Mg), manganese (Mn) are reported from SPEs.^{19,26-30}

Benzyl isothiocyanate (BITC) is considered as the main antimicrobial agent in SPEs.³¹ SPEs with different solvents are well investigated against numerous micro-organisms. This confirms their potent antimicrobial activity against the yeasts (such as *Candida* spp. as one of the prevalent oral pathogen) and both gram negative and positive bacteria (such as *Staphylococcus* spp., *Streptococcus* spp., *Mycobacteria* spp., *Escherichia* spp.).³²⁻³⁴

Based on what was mentioned above, the production of the natural and bio-based antimicrobial composite materials especially for what human beings deal with on a regular basis (eg, food and drug stuffs) can be a promising way toward promoting human safety both environmentally and

biologically. On the other hand, to the best of our knowledge, there has not been a report for synchronous application of TONP and SPE into the CMC matrix. Therefore, 2% of TONP (as the nanofiller for mechanical and structural reinforcement) and different contents of SPE (as the antimicrobial agent) were blended into the CMC matrix in which the films were produced through casting method. Microstructure, elemental analysis, thermo-physical characterizations were studied along with the barrier and antimicrobial properties of the films.

Materials and methods

Materials

S. persica roots were collected from the Sistan and Baluchestan in the South-east of Iran. Sodium carboxymethyl cellulose (Na-CMC) (substituent degree of 0.7 and the molecular weight of 40,000 kDa) was purchased from Caragum Parsian Co (Tehran, Iran). Calcium sulfate (CaSO₄), magnesium nitrate hexahydrate [Mg(NO₃)₂·6H₂O], potassium chloride (KCl), sodium chloride (NaCl), glycerol, ethanol and microbial culture media of Muller Hinton Agar (MHA) and Tryptic Soy Broth (TSB) were purchased from Merck (Darmstadt, Germany). TiO₂ nanoparticles (TONP) in Rutile phase was provided from US Research Nanomaterials Inc. (Hauston, USA) with the average diameter of ~30 nm and the purity of 99.9% as mentioned in the producer data sheet.

Salvadora persica L. extract (SPE)

preparation

S. persica roots were washed and dried and after removing the bark it was suitably grinded into small particles. According to the method of Mohamed and Khan²¹ with some modifications, grinded root was blended into the ethanol–water solvent (50–50% v/v) where the ratio of root to solvent was 1:10. The suspension was stirred for 24 hrs at 25°C in a dark condition. The suspension was filtered through Whatman filter paper No. 1442, two times and the clarified liquid was then concentrated in a vacuum rotary evaporator (Heidolph, Germany) at ambient temperature. Finally, dark-brownish extract was gained from freeze-drying and stored at –70°C until application.

Film preparation

CMC film (control 1) preparation

2 g of CMC was added to 100 ml of water and was first stirred at 90°C and 900 rpm for 30 mins and then stirring

was continued for 12 hrs at room temperature. Then, glycerol (1 ml or 50 wt% of CMC) was slowly added to the solution (at 40°C and 700 rpm) and mixed for 10 mins. Just after that, the solution was cooled to ambient temperature and air bubbles were removed, the film solution was casted into 20×14 mm² polyethylene plastic plate. The solution was dried at 40°C for 20 hrs in an air-circulating condition. The dried films were taken from the container and put in the closed bag until the test time. This control film was coded to C1.

CMC/TiO₂-2% film (control 2) preparation

Same as C1, TiO₂-2% suspension was added to the solution prior to addition of glycerol. TiO₂-suspension was treated by sonication at ~40°C, 40 kHz and 100 w for 10 mins in water-bath ultrasound chamber (AS ONE, Japan) and 10 mins by probe ultrasound (Hielscher UP 200H, Germany) at ~40°C, 70% efficiency, 200 w and 24 kHz for 10 mins. Then, it was stirred at 1,000 rpm, 40°C for 10 mins before adding to C1 solution. This second control film was coded to C2 and considered as a base for SPE.

Active nanocomposite film (CMC/TiO₂-2%/SPE) preparation

The same method for C2 was applied in which 3 ml of 150, 300 and 450 mg/mL SPE [higher than minimal bactericidal concentration (MBC)] was blended into C2 solution (at 40°C) and stirred (10 mins at 700 rpm) before the addition of glycerol. These active films were coded to SPE150, SPE300 and SPE450, respectively.

All produced films with their codes and formulations are presented in Table 1.

Characterization

X-ray diffraction (XRD) analysis

XRD analysis in the angular region (2θ) of 15–80° was recorded by X'pert Pro diffractometer (Panalytical, Netherland) with Cu anode, at room temperature, step size of 0.026°, the rate of 1°/min, the wavelength of 0.154056 nm, 40 kV and 40 mA.

TONP crystallite size (D) was obtained through Debye-Scherrer equation (Eq. (1)).

$$D = \left[K\lambda / \left(\beta_{1/2} \cos \theta \right) \right] \quad (1)$$

where K, λ, β_{1/2} and θ are the Scherrer constant (0.94), X-ray wavelength (λ=0.154056 nm), full width at half maximum (FWHM) of TONP peak (by radian) and

Table 1 The formulation and the codes of all samples in 100 mL deionized water

Samples name	Components				
	CMC powder (g)	Plasticizer (v/w %, to CMC)	Nanofiller (TiO ₂) (wt % to CMC)	SPE concentration (mg/mL)	Casted solution (gr)
C1	2	50	0	0	50
C2	2	50	2	0	50
SPE150	2	50	2	150	50
SPE300	2	50	2	300	50
SPE450	2	50	2	450	50

Abbreviations: C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; SPE, *Salvadora persica* L. root extract; SPE150, SPE300 and SPE450: CMC/TONP-2% containing 150, 300 and 450 mg/ml of SPE, respectively; TONP, TiO₂ nanoparticles.

Bragg's angle, respectively. The $2\theta=27.48^\circ$ was the reference for calculation of FWHM of TiO₂ nanoparticles.

The crystallinity index (CI) of TONP and all samples were calculated by following Segal equation (Eq. (2)):

$$CI(\%) = [(I_{\max} - I_{\min})/I_{\max}] \times 100 \quad (2)$$

where I_{\max} and I_{\min} (or I_{am}) are the maximum and minimum intensity of the specific peak (for TONP: $2\theta=27.48^\circ$).

Fourier transform infrared (FTIR) spectroscopy

The KBr-pellet method and FTIR spectrophotometer (Bruker Tensor 27, Germany) were utilized to evaluate freeze-dried SPE and film samples from 4000 to 400 cm^{-1} .

Scanning electron microscopy (SEM), energy-dispersive X-ray (EDX) spectroscopy and elemental mapping analysis (MAP)

MIRA3 FESEM microscope (Tescan, Czech) was employed for SEM analysis. Surface and cross-section (cryo-fractured in liquid nitrogen) morphology of the films was scanned at room temperature, 5 kV and the scale of 50 μm .

The EDXs and MAPs were recorded for all samples to investigate the distribution quality of nanofiller (TiO₂ nanoparticles) and also dominant and specific elements of SPE based on SPE studies literature (Na, K, Cl, N, S, N and F).

Thickness

The digital micrometer with the accuracy of 0.001 mm (Insize 3109-25, China) was employed for evaluation of thickness in 6 random points for each film.

Moisture content (MC)

About 1 g of conditioned film was placed in 105°C up to reaching a constant weight. MC values were calculated by the following equation:

$$MC = (W_i - W_u/W_i) \times 100 \quad (3)$$

As the W_i and W_u are the initial and ultimate weight of the films, respectively.

Water vapor permeability (WVP)

ASTM E96-05³⁵ with some modifications was used to evaluate WVP. Round head vials (cylindrical) with 4.5×2.5 (cm^2) (height \times diameter) dimensions were considered as the permeation cells. 4 g of dry CaSO₄ (RH=0%) was poured into the cells and the conditioned round-shaped films specimens were placed under their head, respectively. Then, after recording the weight of the permeation cell (including both CaSO₄ and film specimen), they were inserted in a chamber with RH of 86% (containing KCl saturated solution) at $25 \pm 0.1^\circ\text{C}$ which causes driving force (ΔP) of 2725.34 Pa. Every 2 hrs, the weight-gain of the vials were recorded for 24 hrs and plotted as a function of time (weight vs time). Then, the slope of weight-time curves was calculated by linear regression. The water vapor transmission rate (WVTR) and WVP were, respectively, computed by

$$WVTR = (\text{g/h})/\text{film transfer area (m}^2) \quad (4)$$

$$WVP = (WVTR/\Delta P) \times X \quad (5)$$

where ΔP and x are driving force (2725.34 Pa) and film thickness (m), respectively.

ΔP is driving force which was gained from:

$$\Delta P = P(H_1 - H_2) \quad (6)$$

where P , H_1 and H_2 are saturated vapor pressure of water (3169 Pa) at test temperature (25°C), RH in desiccator (86% or 0.86), RH within the vial (0%).

Mechanical properties

Mechanical properties including Young's modulus (or elasticity modulus: E), stress at break (or ultimate tensile stress: UTS) and elongation (or strain) at breakpoint (EB)

were obtained from ASTM D882-10.³⁶ Three dumbbell-shaped of each film (8×0.5 cm) were prepared and conditioned. Then, the samples were tested in a tensile tester (Sanaf Co., Iran) with initial grip distance of 50 mm and the speed of 10 mm/min at ambient temperature.

Thermal gravimetric analysis (TGA)

The thermal analyzer model Linseis STA PT-1000 (Germany) was employed for TGA evaluation. The test was carried out from room temperature to 500°C, at the steady rate of 10°C/min under 50 cm³/min nitrogen flows. 15 mg of any sample was put into an aluminum pan in which empty aluminum pan was the reference. TGA and TGA-derivative (TGA and DTG) are both reported.

UV-vis barrier properties (opacity and light transmittance)

A UV-vis spectrophotometer model Shimadzu UV 2450 (Tokyo, Japan) was applied to read UV-vis transmission (200–900 nm) through the samples. The composites were inserted in the chamber and the air was selected as the reference. The opacity of the samples was obtained by

$$\text{Opacity} = \text{Abs}_{600}/x \quad (7)$$

where Abs₆₀₀ and x are an absorbance at 600 nm and film thickness (mm), respectively. The Lambert-Beer equation was used for conversion the transmittance (Tr) value to absorbance (Abs).

Antibacterial assay

Antibacterial activity of pure SPE

Escherichia coli (*E. coli*) ATCC® 25922™ and *Staphylococcus aureus* (*S. aureus*) ATCC® 29213™ were selected as the typical species of gram negative and positive bacteria, respectively. The minimal inhibition concentration (MIC) and minimal bactericidal concentration (MBC) of SPE were evaluated against both *S. aureus* and *E. coli* according to CLSI protocol by microdilution method in a 96-well micro-plate and using culture media of TSB.³⁷

Antibacterial assay of control and SPE-activated films

The disk shape of each sample with the diameter of 1 cm was placed on a MHA media which was previously whole-inoculated with targeted bacteria. Inhibition zone was considered where the bacteria had no growth around the film.

Statistical analysis and samples conditioning

The samples were conditioned before the physical tests at RH~52% and 25±0.2°C for 24 hrs. Completely randomized design with one-way analysis of variance (ANOVA) was applied by SPSS (version 16). Duncan's multiple range test at the 5% of probability ($p \leq 0.05$) was employed for mean comparisons and significant difference among samples. The physical evaluations were repeated three times.

Results and discussion

X-ray diffraction (XRD) analysis

The XRD analysis is plotted at Figure 1A. TiO₂ nanoparticles in Rutile phase showed some peaks at 2θ of 27.48, 36.1, 39.21, 41.22, 44.07, 54.22, 56.48, 62.35, 63.82, 68.75, 69.74° which is consistent with other studies.^{38–41} The crystallite size (D) and crystallinity index (CI) of TONP were estimated around 19.77 nm and ~85% by Debye-Scherrer and Segal equations.

In CMC-film pattern, no sharp peak was observed that indicated its complete amorphous structure. The amorphous nature of CMC films is previously reported by other studies.^{42,43} The blending of TONP into CMC and SPE (150, 300 and 450 mg/mL) into CMC-TiO₂ matrix caused no shift or appearance of a new peak. The CI of C1, C2, SPE150, SPE300 and SPE450 was 90.57%, 46.19%, 41.73%, 40.27% and 35.12%, respectively. Disappearance of all specific peaks of TONP and simultaneously the reduction of the CI in C2 clearly confirmed the well-distributedness of TiO₂ nanoparticles in CMC matrix. Moreover, the addition of SPE into the CMC-TONP matrix resulted in more decrease in CI of SPE-functionalized nanocomposites dose-dependently. The reduction in the CI of SPE-containing samples, especially in higher contents, is related to insertion of the extract between polymeric chains. This led to higher CMC inter-chains distance and chains sliding (plasticizing effect of SPE). Therefore, the lower tensile strength of active nanocomposites particularly in higher contents of SPE is probable. This plasticizing effect of SPE was in line with our recent research.³

XRD results indicated the well-miscibility of all applied materials, no agglomeration of TONP and SPE, homogenous structure of all nanocomposites and favorable efficiency of film preparation method.

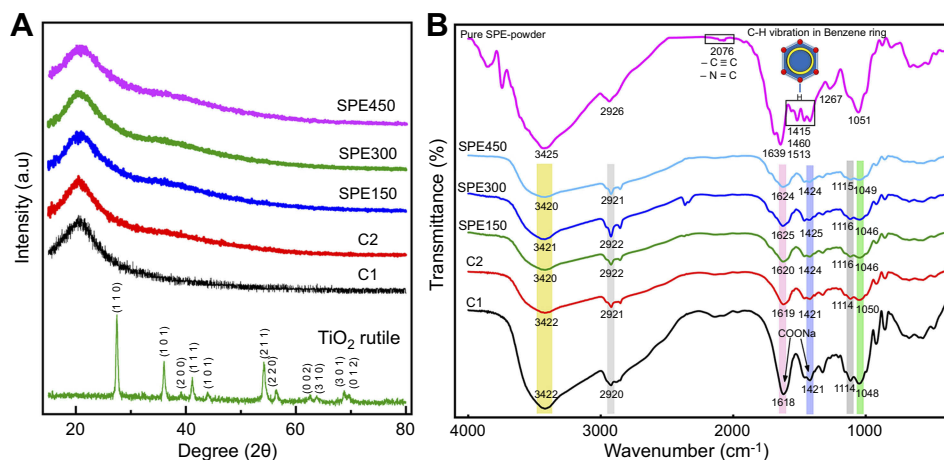


Figure 1 XRD patterns (A) of TiO₂ nanoparticles in Rutile phase, CMC film (C1), CMC/TiO₂-2% (C2) nanocomposite and activated nanocomposites with 150 (SPE150), 300 (SPE300) and 450 mg/mL (SPE450) Miswak extract and FTIR analysis (B) of pure SPE-powder and all mentioned samples.

Abbreviations: C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; FTIR, Fourier transform infrared spectroscopy; SPE, *Salvadora persica* L. root extract; SPE150, SPE300, and SPE450: CMC/TONP-2% containing 150, 300, and 450 mg/ml of SPE, respectively; TONP, TiO₂ nanoparticles; XRD, X-ray diffraction analysis.

Fourier transform infrared (FTIR) spectroscopy

FTIR patterns are presented in Figure 1B. For pure SPE-powder, isothiocyanate (R–N=C=S) and isocyanate (R–N=C=O) showed a weak peak at 2076 cm⁻¹ due to stretching N=C.⁴⁴

The peaks at 1415, 1460 and 1513 cm⁻¹ are related to C–H vibration in benzene ring skeleton⁴⁵ of various compounds in SPE such as lignans, salvadorine, benzyl amides, benzyl cyanates and other phenol-ring-containing bioactive compounds. The peak at 1415 cm⁻¹ can be also due to S=O stretching.^{29,30}

The wide peak around 3425 cm⁻¹ is due to hydrogen bonds, stretch vibration of N–H (in amide and amine compounds such as benzyl amides and three methylamine) and O–H stretch of phenols.

For CMC film (C1), some peaks at 3422, 2920, 1618, 1421, 1323 and 1048 cm⁻¹ were recorded that are attributed to O–H stretch and also inter/intra-molecular hydrogen bond, stretching C–H, stretching COO⁻ (–C=O in asymmetrical carboxyl), scissor-like –CH₂, OH bending and asymmetric stretching –C–O– in ether group (–CH–O–CH₂).

No shift was observed for CMC-TONP (C2) nanocomposite compared to CMC film (C1) which is due to low content of TONP and implied its well physical-insertion within the CMC chains.

In SEP-activated nanocomposites, the pure-SPE specific peaks were not observed and no change happened compared to CMC-TONP peaks. Therefore, no new chemical bond was

formed by the addition of SPE but indirectly confirmed its well physical insertion within CMC-TONP structure.

Scanning electron microscopy (SEM) and energy-dispersive X-ray (EDX) spectroscopy

The surface and cross-section micrograph of CMC film (Figure 2A-1 and A-2) was completely smooth and flat with no crack or pore. Adding the TONP to CMC matrix led to no significant change in both surface and cross-section that may be due to the TONP low content (2%) (Figure 2B-1 and B-2). The active nanocomposite with lowest (SPE150) and highest content of SPE (SPE450) showed a bit roughness at the surface while no considerable change or any crack or pore was observed in cross-sections in surfaces of both SPE150 and 450 (Figure 2C-1 and C-2; D-1 and D-2). According to SEM results, homogenous structure implied good miscibility of CMC, TONP and SPE. The EDX results showed Na, C and O elements for Na-CMC film (Figure 2A-3). As expected, Ti appeared in low amount for CMC-TONP nanocomposite (Figure 2B-3). For SPE150 (with lowest SPE content), elements of S, Cl and K appeared and also the amount of Na was increased (Figure 2C-3). This confirmed the existence of salts like NaCl and KCl and also sulfur-containing materials in SPE (like BITC). The EDX of SPE450 (Figure 2D-3) not only showed mentioned elements in SPE150 but also showed F in a low content (0.2%). The appearance of F element in SPE450 compared to SPE150 is probably because of its low content in SPE150 that EDX spectroscopy can detect in

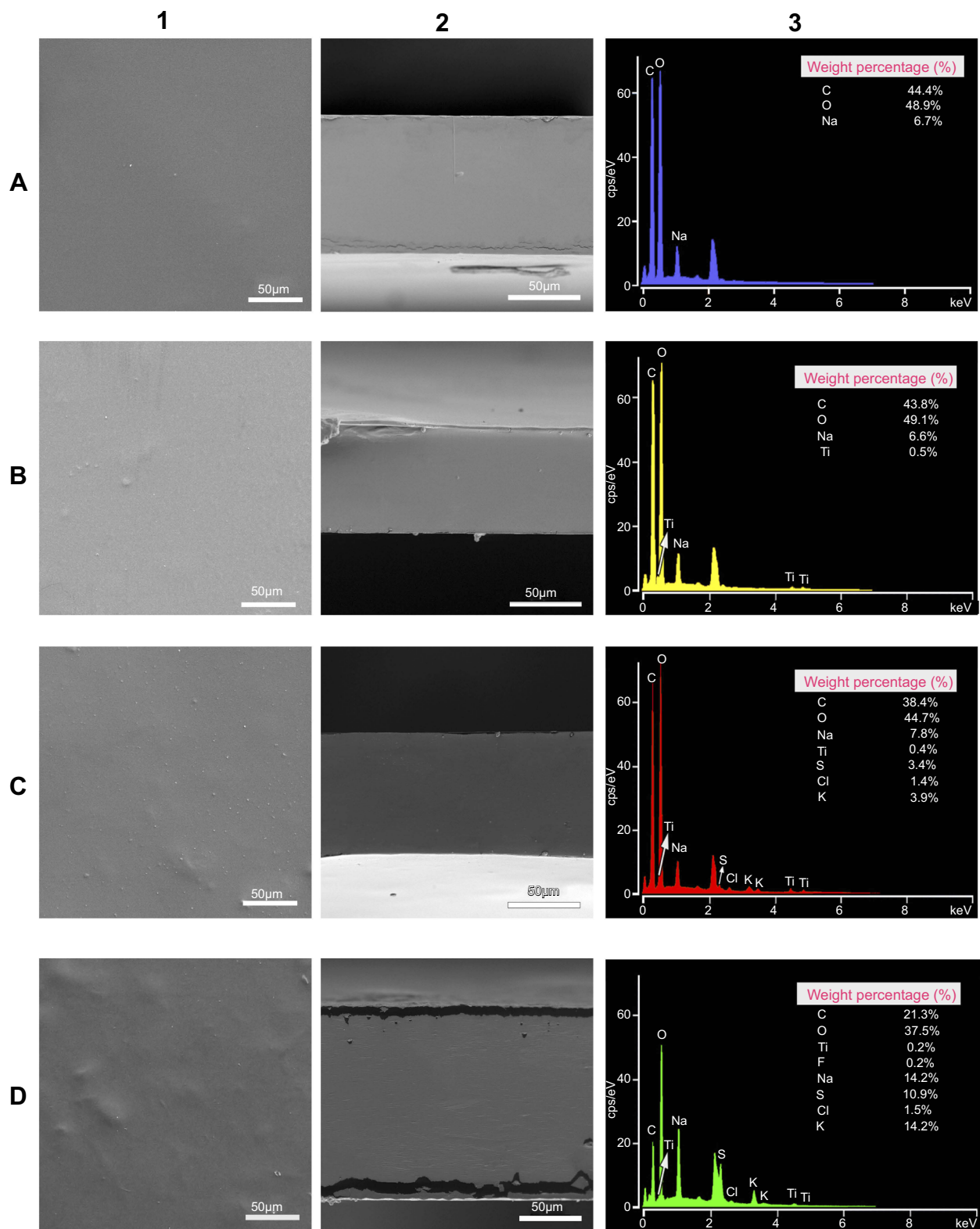


Figure 2 The SEM micrographs of surface (column 1) and cross-section (column 2); and EDX (column 3) for CMC film (**A**), CMC/TiO₂-2% nanocomposite (**B**) and active nanocomposites with lower (**C**) and highest (**D**) amounts of Miswak extract, ie 150 (SPE150) and 450 mg/mL (SPE450), respectively.

Abbreviations: C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; EDX, energy-dispersive X-ray spectroscopy; SEM, scanning electron microscopy; SPE, *Salvadora persica* L. root extract; SPE150, SPE300, and SPE450, CMC/TONP-2% containing 150, 300, and 450 mg/ml of SPE, respectively; TONP, TiO₂ nanoparticles.

a higher content than 150 mg/mL, where F atoms are more exposed to X-ray at the surface of the active nanocomposite. The weight percentage of Na, S, Cl and K in SPE450 was higher than SPE150 which is due to higher amount NaCl, KCl and Sulfur-containing compounds in a higher content of SPE. Dose-dependently, SPE caused to decrease in Ti percentage which probably is due to covering the TONP by SPE which decreased its exposure to X-ray bombardment. This interpretation is approved by a dose-dependent increase in amounts of specific elements in SPE (Na, S, Cl and K) for both SPE150 and SPE450.

Elemental mapping analysis (MAP)

The MAP analysis (Figure 3) of C, O and Na elements for C1; C, O, Na and Ti for C2; and C, O, Na, Ti, S, Cl, K, F and N for both SPE150 and SPE450 showed perfect homogenous distribution of applied material in all control and active films. It is clearly shown in Ti element MAP of all samples that TiO₂ nanoparticles have no agglomeration neither before addition (in C2) of SPE nor after it up to 450 mg/mL (in SPE150, SPE300 and SPE450).

MAP also confirmed the EDX results for dose-dependent increase in Na, F, S, Cl and K by increase in SPE amounts. In addition to detected elements in EDX of SPE-activated nanocomposites, the existence of nitrogen-containing compounds and their dose-dependent increase in SPE was shown in MAP distribution.

Totally, all SEM, EDX and MAP results implied that both TiO₂ nanoparticles at 2% and SPE up to 450 mg/mL could make a uniform and homogeneous structure. It implied that the employed method for film preparation in this study, ie the steps of addition of materials (TONP, SPE and glycerol, respectively); and the time and temperature of sonication, stirring and cast-drying were completely suitable for fabrication of produced films.

The most outstanding reason for the formation of homogeneous structure and well-miscibility of the applied materials is probably their hydrophilic nature which can prevent phase separation during preparation and drying.

Moisture content (MC)

Figure 4A shows MC content. The addition of TONP led to a slight decrease in MC from 31.37% for C1 film to 30.25% for C2 nanocomposite. A similar report is previously reported for the starch-kefir matrix.⁴⁶ Addition of SPE in all content showed lower MC compared to C2 which is due to the higher total solid mass of SEP-

activated nanocomposites. It is mainly related to some organic, minerals and metals that EDX and MAP analysis confirmed their existence within the structure of SPE-containing nanocomposites. Therefore, good entrapment of these components within the structure led to the higher ejection of water molecules from the polymeric matrix during the drying time of the casted solutions. Other studies reported similar results for tea extract,⁴⁷ and mineral and vitamin⁴⁸ on chitosan-based film. Our previous study has also showed such behavior in MC test after addition of SPE.³

Water vapor permeability (WVP)

The WVP is one of the important parameters for moisture-stability and shelf life of the stuff within the nanocomposites or films. The WVP is dependent on different parameters such as the nature of the polymeric matrix, additives, activation agent, water content and the structure and morphology of the films and nanocomposites.⁴⁹ Figure 4B presents the WVP results. Adding TONP to CMC led to the non-significant decrease in WVP. The existence of water-insoluble TONP within the nanocomposite structure as a nanofiller and blocking the paths may form longer routs for transferring of moisture molecules and lower WVP consequently. Deka and Maji⁵⁰ obtained similar results for blending TONP in wood-based composites. They claimed that the early hydrophilic interactions between TONP and matrix would diminish water permeation from the cross-paths of the matrix. Similar to our research, Oleyaei et al.⁵¹ also reported the decrease in WVP by adding TONP (up to 2%) into the potato starch nanocomposites.

The permeability of biopolymeric films will be changed by the addition of plant extracts. Depending on the hydrophilic/hydrophobic nature of the applied biopolymer and extract components, the WVP will differ.⁴⁹ In this study, the addition of the lowest concentration of SPE (150 mg/mL) diminished the WVP while higher concentrations (300 and 450 mg/mL) increased it. Some previous studies reported the WVP decrease by blending mango kernel extract into soy protein isolate,⁵² green tea extract into chitosan⁵³ and rosemary extract into κ-carrageenan⁵⁴ nanocomposites. The main reason for improved moisture barrier properties of extract-containing films is the formation of hydrogen bonds and hydrophobic interactions between biopolymeric matrix and extract or essential oil components. In contrast, it is reported that the WVP is increased after adding grape

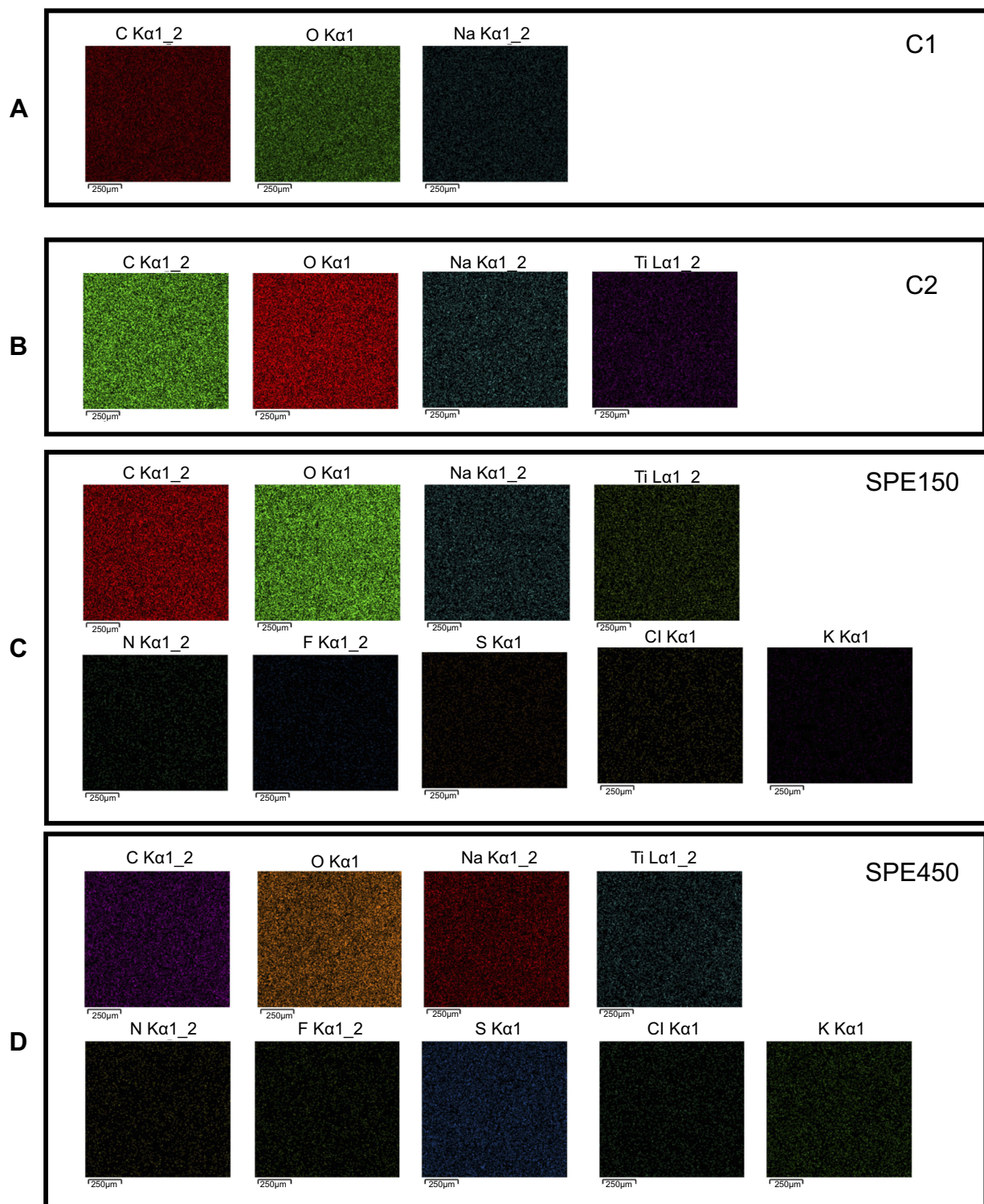


Figure 3 The elemental mapping analysis (MAP) for CMC film (A), CMC/TiO₂-2% (B) nanocomposite and active nanocomposites with 150 (SPE150) (C) and 450 mg/mL (SPE450) (D) Miswak extract. The distribution of any element is separately presented by a single photo.

Abbreviations: C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; MAP, elemental mapping analysis; SPE, *Salvadora persica* L. root extract; SPE150, SPE300, and SPE450, CMC/TONP-2% containing 150, 300, and 450 mg/ml of SPE, respectively; TONP, TiO₂ nanoparticles.

fruit seed extract into carrageenan-based films.⁵⁵ Nouri and Nafchi⁵⁶ also reported the WVP enhancement by addition of betel leaf extract. The WVP decrease in

SPE150 can be attributed to insertion of SPE components in free volume between CMC polymeric chains which act as a filler. For SPE300 and SPE450, as it is in accordance

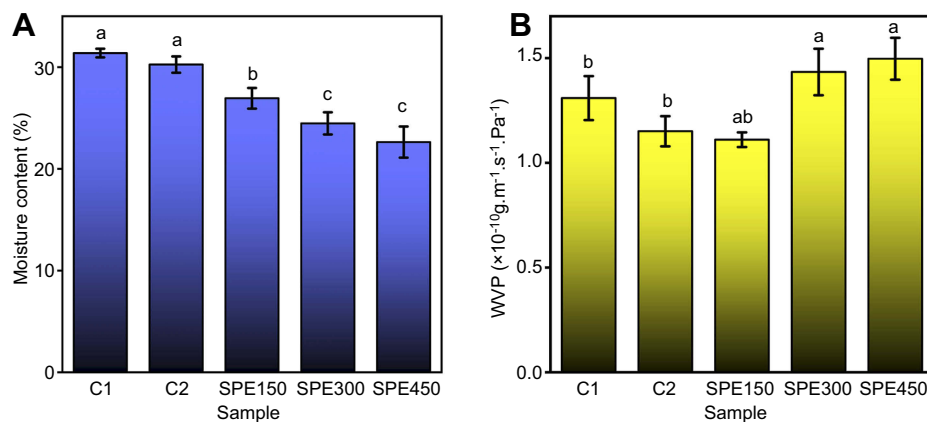


Figure 4 The moisture content (MC) (A) and water vapor permeability (WVP) (B) for CMC (C1) and CMC-TiO₂-2% (C2) control groups and all activated nanocomposites by 150 (SPE150), 300 (SPE300) and 450 mg/mL (SPE450) of *Miswak* extract. Various small letters present significant difference between mean values ($P < 0.05$).

Abbreviations: C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; MC, moisture content; SPE, *Salvadora persica* L. root extract; SPE150, SPE300, and SPE450, CMC/TONP-2% containing 150, 300, and 450 mg/ml of SPE, respectively; TONP, TiO₂ nanoparticles; WVP, water vapor permeability.

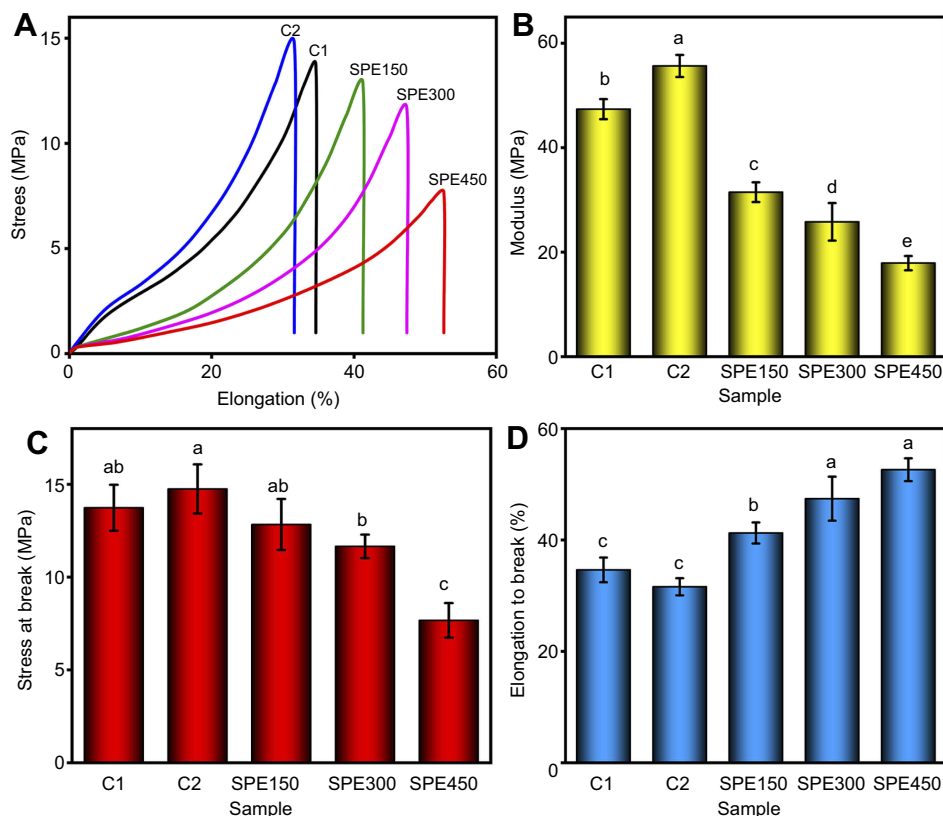


Figure 5 The mechanical behavior patterns (A), modulus (B), stress at break (C) and elongation to break (D) of CMC film (C1), CMC/TiO₂-2% (C2) nanocomposite and all active nanocomposites with 150 (SPE150), 300 (SPE300) and 450 mg/mL (SPE450) *Miswak* extract. Various small letters present significant difference between mean values ($P < 0.05$).

Abbreviations: C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; SPE, *Salvadora persica* L. root extract; SPE150, SPE300, and SPE450, CMC/TONP-2% containing 150, 300, and 450 mg/ml of SPE, respectively; TONP, TiO₂ nanoparticles.

with MC results, dominating the hydrophilic and hygroscopic nature of SPE could increase water vapor molecules adsorption and absorption. On the other hand, the higher gap between CMC polymeric chains (higher space

for passing small molecules) and therefore lower polymeric chains integrity could facilitate the penetration of water vapor molecules across the SPE-nanocomposites structure.

Mechanical behaviors and properties

Being one of the important parameters for packaging materials, mechanical properties are always center of attention in marketing and processing aspects of products. Figure 5A–D indicates mechanical behaviors and parameters. In C2 nanocomposite, mechanical behavior was the same as C1. However, significant increase in STB (17.46%, from 47.38 to 55.65 M.Pa) and YM (7.04%, from 47.38 to 55.65 M.Pa) and a non-significant decrease in EB (8.76%) was observed. This can be related to the interaction between TiO₂ positive charges and the negative charge of carboxylic groups of CMC matrix. Similarly, previous studies reported the increase in TSB and decrease in EB by adding TONP into chitosan,⁵⁷ potato starch¹⁸ and soluble soybean polysaccharide matrix.⁵⁸ Dependent on the plant extract components, the effects of plant extracts are variable. Generally, the polyphenol-rich extract decreases the mechanical properties of the films and nanocomposites.⁴⁹ The blending of SPE led to significant reduction in UTS and YM and enhancement of EB, dose-dependently. These results could be related to insertion of SPE components among CMC chains and broadening the distance between adjoining polymeric chains (and decrease in total polymeric chains entanglements) which finally led to lower UTS and YM and higher EB due to higher chains sliding and mobility. These interpretations are in accordance with XRD results in part 3. 1, Figure 1A, where the intensity of the broad peak around 20–20° was diminished (ie the reduction in crystallinity

index of SPE-activated nanocomposites). In some other studies, similar results were observed for blending *mutra* extract into methyl cellulose⁵⁹ and grape fruit seed extract into carrageenan films.⁵⁵

Thermal gravimetric analysis (TGA-DTG)

TGA (Figure 6A) DTG (Figure 6B) curves show the thermal stability behaviors of control and active films. The proximate weight loss (Δw) and maximum temperature of degradation (T_{max}) were obtained from TGA and DTG curves, respectively. Three steps of weight loss (Δw_1 , Δw_2 and Δw_3 , respectively) were observed for C1 and C2 while the second step was overlapped with third and just two steps were recorded for SPE150, SPE300 and SPE450.

The Δw_1 happened at 45–110°C, the Δw_2 at 110–230°C and Δw_3 at 230–320°C. The Δw_1 is attributed to free water evaporation,⁶⁰ Δw_2 to organic ingredients of SPE^{59,61} and free glycerol⁶² evaporation and the Δw_3 to polymer branches degradation⁵ and the rest of glycerol and SPE volatile components.

The Δw_1 of 5.2%, 5%, 4.8%, 4.6% and 4.2% was estimated for C1, C2, SPE150, SPE300 and SPE450, respectively. The T_{max1} for C1 was 71.4°C which was changed to 72.6°C by blending TONP (for C2). The blending of SPE also increased the T_{max1} to 78.3°C, 82.8°C and 87.3°C for SPE150, SPE300 and SPE450, respectively.

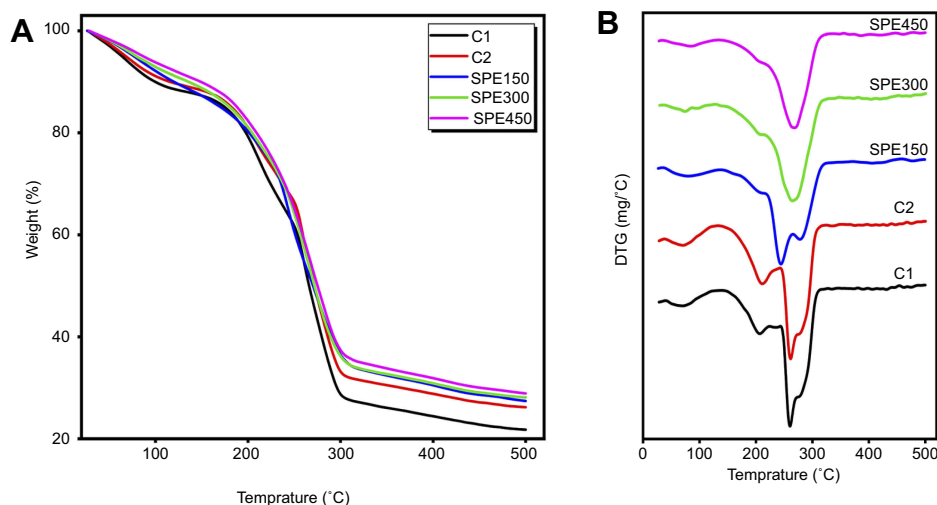


Figure 6 The TGA (A) DTG (B) curves of CMC film (C1), CMC/TiO₂-2% (C2) nanocomposite and active nanocomposites with 150 (SPE150), 300 (SPE300) and 450 mg/mL (SPE450) of *Miswak* extract.

Abbreviations: C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; DTG, TGA-derivative or the derivative of TGA; SPE, *Salvadora persica* L. root extract; SPE150, SPE300, and SPE450, CMC/TONP-2% containing 150, 300, and 450 mg/ml of SPE, respectively; TGA, thermal gravimetric analysis; TONP, TiO₂ nanoparticles.

Table 2 The UV-vis transmittance and opacity of CMC (C1), CMC/TiO₂-2% (C2) and activated nanocomposites by 150, 300 and 450 mg/mL *S. persica* extract

Sample	Tr ₂₈₀ (%)	Tr ₆₀₀ (%)	Opacity
C1	64.34±0.25 ^a	92.43±0.94 ^a	0.43±0.03 ^d
C2	50.64±0.17 ^b	80.04±0.7 ^b	0.94±0.04 ^c
SPE150	0.00±0 ^c	62.04±0.21 ^b	1.86±0.09 ^c
SPE300	0.00±0 ^c	39.69±0.15 ^c	3.54±0.26 ^b
SPE450	0.00±0 ^c	30.86±0.15 ^d	4.15±0.11 ^a

Abbreviations: C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; SPE, *Salvadora persica* L. root extract; SPE150, SPE300, and SPE450, CMC/TONP-2% containing 150, 300, and 450 mg/ml of SPE, respectively; TONP, TiO₂ nanoparticles; Tr₂₈₀, light transmittance at 280 nm; Tr₆₀₀, light transmittance at 600 nm; UV-vis, ultraviolet-visible light.

The Δw_2 and Δw_3 of 21.1% and 43.8% for C1 were decreased to 20.4% and 42.7% for C2, respectively. The T_{max2} and T_{max3} were enhanced from 206.5°C and 260.4°C for C1 to 211.4°C and 262°C for C2, respectively.

The Δw_2 of 61.4%, 61.1% and 61.4% and T_{max2} of 246.2°C, 266.6°C and 271.2°C was recorded for SPE150, SPE300 and SPE450, respectively.

Compared to C1, the O-H interaction between polymer and TONP¹⁸ and the impurity of water and glycerol by adding TONP and SPE is the most probable reason for higher T_{max1} & T_{max2} in C2 and SPE-activated samples. It means that the TONP and SPE enhanced the evaporation temperatures of entrapped water and glycerol.

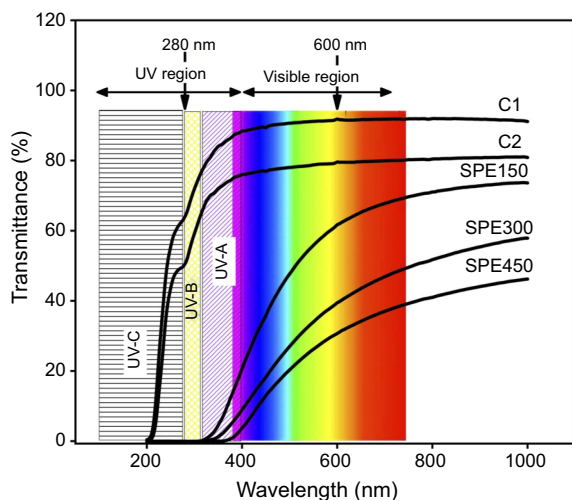


Figure 7 The UV-vis wavelengths transmission curves through the CMC film (C1), CMC/TiO₂-2% (C2) nanocomposite and all active nanocomposites with 150 (SPE150), 300 (SPE300) and 450 mg/mL (SPE450) of *Miswak* extract.

Abbreviations: C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; SPE, *Salvadora persica* L. root extract; SPE150, SPE300, and SPE450, CMC/TONP-2% containing 150, 300, and 450 mg/ml of SPE, respectively; TONP, TiO₂ nanoparticles; UV-vis: ultraviolet-visible light.

Similar to the previous study by Ahmadi et al.³ The lower Δw_2 and Δw_3 For C2 and lower Δw_2 for SPE-containing samples were likely due to the increase in total solid mass of nanocomposites by adding 2% TONP and different SPE content in comparison to C1, respectively.

The weight loss at higher temperatures (up to 500°C) gave no certain peak which means that there is no separate or major phase in composite ingredients with a specific TGA behavior. In the other words, slight weight loss from 320°C to 500°C belongs to the remaining fraction of volatile compounds of SPE and not-degraded CMC side-branches.

The total remaining weight percentage at 500°C (R_{500}) was enhanced by addition of TONP and dose-dependently by blending the SPE. R_{500} was 21.8%, 26.1%, 27.4%, 28.1% and 28.9% for C1, C2, SPE150, SPE300 and SPE450, respectively. The increase in total solid mass and the higher degradation temperature of TiO₂ led to higher heat stability of C2 and SPE-activated nanocomposite. Higher heat stability of SPE-activated nanocomposites compared to C1 and C2 is most probably due to the existence of minerals which was approved by EDX and MAP analysis as well as the hinted inorganic compounds in the literature of *Miswak*. These elements may create an ionic network around CMC polymeric chains in earlier viscose SPE nanocomposite solution and acted as a heat-insulation layer for inner fractions.

Uv-vis barrier properties (opacity and light transmittance)

The UV-vis transmittance patterns of all samples are presented in Figure 7. Opacity and transmittance at the wavelength of 280 nm (Tr₂₈₀) and 600 nm (Tr₆₀₀) are also mentioned in Table 2. Adding 2% of TONP led to a significant decrease in transmittance of UV (Tr₂₈₀) and visible light (Tr₆₀₀) while the opacity was increased. Decreasing the UV/visible light transmittance by TONP is reported for whey protein isolate,⁶³ starch,¹⁸ starch/CMC⁵¹ and kefiran-whey protein⁶⁴ bio-nanocomposites. The dramatic decrease can happen by higher amounts of TONP⁶³ but the intrinsic nature of the polymer, the crystallinity of the final film and co-inside existence of other nanoparticles can lead to different results.

Various small letters indicate significant difference of mean values ($P < 0.05$).

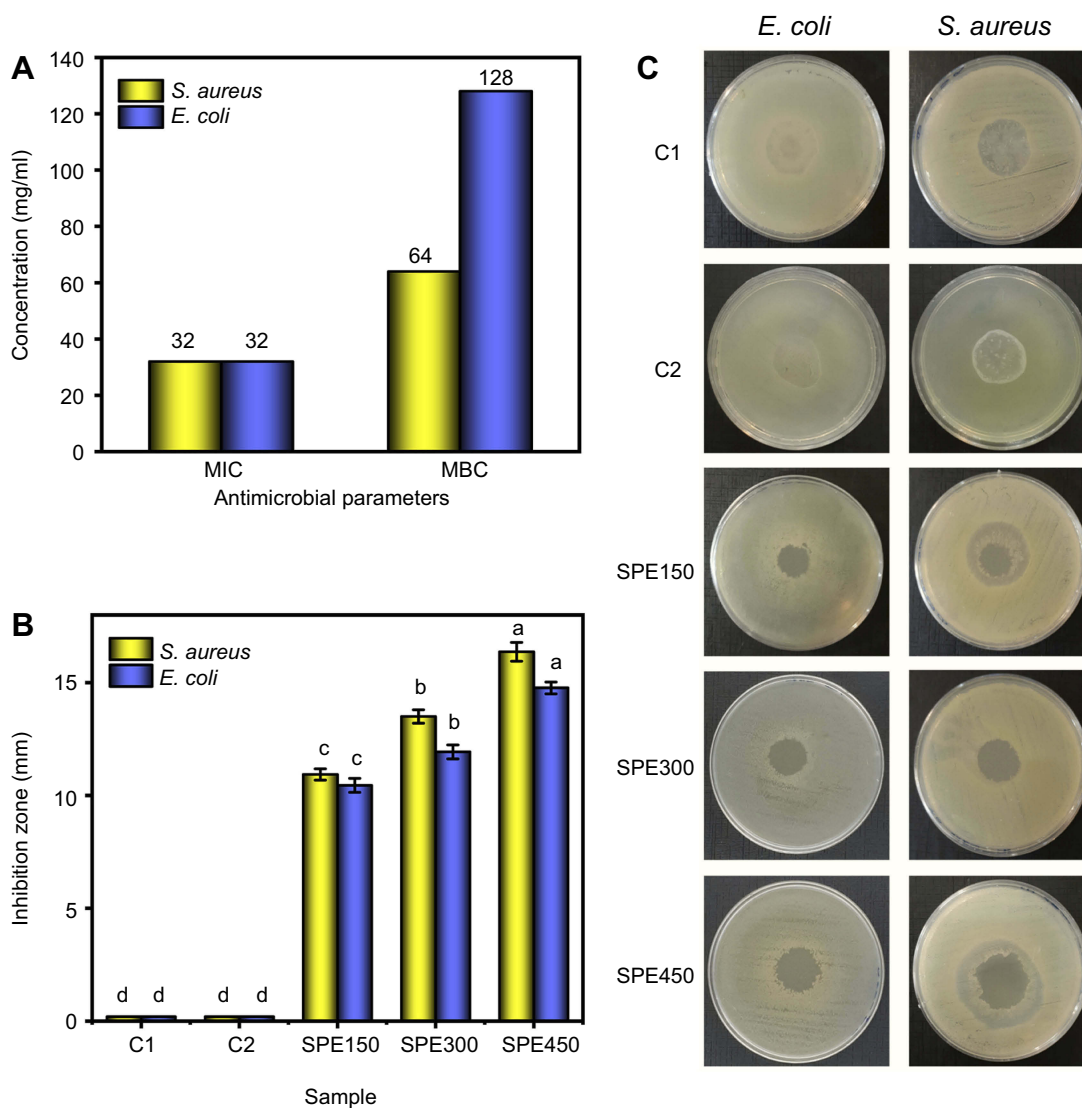


Figure 8 The MIC and MBC test for pure-SPE (A); recorded inhibition zone from disk diffusion test (B); and digital photos of disk diffusion test (C) for all samples. **Abbreviations:** C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC, carboxymethyl cellulose; *E. coli*, Escherichia coli bacterium; MBC, minimum bactericidal concentration; MIC, minimum inhibitory concentration; *S. aureus*, Staphylococcus aureus bacterium; SPE, *Salvadora persica* L. root extract; SPE150, SPE300, and SPE450: CMC/TONP-2% containing 150, 300, and 450 mg/ml of SPE, respectively; TONP, TiO₂ nanoparticles.

A dose-dependent decrease in Tr_{600} and increase in opacity was observed for SEP-containing nanocomposites which were significant ($P < 0.05$) in SPE300 and SPE450. It means that the transparency and opacity of SPE150 is still near to C2 control film (CMC/TiO₂-2%). The opacity enhancement in extract-activated films is previously reported for borage seed, rosemary and oregano extracts on gelatin films⁶⁵ and green tea extract (GTE) on chitosan film.⁵³ The existence of polyphenols are reported as the main reason for higher opacity.

Incredibly, all three SPE-containing nanocomposites completely blocked UV-light at 280 nm ($Tr_{280} = 0\%$).

This high UV-barrier potential is very important for the shelf life of UV-sensitive products such as foods and drugs. It also implies that SPE can play a promising role in UV-blocker nanocomposites or coatings.

Antimicrobial activity of pure SPE and SPE-activated nanocomposites

Pure SPE showed antibacterial activity against both *S. aureus* and *E. coli* as the MIC for both bacteria was 32 mg/mL while the MBC was obtained at 64 mg/mL for *S. aureus* and at 128 mg/mL for *E. coli* (Figure 8A and B). It reveals that *E. coli* is more resistant to current SPE. As expected, the disk diffusion test for TiO₂-containing control group nanocomposite

(C2) did not show any antimicrobial effect on both bacteria. This could be due to its low content (2%) and the fact that it is not performing UV-functionalization on TONP (since it was not considered here as an antimicrobial agent). However, SPE-containing nanocomposites (CMC-TiO₂-SPE) showed a dose-dependent increase in antibacterial activity (Figure 8C). The disc diffusion test for SPE-activated nanocomposites showed higher inhibition zone against *S. aureus* at the same SPE content. There are some contradictory results which showed that gram negative bacteria were more sensitive to *Miswak* extract.^{25,31} On the other hand, similar to our results, some studies have reported higher sensitivity of gram positives (such as *S. aureus*) to SPE.³⁴ The contradictory reports may be due to various extraction methods (such as temperature, solvent, time, etc.), the region and climate of the places where *Miswak* roots were gathered affects the bioactive components and antimicrobial activities. In lots of studies, BITC is reported as the main antimicrobial component in different SPE. The known mechanism for antibacterial activity of BITC is attributed to its electro-lipophilic properties in which the interaction between BITC and bacterium membrane led to the dysfunction of the redox system. Finally, the bacterial membrane disintegration and potential changes are the main reasons for SPE antibacterial activity.³¹ As the presence of minerals and salts was proved by EXD and MAP analysis for this study, these components may also have changed the osmosis potential and the optimal parameters for bacterial growth. These findings suggest that pure SPE and SPE-containing nanocomposites have the potential to be considered in antimicrobial biomaterials.

Conclusion

Due to good biological properties of *Miswak* (*Salvadora persica* L.) extract (SPE) especially for antimicrobial purposes and on the other hand, the application of TiO₂ nanoparticles (TONP) as a nanofiller for biopolymeric nanocomposites, the active bio-nanocomposites using TONP and SPE were properly fabricated. The UV-barrier properties showed that TONP diminished UV-transmittance while simultaneous presence of TONP and SPE in carboxymethyl cellulose (CMC) matrix completely blocked it in all applied concentrations of SPE. Elemental mapping analysis (MAP) and energy-dispersive X-ray (EDX) spectroscopy showed the existence of Na, Cl, K, S, N, Cl, Ti and F in SPE-containing nanocomposite. FTIR, MAP, SEM and XRD showed a favorable nanocomposite structure where there was no pore, crack and agglomeration or accumulation of all applied materials.

Both TONP and SPE improved the heat stability of CMC through lowering mass loss and increase in the maximum temperature of degradation. Pure SPE and SPE-activated nanocomposites showed a good antimicrobial activity against both gram positive (*S. aureus*) and negative (*E. coli*) bacteria. Totally, our results show that SPE and TONP have good miscibility within CMC matrix that highlights their usage in different active antimicrobial biomaterials particularly in the area of packaging applications of product with low moisture content.

Abbreviations list

BITC, benzyl isothiocyanate; C1, CMC film; C2, CMC/TONP-2% nanocomposite; CMC or Na-CMC, carboxymethyl cellulose; EB, elongation to break (flexibility); MBC, minimum bactericidal concentration; MIC, minimum inhibitory concentration; SB, stress at break point; SPE, *Salvadora persica* L. root extract; SPE150, SPE300 and SPE450, CMC/TONP-2% containing 150, 300 and 450 mg/mL of SPE, respectively; TONP, TiO₂ nanoparticles; YM, Young's modulus.

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Disclosure

The authors report no conflicts of interest in this work.

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