

# NiO<sub>x</sub> Hole Transport Layer for Perovskite Solar Cells with Improved Stability and Reproducibility

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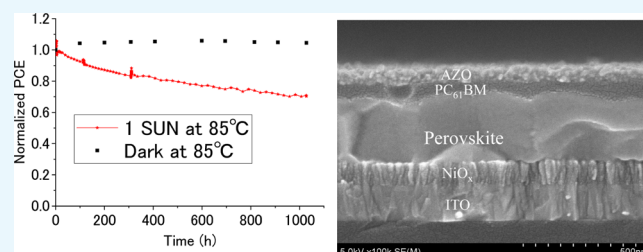
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## Supporting Information

**ABSTRACT:** In this study, highly stable, low-temperature-processed planar lead halide perovskite (MAPbI<sub>3-x</sub>Cl<sub>x</sub>) solar cells with NiO<sub>x</sub> interfaces have been developed. Our solar cells maintain over 85% of the initial efficiency for more than 670 h, at the maximum power point tracking (MPPT) under 1 sun illumination (no UV-light filtering) at 30 °C, and over 73% of the initial efficiency for more than 1000 h, at the accelerating aging test (85 °C) under the same MPPT condition. Storing the encapsulated devices at 85 °C in dark over 1000 h revealed no performance degradation. The key factor for the prolonged lifetime of the devices was the sputter-deposited polycrystalline NiO<sub>x</sub> hole transport layer (HTL). We observed that the properties of NiO<sub>x</sub> are dependent on its composition. At a higher Ni<sup>3+</sup>/Ni<sup>2+</sup> ratio, the conductivity of NiO<sub>x</sub> is higher, but at the expense of optical transmittance. We obtained the highest power conversion efficiency of 15.2% at the optimized NiO<sub>x</sub> condition. The sputtered NiO<sub>x</sub> films were used to fabricate solar cells without annealing or any other treatments. The device stability enhanced significantly compared to that of the devices with PEDOT:PSS HTL. We clearly demonstrated that the illumination-induced degradation depends heavily on the nature of the HTL in the inverted perovskite solar cells (PVSCs). The sputtered NiO<sub>x</sub> HTL can be a good candidate to solve stability problems in the lead halide PVSCs.



## INTRODUCTION

After the invention of organic–inorganic hybrid perovskite solar cell (PVSC) with 3.81% power conversion efficiency (PCE) by Kojima et al.,<sup>1</sup> it has attracted much attention because of its low cost and easy fabrication. Within a very short period of time, PVSCs have achieved a rapid development and the PCE exceeded 20%, comparable to those of the conventional silicon solar cells.<sup>2,3</sup> There have also been a growing interest in the development of low-temperature-processed PVSCs with inverted device structures because of their lower fabrication cost and new solar cell applications with their light weight and flexibility. However, it is still a great challenge to fabricate PVSCs with long lifetime using low-temperature processes for successful commercialization. Although the impressive high efficiency of PVSCs stands up to those of other existing PV technologies with >20% efficiency, the rapid degradation phenomena broadly observed for PVSCs overshadow the future of this PV technology.<sup>4</sup> The literature concerning the real performance of the PVSCs in long-term operations is still limited, and many studies on the perovskite stabilities focus on the stored lifetime (shelf-life).<sup>5,6</sup> As far as we know, there have been only a few reports on the stability of the operating conditions of PVSCs under light.<sup>7–12</sup> The purpose of this study is to examine the effect of an interface layer on the

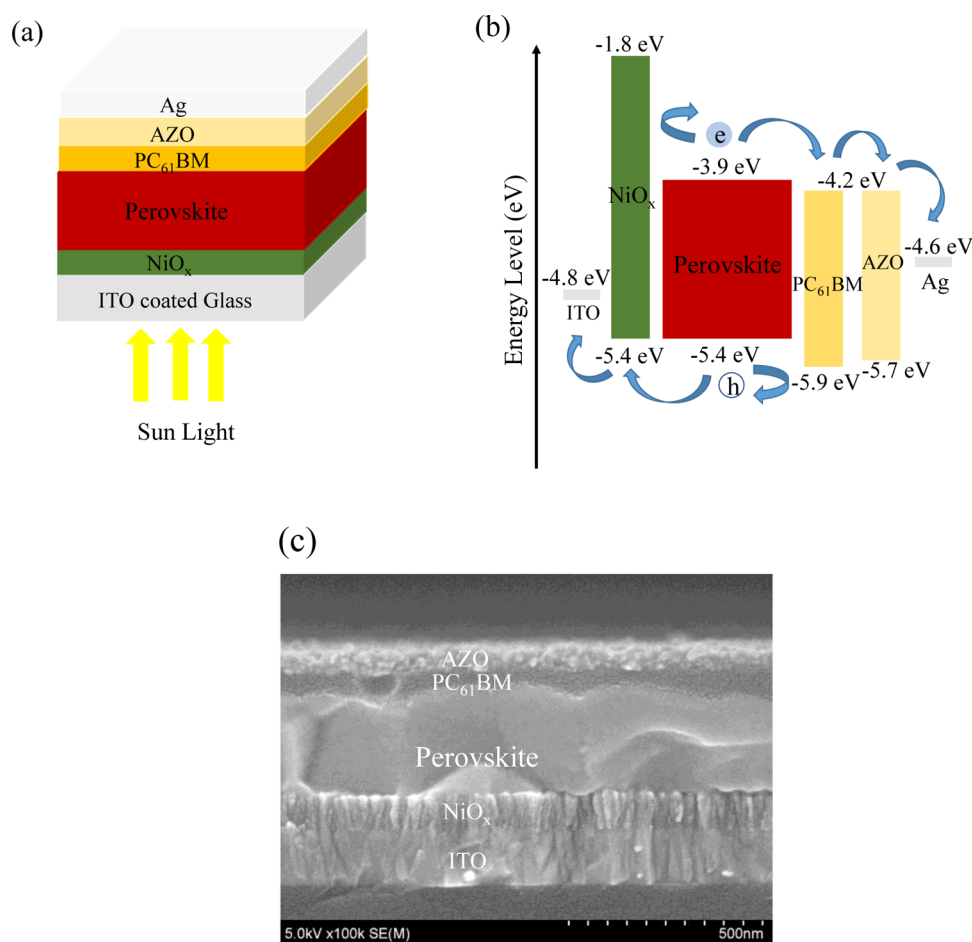
long-term operation of low-temperature-processed PVSCs under real working conditions, with maximum power point tracking (MPPT) under 1 sun (AM1.5G) illumination (no UV-light filtering) and in the accelerated aging at 85 °C under the same MPPT condition.

One of the possible reasons for the instability of the PVSCs is the instability and/or adverse effects of the organic hole transport layers (HTLs),<sup>13–16</sup> and we set out to develop an ideal HTL that has a suitable energy level with perovskite, high optical transparency in the visible range, and high stability and supports the fabrication of thick and high-quality perovskite films. Because of their large band gap (~3.6 eV), deep valance band edge (~5.4 eV), ease of controlling composition, and low cost with superior thermal and chemical stabilities, NiO derivatives were identified to achieve our goal. In fact, NiO-based HTLs have been used for the fabrication of inverted and other type of PVSCs, and the stability issue often became a focus in studies concerning them (Table S1).<sup>7,14,16–36</sup> In this study, we demonstrate inverted planar PVSCs on the basis of the sputter-deposited polycrystalline NiO<sub>x</sub> hole transport

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**Figure 1.** (a) Device structure consisting of ITO-coated glass/ $\text{NiO}_x$ /perovskite ( $\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$ )/ $\text{PC}_{61}\text{BM}$ /AZO/Ag and (b) the corresponding energy band diagram and (c) cross-sectional SEM image.

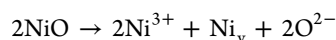
material. We prepared compact and homogeneous  $\text{NiO}_x$  films on indium tin oxide (ITO)-coated glass by radio frequency (rf) magnetron sputtering, allowing us to control the oxygen composition and thickness, with high reproducibility.<sup>20</sup> We revealed in detail the effect of  $\text{NiO}_x$  composition and thickness (from 20 to 250 nm) on the device performance and showed that fine-tuning of the composition ( $\text{Ni}^{3+}/\text{Ni}^{2+}$  ratio) and thickness resulted in high-performance PVSCs with over 15% efficiency and unprecedented stability for low-temperature-processed  $\text{MAPbI}_{3-x}\text{Cl}_x$  devices. We also compared the stabilities of the  $\text{NiO}_x$ -based devices and the PEDOT:PSS-based devices under continuous illumination from a class AAA solar simulator and MPPT condition. Surprisingly, the  $\text{NiO}_x$ -based devices maintained about 73% of the initial efficiency after 1000 h of continuous operation under the 85 °C accelerated aging condition, whereas the PEDOT:PSS-based devices maintained only <20% of the initial efficiency at 30 °C within 400 h of operation. We clearly demonstrated that the illumination-induced degradation significantly depends on the nature of the HTL in the inverted PVSCs. The sputtered  $\text{NiO}_x$  HTL can be a good candidate to solve stability problems in the low-temperature-processed lead halide PVSCs.

## RESULTS AND DISCUSSION

The device structure of our PVSC is shown in Figure 1a. This inverted structure consists of ITO-coated glass/ $\text{NiO}_x$ /perovskite ( $\text{CH}_3\text{NH}_3\text{PbI}_{3-x}\text{Cl}_x$ )/ $\text{PC}_{61}\text{BM}$ /aluminum-doped zinc

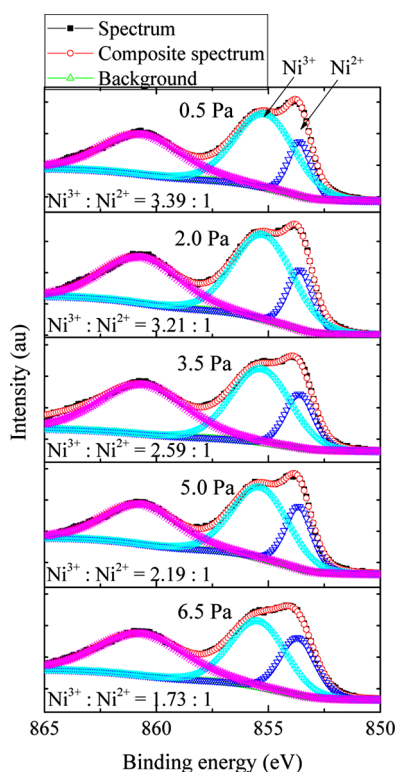
oxide (AZO)/Ag, where the  $\text{NiO}_x$  is the HTL (or electron-blocking layer) and  $\text{PC}_{61}\text{BM}$ /AZO is the electron transport layer. The corresponding band diagram and energy levels are shown in Figure 1b. The scanning electron microscopy (SEM) cross section of the device is illustrated in Figure 1c. Sealing glass was used to encapsulate the device, to protect it from moisture and oxygen.

**Properties of  $\text{NiO}_x$  Thin Films.**  $\text{NiO}_x$  films were fabricated at different Ar pressures of the deposition chamber; the optical and electrical properties of the resulting  $\text{NiO}_x$  films depend on the deposition conditions, and it was found that at lower pressure, because of Ni vacancy/excess oxygen in the films, they became blackish (Figure S1). This observation parallels to that of the previous study with compact  $\text{NiO}_x$  electron-blocking layers prepared by sputtering.<sup>20</sup> The oxygen content of the black form was slightly greater than that of its green counterpart.<sup>37</sup> The defect of  $\text{NiO}$  films is due to the interstitial oxygen or  $\text{Ni}^{2+}$  vacancy that occurs as a result of the creation of  $\text{Ni}^{3+}$  ions. For each  $\text{Ni}^{2+}$  vacancy ( $\text{Ni}_v$ ), two  $\text{Ni}^{2+}$  ions must be converted to  $\text{Ni}^{3+}$  to preserve the overall charge neutrality in the crystal. This ensures excess oxygen compared to the number of nickel ions in the crystal. Finally, the creation of defects in  $\text{NiO}_x$  crystals can be expressed with the following equation



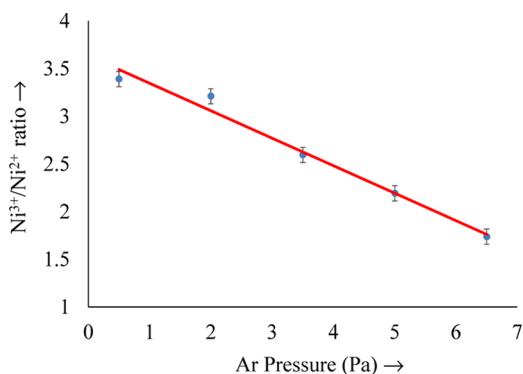
Now, if an electron moves from a  $\text{Ni}^{2+}$  site to a  $\text{Ni}^{3+}$  site, it is like the movement of a hole in the opposite direction through the  $\text{Ni}^{2+}$  sites. These holes contribute to the electrical conductivity of undoped  $\text{NiO}_x$  crystals. Therefore,  $\text{NiO}_x$  with excess oxygen is a p-type semiconductor.<sup>38</sup>

To analyze the  $\text{Ni}^{3+}/\text{Ni}^{2+}$  ratio of the sputtered  $\text{NiO}_x$  thin films, they were characterized by X-ray photoelectron spectroscopy (XPS). As shown in Figure 2, the peak at 860.8 eV is



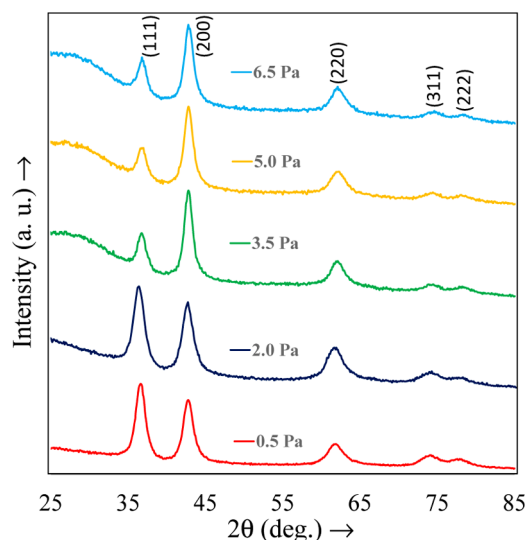
**Figure 2.** XPS spectra representing the  $\text{Ni } 2p_{3/2}$  peak with deconvolution of  $\text{Ni}^{3+}$  and  $\text{Ni}^{2+}$  peaks of  $\text{NiO}_x$  films prepared at 0.5, 2.0, 3.5, 5.0, and 6.5 Pa Ar pressures.

due to the shake-up process of the  $\text{NiO}$  structure, the peak at 853.8 eV indicates  $\text{Ni}^{2+}$  ion, and the peak at 855.5 eV indicates  $\text{Ni}^{3+}$  ion.<sup>23</sup> The compositions of  $\text{Ni}^{2+}$  and  $\text{Ni}^{3+}$  in the crystals were determined by calculating the integral area of the fitting curve of the XPS spectra, and the  $\text{Ni}^{3+}/\text{Ni}^{2+}$  ratio of  $\text{NiO}_x$  films is shown in Figure 3. It is evident from the figure that the  $\text{Ni}^{3+}/\text{Ni}^{2+}$  ratio decreased with increasing pressure.



**Figure 3.** Composition of  $\text{NiO}_x$  films at different Ar pressures.

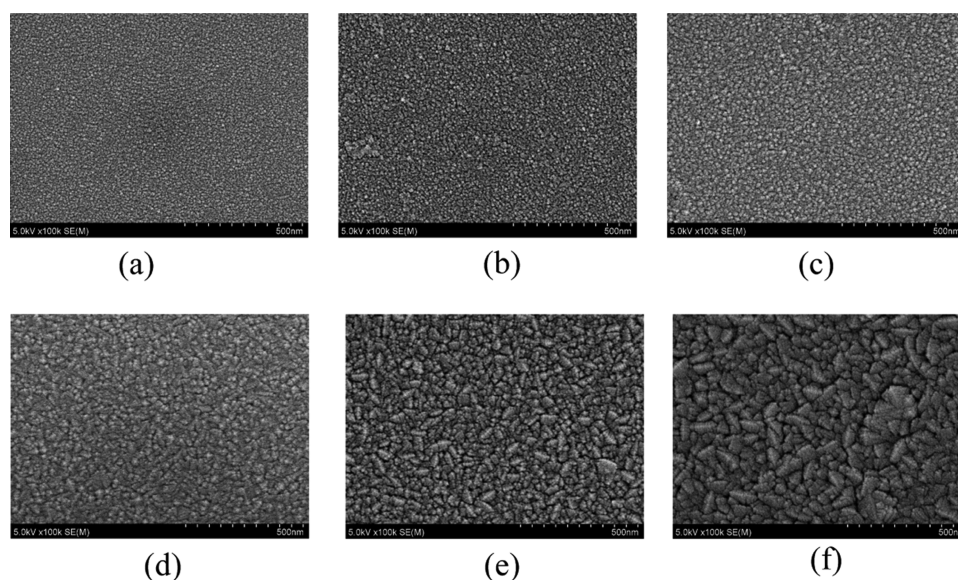
The films prepared by sputtering methods are polycrystalline in nature and have cubic structure, as shown in Figure 4. At



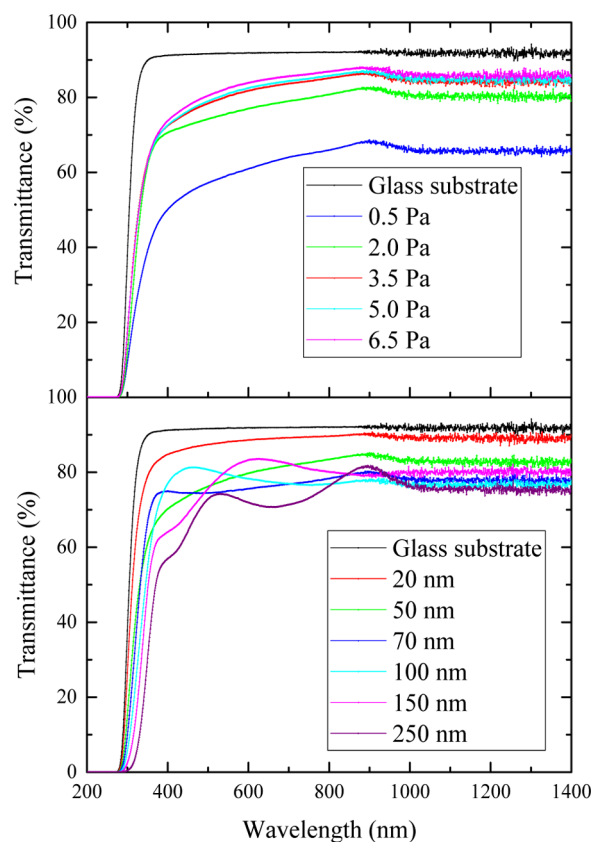
**Figure 4.** XRD patterns of the  $\text{NiO}_x$  films at different Ar pressures.

lower pressure, the (111) peak is dominating, and with increasing pressure, (200) peak becomes dominating. Broader X-ray diffraction (XRD) peaks indicate smaller crystals of the fabricated films. Furthermore, we studied the surface morphology of  $\text{NiO}_x$  films by SEM, as shown in Figure 5. The SEM images demonstrate small grains of  $\text{NiO}_x$  films, and the grain size is dependent on the thickness of the films. At a lower thickness ( $\sim 20$  nm), the grain size is quite small; however, with increasing thickness (up to 250 nm), the grain size gradually increases. The Ar pressure in the sputtering chamber has little effect on the grain size (Figure S2), and with increasing Ar pressure, the grain size increases. With increasing Ar pressure in the sputtering chamber, the transmittance of the prepared films also increases, as shown in Figure 6. The films prepared at 0.5 Pa have a lower transmittance of 60% at the 550 nm wavelength of the visible range of the spectrum, whereas the films prepared at or above 3.5 Pa show more than 80% transmittance at the 550 nm wavelength of the spectrum. It is notable that the difference of transmittance between the films prepared at 0.5 and 3.5 Pa is very large ( $\sim 20\%$ ), whereas the effect is very little ( $<5\%$ ) for the films prepared at 3.5–6.5 Pa. Because of increasing Ni vacancy/excess oxygen, the films became blackish and their transmittance decreased at lower Ar pressure.

Figure 6 also compares the transmittances of the fabricated  $\text{NiO}_x$  films at different thicknesses. It is evident from the figure that transmittance declines with increasing thickness, as expected, and the only exception is at the 150 nm thickness, where the transmittance is higher than that at the 70 nm thickness. The reason may be the large grain size, as shown in Figure 5, which might reduce the light scattering by the grain boundary. The films prepared at 0.5 and 2.0 Pa have resistivities of  $3.28 \times 10^2$  and  $2.92 \times 10^3 \Omega \cdot \text{cm}$ , respectively. However, the films prepared at the pressure range of 3.5–6.5 Pa have very high resistivities, which we could not measure using our linear four-probe system. The low resistivity of the films prepared at lower pressure possibly indicated the increased Ni vacancy/excess oxygen in the crystals, which act as the hole and contribute to the electrical conductivity. This effect is quite



**Figure 5.** SEM images of NiO<sub>x</sub> films fabricated at 3.5 Pa with different thicknesses of (a) 20 nm, (b) 50 nm, (c) 70 nm, (d) 100 nm, (e) 150 nm, and (f) 250 nm.



**Figure 6.** Transmission spectra of NiO<sub>x</sub> films on glass substrate at different Ar pressures (film thickness: 60–70 nm for 0.5–3.5 Pa and ~40 nm for 5.0–6.5 Pa) and at different thicknesses at an Ar pressure of 3.5 Pa.

consistent with the XPS and XRD results. The resistivity values of the (200)-orientated NiO<sub>x</sub> films are higher than those of the (111)-orientated films.<sup>39</sup>

**Device Performances.** We revealed the dependence of device performances on the Ar pressure during sputter deposition of NiO<sub>x</sub>, and the results are summarized in Table

1 and Figure 7. It is observed that with increasing Ar pressure up to 3.5 Pa, the device performance enhanced and beyond that pressure, it declined (Figure 7a). This phenomenon can be explained with the optical and electrical properties of the sputtered NiO<sub>x</sub> thin films. As shown in Figure 6, at lower pressure, the NiO<sub>x</sub> thin films absorbed some part of the incident sunlight. At higher pressure, although the transmittance is better, higher resistance of the films reduces the device performance, as seen from the increased series resistance ( $R_s$ ) for 5.0 and 6.5 Pa. Therefore, we selected the devices with NiO<sub>x</sub> HTL prepared at 3.5 Pa Ar pressure for further study of thickness-dependent device performance, and the results are summarized in Table 2.

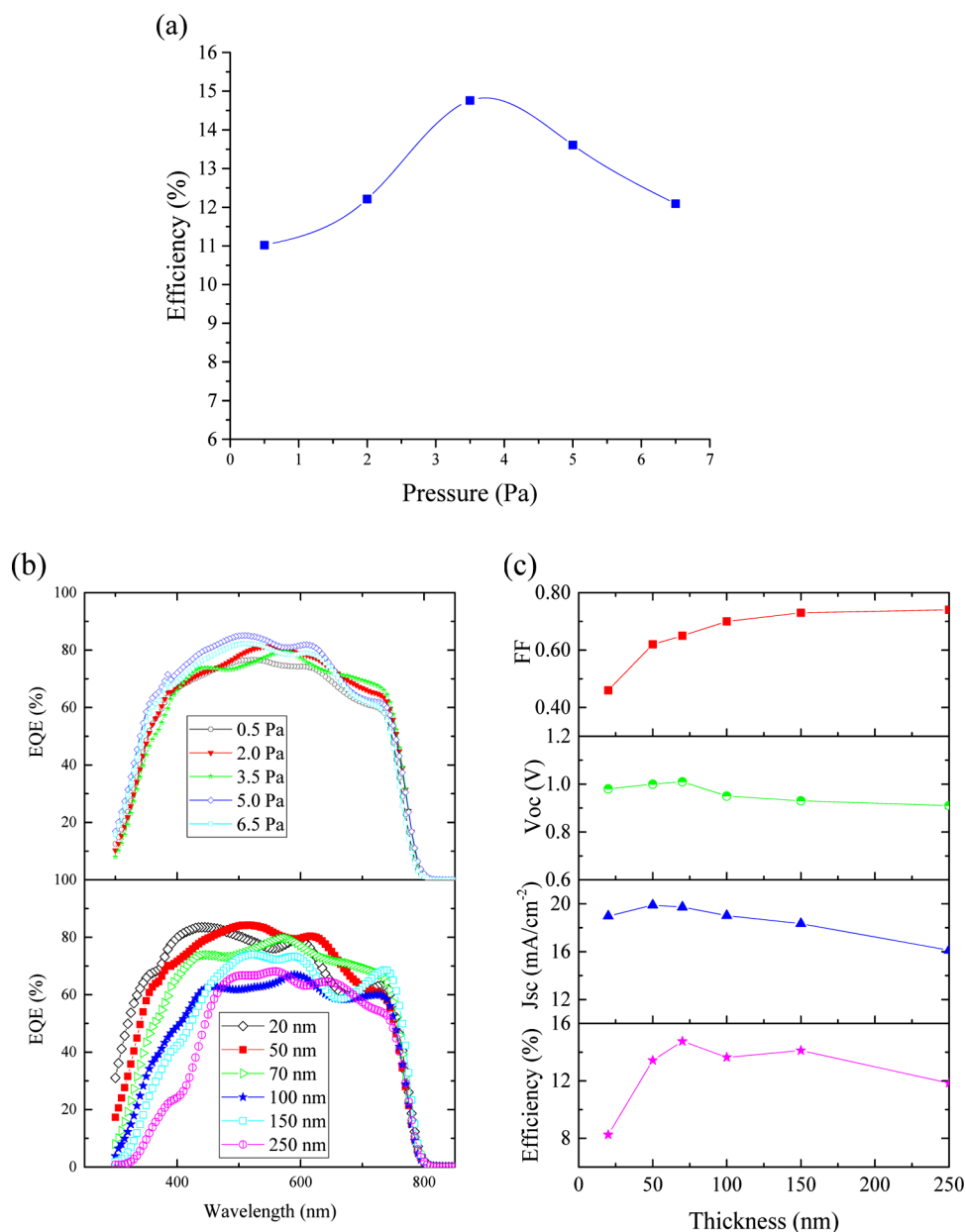
It was found that when the NiO<sub>x</sub> layer was too thin (e.g., 20 nm), the devices showed lower PCE. The very thin film may be not sufficient to block the photogenerated electrons because of the insufficient coverage of the ITO film with the NiO<sub>x</sub> layer, which in turn decreases the FF.<sup>22,29</sup> In fact, the improved FF values with increasing thickness were observed possibly due to the elimination of pinholes. However, with a very thick NiO<sub>x</sub> HTL, the FF was satisfactory, but the PCE was again low. With higher thickness, the transmittance decreased, with a small deviation at 150 nm thickness. From Figure 7b, we can see that with higher thickness, the EQE of the devices decreased significantly at lower wavelength, which reduces their short-circuit current density ( $J_{sc}$ ). Devices with a NiO<sub>x</sub> HTL thickness of  $70 \pm 3$  nm showed the best performance (Figure 8). Although the hysteresis behaviors of the PVSCs are an important issue, which is frequently observed and reported in the literature,<sup>40</sup> our NiO<sub>x</sub> devices showed almost no hysteresis behaviors. The devices showed a good reproducibility with a limited deviation of PCE, as shown in Figure 9. Histograms of solar cell efficiencies were collected from 32 cells with NiO<sub>x</sub> HTL of thickness  $70 \pm 3$  nm prepared at 3.5 Pa Ar pressure.

No significant performance degradation was observed for encapsulated devices with NiO<sub>x</sub> HTL stored at ambient temperature under dark condition for 5 months and 85 °C for 1000 h. The significant improvement of the stability over previous NiO<sub>x</sub>-based devices (Table S1) was achieved possibly due to the synergy of the NiO<sub>x</sub> HTL, with the high

**Table 1.** Performance of the Devices with Different Ar Pressures of the Deposition Chamber during Sputter Deposition of NiO<sub>x</sub><sup>a</sup>

| Ar pressure (Pa) | $\eta$ (%)       | $J_{sc}$ (mA/cm <sup>2</sup> ) | $V_{oc}$ (V)    | fill factor (FF) | $R_c$ ( $\Omega\cdot\text{cm}^2$ ) | $R_{sh}$ ( $\Omega\cdot\text{cm}^2$ ) $\times 10^3$ |
|------------------|------------------|--------------------------------|-----------------|------------------|------------------------------------|---|
| 0.5              | 11.02 $\pm$ 0.46 | 17.67 $\pm$ 0.71               | 0.97 $\pm$ 0.01 | 0.63 $\pm$ 0.03  | 5.01 $\pm$ 0.58                    | 2.07 $\pm$ 0.19                                     |
| 2.0              | 12.21 $\pm$ 0.58 | 18.70 $\pm$ 0.83               | 0.98 $\pm$ 0.01 | 0.63 $\pm$ 0.02  | 5.16 $\pm$ 0.47                    | 2.35 $\pm$ 0.72                                     |
| 3.5              | 14.76 $\pm$ 0.39 | 19.86 $\pm$ 0.85               | 1.01 $\pm$ 0.02 | 0.68 $\pm$ 0.02  | 5.41 $\pm$ 0.62                    | 3.25 $\pm$ 0.33                                     |
| 5.0              | 13.61 $\pm$ 0.61 | 19.79 $\pm$ 0.69               | 0.98 $\pm$ 0.01 | 0.66 $\pm$ 0.03  | 8.85 $\pm$ 0.83                    | 2.30 $\pm$ 0.46                                     |
| 6.5              | 12.09 $\pm$ 0.57 | 18.88 $\pm$ 0.48               | 0.98 $\pm$ 0.01 | 0.65 $\pm$ 0.04  | 8.74 $\pm$ 0.92                    | 1.23 $\pm$ 0.57                                     |

<sup>a</sup>Data collected from at least 12 cells for each condition (NiO<sub>x</sub> film thickness: 60–70 nm for 0.5–3.5 Pa and ~40 nm for 5.0–6.5 Pa).



**Figure 7.** Device performances. (a) Ar-pressure-dependent PCE, (b) external quantum efficiency (EQE) of the devices with NiO<sub>x</sub> HTL prepared at different Ar pressures and thicknesses, (c) PCE,  $J_{sc}$ ,  $V_{oc}$ , and FF depending on NiO<sub>x</sub> HTL thicknesses.

optoelectronic quality of the MAI-treated perovskite layer.<sup>41</sup> We also observed that the PCE and open-circuit voltage ( $V_{oc}$ ) gradually increase with time at ambient temperature and under dark condition. The improvement can be explained by the ion migration and chemical doping of the PCBM layer by iodide.<sup>8,42</sup> On the other hand, under continuous 1 sun

illumination (no UV-light filtering) and MPPT condition at 30 °C, the performance first degraded gradually and then the degradation rate decreased; it eventually reached 87% of the initial efficiency after 670 h of operation, as shown in Figure 10a. The PEDOT:PSS-based devices degrade rapidly, and within 400 h, they retain only <20% of the initial PCE, possibly

Table 2. Performance of the Devices with Different Thicknesses of NiO<sub>x</sub> Prepared at 3.5 Pa<sup>a</sup>

| NiO <sub>x</sub> thickness (nm) | η (%)        | J <sub>sc</sub> (mA/cm <sup>2</sup> ) | V <sub>oc</sub> (V) | FF          | R <sub>s</sub> (Ω·cm <sup>2</sup> ) | R <sub>sh</sub> (Ω·cm <sup>2</sup> ) × 10 <sup>3</sup> |
|---------------------------------|--------------|---------------------------------------|---------------------|-------------|-------------------------------------|--|
| 20 ± 2                          | 8.25 ± 0.41  | 18.47 ± 0.23                          | 0.98 ± 0.01         | 0.46 ± 0.11 | 21.48 ± 1.31                        | 0.86 ± 0.15  |
| 50 ± 2                          | 13.43 ± 0.56 | 19.89 ± 0.65                          | 1.00 ± 0.01         | 0.62 ± 0.02 | 5.07 ± 0.45                         | 1.47 ± 0.38  |
| 70 ± 3                          | 14.76 ± 0.39 | 19.86 ± 0.85                          | 1.01 ± 0.02         | 0.68 ± 0.02 | 5.41 ± 0.62                         | 3.25 ± 0.33  |
| 100 ± 5                         | 13.64 ± 0.67 | 19.01 ± 0.41                          | 0.95 ± 0.01         | 0.69 ± 0.01 | 5.82 ± 0.33                         | 3.05 ± 0.46  |
| 150 ± 5                         | 14.12 ± 0.35 | 18.74 ± 0.64                          | 0.93 ± 0.02         | 0.73 ± 0.01 | 5.35 ± 0.52                         | 2.83 ± 0.63  |
| 250 ± 7                         | 11.85 ± 0.58 | 16.11 ± 0.72                          | 0.90 ± 0.02         | 0.74 ± 0.01 | 5.80 ± 0.61                         | 2.76 ± 0.29  |

<sup>a</sup>Data collected from at least 12 cells for each thickness.

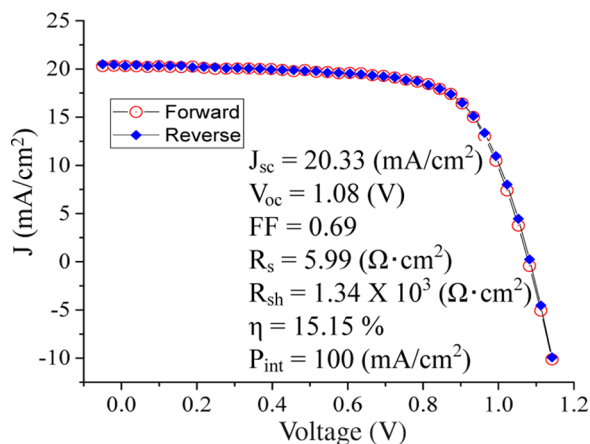


Figure 8. *J*–*V* curve of the best device with NiO<sub>x</sub> HTL (3.5 Pa, 70 nm) under 1 sun condition measured at forward scan (−0.05 → 1.2 V; step, 0.02 V; delay time, 200 ms) and reverse scan (1.2 → −0.05 V; step, 0.02 V; delay time, 200 ms).

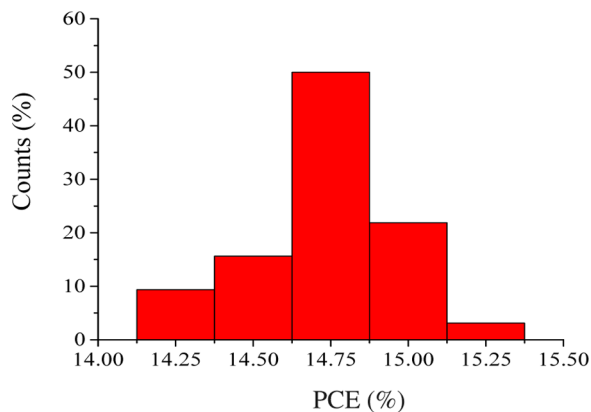


Figure 9. PCE distribution histogram of devices with NiO<sub>x</sub> HTL prepared at 3.5 Pa.

due to the chemical nature of the PEDOT:PSS layer.<sup>13–16</sup> The lifetime of solar cells may be defined as the operation time until the output of the device has fallen below a certain level, that is, 70% of nominal efficiency for more than 40 years was expected from some commercial Si solar cells, and it often requires accelerated aging conditions to predict their lifetime in a reasonable testing time. In fact, the NiO<sub>x</sub>-based devices showed surprisingly high stability and it would require significant testing time to observe degradation at 30 °C (Figure 10a). Thus, we decided to accelerate the aging by increasing the testing temperature up to 85 °C (Figure 10b). The degradation rate was indeed increased compared to that of the 30 °C testing, and the initial 14% efficiency at room temperature reduced to about 13% because of the negative-efficiency

temperature coefficient of the perovskite devices.<sup>8</sup> Further continuation of the high-temperature testing induced significant efficiency drop from 13.0% to below 9.5%, which corresponds to 73% of the initial efficiency after 1000 h. According to the definition of the acceleration factor, *K*, under the assumption of an Arrhenius model, defined by the equation<sup>43</sup>

$$K = \exp \left[ \frac{E_a}{k_B} \left( \frac{1}{T_{\text{high}}} - \frac{1}{T_{\text{low}}} \right) \right] \quad (1)$$

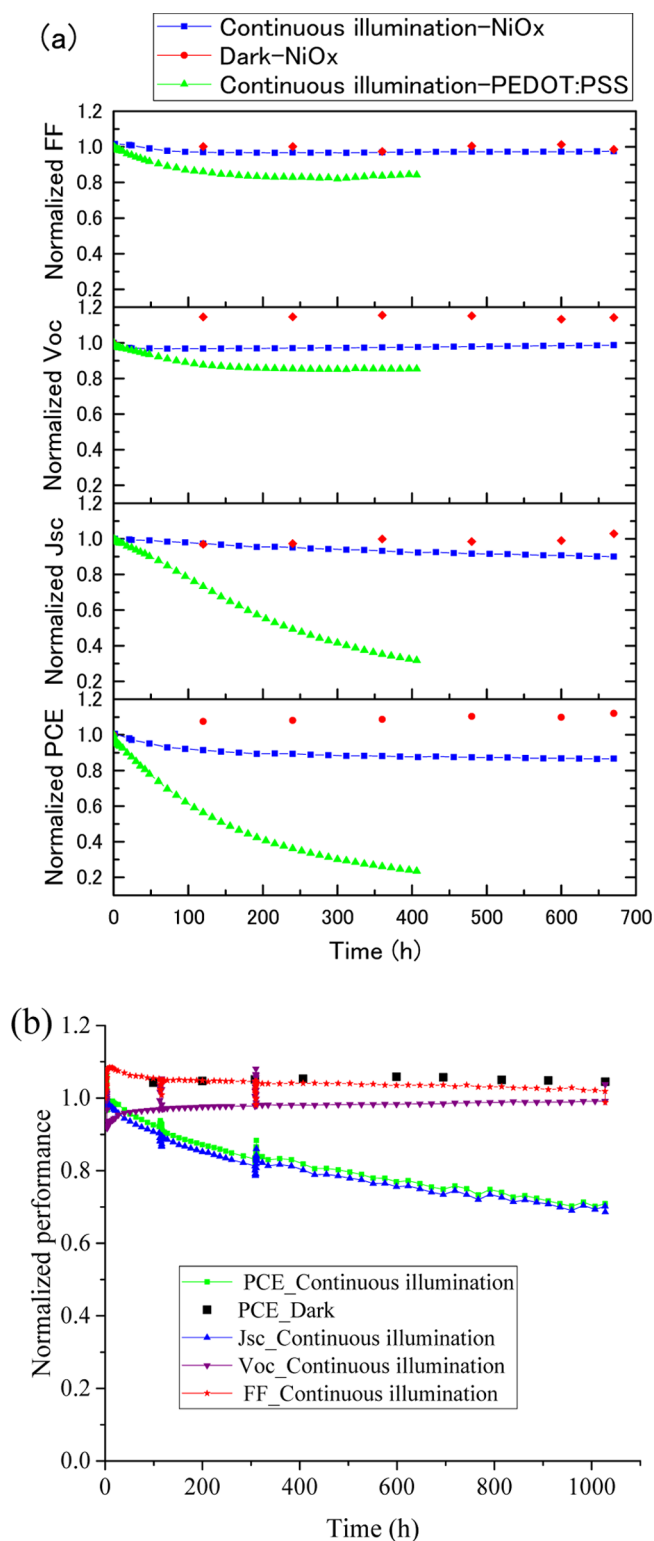
where *E<sub>a</sub>* is the activation energy for the degradation processes in electron volts (eV), *k<sub>B</sub>* is the Boltzman constant, and two testing temperatures (*T<sub>high</sub>* = 85 °C and *T<sub>low</sub>* = 25 °C), the result of the accelerated testing suggested that the NiO<sub>x</sub>-based devices would operate over 7000 h (*K* = 7) at room temperature (*T<sub>low</sub>* = 25 °C) before the output falls below 73% of the initial efficiency, if the activation energy (*E<sub>a</sub>*) for a degradation path of these devices was 0.3 eV, which is a lower-end value estimated for a polymer solar cell.<sup>43</sup> In the similar manner, the device would operate over 3000 h (*K* = 3) at 50 °C (Figure S3), which is a typical working condition of a solar cell operating on the roof. As shown in Figure S3, the predicted lifetime based on the acceleration factor (eq 1) of the solar cells strongly relies on the activation energy (*E<sub>a</sub>*); thus, accelerated aging tests at several different temperature conditions will be necessary for more conclusive discussions in future stability studies.

## CONCLUSIONS

In summary, we have successfully developed efficient and hysteresis-free inverted planar lead halide PVSCs with improved stability and reproducibility using NiO<sub>x</sub> HTLs. The NiO<sub>x</sub> layers were prepared by rf magnetron sputtering without postdeposition annealing. Ni<sup>3+</sup>/Ni<sup>2+</sup> ratio in the NiO<sub>x</sub> film not higher than ~3 with 70 ± 3 nm thickness showed superior quality as an HTL. The synergy of the NiO<sub>x</sub> HTL with the high optoelectronic quality of the MACl-treated perovskite layer<sup>41</sup> resulted in no performance degradation at 85 °C in dark. More importantly, although it was found that these encapsulated devices showed no degradation under dark condition at 85 °C over 1000 h, they still degraded under continuous 1 sun illumination at 30 and 85 °C and under MPPT operation. Nevertheless, our results demonstrated that NiO<sub>x</sub> as HTL is a good candidate to solve stability problems in the low-temperature-processed inverted PVSCs. Further investigation is necessary to improve the solar cell performance under continuous illumination for commercial applications.

## METHODS

**NiO<sub>x</sub> Film Deposition.** The NiO<sub>x</sub> thin films were prepared on the commercially available precleaned and prepatterned



**Figure 10.** (a) Stability of the encapsulated device at 30 °C (~50% RH) under ambient and dark condition and under MPPT condition (1 sun) (the device was kept under MPPT condition between the periodical  $J$ - $V$  measurements). (b) Accelerated aging test at 85 °C (~5% RH) in dark and under MPPT condition (the device was kept under MPPT condition between the periodical  $J$ - $V$  measurements).

ITO-coated glass substrates using an rf magnetron sputtering system (SVC-700 RFIINA; Sanyu Electron, Japan). Substrates for devices and glasses for characterizing the structural, optical,

electrical, and compositional properties were deposited in the same batch for side-by-side comparisons. All of the substrates were treated with ultraviolet ozone for 20 min and immediately loaded in the deposition chamber. Before deposition, the chamber was evacuated until the pressure inside it becomes  $<2 \times 10^{-3}$  Pa; then, pure argon gas was introduced at the rate of 20 sccm. Sputter deposition was carried out in an argon gas pressure of 0.5–6.5 Pa and a rf power supply of 50 W. The thickness of the NiO<sub>x</sub> films for pressure  $<3.5$  Pa was about 60–70 nm and that for  $>5.0$  Pa was about 40 nm. Commercially available sintered 99.9% pure NiO was used as the target (Kojundo Chemical Laboratory Co. Ltd, Japan). The thicknesses of the NiO<sub>x</sub> films were controlled by regulating the deposition time from 20 min to 2 h. All procedures were carried out in room temperature (no intentional heating).

**Materials and Characterizations.** All chemicals were purchased from commercial suppliers and used as received, unless stated otherwise. Perovskite precursor solutions were prepared by dissolving PbI<sub>2</sub> (Kanto Chemical, 98% purity) in anhydrous *N,N*-dimethylformamide (400 mg mL<sup>-1</sup>), and methylammonium iodide (MAI) and methylammonium chloride (MACl) (Wako Chemicals, battery grade) in ethanol (50 mg mL<sup>-1</sup>, 19:1 ratio). PC<sub>61</sub>BM (Sigma-Aldrich, 99% purity) solution (2 wt %) dissolved in anhydrous chlorobenzene was used for coating the electron transport layer. All solutions were filtered through 0.45 μm syringe filters to avoid the risk of particle formation. AZO nanoparticle ink (Nano-grade N-21X) was used to prepare the AZO layer. The XRD patterns were collected using an X-ray diffractometer (Rigaku SmartLab, Japan) (Cu Kα radiation,  $\lambda = 1.54050$  Å). Top-surface and cross-sectional images were taken using a high-resolution scanning electron microscope (Hitachi, S-4800) at a 5 kV accelerating voltage carefully to avoid damage to the samples. XPS (ULVAC-PHI, VersaProbe II, Japan) was used to analyze the elemental composition of the NiO<sub>x</sub> films. The UV-vis absorption spectra were recorded on a UV-vis NIR spectrophotometer (Jasco V-7200). The resistivity of the films was measured by a linear four-probe method. The current density–voltage ( $J$ - $V$ ) characteristics (FF,  $R_s$ , and  $R_{sh}$ ) were analyzed by commercial software (SYSTEMHOUSE SUNRISE Corp.), and the incident monochromatic IPCE spectra and EQE were measured using a spectrometer (SM-250IQE; Bunko-keiki, Japan). For stability testing, the encapsulated devices were evaluated under 1 sun illumination (AM1.5G, no UV-light filtering) and MPPT condition using a solar simulator system equipped with a temperature-controlled oven (BIR-50; Bunko-keiki, Japan). A thermocouple was placed near the sample surface to monitor the testing condition. Histograms of 32 cells with NiO<sub>x</sub> HTL ( $70 \pm 3$  nm) prepared at 3.5 Pa were deduced from the devices, with an area of 0.19 cm<sup>2</sup> defined by an aperture mask.

**Device Fabrication.** A thin layer (~30 nm) of PEDOT:PSS (Clevios, Al4083) was formed by spin coating at 3000 rpm and subsequently dried at 120 °C for 15 min on a hot plate in ambient air. Sputter-deposited NiO<sub>x</sub> and PEDOT:PSS substrates were transferred to a nitrogen-filled glovebox ( $<1.0$  ppm of O<sub>2</sub> and H<sub>2</sub>O), inside which the remaining steps were performed. A PbI<sub>2</sub> film was spin-coated at 3000 rpm for 90 s and then a mixture of MAI and MACl was spun onto the PbI<sub>2</sub> layer at 4000 rpm for 90 s for Cl-mediated interdiffusion.<sup>44</sup> Then, to promote crystallization, those as-grown CH<sub>3</sub>NH<sub>3</sub>PbI<sub>3-x</sub>Cl<sub>x</sub> perovskite films were placed inside a Petri dish with MACl powders on a hot plate at 100 °C for

MAcI treatment.<sup>41</sup> A PC<sub>61</sub>BM layer was spun onto the perovskite layer at 700 rpm for 60 s, followed by coating with an AZO layer at 3000 rpm for 30 s. The samples were then transferred to the rf magnetron sputtering chamber outside the glovebox for metal contact (silver) deposition. Ag (150 nm) was sputtered at an Ar pressure of 0.15 Pa. The devices were sealed by encapsulation glasses and UV-curable resins (UV RESIN XNR5516Z; Nagase ChemteX, Japan) before measurement under ambient conditions and the stability test.

## ■ ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acsomega.7b00538.

Additional details of the photographs and SEM images of the NiO<sub>x</sub> films prepared at different Ar pressures of the deposition chamber and the calculation results on the acceleration factors of devices (PDF)

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### Notes

The authors declare no competing financial interest.

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