

Thickness-Dependent Perovskite Octahedral Distortions at Heterointerfaces

Jennifer Fowlie,^{*,†,‡} Céline Lichtensteiger,[†] Marta Gibert,^{†,||} Hugo Meley,[†] Philip Willmott,^{‡,§} and Jean-Marc Triscone[†]

[†]Department of Quantum Matter Physics, University of Geneva, 24 Quai Ernest-Ansermet, 1211 Geneva, Switzerland

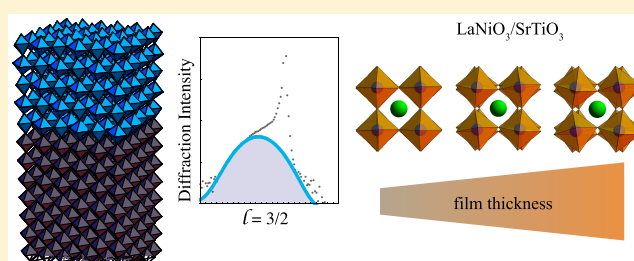
[‡]Swiss Light Source, Paul Scherrer Institut, 5232 Villigen, Switzerland

[§]Physik Institut, University of Zürich, 190 Winterthurerstrasse, 8057 Zürich, Switzerland

Supporting Information

ABSTRACT: In this study, we analyze how the octahedral tilts and rotations of thin films of LaNiO_3 and LaAlO_3 grown on different substrates, determined using synchrotron X-ray diffraction-measured half-integer Bragg peaks, depend upon the total film thickness. We find a striking difference between films grown on SrTiO_3 and LaAlO_3 substrates which appears to stem not only from the difference in epitaxial strain state but also from the level of continuity at the heterointerface. In particular, the chemically and structurally discontinuous $\text{LaNiO}_3/\text{SrTiO}_3$ and $\text{LaAlO}_3/\text{SrTiO}_3$ interfaces cause a large variation in the octahedral network as a function of film thickness whereas the rather continuous $\text{LaNiO}_3/\text{LaAlO}_3$ interface seems to allow from just a few unit cells the formation of a stable octahedral pattern corresponding to that expected only given the applied biaxial strain.

KEYWORDS: Perovskite, nickelate, thin film, heterostructure, X-ray diffraction, structure



Owing to the highly directional and electronically localized nature of the d orbital wave functions, transition metal oxides (TMOs) often exhibit a strong coupling between the lattice and other degrees of freedom.¹

Many TMOs stabilize in the perovskite structure with the formula unit ABO_3 .² The lattice can be visualized as a network of corner-sharing O_6 octahedra with the B cation situated in the center of each. One octahedron represents the perovskite pseudocubic (pc) unit cell (u.c.) and can be seen as the functional unit. In almost all perovskites, however, the primitive unit cell is larger than this as the lattice is distorted, most commonly through antiferrodistortive rotations and tilts of the O_6 units, doubling the unit cell in real space. These often subtle rotations of oxygen octahedra carry direct implications for the wider physics of the material affecting, for example, magnetic exchange and electronic hopping.^{3–7} In the family of rare earth nickelates, RNiO_3 , for instance, the Ni—O—Ni bond angle determines the temperature of their famous metal-to-insulator transition.^{8,9}

From the wide range of available chemistries given by the distortion of the octahedral functional unit, a correspondingly wide range of materials functionalities emerges. One main challenge is in the development of new approaches by which the octahedral network can be engineered.

In bulk systems, octahedral engineering is largely understood within an empirical framework, for example, via the tolerance factor.¹⁰ Then, for a given ABO_3 compound further tuning the

octahedral lattice can be achieved by chemical or isotope substitution,^{8,11,12} nonhomogenous deformations such as bending,¹³ nonequilibrium excitation with light,¹⁴ or by pressure effects, either hydrostatic¹⁵ or internal.¹⁶

In epitaxial perovskite oxide heterostructures, direct and static lattice tuning can be achieved without relying on external stimuli or chemistry through heterostructure effects both brought about by the substrate and by the heterointerfaces with the other functional layers.^{17,18}

The primary mechanism of heterostructure engineering of the lattice is through epitaxial strain imposed by the substrate. Figure 1 illustrates schematically the expected effect of the biaxial strain when considering either pure bond length modifications (panel a) or pure rotations of octahedra (panel b). In reality, a combination of the two responses would most likely be expected. The effect of heterostructuring, however, goes beyond simple biaxial strain as, very often, the substrate itself has a characteristic octahedral network, which can affect the film through the interfacial coupling of distortions. This added texturing should also bring a thickness-dependent modulation to the film structure, even with the retention of complete biaxial strain that is usually observed up to much higher thickness limits. As many functional properties, often

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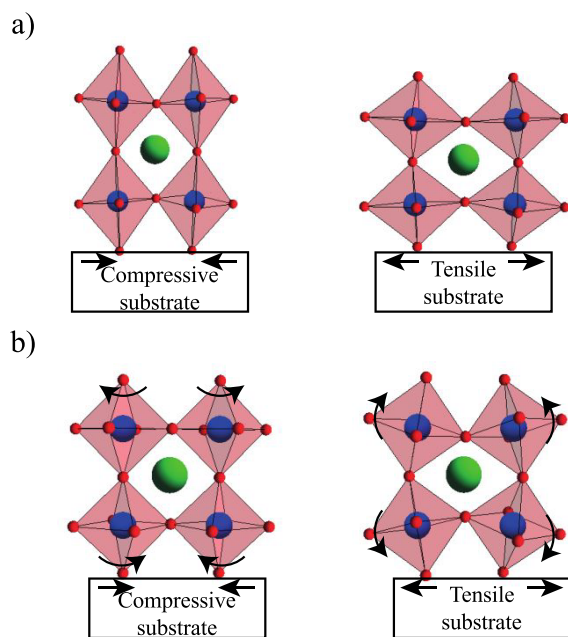


Figure 1. A simplified schematic of the response to biaxial strain through (a) pure bond length modifications and (b) pure oxygen octahedral rotations.

unavailable in bulk, stem from heterointerfaces, the study of the lattice structure toward the ultrathin limit is highly appealing from the perspective of tailor-made nanoscale materials and may reveal new engineering routes.

Today, an entire field of research is dedicated to measuring collective lattice distortions. Structural modulations have often been observed at heterointerfaces to vary on a length scale of 3–8 pseudocubic unit cells.^{19–25} Our previous work on LaNiO₃ thin films, utilizing TEM and ab initio calculation, found that the film structure has a tendency to adopt the structure of the LaAlO₃ substrate close to the interface with strong distortions on the film surface.²³ This coexistence of different structures was shown to be the likely origin for a thickness-dependent conductivity enhancement in those ultrathin films, illustrating how important the lattice distortion can be for the physical properties of TMOs.

One of the primary challenges in further developing tools and approaches to measure distortions of the octahedral functional unit itself lies in the fact that they usually stem only from displacements of light oxygen atoms which, when combined with the fact that heterostructuring tends to involve low sample volumes, make their detection rather arduous. Here we describe an approach to quantify rotational distortions of the perovskite octahedral functional unit by synchrotron X-ray diffraction. Our results show that this technique is well-suited to this type of problem even when the film thickness is as low as 2 nm. Resulting from our analysis, a striking contrast in behavior is observed when the heterointerface is quite structurally and chemically continuous (i.e., between very similar materials) versus when it is rather discontinuous. In the latter case, a thickness-dependence of the collective distortion is uncovered and a potential control knob for octahedral distortion magnitude is revealed.

The approach employed here leverages the fact that distortions of the functional unit double the real space unit cell and, therefore, halve the reciprocal space periodicity,

producing extra peaks at half-integer points when probed by X-ray diffraction (XRD). This method of obtaining the precise pattern of tilts and rotations of the octahedra was introduced by Glazer along with a convenient shorthand to describe this collective distortion.²⁶ First, a definition of the functional unit distortion is that α and β are the rotation magnitudes of the octahedron around the a - and b -axes, respectively, often referred to as “tilts”, while γ represents the rotation magnitude around the c -axis. A schematic definition can be found in the [Supporting Information](#). Glazer notation uses three labels, each of one letter and one superscript symbol, to describe the rotations around the three respective crystallographic axes. The superscript “0” denotes no rotation around that axis, “–” denotes an out-of-phase rotation (adjacent octahedra rotate in the opposite direction), and “+” denotes an in-phase rotation (all octahedra rotate in the same direction). If more than one label has the same letter, then it means that the rotations around those axes are equal in magnitude. For example, the common Glazer system a^-a^- represents out-of-phase-rotations around the three axes, all of the same magnitude, that is, $\alpha = \beta = \gamma$. By recording which half-integer Bragg peaks (HIBP) are present and which are absent, the sample symmetry is readily reasoned.

As a further step, the magnitudes of the tilts and rotations can be obtained by more detailed analysis of the relative intensities of the HIBP in XRD.²⁷ This method has been successfully applied to perovskite systems to examine the effect of biaxial strain^{28,29} and symmetry mismatch³⁰ as well as more complex systems such as multicomponent superlattices^{24,31} and structures that include cation displacements.³² In an original study, May and co-workers illustrated how a typical perovskite, LaNiO₃, adapts its NiO₆ network to the epitaxial strain from the substrate. In particular, the authors found that when compressive (tensile) strain is applied, the in-plane rotation increases (decreases) in magnitude whereas the out-of-plane tilts decrease (increase) in magnitude, as sketched in [Figure 1](#). In that work, the films were 20–40 pc u.c. in thickness (1 pc u.c. \approx 3.85 Å). Here, we apply a rather similar approach to films close to the ultrathin limit demonstrating for the first time to our knowledge how distortions evolve from ultrathin to thin films depending on the substrate, providing important information on the interfacial strain, structural, and chemical effects potentially giving new engineering tools for future material designs.

Because of the possibility that ultrathin films may allow the stabilization of more extreme octahedral rotations, the simulation aspect of this work was adapted and generalized from the one employed in the past and details of this can be found in the [Supporting Information](#).

The systems investigated were tensile LaNiO₃ on (001)-oriented SrTiO₃ (STO) substrate and compressive LaNiO₃ grown on (001)-oriented LaAlO₃ substrate, both series of various film thicknesses, as low as 5 u.c. (20 Å), as well as the analogous compressive system of LaAlO₃ on SrTiO₃ substrate. [Table 1](#) summarizes the room-temperature structural properties of the bulk materials involved.

The growth and characterization for the LaNiO₃ and LaAlO₃ samples can be found in the [Experimental Section, Supporting Information](#), and elsewhere.^{35–37}

For all samples, the film thickness was obtained by recording a specular CTR (described in the [Experimental Section](#) and shown in [Supporting Information](#)) and using a specialized simulation to extract the number of unit cells and the c -axis

Table 1. Summary of Bulk, Room Temperature, Structural Parameters of the Materials Used^{33,34}

	LaNiO ₃	LaAlO ₃	SrTiO ₃
pc lattice parameter	3.838 Å	3.789 Å	3.905 Å
space group	$R\bar{3}c$	$R\bar{3}c$	$Pm\bar{3}m$
Glazer system	$a^-a^-a^-$	$a^-a^-a^-$	$a^0a^0a^0$
O ₆ rotation/tilt $\alpha = \beta = \gamma$	5.2°	5.6°	0°

parameter through fitting the finite thickness fringes.³⁸ Figure 2 displays the results of this analysis.

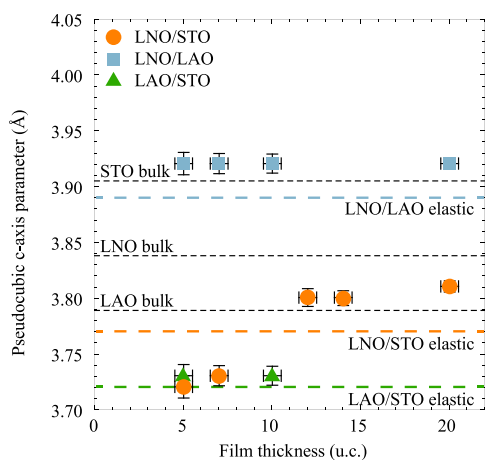


Figure 2. Pseudocubic *c*-axis lattice parameter as a function of total film thickness for the three heterostructure types, LaNiO₃/SrTiO₃, LaNiO₃/LaAlO₃, and LaAlO₃/SrTiO₃. Bulk lattice parameters for all materials are denoted by the black dashed line. The wider dashed lines show the film *c*-axis parameters that would be expected for a given heterointerface assuming only a volume change governed by the Poisson ratio ($\nu_{\text{LNO}} = 0.34$ from ref 39 and $\nu_{\text{LAO}} = 0.22$ from ref 40). Error bars derive from the reliability of fitting the finite thickness fringes in InteractiveXRDFit, which increases with film thickness.

The out-of-plane lattice parameters primarily reflect the strain state of the films where LaNiO₃/SrTiO₃ and LaAlO₃/SrTiO₃, both under tensile strain, see a reduction in their *c*-axis parameter as compared to the bulk state, whereas the opposite effect is observed in LaNiO₃/LaAlO₃ samples, which are compressively strained. Indeed, the measured *c*-axis parameters are close to that predicted by elastic theory for almost all the samples. The LaNiO₃/LaAlO₃ series does not display a thickness dependence of the *c*-axis parameter over the range examined. The LaNiO₃/SrTiO₃ series, on the other hand, shows an overall decrease of the out-of-plane lattice parameter as the thickness of the film is reduced. The two samples of LaAlO₃/SrTiO₃ are entirely consistent with a previous report on the thickness-dependence of the *c*-axis parameter where a strong modulation due to the electrostrictive effect and intermixing is observed for films of less than 4 u.c. and a relaxation of the lattice parameter back to that of bulk LaAlO₃ is seen for films over 40 u.c.⁴¹ The two thicknesses of LaAlO₃/SrTiO₃ measured here, however, belong to a thickness regime where the *c*-axis parameter was found by Cancellieri and co-workers to be constant at between 3.73–3.74 Å.⁴¹ The variation of *c*-axis parameter with thickness for the LaNiO₃/SrTiO₃ series seen here cannot be reasoned in the same way as the LaAlO₃/SrTiO₃ series was previously. In the case of LaNiO₃/SrTiO₃, there is no two-dimensional (2D) electron

system at the interface and the film remains fully strained to the substrate for all the thicknesses studied, as confirmed by analyzing the in-plane diffraction peaks. The intriguing thickness dependence of the *c*-axis in the LaNiO₃/SrTiO₃ series, however, may be correlated with a strong thickness-dependence of the lattice of oxygen octahedral distortions, as will be discussed.

To access information about the rotations and tilts of the oxygen octahedral network, a series of off-specular CTRs were recorded around 6–10 half-integer Bragg peaks (HIBPs) belonging to different families of diffraction planes. All measurements reported were carried out at room temperature.

For all samples, all HIBPs of the form $(\frac{h}{2}, \frac{k}{2}, \frac{l}{2})$ with *h*, *k*, *l* odd integers were observed except those that satisfy $h = k = l$, which are forbidden. No intensity was observed at conditions where one of the *h*, *k*, or *l* is an even integer, peaks that would arise from in-phase rotations of octahedra only, thus no orthorhombicity is present. This confirms that all the films adhere to the Glazer system $a^-b^-c^-$, at lowest possible symmetry. As these samples are biaxially strained, it is reasonable to increase the symmetry to $a^-a^-c^-$, reflecting the assumed equivalence of the two in-plane directions. This assumption is justified by the observation of equivalent intensities in all four quadrants $(\frac{h}{2}, \frac{k}{2}, \frac{l}{2})$, $(\frac{-h}{2}, \frac{k}{2}, \frac{l}{2})$, $(\frac{h}{2}, \frac{-k}{2}, \frac{l}{2})$, and $(\frac{-h}{2}, \frac{-k}{2}, \frac{l}{2})$. The condition $\alpha = \beta$ will henceforth be enforced.

In order to obtain the magnitudes of the octahedral tilts and rotations, it was necessary to compare measured and simulated relative diffraction intensities of HIBPs. The measured intensity for a given peak was taken as the integrated intensity of a Gaussian that has been fitted to the data including a linear slope offset and a constant background, as shown in the Supporting Information. Figure 3 shows the evolution of the

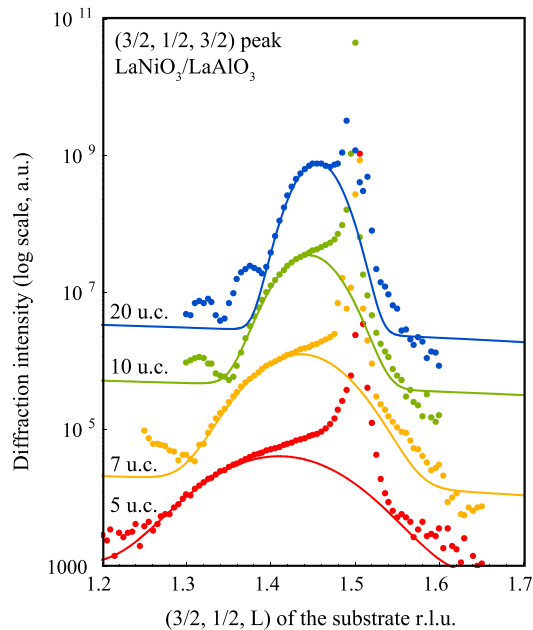


Figure 3. Diffraction intensity close to the $(\frac{3}{2}, \frac{1}{2}, l)$ reflection as an example of half integer Bragg peaks for the series of LaNiO₃/LaAlO₃ samples showing the evolution with film thickness. Solid lines are the fits to the data of a Gaussian function plus a linear slope and a constant background. The data was vertically displaced for clarity.

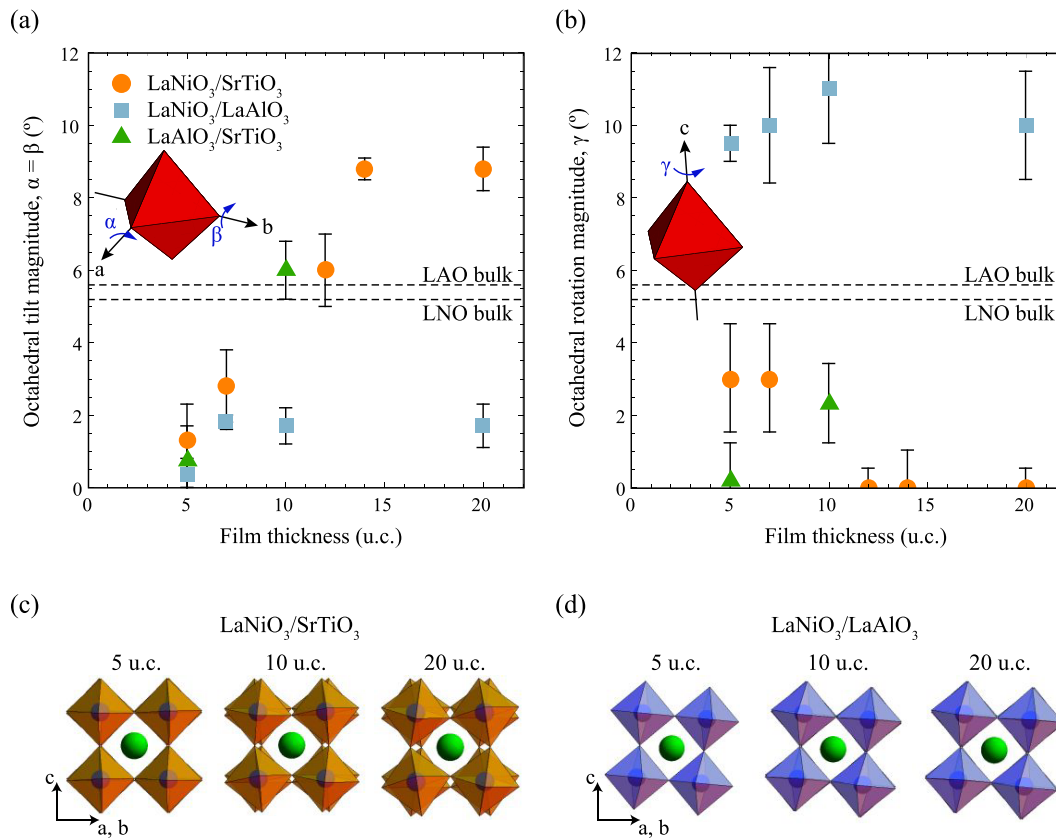


Figure 4. Oxygen octahedral tilt (a) and rotation (b) magnitudes as a function of total film thickness in pc u.c. for the three types of heterostructure system analyzed. Dashed lines indicate the octahedral tilt and rotation magnitudes for the bulk materials LaNiO_3 and LaAlO_3 (SrTiO_3 having no room-temperature octahedral distortions). The uncertainties are generated as described in the Supporting Information. An inset on each of the figures illustrates the relevant Glazer angles, α , β , and γ . Panels c and d are side view sketches of the octahedral tilt systems determined for, respectively, the $\text{LaNiO}_3/\text{SrTiO}_3$ and $\text{LaNiO}_3/\text{LaAlO}_3$ films, with the true measured angles. Three representative thicknesses of each series are shown. A strong evolution with thickness is seen for $\text{LaNiO}_3/\text{SrTiO}_3$ (panel c), whereas in panel d there is little variation with thickness for the $\text{LaNiO}_3/\text{LaAlO}_3$ series.

diffraction intensity close to the $(\frac{3}{2}, \frac{1}{2}, l)$ reflection for the series of $\text{LaNiO}_3/\text{LaAlO}_3$ samples and illustrates the fits that are obtained. When the thickness is reduced to only 5 u.c. (2 nm), the intensity is still clear and easily quantifiable, even when directly adjacent to the much stronger substrate contribution. This represents a serious technical achievement as this intensity derives only from small displacements of light oxygen atoms in a film only 2 nm in thickness. Also observed are the oscillations resulting from the finite thickness effect in the films, confirming the high crystalline quality of the oxygen sublattice.

The simulated diffraction intensity was computed from given atomic positions, where only the oxygen positions are considered, as the cations do not contribute to these HIBPs. Simulating the atomic positions was done on a Mathematica 9 notebook with the only input parameters being the Glazer rotation system (i.e., how each octahedron rotates with respect to its neighbors), assuming the oxygen octahedra to be rigid. The tilt and rotation magnitudes are always defined with respect to the internal symmetry axes of the octahedron and not to the original, unrotated Cartesian reference frame. To avoid possible issues arising from the non-Abelian nature of the $\text{SO}(3)$ group, the Rodrigues formula for three-dimensional rotations around a vector is applied iteratively around the three internal axes of octahedral symmetry, each time by an increment of the total rotation angle of that axis. More detail

on this process is given in the Supporting Information. After the rotation operation on one octahedron is complete for a given set of input angles, α , β , γ , the full unit cell can be built by generating seven additional octahedra, rotated by the same magnitude of angles but in different directions depending upon the Glazer rotation system.

After obtaining a simulated doubled perovskite unit cell in three dimensions, the diffracted intensity can be calculated from the structure factor for the comprising 24 oxygen atoms via

$$I_{\text{sim}} = \sum_{p=1}^4 d_p [F_{h,k,l}]_p^2$$

where d_p expresses the relative domain population of four geometrically equivalent domains and with the structure factor being given by

$$F_{h,k,l} = f_{\text{O}^{2-}} \sum_{n=1}^{24} \exp(2\pi i(hx_n + ky_n + lz_n))$$

where $f_{\text{O}^{2-}}$ is the form factor for the O^{2-} oxygen ion and the n th oxygen is situated at (x_n, y_n, z_n) in coordinates relative to the doubled unit cell. The domains indexed by p arise due to the equivalence of the final atomic positions when the octahedron

undergoes rotations of the type (α, β, γ) , $(-\alpha, \beta, \gamma)$, $(\alpha, -\beta, \gamma)$ and $(\alpha, \beta, -\gamma)$.

Once both the simulated and experimental intensities have been obtained they are normalized to the same half integer (h, k, l) usually the one corresponding to the highest experimental intensity, giving two comparable sets. If the normalized simulated intensity for the n th diffraction peak is $I_{\text{sim}}(h, k, l)_n$ and the normalized experimentally measured intensity for the n th diffraction peak is $I_{\text{exp}}(h, k, l)_n$, then the residual sum of squares (RSS) is defined as

$$\text{RSS} = \sum_n |I_{\text{sim}}(h, k, l)_n - I_{\text{exp}}(h, k, l)_n|^2$$

and from this, the input parameters (α, β, γ) and the d_p can be varied until the RSS is minimized, determining the best fit structure. This was done for the two samples of LaAlO₃/SrTiO₃, the series of LaNiO₃/SrTiO₃, and the series of LaNiO₃/LaAlO₃ and the resultant rotation magnitudes are plotted in Figure 4. More detail on the fitting procedure is shown in the Supporting Information.

For the thicker (>14 u.c.) LaNiO₃ samples, on both SrTiO₃ and LaAlO₃ substrate, the angles reflect what would be expected from only considering the effect of biaxial strain, as sketched in panel b of Figure 1. On SrTiO₃, the strain is tensile so the LaNiO₃ oxygen octahedra would be expected to straighten their equatorial Ni—O—Ni bonds laterally to maximize the in-plane space, meaning a rotation around the c -axis toward a smaller γ of nearly 0°. The α and β , on the other hand, are much larger at almost 9°. The octahedra reduce their out-of-plane extension by increasing α and β to account for the tetragonal distortion stemming from tensile biaxial strain.

The opposite situation is observed for the compressive LaNiO₃/LaAlO₃; here, γ is forced to around 10° by in-plane restriction while α and β decrease to accommodate the new tetragonal distortion that is longer out-of-plane than in-plane. These Glazer rotation magnitudes would therefore likely be found for continually increasing thickness until strain relaxation, where the bulk of 5.2° would be preferred. A previous report finds, for slightly thicker LaNiO₃ on both LaAlO₃ and SrTiO₃ substrates, Glazer angles that are in good agreement with what we see here.²⁷ Theoretical calculations on LaAlO₃ films also reproduce these higher thickness, biaxially strained structures and LaNiO₃ would be expected to be similar.^{29,42}

Now we perform the same analysis on films with thicknesses as low as 5 u.c. (2 nm) and in doing so, we have unveiled a striking behavior toward the atomic limit that suggests that the nature of the specific heterointerface plays a role in determining the film structure far beyond the simple consideration of biaxial strain.

First, in the case of the films on LaAlO₃ substrate, the structure similar to the thicker LaNiO₃/LaAlO₃ films holds down to the lowest thicknesses measured, 5 u.c., similar to how the c -axis parameter appears to be independent of the total thickness, as shown in Figure 2. In our previous work, we found that although the depth-resolved structure of LaNiO₃ films on LaAlO₃ substrate varies the depth-averaged structure is independent of the total film thickness, as is observed here.²³

By contrast, the octahedral rotations in films on SrTiO₃ substrate depend much more strongly on the total thickness. In particular, as the thickness is decreased, the α - and β -tilts tend toward 0°, which, it should be noted, is the room temperature tilt angle in the cubic SrTiO₃ substrate. It is therefore likely

that the film structure is inclined to adopt the same out-of-plane tilts as the substrate through octahedral coupling at the heterointerface. The effect of the interfacial octahedral coupling reduces when the thickness increases to 12–15 u.c., the same length scale over which the c -axis parameter changes in the same system. The actual interfacial coupling length could be much less than 12 u.c., however, as it should be kept in mind that these measurements are sensitive to the entire film thickness and so give an average structure. Indeed, several studies find an interfacial coupling of the octahedral tilt magnitudes over around four unit cells.^{43–45} Other works also find a highly distorted surface structure extending over around 2 u.c. which, although it would contribute to the macroscopic measurements reported here, does not become relevant unless films even thinner than those measured in this work, the so-called ultrathin regime, are investigated.^{23,46} Interesting behavior emerges in ultrathin films. LaNiO₃ films are found to become insulating under 6 u.c. thickness when grown on SrTiO₃ substrate but retain metallicity until as thin as 3 u.c. on LaAlO₃ substrate.^{36,47} This thickness-dependent metal–insulator transition is not fully understood but the substrate effect may be related to the type of interfacial structural coupling observed here.

The $\alpha = \beta$ tilts determined for the two LaAlO₃/SrTiO₃ films compare well to those obtained from a microscopy study²⁰ and fit well with the LaNiO₃/SrTiO₃ series, suggesting that LaAlO₃ and LaNiO₃ react to the SrTiO₃ interface in a similar way (except for the 2D electron system). This is most likely as a result of their structural similarities in bulk (both are rhombohedral with a similar distortion magnitude), as summarized in Table 1. Comparing the bulk similarities of LaNiO₃ and LaAlO₃ also aids in the understanding of the LaNiO₃/LaAlO₃ interface.

In the case of the LaNiO₃ films on LaAlO₃ substrate, the two materials are much closer in lattice parameter and the interface has, nominally, no polar discontinuity, no A-site discontinuity, and no symmetry discontinuity so the LaNiO₃ can adopt a more stable, low energy, structure. In the LaNiO₃/SrTiO₃ system, however, there are chemical, polar, and symmetry discontinuities so it could be that the LaNiO₃ is more frustrated and has to respond to the interface in a more complex way as the thickness is increased, first by octahedral coupling and then by accommodating the biaxial strain. Such interface effects are not unusual.^{18,21,48} Although strain relaxation on the scale of the thicknesses measured here can be ruled out by the in-plane coherency of the diffraction peaks between the substrate and the film, there is still the possibility that the LaNiO₃/SrTiO₃ interface has a higher propensity to develop point defects and this may have an effect as well. Regardless of the origin, the smooth evolution of the thickness dependence of the octahedral rotations in the LaNiO₃/SrTiO₃ system may have an application in atomic-scale structural engineering through integration in a superlattice structure. This is especially appealing given the strong relationship between the Ni—O—Ni bond angle and electronic and magnetic properties of rare earth nickelates in general.

In summary, a simulation designed to give the Glazer rotation system and angles through fitting half integer Bragg peaks measured by synchrotron X-ray diffraction was used for three different perovskite oxide heterostructures; LaNiO₃/LaAlO₃, LaNiO₃/SrTiO₃, and LaAlO₃/SrTiO₃.

LaNiO₃/LaAlO₃ appears to be a system where the compatibility between the two component materials leads to

a general film structure that is independent of the total film thickness and the Glazer rotation angles reflect an almost pure biaxial strain effect.

LaNiO₃/SrTiO₃ and LaAlO₃/SrTiO₃ are very similar, possibly owing to the corresponding similarity between bulk LaNiO₃ and LaAlO₃. In the LaNiO₃/SrTiO₃ system, the stark differences between the ideal film structure and the substrate are such that the film appears to accommodate the heterointerface in a complex and thickness-dependent way, first by undoing the octahedral tilting to match the substrate and then for thicker films by octahedral rotation to fill the tetragonality imposed by biaxial strain.

This demonstrated ability of using layer thickness as a tool to control the atomic positions on such a fine scale is certainly appealing from a materials engineering standpoint and is likely to be generalizable to many perovskite/perovskite systems with a high level of structural and chemical discontinuity.

■ EXPERIMENTAL SECTION

The substrates used were SrTiO₃ and LaAlO₃, both (001)-oriented and distributed by Crystec GmbH. The SrTiO₃ is treated by the manufacturer to obtain a TiO₂ terminated surface while the LaAlO₃ is annealed in flowing oxygen at 1000 °C to obtain an AlO₂ terminated surface.

The LaNiO₃ films were grown by radio frequency off-axis magnetron sputtering at a temperature of 510 °C and in a 7:2 Ar/O₂ mix at a pressure of 0.24 mbar. The LaAlO₃ films were grown by pulsed laser deposition in a pressure of 8×10^{-5} mbar at a temperature of 800 °C and a laser repetition rate of 1 Hz with a postannealing stage in 200 mbar of oxygen at 530 °C for an hour. Characterization can be found in the [Supporting Information](#).

The XRD measurements took the form of crystal truncation rods (CTRs) and were performed at the Materials Science beamline X04SA, Surface Diffraction endstation at the Swiss Light Source, Paul Scherrer Institut.⁴⁹ The samples were mounted on a hexapod in a 6-circle diffractometer and, for all off-specular conditions, the incident angle was fixed at 4°. The two-dimensional Pilatus-II pixel detector was used to capture the diffracted beam. All measurements were carried out at room temperature and under flowing helium and at photon energies between 13 and 15.5 keV.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: [10.1021/acs.nanolett.9b01772](https://doi.org/10.1021/acs.nanolett.9b01772).

Additional sample characterization. Details on the simulation of atomic positions. The determination of the experimental intensities and associated coherence thickness. The goodness of the simulation fits and a summary of the output structural parameters. Supplementary structural parameters (Ni—O—Ni bond angle and Ni—O bond lengths) ([PDF](#))

■ AUTHOR INFORMATION

Corresponding Author

*E-mail: jennifer.fowlie@unige.ch.

ORCID

Jennifer Fowlie: [0000-0002-3528-6390](https://orcid.org/0000-0002-3528-6390)

Present Address

^{||}(M.G.) Physik Institut, University of Zürich, 190 Winterthurerstrasse, 8057 Zürich, Switzerland.

Notes

The authors declare no competing financial interest.

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■ REFERENCES

- (1) Khomskii, D. I. *Transition Metal Compounds*; Cambridge University Press: Cambridge, 2014.
- (2) Mitchell, R. H. *Perovskites: Modern and Ancient*; Almaz: Thunder Bay, 2003.
- (3) Rondinelli, J. M.; Spaldin, N. a. Substrate Coherency Driven Octahedral Rotations in Perovskite Oxide Films. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2010**, *82* (11), 113402.
- (4) Rondinelli, J. M.; May, S. J.; Freeland, J. W. Control of Octahedral Connectivity in Perovskite Oxide Heterostructures: An Emerging Route to Multifunctional Materials Discovery. *MRS Bull.* **2012**, *37* (03), 261–270.
- (5) Islam, M. A.; Rondinelli, J. M.; Spanier, J. E. Normal Mode Determination of Perovskite Crystal Structures with Octahedral Rotations: Theory and Applications. *J. Phys.: Condens. Matter* **2013**, *25* (17), 175902.
- (6) Balachandran, P. V.; Rondinelli, J. M. Interplay of Octahedral Rotations and Breathing Distortions in Charge-Ordering Perovskite Oxides. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2013**, *88*, 054101 DOI: [10.1103/PhysRevB.88.054101](https://doi.org/10.1103/PhysRevB.88.054101).
- (7) Pavarini, E.; Biermann, S.; Poteryaev, A.; Lichtenstein, A. I.; Georges, A.; Andersen, O. K. Mott Transition and Suppression of Orbital Fluctuations in Orthorhombic 3 d 1 Perovskites. *Phys. Rev. Lett.* **2004**, *92*, 4–7.
- (8) Medarde, M. L. Structural, Magnetic and Electronic Properties of RNiO₃ Perovskites (R Equals Rare Earth). *J. Phys.: Condens. Matter* **1997**, *9* (8), 1679–1707.
- (9) Catalan, G. Progress in Perovskite Nickelate Research. *Phase Transitions* **2008**, *81* (7–8), 729–749.
- (10) Goodenough, J. B. Electronic and Ionic Transport Properties and Other Physical Aspects of Perovskites. *Rep. Prog. Phys.* **2004**, *67* (11), 1915–1993.
- (11) Dabrowski, B.; Kolesnik, S.; Baszczuk, A.; Chmaissem, O.; Maxwell, T.; Mais, J. Structural, Transport, and Magnetic Properties of RMnO₃ perovskites (R = La, Pr, Nd, Sm, 153Eu, Dy). *J. Solid State Chem.* **2005**, *178* (3), 629–637.
- (12) Medarde, M.; Lacorre, P.; Conder, K.; Fauth, F.; Furrer, A. Giant ¹⁶O-¹⁸O Isotope Effect on the Metal-Insulator Transition of RNiO₃ Perovskites (R = Rare Earth). *Phys. Rev. Lett.* **1998**, *80*, 2397–2400.
- (13) Koirala, P.; Mizzi, C. A.; Marks, L. D. Direct Observation of Large Flexoelectric Bending at the Nanoscale in Lanthanide Scandates. *Nano Lett.* **2018**, *18* (6), 3850–3856.
- (14) Subedi, A.; Cavalleri, A.; Georges, A. Theory of Nonlinear Phononics for Coherent Light Control of Solids. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2014**, *89* (22), 1–5.
- (15) Xiang, H. J.; Guennou, M.; Ñiguez, J.; Kreisel, J.; Bellaiche, L. Rules and Mechanisms Governing Octahedral Tilts in Perovskites under Pressure. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2017**, *96* (5), 1–11.

- (16) Herklotz, A.; Wong, A. T.; Meyer, T.; Biegalski, M. D.; Lee, H. N.; Ward, T. Z. Controlling Octahedral Rotations in a Perovskite via Strain Doping. *Sci. Rep.* **2016**, *6*, 1–7.
- (17) Hwang, H. Y.; Iwasa, Y.; Kawasaki, M.; Keimer, B.; Nagaosa, N.; Tokura, Y. Emergent Phenomena at Oxide Interfaces. *Nat. Mater.* **2012**, *11* (2), 103–113.
- (18) Zubko, P.; Gariglio, S.; Gabay, M.; Ghosez, P.; Triscone, J.-M. Interface Physics in Complex Oxide Heterostructures. *Annu. Rev. Condens. Matter Phys.* **2011**, *2* (1), 141–165.
- (19) Aso, R.; Kan, D.; Shimakawa, Y.; Kurata, H. Atomic Level Observation of Octahedral Distortions at the Perovskite Oxide Heterointerface. *Sci. Rep.* **2013**, *3*, 1–6.
- (20) Gazquez, J.; Stengel, M.; Mishra, R.; Scigaj, M.; Varela, M.; Roldan, M. A.; Fontcuberta, J.; Sánchez, F.; Herranz, G. Competition between Polar and Nonpolar Lattice Distortions in Oxide Quantum Wells: New Critical Thickness at Polar Interfaces. *Phys. Rev. Lett.* **2017**, *119* (10), 1–6.
- (21) Borisevich, A. Y.; Chang, H. J.; Huijben, M.; Oxley, M. P.; Okamoto, S.; Niranjana, M. K.; Burton, J. D.; Tsymbal, E. Y.; Chu, Y. H.; Yu, P.; et al. Suppression of Octahedral Tilts and Associated Changes in Electronic Properties at Epitaxial Oxide Heterostructure Interfaces. *Phys. Rev. Lett.* **2010**, *105* (8), 1–4.
- (22) Jia, C. L.; Mi, S. B.; Faley, M.; Poppe, U.; Schubert, J.; Urban, K. Oxygen Octahedron Reconstruction in the SrTiO₃/LaAlO₃ Heterointerfaces Investigated Using Aberration-Corrected Ultrahigh-Resolution Transmission Electron Microscopy. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2009**, *79* (8), 2–5.
- (23) Fowlie, J.; Gibert, M.; Tieri, G.; Gloter, A.; Íñiguez, J.; Filippetti, A.; Catalano, S.; Gariglio, S.; Schober, A.; Guennou, M.; et al. Conductivity and Local Structure of LaNiO₃ Thin Films. *Adv. Mater.* **2017**, *29* (18), 1605197.
- (24) May, S. J.; Smith, C. R.; Kim, J. W.; Karapetrova, E.; Bhattacharya, A.; Ryan, P. J. Control of Octahedral Rotations in (LaNiO₃)_n/(SrMnO₃)_m Superlattices. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2011**, *83* (15), 2–5.
- (25) Moon, E. J.; Balachandran, P. V.; Kirby, B. J.; Keavney, D. J.; Sichel-Tissot, R. J.; Schlepütz, C. M.; Karapetrova, E.; Cheng, X. M.; Rondinelli, J. M.; May, S. J. Effect of Interfacial Octahedral Behavior in Ultrathin Manganite Films. *Nano Lett.* **2014**, *14* (5), 2509.
- (26) Glazer, A. M. Simple Ways of Determining Perovskite Structures. *Acta Crystallogr., Sect. A: Cryst. Phys., Diffraction, Theor. Gen. Crystallogr.* **1975**, *31* (6), 756–762.
- (27) May, S. J.; Kim, J.-W.; Rondinelli, J. M.; Karapetrova, E.; Spaldin, N. A.; Bhattacharya, A.; Ryan, P. J. Quantifying Octahedral Rotations in Strained Perovskite Oxide Films. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2010**, *82* (1), 014110.
- (28) Rotella, H.; Lüders, U.; Janolin, P.-E.; Dao, V. H.; Chateigner, D.; Feyerherm, R.; Dudzik, E.; Prellier, W. Octahedral Tilting in Strained LaVO₃ Thin Films. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2012**, *85* (18), 184101.
- (29) Johnson-Wilke, R. L.; Marincel, D.; Zhu, S.; Warusawithana, M. P.; Hatt, A.; Sayre, J.; Delaney, K. T.; Engel-Herbert, R.; Schlepütz, C. M.; Kim, J.-W.; et al. Quantification of Octahedral Rotations in Strained LaAlO₃ Films via Synchrotron X-Ray Diffraction. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2013**, *88* (17), 174101.
- (30) Kan, D.; Wakabayashi, Y.; Tajiri, H.; Shimakawa, Y. Interfacially Engineered Oxygen Octahedral Rotations and Their Impact on Strain Relief in Coherently Grown SrRuO₃ Films. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2016**, *94* (2), 1–6.
- (31) Kinyanjui, M. K.; Lu, Y.; Gauquelin, N.; Wu, M.; Frano, A.; Wochner, P.; Reehuis, M.; Christiani, G.; Logvenov, G.; Habermeier, H. U.; et al. Lattice Distortions and Octahedral Rotations in Epitaxially Strained LaNiO₃/LaAlO₃ Superlattices. *Appl. Phys. Lett.* **2014**, *104* (22), 221909.
- (32) Brahlek, M.; Choquette, A. K.; Smith, C. R.; Engel-Herbert, R.; May, S. J. Structural Refinement of Pbnm-Type Perovskite Films from Analysis of Half-Order Diffraction Peaks. *J. Appl. Phys.* **2017**, *121* (4), 045303.
- (33) García-Muñoz, J. L.; Rodríguez-Carvajal, J.; Lacorre, P.; Torrance, J. B. Neutron-Diffraction Study of RNiO₃ (R = La, Pr, Nd, Sm): Electronically Induced Structural Changes across the M. *Phys. Rev. B: Condens. Matter Mater. Phys.* **1992**, *46* (8), 4414–4425.
- (34) Müller, K. A.; Berlinger, W.; Waldner, F. Characteristic Structural Phase Transition in Perovskite-Type Compounds. *Phys. Rev. Lett.* **1968**, *21* (12), 814–817.
- (35) Scherwitzl, R.; Zubko, P.; Lichtensteiger, C.; Triscone, J. M. Electric-Field Tuning of the Metal-Insulator Transition in Ultrathin Films of LaNiO₃. *Appl. Phys. Lett.* **2009**, *95* (22), 222114.
- (36) Scherwitzl, R.; Gariglio, S.; Gabay, M.; Zubko, P.; Gibert, M.; Triscone, J. M. Metal-Insulator Transition in Ultrathin LaNiO₃ Films. *Phys. Rev. Lett.* **2011**, *106* (24), 3–6.
- (37) Cancellieri, C.; Reyren, N.; Gariglio, S.; Caviglia, A. D.; Fête, A.; Triscone, J. M. Influence of the Growth Conditions on the LaAlO₃/SrTiO₃ Interface Electronic Properties. *Epl* **2010**, *91* (1), 17004.
- (38) Lichtensteiger, C. InteractiveXRDFit: A New Tool to Simulate and Fit X-Ray Diffractograms of Oxide Thin Films and Heterostructures. *J. Appl. Crystallogr.* **2018**, *51* (6), 1745–1751.
- (39) Masys; Jonauskas, V. Elastic Properties of Rhombohedral, Cubic, and Monoclinic Phases of LaNiO₃ by First Principles Calculations. *Comput. Mater. Sci.* **2015**, *108* (PA), 153–159.
- (40) Luo, X.; Wang, B. Structural and Elastic Properties of LaAlO₃ from First-Principles Calculations. *J. Appl. Phys.* **2008**, *104* (7), 073518.
- (41) Cancellieri, C.; Fontaine, D.; Gariglio, S.; Reyren, N.; Caviglia, A. D.; Fête, A.; Leake, S. J.; Pauli, S. A.; Willmott, P. R.; Stengel, M.; et al. Electrostriction at the LaAlO₃/SrTiO₃ Interface. *Phys. Rev. Lett.* **2011**, *107* (5), 1–4.
- (42) Hatt, A. J.; Spaldin, N. A. Structural Phases of Strained LaAlO₃ Driven by Octahedral Tilt Instabilities. *Phys. Rev. B: Condens. Matter Mater. Phys.* **2010**, *82* (19), 1–5.
- (43) Aso, R.; Kan, D.; Shimakawa, Y.; Kurata, H. Control of Structural Distortions in Transition-Metal Oxide Films through Oxygen Displacement at the Heterointerface. *Adv. Funct. Mater.* **2014**, *24* (33), 5177–5184.
- (44) Liao, Z.; Gauquelin, N.; Green, R. J.; Müller-Caspary, K.; Lobato, I.; Li, L.; van Aert, S.; Verbeeck, J.; Huijben, M.; Grisolia, M. N. Metal – Insulator-Transition Engineering by Modulation Tilt-Control in Perovskite Nickelates for Room Temperature Optical Switching. *Proc. Natl. Acad. Sci. U. S. A.* **2018**, *115* (38), 9515.
- (45) Liao, Z.; Huijben, M.; Zhong, Z.; Gauquelin, N.; Macke, S.; Green, R. J.; Van Aert, S.; Verbeeck, J.; Van Tendeloo, G.; Held, K.; et al. Controlled Lateral Anisotropy in Correlated Manganite Heterostructures by Interface-Engineered Oxygen Octahedral Coupling. *Nat. Mater.* **2016**, *15* (4), 425–431.
- (46) Kumah, D. P.; Malashevich, A.; Disa, A. S.; Arena, D. A.; Walker, F. J.; Ismail-Beigi, S.; Ahn, C. H. Effect of Surface Termination on the Electronic Properties of LaNiO₃ Films. *Phys. Rev. Appl.* **2014**, *2* (5), 1–7.
- (47) Son, J.; Moetakef, P.; Lebeau, J. M.; Ouellette, D.; Balents, L.; Allen, S. J.; Stemmer, S. Low-Dimensional Mott Material: Transport in Ultrathin Epitaxial LaNiO₃ Films. *Appl. Phys. Lett.* **2010**, *96* (6), 062114.
- (48) Vailionis, A.; Boschker, H.; Liao, Z.; Smit, J. R. A.; Rijnders, G.; Huijben, M.; Koster, G. Symmetry and Lattice Mismatch Induced Strain Accommodation near and Away from Correlated Perovskite Interfaces. *Appl. Phys. Lett.* **2014**, *105* (13), 131906.
- (49) Willmott, P. R.; Meister, D.; Leake, S. J.; Lange, M.; Bergamaschi, A.; Böge, M.; Calvi, M.; Cancellieri, C.; Casati, N.; Cervellino, A.; et al. The Materials Science Beamline Upgrade at the Swiss Light Source. *J. Synchrotron Radiat.* **2013**, *20* (5), 667–682.