

«Review»

Selenium in Poultry Nutrition: from Sodium Selenite to Organic Selenium Sources

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Selenium (Se) is an essential element in poultry nutrition and its bio-efficacy depends on its chemical form. A growing body of research proves that organic forms of Se, mainly selenomethionine (SeMet), in poultry diets have a range of important advantages over traditional sodium selenite. In fact, the organic Se concept considers SeMet as a storage form of Se in the chicken body. As chickens are not able to synthesize SeMet, its provision through diet is a key strategy to fight commercially relevant stresses. Indeed, in stress conditions, when increased selenoprotein expression requires additional Se, while its provision via feed usually decreases due to a reduction in feed consumption, Se reserves in the body (mainly in the muscles) could help maintain an effective antioxidant defense and prevent detrimental consequences of stresses. The poultry industry is looking for the most effective sources of organic Se for commercial use. In this review, advantages and disadvantages of main organic Se sources for poultry (Se-yeast, SeMet, and OH-SeMet) are analyzed, and future directions for the development of new Se sources are identified.

Key words: chicken, OH-SeMet, poultry, selenium, SeMet, Se-yeast

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Introduction

Selenium (Se) is an essential element for poultry and in farm animal nutrition that was discovered by the Swedish chemist Berzelius 200 years ago. Interest in this enigmatic element is growing since 1957, when its essentiality was described by Schwarz and Foltz (1957, 1958; Fig. 1). Severe Se deficiency has been shown to be associated with the development of various disorders that do no longer occur in modern commercial poultry industry. However, decreased productive and reproductive performance due to suboptimal levels of dietary Se and inadequate antioxidant defenses in stress conditions can still be observed in farms. The introduction of modern genetics to the poultry industry has substantially improved the growth rates of chickens. However, a major downside of such improvements in performance is that the birds are often highly sensitive to various

stresses. Therefore, in modern poultry production, there is an important movement from prevention of Se deficiency to meeting the exact Se requirement of birds for optimizing their performance. In particular, the discovery and characterization of major selenoproteins, as well as a better understanding of the relationships between different antioxidants of the antioxidant system, provide new insights in this area. It is generally accepted that in biological systems, Se participates in various biochemical pathways and physiological functions as an integral part of a range of important selenoproteins. In chickens, 26 genes encoding different selenocysteine (SeCys)-containing proteins have been identified (Lei, 2017; Zhao *et al.*, 2017). Interestingly, more than half of the known selenoproteins are directly or indirectly involved in antioxidant defenses and maintaining redox balance in the cell (Fig. 2). For example, selenoproteins are involved in glutathione-dependent hydroperoxide removal, reduction of thioredoxins, selenophosphate synthesis, activation and inactivation of thyroid hormones, thioredoxin-dependent repair of oxidized methionine residues, endoplasmic reticulum-associated protein degradation, and other important biochemical processes. This explains the role of Se

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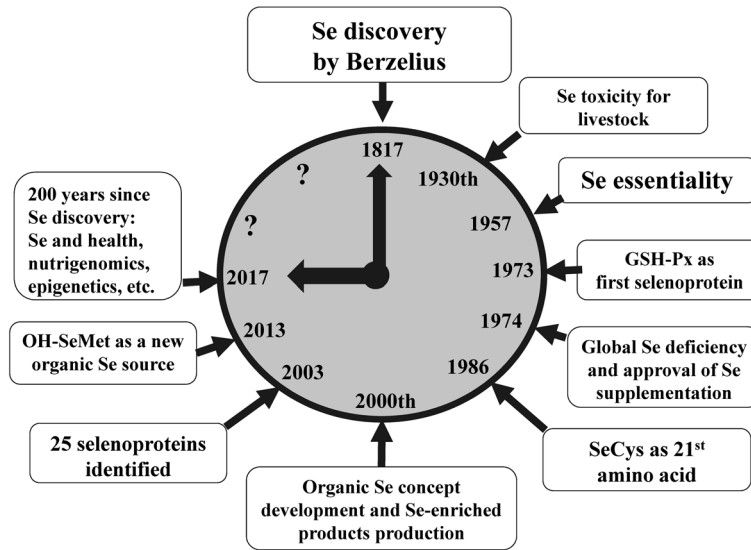


Fig. 1. **History of Se research and usage.** Selenium was discovered by the Swedish chemist Berzelius in 1817. In 1930, Se toxicity for livestock was described (Surai, 2006). In 1957, Se essentiality was discovered by Schwarz and Foltz (1957, 1958). The first selenoprotein, GSH-Px was described by Rotruck *et al.* (1973). In 1970, a global Se deficiency in livestock was admitted and the FDA approved Se supplements for poultry and swine in 1974 in the form of selenite or selenate. In 1986, SeCys was identified as the 21st amino acid encoded by the stop codon TGA (Chambers *et al.*, 1986). The organic Se concept was developed in 2000 and a range of Se-enriched products appeared on the market (Surai, 2006). In 2003, mammalian proteomes were characterised and 25 selenoproteins were identified (Kryukov *et al.*, 2003). Later, 26 genes encoding different selenoproteins were identified (Lei, 2017; Zhao *et al.*, 2017). Organic selenium sources found their way into animal/poultry nutrition, and a new effective source of organic Se (OH-SeMet) combining major advantages of Se-yeast and pure SeMet was successfully tested and found its way to poultry/animal industry (Briens *et al.*, 2013, 2014; Jjali *et al.*, 2013). Two hundred years have passed since the discovery of Se, and interest in the chemistry, biochemistry, and practical application of this element in poultry/animal industry increases steadily.

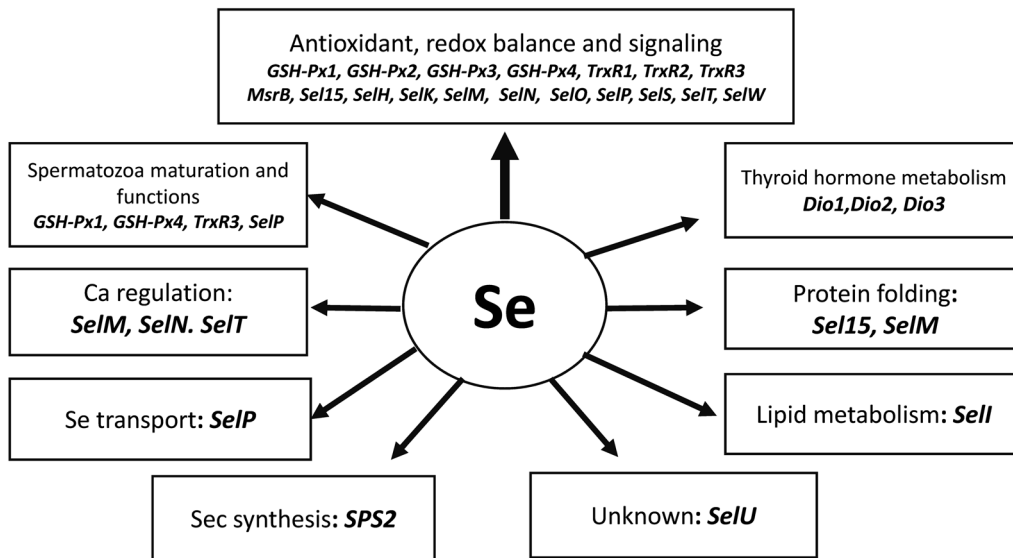


Fig. 2. **Established and suggested selenoprotein functions.**

in animal health, including gut health and immune system regulation. Selenoprotein expression is characterised by high tissue specificity, depends on Se availability, can be regulated by hormones, and contributes to various pathological conditions if compromised (Surai, 2006).

During the last 40 years, a mandatory approach to regulate Se supplementation in commercial feed has been used in the poultry industry worldwide. To meet Se requirements, the poultry industry has relied completely on supplemental Se delivered with premixes (usually at 0.2–0.3 mg/kg diet), while the Se levels in the feed ingredients have not been taken into account (Surai, 2006; Surai and Fisinin, 2014).

Organic Se Concept Development

A growing number of studies over the last three decades have suggested that the dietary form of Se is a major determinant of its efficiency in meeting the Se requirement in poultry. There are two major Se sources for poultry: inorganic Se, mainly selenite or selenate, and organic Se, mainly in the form of selenomethionine (SeMet). In major feed ingredients, including grains, soya, and oilseed, Se is found exclusively in organic forms (with SeMet comprising more than 50% of total Se), while the major commercial Se supplements in use for the last 40 years are selenite and selenate (Surai, 2006). Studies have suggested that organic Se naturally occurs in the form of various seleno-amino acids in poultry diets and it seems likely that the digestive system adapted to these organic Se forms during evolution, thereby explaining principal differences in assimilation and metabolism between organic and inorganic forms of Se (Surai, 2006; Surai and Fisinin, 2014).

In 1974, the FDA approved Se supplements for poultry and swine in the form of selenite or selenate (Surai, 2006). Since then, a great body of evidence showing disadvantages of using inorganic Se in poultry nutrition has been accumulated.

Firstly, nutrients in the feed interact with each other, and sodium selenite can be reduced to the unavailable form (elemental Se) by various nutrients possessing reducing activity. For example, the chemical reaction between sodium selenite and ascorbic acid (AA) in premix, feed, or gut can lead to selenite reduction to elemental Se, which cannot be absorbed in the digestive tract of chickens, and the oxidation of AA, which thereby loses its biological activity (Robinson *et al.*, 1985; Ip, 1986; Gosetti *et al.*, 2007). In fact, pink particles in the premix after long storage very often represent elemental Se produced as a result of the aforementioned reaction. This was clearly demonstrated when solutions of selenite and AA were mixed in proportions similar to those in supplements: only about 50% selenite was recovered after 2 h, and selenite was not detectable after 24 h storage (Gosetti *et al.*, 2007). A similar reaction was shown to occur in the chicken gut (Mykkänen and Mutanen, 1983). When premixes containing sodium selenite were prepared with glucose monohydrate, cornstarch, or sucrose and stored at room temperature, changes in odor and/or color were observed (Groce *et al.*, 1973). The odor was musty and sweetish in character,

while pink to dark-red particles appeared in the originally white matrix. Furthermore, when pigs were fed old premix, Se retention decreased, while Se excretion substantially increased. Decreased Se selenite availability due to its reduction has been also demonstrated in other studies. For example, in rats, the protective effect of selenite on tumorigenesis was prevented by vitamin C, due to the formation of unavailable elemental Se, while the chemopreventive action of SeMet was not affected (Ip, 1986). In a human trial, when selenite and AA were administered together, Se availability was dramatically reduced (Robinson *et al.*, 1985). In contrast, it seems likely that AA can enhance SeMet assimilation from the diet (Mutanen and Mykkänen, 1985).

Secondly, sodium selenite can be lost as a vapor after its conversion to a volatile form. For example, it has been shown that selenite can be dissolved when dispersed in feeds with relatively high water activity. When dissolved, it may form selenious acid and disperse as a vapor (Eisenberg *et al.*, 2007).

Thirdly, pro-oxidant effects of sodium selenite (Spallholz, 1997; Drake, 2006; Xiang *et al.*, 2009; Brozmanová *et al.*, 2010) can have detrimental consequences in the chicken gut. It has been recently shown that sodium selenite at 0.3 ppm can damage the gut structure in chickens. Chickens fed 0.3 ppm sodium selenite showed vacuolar degeneration in the epithelial cells lining of the intestinal crypts of the duodenum and excess mononuclear cell infiltration and aggregation in between degenerated and necrotic intestinal glands in the ileum (Attia *et al.*, 2010). In great contrast to sodium selenite, SeMet has been suggested to possess antioxidant properties (Schrauzer, 2000; Suryo Rahmanto and Davies, 2011).

Fourthly, sodium selenite is poorly transferred to the egg and developing embryo (Surai, 2006) and thus has a limited ability to improve antioxidant defenses against hatching-imposed oxidative stress (Surai *et al.*, 2016).

Finally, sodium selenite cannot be stored in Se reserves in the body, which are needed to maintain effective antioxidant defenses in stress conditions (Surai and Fisinin, 2014).

Thus, the use of sodium selenite in poultry diets has been questioned recently, and the organic Se concept has been developed and successfully introduced in the poultry industry (Surai, 2006; Fisinin *et al.*, 2008; Surai and Fisinin, 2014). Compared with inorganic Se, organic sources of Se appear to better meet the needs of modern poultry. Main advantages of organic Se sources for poultry have been recently reviewed (Surai and Fisinin 2014). In particular, advantages for breeders include better transfer to the egg and developing embryo, which results in stronger antioxidant defenses and benefits in terms of hatchability and viability of newly hatched chicks (Fisinin *et al.*, 2008; Surai and Fisinin, 2014). It has been suggested that Se delivered to the egg might have an epigenetic effect on the developing progeny, which awaits detailed investigation (Fisinin *et al.*, 2016). Beneficial effects of organic Se in layers are associated with the higher efficacy of Se transfer to the egg (Jlali *et al.*, 2013; Tufarelli *et al.*, 2016), which has a positive effect on internal egg quality (Haugh units) via the activation of methionine sulfoxide

reductase B (MsrB), a selenoprotein responsible for the prevention of protein oxidation and for maintaining the water-holding capacity of albumen. It has been suggested that the reversal methionine oxidation in proteins can alter their structure and functions, to regulate redox signaling (Kaya *et al.*, 2015). Therefore, the reduction of oxidized methionine into an active form by MsrB is considered to be key to the prevention/repair of detrimental consequences of oxidative stress on protein structure and functions (Lee, 2016). Furthermore, Se has a positive effect on eggshell quality, including shell breaking strength, via the modulation of organic matrix formation (Surai, 2006). Advantages of organic Se in the broiler diet include increased Se transfer to the muscles and the build-up of Se reserves in the body, which is expected to result in improved chicken resistance to various stresses and to have a positive effect on immunity, gut health, and meat quality (Surai, 2006; Fisinin *et al.*, 2008; Surai and Fisinin, 2014). In particular, it has been recently shown that Se deficiency in chickens promotes gut inflammation (Gao *et al.*, 2016; Wu *et al.*, 2016). Furthermore, specific emphasis has been placed on protein oxidation in relation to meat quality, including water-holding capacity of meat and drip loss. Indeed, increased activity of MsrB in the muscles as a result of improved Se status owing to the usage of hydroxy-selenomethionine (OH-SeMet) in the diet led to a decrease in protein oxidation (Zhao *et al.*, 2017) and could reduce drip loss. Se has been suggested to have immunomodulatory properties as well, which remain to be explored (Surai, 2006).

There are two important questions to be answered. First, what is an optimal Se concentration in the chicken egg to provide maximal protection against oxidative stress caused by the process of hatching? Second, how to achieve that concentration in commercial poultry production? The first question might be answered by analyzing the Se concentration in egg yolks from wild birds having a free choice of food. Se concentrations of 100–200 ng/g (Pappas *et al.*, 2005) or 300 ng/g wet yolk (Surai, 2000) have been found in egg yolk of chickens maintained on a standard diet (without supplementary Se and containing approximately 0.1 or 0.17

ppm Se respectively; Pappas *et al.*, 2005; Surai, 2000). In this case, Se is derived from the basic dietary constituents including wheat, maize, and soybean meal, and the levels in the body are quite variable depending on dietary composition and the place of origin of the feed ingredients. In our study, the level of Se in avian yolks was found to differ widely among free-living species, and yolk Se concentrations were substantially higher in eggs of wild species than in those of domestic chickens (Pappas *et al.*, 2006a). However, after organic Se supplementation at 0.4–0.5 ppm (Surai, 2000; Pappas *et al.*, 2005), Se levels in the chicken egg yolk were quite close to the levels found in wild birds, suggesting that there is scope for breeders to increase Se supplementation. Additional evidence of high Se concentrations in eggs from wild birds was found in little egrets, black-crowned night herons, and bridled terns from coastal areas of Hong Kong (Lam *et al.*, 2005). Furthermore, comparatively high Se levels were found in eggs from tree swallow, bank swallow, and house wren (Dickerson *et al.*, 2002). Similarly, in tissues of birds from various areas, including the Barents Sea, Alaska (Savinov *et al.*, 2003), and arctic Russia (Stout *et al.*, 2002), as well as in bald eagles from Adak Island, Alaska (Stout and Trust, 2002), Se concentrations were shown to be several fold higher than those found in domestic chickens. It is clear that effective Se dietary supplements are needed to meet the Se requirement of modern breeds of egg and meat types of poultry.

Accordingly, a range of supplements claimed to be organic Se sources can be found on the world market (Table 1). However, strictly speaking, only those products providing SeMet or its precursor could be considered as sources of organic Se, as SeMet is the only form that allows building Se reserves in the body (mainly in the muscles) (Surai, 2006). As avian species as well as other animals cannot synthesize SeMet, it has to be provided through diet. True organic Se sources include Se-enriched yeast (Se-yeast), SeMet, Zn-SeMet, and OH-SeMet, while Se chelates, glycinate, and proteinates should not be included into this category of supplements.

Of interest, Se supplements available for humans include

Table 1. Sources of organic Se available on the market

Source	Comments	SeMet*	References
Se-yeast	Well researched, 50–70% SeMet	Yes	Schrauzer, 2006
Se-Met	Not stable, >95% SeMet	Yes	Liu <i>et al.</i> , 2017
Zn-SeMet	Not stable, >95% SeMet	Yes	Geraert <i>et al.</i> , 2015
OH-SeMet	Stable, >95% OH-SeMet	Yes	Liu <i>et al.</i> , 2017
Se-proteinates	Chemistry is not proven	No	Liu <i>et al.</i> , 2017
Se-glycinates	Chemistry is not proven	No	Liu <i>et al.</i> , 2017
Se-chelates	Chemistry is not proven	No	Kubachka <i>et al.</i> , 2017
Se-homolanthionine	Lack of research data	No	Anan <i>et al.</i> , 2011
Other (nano-Se, etc.)	Lack of data on molecular mechanisms of action	No	Pelyhe and Mezes, 2013; Sarkar <i>et al.</i> , 2015

* As only SeMet can be non-specifically incorporated into proteins in measurable amounts, it is an active component of organic Se sources.

mainly sodium selenite, Se-yeast, and pure L-SeMet (Surai, 2006). Similar to animal studies, studies in humans have indicated that SeMet is the major metabolizable form of Se for humans (Schrauzer, 2000, 2001, 2003; Schrauzer and Surai, 2009). Sodium selenite, Se-yeast, and SeMet have shown cancer-preventive effects in various clinical trials (Surai, 2006). However, the most comprehensive clinical trial on cancer prevention by using Se (SELECT), where Se was supplemented in the form of L-SeMet (200 µg/day), did not show any benefit and was terminated prematurely (Nicastro and Dunn, 2013). The main lesson from the SELECT trial is that there is a need for further research to better understand the molecular mechanisms of Se action, elucidate an optimal dosage of Se supplementation, and determine the most effective forms of Se for dietary supplementation.

Building Se Reserves in the Body

After matrix digestion, similar to Met, ingested SeMet released from selenium-enriched yeast or pure SeMet is absorbed in the small intestine via a single, Na⁺-dependent, carrier-mediated process (Wolffram *et al.*, 1989). Part of the absorbed SeMet is trans-selenated to SeCys, which is converted to hydrogen selenide (H₂Se) by β-lyase, and H₂Se is further converted to selenophosphate (HSePO₃²⁻) by selenophosphate synthetase (SPS). SPS provides the active Se donor in the form of selenophosphate for the synthesis of SeCys, the 21st amino acid in the genetic code. The terminal reaction of SeCys synthesis is catalyzed by SeCys synthase when serine or phosphoserine is converted into SeCys (Dobosz-Bartoszek and Simonovic, 2016). SeCys synthesis involves complex cellular machinery, including specialized tRNA. In fact, selenophosphate first reacts with tRNA-bound serinyl residues to form SeCys-bound tRNA, and the incorporation of SeCys into selenoproteins occurs via UGA codon recognition (Gladishev *et al.*, 2016). In contrast to the other amino acids, no free pool of SeCys exists in the cell and only newly synthesized SeCys is incorporated into selenoproteins (Surai, 2006).

As mentioned above, humans and higher animals, including poultry, cannot synthesize SeMet, and it is the only seleno-amino acid that can be significantly stored in organs and tissues, building Se reserves in the body (Schrauzer, 2003, 2006). In contrast, there is no evidence of non-specific incorporation of Se into plasma proteins when administered as selenate or as SeCys (Surai, 2006). A growing body of evidence shows that SeMet is metabolized as a constituent of the methionine pool. In fact, in humans, approximately 46.9% of the total Se in the body is located in skeletal muscles, while only 4% of body Se is found in the kidneys (Oster *et al.*, 1988). In chicken breast and leg muscles, SeMet is the major Se form, comprising more than half (66.7% and 56.1%) of total Se (Bierla *et al.*, 2008). The authors also showed that in chickens fed a diet containing high doses of organic Se, the SeMet fraction in the muscles increased up to 99%, which confirmed non-specific incorporation of SeMet into muscle proteins. In muscles, the

content of Met is always in large excess over that of SeMet. For example, in humans, the SeMet/Met ratio in skeletal muscle is approximately 1:7,000, which is similar to the ratio of 1:>6,000 found in chicken breast muscle (Schrauzer and Surai, 2009). We calculated that in the egg yolk, the SeMet/Met ratio is approximately 1:160,000, and in egg white, it is close to 1:87,000. When organic Se is included in the breeder diet, this ratio can change substantially. It is interesting to note that SeMet comprises 53–71% of total Se in the egg albumen, while its proportion in the egg yolk (12–19%) is much lower (Lipiec *et al.*, 2010). This explains why albumin response to dietary organic Se is substantially stronger than that in the egg yolk (Surai, 2006). SeMet reportedly has a lower turnover rate in the body than sodium selenite and is characterised by greater Se re-utilization efficiency (Swanson *et al.*, 1991). The confirmation of chicken muscle being a storage site for Se came from a recent study by Brandt-Kjelsen *et al.* (2014) who showed that organic selenium has the longest half-life in muscles (12 days) and in the brain and lungs (13 days), while the shortest half-lives were found in the liver, kidneys, and pancreas (about 4 days). In comparison, the average half-lives of SeMet and selenite in the human body are 252 and 102 days, respectively (Patterson *et al.*, 1989). It was proven long ago that Se from both selenite and SeMet is readily available for the synthesis of the selenoenzyme glutathione peroxidase (GSH-Px) in rat tissues (Pierce and Tappel, 1977). Furthermore, it has been shown recently *in vitro* and *in vivo* that the Se availability from SeMet and sodium selenite is similar, while SeMet was more efficiently transported than sodium selenite in an *in vitro* membrane permeability study using Caco-2 cells (Takahashi *et al.*, 2017).

The idea that Se stored in tissues in the form of SeMet is available for selenoprotein synthesis has been proven in a range of studies (for review see Surai, 2006). In particular, chicks hatched from eggs enriched with Se were characterised by increased liver GSH-Px activity not only at hatching, but also at 5 days posthatch (Surai, 2000). Similarly, in quail (Surai *et al.*, 2006) and chickens (Pappas *et al.*, 2005), the Se concentration in the livers of the progeny was alleviated up to 2–4 weeks posthatch owing to organic Se in the breeder diet and increased Se levels in the egg. In addition, GSH-Px activity was increased in chicken muscle several weeks posthatch owing to maternal organic Se supplementation (Pappas *et al.*, 2005). In an elegant study with chickens, it was proven that endogenous Se accumulated in tissues through organic dietary Se supplementation was available for maintaining the GSH-Px activity when Se supplementation ceased (Payne and Southern, 2005). Similarly, in SeMet or Se-yeast supplemented mice, GSH-Px activities in the liver decreased slower during Se depletion than in mice given selenite (Spallholz and Rafferty, 1987). The half-life of GSH-Px was calculated to be 4.2 and 9.1 days in rats that received 3 ppm Se as selenite or SeMet, respectively (Ip and Hayes, 1989). In a human study, once supplements were withdrawn, GSH-Px activity in platelets (Levander *et al.*, 1983) or selenoprotein P concentration (Persson-Maschos *et*

al., 1998) decreased slower in the group given wheat Se than in the selenite-supplemented group. In New Zealanders who consumed high-Se bread, plasma Se remained elevated when Se supplementation ceased (Robinson *et al.*, 1985). Furthermore, Se-yeast was shown to provide a long-lasting body pool of Se in children (Alfthan *et al.*, 2000).

It seems likely that the protective effect of organic Se is most pronounced in stress conditions. Indeed, our previous study (Pappas *et al.*, 2006b) showed that mortality of chick embryos in week 3 of egg incubation was 3.5% for the control group and 10.6% in fish-oil supplemented (stressed) 27-week-old breeders. Inclusion of Se-yeast (0.4 ppm) in the breeder diets decreased mortality to 3.0% and 6.2%, respectively. In addition, stress associated with fish oil supplementation in the breeder diet reduced both hatchability and 1-day-old chick weight (Pappas *et al.*, 2006b). Again, the inclusion of organic Se into the breeder diet prevented some of these adverse effects. Therefore, Se reserves in the chicken body (mainly in the muscles) built as a result of non-specific

SeMet incorporation into the proteins in place of Met are considered to be an important adaptation of chickens to various stresses. This strategy might prevent decreases in productive and reproductive performance in stressful commercial conditions of chicken production (Surai and Fisinin, 2016 a, b). In general, the metabolic pathway of Se in birds (quails) is the same as that in rodents, but the metabolic capacity for Se seems to be larger in quails than in rodents. Indeed, the concentrations of exogenous Se in all organs and tissues of SeMet-administered quails were significantly higher than those in selenite-administered quails 3 h after the administration, suggesting that SeMet is more rapidly and/or efficiently incorporated into the quail body than selenite (Anan *et al.*, 2014). It should be noted that only a small portion of the methionine pool can be replaced with SeMet, and protein turnover prevents accumulation of SeMet to toxic levels in the organism (Schrauzer, 2003). A model of organic Se action in poultry is shown in Fig. 3.

The mechanisms regulating SeMet conversion to H₂Se and

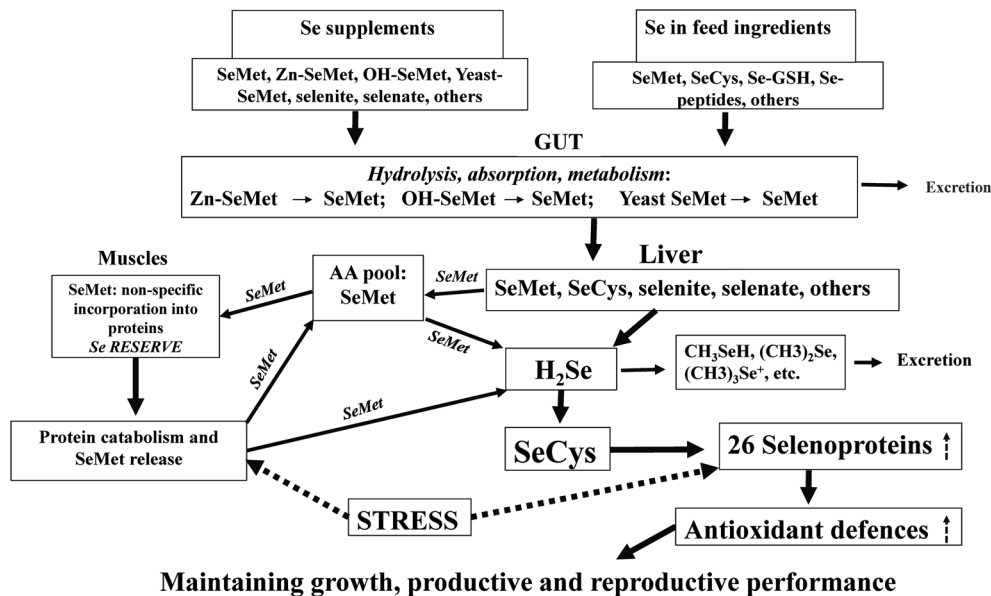


Fig. 3. Organic Se in action. A range of Se-containing compounds, including selenite, selenate, SeMet, Zn-SeMet, OH-SeMet, Se-yeast, SeCys, Se-GSH, and Se-peptides, can be included into premixes. All those Se forms come to the intestine where initial hydrolysis (Se-Met will be released from Se-yeast or Zn-SeMet; OH-SeMet will be converted into SeMet) and some metabolic changes will take place. This includes excretion of Se metabolites via bile, feces, and urine. Further, selenite, selenate, SeMet and some other Se forms will be delivered to the liver for metabolization and distribution. In parallel, a fraction of SeMet will go to the free amino acid pool and be used for building Se reserves mainly in muscles. The next step of Se assimilation and metabolism includes the conversion of all major forms of Se into H₂Se, from which SeCys will be synthesized and incorporated into 26 newly synthesized selenoproteins that are integral part of the antioxidant system of the body. Under stress conditions, protein catabolism will take place, which will release some SeMet incorporated into those proteins, and this SeMet will be converted into H₂Se and further into newly synthesized SeCys and 26 selenoproteins. Additional sources of Se will be responsible for the upregulation of selenoprotein genes and additional synthesis of selenoproteins, which will upregulate antioxidant defenses and will aid the body to adapt to and overcome the stress with minimal negative consequences. When only selenite is present in the diet, Se reserves in the muscles will not be built and therefore, the ability of the body to adapt to stress will be restricted.

further to SeCys and respective selenoproteins are not clear at present, but it seems likely that changes in redox status of the cells/tissues and activation of proteasomal protein degradation are involved. Indeed, ATP- and ubiquitin-independent proteolysis by the 20S proteasome is responsible for the selective degradation of oxidized proteins. *In vitro*, the 20S proteasome shows increased proteolytic activity toward oxidized polypeptides. In cells overexpressing GSH-Px-1, chymotrypsin-like activity of the proteasome was decreased by 30% (Kretz-Remy and Arrigo, 2003). This observation was correlated with a 2-fold increase in the half-life of κ B- α , a protein whose basal turnover is 20S proteasome-dependent. Furthermore, following exposure to H₂O₂, human T47D cells overexpressing GSH-Px showed Se-dependent decreases in intracellular ROS accumulation and 20S proteasome chymotrypsin-like activity. Moreover, exposure of HeLa cells to antioxidant compounds reduced the proteasome 20S chymotrypsin-like activity. These results suggest that GSH-Px activity or pro-reducing conditions can down-regulate basal 20S proteasome activity (Kretz-Remy and Arrigo, 2003). This implies that selenium is a key element in proteasome activity control. There might be a feedback mechanism of recognition of SeMet as a source of Se for selenoprotein synthesis. In stress conditions, some amino acids inside muscle proteins can be oxidized (Davies, 2016), and this will trigger an increase in proteasome activity to degrade such proteins (Pajares *et al.*, 2015) and release SeMet to be available as an additional source of Se for selenoprotein synthesis. When the antioxidant/pro-oxidant equilibrium is restored, increased GSH-Px activity would decrease proteasome activity and protein degradation.

Se-yeast: Advantages and Disadvantages

It is generally accepted that chemical and physical properties of Se and sulphur are very similar. Enzymatic systems in plants are not able to distinguish between these two elements; SeMet instead of Met was synthesized when sulphur in the incubation medium was replaced with Se (Surai, 2006). The commercial technology of production of Se-yeast is based on the aforementioned features. *Saccharomyces cerevisiae* is the main yeast strain used for aerobic fermentation in a Se-enriched medium containing beet or cane molasses, vitamins, nutritional salts, and sodium selenite as the Se source to produce Se-yeast (Esmaili and Khosravi-Darani, 2014). An important limitation to Se-yeast production is the toxicity of sodium selenite to yeast cells, which results in dramatic reductions in yeast growth. Recent results obtained by scanning electron microscopy confirmed a damaging effect of increasing concentrations of sodium selenite to yeast cells (Rajashree and Muthukumar, 2013). Therefore, industrial Se-yeast production requires a long selection process for Se-tolerant yeast strains. Theoretically, Se-yeast can accumulate up to 6,000 ppm Se; however, full replacement of Met in the yeast cells is not possible and therefore, Se concentrations in current commercial products vary between 1,000 and 3,000 ppm (Schrauzer, 2006). In many cases, industrial Se-yeast production technologies

are patented and differ between producers, which represents an additional source of variation in the composition of Se-yeast supplements found on the world market.

After harvest, Se-yeast cream is usually pasteurized and then dried. Final product characteristics include water content (5–7%), protein (40–50%), carbohydrate (11–48%), fat (2–8%), and residual ash (5–10%) (EFSA, 2008). The technology of Se-yeast manufacturing was developed more than four decades ago (for review, see Schrauzer, 2006) and today, Se-yeast is produced by a number of companies worldwide and widely used in poultry production (for review see Surai 2006; Surai and Fisinin 2014). As mentioned above, SeMet is an active component of Se-yeast, but its proportion in Se-yeast is not shown on the label, as only total Se and the inorganic proportion in Se-yeast are officially regulated.

Consequently, there are several points that remain to be addressed for commercial usage of Se-yeast. Firstly, based on modern Se-yeast production technology, it is not possible to guarantee an exact percentage of SeMet in the final product, as its composition depends on fermentation conditions, including yeast strain, source of inorganic Se, Se concentration, protocol of Se addition, base medium, energy sources (molasses), pH, temperature, shaking speed, aeration, inoculum size, and incubation time (Esmaili and Khosravi-Darani, 2014). Indeed, the Se-metabolite profile of Se-yeast is influenced by cultivation conditions (Rao *et al.*, 2010). For example, it has been recently shown that Se toxicity and Se-metabolite profile in Se-yeast are significantly affected by the amount of sulphur provided during the fermentation (Mapelli *et al.*, 2012). Secondly, analytical limitations to SeMet determination in Se-yeast further complicate this issue (Kubachka *et al.*, 2017). Furthermore, there are principal differences in composition and probably in assimilation efficacy of yeasts produced by different technologies/companies. Indeed, the fraction of water-soluble Se in various yeasts varies from 11.5% up to 28.0% (Encinar *et al.*, 2003). It has been confirmed that significant differences exist between Se-yeast products in terms of Se deposition in different cell fractions and Se bioavailability (Fagan *et al.*, 2015).

Se-yeast has been reported to contain more than 60 (Arnaudguilhem *et al.*, 2012) or even 100 (Gilbert-Lopez *et al.*, 2017) unique Se species, but only SeMet has been well researched and has been proven to be an active compound of Se-yeast. Possible roles and effects of these other Se compounds in animals/poultry await further investigation, but based on current knowledge, it can be concluded that their bio-potency and efficacy in most cases are not different from those of sodium selenite. Variability in SeMet proportion in Se-yeast is the major concern for poultry nutritionists. In fact, it would best to balance Se in the poultry diet not based on total Se, but on SeMet, which is impossible when using Se-yeast because of the aforementioned limitations in Se-yeast production and analysis. For example, Rayman (2004) reported percentages of SeMet of 84, 69, 75, 81, 83, 61, and 60 in seven different supplements. The author concluded that, despite the fact that in commercial Se-yeast products

found on the market nearly all Se is in the organic form, only 55–75% is SeMet. Indeed, great variability in SeMet content of Se-yeast products has been reported: 54–60% (Larsen *et al.*, 2004); 60–61% (Kotrebai *et al.*, 2000), 63% (EFSA, 2016), 63.4–66.6% (Burdock and Cousins, 2010), 64% (McSheehy *et al.*, 2005), 65% (Wrobel *et al.*, 2003), 73% (Wolf *et al.*, 2001), 74.8% (Yoshida *et al.*, 2002), 84% (Uden *et al.*, 2004), 85% (Ip *et al.*, 2000; Fan *et al.*, 2003). In general, the criterion for commercial Se-yeast is >60% SeMet (Bierla *et al.*, 2012). When Se-yeast samples of five leading manufacturers worldwide were analyzed (Casal *et al.*, 2010), the proportion of water-extractable Se varied between 16–35%, and the SeMet fraction also greatly varied. Recently, considerable incorporation of SeCys in proteins of the yeast proteome, despite the absence of the UGA codon, was demonstrated (Bierla *et al.*, 2013). The authors showed that 10–15% of Se present in Se-yeast is SeCys. This means that if all the Se in Se-yeast is accounted for, the maximum SeMet fraction would not exceed 85%, but in many cases will be much lower than that. In accordance herewith, in a batch production analysis, SeMet in the product varied between 58–69%, and SeCys accounted for 15–22% (EFSA, 2016). However, dietary SeCys, similar to sodium selenite, is not effective in increasing the tissue Se concentration and building Se reserves in monogastric animals as shown in rats by Deagen *et al.* (1987). Using ICP-MS, the UT2A Lab in France analyzed 12 Se-yeast products of eight commercial brands, collected from end users in the Asia-Pacific region. The results revealed that Se-yeast products largely vary in terms of total Se and SeMet contents. For a number of samples, SeMet accounted for approximately 60% of total Se. However, certain products showed extremely high variation in or even lacked SeMet, being a simple mixture of yeast and mineral selenium (Liu *et al.*, 2017). Recently, a range of commercially available Se supplements was evaluated, and discrepancies between labeled ingredients and detected species were noted (Kubachka *et al.*, 2017).

It should be noted that absorption of dietary Se (organic Se) is generally believed to be good (~80%; Reilly, 2006). Absorption and retention of Se from Se-yeast, measured in 12 volunteers fed ⁷⁷Se-labeled Se-yeast was shown to be 75–90%; however, other Se-yeasts gave different results (50–60%) (Sloth *et al.*, 2003). Intestinal absorption of SeMet in women was shown to be 95.7–97.3% of the administered dose (Griffiths *et al.*, 1976). The fact that Se-yeast efficacy is determined by the fraction of SeMet was proven in a study in broilers using two commercially available Se-yeast preparations, containing SeMet at 63% and 56.7% (Simon *et al.*, 2013). Se content in the muscles of broilers fed a diet supplemented with sodium selenite was 133 µg/kg, while that in broilers fed a diet supplemented with Se-yeast containing 26% or 69% SeMet was 161 and 267 µg/kg Se, respectively, indicating that high SeMet content results in higher Se deposition in muscle tissues (Van Beirendonck *et al.*, 2016a, b). Similar results have been reported by Liu *et al.* (2017), who compared two Se-yeast products with significantly different SeMet content (74% vs. 33%). In brief, Se accu-

mulation in the muscles is proportional to the level of SeMet consumption through diet.

SeMet and Zn-SeMet

SeMet exists in various forms, including the naturally occurring L-SeMet and two synthetic forms, D-SeMet and DL-SeMet (Schrauzer, 2003). Pure SeMet can also be used as a dietary supplement (Schrauzer 2000, 2001, 2003; Schrauzer and Surai, 2009). There is a range of publications showing the beneficial effects of SeMet in poultry diets. For example, dietary L-SeMet is more effectively transferred to the egg than Se-yeast (Delezie *et al.*, 2014). It seems likely that both L-SeMet and D-SeMet can be used as Se sources for poultry with better efficacy than sodium selenite (Wang *et al.*, 2011). However, L-SeMet is more effective in improving antioxidant defenses in chickens as well as average daily gain and feed conversion ratio (FCR) than D-SeMet. On the other hand, a study analyzing major indexes of growth, development, and antioxidant defense system in plasma and breast muscles in broilers supplemented with either sodium selenite or DL-SeMet at 0.15 mg/kg showed that DL-SeMet was not different from sodium selenite in terms of daily gain, FCR, total antioxidant activity, GSH-Px activity (Jiang *et al.*, 2009). Only total superoxide dismutase activity and GSH levels in breast muscle were significantly upregulated upon organic Se supplementation (Jiang *et al.*, 2009). DL-SeMet was more effective than sodium selenite in improving chicken immunity (Wang *et al.*, 2016). More recently, DL-SeMet in the chicken diet was shown to more strongly increase GSH-Px and total antioxidant activity in thigh muscle and liver than sodium selenite (Bakhshalinejad *et al.*, 2017). There is a need for more studies comparing the efficiencies of L-, D-, DL-SeMet and other forms of organic Se in poultry diets.

Unfortunately, scientific information on Zn-SeMet is quite limited and this product is not registered for use in the EU, although it is available on other markets, including Asian countries. This product is sometimes called “chelated selenium,” but in fact, the Zn is chelated by SeMet. Based on the chemical structure of the product, it seems likely that it does not differ from pure SeMet in terms of stability, though its absorption efficiency might be lower (Geraert *et al.*, 2015). Inclusion of Zn-SeMet into the laying hen diet at 0.2 and 0.4 mg/kg was shown to have positive effects on egg production and quality in comparison to non-supplemented layers, and GSH-Px activity in layer plasma was increased only at 0.2 ppm Zn-SeMet supplementation (Laika and Jahanian, 2015). When Zn-SeMet was compared to sodium selenite at 0.4 ppm supplementation in breeder diet, the reproductive performance of birds in both groups was the same. The only significant difference induced by Zn-SeMet was a heavier weight from hatchling until egg production peak (33 weeks) (Urso *et al.*, 2015). Similarly, Se source (sodium selenite or Zn-SeMet at 0.3 mg/kg) in layer diets did not affect egg production and quality parameters (FCR, egg weight, Haugh units, and eggshell thickness). Replacement of sodium selenite in the layer diet with Zn-SeMet was associated with an

increase in the Se level in the egg albumin (1.92 vs. 1.35 mg/kg), but did not affect Se level in the egg yolk (Chantiratikul *et al.*, 2008). SeMet and Se-yeast efficacies have been compared in dogs using six biochemical markers, including intraprostatic dihydrotestosterone, testosterone, dihydrotestosterone:testosterone ratio, and epithelial cell DNA damage, proliferation, and apoptosis (Waters *et al.*, 2012). By comparing dogs that achieved equivalent intraprostatic Se concentration, it was concluded that there was no significant difference in potency of the tested Se supplements. Indeed, SeMet is the main active ingredient of Se-yeast.

However, SeMet in purified form can be easily oxidized. For example, in freeze-dried oyster samples, total Se and the Se species detected were shown to be stable for at least 12 months; however, after purification of Se species, SeMet in the enzymatic extracts is only stable for 10 days if stored at 4°C in Pyrex containers (Moreno *et al.*, 2002). SeMet is also partially (~20%) lost after storage in water solution for 30 days at 20°C (Lindemann *et al.*, 2000). SeMet was identified as the major compound in gastrointestinal extract and its oxidation product (selenoxide, SeMetO) was the main degradation product formed after medium and long-term sample storage, respectively (Reyes *et al.*, 2006). Indeed, pure SeMet is chemically oxidized to SeMetO in the small intestine (Lavu *et al.*, 2016). According to a report by the EFSA (2013a), there is a problem with the stability of SeMet in premixtures containing trace elements; SeMet recovery from premixes after 3, 6, and 9 months of storage was 55%, 54%, and 37%, respectively.

OH-SeMet

Recently, a new organic Se source, OH-SeMet (2-hydroxy-4-methylselenobutanoic acid or HMSeBA), has been developed (Briens *et al.*, 2013, 2014). It is a precursor of SeMet and after its dietary consumption, OH-SeMet is easily converted into SeMet and metabolized in the same way as pure SeMet, including building Se reserves in muscles. OH-SeMet is synthesized in seven steps, and the solid form of the product is manufactured by spraying a 40% solution of OH-SeMet on a carrier (e.g. silica) and mixing. The liquid form of additive is prepared by dissolving solid OH-SeMet in distilled water (5% solution; EFSA, 2013b). In commercial products, OH-SeMet is shown to be more stable than pure SeMet (Geraert *et al.*, 2015). It seems likely that OH-SeMet during storage behaves like the hydroxylated analog of Met, which is a stable product.

In a comparison of the effect of OH-SeMet with that of sodium selenite and Se-yeast, all three Se sources increased muscle Se compared with control chickens, with a significant ($P < 0.05$) source effect in the order OH-SeMet > Se-yeast > sodium selenite and the effect of OH-SeMet being 1.5-fold stronger than that of Se-yeast (Briens *et al.*, 2014). Selenium speciation in tissues indicates that muscle Se is exclusively present as SeMet or SeCys, indicating a full conversion of OH-SeMet to active Se species in the bird. These results corroborate the higher bioavailability of organic Se than that of mineral Se. Based on muscle Se levels, it is clear that OH-

SeMet is characterised by a better bio-efficiency than Se-yeast; OH-SeMet was 39% more available than Se-yeast. It seems likely that the higher SeMet content in OH-SeMet supplement (nearly 100% in OH-SeMet vs. 60–70% in Se-yeast) is responsible for the aforementioned differences. Interestingly, OH-SeMet in chicken diet increased the breast-muscle Se concentration compared to hatch value at day 21, while Se-yeast supplement only maintained the original Se concentration and sodium selenite significantly decreased the muscle Se concentration (Couloigner *et al.*, 2015). Furthermore, OH-SeMet in turkey diet improved GSH-Px activity in thigh muscles and decreased lipid peroxidation (Briens *et al.*, 2016). In addition, compared with sodium selenite or Se-yeast, OH-SeMet demonstrated an increased efficacy to enrich SeMet and total Se depositions, to induce mRNA expression of selenoprotein S and MsrB, thioredoxin reductase activity, and protein expression of GSH-Px4, selenoprotein P, and selenoprotein U in the tissues of chicks (Zhao *et al.*, 2017). Similar to growing chickens, breeders would benefit from OH-SeMet supplementation, as laying hens fed a diet with 0.2 ppm Se as OH-SeMet transferred more Se to their eggs (+28.8%) and built bigger Se reserves in muscles (+28%) than birds fed the diet supplemented with 0.2 ppm Se in the Se-yeast form (Jlali *et al.*, 2013).

Recently, in the EU, supplementation with Se-yeast has been limited to 0.2 mg Se/kg complete feed for reasons of consumer safety (Commission Implementing Regulation No. 427/2013 of 8 May 2013). Taking into account research data on the effects of organic Se on breeders, which are mainly related to dietary supplementation at 0.3 ppm, and previous commercial usage of such levels of Se supplementation, commercial premix and feed companies are looking for the most effective sources of organic Se. The idea is to deliver the same amount of Se to the egg and build the same Se reserves in muscles using 0.2 ppm Se dietary supplementation as can be achieved with 0.3 ppm in the form of Se-yeast. Indeed, the aforementioned results clearly showed that OH-SeMet supplied in the same dose as Se-yeast in the chicken diet would provide additional benefit in terms of Se reserves in the muscles as well as Se transfer to the egg and probably, to the developing embryo. This improvement in Se status could potentially be translated into higher selenoprotein expression and better antioxidant protection in stressful conditions of commercial poultry production (Zhao *et al.*, 2017; Lei *et al.*, 2017).

Se-homolanthionine (SeHLan)

A specific Se-yeast is produced based on *Torula* yeast (*Cyberlindnera jadinii*). Its specificity lies in the fact that it contains mainly SeHLan. However, it is not clear at present how SeHLan functions in the body. When pure SeMet and SeHLan were compared in a study in mice, SeHLan was significantly less effective in liver Se accumulation than SeMet (Anan and Ogra, 2013). Indeed, SeHLan can be metabolized in poultry/animals similar to SeMet, but it cannot build Se reserves in the body (Anan *et al.*, 2011) as only SeMet can be non-specifically incorporated into body

proteins without altering their functions (Schrauzer and Surai, 2009).

Chelated Se Products

There is a range of products on the market that are claimed to contain chelated Se (including Se-glycinates, Se-proteinates, and Se-amino acid complexes); however, the chemical position of Se in the periodic table of elements indicates that Se is not a true metal, more a metalloid, and therefore, its chelating ability is questioned. Indeed, attempts to detect chelated Se in such products resulted in the detection of only inorganic Se (selenite and selenate) (Amoako *et al.*, 2009; Kubachka *et al.*, 2017). Recent bioassay results confirmed that those products (glycinates and proteinates) have a similar efficacy as sodium selenite (Liu *et al.*, 2017). Similarly, Givens *et al.* (2004) showed that a chelated selenium–amino acid complex in cow diet was not different from sodium selenite when the efficacy of Se transfer to the milk was assessed. Indeed, chelated Se products are not related to SeMet and should probably not be included in the organic Se category.

Selenium Nanoparticle (nano-Se) Products

Recently, nano-Se have received substantial attention as potential novel nutritional supplements because of their low toxicity and novel characteristics, such as great specific surface area, high surface activity, and high catalytic efficiency (Pelyhe and Mezes, 2013; Sarkar *et al.*, 2015; Wadhvani *et al.*, 2016; Skalickova *et al.*, 2017). In addition, it has been suggested that nano-Se can have important applications in the field of medicine owing to their antibacterial and anticancer properties (Wang *et al.*, 2007; Ahmed *et al.*, 2014; Wadhvani *et al.*, 2016). Nano-Se has an ability to increase selenoenzyme activity comparable to that of SeMet. However, the question remains how elemental Se can be involved in SeCys synthesis and selenoprotein expression. There are several reports of successful testing of nano-Se in broiler nutrition (Wang, 2009; Zhou and Wang, 2011; Cai *et al.*, 2012; Gulyas *et al.*, 2017). In many cases, its low Se toxicity is considered the main advantage of nano-Se. However, one should realize that Se

toxicity is not a major problem in poultry industry and Se in the form of sodium selenite or organic Se (SeMet, Se-yeast or other preparations) is an essential part of premixes produced worldwide. While it seems likely that nano-Se could be a new chemopreventive agent for the treatment of various diseases (Zhang *et al.*, 2008), including cancer, its value as a feed supplement for the poultry industry is questionable.

Conclusion

Se is an essential element in animal/poultry nutrition and its dietary supplementation in an optimal form and amount is key to maintaining animal health, productive, and reproductive performance. It has been proven that organic Se (mainly SeMet) in animal/poultry diets has a range of advantages over traditional sodium selenite. In fact, the organic Se concept considers SeMet as a storage form of Se in the chicken body. As animals/poultry are not able to synthesize SeMet, its provision through diet is a key strategy to fight commercially relevant stresses. Indeed, in stress conditions, when increased selenoprotein expression requires additional Se, but its provision with feed is decreased because of reduced feed consumption, Se reserves in the body (mainly in muscles) help maintain an effective antioxidant defense. It is well appreciated that via selenoprotein expression, Se is involved in the protection against oxidative stress and regulation of cell growth, apoptosis, and modification of cell signaling systems and transcription factors (Fig. 4). Therefore, its adequate dietary supply is crucial for various physiological processes in the animal/chicken body. Furthermore, it seems likely that increased muscle reserves of Se may enhance chicken resistance to stress and disease.

The poultry industry is looking for the most effective sources of organic Se for commercial use. Several selenium compounds have been authorized as feed additives in the EU, including sodium selenite, sodium selenate, Se-yeast, L-SeMet, DL-SeMet, and OH-SeMet. In accordance with EU regulations, the total dietary Se is limited to 0.5 mg/kg and the addition of the aforementioned organic Se compounds is limited to 0.2 mg Se/kg (EFSA, 2014). Se-yeast has received substantial attention and is widely used in commercial products as a reliable source of organic Se. Generally speak-

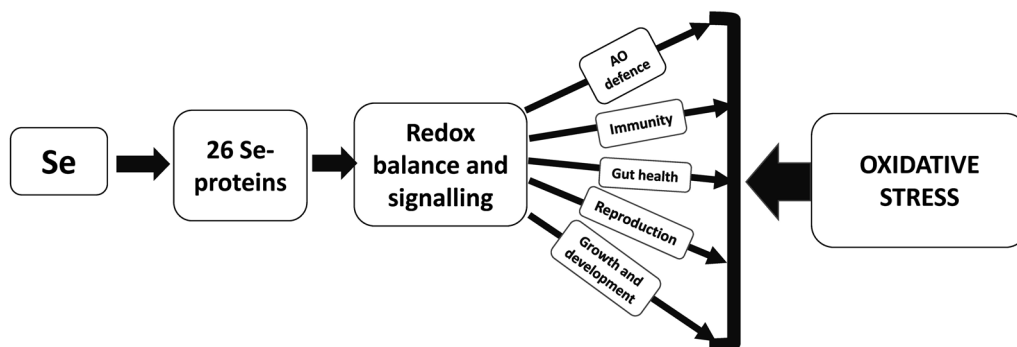


Fig. 4. Protective effects of Se in poultry.

ing, there are three generations of Se supplements on the worldwide market. The first generation includes inorganic Se sources, mainly selenite and selenate. We would consider this generation of Se supplements as outdated. They are on the market since 1970 and helped animal and poultry industry get rid of real Se deficiency, but with today's precision nutrition concept, they have limited value and their only advantage is their comparatively low price. The second generation of Se supplements for poultry includes Se-yeast, SeMet, and Zn-SeMet. Usage of such Se supplements revolutionized Se nutrition over the last 20 years. However, a main disadvantage of Se-yeast is that the level of the active compound (SeMet) is quite variable (60% on average). There is no proof that the remaining fraction (30–50%) of Se, e.g., in the form of SeCys or MeSeCys, has any additional benefits to poultry in comparison to sodium selenite. From the production point of view, it is difficult to guarantee a certain percentage of SeMet in Se-yeast, as a range of production-related factors, including yeast strain, medium composition, selenite concentration, and temperature, affect the composition of the end product. Analytical difficulties in precisely determining the SeMet content in yeast-based products further complicate the issue. The main disadvantage of pure SeMet is its instability when included in pre-mixes and feeds. The third generation of Se supplements for poultry is represented by OH-SeMet, which combined advantages of both Se-yeast (stability) and Se-Met (>95% of active Se compound, i.e., SeMet). Clearly, there is a need for more research devoted to organic Se sources in poultry and farm animal nutrition.

There is a need for more research devoted to comparative studies of different forms of organic Se in poultry/animal nutrition to better understand the advantages and limitations of each form. In addition, there is a need for further clarification in which form Se is present in products such as Se-chelates, Se-proteinates, Se-glycinates to place them in a proper category of Se supplements. Indeed, Se-homolanthionine is commercially available but awaits more research to explain the molecular mechanisms of its metabolization and possible advantages to inorganic forms of Se. A quickly developing area of nano-Se research and application urgently needs well-designed studies with usage of modern molecular techniques to explain how metallic nano-Se particles are metabolized in the chicken/animal/human body to become a source of Se for synthesis of selenoproteins, which represent biological activity involving main active Se forms in biological systems.

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