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Enhanced Photoluminescence of Solution-Exfoliated Transition Metal Dichalcogenides by Laser Etching

Mark A. Bissett,^{*,†}⁶ Andrew G. Hattle,[‡] Alexander J. Marsden,[†] Ian A. Kinloch,[†] and Robert A. W. Dryfe[‡]⁶

[†]School of Materials and [‡]School of Chemistry, University of Manchester, Oxford Road, M13 9PL Manchester, U.K.

Supporting Information

ABSTRACT: Using a conventional Raman experimental apparatus, we demonstrate that the photoluminescent (PL) yield from ultrasonication-exfoliated transition metal dichalcogenides (TMDs) (MoS_2 and WS_2) can be increased by up to 8-fold by means of a laser etching procedure. This laser etching process allows us to controllably pattern and reduce the number of layers of the solution-exfoliated material, overcoming the key drawback to solvent-based exfoliation of two-dimensional (2D) semiconducting materials for applications in optoelectronics. The successful laser thinning of the exfoliated 2D crystals was investigated systematically by changes in both Raman and PL spectra. A simple proof-of-principle of the scalability of this laser etching technique for solution-exfoliated TMD crystals was also demonstrated. As well as being applicable for individual materials, it is also possible to use this simple laser etching technique to investigate the structure of solution-generated van der Waals heterostructures, consisting of layers of both MoS_2 and WS_2 .



INTRODUCTION

Research into two-dimensional (2D) materials has continued to gain interest, first with graphene and then with other layered materials, such as the semiconducting transition metal dichalcogenides (TMDs), molybdenum disulfide (MoS₂), and tungsten disulfide $(WS_2)^{.1-4}$ TMDs are of particular interest due to their unique electronic structure, and they exhibit a significant change in the band structure when reduced in thickness from bulk (indirect gap) to a monolayer (direct gap).⁵⁻⁷ The transition to a direct gap semiconductor means that photons with an energy equal to the band gap can be easily absorbed or emitted, whereas the indirect band gap material requires an additional phonon absorption or emission to compensate for the difference in the momentum, making absorption much less efficient.8 This change in electronic structure can typically be experimentally detected by a large increase in the photoluminescent (PL) quantum yield due to the increased efficiency of the direct band gap monolayer material.^{9,10} This unique optical property is particularly important for photonic and optoelectronic applications including photovoltaics, light-emitting diodes, photodetectors, and displays.^{6,7,11} Because of the atomically thin nature, it may also be possible to develop flexible and transparent optoelectronics.

Exfoliated monolayer TMDs can be produced by several methods including micromechanical exfoliation,¹ chemical vapor deposition (CVD),^{12–14} chemical exfoliation,¹⁵ electrochemical lithium intercalation,^{16–18} and solvent-assisted ultrasonication.^{19,20} Mechanical exfoliation produces the highestquality crystals, with lateral dimensions of over 50 μ m possible; however, the method is expensive, time consuming, and cannot be easily scaled for commercial applications. CVD growth can also produce high-quality large-area films of TMDs and can be scaled with continued progress, but it is also expensive and requires high temperatures for the growth, which limits the available substrates. Solvent exfoliation by ultrasonication, in contrast, is the only current method for economical production of large volumes of dispersed TMD sheets; however, the biggest drawback to solvent exfoliation is the reduced quality of the subsequent flakes due to the small lateral size, residual solvent, and a low monolayer content. Because of the low monolayer content, combined with the effect of contamination from the residual solvent molecules, very few studies have reported the PL response from ultrasonically exfoliated TMD flakes. To increase the applicability of the ultrasonic production method for use in optoelectronic applications, some form of post-processing is required to tune the layered structure.

It has previously been demonstrated that both mechanically exfoliated²¹ and CVD-grown films^{22,23} of MoS_2 can be thinned by laser etching in a controllable and reproducible manner using a simple laser setup, allowing for control over the number of layers as well as patterning into desired shapes. A similar process of laser etching has also previously been applied to other 2D materials such as graphene²⁴ and black phosphorus.^{25,26} However, its application to solution-exfoliated TMDs would allow for a truly scalable production of efficient

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Figure 1. (a) Schematic showing the laser etching of an assembled stack of solution-exfoliated TMD layers. (b-c) Optical images showing a patch of agglomerated WS₂ flakes deposited on a SiO₂ wafer; the area marked by the box shows the obvious change in thickness, and consequently optical contrast after laser etching is performed. (d) PL spectra of the same area before (black) and after (red) laser etching showing the significant increase in signal intensity.

optoelectronic devices. The use of a laser etching process to perform the thinning and patterning of the sample has the benefit of avoiding the need for chemical photoresists or masks that can contaminate the flakes and alter the electronic structure. The mechanism behind the etching has been attributed to thermal sublimation of the upper TMD layers, with the heat being poorly dissipated through the structure due to weak van der Waals cohesive forces.^{21,22} Because of this heat sink effect, the bottommost lasers directly in contact with the Si/SiO₂ wafer become increasingly difficult to remove and require an increased laser dosage when the laser irradiation is performed under ambient conditions. In addition to thermal sublimation of the crystals, there have been some reports of partial oxidation of the material (e.g., conversion of MoS₂ into MoO_3) due to the etching being performed under ambient conditions.²² Chemical doping can also occur due to the freshly etched TMD layers being exposed to ambient atmospheric conditions (e.g., O2, H2O, and N2). Previously, it has been observed that laser irradiation under ambient conditions leads to oxygen and water doping of the MoS₂ film and subsequent improvement of the photocurrent response.²² Doping of the ntype semiconducting TMDs (e.g., MoS_2 and WS_2) by exposure to atmospherics molecules, such as O₂ and H₂O, is known to contribute to the enhancement of the PL intensity due to electron withdrawal by the dopant molecules stabilizing excitons and providing a new radiative recombination pathway.^{27,28} This combined effect of chemical doping and morphological thinning can lead to an effective engineering of the electronic structure as well as providing tuning of the thickness.²⁹ This is of particular importance as chemical doping can be used to effectively tune the optoelectronic properties of the exfoliated flakes.

Heterostructures of different semiconducting 2D materials are also of great interest because of the ability to modify the electronic structure, thus introducing novel properties, and because of the fundamental interactions between each of the layers stacked together by van der Waals forces.^{30–32} Owing to the close interaction between each of the exfoliated layers, it is possible for excited electrons and holes to transfer between the respective layers. These heterostructures have been demonstrated experimentally using mechanically exfoliated flakes manually stacked on top of one another;^{33–35} however, this process is time consuming and lacks scalability. In addition to manual stacking, CVD-grown films grown on top of one another have been realized;^{36–38} this technique shows promise but remains difficult to control and has a limited number of suitable growth substrates.

Here, we demonstrate that the PL intensity of solventexfoliated dispersions of both MoS₂ and the similarly structured WS_2 can be significantly increased (up to $8\times$), overcoming the key drawback of solvent-based exfoliation for applications in optoelectronics, by reducing the number of layers in a scalable manner. The optical properties of the flakes before and after etching are thoroughly characterized using Raman and PL spectroscopy, and the morphology of the flakes is analyzed using scanning electron microscopy (SEM). The laser etching of the solution-exfoliated TMD flakes is easily scalable as demonstrated over an arbitrary 250 μ m × 250 μ m area and can be applied to a wide variety of similarly structured 2D layered materials. We also investigate the properties of mixed dispersions of exfoliated MoS_2 and WS_2 flakes to create heterostructures. After depositing onto a silicon substrate, mixed heterostructures consisting of randomly stacked MoS₂ and WS₂ flakes are formed with differing thicknesses and stacking orders. The resultant stacks can then be modified through the use of laser etching, and the resultant 3D layered structure can be characterized.



Figure 2. (a) SEM image showing deposited WS₂ nanoflakes after laser etching of a 20 μ m × 20 μ m area. (b) Map showing the Raman intensity of the WS₂ E_{2g} intensity over the same area showing the reduced signal intensity where etching has occurred. (c) Map showing the normalized PL intensity of the same area where the increased signal from inside the etched area is visible. PL spectra of the WS₂ (d) and MoS₂ (e) flakes with repeated laser etching. The as-deposited flakes are shown in green, followed by one cycle of laser etching in red and a second cycle of laser etching in black. The spectra in each figure are fitted with Lorentzian components, with the peak center labeled along with the full width at half-maximum (fwhm) (meV) in parentheses. The peak intensity has been normalized to the intensity of the Raman A_{1g} peak in each case. Insets show the optical image of the individual flakes before laser etching; the scale bar in each inset is 2 μ m.

RESULTS AND DISCUSSION

After the formation of solvent-stabilized dispersions, the TMD flakes can then be deposited onto Si/SiO₂ wafers to act as a supporting substrate. The deposited flakes can form aggregates with lateral sizes of several microns, as seen in the typical optical image of the WS₂ flakes in Figure 1b, wherein repeated deposition was used to ensure that large-scale aggregates were clearly visible. A confocal Raman setup was used as both the etching laser source and the characterization tool, allowing rapid characterization of the same areas before and after etching. Etching was achieved by rastering the laser with a high incident power density over a designated area as seen schematically in Figure 1a. By controlling the exposure time, laser power, and scanning over designated areas in repeated cycles, it was possible to etch away the TMD flakes. After etching there is a clear change in the optical contrast of the deposited flakes, as seen in Figure 1c, indicating a reduction in the thickness of the flakes, and this can be used to approximate the thickness of the flakes.³⁹ When we compare the PL spectrum of the aggregated flakes before etching to that of the same area after the laser etching, as seen in Figure 1d, we see that there is a significant improvement in the signal intensity

due to the reduced thickness of the flakes. Because of the large variation in the initial thickness of the aggregated flakes, some areas are reduced in thickness more than others. This is due to the mechanism of the laser etching, as the incoming laser light is absorbed and converted to heat, and the poor interlayer coupling causes the uppermost layers to be sublimed. The thinner flakes are in better thermal contact with the underlying substrate and can more strongly resist the localized heating.^{21,40} Thus, the rate of etching is more pronounced on the thicker flakes compared to that of the thinner flakes for an equal exposure time.

After the ultrasonic exfoliation, the average MoS₂ and WS₂ flake dimensions are on the order of 200–300 nm, making them smaller than the size of the laser spot (~1 μ m); this increases the exposure of the more reactive edges to the laser and may lead to the selective removal of highly defective edge structures. A similar effect has been observed previously for graphene films, which showed that, contrary to expectations, after laser etching the defect-related D band decreased in intensity due to selective removal of the highly reactive edges and defect sites.²⁴



Figure 3. (a) Comparison of the Raman spectra of the bulk WS₂ powder (blue) and the solution-exfoliated WS₂ flakes (black) after deposition onto SiO₂, along with Lorentzian fitting and labeling according to literature peak assignments.⁴⁴ (b) Raman spectra of the same WS₂ flake after repeated laser etching. The spectra are normalized to the intensity of the A_{1g} peak. (c) Comparison of the Raman spectra of the bulk MOS₂ powder (blue) and the solution-exfoliated MOS₂ flakes (black) after deposition onto SiO₂, along with Lorentzian fitting and labeling according to literature peak assignments.⁴⁷ (d) Raman spectra of the same MOS₂ flake after repeated laser etching. The spectra are normalized to the intensity of the A_{1g} peak.

To further investigate the applicability of this laser etching procedure on solution-exfoliated TMD flakes, a thick coating of exfoliated WS₂ was deposited onto a Si/SiO₂ wafer and laser etched before analysis. Figure 2a shows an SEM image of the WS₂ flakes aggregated on the substrate after laser etching. The area ($\sim 20 \ \mu m \times 20 \ \mu m$) where the laser was rastered is clearly noticeable, indicating that this technique can be applied to large-scale samples as well as individual nanocrystals with a size of several hundred nanometers. By measuring the Raman spectrum across every point over this area, we can create a map of the E_{2σ} intensity and compare how this changes across the etched area, as seen in Figure 2b. Clearly, there is reduced signal intensity from inside the laser-etched area, indicating the reduced thickness of the flakes. Interestingly, no peaks associated with tungsten oxides, WO₃⁴¹ (or molybdenum oxide, MoO_3 , in the case of MoS_2), were observed for the thin flakes, indicating that the etching process does not destroy the WS₂ or MoS₂ structure. However, in some areas of the thicker flakes, oxide peaks were occasionally detected (see Figure S2). The PL intensity was also mapped across the etched area, as seen in Figure 2c, and we see again an increased signal intensity within the etched areas. These results demonstrate that laser

etching of solution-exfoliated TMDs can be performed over a designated area of an arbitrary size.

To further investigate this systematically, low-concentration dispersions were used to create small aggregates that were etched repeatedly, and the PL spectra were measured. The PL spectrum, normalized to the height of the A1g Raman peak, of both the solution-exfoliated WS₂ and MoS₂ flakes, after dilution and deposition onto a silicon wafer, are shown in green in Figure 2d. Optical images showing the individual flakes as opposed to the large-scale aggregates are shown in the insets. The monolayer WS_2 PL emission typically contains a single peak, originating from the A exciton at ~ 1.95 eV.^{8,42} The asdeposited WS₂ flakes shown in Figure 2d show a single lowintensity peak in the PL spectrum centered at 648 nm (1.91 eV), which corresponds to the energy expected for an approximately three-layer-thick WS_2 flake. $\overset{42}{}$ The fwhm of the PL peak can also provide information regarding the electronic structure of the TMD flakes. Typically, the fwhm of monolayer WS₂ is ~50-75 meV and that of MoS₂ is ~90 meV.^{8,27,42,43} However, this value can vary depending on the doping of the flake and the localized strain or defects. The calculated fwhm (in meV) is shown in parentheses in Figure 2d,e for both the WS₂ and the MoS₂ flakes. For the as-deposited WS₂ flakes (shown in green, Figure 2d), the fwhm is much larger than expected for the monolayer at ~150 meV, again indicative of \sim 3–5 layers. However, after laser etching of the WS₂ flakes there is a significant change in the PL response. After the first cycle of laser etching (shown in red), we notice a significant increase in the PL intensity, as may be expected due to the combination of reduced flake thickness combined with chemical doping by freshly exposing the etched surface to an ambient atmosphere, as mentioned previously.²⁷ The increase in the PL intensity after the first laser-etching cycle is by a factor of 5, and after repeated etching (shown in black), it increases further by a factor of 8 compared to that of the asdeposited flakes. This is a significant increase in the photoresponse of the solution-exfoliated flakes and could be valuable for use in applications in which the solution-exfoliated WS₂ flakes can be mass-produced and easily deposited as films on surfaces, followed by laser etching to increase the photoresponse as demonstrated previously for CVD-grown films.²² As mentioned previously, as the laser etching removes the topmost WS₂ layers, it freshly exposes the underlying layer, providing a highly reactive surface that is rapidly passivated by atmospheric oxygen and water; these dopants increase the free carrier concentration of the WS₂ flakes, which can also increase the PL intensity. The width of the peaks also narrows significantly with laser etching, decreasing from ~150 meV for the as-deposited flakes, to ~70 meV after the first etching cycle, and finally to \sim 66 meV after the second etching cycle, in agreement with the reduced number of layers and indicating the high quality of the thinned flakes.

To verify that the laser etching treatment is indeed reducing the thickness of the flakes, atomic force microscopy (AFM) was used to statistically measure the thickness of a number of individual WS₂ flakes before and after laser etching (see Figure S3). These flakes were prepared in the identical manner as described previously and deposited onto Si/SiO₂ wafers for analysis. It was found that after a single cycle of laser etching the average thickness of the flakes was decreased significantly, in agreement with the spectroscopic results. Also noticeable was the exposure of terrace-like structures within the thicker flakes. This indicates that the etching mechanism removes the uppermost layers preferentially before the lower layers.

A similar trend in PL response is observed for the MoS₂ flake, as seen in Figure 2e. Initially, the as-deposited MoS₂ flakes, shown in green, consist of two separate peaks referred to as the A and B excitons.^{9,10} Similar to the WS₂ flakes, the asdeposited PL response of the solution-exfoliated MoS₂ exhibits a low intensity and very broad peaks, due to the thickness of the aggregated flakes. The literature values for the PL response for the monolayer MoS_2 flakes are 670 nm (~1.85 eV) for the A exciton and 620 nm (~2.00 eV) separated by ~145 meV due to spin-orbit band splitting.^{5,9,10,15} The PL response of the asdeposited flakes in Figure 2e matches fairly closely to the expected peak positions. After laser etching, the PL response of the MoS₂ flakes increases in intensity and narrows in width, indicating the reduced thickness of the flakes. After etching, the PL intensity increases initially by a factor of 1.5 and then by a factor of 2 compared to that of the as-deposited flakes. This is a lower enhancement of the PL intensity when compared to that of the WS₂ flakes and indicates that the PL enhancement is more effective in the case of the WS₂ flakes. This may also be related to the lower inherent PL intensity achieved by the MoS₂

flakes compared to that by the WS_2 flakes, making it more difficult to resolve the effects of atmospheric doping.

To systematically analyze the etching process on the solventexfoliated flakes, we can also look at the changes to the Raman spectra. Figure 3 shows the Raman spectra of the bulk powder and the solution-exfoliated WS_2 (a, b) and MoS_2 (c, d) flakes deposited onto the SiO₂ wafers after solvent removal. Lowconcentration dispersions were used to minimize aggregation and allow the investigation of individual flakes. In Figure 3a, the Raman spectrum of WS₂ is shown with each of the peaks fitted with Lorentzian curves and labeled on the basis of literature peak assignments. The most significant peaks are the first-order A_{1g} (out-of-plane phonon) and E^{1}_{2g} (in-plane phonon) peaks that have been studied in detail previously; we also observe a rich variety of second-order peaks of WS2, which are labeled according to the literature.⁴⁴ When we compare the spectrum of the bulk crystal to that of the solution-exfoliated flakes, we can immediately see a large difference in the intensity ratio of the A_{1g} and the overlapping E_{2g}^1 and 2LA peaks. This intensity ratio can be used to determine the number of WS₂ layers present in the sample, as this ratio increases as the flakes are reduced in thickness.^{44,45} In the spectra shown in Figure 3a, the $I_{\rm 2LA}/I_{\rm A_{e}}$ ratio is 0.8 for the bulk sample, whereas it is 3.4 for the exfoliated flake, indicating the few-layered ($\sim 2-3$) structure. With continued laser-etching cycles the amount of material is reduced, and this leads to a decrease in the signal intensity after each cycle. Thus, the spectra shown have been normalized to the intensity of the A_{1g} peak to allow for easy comparison. Figure 3b shows the Raman spectra of the as-deposited ultrasonication-exfoliated WS₂ flake, as well as after successive cycles of laser etching. When we compare the frequency of the Raman peaks after etching, we see very little shift in the peak position; however, we do see a change in the $I_{\rm 2LA}/I_{\rm A_{1g}}$ ratio with continued etching. After the first etching cycle, the $I_{\rm 2LA}/I_{\rm A_{\rm 1g}}$ ratio increases significantly to 5.1, and with the next etching cycle it increases slightly more to 5.2, indicating the reduced thickness of the deposited WS₂ flakes.

Figure 3c shows the comparison between the Raman spectrum of the bulk MoS₂ powder and the subsequent exfoliated flake, with the A_{1g} and E_{2g}^{1} phonon peaks labeled. We see a significant shift in the peak frequency when we compare the spectra before and after exfoliation, and this matches previous observations for similarly produced solution-exfoliated flakes.⁴⁶ This shift in phonon frequency is caused by the reduction in the number of layers of the MoS₂ and can be used to determine the thickness of the exfoliated flakes.^{47,48} As seen in the case of WS₂, the as-deposited exfoliated MoS₂ flakes are predominantly few layered ($\sim 2-3$ layers), as expected from literature based on the sonication time and the centrifuge conditions used.⁴⁹ Figure 3d shows the Raman spectra of an asdeposited MoS₂ flake with repeated laser etching, similar to the WS₂ flake, seen previously. These spectra were obtained in the same x-y position after etching to ensure that the same flake was being analyzed. Again there is very little noticeable shift in the position of the peaks with continued laser etching. However, although the shift in the Raman frequency is only minor for both the WS_2 and the MoS_2 flakes, we do see a significant change in the PL spectra after the laser etching as discussed previously.

As discussed previously, there is also a great deal of interest in the formation of vertically stacked van der Waals heterostructures, due to the unique electronic structure that



Figure 4. (a) Optical image of large-scale ($250 \ \mu m \times 250 \ \mu m$) laser etching performed on the solution-exfoliated WS₂ flakes deposited onto a Si/SiO₂ wafer. (b) SEM image showing the same patterned area. (c) Increased magnification image showing the interface between the laser-etched and pristine deposited WS₂ flakes.

can be formed through the nonradiative transfer of excitons within each layer.^{30–32,34–38} By creating these heterostructures by solution exfoliation, the interaction between the two materials can be analyzed easily. To investigate these heterostructures, dispersions of both exfoliated MoS₂ and WS₂ nanocrystals were mixed together and deposited onto substrates to create a series of heterostructures with random interlayer stacking as discussed in further detail in the Supporting Information (Figure S3).

■ LARGE-SCALE ETCHING

The key benefits of using solvent-based exfoliation over the previously demonstrated mechanical exfoliation and CVD growth of TMDs include both the low cost and the simple scalability. By creating large volumes of solvent-stabilized flakes and subsequently drop-casting or spin-coating them onto desired substrates, it is possible to demonstrate the large-scale applicability of this method. Previous examples showing a similar laser-etching process have only demonstrated this over comparatively small areas of several 10 s of micrometers due to sample size limitations.^{21,22} In Figure 4, we demonstrate a proof-of-principle for the scalability of using liquid-exfoliated material by first simply drop-casting WS₂ flakes over a silicon wafer to create a large-scale coverage, which was then laser etched over a designated ~250 μ m × 250 μ m area. Figure 4a shows the optical image of the etched area showing the distinct difference in optical contrast where the laser was rastered. Because of the use of simple drop-casting to deposit the flakes, there is a random assortment of thicknesses present on the surface; however, as the laser etching appears to more effectively remove thicker flakes due to poor thermal conductivity with the underlying substrate, repeated etching cycles can be used to produce an area of an approximately similar thickness. Figure 4b shows the SEM image of the same area where laser etching has occurred. In Figure 4c, an area of increased magnification shows the interface between the pristine as-deposited flakes and the laser-etched material, indicating a change in flake morphology. The selective etching of the crystal edges is also visible as the laser-etched flakes are seen to all have smoother edges than the pristine flakes as well as a reduced thickness. With further optimization, such as carefully controlling the initial concentration of the TMD dispersion and deposition conditions, it would be possible to create homogenous large-scale coatings that could then be postprocessed using this laser-etching technique to enhance their mobility²¹ and sensitivity to photons²² or be patterned for further functionalization.^{23,50}

EXPERIMENTAL SECTION

Dispersions of MoS_2 and WS_2 were produced following the established literature protocols.^{19,49} Briefly, commercially available TMD powders (MoS_2 and WS_2) were each suspended in N-methylpyrrolidone (NMP) at a concentration of 10 mg mL^{-1} . To investigate the effect of the residual solvent, similar dispersions were also produced using a mixed solvent system of isopropanol (IPA) and water in a ratio of 7:3 (MoS₂) or 1:1 (WS_2) as this has previously been shown to match the exfoliation performance of NMP.^{51,52} The suspension was then bath sonicated (Elmasonic P, 40% power, 37 kHz) for 12 h at a constant temperature of 20 °C. The resultant dispersion was then centrifuged at 6000 rpm (3139g) for 30 min, and the supernatant was decanted to provide a stable TMD dispersion. This technique has been shown to produce a distribution of flake sizes and thicknesses, ranging from a monolayer up to several layers (3-5) thick with lateral dimensions of 200-300 nm.^{19,49} To characterize the flakes, the NMP suspension was first diluted by a factor of \sim 50 in IPA to minimize the NMP and prevent agglomeration before drop-casting onto Si/SiO2 wafers. The residual solvent was first evaporated on a hot plate at 50 °C before transferring into a vacuum oven and annealed at 200 °C at a pressure of 5×10^{-5} mbar to ensure thorough removal of the residual solvent. This vacuum annealing step was found to be vital in achieving the enhanced PL intensity after the laser etching, as without this step the residual solvent (NMP) molecules that are trapped between the TMD layers cause a large amount of amorphous carbon to appear in the Raman spectrum (seen in Figure S1). This amorphous carbon significantly suppresses the PL intensity of the MoS₂ or WS₂ flakes; however, after the vacuum annealing very little to no amorphous carbon was detected after the laser etching, indicating the successful removal of trapped solvent molecules and the increased cleanliness of the TMD flakes. In contrast, the IPA/H₂O dispersions were deposited as prepared onto the Si/SiO₂ wafers and left to dry, and little to no amorphous carbon was detected after etching.

Confocal Raman spectroscopy and PL measurements were conducted using a Renishaw inVia microscope with a 532 nm (2.33 eV) excitation at a power of 1.8 mW with a 100× objective and a grating of 1800 l/mm to achieve a spectral resolution of ~1 cm⁻¹. Laser etching was performed using the same Raman setup, but with an incident laser power of ~20 mW (power density: ~2.45 MW cm⁻²). The laser was rastered over a designated area in 500 nm steps with a typical exposure time of 100 ms per step, although this could be increased as

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desired. For the large etching area, a lower magnification objective was used $(20\times)$, and a longer exposure time $(\sim 1 \text{ s})$ compensated for the lower power density. The morphology of the flakes after etching, as well as the individual crystal flake size of the TMDC materials, was characterized using a Philips/FEI XL30 FEG ESEM SEM. All images were obtained with an accelerating voltage of 15 kV, under high vacuum conditions utilizing secondary electron detection. AFM images were taken using a Bruker Multimode using the PeakForce Tapping mode under ambient conditions. After acquiring images of the pristine flakes, the sample was laser etched using the same conditions listed previously.

CONCLUSIONS

We have demonstrated that using a conventional Raman experimental setup it is possible to enhance the PL yield, by up to 8-fold, of ultrasonication-exfoliated MoS₂ and WS₂ using a simple laser-etching procedure. This laser-etching procedure, which had previously only been applied for mechanically exfoliated and CVD-grown MoS₂, has been demonstrated to be applicable for other solution-exfoliated 2D materials. The mechanism behind this PL improvement is a combination of the thinning of the material through thermal sublimation of the uppermost layers and atmospheric doping by H₂O and O₂ that occurs on the freshly exposed surface. This technique allows the cheap and scalable production of optoelectronic devices, which require the high quantum yield that monolayer flakes of these TMDs provide, while still using solution-based exfoliation. A simple proof-of-principle of the scalability of this laser-etching technique for solution-exfoliated TMD crystals was also demonstrated, indicating that there are many applications in which this would be beneficial, such as, photodetectors or other optoelectronics. Finally, this laser-etching technique was used to modify heterostructures (MoS₂ and WS₂) produced by mixing dispersions of each TMD, providing information about the structure and the ability to tune the optical properties as desired.

ASSOCIATED CONTENT

S Supporting Information

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Further Raman analysis of laser-etched samples; AFM analysis of pristine and laser-etched flakes; Raman and PL characterization of solution-phase-generated MoS₂/WS₂ heterostructures (PDF)

AUTHOR INFORMATION

Corresponding Author

*E-mail: Mark.Bissett@Manchester.ac.uk.

ORCID 0

Mark A. Bissett: 0000-0002-8908-7960 Robert A. W. Dryfe: 0000-0002-9335-4451

Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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