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Shortwave Infrared Imaging with J-Aggregates Stabilized in Hollow Mesoporous Silica Nanoparticles

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Abstract

Tissue is translucent to shortwave infrared (SWIR) light, rendering optical imaging superior in this region. However, the widespread use of optical SWIR imaging has been limited, in part, by the lack of bright, biocompatible contrast agents that absorb and emit light above 1000 nm. J-Aggregation offers a means to transform stable, near-infrared (NIR) fluorophores into red-shifted SWIR contrast agents. Here we demonstrate that J-aggregates of NIR fluorophore IR-140 can be prepared inside hollow mesoporous silica nanoparticles (HMSNs) to result in nanomaterials that absorb and emit SWIR light. The J-aggregates inside PEGylated HMSNs are stable for multiple weeks in buffer and enable high resolution imaging in vivo with 980 nm excitation.

> Optical imaging with shortwave infrared (SWIR, 1000–2000 nm) light has emerged as a powerful method of fluorescence imaging in animals due to the superior resolution and contrast one can achieve with low energy light (Figure 1A).¹ A primary challenge with SWIR imaging is the development of bright, biocompatible, SWIR contrast agents.² Originally, the advantageous qualities of imaging in the SWIR region were showcased with carbon nanotubes,³ quantum dots,⁴ and rare earth nanomaterials.⁵ In efforts to set the stage for clinical translation, the past three years have seen a focus on the synthesis of nontoxic, SWIR-emissive organic fluorophores.⁶ This work has significantly expanded the suite of fluorophores that emit above 1000 nm; however, challenges remain in the stability, delivery,

Supporting Information

The Supporting Information is available free of charge on the [ACS Publications website](https://pubs.acs.org/) at DOI: 10.1021/jacs.9b05195. All experimental procedures, Figures S1–S22, Tables S1–S3, Schemes S1 and S2, Notes S1–S4 ([PDF](https://pubs.acs.org/doi/suppl/10.1021/jacs.9b05195/suppl_file/ja9b05195_si_001.pdf))

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and brightness of SWIR dyes. Consequently, we looked to explore an alternative avenue to create SWIR organic materials: J-aggregation.

J-Aggregation is the slip-stacked alignment of chromophores that leads to constructive coupling of the excited state transition dipoles (Figure 1B).⁷ The photophysical consequences of J-aggregation are bathochromically shifted absorption and emission spectra, narrow absorption and emission bands with small Stokes shifts, enhanced absorbance coeffcients (ε), and shortened fluorescence lifetimes which can result in enhanced quantum yields (Φ_F) and cycling rates. Many J-aggregate characteristics are beneficial qualities for in vivo imaging: red-shifted absorption and emission spectra will enable significant depth penetration during both the excitation and image acquisition, $2,8$ narrow bands can facilitate multi plexed imaging, and increased ε will result in bright materials. Despite the significant photophysical advantages J-aggregates typically have over the monomer, there are few reports of employing J-aggregates for *in vivo* imaging due to the diffculty in obtaining and stabilizing the necessary chromophore alignment in complex settings.⁹

Nanostructures can sequester and protect payloads, rendering nanomaterials a promising approach toward stabilizing J-aggregates in vivo. In 2016, Zheng and co-workers performed image-guided surgery with porphyrin lipids that formed J-aggregates upon self-assembly into nanovesicles.10 The following year, Xu and co-workers prepared pyrrolopyr-role cyanine J-aggregate-containing polymer micelles, which could be visualized after subcutaneous injection.¹¹ In work recently published, Fan and co-workers reported a squaraine J-aggregate, stabilized in polymeric micelles, for SWIR image-guided photothermal therapy.¹² Each of these reports utilizes self-assembled organic nanomaterials, ¹³ which are prone to disassembly when diluted in the presence of hydrophobic biomolecules, leading to destabilization of the J-aggregate.¹⁴ Here, we employ robust, biocompatible, hollow mesoporous silica nanoparticles to stabilize and protect SWIRemissive J-aggregates of IR-140 for in vivo imaging (Figure 1C,D).

Hollow mesoporous silica nanoparticles (HMSNs) have 2–4 nm pores that open into a large, 10–200 nm cavity, allowing these nanostructures to carry significant cargo.¹⁵ The surfaces of the HMSNs can be modified to alter the biodistribution of the nanoparticles.16 Consequently, there are numerous reports of HMSNs as the core scaffold of multifunctional materials. $15a,16b,17$ Included in these studies are the loading or conjugation of visible 18 and nearinfrared^{15a,19} fluorophores and administering the resulting nanomaterials for imaging. However, the controlled assembly of J-aggregates in HMSNs has yet to be demonstrated.

To realize SWIR-emissive J-aggregates inside HMSNs, we utilized the heptamethine dye IR-140 (1). IR-140 is a commercially available NIR fluorophore ($\lambda_{\text{max,abs}} = 826 \text{ nm}$, $\lambda_{\text{max, em}} = 875 \text{ nm}$) that has been applied as a photo-polymerization initiator,²⁰ fluorescent payload,²¹ component of plasmonic arrays,²² as well as a Raman²³ and two-photon²⁴ imaging agent. In 2016, Wang and Weiss reported that introduction of IR-140 to glutathionecoated quantum dots results in J-aggregate formation with two aggregates observed: J1, $(\lambda_{\text{max.abs}} = 965 \text{ nm}, \text{nonemissive})$ and J2 ($\lambda_{\text{max.abs}} = 1040 \text{ nm}, \lambda = 1047 \text{ nm}.^{25}$ We envisioned that similar IR-140 J-aggregates could be formed on the negatively charged pores

and inner surface of HMSNs. Further, once the aggregates were assembled inside the particles, the hydrophobic nature of IR-140 would make them unlikely to disassemble in aqueous environments, rendering J-aggregates stable in vivo.

We prepared HMSNs by synthesizing a mesoporous silica coating on a Stöber sphere core that was subsequently removed via etching with sodium carbonate (Scheme S1, Figure S1). The HMSNs were treated with varying amounts of IR-140 in different solvents (Figure 2A). J-Aggregate formation was assayed by UV/vis/NIR spectroscopy evaluating loss of monomeric IR-140 at 826 nm and formation of the J-aggregates at 965 nm (J1) and 1040 nm (J2). Upon optimization, we found that SWIR J-aggregates could be obtained when IR-140 dissolved in dimethyl sulfoxide (DMSO) was combined with HMSNs and washed. The washing procedure proved essential for obtaining the desired J2 aggregate formation (Figures 2B and S2), with gentle PBS washes yielding the largest amount of the desired J2 aggregate (dark blue, Figure 2B). When these optimized conditions were repeated on Stöber spheres that did not have pores or an inner surface for IR-140 to associate with, only a small J-band was observed (Figure 2B, gray line, Figure S3). Similar results were obtained when loading was performed on mesoporous silica coated Stöber spheres (Figure S2). These control experiments (Figure S5) suggest that the majority of IR-140 is protected inside the HMSN cavity. Through analysis of IR-140 collected after the washing procedures, we calculated the loading of IR-140 to be $\sim 10^3$ molecules/particle (Figure S6, Note S1).

The HMSNs were further characterized through transmission electron microscopy (TEM), which showed \sim 85 nm particles with a distinct cavity and pores (Figure 2C). The pore size was quantified to be 3.2 nm through nitrogen adsorption experiments (Figure S4). While the pores are clearly visible in the TEM of the empty HMSNs, they are darkened after treatment with IR-140 (Figure 2D), suggesting the presence of IR-140. Control experiments in which HMSNs were subjected to PBS washing procedures but no IR-140 show no change in contrast of the pores upon TEM analysis (Figure S7).

After confirming that the HMSNs could facilitate J-aggregation of IR-140, we modified the surface with poly-(ethylene glycol) (PEG) such that they could be suspended in aqueous media. This was accomplished by loading HMSNs that had undergone surface silanization with (3-aminopropyl)-triethoxysilane (APTS) prior to Stöber sphere and surfactant removal (Scheme S2). This procedure resulted in HMSNs that were positively charged on the outside but still contained a negatively charged interior to associate with the cationic IR-140.²⁶ Nitrogen adsorption data (Figure S4) also suggests that the pores are not modified with APTS.27 The introduction of IR-140 into the HMSNs-APTS proceeded similarly to the HMSNs, yielding analogous loading of IR-140 and a higher ratio of J2:J1 (Figure S8). Control experiments performed with Stöber spheres treated with APTS support that IR-140 is protected on the interior of the HMSNs (Figures S3 and S5). After loading, a 23 kDa PEG-carboxylate was conjugated to the amines present on the outer surface of the HMSNs-APTS using carbodiimide chemistry (Scheme S2). Successful PEG conjugation was verified by changes in hydrodynamic diameter and ζ-potential (Figures S9 and S10).

We evaluated the photophysical properties of the PEGylated HMSNs (HMSNs-PEG) containing IR-140 in comparison to IR-140 in solution as the monomer and J-aggregate

(Figure 3A). Monomeric IR-140 has been well-characterized;28 however, the solution Jaggregate of IR-140 had previously not been reported.²⁹ After screening numerous conditions (Figure S11), we found that 35% DMSO/0.9% NaCl in water afforded formation of the desired SWIR J-aggregate with a $\lambda_{\text{max,abs}} = 1042 \text{ nm}$, $\lambda_{\text{max,em}} = 1043 \text{ nm}$, $\varepsilon = 3.9 \times 10^{12} \text{ nm}$ 10^5 M⁻¹ cm⁻¹, and $\Phi_F = 0.01\%$ (Table S1, Notes S2 and S3). The IR-140-containing HMSNs-PEG had similar spectral properties with a $\lambda_{\text{max,abs}} = 1038 \text{ nm}$, $\lambda_{\text{max,em}} = 1047$ nm, although the absorbance was considerably broader, which we attribute to the presence of other nonemissive aggregate states. When solutions of IR-140 in DMSO, IR-140 in 35% DMSO/0.9% NaCl in water, and IR-140 loaded HMSNs-PEG in PBS were excited with a 980 nm laser, the wavelength to be used for in vivo imaging experiments, the IR-140 Jaggregate in solution and in the particles were similarly emissive, while the monomer was not excited by 980 nm light (Figures 3B and S12).³⁰ Thus, J-aggregation is essential for SWIR imaging with low energy excitation.

Next, we analyzed the role of the HMSNs in stabilizing IR-140 J-aggregates. Over 2 weeks in PBS at room temperature, we observed only $a \sim 10\%$ decrease in absorbance from the IR-140 loaded HMSNs-PEG and no evidence that the packing of the IR-140 within the nanoparticles was changing (Figures 3C, blue; S13). Comparatively, only ~8% of the Jaggregate in solution remained after 1 day (Figures 3C, red; S13). Not only do the HMSNs stabilize the assembly of the J-aggregate, but they also enhance the photostability. The fluorescence of solutions containing IR-140 J-aggregate in 35% DMSO/0.9% NaCl in water and HMSNs-PEG containing IR-140 J-aggregate in PBS were continually irradiated with a 980 nm laser (97 mW/cm²) and the fluorescence intensity was measured with an InGaAs camera. The photostability of monomeric IR-140 in DMSO was also evaluated via excitation at 785 nm (97 mW/cm²). As shown in Figure 3D, the J-aggregates within the HMSNs-PEG are 4-fold more stable than the J-aggregates in solution and ~60-fold more stable than the monomer (Table S2, Note S4). This result is consistent with the use of silica shells to overcome the poor photostability that is characteristic of J-aggregates by limiting the amount of reactive oxygen species that can access the aggregate.31 Photobleaching experiments in deoxygenated solvents support that the HMSNs protect the IR-140 J-aggregate from reactive oxygen species (Figure S14). Taken together, our data show that the HMSNs are critical for stabilizing J-aggregates to light and solution.

Finally, with bright SWIR-emissive nanoparticles prepared and characterized, we evaluated their biocompatibility and in vivo imaging performance. In vitro studies showed no cytotoxicity of the IR-140 loaded HMSNs-PEG over 6 h at concentrations up to 200 μg/mL (Figure S15). These data are consistent with other studies regarding mesoporous silica, which is generally considered nontoxic to animals.^{26,32} We performed in vivo imaging experiments using the IR-140 loaded HMSNs-PEG with excitation at 980 nm and collection from 1000–1700 nm. The SWIR-emissive HMSNs-PEG were intravenously injected into nude mice and the mice were immediately imaged (Figure 4). The HMSNs-PEG rapidly clear from the bloodstream and intense signal can be seen in the lungs, liver, and spleen. Fifty minutes after injection, the signal intensity within these organs remained constant (Figure S16).

In summary, we have presented J-aggregation as an approach to prepare biocompatible, SWIR contrast agents and demonstrated this concept by stabilizing J-aggregates of the NIR fluorophore IR-140 inside HMSNs. The bathochromically shifted absorption and emission and small Stokes shifts of the IR-140 J-aggregate allow imaging with 980 nm excitation and 1000–1700 nm acquisition, providing high resolution in vivo images. The modularity of the HMSNs will enable facile exchange of the imaging agent as well as the addition of targeting agents and/or therapeutics, poising these materials to become SWIR theranostics.³³ While we did not observe an enhanced Φ_F with the IR-140 J-aggregate, likely due to disorder or intermolecular vibrations, 34 work is ongoing to access a SWIR J-aggregate that exhibits the superradience phenomena predicted by Kasha.³⁵ Collectively, the use of J-aggregates stabilized in HMSNs as SWIR imaging agents has the potential to overcome the stability, toxicity, and brightness challenges of contrast agents for this compelling region of the electromagnetic spectrum.

Supplementary Material

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Figure 1.

(A) Regions of the electromagnetic spectrum employed for optical imaging. For further details on contrast and resolution within regions of the SWIR, see references 1b, 1c, 8b, 8c. (B) J-Aggregation and characteristic photophysical properties. (C) IR-140. (D) Work reported herein: the stabilization of IR-140 J-aggregates in hollow mesoporous silica nanoparticles (HMSNs) to result in biocompatible SWIR-emissive contrast agents.

Figure 2.

(A) Schematic of loading IR-140 into HMSNs. (B) Washing conditions facilitate Jaggregation. 10 mg/mL HMSNs were combined with 10 mM IR-140 in DMSO and washed with PBS with (green) and without (dark blue) sonication. Prewash spectrum, diluted 1:350 is shown in orange. Loading control for solid, nonporous Stöber spheres is shown in gray. (C,D) TEM images of HMSNs with (D) and without (C) IR-140 treatment.

Figure 3.

(A) Normalized absorption and emission of IR-140 J-aggregate in HMSNs-PEG (blue), Jaggregate in solution (red), and monomer (yellow). (B) Emission (1000–1700 nm) of IR-140 monomer (left), J-aggregate in solution (middle) and J-aggregate in HMSNs-PEG (right) upon 980 nm excitation. (C) Normalized relative absorption of IR-140 J-aggregate in 35% DMSO/0.9% NaCl in water (red) and in HMSNs-PEG in PBS (blue) on day zero (solid) and day 1 or 14 (dotted). (D) Photostability under laser irradiation (97 mW/cm²) at 980 nm for IR-140 J-aggregate in HMSNs-PEG (blue) and IR-140 J-aggregate in 35% DMSO/0.9–% NaCl in water (red), and at 785 nm for monomer in DMSO (yellow).

Figure 4.

Whole-mouse imaging at 16 fps (980 nm, 91 mW/cm² excitation; 1000–1700 nm collection) upon i.v. delivery of IR-140 HMSNs-PEG. Background subtracted stills were averaged over 5 frames at 3 s (A), 8 s (B), 25 s (C), and 120 s (D) postinjection. Scale bar represents 1 cm. Data are representative of two replicate experiments.