



# Crystal structures of the c-di-AMP–synthesizing enzyme CdaA

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Cyclic di-AMP (c-di-AMP) is the only second messenger known to be essential for bacterial growth. It has been found mainly in Gram-positive bacteria, including pathogenic bacteria like *Listeria monocytogenes*. CdaA is the sole diadenylate cyclase in *L. monocytogenes*, making this enzyme an attractive target for the development of novel antibiotic compounds. Here we report crystal structures of CdaA from *L. monocytogenes* in the apo state, in the post-catalytic state with bound c-di-AMP and catalytic  $\text{Co}^{2+}$  ions, as well as in a complex with AMP. These structures reveal the flexibility of a tyrosine side chain involved in locking the adenine ring after ATP binding. The essential role of this tyrosine was confirmed by mutation to Ala, leading to drastic loss of enzymatic activity.

Bacteria have the ability to perceive environmental changes, leading to rapid and effective adaptation by utilizing different proteins as well as second messengers to transduce signals in the cell. In response to external stimuli, the intracellular concentration of second messengers, like cyclic dinucleotides and linear mononucleotides, varies to regulate and coordinate cellular processes (1–3). Cyclic di-AMP (c-di-AMP)<sup>2</sup> is the most recently discovered bacterial signaling nucleotide and, to date, has been found mostly in Gram-positive bacteria. c-di-AMP is involved in different cellular processes, such as DNA integrity scanning, cell wall metabolism, and osmolyte homeostasis (for a review, see Refs. 4–6). c-di-AMP is the only essential second messenger in bacteria because of its role in potassium homeostasis. It regulates potassium importers at high intracellular  $\text{K}^+$  concentrations, whereas c-di-AMP is not essential at low  $\text{K}^+$  concentrations (7). Interestingly, c-di-AMP becomes toxic when its degradation is blocked; hence, a tightly controlled intracellular c-di-AMP concentration is required for bacterial growth (8).

Proteins containing a diadenylate cyclase (DAC) domain have been bioinformatically identified, mainly in Gram-positive bacteria of the phyla Actinobacteria and Firmicutes but also in Gram-negative Cyanobacteria, Chlamydiae, Bacteroidetes, Fusobacteria, and Deltaproteobacteria and even in archaea of the phylum Euryarchaeota (5). Several DAC

domain-containing proteins from various bacterial species have also been experimentally proven to produce c-di-AMP. Many of these bacteria are well-known pathogens, e.g. *Mycobacterium tuberculosis* (9), *Staphylococcus aureus* (10) and *Listeria monocytogenes* (11). In total, eight families of diadenylate cyclases have been identified so far, sharing the highly conserved DAC domain (12). However, DACs differ in their additional domains and domain organization, suggesting that DAC enzymes are regulated by different signals (12).

The three-dimensional structure of a DAC domain was first reported for DisA, a multidomain protein with an N-terminal DAC domain (13). This structure revealed that, within the homo-octameric DisA, two adjacent and properly positioned DAC domains, each with one ATP bound, catalyze the synthesis of c-di-AMP. Based on the homology of all DAC domains, it was proposed that DAC domains with bound ATP need to dimerize in a specific arrangement to catalyze c-di-AMP formation.

The importance of c-di-AMP for the growth of several pathogenic bacteria is marked by an increased resistance to cell wall-targeting antibiotics (10, 14). Its absence in humans makes DAC enzymes an interesting target for the development of novel antibiotics by structure-based drug design. Therefore, CdaA, the only DAC of the human pathogen *L. monocytogenes*, was previously characterized biochemically and structurally. The analysis revealed that CdaA is active with  $\text{Co}^{2+}$  or  $\text{Mn}^{2+}$  ions as cofactors but inactive in the presence of  $\text{Mg}^{2+}$  ions (15). The CdaA crystal structure unveiled the monomeric and catalytically inactive enzyme–substrate complex with bound ATP and  $\text{Mg}^{2+}$ , leaving the structure of a dimeric and active form with a bound  $\text{Co}^{2+}$  or  $\text{Mn}^{2+}$  cofactor still to be determined. Such a crystal structure could shed light on the role of the metal ion in the catalytic reaction.

In this study we report two new crystal structures of CdaA from *L. monocytogenes* at 2.0 Å and 2.8 Å resolution, representing the enzyme in its apo form and the post-catalytic homodimeric enzyme–product complex, respectively. The structure of CdaA with bound c-di-AMP was obtained by co-crystallization of CdaA in the presence of ATP and  $\text{Co}^{2+}$  ions. Comparison of the CdaA structure in the apo state with the ligand-bound forms of CdaA (ATP, AMP, or c-di-AMP) revealed conformational changes of a tyrosine residue present in the active site. Mutation of this tyrosine to alanine abolishes c-di-AMP formation and, thus, demonstrates its functional importance. Furthermore, we confirmed that CdaA is active in the presence of  $\text{Mn}^{2+}$  or  $\text{Co}^{2+}$  ions, with significantly higher activity in the case of  $\text{Mn}^{2+}$ , but it is inactive in the presence of  $\text{Mg}^{2+}$  ions. These

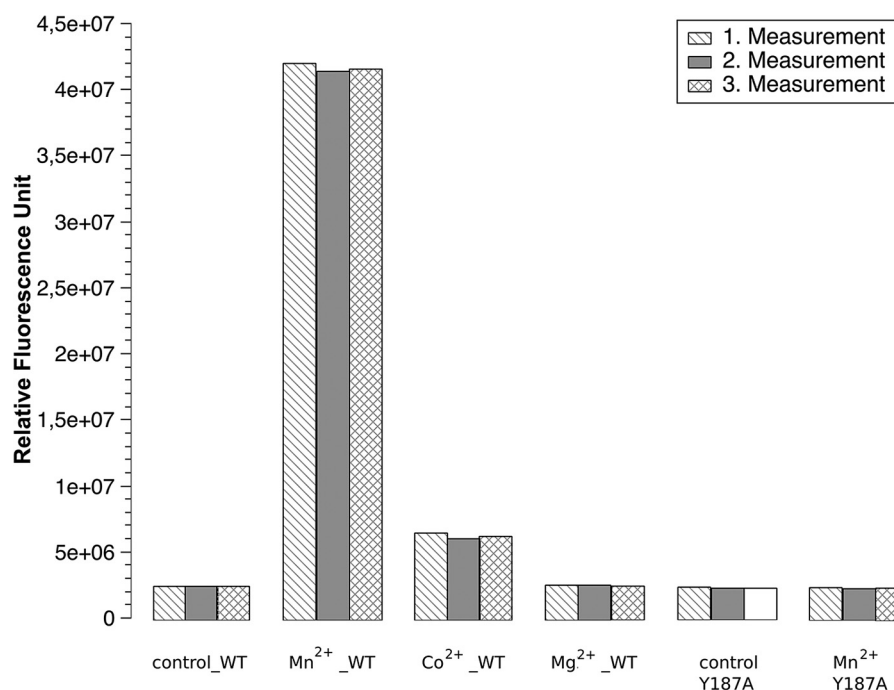
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This article contains Figs. S1–S6.

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<sup>2</sup> The abbreviations used are: c-di-AMP, cyclic di-AMP; DAC, diadenylate cyclase; TM, transmembrane; r.m.s.d., root mean square deviation.

## Structures of the diadenylate cyclase CdaA



**Figure 1. *In vitro* diadenylate cyclase activity of  $\Delta 100\text{CdaA}$ .** Presented is a histogram displaying three independent measurements. A control measurement was performed using WT  $\Delta 100\text{CdaA}$  without addition of any divalent metal cations. The histogram represents the divalent metal cation preferences of WT  $\Delta 100\text{CdaA}$ . The highest amount of *c*-di-AMP was formed in the presence of  $\text{MnCl}_2$ , whereas, in the presence of  $\text{CoCl}_2$ , the amount of the product is significantly reduced. For  $\text{MgCl}_2$  and  $\text{CaCl}_2$ , production of *c*-di-AMP could not be confirmed, as it was within the range of the control. Additionally, it represents the importance of Tyr-187 on catalysis. The mutant Y187A causes a significant reduction (5-fold) of diadenylate cyclase activity, confirming its essential role in *c*-di-AMP synthesis.

new CdaA structures could serve as an important starting point for future rational drug design.

### Results

Structure-based development of novel antibiotic drugs requires high-resolution three-dimensional structures of the targeted enzyme and enzyme–inhibitor complexes. CdaA of the human pathogen *L. monocytogenes* appears to be an attractive target, as *c*-di-AMP synthesis is essential for bacterial growth and CdaA is the only DAC in this pathogenic bacterium, whereas there are no DACs in humans. For this study, truncated  $\Delta 100\text{CdaA}$ , missing the N-terminal transmembrane (TM) helices and the 20 amino acids linking the TM to the DAC domain, was used because the transmembrane helices hamper the solubility of the recombinant full-length protein. We have demonstrated previously that this truncated  $\Delta 100\text{CdaA}$  has preserved its enzymatic activity with a higher enzymatic activity for  $\text{Co}^{2+}$  compared with  $\text{Mn}^{2+}$  but no activity for  $\text{Mg}^{2+}$  (15). Although, in this previous study, the *in vitro* activity was measured by LC-MS/MS, we now applied a direct fluorescence-based measurement of *c*-di-AMP formation by its binding to coralyne (16). In contrast to the results obtained with the LC-MS/MS method, more efficient *c*-di-AMP synthesis was observed in the presence of  $\text{Mn}^{2+}$  compared with  $\text{Co}^{2+}$  (Fig. 1A).

### Structure of apo CdaA

One approach for identification of potential inhibitors is crystallographic fragment screening, which desires crystals of CdaA in its apo state. Therefore,  $\Delta 100\text{CdaA}$  was crystallized in

the absence of ATP and divalent metal ions. Crystals of apo-CdaA were obtained and belong to space group  $\text{P}2_12_12_1$ , containing two  $\Delta 100\text{CdaA}$  molecules per asymmetric unit. The phase problem was solved by means of molecular replacement using the monomeric  $\Delta 100\text{CdaA}$  structure of *L. monocytogenes* (PDB code 4RV7) as a search model. The resulting crystal structure of apo-CdaA was determined at 2.0 Å resolution (Table 1). The CdaA monomer is composed of a slightly twisted central  $\beta$ -sheet made up of seven mixed-parallel and antiparallel  $\beta$ -strands ( $\beta 1$ – $\beta 7$ ), flanked on both sides by five  $\alpha$ -helices ( $\alpha 1$ – $\alpha 5$ ) in total (Fig. 2). The two  $\Delta 100\text{CdaA}$  molecules in the asymmetric unit are structurally very similar, as indicated by the root mean square deviation (r.m.s.d.) of 1.19 Å between all C $\alpha$  positions.

The structure of apo-CdaA closely resembles that of CdaA with bound ATP (PDB code 4RV7), as they exhibit an r.m.s.d. of 1.56 Å, but a few differences are seen in a loop region (residues 137–140) and the C-terminal residues. Careful inspection of the difference electron density map revealed a small molecule bound to the surface of one of two CdaA molecules in the asymmetric unit (Fig. S1). This electron density was interpreted as a sucrose molecule originating from the utilized cryo-protectant solution. In the apo-CdaA crystal structure, the active site is accessible from solvent channels; hence, this crystal form of apo-CdaA appears to be suitable for a fragment screen.

### Structure of the CdaA–*c*-di-AMP complex

To gain more insight into the structure and function of CdaA, we also crystallized  $\Delta 100\text{CdaA}$  in the presence of ATP and the cofactor  $\text{Co}^{2+}$ . The obtained crystals belong to a differ-

**Table 1**  
Crystallographic data collection and refinement statistics

	$\Delta 100\text{CdaA}$ with AMP and c-di-AMP	$\Delta 100\text{CdaA}$ -APO	$\Delta 100\text{CdaA}_{\text{Y187A}}$ -APO
<b>Crystallographic data</b>			
Beamline	Petra III-P14, EMBL, Hamburg	Petra III-P14, EMBL, Hamburg	Petra III-P13, EMBL, Hamburg
Wavelength (Å)	0.97620	0.97620	0.97625
Resolution range (Å) <sup>a</sup>	42.27–2.80 (2.90–2.80)	45.89–2.00 (2.10–2.00)	46.49–2.23 (2.33–2.23)
Unique reflections	9,435	24,884	19,512
Redundancy	5.6 (5.7)	7.1 (7.0)	5.8 (4.2)
Completeness (%)	93.0 (95.4)	99.7 (98.5)	97.1 (79.3)
Space group	H32	$P2_12_12_1$	$P2_12_12_1$
a (Å)	121.90	42.69	46.49
b (Å)	121.90	64.67	65.13
c (Å)	141.59	129.75	131.33
R <sub>merge</sub> (%)	10.9 (80.5)	9.4 (119.0)	8.0 (52.0)
I/σ(I)	12.4 (1.9)	13.6 (2.0)	15.6 (2.8)
CC <sub>1/2</sub>	99.8 (72.6)	99.9 (77.2)	99.8 (80.5)
<b>Refinement statistics</b>			
R <sub>work</sub> /R <sub>free</sub>	0.1875/0.2337	0.1858/0.2245	0.1837/0.2258
No. of atoms	2453	2610	2740
Average B-factor (Å <sup>2</sup> )	58.0	47.6	39.8
Root mean square deviation			
Bonds Å	0.003	0.008	0.006
Angles (°)	0.644	1.003	1.258
Ramachandran plot			
Favored (%)	98.05	98.11	98.79
Allowed (%)	1.95	1.57	1.21
Outlier (%)	0.00	0.31	0.00
<b>PDB codes</b>	6HVL	6HVM	6HVN

<sup>a</sup> Values for the data in the highest-resolution shell are shown in parentheses.

ent space group (H32) than the previously determined structure but also contain two CdaA molecules in the asymmetric unit. The newly obtained crystal structure was determined at 2.8 Å resolution. The two CdaA molecules in the asymmetric unit superpose very well, as the r.m.s.d. calculated between all Cα positions amounts to 0.65 Å. Analysis of the protein contact surfaces in the crystal revealed that one of the two CdaA molecules in the asymmetric unit forms a dimer with a symmetry mate related by a crystallographic two-fold axis (Fig. 3A). This CdaA homodimer corresponds to the catalytically active DAC domain dimers seen in the DisA homo-octamer. The calculated r.m.s.d between superimposed CdaA and DisA dimers amounts to 1.72 Å (198 matched Cα positions, Fig. S4). The CdaA–CdaA dimer interface buries about 605 Å<sup>2</sup> of the accessible surface area (7.3%) and is stabilized by six hydrogen bonds and two salt bridges. However, additional interactions between the monomers are mediated by the ligand bound to the active site (see below). Surprisingly, the difference electron density map clearly revealed the presence of a c-di-AMP molecule and two metal ions bound in the active site of the CdaA crystallographic dimer (Fig. S2A). As only ATP and Co<sup>2+</sup> were added to the protein right before it was subjected to crystallization, the c-di-AMP must have formed during or after crystallization droplets were set up. It appears very likely that the bound metal ion is a Co<sup>2+</sup>, as no other catalytic metal cation was present in the crystallization solution. The Co<sup>2+</sup> is coordinated by the phosphate moiety and the carboxylate group of Glu-224 as well as the carboxylate group of Asp-171 and the imidazole ring nitrogen of His-170 of the symmetry-related subunit (Fig. 3, B and C). The metal–oxygen distances of 2.1 Å for Asp-171 and Glu-224 and 2.3 Å for phosphate correspond to distances observed in other proteins containing a Co<sup>2+</sup> ion (17). Elongated distances observed between Co<sup>2+</sup> and the imidazole

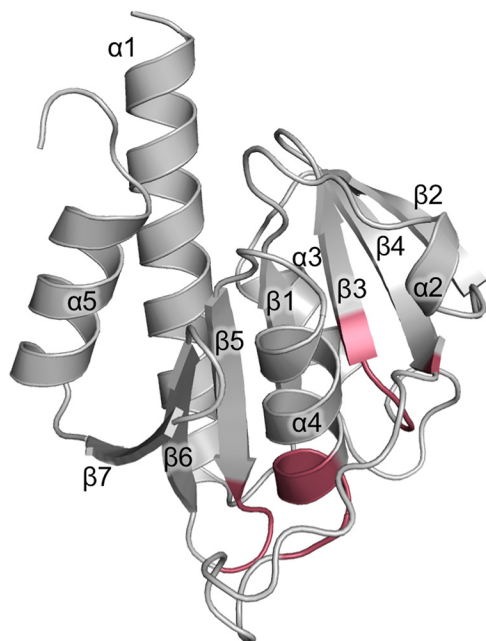
ring of His-170 (3.1 Å) and the c-di-AMP 3'OH group (3.8 Å) indicate that this complex corresponds to the post-catalytic state. To fulfill its catalytic role, the metal ion must be shifted. Only then it can act as Lewis acid to increase the nucleophilicity of the metal-activated 3' hydroxyl group of ATP and enhance the electrophilicity of the phosphorus atom of the adjacent ATP molecule.

The asymmetric unit of the crystal contains a second CdaA molecule that also accommodates a nucleotide bound in the active site but no bound metal ion (Fig. 4). Based on the observed difference omit electron density map (Fig. S2B), the nucleotide was identified as AMP. Because previous crystal structures of DisA and CdaA with bound ATP or 3'-dATP showed well-defined electron density for the β and γ phosphates, it appears likely that the second CdaA molecule indeed has AMP bound, which must have formed out of ATP during the crystallization process. ATP hydrolysis also explains the presence of another difference electron density map peak, which has been interpreted as a free phosphate. This phosphate ion is bound in the vicinity of the c-di-AMP molecule and could potentially mark an exit route of the pyrophosphate molecule on the surface of CdaA.

#### Conformational rearrangements of the active site induced by ligands

The comparison of CdaA in the apo state to CdaA complexed with AMP or c-di-AMP unveils different orientations of the Tyr-187 side chain, which is located in close proximity to the adenine base. In the CdaA apo state, this tyrosine side chain is rotated outward from the active site, leading to an opening of the binding site for the adenine base (Fig. 5). In the monomeric CdaA–AMP complex, the tyrosine is rotated inward at the active site and stacks on the adenine in an almost coplanar orientation. In contrast, in the dimeric c-di-

## Structures of the diadenylate cyclase CdaA



**Figure 2. Crystal structure of  $\Delta 100$ CdaA in the apo state, refined at 2 Å resolution.** The fold of the CdaA DAC domain consists of seven  $\beta$ -strands forming a central  $\beta$ -sheet surrounded by five  $\alpha$ -helices. The positions of residues forming the active site are highlighted in red.

AMP complex, the tyrosine side chain is flipped outward, as the Thr-202 side chain of the other subunit packs against the adenine ring (Fig. 3B).

To investigate whether Tyr-187 plays an important role in *c*-di-AMP formation, a Y187A mutant was generated. This mutation led to a significant reduction (about 80%) in activity, confirming the functional impact of Tyr-187 (Fig. 1B). To exclude that the Y187A mutation perturbed the fold of CdaA, the crystal structure of  $\Delta 100$ CdaA\_Y187A was determined as well (Table 1). Comparison with the structure of WT  $\Delta 100$ CdaA demonstrates no structural changes caused by the mutation.

### Discussion

Synthesis of *c*-di-AMP requires dimerization and proper orientation of two DAC domains, each with one ATP bound and accompanied by the metal ion cofactor. In DisA, this is achieved permanently by the homo-octameric oligomerization state (13). The first structure of CdaA of *L. monocytogenes* showed that the DAC domain crystallized as a monomer even though ATP was bound to the active site (15). However, for the previous study and this one, a truncated CdaA was used. So far, the influence of the missing transmembrane domain on oligomerization and catalytic activity is unknown.

Here, a new crystal form of CdaA was obtained that contains two CdaA molecules with different nucleotides bound. One CdaA molecule forms a catalytically active dimer with a symmetry mate in the crystal. This dimer contains a *c*-di-AMP molecule and two metal ions in the active site; hence, it closely resembles the dimer arrangement of DAC domains seen in DisA (Fig. S4).

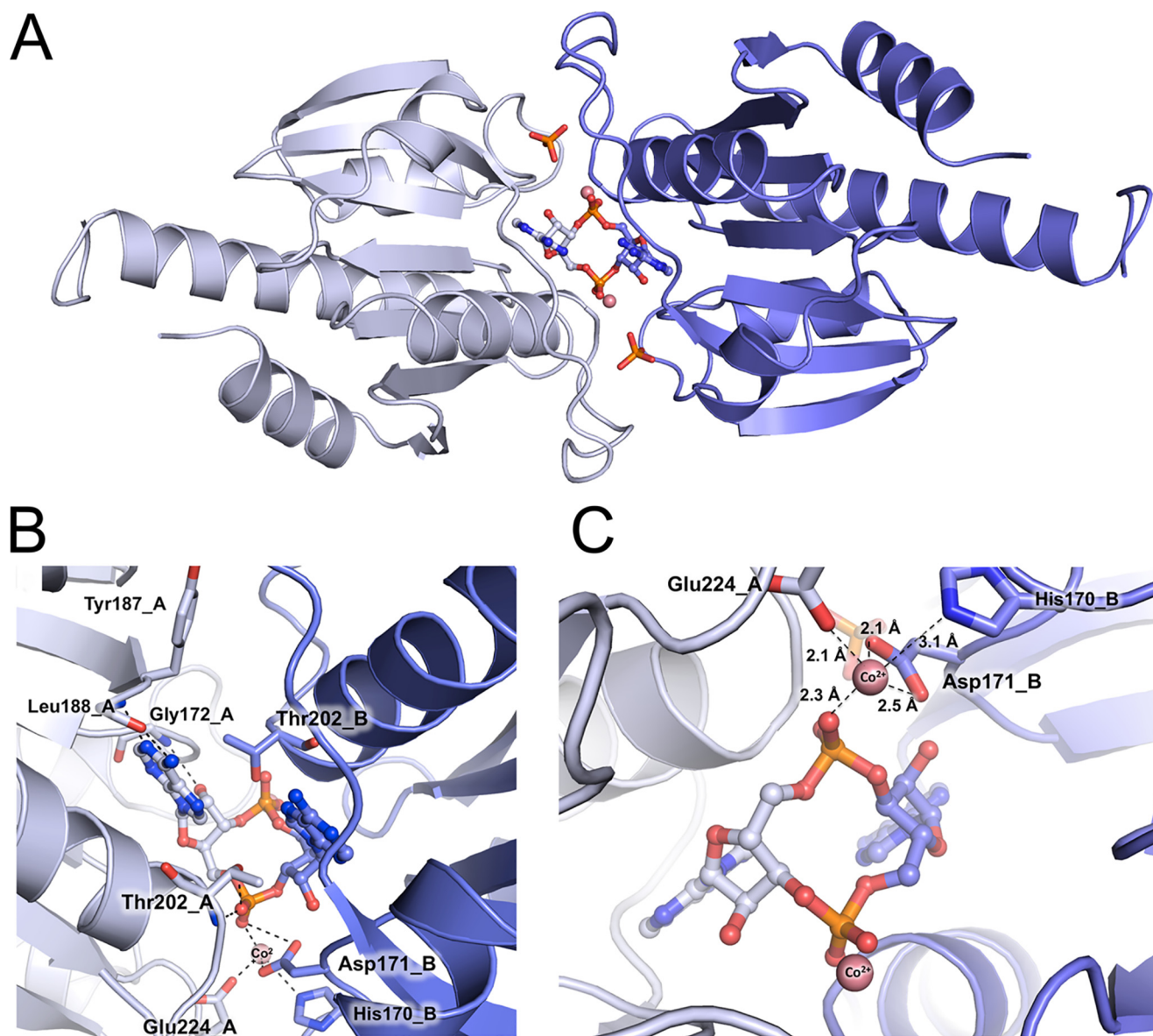
The *c*-di-AMP must have been formed during crystallization, as only ATP and  $\text{Co}^{2+}$  ions were added to CdaA. This

complex corresponds to the enzyme–product complex, which is supposed to have a lower stability. However, *c*-di-AMP mediates multiple contacts between the monomers, increasing the interaction surface area between the monomers, and the catalytically active dimer appears to be caught in the crystalline lattice.

Conserved active-site residues of DisA and CdaA directly involved in substrate binding and catalysis have been identified previously (15, 18). Each of the mutations in DisA (D75N, R130A, RHR (108–110)AAA, T107V+T111V), and in CdaA (D171N, G172A, and T202N) led to a reduction or complete loss of enzymatic activity. However, by analyzing the structure of the monomeric CdaA in the asymmetric unit with bound AMP, we realized that the Tyr-187 side chain might also be involved in substrate binding, as it stacks on the adenine ring, but it is rotated outward in the structure of apo CdaA (Fig. 5). Hence, it appears likely that, upon binding of ATP to monomeric CdaA, Tyr-187 rotates toward the adenine moiety and locks the ATP in the active site by a  $\pi$ – $\pi$  stacking interaction, as observed for many other ATP binding proteins (19). However, upon CdaA dimerization, the tyrosine side chain is replaced by the side chain of Thr-202 of the other subunit, which then stabilizes the bound ATP. The replacement of the Tyr by a side chain from the other subunit might facilitate product release after catalysis, as, upon dimer dissociation, the product can be released more easily. By mutation of Tyr-187 to Ala, which strongly reduced the activity *in vitro*, we demonstrate that Tyr-187 indeed plays an essential role in *c*-di-AMP formation by CdaA. Notably, this Tyr-187 is conserved in most CdaA enzymes but not in other DAC proteins, like DisA, suggesting a slightly different mechanism of substrate binding between different classes of DACs.

A remarkable difference between DisA and CdaA concerns the metal ion specificity in the catalytic center. Although DisA appears to be active with  $\text{Mg}^{2+}$  and  $\text{Mn}^{2+}$  (20), CdaA is not active in the presence of  $\text{Mg}^{2+}$ . Such unexpected differences in metal ion preferences have also been observed for other protein families, e.g. the metal-dependent serine/threonine phosphoprotein phosphatase family (21). Because the catalytic mechanism of phosphoprotein phosphatase enzymes as well as that of DAC proteins does not require the redox potential of  $\text{Mn}^{2+}$  to carry out the catalyzed reaction, it is not clear why some members of the DAC family would prefer  $\text{Mn}^{2+}$  or other divalent cations over  $\text{Mg}^{2+}$  (22). Hence, the observed strict dependence of CdaA on  $\text{Mn}^{2+}$  or  $\text{Co}^{2+}$  ions raised questions concerning its structural basis. The observed metal ion dependence is most likely related to different chemical properties of the cation, e.g. ionic radius, and to the amino acid composition of the active site.

Comparison of the DisA and CdaA structures reveals significant differences in metal ion coordination. In the catalytically active dimer of DisA with bound ATP, the  $\text{Mg}^{2+}$  ion is coordinated by three phosphate groups and the Asp carboxylate. In CdaA, more protein residues contribute to metal binding, resulting in a more crowded active site. In addition to Asp-171 (which corresponds to Asp-75 of DisA), the side chains of Glu-224 and His-170 coordinate the  $\text{Co}^{2+}$  ion. These two residues are not structurally conserved in DisA, as



**Figure 3. The active site of dimeric CdaA with bound c-di-AMP.** A, the catalytically active  $\Delta 100$ CdaA homodimer is depicted as a *cartoon*, and the bound reaction product c-di-AMP is shown as *balls and sticks* (carbon in *pale blue* and *blue*, phosphate in *orange*, oxygen in *red*, and nitrogen in *dark blue*). The two  $\Delta 100$ CdaA monomers are colored according to the c-di-AMP in *pale blue* and *blue*, respectively.  $\text{Co}^{2+}$  ions are depicted as *pale red spheres*. B, detailed view of the CdaA active site. Amino acids involved in binding the c-di-AMP molecule (colored and depicted as in A) are shown as *sticks* (carbon in *pale blue* and *blue*, oxygen in *red*, and nitrogen in *dark blue*). The  $\text{Co}^{2+}$  ions are colored and depicted according to A. For simplicity, only one half of the two-fold symmetric CdaA active site is shown. C, detailed view of the  $\text{Co}^{2+}$  binding site and its coordination sphere.

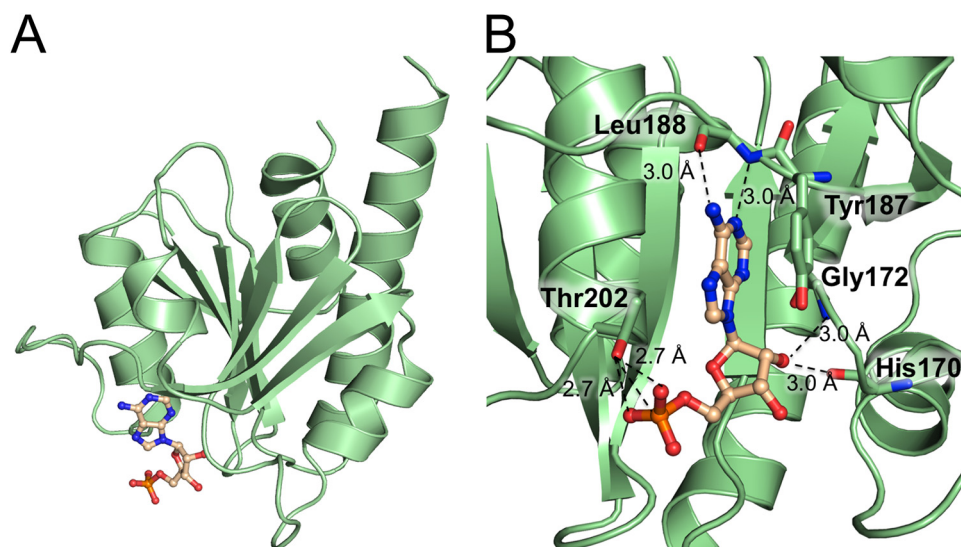
there is an Arg instead of the Glu and a Met instead of the His. The side chains of both Arg and Met are rotated outward from the metal binding site, making it less crowded. Hence, the major difference appears to be presence of the His. Although  $\text{Mg}^{2+}$  strongly prefers coordination by Asp and Glu, the transition metal ions  $\text{Mn}^{2+}$  and  $\text{Co}^{2+}$  are bound as well by His, as deduced from analysis of all metal binding sites in known protein structures (23).

The reason for this difference between DisA and CdaA is most likely related to the fact that DisA contains stably associated, catalytically active dimers, whereas, for CdaA, the catalytic dimer might just exist transiently. Therefore, in DisA, Asp-171, which belongs to the second DAC domain, is sufficient for binding the substrate ATP and the metal ion. In

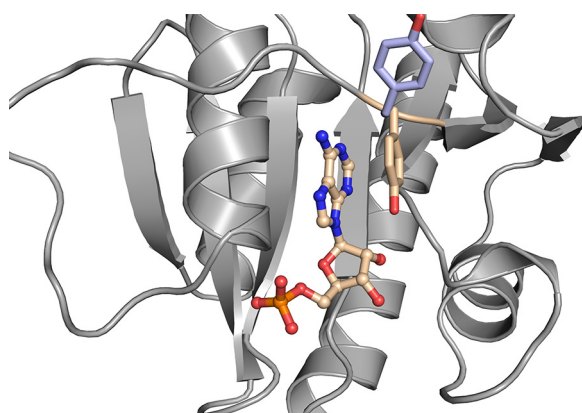
CdaA, ATP and the metal ion are initially bound to the monomeric DAC domain by the Glu-224 side chain, and solvent molecules complete the metal coordination sphere. Upon formation of the catalytically active dimer, the metal ion needs to be slightly repositioned to be further coordinated by His-170 and Asp-171, provided by the second monomer (Fig. S3).

Recently, CdaA from *S. aureus* was characterized structurally and biochemically (24). Surprisingly, in contrast to *L. monocytogenes* CdaA, the *S. aureus* CdaA homolog shows activity not only in the presence of transition metal ions but also in the presence of  $\text{Mg}^{2+}$ . Comparison of the active sites of both available CdaA structures unveils identical positioning of the amino acids directly involved in c-di-AMP and

## Structures of the diadenylate cyclase CdaA



**Figure 4. Structure of the CdaA-AMP complex.** A, CdaA monomer (cartoon, pale green) with a bound AMP molecule depicted as balls and sticks. B, a detailed view of the active site showing the amino acids (sticks, carbon in pale green, oxygen in red, and nitrogen in blue) involved in AMP binding. The bound AMP is shown as a ball-and-stick model (carbon in wheat, phosphate in orange, oxygen in red, and nitrogen in dark blue).



**Figure 5. Conformational switch of Tyr-187 during c-di-AMP synthesis.** For convenience, only the  $\Delta 100\text{CdaA}$  monomer (gray) with bound AMP (carbon in wheat, phosphate in orange, oxygen in red, and nitrogen in dark blue) and Tyr-187 (wheat) is shown. AMP is depicted as a ball-and-stick model. The side chain of Tyr-187 (pale blue) of the  $\Delta 100\text{CdaA}$ -c-di-AMP complex structure is superimposed. Upon ATP binding to monomeric CdaA, Tyr-187 stacks parallel on the adenine base ( $\pi$ - $\pi$  interaction) and stabilizes the protein-substrate complex. Upon homodimer formation, the side chain of Tyr-187 rotates outwards as it is replaced by the Thr-202 side chain of the second monomer in the catalytically active homodimer.

metal ion coordination (Fig. S5). The largest structural differences can be observed for N- and C-terminal  $\alpha$ -helices and a loop connecting  $\beta$ -strand 4 and  $\alpha$ -helix 4. The latter is located in close proximity to the phosphate moiety of ATP and could indirectly alter the metal binding preferences or even metal catalytic efficiency. This could serve as a potential explanation for the metal ion promiscuity of CdaAs showing strict conservation of three residues (His-170, Asp-171, and Glu-224) directly involved in metal binding, as revealed by sequence alignment of ten bacterial CdaAs (Fig. S6). The chemical properties of these three residues are most likely the structural basis for the observed metal ion promiscuity of CdaAs, which has also been observed for other enzymes, e.g. mannosylglycerate synthase (25). The observed capability of utilizing several ions by one enzyme is still one

of many not well-understood marvels of enzymology that require further investigation.

### Experimental procedures

#### Bacterial strains and growth conditions

For cloning procedures and protein overexpression, *Escherichia coli* strains DH5 $\alpha$  and BL21(DE3) were used. The *E. coli* strains were cultivated in 2xYT ((trypton 1.6% (w/v), yeast extract 1.0% (w/v), NaCl 0.5% (w/v)) medium, whereas transformed cells were selected on lysogeny broth-medium plates containing ampicillin (100  $\mu\text{g/ml}$ ).

#### Plasmid construction

For purification, the DAC-type CdaA was equipped with a GST tag. CdaA is known to be a transmembrane protein. The  $\Delta 300\text{cdaA}$  allele, which lacks the TM domain, was amplified using the primer pairs JH004 forward (5'-CCGGATCCTATGATCAAGAATTGAGCG-3')/JH005 reverse (5' GGCTCGAGTCATTCGCTTTTGCCTCCTTCC-3'). As a template, the plasmid pBP33 was used (15). The resulting PCR products were cloned in the pGEX-6P-1 (GE Healthcare) expression vector using the restriction sites XhoI and BamHI, leading to plasmid pGEXpBP33, which encodes for the truncated  $\Delta 100\text{CdaA}$  protein with an N-terminal GST tag.

#### Site-directed mutagenesis

$\Delta 100\text{CdaA}$  mutants were generated with site-directed mutagenesis to identify amino acid residues that have an important function in the catalytic reaction mechanism. The CdaA mutant Y187A was created by PCR using the mutagenesis primer pairs JH\_Y187A\_forward (5'-CAGCAAGTGCCTTGCCA CTTTCAGATAGCCCGTTCTTATCCAAAGAAC-3') and JH\_Y187A\_reverse (5'-GTGGCAAGGCACTTGCTGCCGATGC AATTTCGTTTCTTTAATAATAACTGC-3'), resulting in the plasmid encoding the truncated mutant variant  $\Delta 100\text{CdaA}_{\text{Y187A}}$ .

### Protein expression and purification

*E. coli* BL21(DE3) was used for expression of the fusion protein GST- $\Delta$ 100CdaA. The cells were grown in 1 liter of 2xYT medium at 37 °C. Protein expression was induced after the culture reached an  $A_{600}$  of  $\sim 0.6$  by addition of 1 mM isopropyl 1-thio- $\beta$ -D-galactopyranoside and incubated at 16 °C for 18 h. After harvesting and subsequent to cell disruption with a microfluidizer (M-110S Microfluidizer, Microfluidics) and centrifugation at  $15,600 \times g$  for 30 min to remove cell debris, the lysate was loaded onto a GSH-Sepharose column (GE Healthcare) in 300 mM NaCl, 20 mM Tris/HCl (pH 7.5), and 10 mM EDTA. The target protein GST- $\Delta$ 100CdaA was eluted from the column with 40 mM reduced GSH. The eluate was incubated overnight with PreScission protease (1:100 (w/w)) in cellulose tubing placed in dialysis buffer (100 mM NaCl and 20 mM Tris/HCl (pH 7.5)) at 4 °C to remove the high GSH concentration and to dissect the GST tag from  $\Delta$ 100CdaA. To remove the cleaved-off tag from the truncated CdaA, a second GSH-Sepharose purification step was included.

### Crystallization and cryoprotection

For crystallization, the sitting-drop vapor diffusion method was applied. Initial crystallization trials were performed at 20 °C using  $\Delta$ 100CdaA at a concentration of 4.0 mg/ml supplemented with 500  $\mu$ M CoCl<sub>2</sub> and 500  $\mu$ M ATP. Rectangular crystals grew after approximately 48 h in a 2- $\mu$ l droplet composed of the aforementioned protein solution mixed with reservoir in a 1:1 ratio. The reservoir was composed of 0.2 M Ca(CH<sub>3</sub>COO)<sub>2</sub>, 0.1 M Na-HEPES (pH 7.5), and 10% (w/v) PEG8000. Crystals were cryoprotected by soaking them in a reservoir solution supplemented with 25% PEG8000.

For crystallization of the apo form and the Y187A variant of  $\Delta$ 100CdaA, a protein concentration of 6.0 mg/ml was used, keeping the 2- $\mu$ l droplet size and 1:1 protein-to-reservoir ratio. To facilitate crystal growth, microseeding was performed in combination with small alterations of NaCl concentration. Thin crystal plates were obtained after approximately 18 h in a salt concentration ranging between 3.7–4.5 M NaCl and 0.1 M Na-HEPES (pH 8.5). Crystals were cryoprotected by soaking them in a saturated sucrose solution obtained by solubilizing sucrose in reservoir solution.

### X-ray data collection and processing

Diffraction images were collected at PETRA III EMBL beamlines P13 and P14 (DESY, Hamburg, Germany) and processed with the XDS package (26, 27). Data collection and processing statistics are summarized in Table 1. A trigonal lattice with unit cell parameters of  $a = b = 121.90$  Å,  $c = 141.59$  Å was determined for the crystals containing the CdaA–c-di-AMP complex. Cell content analysis indicated the presence of two CdaA molecules occupying the asymmetric unit ( $V_m = 2.89$  Å<sup>3</sup>/Da, corresponding solvent content of 57.4%). The crystals of apo CdaA and the Y187A mutant exhibited an orthorhombic lattice and the unit cell parameters of  $a = 42.96$  Å,  $b = 64.67$  Å,  $c = 129.75$  Å and  $a = 46.49$  Å,  $b = 65.13$  Å,  $c = 131.33$  Å, respectively. The Matthews coefficient ( $V_m = 2.55$  Å<sup>3</sup>/Da, corre-

sponding solvent content of 51.78%) implicates two molecules occupying the asymmetric unit.

### Structure determination and refinement

The crystallographic phase problem was solved by molecular replacement with PHASER (28) using the structure of the DAC  $\Delta$ 100CdaA from *L. monocytogenes* (PDB code 4RV7) as a search model. Manual model building was performed with Coot (29), and the structure was refined with Refmac (30) and PHENIX (31). To monitor the refinement progress using  $R_{free}$ , 5% of the reflections were selected randomly and excluded from refinement. During the refinement process, the coordination distance for the Co<sup>2+</sup> ion in the ligand-bound structure was restrained to 2.1 Å for the Glu-224 and Asp-171 side chains and to 2.3 Å for the cyclic di-AMP phosphate. The final structure of the CdaA–c-di-AMP complex was refined at a resolution of 2.8 Å to  $R_{work}$  of 18.7% and  $R_{free}$  of 23.4%. The apo-CdaA was refined at a resolution of 2.0 Å to  $R_{work}$  of 18.6% and  $R_{free}$  of 22.5%. The structure of the Y187A mutant was determined at 2.32 Å resolution and refined to 17.5% and 22.0% for  $R_{work}$  and  $R_{free}$ , respectively. Protein contact areas in the crystals were analyzed using “Protein interfaces, surfaces and assemblies” services at the European Bioinformatics Institute using standard settings (32).

### In vitro DAC activity assay

Diadenylate cyclase activity was measured with a quantitative fluorescence assay based on an increased fluorescence signal because of the specific interaction of the fluorescent dye coralyne with c-di-AMP (16). A 200- $\mu$ l reaction mixture contained 40 mM Tris/HCl (pH 7.5), 100 mM NaCl, 10 mM XCl<sub>2</sub> (X = Mg, Co, Mn, or Ca) and 100  $\mu$ M ATP. The reaction was started by addition of 10  $\mu$ M  $\Delta$ 100CdaA and incubated for 1 h at 37 °C. To stop the reaction, the reaction mixture was boiled for 5 min and centrifuged for another 5 min at  $13,400 \times g$  to remove the precipitated protein. For quantification of the synthesized c-di-AMP, 250 mM KBr and 10  $\mu$ M coralyne were added to the reaction mixture. After incubating the samples for 30 min in the dark, c-di-AMP concentration was determined by excitation at a wavelength of 420 nm and measuring the fluorescence emission at a wavelength of 475 nm using a microplate reader (Victor Nivo multimode microplate reader, PerkinElmer Life Sciences).

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