



Characterization and antioxidant activity of pectin from Indonesian mangosteen (*Garcinia mangostana* L.) rind



Nasrul Wathoni^{a,*}, Chu Yuan Shan^a, Wong Yi Shan^a, Tina Rostinawati^b, Raden Bayu Indradi^b, Rimadani Pratiwi^c, Muchtaridi Muchtaridi^c

^a Department of Pharmaceutics and Pharmaceutical Technology, Faculty of Pharmacy, Universitas Padjadjaran, Sumedang 45363, Indonesia

^b Department of Biology Pharmacy, Faculty of Pharmacy, Universitas Padjadjaran, Sumedang 45363, Indonesia

^c Department of Pharmaceutical Analysis and Medicinal Chemistry, Faculty of Pharmacy, Universitas Padjadjaran, Sumedang 45363, Indonesia

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ABSTRACT

Pectin, a natural polysaccharide, has gained increasing attention due to not only its biomaterial properties but also its biomedical activities. One of the abundant sources of pectin is mangosteen (*Garcinia mangostana* L.) rind. In this study, we characterized the pectin from Indonesian mangosteen rind extract and evaluated its antioxidant activity. Pectin was extracted in acid condition and evaluated its physicochemical properties by fourier transform infrared (FTIR), powder X-ray diffractometer (PXRD), water content, ash content, equivalent weight, methoxyl level and of galacturonic acid content. Furthermore, the antioxidant activity of pectin was also observed by 2,2-diphenyl-1-picrylhydrazyl (DPPH) method. Pectin was successfully extracted from dry weight of Indonesian mangosteen rind with yield about $1.16 \pm 0.17\%$, fine powder, brownish and odorless. FTIR and PXRD results showed that pectin from mangosteen rind extract was amorphous and similar characteristic with a commercial pectin. The chemical properties of pectin such as water content, ash content, equivalent weight, methoxyl level and of galacturonic acid level were $9.85 \pm 0.12\%$, $3.91 \pm 0.17\%$, 6330.76 ± 220.43 g/mol, $2.86 \pm 0.05\%$ and $75.98 \pm 0.88\%$, respectively. In addition, pectin showed an antioxidant activity with the IC_{50} about 161.94 ± 31.57 ppm. These results suggest that pectin from Indonesian mangosteen rind has the potential properties as biopolymers for biomedical applications with a low-methylated pectin and a moderate antioxidant activity.

1. Introduction

Mangosteen (*Garcinia mangostana* L.) is one of the tropical fruits that has been widely used as a traditional medicine in Southeast Asia. Indonesia, the Southeast Asia country with the largest mangosteen producer in the world, has abundant mangosteen waste which can be used as pharmaceutical raw material [1, 2]. The important part of mangosteen fruits is rind which contains various bioactive compounds, such as phenolic acid, tannin, xanthone, anthocyanins [3, 4, 5, 6, 7], and pectin [8].

Pectin, a heterogeneous group of polysaccharides, mainly contains 1,4-linked- α -D-galacturonic acid and presents in plant cell walls [9]. Recent studies proved that pectin has various biological properties, such as antioxidants, antitumor, and anti-inflammatory activities [10]. In addition, because of pectin properties in forming a stable gel with simple

gelling mechanism, the use of pectin as a matrix for biomedical applications has increased in the fields of drug delivery, tissue engineering, and wound dressing [11, 12, 13].

The outstanding quality of pectin has a more than 65% of galacturonic acid content [14]. Pectin from mangosteen rind has the highest galacturonic acid content (73.16%), meanwhile, the galacturonic acid content of pectin from lime and mango peels are 72.5% and 56.67%, respectively [15]. These properties lead pectin from mangosteen rind has a high potential to be extracted and investigated for the antioxidant activity.

Apparently, no studies discussed about the characterization of pectin along with its antioxidant activity from Indonesian mangosteen rind. Therefore, in this study, we characterized physicochemical properties of pectin from Indonesian mangosteen rind and evaluated its antioxidant activity using DPPH method to be developed for biomedical applications.

* Corresponding author.

E-mail address: nasrul@unpad.ac.id (N. Wathoni).

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2. Materials and methods

2.1. Materials

Mangosteen fruits were obtained from farmers in Puspahiang (Tasikmalaya, West Java, Indonesia, latitude: -7.422070, longitude: 108.044540). The fruit was harvested in 100 days after anthesis. A commercial pectin as standard was bought from Cargill (Jakarta, Indonesia). Phenolphthalein indicators were obtained from Asia Chemical (Jakarta, Indonesia). 2,2-diphenyl-1-picrylhydrazyl (DPPH) and ascorbic acid were acquired from Sigma Aldrich (Jakarta, Indonesia). All reagents were procured from Emsure (Jakarta, Indonesia).

2.2. Extraction of pectin from mangosteen rind

The rind of mangosteen fruits were dried at 60 °C and milled to 250–500 µm size. 50 g of mangosteen powder was added to 250 mL of distilled water, and acidified with H₂SO₄ at pH 2 in 90 °C for 120 min. The filtrate was then coagulated with 96 % ethanol (ratio 1:1) for 10 h, then, the precipitate was separated by cloth filters. Afterward, the precipitation was washed with ethanol and acetone to remove organic solvents. The pectin was dried in oven at 40 °C for 6 h until constant weight and the pectin yield was calculated. The percentage of yield was determined from the weight ratio of pectin produced and dry raw material [16].

$$\text{Galacturonic acid levels (\%)} : \frac{(\text{meq from NaOH release carbon} + \text{meq from NaOH for methoxyl}) \times 176 \times 100}{\text{weight sample (mg)}} \quad (3)$$

2.3. Scanning electron microscope analysis

Pectins were put on aluminum stubs and coated with gold-palladium for 10 seconds, and analyzed by SEM (SU3500 Hitachi, Tokyo, Japan) [17].

2.4. Fourier-transform infrared spectroscopy analysis

Pectins (2 mg) were mixed with 300 mg of dry KBr crystals and printed with a rotary vacuum pump. The pellet of KBr was scanned by FTIR (IR Prestige-21 Shimadzu, Tokyo, Japan) in the range of 400–4000 cm⁻¹ [16].

2.5. X-ray diffractometric analysis

The X-ray diffraction (XRD) patterns were acquired by an X-ray diffractometer (Bruker, Yokohama, Japan). Briefly, pectins were irradiated with filtered Cu-K (α) at a voltage of 40.0 kV and a current of 40.0 mA. The scanning level was 2°/min at a 2θ diffraction angle which ranged from 5° to 60° [18].

2.6. Determination of moisture and ash contents

A total of 0.5 g of pectin samples were dried in oven (Memmert UN55, Shanghai, China) at 100 °C for 4 h until constant weight. To determine ash content, porcelain crucible was spread in the furnace at a temperature of 600 °C then cooled in a desiccator and weighed as the weight of the container. As much as 0.5 g of pectin was weighed and put in the silicate crucible, then, added to the furnace with a temperature of 600 °C for 4 h until constant weight [19].

2.7. Determination of equivalent weight, methoxyl level, and galacturonic acid level

The equivalent weight was measured by weighing 0.25 g of pectin in a tube and moistened with 2.5 mL ethanol. One gram of NaCl, carbon free distillation water (100 mL), and six drops of phenolphthalein indicator were added. The mixture was then stirred quickly and the titration was done slowly until the indicator color changed to pink (pH 7.5). The neutralized solution was used to measure methoxyl. The determination of methoxyl content was carried out by adding 12.5 mL of 0.25 N NaOH to a neutral solution, then shaken correctly and left for 30 min at room temperature in a closed Erlenmeyer. Afterward, 12.5 mL of 0.25 N HCl and the phenolphthalein indicator were added and titrated with 0.1 N NaOH titrant until the solution turned to pink [19]. Galacturonic level (% AUA) was calculated from the volume of NaOH obtained from the determination of BE (equivalent weight) and methoxyl content as shown in Eqs. (1), (2), and (3) [20].

$$\text{Equivalent Weight} : \frac{\text{weight pectin(mg)}}{\text{ml NaOH} \times \text{N NaOH}} \quad (1)$$

$$\text{Methoxyl levels (\%)} : \frac{\text{ml NaOH} \times 31 \times \text{N NaOH} \times 100}{\text{weight sample(mg)}} \quad (2)$$

2.8. Antioxidant activity of pectin

Ascorbic acid was used as the reference for antioxidant activity of pectin. 2 mL of pectin solutions in ethanol (500 µg/mL) were reacted with a DPPH solution (40 µg/mL) in 5 mL of volumetric flask and then incubated for 30 min. The absorbance was measured at 400–600 nm by UV-visible spectrophotometer (Analytic Jena Specord 200, Germany). The antioxidant activity was carried out with the variation of concentration of 2, 4, 8, 16, and 32 µg/mL for ascorbic acid and 500, 1000, 1500, 2000, 2500 µg/mL for pectin. In this case, ethanol was used as blank reagent. The IC₅₀ level was calculated from the linear regression curve and Eq. (4) below [21]:



Fig. 1. Macroscopic image of pectin from mangosteen rind extract.

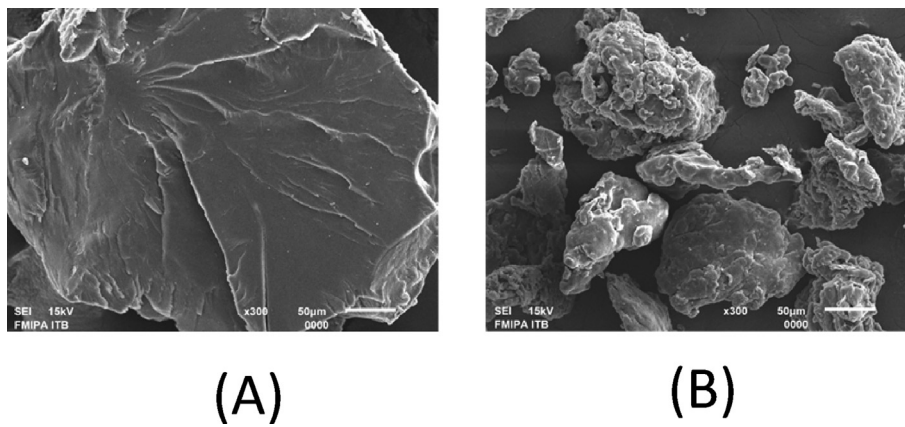


Fig. 2. SEM images of pectins (A) standard pectin (B) extracted pectin.

$$\% \text{ inhibition} = 1 - (\text{Absorbance sample} / \text{Absorbance blank DPPH}) \times 100\% (4)$$

3. Results

3.1. Characterization of pectin from mangosteen rind extract

To obtain mangosteen powder, we successfully extracted pectin from mangosteen rind with a fine powder, a brownish color and odorless (Fig. 1). The yield of extraction was $1,16 \pm 0,17\%$.

We next investigated the morphology of the pectin by using SEM. Based on the results in Fig. 2 show that the pectin from mangosteen peel extract had the smaller in size than the standard.

To confirm the functional group of pectin from mangosteen rind extract and standard pectin, pectin was analyzed by FTIR. The result in Fig. 3 describes that extracted pectin and standard pectin had similar FTIR spectrums. PXRD analysis was performed to investigate the crystallinity of the pectins. Fig. 4 describes that the X-ray diffraction patterns of extracted pectin and standard pectin were amorphous.

To observe the purity of pectins, we next analyzed moisture content, ash content, equivalent weight, methoxyl level, and galacturonic acid level as shown in Table 1.

3.2. Antioxidant activity of pectin

In this examination, DPPH solution was stable for 30 min at the

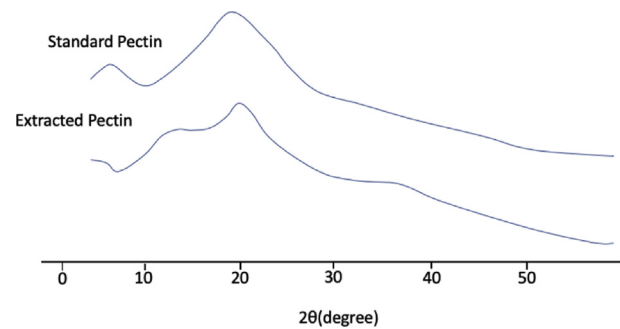


Fig. 4. PXRD patterns of pectins (A) standard pectin (B) extracted pectin.

Table 1
Purity of pectin from mangosteen rind extract.

Parameter	Result
Moisture content (%)	$9.85 \pm 0.12\%$
Ash content (%)	$3.91 \pm 0.17\%$
Equivalent weight (g/mol)	$6330.76 \pm 220.43 \text{ g/mol}$
Methoxyl level (%)	$2.86 \pm 0.05\%$
Galacturonic acid level (%)	$75.98 \pm 0.88\%$

maximum wavelength at 516 nm. The antioxidant activity result as shown in Table 2. The ascorbic acid had the IC_{50} value of 3.39 ± 0.20 and 161.93 ± 31.57 for pectin from mangosteen rind extract.

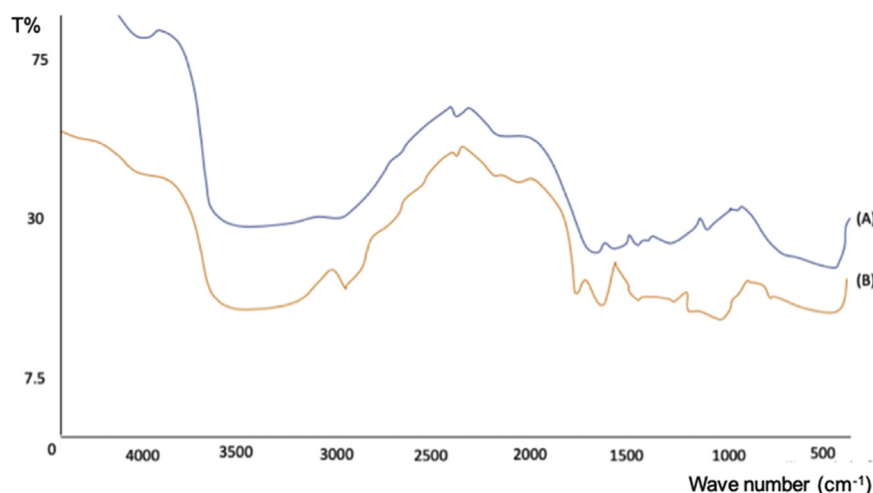


Fig. 3. FTIR spectra of pectins (A) standard pectin (B) extracted pectin.

Table 2
IC50 values of ascorbic acid and extracted pectin.

Ascorbic Acid			Pectin from mangosteen rind extract		
Concentration (ppm)	% Inhibition	IC ₅₀	Concentration (ppm)	% Inhibition	IC ₅₀
2	33.30	3.39	500	74.15	161.93
4	53.39	±	1000	86.09	± 31.57
8	82.38	0.20	1500	86.88	
16	92.70		2000	91.49	
32	96.53		2500	95.17	

4. Discussion

4.1. Characterization of pectin from mangosteen rind extract

To obtain the optimum yield of pectin, the extraction of pectin was done by heating at low pH. This condition increased the hydrolysis of protopectin to pectin [21]. The pectin yield from mangosteen rind extract was $1.16 \pm 0.17\%$. The yield and physicochemical properties of pectin depend on the sources and the extraction process [22].

The physicochemical properties of pectin from Indonesian mangosteen rind were observed to compare with a commercial pectin standard. Firstly, to observe the morphology and particle size of the pectin, we next analyzed pectin by SEM. Based on Fig. 2, the surface of control was smoother than that of pectin from mangosteen rind extract. In addition, extracted pectin had a surface with relatively smaller particles compared to standard pectin. The standard pectin is a marketable pectin which has been modified by the addition of sugar, therefore, the sugar may affect its morphology. Interestingly, Malviya and Kulkarni identified pectin from mango peel extract and it had similar morphology with pectin from Indonesian mangosteen rind [23].

We next investigated the chemical structure of pectin by FTIR analysis (Fig. 3). The major peaks of hydroxyl group were appeared at 3402.43 cm^{-1} and 3367.71 cm^{-1} for pectin and standard pectin, respectively. The aliphatic C–H groups and C=O carbonyl group were shown in 2927.94 cm^{-1} and 1747.51 cm^{-1} for pectin, and 2939.52 cm^{-1} and 1639.49 cm^{-1} for standard pectin. The peak at 1442.7 cm^{-1} describes the presence of a CH methyl group for both of extracted pectin and standard pectin. In addition, 1161.15 cm^{-1} and 1153.43 cm^{-1} indicate the presence of C–O bonds in alcohols, esters, and carboxylic acids. These results suggest that FTIR spectrums of extracted pectin was similar with pectin standard. The result of X-ray diffraction analysis corroborated the FTIR study which indicated amorphous system in both extracted pectin and standard pectin (Fig. 4).

4.2. Purity of pectin

The quality of pectin is necessary to be investigated by observing its purity. The purity of pectin was analyzed by calculating the moisture content, ash content, equivalent weight, methoxyl level and galacturonic acid level. The result of chemical properties of pectin from mangosteen rind is shown in Table 1. The moisture and ash content of pectin were $9.85 \pm 0.12\%$ and $3.91 \pm 0.17\%$, respectively. The moisture content is closely related to the water content in mangosteen rind sources. Mai and Dui successfully extracted pectin from vietnamese mangosteen with moisture content of 4.64% [8]. The ash content represents the purity of the pectin. The lower the ash content, the higher the purity of the pectin [20]. In addition, the ash content in pectin extracted from dragon fruits were varied from 7 to 10% depend on extraction time [16]. These results suggest that pectin from Indonesian mangosteen rind has a high moisture content and a high purity of pectin.

Equivalent weight is a measurement of the free galacturonic acid content in the pectin. Pure pectin acid, a pectate, is composed entirely of polygalacturonic acid which is free of methyl ester groups [19]. The equivalent weight of pectin depends on the type of plant, quality of raw

material, method of extraction and extraction process. Pectin extracted from the Indonesian mangosteen rind had a great equivalent weight, which was $6330.76 \pm 220.43 \text{ g/mol}$.

Methoxyl and galacturonic acid content describe the quality of pectin. Their level in pectin can affect the structure and texture of the pectin gel formed [24]. The methoxyl level and galacturonic acid contents of pectin were $2.86 \pm 0.05\%$ and $75.98 \pm 0.88\%$, respectively. These results were in acceptable range as stated in Indonesian Pharmacopeia and Handbook of Pharmaceutical Excipient ($<6.7\%$ for methoxyl level and $\leq 74\%$ for galacturonic acid) [25,26]. In addition, these result suggest that the methoxyl level of pectin from Indonesian mangosteen rind extract is low-methylated pectin [27]. Low-methoxy pectin is gelled with calcium ions and independent on the presence of acid or high solids content [25]. On the other hand, the galacturonic acid level is quite good in quality.

4.3. Antioxidant activity of pectin

Antioxidant effect of pectin is important for biomedical application. Therefore, we determined the antioxidant activity of pectin by DPPH method. The compound 1,1-diphenyl-2-picryl-hydrazil (DPPH) is a free radical molecule that is stable with the delocalization of electrons around its molecules, therefore the DPPH solution is purple. When reacting with antioxidant compounds, DPPH will be reduced and change to yellow due to the formation of DPPH-H. DPPH acts as a hydrogen donor for antioxidant compounds [28]. Pectin has been found to have antioxidant activity depending on the structure. It predict that hydroxyl groups of polysaccharides such as pectin can show good antioxidant activity when the viscosity is not too high [29]. The result of antioxidant activity in Table 2 show that pectin from Indonesian mangosteen rind extract was moderate antioxidant with IC₅₀ value was $161.93 \pm 31.57 \text{ ppm}$ [30].

5. Conclusion

In conclusion, pectin was successfully extracted from Indonesian mangosteen rind and had a distinctive characteristic. It was amorphous form and classified as low-methylated pectin. The antioxidant activity of this pectin was moderate antioxidant compare to ascorbic acid. These results suggest that pectin from Indonesian mangosteen has a good quality to be developed as biomaterial for biomedical application.

Declarations

Author contribution statement

Nasrul Wathoni: Conceived and designed the experiments; Analyzed and interpreted the data; Contributed reagents, materials, analysis tools or data; Wrote the paper.

Chu Yuan Shan & Wong Yi Shan: Performed the experiments; Analyzed and interpreted the data.

Tina Rostinawati, Raden Bayu Indradi, Rimadani Pratiwi & Muchtaridi Muchtaridi: Analyzed and interpreted the data; Wrote the paper.

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Competing interest statement

The authors declare no conflict of interest.

Additional information

No additional information is available for this paper.

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